



Metallothermic Reactions

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Message from the Guest Editor

Dear Colleagues,

Metallothermic reactions are the basis of most metallurgical processes, during which the redox interaction characterized by heat release due to exothermic reactions is realized. This class of reactions can be used to create energy-saturated systems and to produce materials for various purposes: ceramics, heat-resistant alloys, and functional gradient materials.

The practical aim of such metallothermic reactions has a wide scope, from synthesis of pure materials from systems with high oxophilicity (Ti, Ta, Nd, etc.) to processing of recycled materials in the metallurgical industry, including the solution of environmental problems and obtaining of catalyst precursors. Topics of interest include, but are not limited to, the following:

- Reaction control methods: imposition of gravitational and electromagnetic fields.
- Experimental research methods and mathematical modeling based on approaches of mechanics of multiphase, multicomponent, chemically active continuous media.
- Controlling thermodynamic parameters (pressure and temperature).





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Message from the Editorial Board

Metallic materials play a vital role in the economic life of modern societies; contributions are sought on fresh developments that enhance our understanding of the fundamental aspects related to the relationships between processing, properties and microstructure – disciplines in the metallurgical field ranging from processing, mechanical behavior, phase transitions and microstructural evolution, nanostructures, as well as unique metallic properties – inspire general and scholarly interest among the scientific community.

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