

Biosorption and Bioleaching of Heavy Metals from Electronic Waste Varied with Microbial Genera

Table S1. Advantages and Disadvantages of conventional and biological methods for efficient metal recovery from e-waste.

Conventional methods	Advantages	Disadvantages	References
Chemical precipitation	<ul style="list-style-type: none"> Integrated physico-chemical process, low capital cost, simple in operation, adaptive to high pollutant loads, efficient for removal of metals and fluorides 	<ul style="list-style-type: none"> Inefficient for metals present in less amount, pH dependent requires chemicals such as lime, oxidants and hydrogen sulfide, oxidation required if metals are complexed, maximum sludge generation during the entire process 	[5,6]
		<ul style="list-style-type: none"> requires extra cost for sludge management 	
Coagulation/Flocculation	<ul style="list-style-type: none"> Integrated physico-chemical technique, simple process, low capital cost, efficient for removal of soluble, insoluble metals and colloidal particles 	<ul style="list-style-type: none"> pH dependent requires non-reusable chemicals, maximum sludge generation requires extra cost for sludge management not efficient for As removal 	[7,8]
		<ul style="list-style-type: none"> High operational cost (maintenance and initial cost of the resin), 	
Ion exchange	<ul style="list-style-type: none"> Technologically simple process, rapid, efficient process for removal of valuable metals up to ppb levels, can be integrated with other processes (precipitation and filtration) 	<ul style="list-style-type: none"> not efficient for dispersal dyes and drugs, large volume requires large columns, rapid saturation and clogging of the reactor, easily contaminated with organic matter, oils 	[9]
		<ul style="list-style-type: none"> requires physico-chemical pretreatment 	
Filtration	<ul style="list-style-type: none"> Simple process, smaller space requirement, high separation selectivity with fast kinetics, efficient for removal of wide range of pollutants 	<ul style="list-style-type: none"> High operational cost due to reactor clogging, require several types of adsorbents, regeneration is quite costly and results in loss of material, 	[6]

		<ul style="list-style-type: none"> not efficient for certain metals and dye stuffs 	
Electro-dialysis	<ul style="list-style-type: none"> Simple process, efficient even at high concentration, no chemicals required, efficient for removal of dyes, salts, minerals and dissolved inorganic matter with high separation selectivity 	<ul style="list-style-type: none"> High maintenance cost high energy requirement, specific process, rapid clogging of membrane, limited flow rates low throughput 	[10]
Photo-catalysis	<ul style="list-style-type: none"> Simple, highly efficient process for removal of organic pollutants and metals simultaneously, production of energy 	<ul style="list-style-type: none"> Initial investment costs, formation of dioxins and other pollutants, high operational and maintenance cost, limited applications 	[11]
Biological methods	<ul style="list-style-type: none"> Low cost, high efficiency, potential metal recovery, regeneration of biosorbent, economically attractive, non-requirement of additional nutrient, utilization of large number of microbial species either in pure or mixed cultures, environment-friendly, minimization of sludge 	<ul style="list-style-type: none"> Possibility of bio-transformation or biodegradation, slow process, complexity of microbiological mechanisms, necessary to maintain optimized conditions, possible sludge bulking and foaming 	[12–14]

Table S2. Research studies emphasizing the potential of biological remediation of the E-waste.

Physico chemical practices	Biological practices	Remarks
<ul style="list-style-type: none"> Hydrometallurgy: Gold mining involves use of leach reagent i.e. cyanide <ul style="list-style-type: none"> Exhibits high gold recovery Robust and low cost environmental damage mismanagement produces noxious side products which tempers recycling process <ul style="list-style-type: none"> Cu, Pb and Sn recovery from scrap printed circuit boards (PCBs) by combination of leaching, electrochemical ion exchange and electrodeposition Use of aq. HNO_3 stripping solution for Cu and Pb extraction Sn precipitation as 2SnO_3 at high acid concentrations ($>4 \text{ M dm}^{-3}$) 	<ul style="list-style-type: none"> No leaching agents are required Microorganisms themselves act as leaching agents Environment-prudent No huge capital investment is required 	<p>[25]</p> <p>[26]</p>

- Poor current efficiency due to cathodic Pb deposition
 - Requirement of huge capital investment
 - Physical processes require low capital
 - suffer from huge metal loss (10-35%)
 - brominated flame retardant dust problem
 - No Brominated flame retardant dust problem was observed
- [27]
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