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# **Physical Chemistry of Aqueous Solutions and Glass Forming Systems**

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## **Message from the Guest Editors**

The thermodynamical behaviors of aqueous solutions resemble those of glass-forming systems and can be studied in the same theoretical framework, for example, that of the mode coupling theory. In fact, the hydrogen bonding ability of water, which is progressively enhanced by lowering the temperature, allows the formation of local clusters and dynamical heterogeneities, as in glass-forming systems. This holds also for aqueous solutions, such as hydrated proteins or water/alcohol mixtures. Thus, the importance of establishing a rigorous picture for these systems is at the borderline among physics, chemistry, biology, technology, and life science.

This Special Issue aims to cover recent advances in the experiments, theoretical modeling, and simulations within this area and toward nanotechnologies. Water, in fact, is the medium par excellence for the "development" of nanosystems, mainly polymers, with both hydrophobic and hydrophilic moieties showing competing properties.













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# **Message from the Editor-in-Chief**

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