

Antiproliferative copper(II) complexes bearing mixed chelating ligands: structural characterization, ROS scavenging, antitumor activity and *in silico* studies

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Table S1. IR bands (cm^{-1}) together with corresponding assignments.

(1)	(2)	(3)	(4)	Assignments
-	3475m	-	-	$\nu(\text{H}_2\text{O})$
1615 m	1615 m	1605 m	1605 m	$\nu(\text{C=O})$
1590 m	1590 m	1590 m	1590 m	$\nu(\text{C=N})$
1560 s	1560 s	1560 m	1560 s	
-	1520 s	-	1520 s	$\nu_3(\text{NO}_3)$
	1320 m		1315 s	
1485 m	1485 m	1485 m	1485 m	$\nu(\text{C=C})$
1450 m	1450 m	1450 m	1450 m	
1425 m	1425 m	1435 m	1435 m	
1105 vs 1065 s	1020 m	1090 vs	1025 m	$\nu_3(\text{ClO}_4)/ \nu_1(\text{NO}_3)$
960 m	-	-	-	$\nu_1(\text{ClO}_4)$
-	850 m	-	850 m	$\nu_2(\text{NO}_3)$
-	715 m	-	715 m	$\nu_4(\text{NO}_3)$
650 w 620 m	-	620 m	-	$\nu_4(\text{ClO}_4)$
445 w	445 w	450 w	460 w	$\nu(\text{Cu-N})$
430 w	430 w	440 w	435 w	$\nu(\text{Cu-O})$

s-strong; m-medium, w-weak.

Table S2. Thermal data for complexes in air atmosphere. Representative mass peaks appear on Supplementary Figure S2.

Compound	Step	Thermal effect	Temperature range / °C	$\Delta m_{\text{exp}} / \%$	Identified product by EGA with corresponding <i>m/z</i> values
(1)	1.	Exothermic	125-292	37.75	moieties from bzac (C_6H_5 (77, 78), 105, (51, 50)), moieties from phen ($C_6H_{13}N_2$ (112, 113), 63, 50), perchlorate (Cl (35), CHCl (49), ClO (51, 52), H_2O (17, 18), CO_2 (44))
	2.	Exothermic	292-352	5.07	moieties from bzac and phen (77, 78, 51, 50), H_2O (17, 18), CO_2 (44)
	3.	Exothermic	352-569	41.77	perchlorate (Cl (35), CHCl (49), ClO (50, 51), Cl_2 (70)), H_2O (17, 18), CO_2 (44), still some organic moiety
(2)	1.	Endothermic	25-155	2.96	H_2O (17, 18)
	2.	Exothermic	155-214	7.73	moieties from bzac (C_4H_5O (68), C_6H_5 (77, 78), 51), H_2O (17, 18), CO_2 (44))
	3.	Exothermic	214-304	23.99	moieties from bzac (C_4H_5O (68), C_6H_5 (77, 78), 51), moieties from phen ($C_6H_{13}N_2$ (112, 113), 63, 50), nitrate (NO (30), NO_2 (46)), H_2O (17, 18), CO_2 (44))
	4.	Exothermic	304-509	48.65	moieties from bzac (C_4H_5O (68), C_6H_5 (77, 78), moieties from phen ($C_6H_{13}N_2$ (112, 113), 63, 50), nitrate (NO (30), NO_2 (46)), H_2O (17, 18), CO_2 (44))
(3)	1.	Exothermic	251-320	54.36	moieties from bzac (C_4H_5O (68), C_6H_5 (77, 78), 105, 85, 43, (51, 50)), moieties from bpy (C_5H_6N (78, 79), (51, 52, 50)), perchlorate (Cl (35), CHCl (49), ClO (51, 52)), H_2O (17, 18), CO_2 (44))
	2.	Exothermic	320-600	31.65	perchlorate (Cl (35), CHCl (49), ClO (51, 52), Cl_2 (70)), CO_2 (44))
(4)	1.	Exothermic	150-278	37.75	moieties from bzac (C_4H_5O (68, 69), C_6H_5 (77, 78), 60, 51), nitrate (NO (30), NO_2 (46)), H_2O (17, 18), CO_2 (44))
	2.	Exothermic	278-304	4.70	moieties from bzac (C_6H_5 (77, 78), 105, 51, 43) moieties from bpy (C_5H_6N (78, 79), (51, 52, 50)), nitrate (NO (30), NO_2 (46)), H_2O (17, 18), CO_2 (44))
	3.	Exothermic	304-337	11.29	still some organic moiety, nitrate (NO (30), NO_2 (46)), H_2O (17, 18), CO_2 (44))
	4	Exothermic	337-538	30.03	nitrate (NO (30), NO_2 (46)), H_2O (17, 18), CO_2 (44))

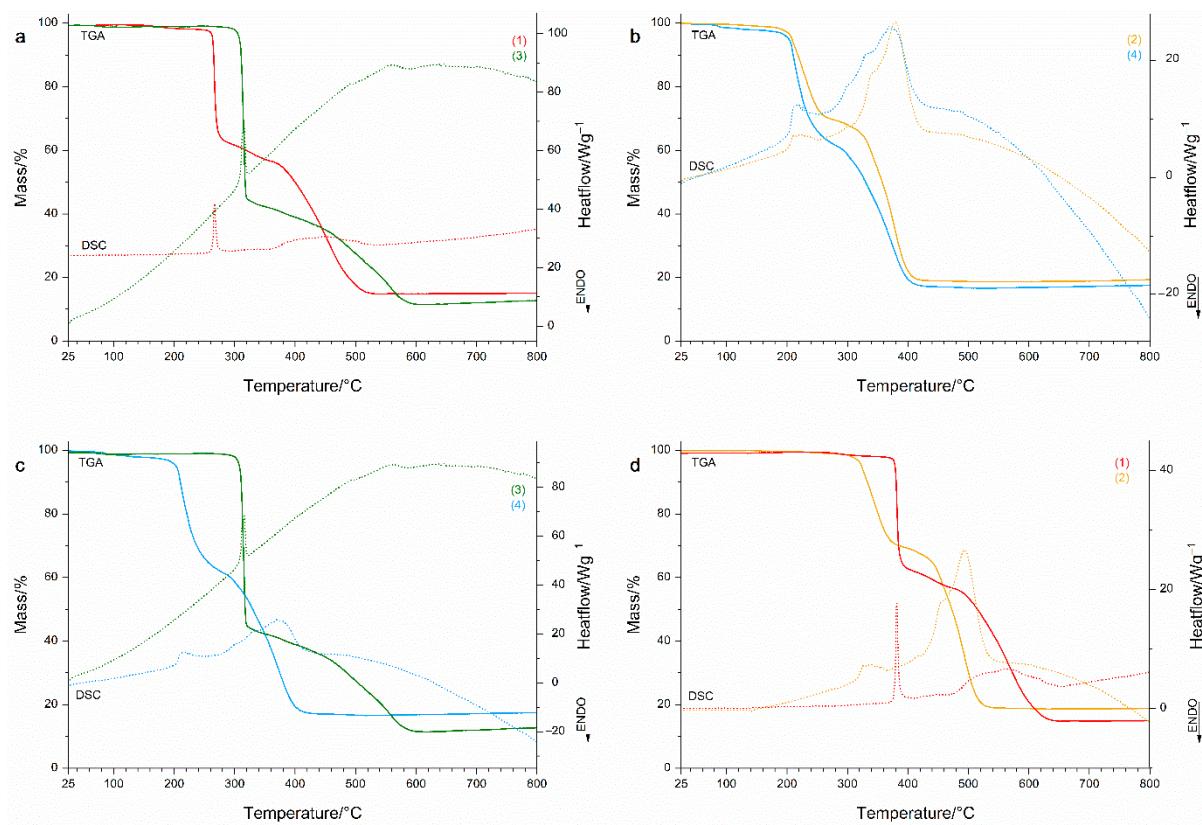


Figure S1. TG-DSC graphs in air atmosphere from 25 °C to 600 °C in various combinations: **a)** complexes with perchlorate ligand, **b)** complexes with nitrate ligand, **c)** complexes with bpy ligand, and **d)** complexes with phen ligand.

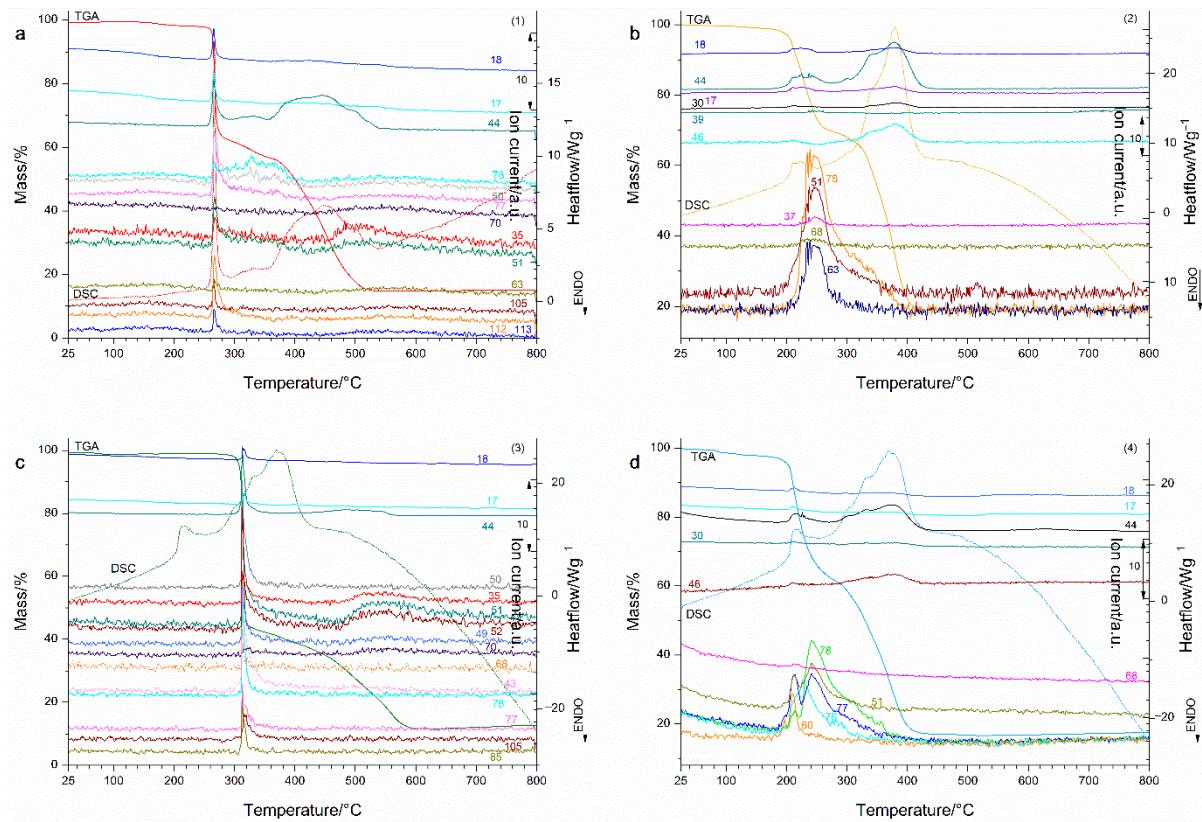


Figure S2. TG/DSC–MS graphs of all complexes from 25 to 600 °C in air atmosphere: **a)** complex (1), **b)** complex (2), **c)** complex (3), and, **d)** complex (4).