

Supporting Information:

Catalytic performances of Sn-Beta catalysts prepared from different heteroatom-containing Beta zeolites for the retro aldol fragmentation of glucose

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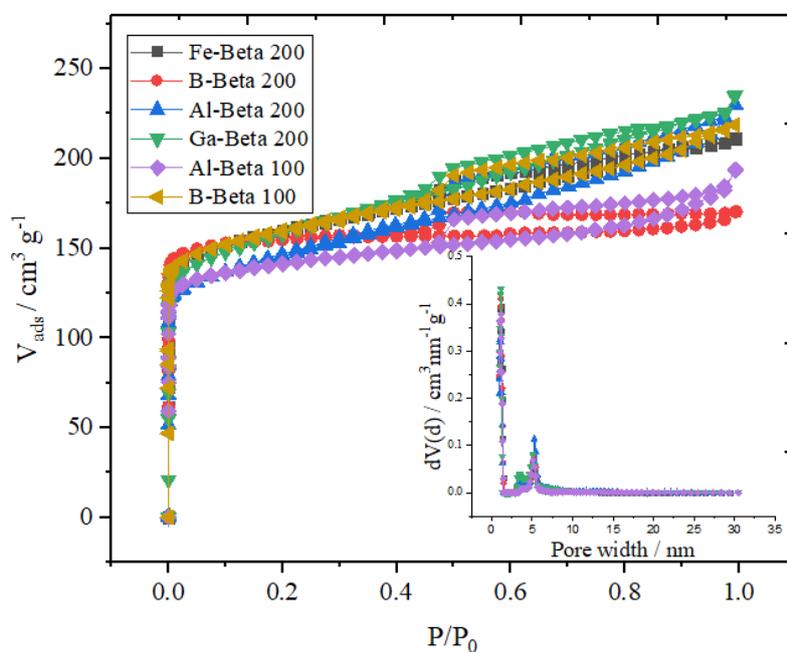
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Table S1. Principal modes and assignments of hydrothermal M-Beta zeolites.

Zeolite	Assignment (cm ⁻¹)						
	ν O-T-O _{Int-T} ^a	$\nu_{(s)}$ O-T-O _{Ext} ^b	ν SiO ₄ (UC) ^c	ν O-T-O _{Int} ^d	$\nu_{(as)}$ O-T-O _{Ext} ^e	ν SiOH _{Int} ^f	ν SiOH _{Term} ^g
Al-Beta 100	736	~800	-	1042	1208	-	3748
B-Beta 100	737	~800	916 948	1040	1209	3734	3747
Ga-Beta 200	736	~800	942	1029	1209	-	3748
Al-Beta 200	737	~800	941	1039	1208	-	3747
B-Beta 200	736	~800	917	1033	1208	3733	3747
Fe-Beta 200	736	~800	-	1031	1207	3735	3748

[a] Intertetrahedral O-T-O; [b] Symmetric external O-T-O; [c] Vibrations of Si-O belonging to uncoupled SiO₄ tetrahedra [d] Symmetric internal O-T-O; [e] Asymmetric external O-T-O; [f]; Internal Si-OH groups present at framework defects; [h] Terminal Si-OH groups.

**Figure S1.** Nitrogen sorption isotherm and pore size distribution using DFT model (inset) of parent M-Beta zeolites.

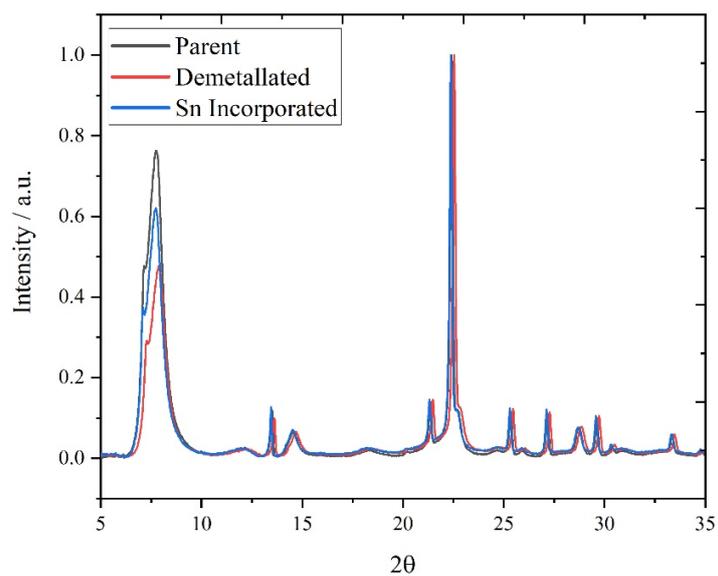


Figure S2. X-Ray Diffraction patterns of calcined B-Beta parent material, demetallated Beta derived from B-Beta, and Sn-Beta achieved by SSI of demetallated B-Beta.

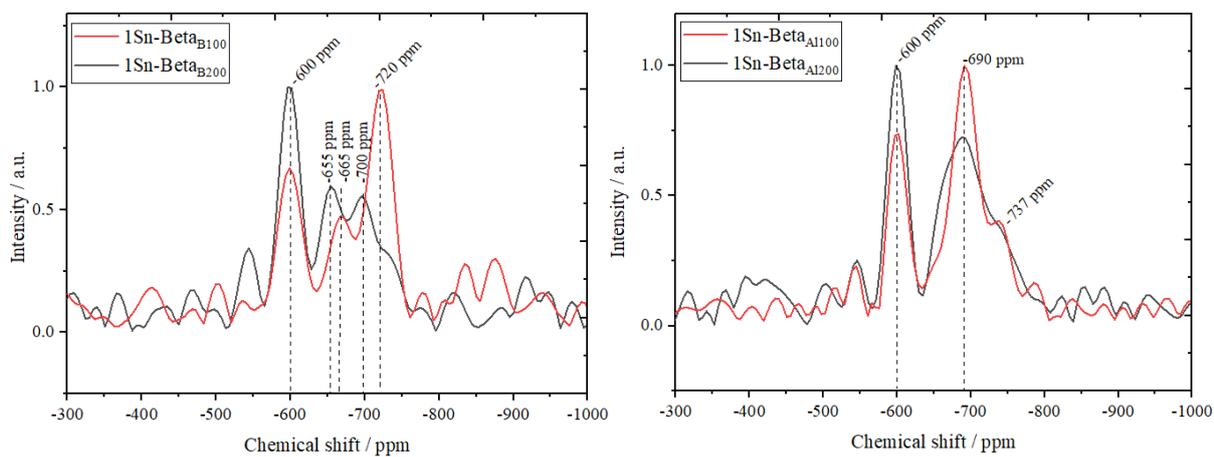


Figure S3. ^{119}Sn De CPMG MAS NMR spectra of 1Sn-Beta₁₀₀ and 1Sn-Beta₂₀₀ where M=B(left) and M=Al (right).

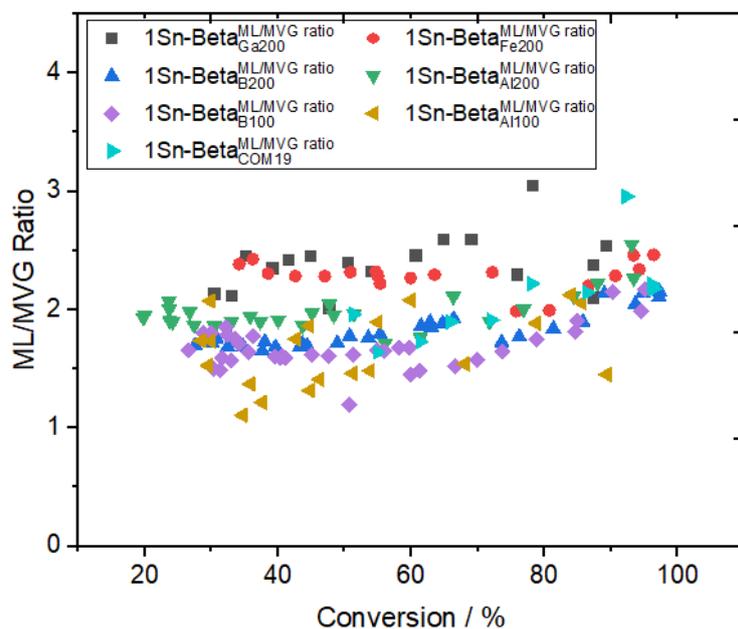


Figure S4. ML/MVG ratio over glucose conversion, for the 1Sn-Beta_{MX} series and 1SnBeta_{COM19}.

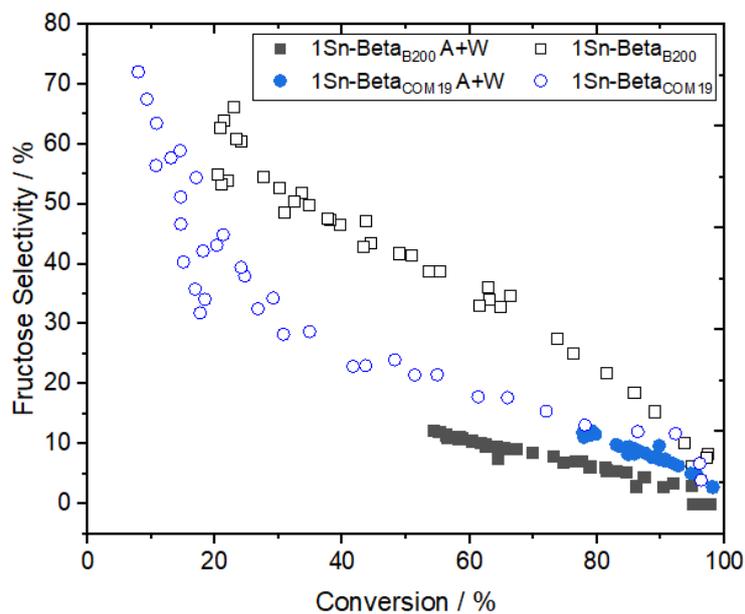


Figure S5. Comparison of the fructose selectivity over glucose conversion of original ML fragmentation and ML fragmentation with alkali salts and water (A+W).

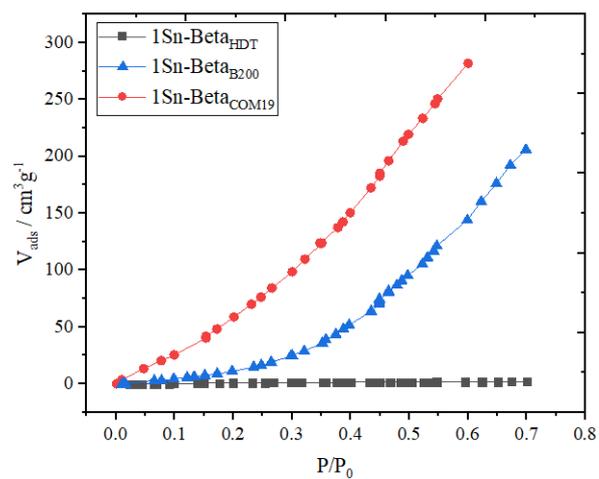


Figure S6. Vapour sorption data of various 1Sn-Beta catalysts, including (red/circles) 1Sn-Beta_{COM19}, (blue/triangles) 1Sn-Beta_{B200} and (squares/black) hydrothermally synthesised 1Sn-Beta_{HDT}, which is provided for reference purposes only.