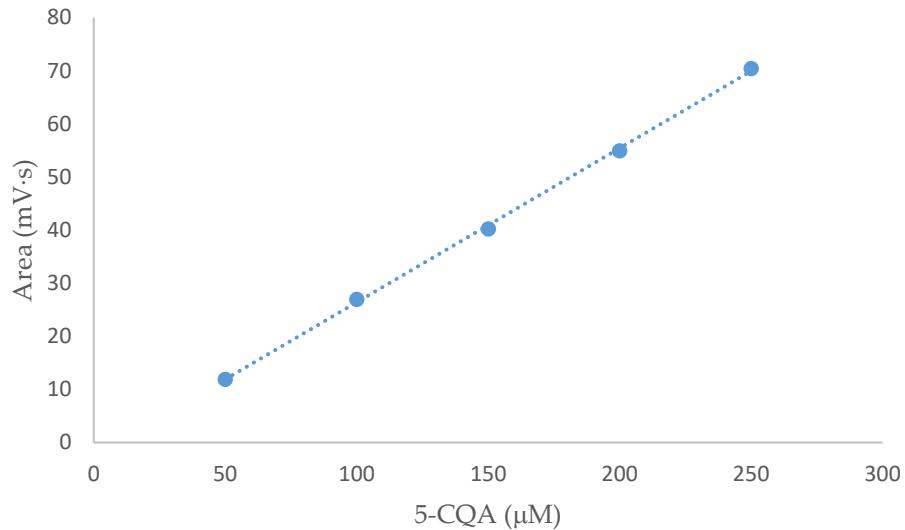


# Fluorescent Imprinted Nanoparticles for Sensing of chlorogenic acid in coffee extracts

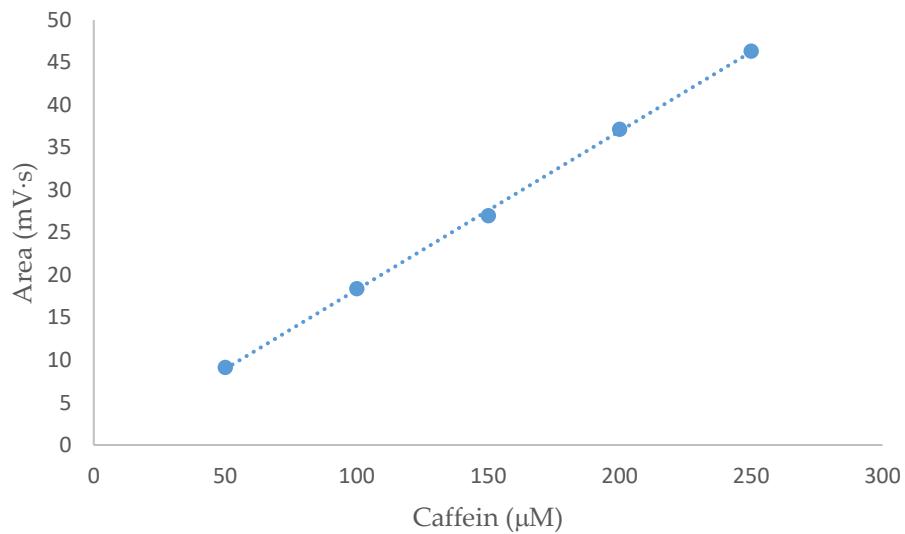
Anggy Lusanna Gutierrez-Ortíz<sup>1</sup>, Veronica Vida<sup>1</sup>, Matjaz Peterka<sup>2</sup>, Jasmina Tušar<sup>2</sup>, Federico Berti<sup>1</sup>, Luciano Navarini<sup>3</sup>, Cristina Forzato <sup>1,\*</sup>

## Supplementary Material

### Calibration curves



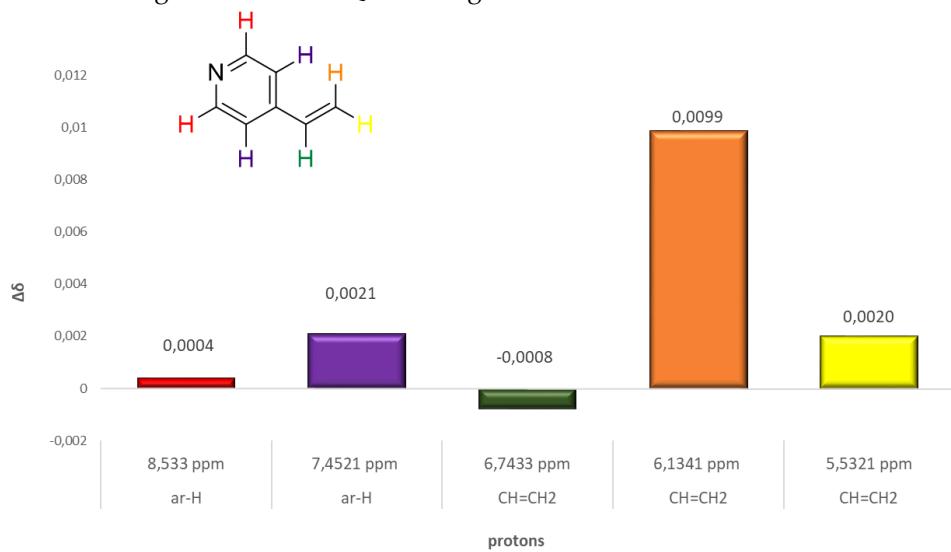
**Figure S1.** HPLC calibration for 5-CQA, average of triplicate measures. Linear regression for 5-CQA:  $r^2$  0.999, slope 0.29, intc. 2.61. Linear regression for HT:  $r^2$  0.997, slope 47.01, intc. -29.15



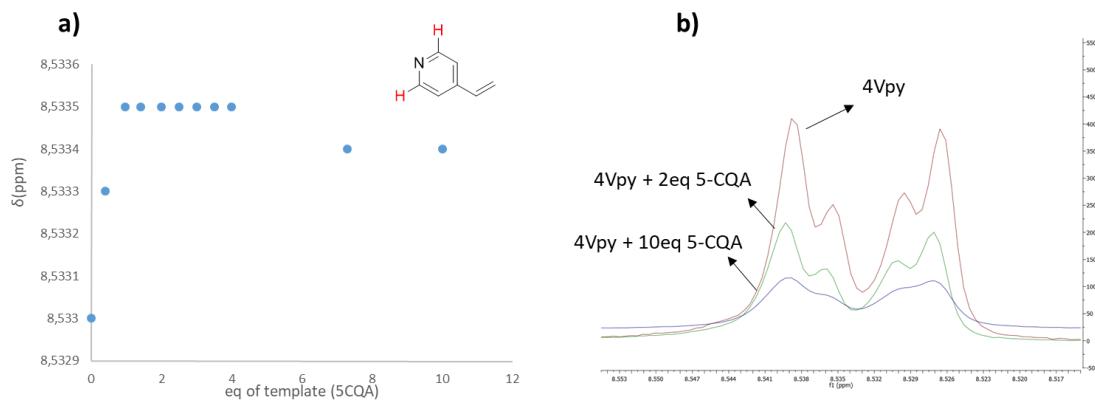
**Figure S2.** HPLC calibration for caffeine, average of triplicate measures. Linear regression:  $r^2$  0.999, slope 0.19, intc. 0.37

*Interactions of functional monomers with the targets: NMR titrations*

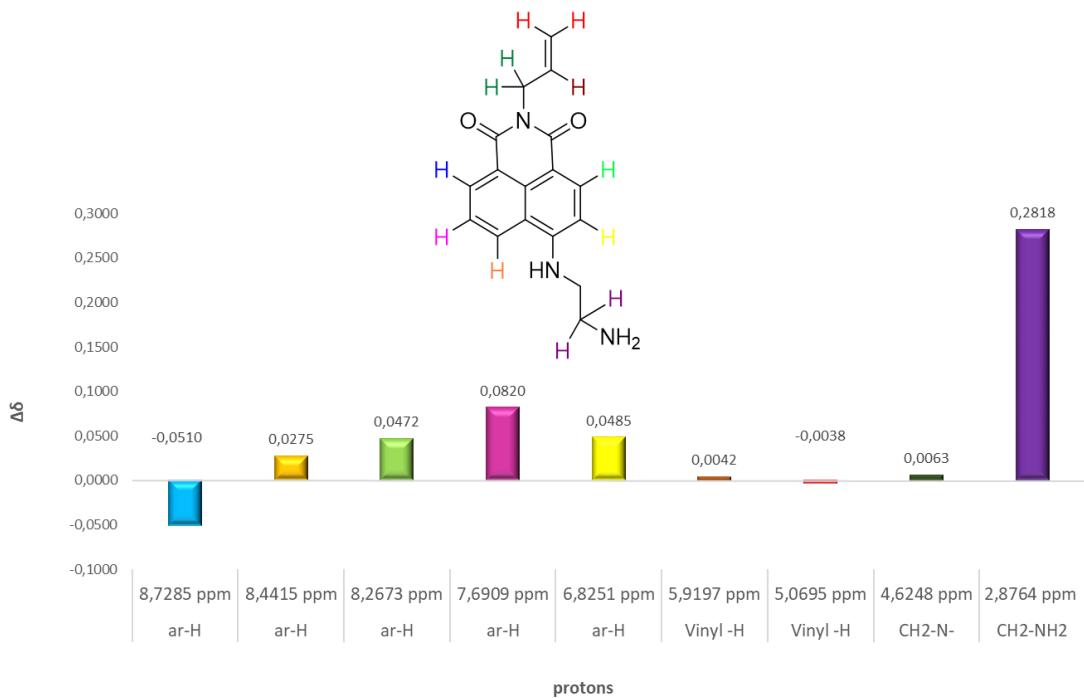
In figures S3 – S7 are reported the most significant changes in the NMR spectra of 4VP and monomer 7 upon titration with increasing amount of 5-CQA leading to the most relevant effect.



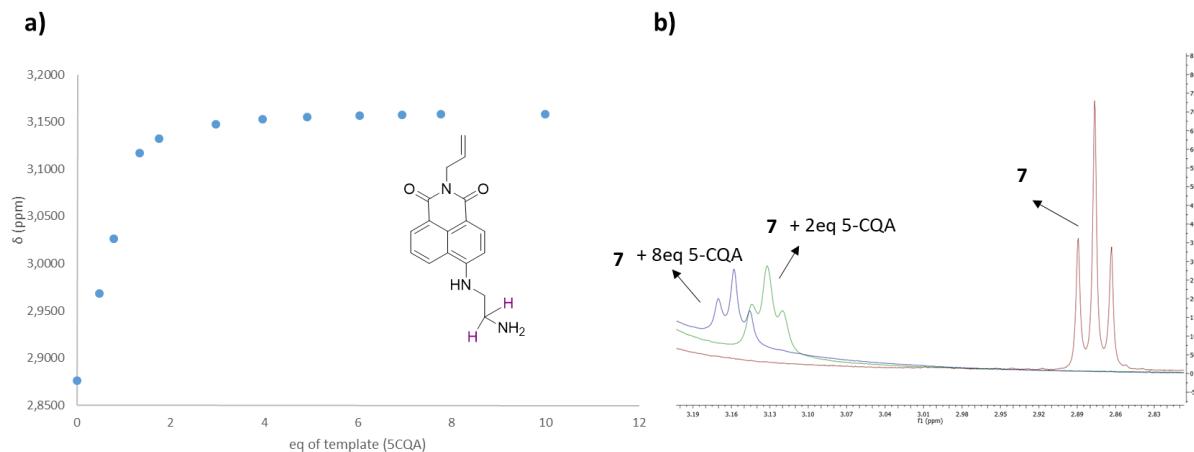
**Figure S3.** Histogram of the chemical shift variations in the  $^1\text{H}$ -NMR spectrum of 4VP upon addition of 10 equivalents of 5-CQA



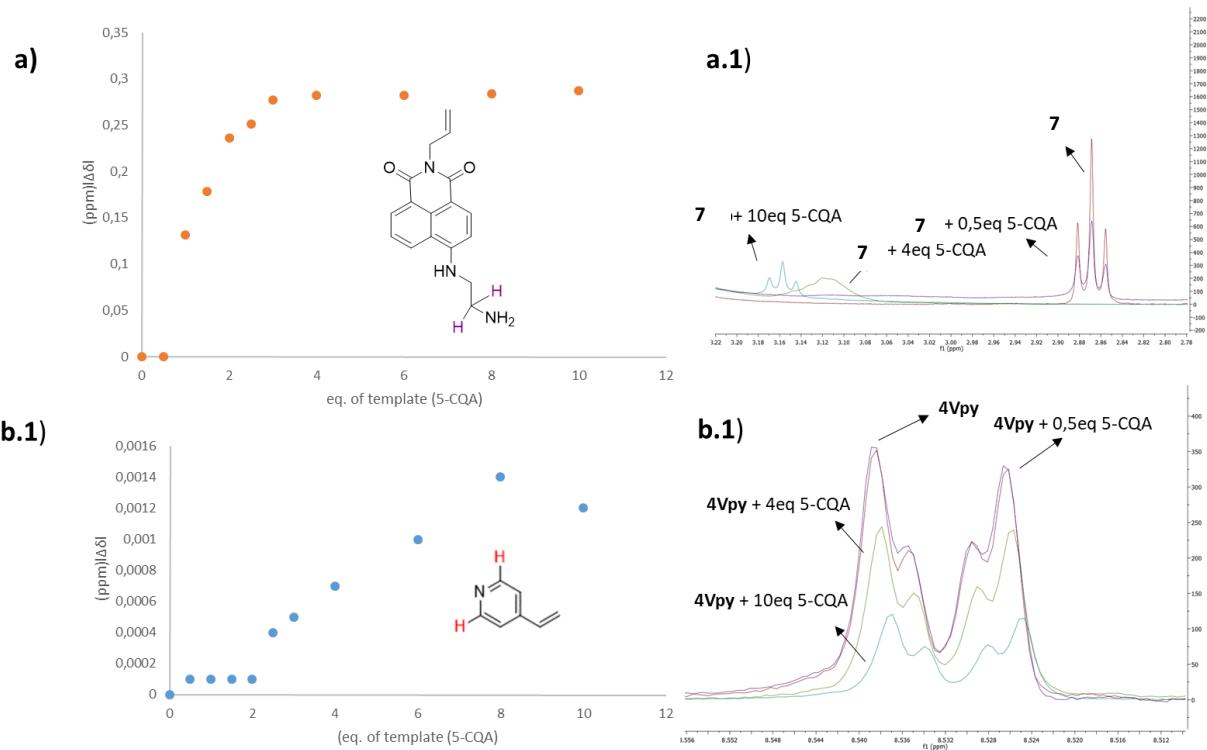
**Figure S4.** a) Chemical shift variation of aromatic proton of 4-vinylpyridine (4Vpy) upon interaction with template molecule (5-CQA). b) progressive shift of the proton aromatic protons adjacent to the nitrogen atom of the pyridine ring in the  $^1\text{H}$ -NMR spectra. Pure 4VPy (red), addition of 2eq of 5-CQA (green) and 10eq (blue).



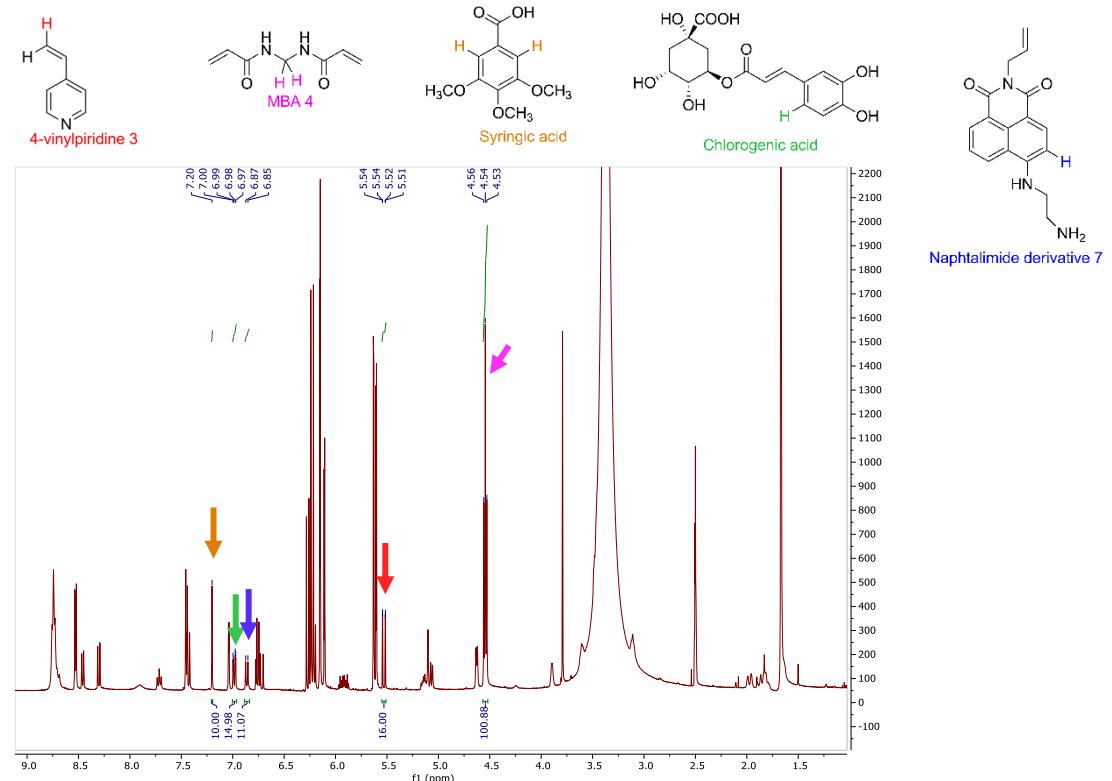
**Figure S5.** Histogram of the chemical shift variations in the  $^1\text{H}$ -NMR spectrum of **7** upon progressive additions of 10eq of 5-CQA



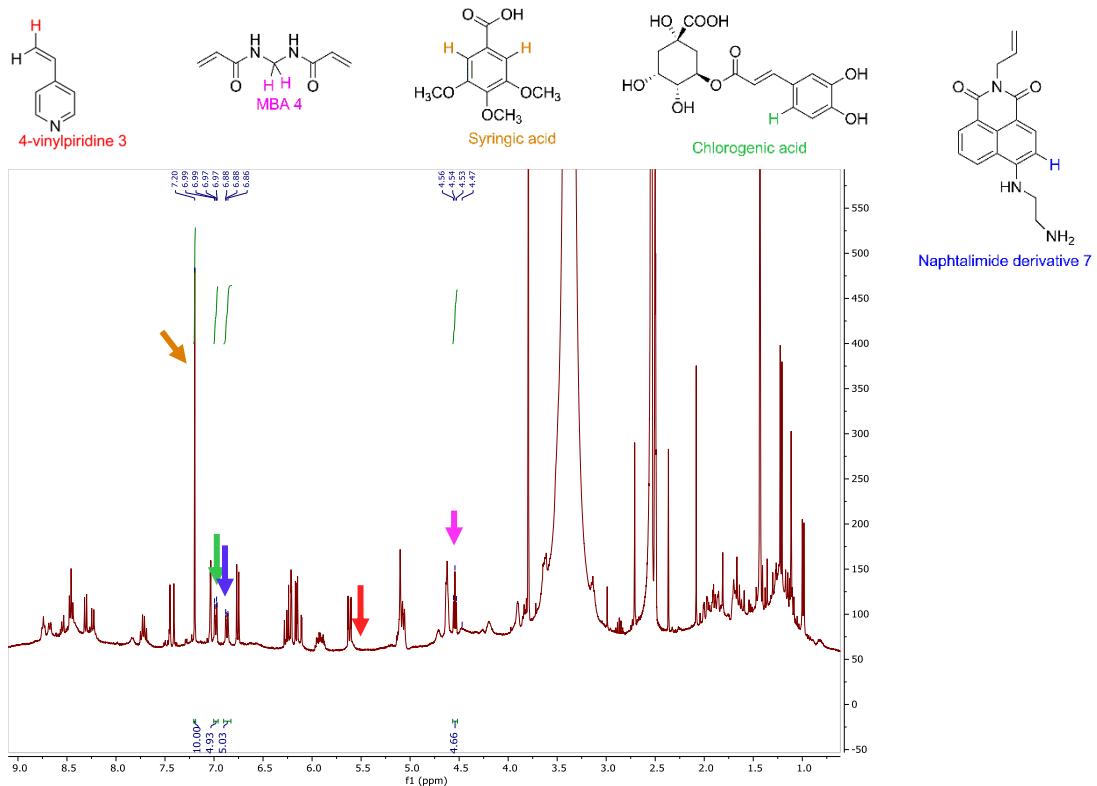
**Figure S6.** a) Chemical shift variation of  $\text{CH}_2\text{NH}_2$  protons of monomer **7** upon interaction with 5-CQA; b) progressive shift of protons  $\text{CH}_2\text{NH}_2$  in the  $^1\text{H}$ -NMR spectra.



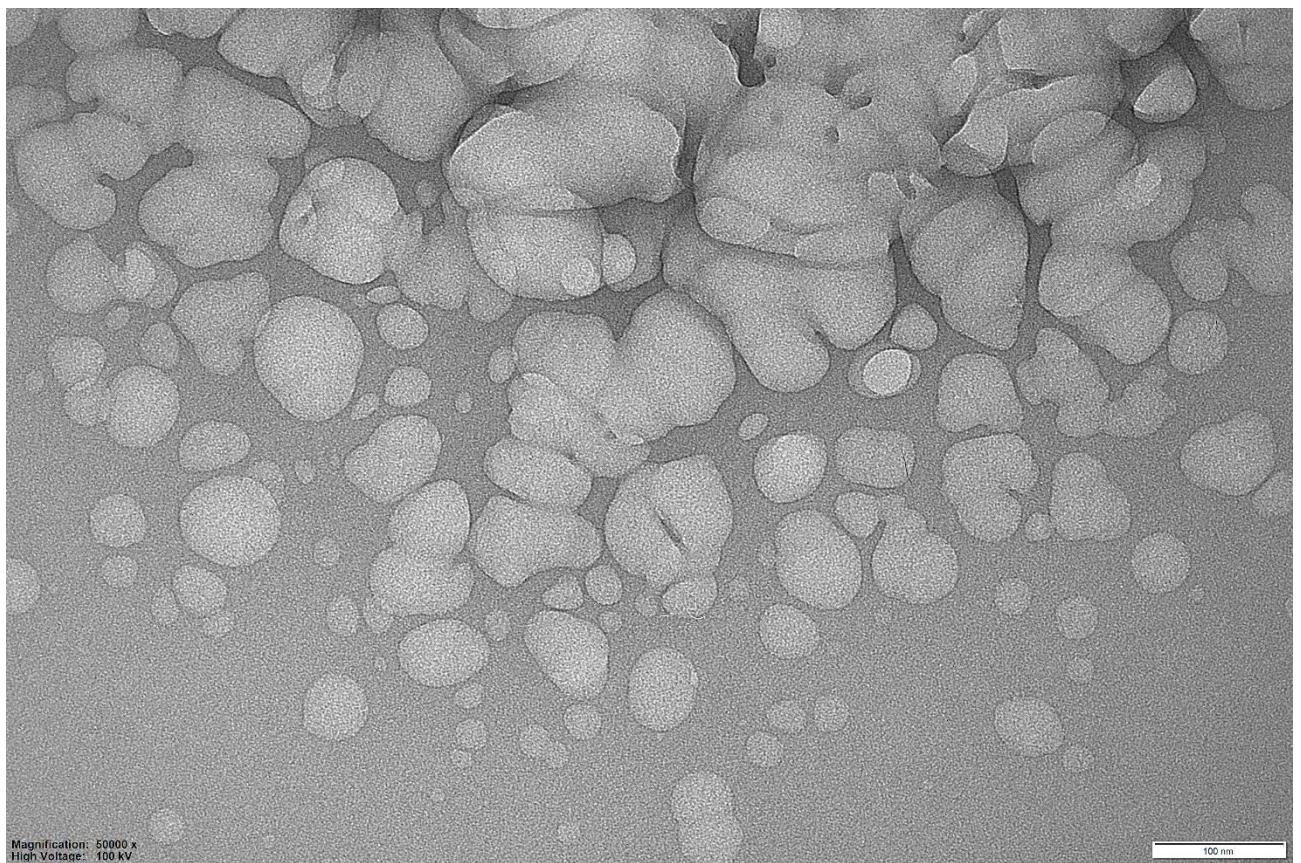
**Figure S7:** a) Chemical shift variation of  $\text{CH}_2\text{NH}_2$  protons of monomer 7 upon interaction of a monomers mix with template molecule (5-CQA). a.1) progressive shift of the protons  $\text{CH}_2\text{NH}_2$  in the  $^1\text{H}$ -NMR spectra: 7 (red), addition of 0.5eq of 5-CQA (purple), 4eq (green) and 10q (blue). b) Chemical shift variation of aromatic proton of 4VP upon interaction of a mixture of monomers and the template molecule (5-CQA). b.1) progressive shift of the proton Ar-H in the  $^1\text{H}$ -NMR spectra. 4VP (red), addition of 0.5eq of 5-CQA (purple), 4eq (green) and 10q (blue).



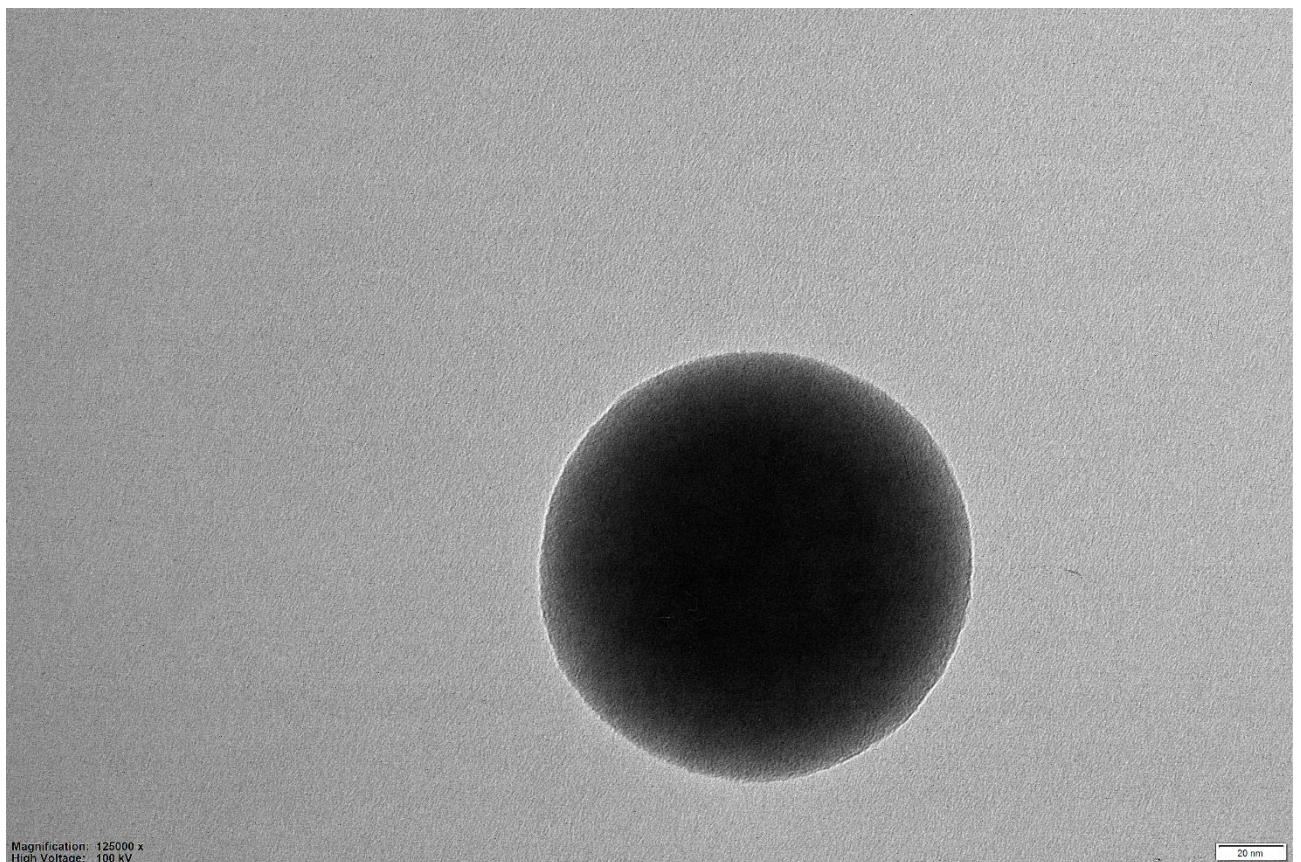
**Figure S8:** Mixture of monomers 4VP **3** and napthalimide derivative **7**, cross-linker MBA **4**, syringic acid (internal standard) and 5-CQA in DMSO-d<sub>6</sub> before polymerization



**Figure S9:** Mixture of monomers 4VP **3** and napthalimide derivative **7**, cross-linker MBA **4**, syringic acid (internal standard) and 5-CQA in DMSO-d<sub>6</sub> after polymerization



**Figure S10:** TEM image of fMIP03



**Figure S11:** TEM image of fMIP03