



# Supplementary Material: Thermal Depolymerization of $\alpha$ -Methylstyrene/Styrene Resins Inducing SBR Crosslinking and Self-Compatibilization

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# 1. Molecular characteristics of poly(@MSt-co-St) resin

 $\alpha$ -methylstyrene/styrene composition is equal to 45:55 (determined by <sup>1</sup>H NMR). Mn = 1820 g/mol, Mw = 2140 g/mol, Đ = 1.14 (determined by GPC, polystyrene standard calibration). Glass transition temperature was measured at 65.1 °C (determined by G" peak in rheological measurements at 1 Hz, cooling ramp 5°C/min).

# 2. Summary of the poly(@MSt-co-St) resins blended in SBR

Table S1. Summary of the poly(αMSt-co-St) resins blended in SBR.

Sample reference	m <sub>resin</sub> (g)	msbr (g)	Phr concentration
SBR/poly(aMSt-a-St)25	0.0146	0.0532	27
SBR/poly(aMSt-a-St)35	0.0187	0.0543	34
SBR/poly(aMSt-a-St)40	0.0241	0.0585	41
SBR/poly(aMSt-a-St)60	0.0307	0.0502	61
SBR/poly(aMSt-a-St)80	0.0378	0.0466	81
SBR/poly( $\alpha$ MSt-a-St)100	0.0452	0.0452	100
SBR/poly(aMSt-a-St)150	0.0575	0.0383	150

# 3. Evidence of the irreversibility of resin compatibilization in SBR



**Figure S1.** Evidence of the irreversibility of resin compatibilization in SBR after the heat treatment at 215 °C. Identical viscoelastic properties are retained after multiple heat–cool–heat measurements.

#### 4. AFM measurements

Conditions for the sample preparations and AFM measurements explained in the main text Section 2.5.



**Figure S2.** Evidence of the resin/rubber phase separation at 160 °C and after cooling back the sample at 25 °C. Scale bars on the images correspond to 2  $\mu$ m.



Figure S3. Supplemental images taken at each temperature. Evidences of the phase separation in other sample location. Scale bars on the images correspond to 2  $\mu$ m.



Figure S4. Nanomechanical properties of the different samples.

#### 5. SBR thermal stability



Figure S5. Thermal stability of neat SBR up to 225 °C.

### 6. NMR experiments

The liquid- and solid-state NMR spectra were measured on a Bruker Avance III HD 600 MHz (proton Larmor frequency) spectrometer. All the chemical shifts were referenced to tetramethylsilane (TMS) by referencing the residual d-chloroform signal to 7.26 ppm (liquid state) or setting the adamantane methylene signal to 37.77 ppm (solid state). For liquid-state NMR, ~ 20 mg of neat SBR was dissolved in ~600  $\mu$ L of deuterated toluene-d8 (99.6%D, Sigma Aldrich) in a 5 mm NMR tube. <sup>13</sup>C spectra with inverse gated <sup>1</sup>H decoupling were performed on solution samples with a repetition delay of 6 s and a total accumulation of 10,240 scans. Prior to the swollen solid-state NMR experiments, samples were swollen and washed at least 3 times in 100 mL of toluene. To obtain the solid-state <sup>13</sup>C NMR spectra, ~ 7 mg of crosslinked SBR (either with the resin CrR-SBR or with dicumyl peroxide CrDCP-SBR) was swollen by ~ 24 mg of toluene-d8 in a disposable HRMAS insert (B4493, Bruker), which was inserted into a 4 mm ZrO2 rotor and spun at 7 kHz spinning rate on a double-resonance MAS probe. One-dimensional <sup>13</sup>C direct polarization with high-power decoupling NMR experiments were performed on the CrR-SBR and CrDCP-SBR samples with a repetition delay of 4 s, and a total accumulation of 3288 and 2499 scans, respectively.

## 7. Radical SBR crosslinking with dicumyl peroxide (CrDCP-SBR)

The SBR blend was prepared with 0.1 phr of dicumyl peroxide (DCP). SBR was crosslinked in an oven at 160 °C during 20 min. The crosslinking density of CrDCP-SBR was measured by swelling test ( $V_e = 1.6 \times 10^{-3}$ ).



# 8. Viscoelastic behavior of resin-crosslinked SBR (CrR-SBR)

Figure S6. Comparison of the viscoelastic behavior of neat SBR versus CrR-SBR.