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*Supplementary Materials*

# **A Systematic Study of the Antioxidant Capacity of Humic Substances against Peroxyl Radicals: Relation to Structure**

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**Table S1.** Pearson correlation coefficients between physical-chemical HS properties and their AOC<sup>1</sup>.

	AOC	O/C	H/C	C/N	C <sub>C=O</sub>	C <sub>COO</sub>	C <sub>ArO</sub>	C <sub>Ar</sub>	C <sub>OCO</sub>	C <sub>CHO</sub>	C <sub>CH2O</sub>	C <sub>CH3O</sub>	C <sub>CHn</sub>	ΣC <sub>Ar</sub>	ΣC <sub>Carb</sub>	ΣC <sub>Alk</sub>	ΣC <sub>Alk-O</sub>	ΣC <sub>Ar</sub> /ΣC <sub>Alk</sub>	TP
AOC	1.00																		
O/C	0.36	1.00																	
H/C	0.08	0.21	1.00																
C/N	<b>0.65</b>	0.18	-0.14	1.00															
C <sub>C=O</sub>	0.11	<b>0.42</b>	0.21	0.00	1.00														
C <sub>COO</sub>	-0.29	0.23	-0.11	-0.11	0.24	1.00													
C <sub>ArO</sub>	-0.39	-0.17	0.05	-0.16	-0.21	0.13	1.00												
C <sub>Ar</sub>	-0.19	<b>-0.73</b>	<b>-0.64</b>	0.03	<b>-0.42</b>	-0.22	0.22	1.00											
C <sub>OCO</sub>	<b>0.44</b>	0.37	0.37	0.17	<b>0.43</b>	<b>-0.45</b>	-0.33	<b>-0.51</b>	1.00										
C <sub>CHO</sub>	<b>0.46</b>	<b>0.52</b>	<b>0.40</b>	0.13	-0.04	-0.39	-0.34	<b>-0.65</b>	<b>0.68</b>	1.00									
C <sub>CH2O</sub>	0.10	0.32	<b>0.53</b>	-0.14	0.03	<b>-0.45</b>	-0.15	<b>-0.60</b>	<b>0.65</b>	<b>0.74</b>	1.00								
C <sub>CH3O</sub>	<b>-0.56</b>	-0.27	0.37	<b>-0.40</b>	-0.03	-0.10	0.38	0.02	-0.06	-0.23	0.31	1.00							
C <sub>CHn</sub>	0.00	0.05	0.06	0.04	-0.08	0.30	-0.29	-0.10	<b>-0.50</b>	-0.24	-0.31	-0.24	1.00						
ΣC <sub>Ar</sub>	-0.27	<b>-0.72</b>	<b>-0.58</b>	-0.01	<b>-0.44</b>	-0.18	<b>0.43</b>	<b>0.98</b>	<b>-0.54</b>	<b>-0.68</b>	-0.59	0.11	-0.16	1.00					
ΣC <sub>Carb</sub>	<b>0.46</b>	<b>0.50</b>	<b>0.44</b>	0.12	0.10	<b>-0.45</b>	-0.35	<b>-0.67</b>	<b>0.83</b>	<b>0.97</b>	<b>0.81</b>	-0.14	-0.35	<b>-0.69</b>	1.00				
ΣC <sub>Alk</sub>	0.39	<b>0.52</b>	<b>0.59</b>	0.07	0.07	-0.38	<b>-0.44</b>	<b>-0.77</b>	<b>0.66</b>	<b>0.90</b>	<b>0.81</b>	-0.06	0.04	<b>-0.82</b>	<b>0.91</b>	1.00			
ΣC <sub>Alk-O</sub>	0.36	<b>0.45</b>	<b>0.52</b>	0.05	0.10	<b>-0.47</b>	-0.28	<b>-0.67</b>	<b>0.83</b>	<b>0.93</b>	<b>0.87</b>	0.05	<b>-0.40</b>	<b>-0.68</b>	<b>0.98</b>	<b>0.90</b>	1.00		
ΣC <sub>Ar</sub> /ΣC <sub>Alk</sub>	-0.27	<b>-0.71</b>	<b>-0.64</b>	0.05	-0.35	0.06	0.37	<b>0.94</b>	<b>-0.61</b>	<b>-0.76</b>	<b>-0.70</b>	0.02	-0.13	<b>0.95</b>	<b>-0.78</b>	<b>-0.91</b>	<b>-0.78</b>	1.00	
TP	<b>0.64</b>	-0.02	0.10	0.23	0.19	<b>-0.54</b>	<b>-0.46</b>	0.09	0.27	0.12	0.08	-0.19	0.17	-0.02	0.07	0.23	0.14	-0.12	1.00

<sup>1</sup> The values in bold denote statistically significant correlation coefficients at  $p < 0.05$ .

H/C, O/C and N/C ratios are calculated on ash- and water-free basis.

Content of carbon in the structural fragments is determined by <sup>13</sup>C NMR spectroscopy as the integral intensity (%) of the following spectral regions (ppm): 220–189 (C<sub>C=O</sub>), 189–168 (C<sub>COO</sub>), 168–145 (C<sub>ArO</sub>), 145–108 (C<sub>Ar</sub>), 108–91 (C<sub>OCO</sub>), 91–66 (C<sub>CHO</sub>), 66–59 (C<sub>CH2O</sub>), 59–48 (C<sub>CH3O</sub>), 48–0 (C<sub>CHn</sub>).

TP is measured in μmol TE mg<sup>-1</sup>.

