

## **Supplementary material**

# **Montmorillonite-Based Natural Adsorbent from Colombia for the Removal of Organic Pollutants from Water: Isotherms, Kinetics, Nature of Pollutants, and Matrix Effects**

**Marcela Paredes-Laverde <sup>1,2</sup>, Diego F. Montaño <sup>3</sup> and Ricardo A. Torres-Palma <sup>1,\*</sup>**

<sup>1</sup> Grupo de Investigación en Remediación Ambiental y Biocatálisis (GIRAB), Instituto de Química, Facultad de Ciencias Exactas y Naturales, Universidad de Antioquia UdeA, Calle 70 No. 52-21, Medellín 50011, Colombia;

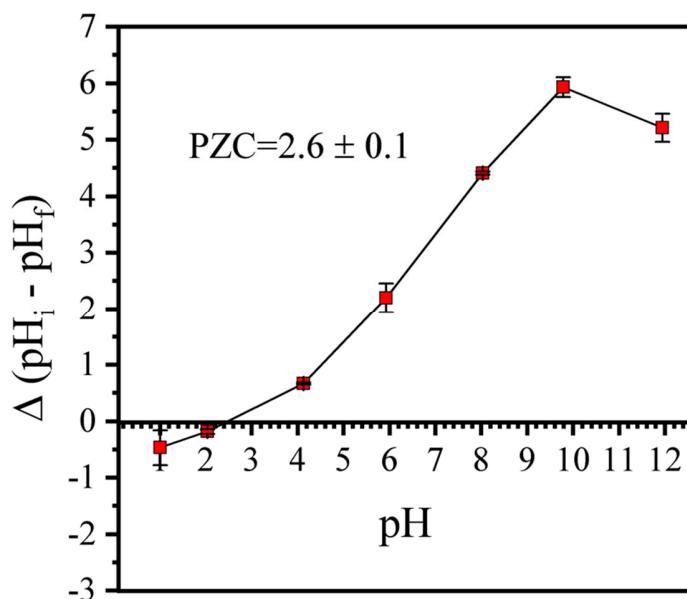
<sup>2</sup> Grupo de Investigación Cuidados de la Salud e Imágenes Diagnósticas, Facultad de Ciencias de la Salud, Fundación Universitaria Navarra—Uninavarra, Calle 10 No. 6-41, Neiva 31008, Colombia

<sup>3</sup> Grupo de Investigación Ciencia y Diseño de Materiales (CiDiMat), Departamento de Química, Facultad de Ciencias Básicas, Universidad de Pamplona, Km 1 Vía Bucaramanga Ciudad Universitaria, Calle 5 No. 3-93, Pamplona 26909, Colombia;

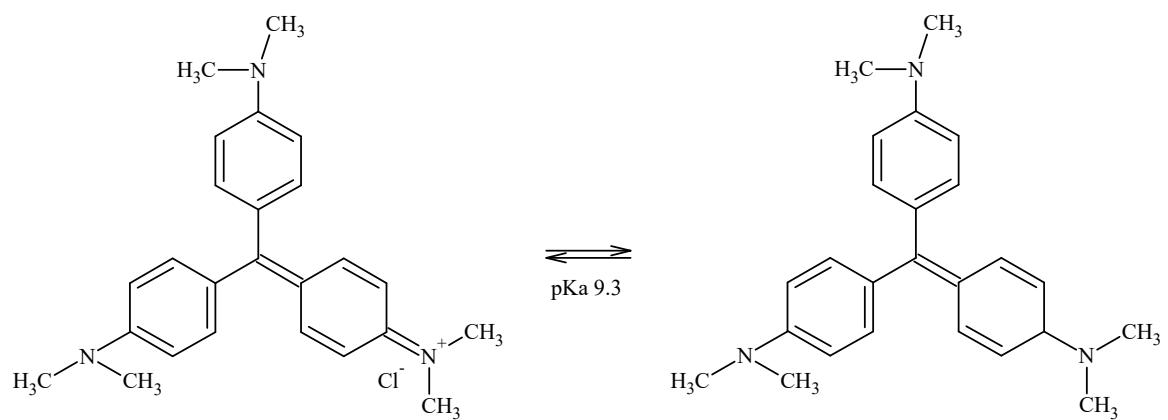
\* Correspondence: ricardo.torres@udea.edu.co; Tel.: +57-315-314-98-76

**Table S1.** Chemical composition of fresh urine and textile wastewater.

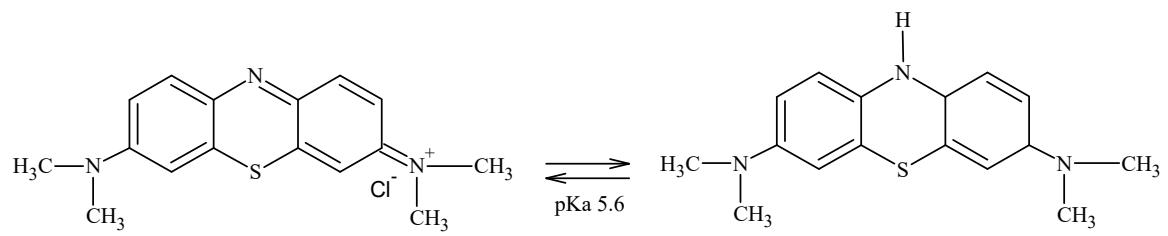
Fresh Urine		Textile wastewater	
Concentration [mg L <sup>-1</sup> ]		Concentration [mg L <sup>-1</sup> ]	
Urea	16000	Dye	Required concentration
Na <sub>2</sub> SO <sub>4</sub>	2300	NaCl	1500
NH <sub>4</sub> Cl	1800	Na <sub>2</sub> CO <sub>3</sub>	500
NaH <sub>2</sub> PO <sub>4</sub>	2900	NaHCO <sub>3</sub>	500
KCl	4200	NaOH	500
MgCl <sub>2</sub> •6H <sub>2</sub> O	790	H <sub>2</sub> SO <sub>4</sub>	800
CaCl <sub>2</sub> •2H <sub>2</sub> O	680	Starch	500
NaOH	120		



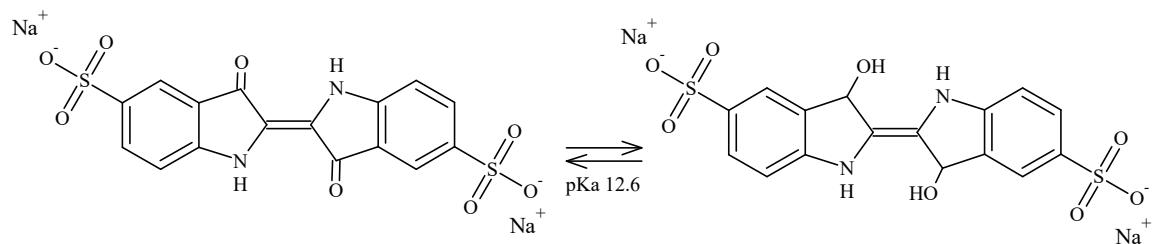
**Figure S1.** Determination of PZC value of MMT



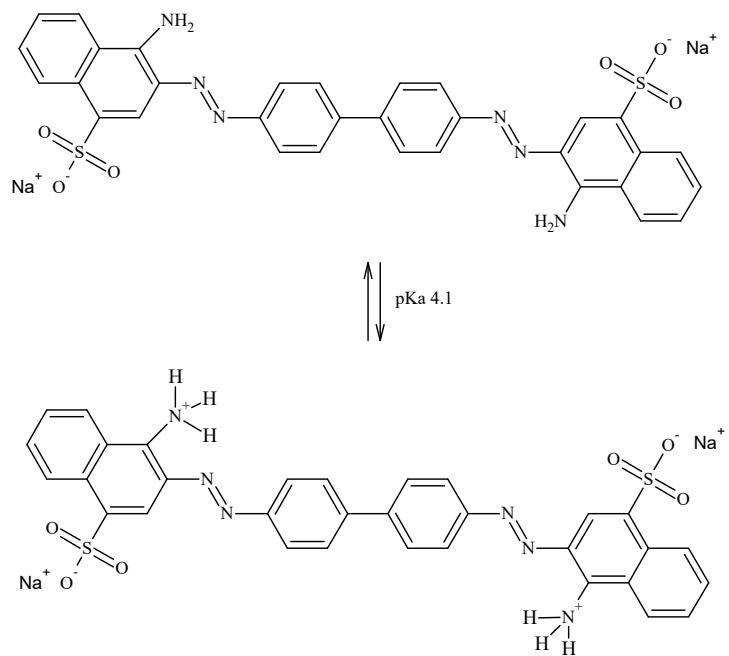
**Figure S2.** Structure CV



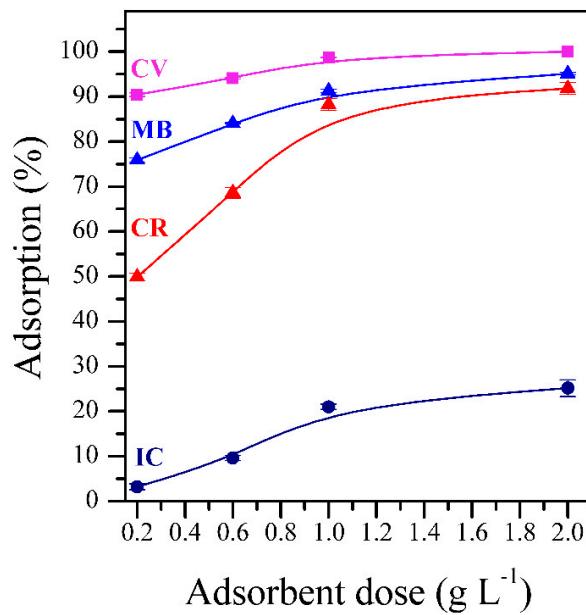
**Figure S3.** Structure MB



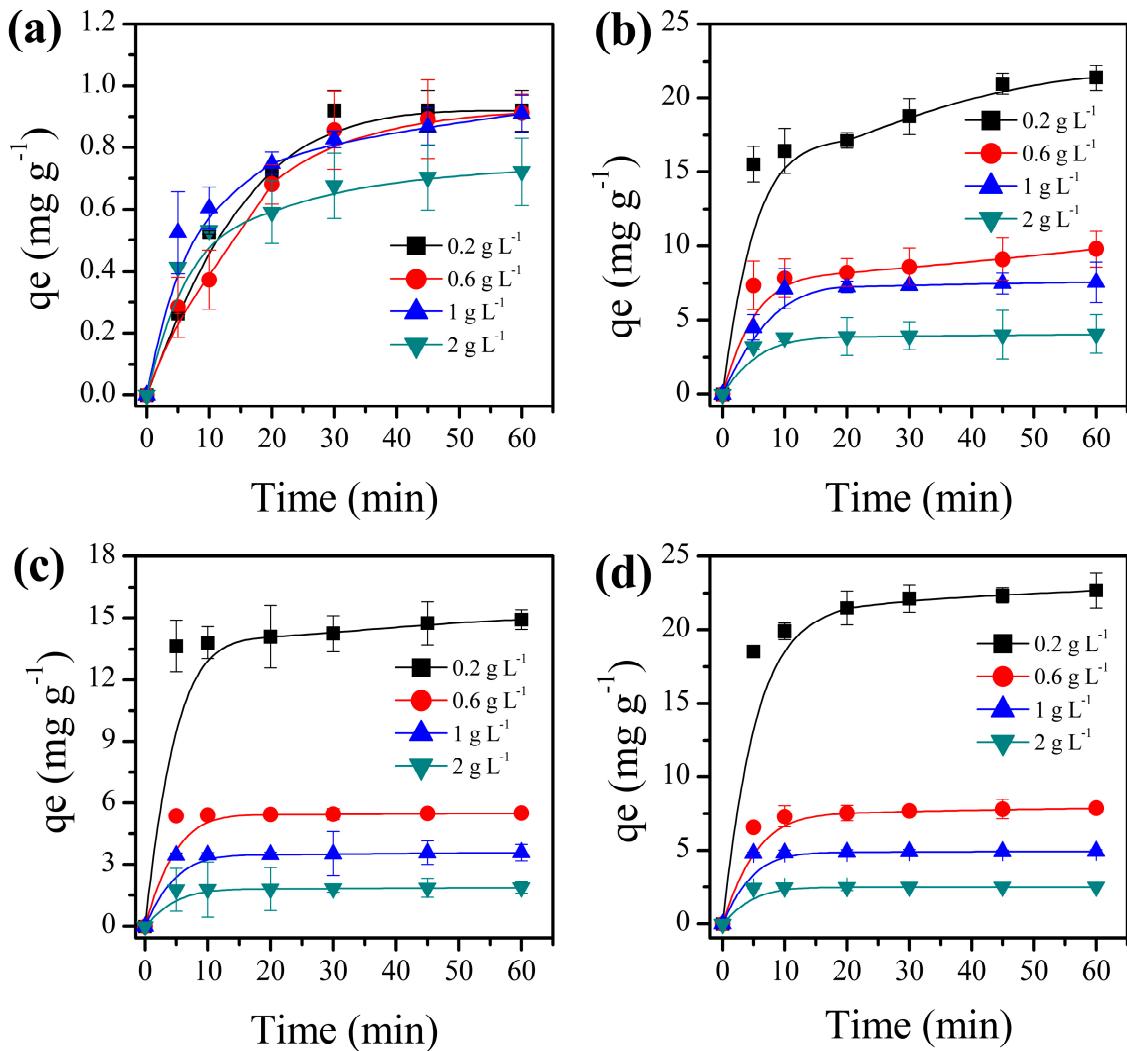
**Figure S4.** Structure IC



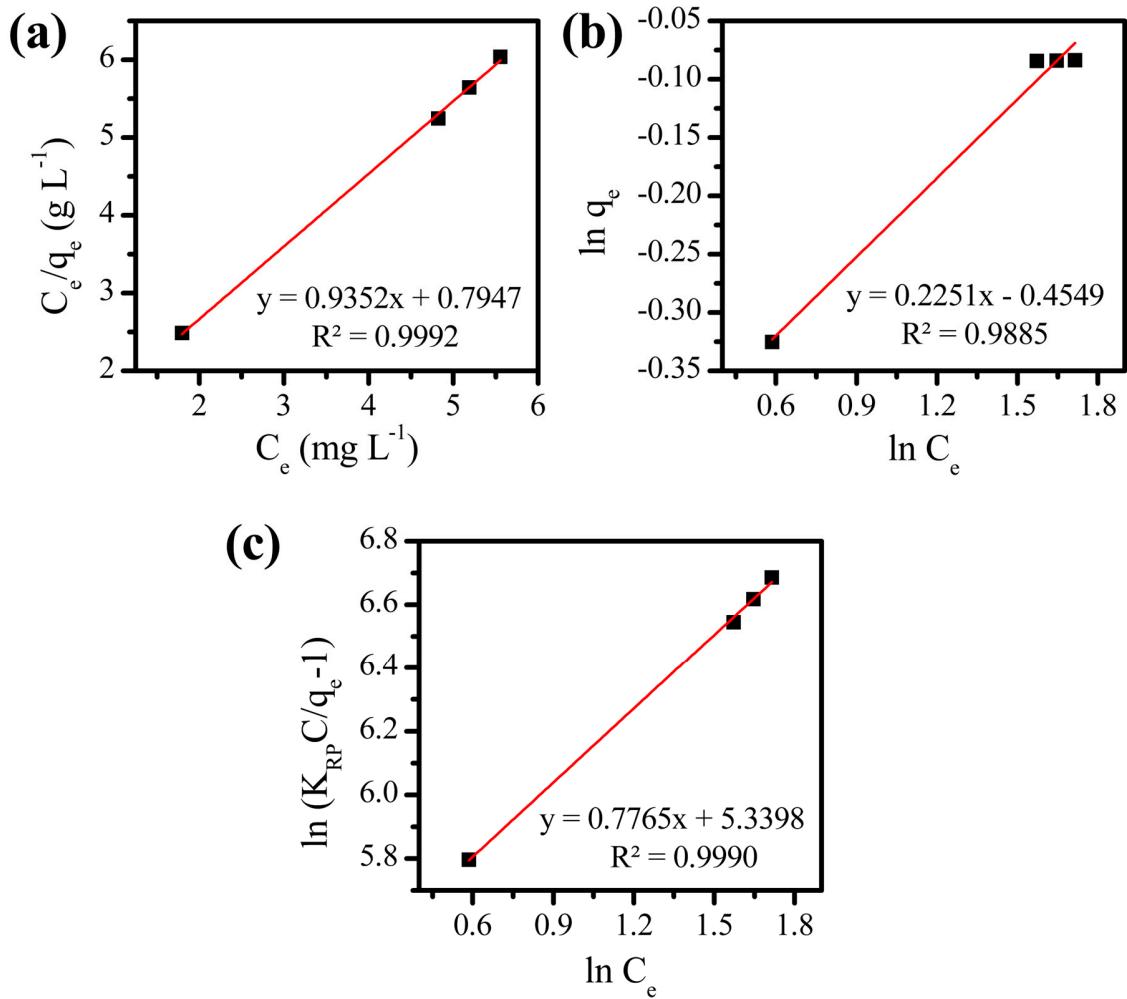
**Figure S5.** Structure CR



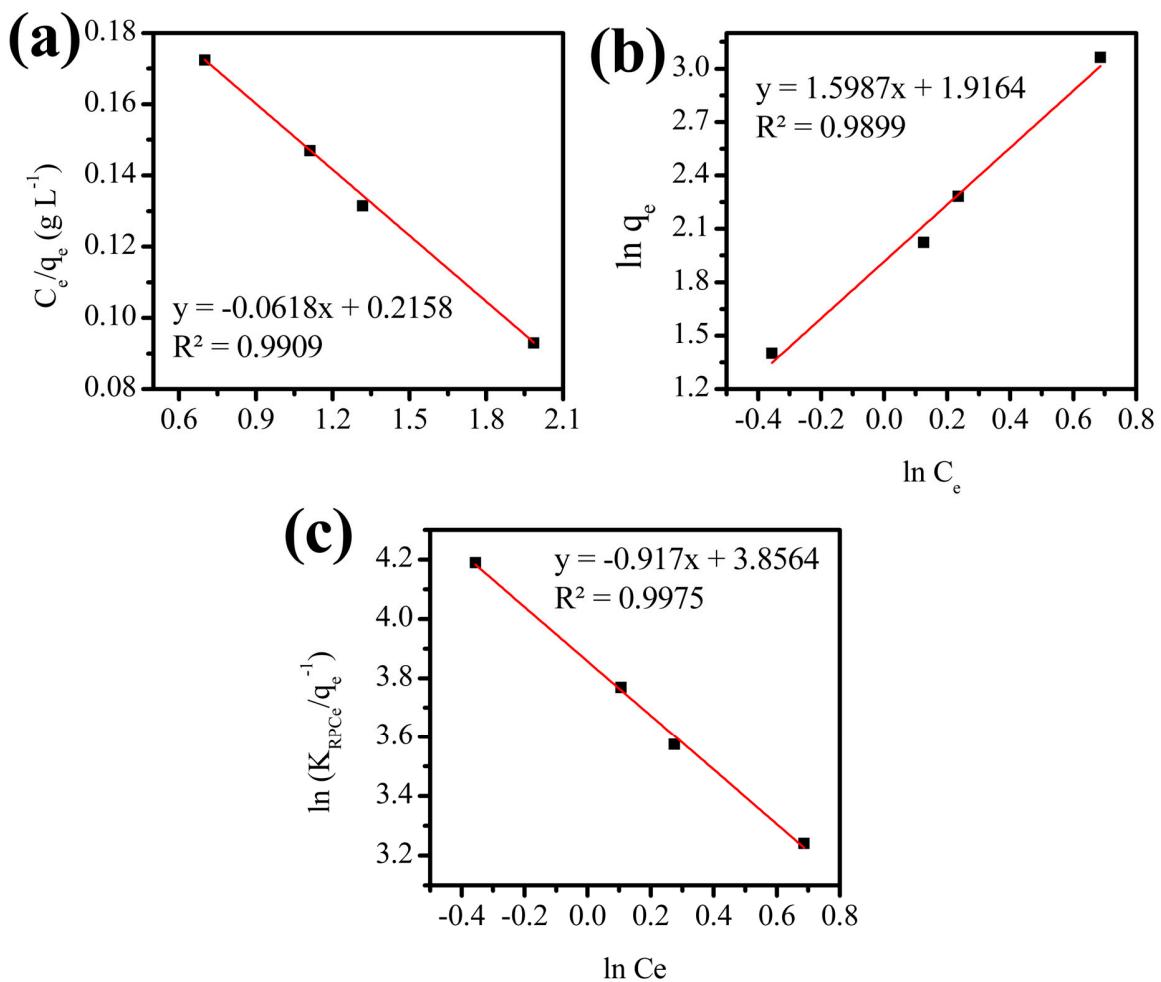
**Figure S6.** Effect of MMT dose in the adsorption percentage of CV, MB, CR and IC after 60 min. Conditions: Dyes concentration  $1.23 \times 10^{-2}$  mmol L<sup>-1</sup>, adsorbent dose 0.2 – 2 g L<sup>-1</sup>, particle size <200 µm, pH: IC: 5.7; CR: 6.5; MB: 5.6; CV: 7.3, temperature 25°C, stirring rate 200 rpm.



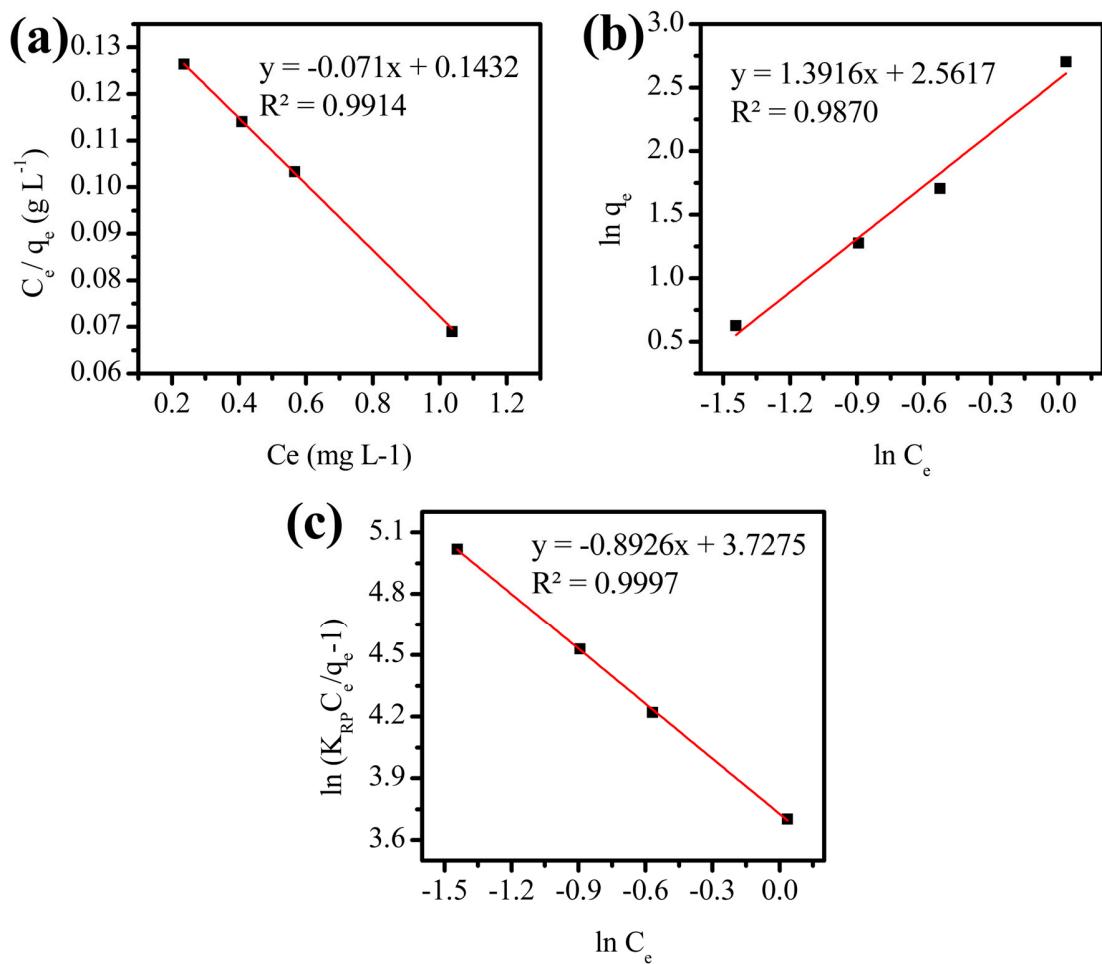
**Figure S7.** IC, CR, MB and CV removal in distilled water using different doses of MMT. Conditions: dyes concentration  $1.23 \times 10^{-2}$  mmol L<sup>-1</sup> (IC: 5.74 mg L<sup>-1</sup>, CR: 8.57 mg L<sup>-1</sup>, MB: 3.93 mg L<sup>-1</sup> and CV: 4.85 mg L<sup>-1</sup>), adsorbent dose 0.2 - 2 g L<sup>-1</sup>, particle size <200  $\mu\text{m}$ , pH: IC: 5.7; CR: 6.5; MB: 5.6; CV: 7.3, temperature 25°C, stirring rate 200 rpm.



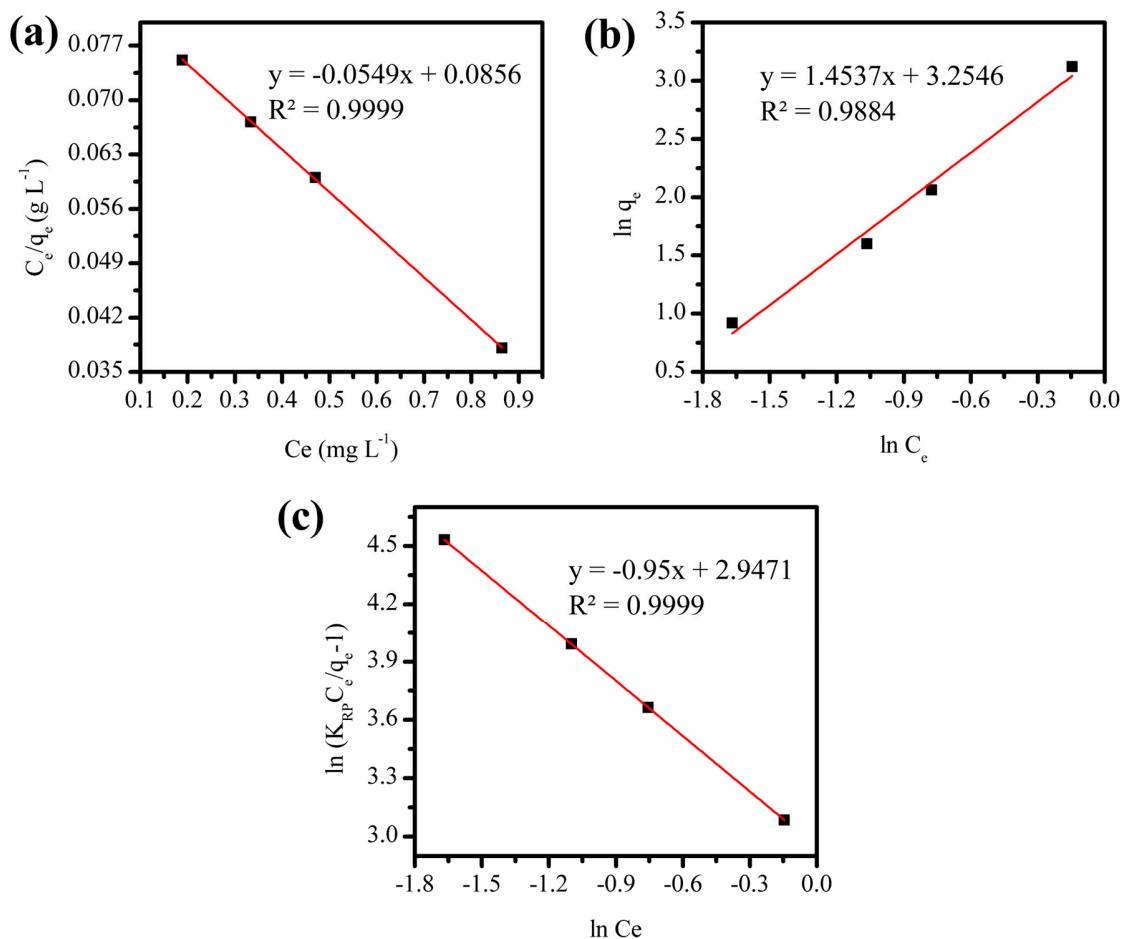
**Figure S8.** Adsorption isotherms for IC removal in distilled water using MMT as an adsorbent. (a) Langmuir model; (b) Freundlich model; (c) Redlich-Peterson model. Conditions: dye concentration  $1.23 \times 10^{-2}$  mmol  $\text{L}^{-1}$  ( $5.74 \text{ mg L}^{-1}$ ), adsorbent dose 0.2 - 2 g  $\text{L}^{-1}$ , particle size  $<200 \mu\text{m}$ , pH: 5.7, temperature 25°C, stirring rate 200 rpm, time 60 min.



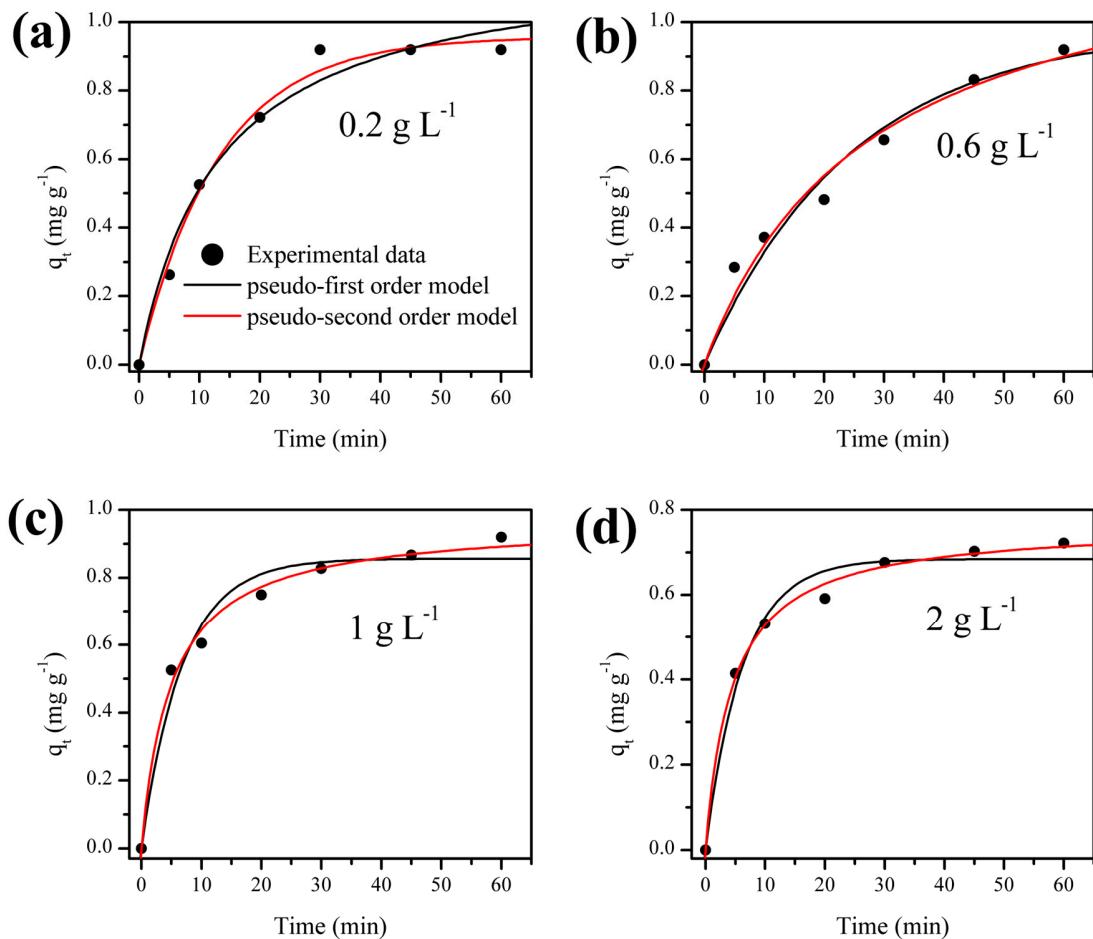
**Figure S9.** Adsorption isotherms for CR removal in distilled water using MMT as an adsorbent. (a) Langmuir model; (b) Freundlich model; (c) Redlich-Peterson model. Conditions: dye concentration  $1.23 \times 10^{-2} \text{ mmol L}^{-1}$  ( $8.57 \text{ mg L}^{-1}$ ), adsorbent dose  $0.2 - 2 \text{ g L}^{-1}$ , particle size  $<200 \mu\text{m}$ , pH: 6.5, temperature  $25^\circ\text{C}$ , stirring rate 200 rpm, time 60 min.



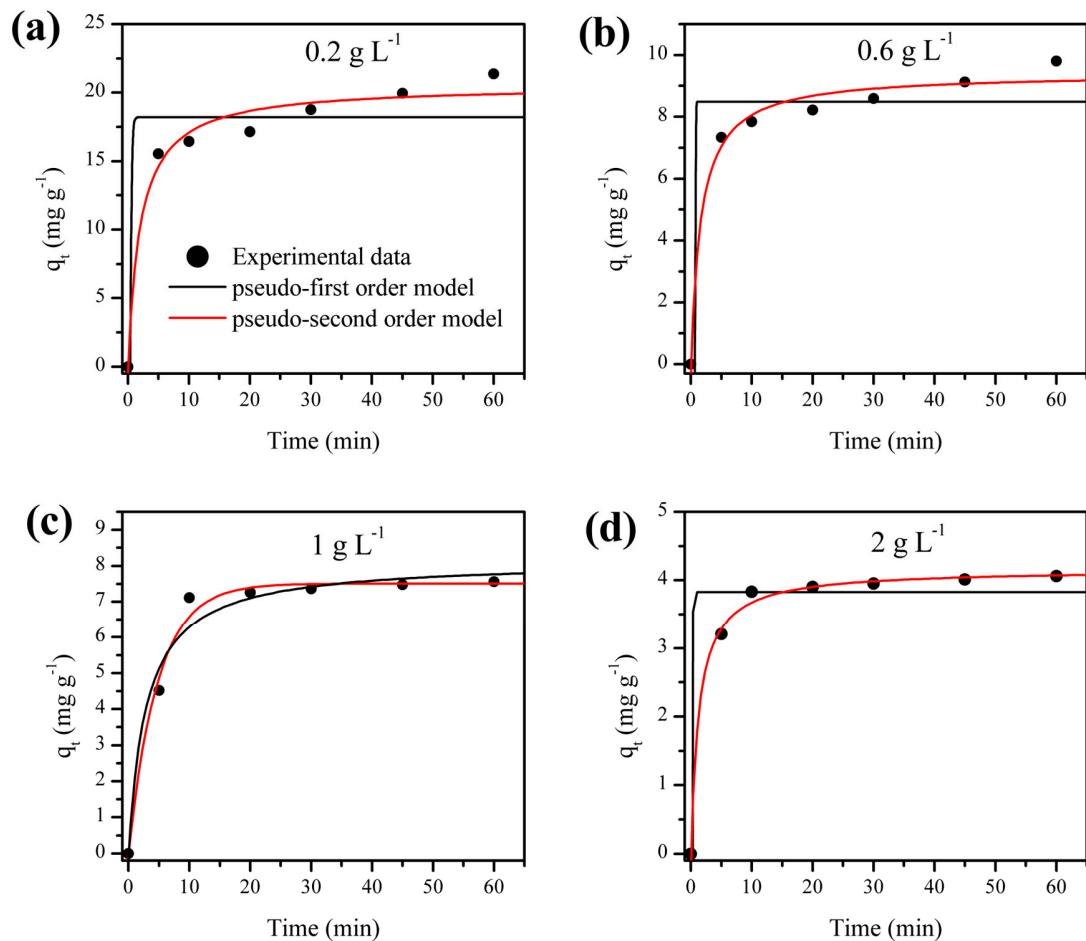
**Figure S10.** Adsorption isotherms for MB removal in distilled water using MMT as an adsorbent. (a) Langmuir model; (b) Freundlich model; (c) Redlich-Peterson model. Conditions: dye concentration  $1.23 \times 10^{-2}$  mmol L<sup>-1</sup> (3.93 mg L<sup>-1</sup>), adsorbent dose 0.2-2 g L<sup>-1</sup>, particle size <200 µm, pH: 5.6, temperature 25°C, stirring rate 200 rpm, time 60 min.



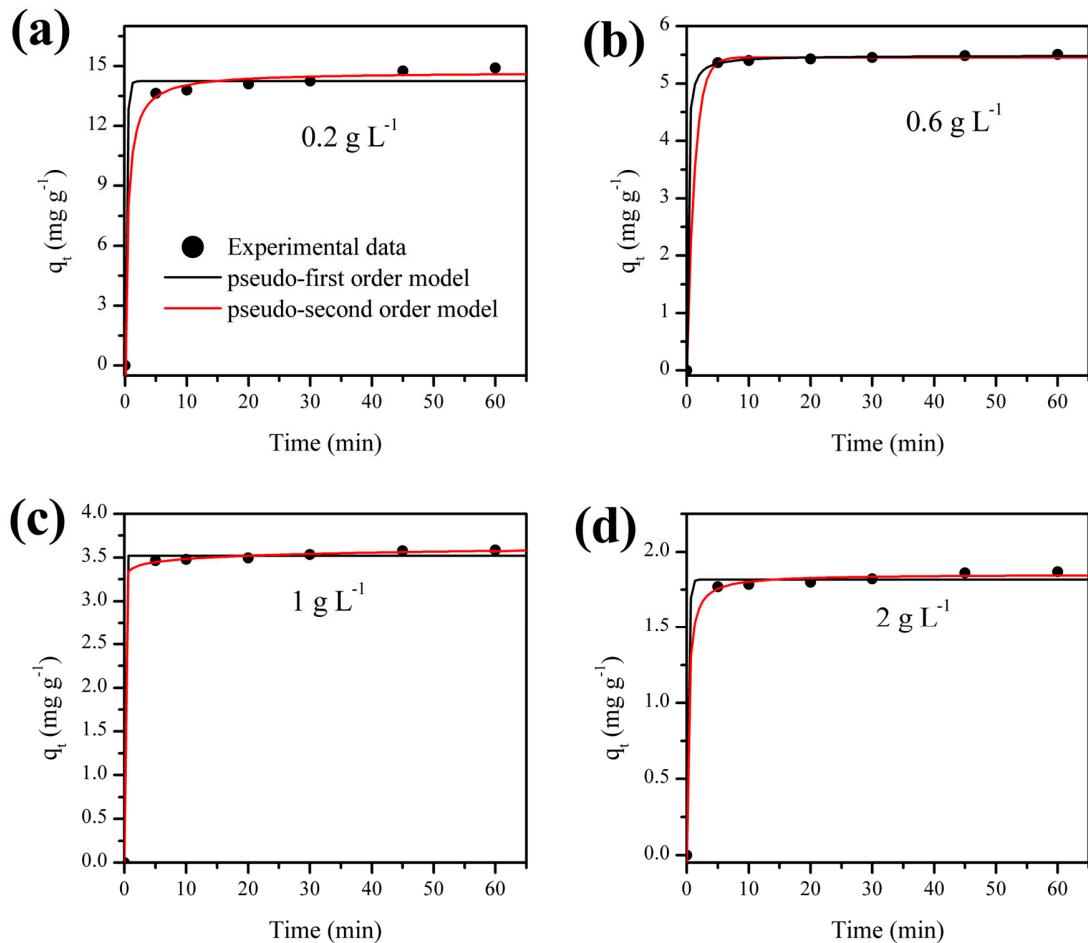
**Figure S11.** Adsorption isotherms for CV removal in distilled water using MMT as an adsorbent. (a) Langmuir model; (b) Freundlich model; (c) Redlich-Peterson model. Conditions: dye concentration  $1.23 \times 10^{-2}$  mmol L<sup>-1</sup> (4.85 mg L<sup>-1</sup>), adsorbent dose 0.2 - 2 g L<sup>-1</sup>, particle size <200 μm, pH: 7.3, temperature 25°C, stirring rate 200 rpm, time 60 min.



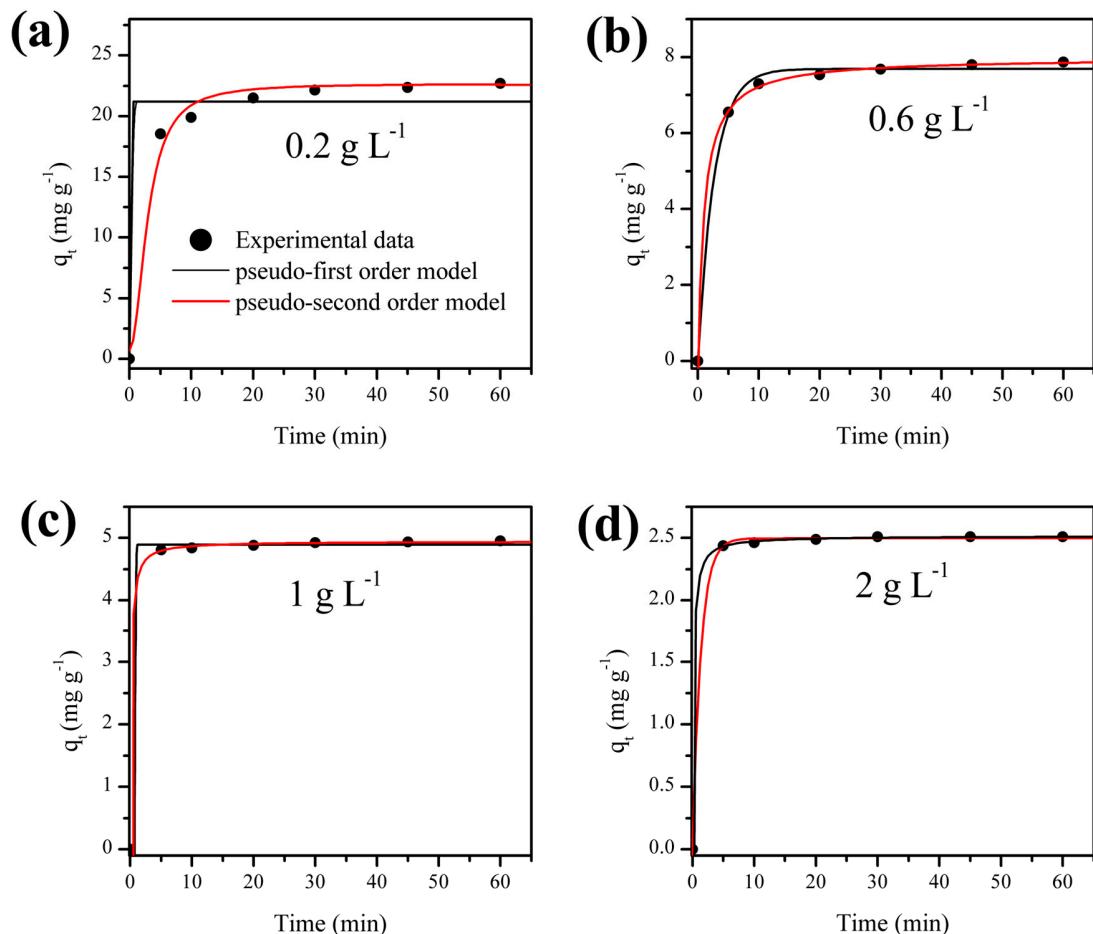
**Figure S12.** Kinetics of pseudo-first order model and pseudo-second order model for IC removal in distilled water using MMT as an adsorbent. (a)  $0.2 \text{ g L}^{-1}$ ; (b)  $0.6 \text{ g L}^{-1}$ ; (c)  $1 \text{ g L}^{-1}$ ; (d)  $2 \text{ g L}^{-1}$ . Conditions: dye concentration  $1.23 \times 10^{-2} \text{ mmol L}^{-1}$  ( $5.74 \text{ mg L}^{-1}$ ), particle size  $<200 \mu\text{m}$ , pH: 5.7, temperature  $25^\circ\text{C}$ , stirring rate 200 rpm, time 60 min.



**Figure S13.** Kinetics of pseudo-first order model and pseudo-second order model for CR removal in distilled water using MMT as an adsorbent. (a)  $0.2 \text{ g L}^{-1}$ ; (b)  $0.6 \text{ g L}^{-1}$ ; (c)  $1 \text{ g L}^{-1}$ ; (d)  $2 \text{ g L}^{-1}$ . Conditions: dye concentration  $1.23 \times 10^{-2} \text{ mmol L}^{-1}$  ( $8.57 \text{ mg L}^{-1}$ ), particle size  $<200 \mu\text{m}$ , pH: 6.5, temperature  $25^\circ\text{C}$ , stirring rate 200 rpm, time 60 min.



**Figure S14.** Kinetics of pseudo-first order model and pseudo second-order model for MB removal in distilled water using MMT as an adsorbent. (a)  $0.2 \text{ g L}^{-1}$ ; (b)  $0.6 \text{ g L}^{-1}$ ; (c)  $1 \text{ g L}^{-1}$ ; (d)  $2 \text{ g L}^{-1}$ . Conditions: dye concentration  $1.23 \times 10^{-2} \text{ mmol L}^{-1}$  ( $3.93 \text{ mg L}^{-1}$ ), particle size  $<200 \mu\text{m}$ , pH: 5.6, temperature  $25^\circ\text{C}$ , stirring rate 200 rpm, time 60 min.



**Figure S15.** Kinetics of pseudo-first order model and pseudo-second order model for CV removal in distilled water using MMT as an adsorbent. (a)  $0.2 \text{ g L}^{-1}$ ; (b)  $0.6 \text{ g L}^{-1}$ ; (c)  $1 \text{ g L}^{-1}$ ; (d)  $2 \text{ g L}^{-1}$ . Conditions: dye concentration  $1.23 \times 10^{-2} \text{ mmol L}^{-1}$  ( $4.85 \text{ mg L}^{-1}$ ), particle size  $<200 \mu\text{m}$ , pH: 7.3, temperature 25°C, stirring rate 200 rpm, time 60 min.