

Supplementary materials

Microwave-Assisted Synthesis of Pt/SnO₂ for the Catalytic Reduction of 4-Nitrophenol to 4-Aminophenol

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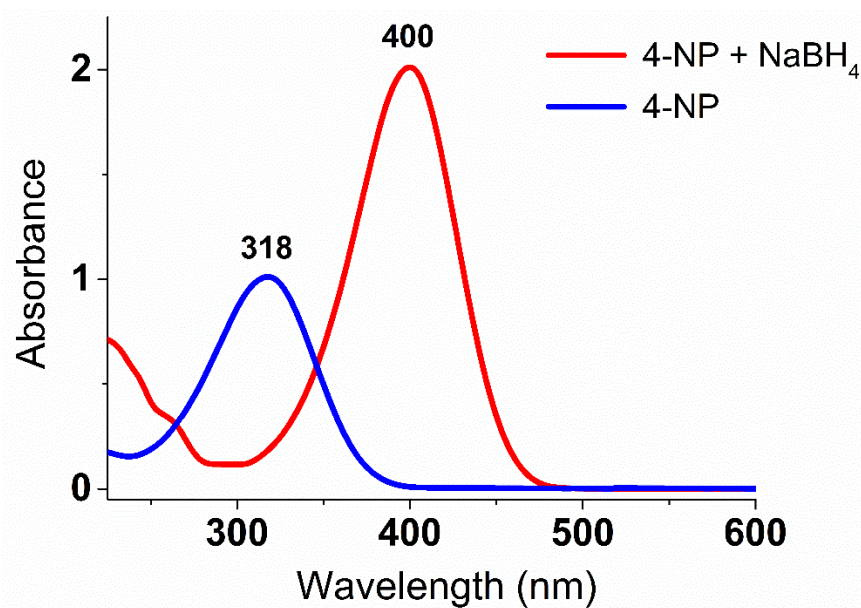


Fig. S1 UV-Vis spectra of the aqueous solution of pure 4-nitrophenol with the maximum at 318 nm and 4-nitrophenolate ions after addition of NaBH₄ (alkaline aqueous solution) with maximum at 400 nm.

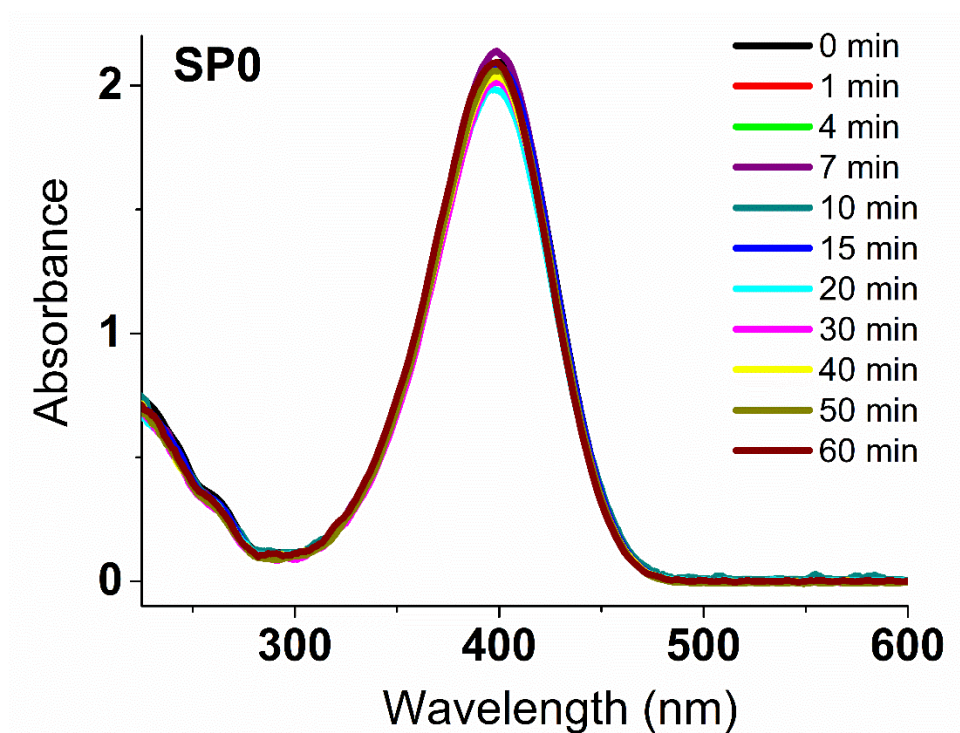


Fig. S2 Catalytic reduction of 4-nitrophenol (4-NP) to 4-aminophenol (4-AP) as a function of time using sample SP0, which does not contain platinum. This sample is completely inactive for the reduction of 4-NP to 4-AP.

Table S1 The peak positions and relative proportions (%) of Pt⁴⁺, Pt²⁺, Pt⁰, Sn⁴⁺ and Sn²⁺ in the synthesized samples based on the deconvoluted Pt 4f and Sn 3d spectra.

Sample	Electron configuration	Binding energy / eV	Pt / %	Electron configuration	Binding energy / eV	Sn / %
SP1	Pt ⁴⁺ 4f _{5/2}	—	Pt ⁴⁺ / 0	Sn ⁴⁺ 3d _{3/2}	495.2	Sn ⁴⁺ / 96.6
	Pt ⁴⁺ 4f _{7/2}	—		Sn ⁴⁺ 3d _{5/2}	486.8	
	Pt ²⁺ 4f _{5/2}	75.7	Pt ²⁺ / 36.8	Sn ²⁺ 3d _{3/2}	493.8	Sn ²⁺ / 3.4
	Pt ²⁺ 4f _{7/2}	72.5		Sn ²⁺ 3d _{5/2}	485.4	
	Pt ⁰ 4f _{5/2}	74.0	Pt ⁰ / 63.2	Sn ⁰ 3d _{3/2}	—	Sn ⁰ / 0
	Pt ⁰ 4f _{7/2}	70.8		Sn ⁰ 3d _{5/2}	—	
	Pt 4f	eV	Pt / %	Sn 3d	eV	Sn / %
SP3	Pt ⁴⁺ 4f _{5/2}	—	Pt ⁴⁺ / 0	Sn ⁴⁺ 3d _{3/2}	495.1	Sn ⁴⁺ / 97.1
	Pt ⁴⁺ 4f _{7/2}	—		Sn ⁴⁺ 3d _{5/2}	486.6	
	Pt ²⁺ 4f _{5/2}	75.7	Pt ²⁺ / 19.9	Sn ²⁺ 3d _{3/2}	493.6	Sn ²⁺ / 2.9
	Pt ²⁺ 4f _{7/2}	72.4		Sn ²⁺ 3d _{5/2}	485.2	
	Pt ⁰ 4f _{5/2}	74.0	Pt ⁰ / 80.1	Sn ⁰ 3d _{3/2}	—	Sn ⁰ / 0
	Pt ⁰ 4f _{7/2}	70.8		Sn ⁰ 3d _{5/2}	—	
	Pt 4f	eV	Pt / %	Sn 3d	eV	Sn / %
SP5	Pt ⁴⁺ 4f _{5/2}	77.5	Pt ⁴⁺ / 7.8	Sn ⁴⁺ 3d _{3/2}	495.2	Sn ⁴⁺ / 96.1
	Pt ⁴⁺ 4f _{7/2}	74.3		Sn ⁴⁺ 3d _{5/2}	486.8	
	Pt ²⁺ 4f _{5/2}	75.8	Pt ²⁺ / 24.6	Sn ²⁺ 3d _{3/2}	493.6	Sn ²⁺ / 3.9
	Pt ²⁺ 4f _{7/2}	72.5		Sn ²⁺ 3d _{5/2}	485.2	
	Pt ⁰ 4f _{5/2}	74.1	Pt ⁰ / 67.6	Sn ⁰ 3d _{3/2}	—	Sn ⁰ / 0
	Pt ⁰ 4f _{7/2}	70.8		Sn ⁰ 3d _{5/2}	—	
	Pt 4f	eV	Pt / %	Sn 3d	eV	Sn / %
SP10	Pt ⁴⁺ 4f _{5/2}	77.5	Pt ⁴⁺ / 8.3	Sn ⁴⁺ 3d _{3/2}	495.1	Sn ⁴⁺ / 97.1
	Pt ⁴⁺ 4f _{7/2}	74.3		Sn ⁴⁺ 3d _{5/2}	486.7	
	Pt ²⁺ 4f _{5/2}	75.8	Pt ²⁺ / 25.8	Sn ²⁺ 3d _{3/2}	493.7	Sn ²⁺ / 2.9
	Pt ²⁺ 4f _{7/2}	72.6		Sn ²⁺ 3d _{5/2}	485.3	
	Pt ⁰ 4f _{5/2}	74.2	Pt ⁰ / 65.9	Sn ⁰ 3d _{3/2}	—	Sn ⁰ / 0
	Pt ⁰ 4f _{7/2}	70.9		Sn ⁰ 3d _{5/2}	—	
	Pt 4f	eV	Pt / %	Sn 3d	eV	Sn / %
SP15	Pt ⁴⁺ 4f _{5/2}	77.5	Pt ⁴⁺ / 16.3	Sn ⁴⁺ 3d _{3/2}	495.3	Sn ⁴⁺ / 92.2
	Pt ⁴⁺ 4f _{7/2}	74.3		Sn ⁴⁺ 3d _{5/2}	486.8	
	Pt ²⁺ 4f _{5/2}	75.7	Pt ²⁺ / 29.8	Sn ²⁺ 3d _{3/2}	493.4	Sn ²⁺ / 7.8
	Pt ²⁺ 4f _{7/2}	72.5		Sn ²⁺ 3d _{5/2}	485.0	
	Pt ⁰ 4f _{5/2}	74.1	Pt ⁰ / 53.9	Sn ⁰ 3d _{3/2}	—	Sn ⁰ / 0
	Pt ⁰ 4f _{7/2}	70.8		Sn ⁰ 3d _{5/2}	—	

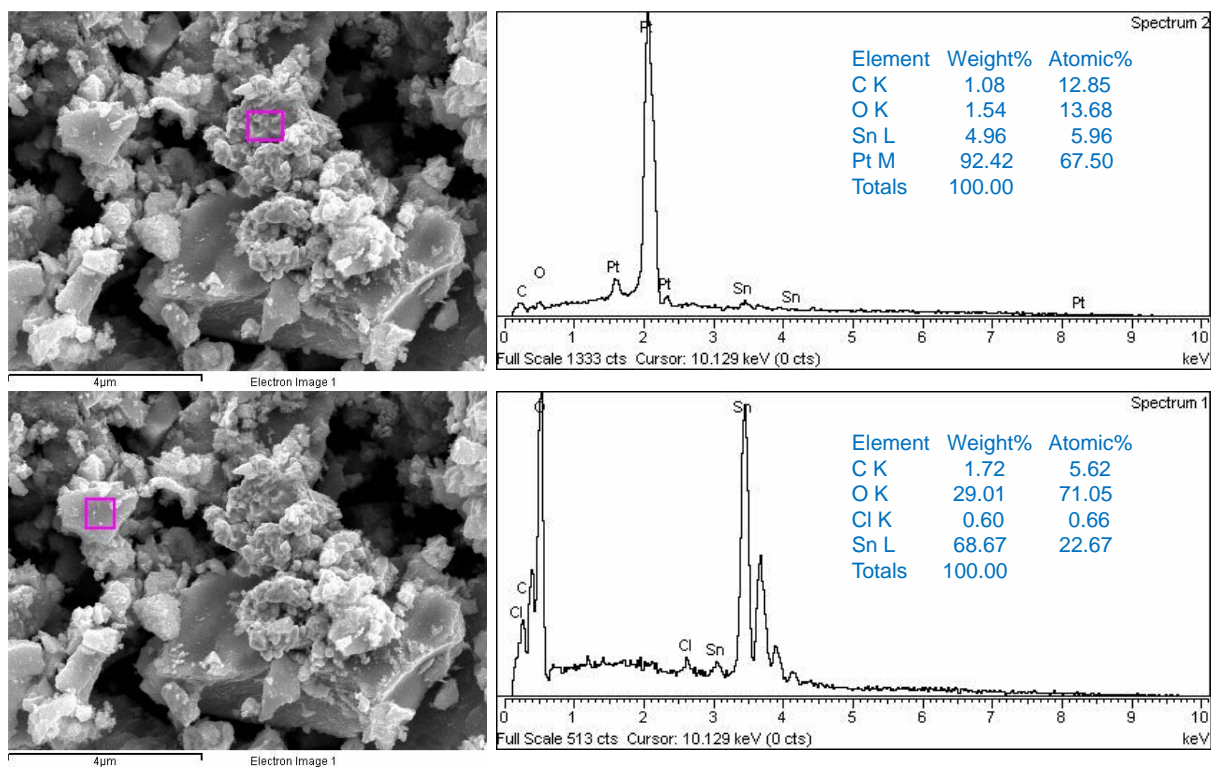


Fig. S3 SEM EDS results of sample SP3. EDS analyses were taken from two different marked sites showing the platinum rich region (upper panel) and SnO₂ rich region (lower panel).

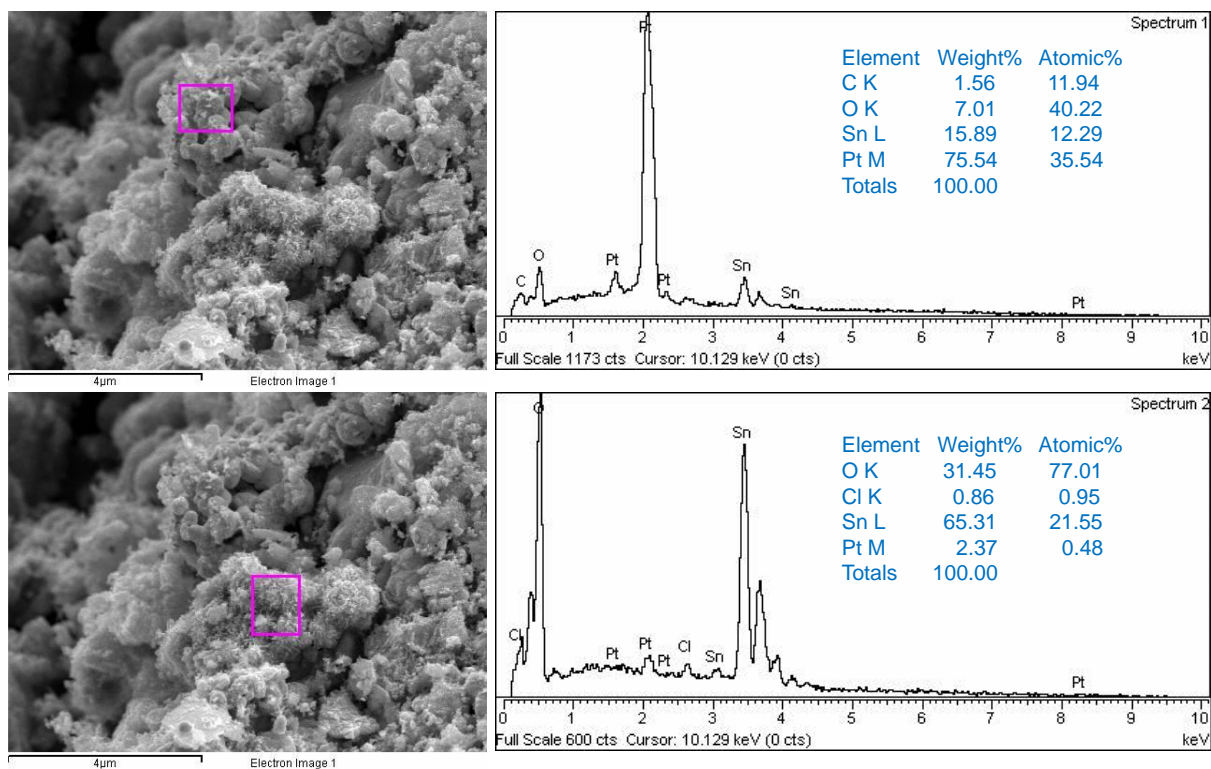


Fig. S4 SEM EDS results of sample SP5. EDS analyses were taken from two different marked sites showing the platinum rich region (upper panel) and SnO₂ rich region (lower panel).

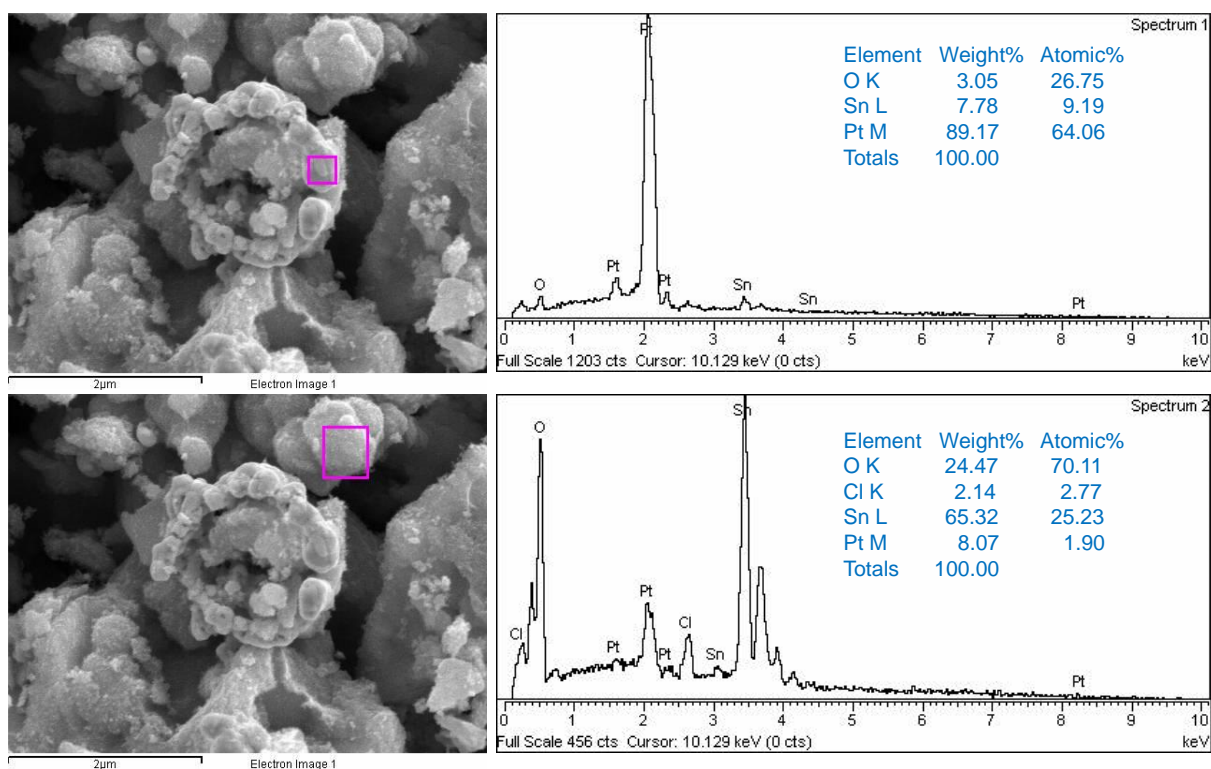


Fig. S5 SEM EDS results of sample SP10. EDS analyses were taken from two different marked sites showing the platinum rich region (upper panel) and SnO₂ rich region (lower panel).

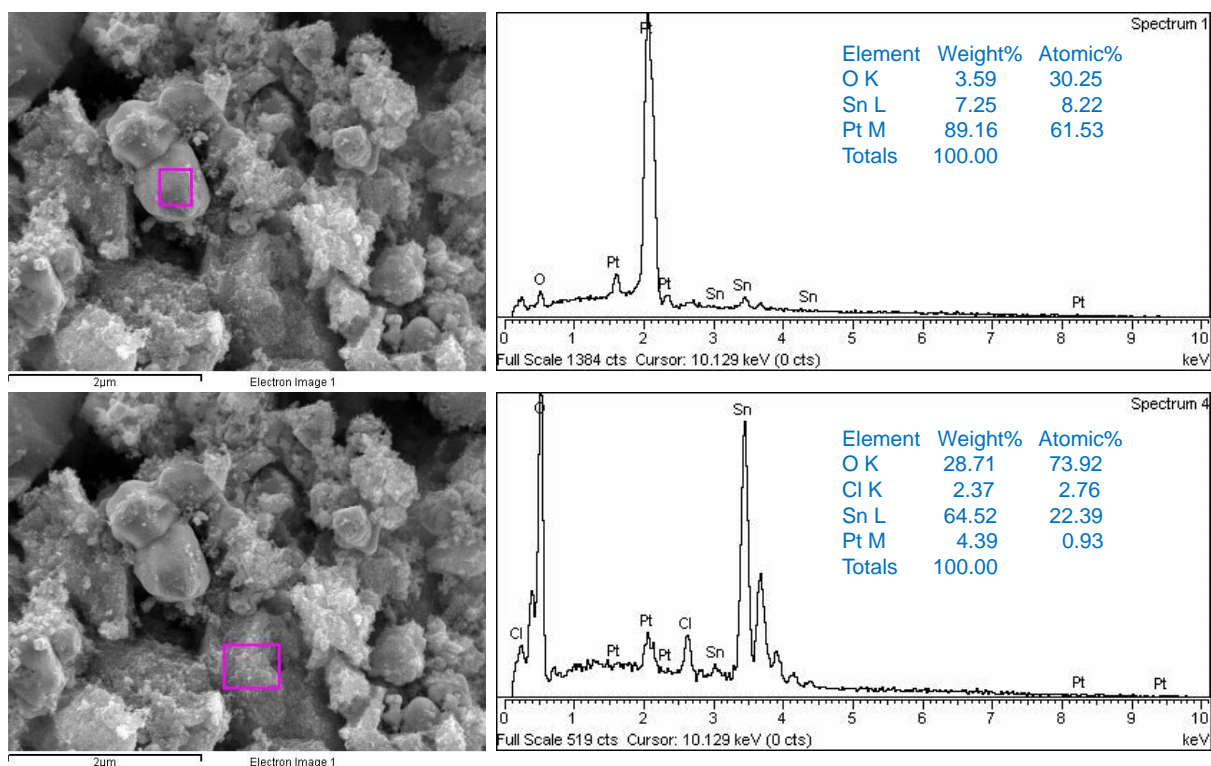


Fig. S6 SEM EDS results of sample SP15. EDS analyses were taken from two different marked sites showing the platinum rich region (upper panel) and SnO₂ rich region (lower panel).

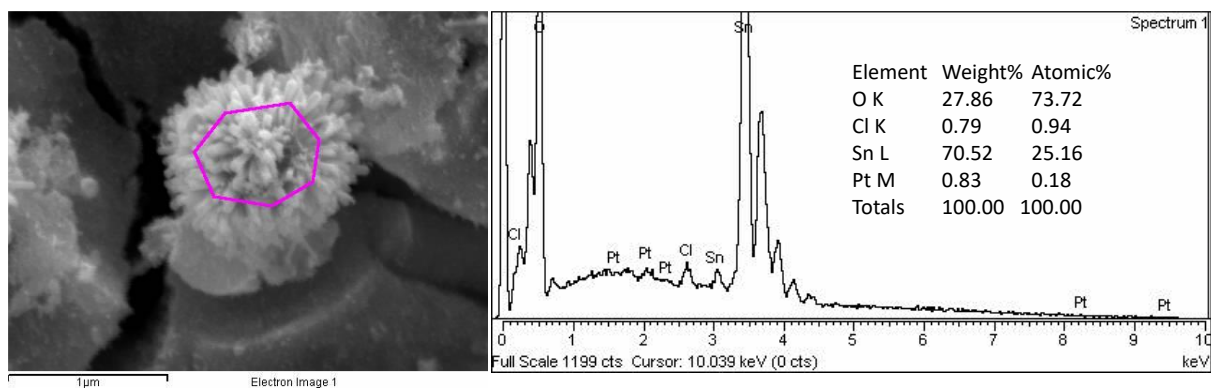


Fig. S7 SEM EDS results of urchin-like particle in sample SP5. The image shows anisotropic urchin-like particles. EDS analyses taken from the marked sites showing that the urchin-like particle mainly consisted of SnO_2 .