

# **Utilization of waste straw biomass in suspension magnetization roasting of refractory iron ore: iron recovery, gas analysis and roasted products characterization**

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## **Captions**

Figure S1. Preparation process of straw powder.

Figure S2. The general ion diagrams of straw-type and hematite ores mixture at different characteristic temperature: (a) 329 °C, and (b) 820 °C.

Table S1. Peak areas of evolved compounds at 329 °C in SMR process by GC/MS.

Table S2. Peak areas of evolved compounds at 820 °C in SMR process by GC/MS.

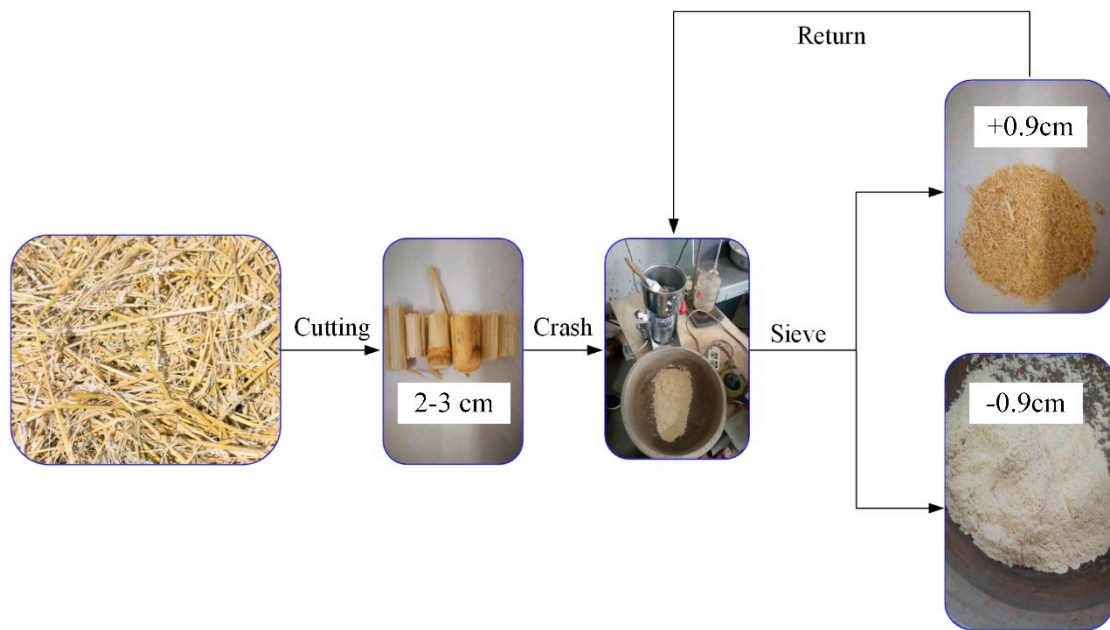


Figure S1. Preparation process of straw powder.

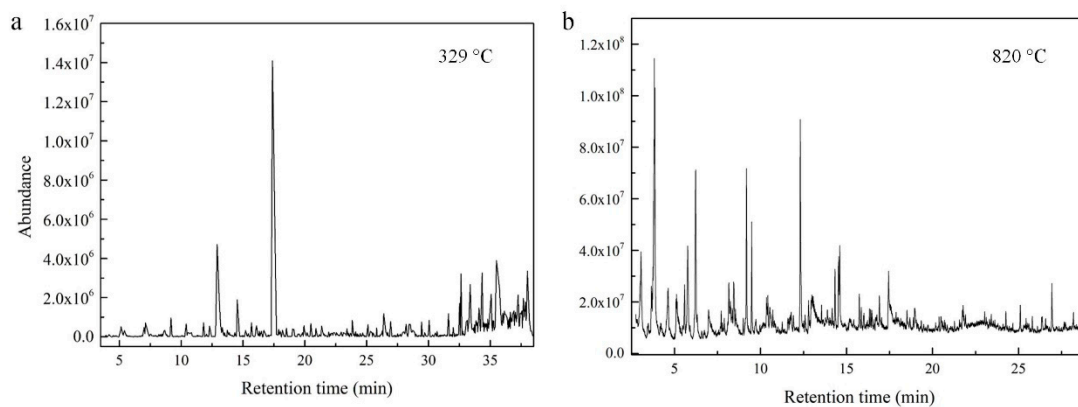


Figure S2. The general ion diagrams of straw-type and hematite ores mixture at different characteristic temperature: (a) 329 °C, and (b) 820 °C.

Table S1. Peak areas of evolved compounds at 329 °C in SMR process by GC/MS.

Retention time (min)	Molecular formula	Compounds	Relative content (%)
12.899	C <sub>8</sub> H <sub>8</sub> O	4-methylbenzaldehyde	2.73
13.059	C <sub>8</sub> H <sub>8</sub> O	Vinyl phenyl ether	1.22
14.534	C <sub>9</sub> H <sub>10</sub> O <sub>2</sub>	2-methyl-4-hydroxyacetophenone	1.10
17.375	C <sub>6</sub> H <sub>10</sub> O <sub>5</sub>	1,6-anhydro-β-D-glucose	8.16
26.389	C <sub>15</sub> H <sub>24</sub> O <sub>2</sub>	3,5-di-tert-butyle-4-hydroxyanisole	0.68
31.621	C <sub>17</sub> H <sub>30</sub> OSi	Trimethyl [4-(1,1,3,3-tetramethylbutyl) phenoxy] silane	0.68
32.512	C <sub>22</sub> H <sub>46</sub>	6,6-diethylhooctadecane	0.99
32.637	C <sub>30</sub> H <sub>50</sub>	Squaleme	1.86
33.387	C <sub>6</sub> H <sub>18</sub> O <sub>3</sub> Si <sub>3</sub>	Hexamethyl-cyclotrisiloxane	1.55

Table S2. Peak areas of evolved compounds at 820 °C in SMR process by GC/MS.

Retention time (min)	Molecular formula	Compounds	Relative content (%)
3.054	C <sub>5</sub> H <sub>8</sub> O <sub>2</sub>	Methyl methacrylate	3.75
3.650	C <sub>21</sub> H <sub>15</sub> O <sub>9</sub> NF <sub>12</sub>	Tetrakis	0.90
3.825	C <sub>7</sub> H <sub>8</sub>	Toluene	10.63
4.635	C <sub>6</sub> H <sub>18</sub> O <sub>3</sub> Si <sub>3</sub>	Hexamethyl-cyclotrisiloxane	2.37
5.115	C <sub>8</sub> H <sub>16</sub> O <sub>2</sub>	CIS-1,2-cyclohexanedimethanol	1.19
5.585	C <sub>8</sub> H <sub>10</sub>	Ethylbenzene	1.10
5.766	C <sub>8</sub> H <sub>10</sub>	P-xylene	3.19
6.226	C <sub>15</sub> H <sub>14</sub> O <sub>2</sub>	Benzoic acid	4.44
6.986	C <sub>8</sub> H <sub>14</sub> ONF <sub>3</sub>	1-(3,3,3-trifluoro-2-hydroxypropyl) piperrazine	1.30
8.437	C <sub>9</sub> H <sub>10</sub>	2-propenyl-benzene	1.24
9.187	C <sub>10</sub> H <sub>16</sub>	D-limonene	3.09
9.482	C <sub>9</sub> H <sub>8</sub>	3-methylphenylacetylene	1.72
12.313	C <sub>10</sub> H <sub>8</sub>	Naphthalene	4.59
13.049	C <sub>8</sub> H <sub>10</sub> O <sub>2</sub>	1,2-benzenedimethanol	0.94
14.329	C <sub>11</sub> H <sub>10</sub>	2-methyl-naphthalene	1.28
14.539	C <sub>9</sub> H <sub>10</sub> O <sub>2</sub>	2-methoxy-4-vinylphenol	1.47
14.604	C <sub>11</sub> H <sub>10</sub>	1-methyl-naphthalene	1.81
17.445	C <sub>6</sub> H <sub>10</sub> O <sub>5</sub>	1,6-anhydro-beta-D-glucopyranose	2.10
17.560	C <sub>6</sub> H <sub>12</sub> O <sub>6</sub>	D-allose	1.28
26.939	C <sub>18</sub> H <sub>18</sub>	Methyl-1-isopropyl-7-phenanthrene	0.97