

New (Iso)Quinoliny-Pyridine-2,6-dicarboxamide G-Quadruplex Stabilizers. A Structure-Activity Relationship Study

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1. General

All the reagents were purchased from Sigma-Aldrich and Merck and used without further purification.

NMR-Spectra were recorded on a Bruker 300 Ultrashield 300MHz, using 300 MHz scan for ¹H-NMR spectra and 75 MHz for ¹³C.

Purity of compounds submitted to biophysical, biochemical or biological tests were in all cases > 90 % as determined by HPLC–MS (HPLC Waters Alliance 2695 coupled to a Photodiode Array Detector Waters 996 PDA and a Triple Quadrupole MS Micromass Quattro Micro API), using a Phenomenex Luna 5u HILIC 200A (100x3mm; 5µm) column and a ACN-HCOOH 0.5% (40:60) elution system at 0.2 ml/min.

2. FRET melting experiments

FRET melting assays were performed on a 7300 RT-PCR equipment from Applied Biosystems. The test compound solutions (50 µL) were distributed across 96-well RT-PCR plates (PCR-96-FLT-C, Axygen, Inc.). Experiments were performed in cacodylate buffer, 60 mM K⁺, pH 7.4, with 0.2 µM of oligonucleotide.

Standard and labelled HPLC-purified oligonucleotides were purchased from STABVIDA (Portugal), and sequences are depicted in table **S1**.

Table S1. Sequences used in FRET-melting experiments.

Name	Sequence	Topology	T _m (°C)
<i>k</i> -RAS	5'-FAM-AGGGCGGTGTGGGAAGAGGGA-TAMRA-3'	Parallel G4	49.0 ± 0.2
<i>h</i> -Telo	5'-FAM-GGGTTAGGGTTAGGGTTAGGG-TAMRA-3'	Hybrid G4	56.9 ± 0.2
<i>T</i> -Loop	5'-FAM-TATAGCTATATTTTTTATAGCTATA-TAMRA-3'	dsDNA	53.2 ± 1.0
26merA 26merB	5'-CAATCGGATCGAATTCGATCCGATTG-3' 5'-GTTAGCCTAGCTTAAGCTAGGCTAAG-3'	Self-hybridizing duplex DNA	

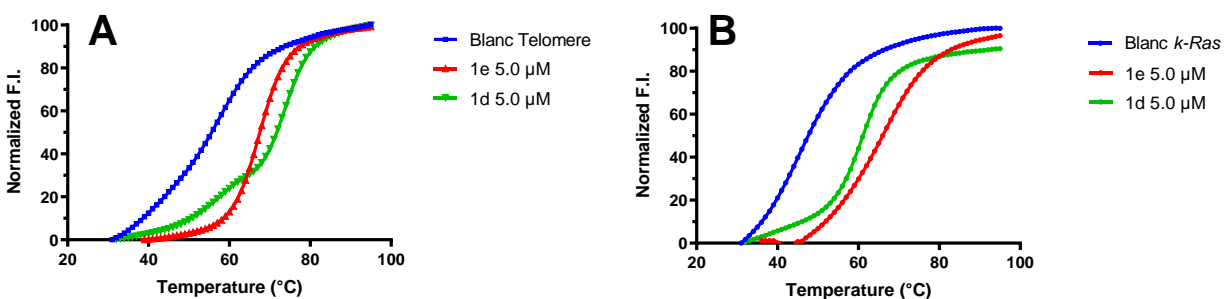


Figure S1. FRET-Melting curves of *h*-Telo (A, blue curve) and *k*-RAS (B, blue curve) in presence of compound **1e** (red curves) and **1d** (green curves) at 5 µM concentration (25 eq.).

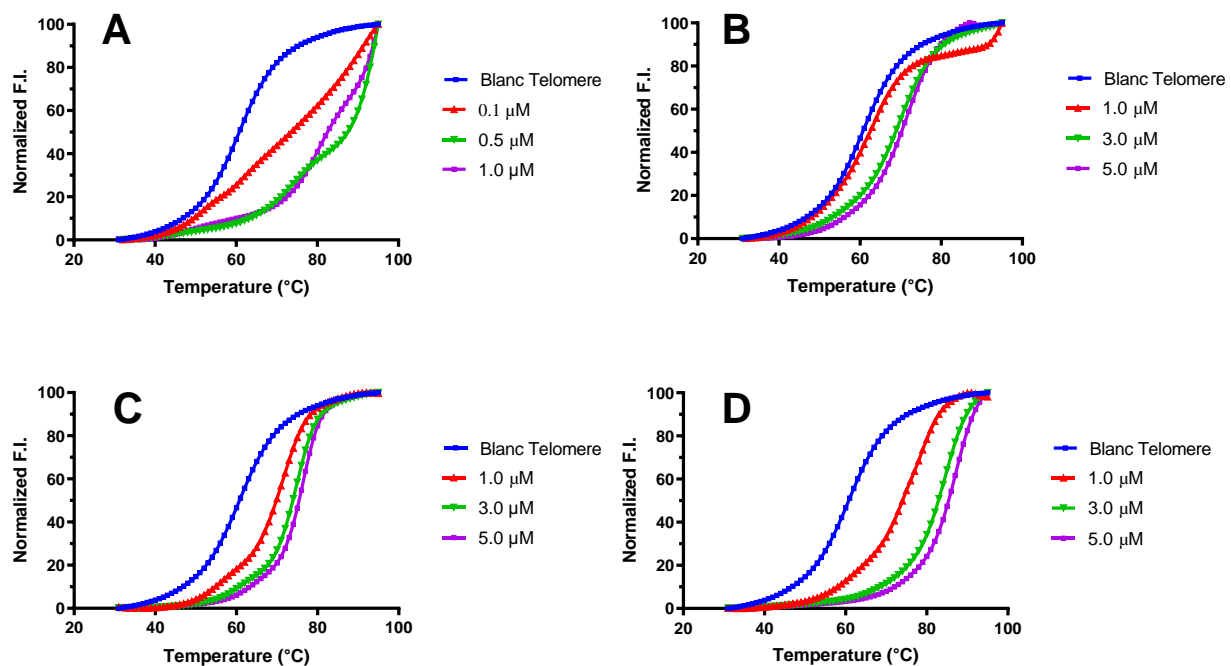


Figure S2. FRET-Melting curves of h-Telo (blue curves), in presence of increasing concentration of compounds **2a** (A), **2b** (B), **2c** (C) and **2d** (D).

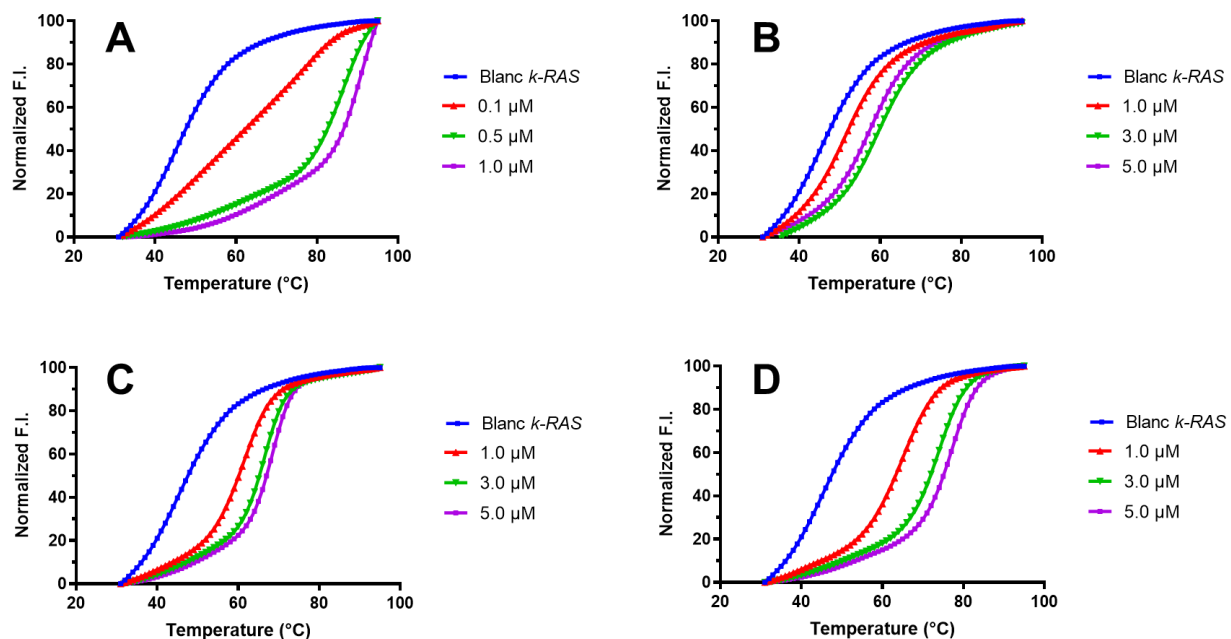


Figure S3. FRET-Melting curves of k-RAS (blue curves), in presence of increasing concentration of compounds **2a** (A), **2b** (B), **2c** (C) and **2d** (D).

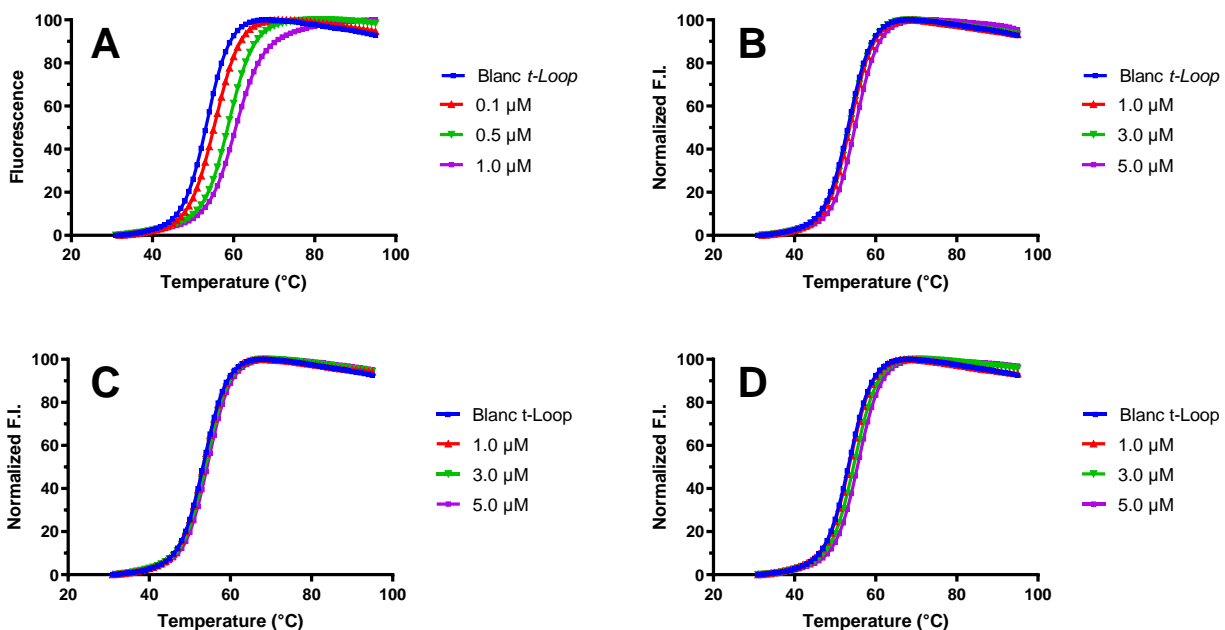


Figure S4. FRET-Melting curves of double-stranded t-Loop (blue curves), in presence of increasing concentration of compounds **2a** (A), **2b** (B), **2c** (C) and **2d** (D).

3. CD Titrations

CD titrations and melting assays were performed on Jasco J-815 spectropolarimeter equipped with a Peltier-type temperature control system (model CDF-426S/15), using an instrument scanning speed of 200 nm/min with a response time of 1 s in wavelengths ranging from 200 to 340 nm. CD-melting experiments were performed with 10 μ M of oligonucleotide in 20 mM lithium cacodylate buffer supplemented with KCl and LiCl as described in Material and Methods (& 4.4), keeping a constant ionic strength of 100 mM. The DNA sequences used for the experiments, and their respective melting temperatures are depicted in table S2.

Table S2. Sequences used in CD experiments.

Name	Sequence	Topology	T _m (°C)
<i>k</i> -RAS	5'-AGGGCGGTGTGGGAAGAGGGA-3'	Parallel G4	48.3 \pm 0.2
<i>H</i> -Telo	5'-GGGTTAGGGTTAGGGTTAGGG-3'	Hybrid G4	59.6 \pm 0.9
<i>c</i> -MYC	5'-TGGGGAGGGTGGGGAGGGTGGGGAAGG-3'	Parallel G4	50.4 \pm 1.9

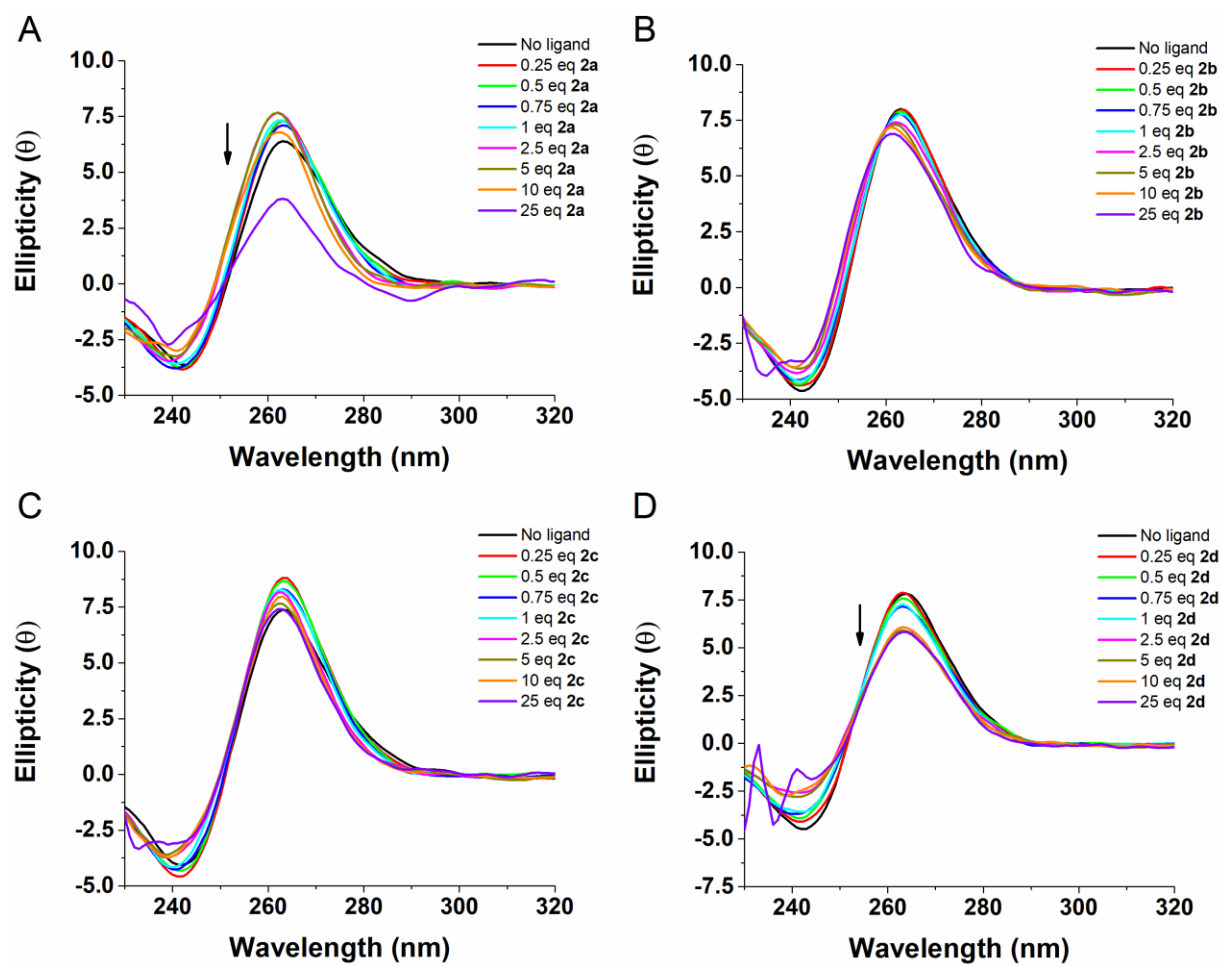


Figure S5. CD titration of c-MYC in presence of increasing equivalents (0-25) of compound 2a (A), 2b (B), 2c (C), 2d (D), at a concentration of 10 μ M, performed in 20 mM lithium cacodylate containing 100 mM LiCl.

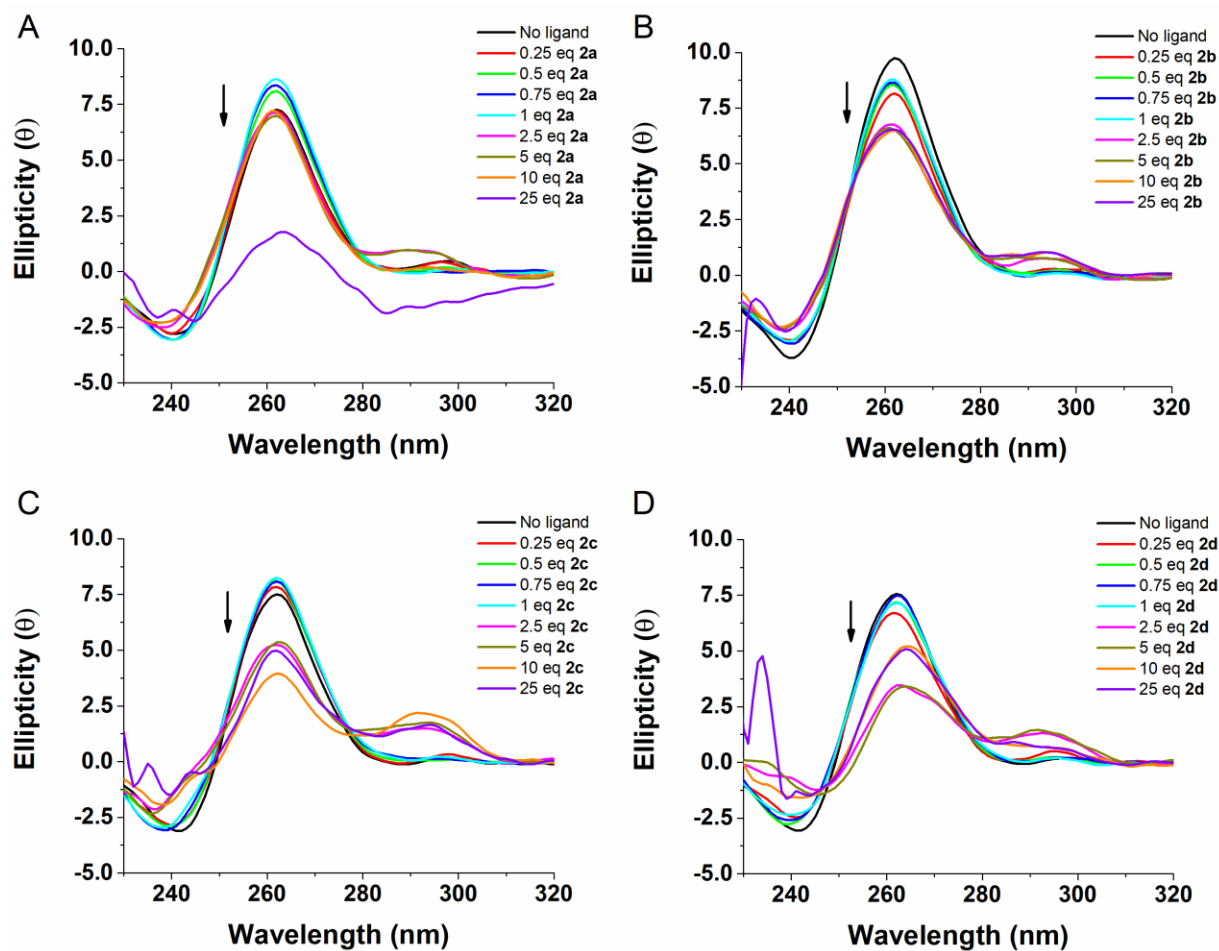


Figure S6. CD titration of *k*-RAS in presence of increasing equivalents (0-25) of compound 2a (A), 2b (B), 2c (C), 2d (D), at a concentration of 10 μ M, performed in 20 mM lithium cacodylate containing 50 mM KCl and 50 mM LiCl.

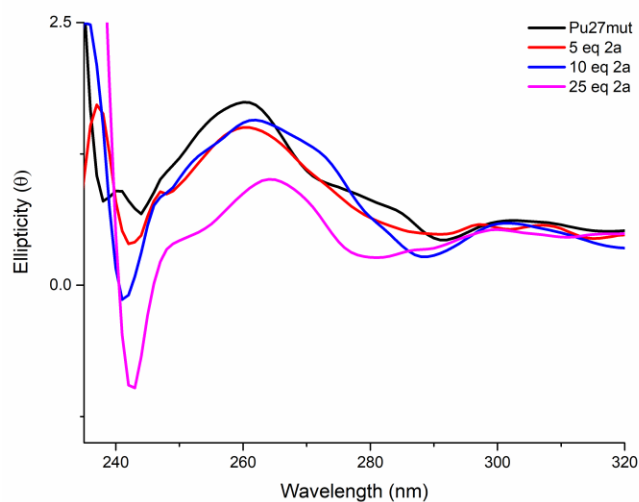


Figure S7. CD titration of c-MYC Pu27mut (sequence in table S3) in presence of increasing equivalents of compound 2a, performed in ThermoPol buffer (20mM Tris-HCl, 10mM $(\text{NH}_4)_2\text{SO}_4$, 10mM KCl, 2mM MgSO_4 , 0.1% Triton[®] X100, pH 8.8).

4. PCR-Stop assay

The primer sequences used in PCR-stop assay are depicted in table S3. Sequence-design was adapted from [1].

Table S3. Sequences used in PCR-stop assay

Name	Sequence
Pu27	5'-TGGGGAGGGTGGGGAGGGTGGGGAAGG-3'
Pu27mut	5'-TGGGGAGGGTGGAAAGGGTGGGGAAGG-3'
Pu27REV	5'-ATCGAATCGCTTCTCGTCCTTCCCCA-3'
Pu27mut2	5'-TGGGGAGGGTGGAAAGGGTTTGAAGG-3'
Pu27REV2	5'-ATCGATCGCTTCTCGTCCTTCCAAA-3'

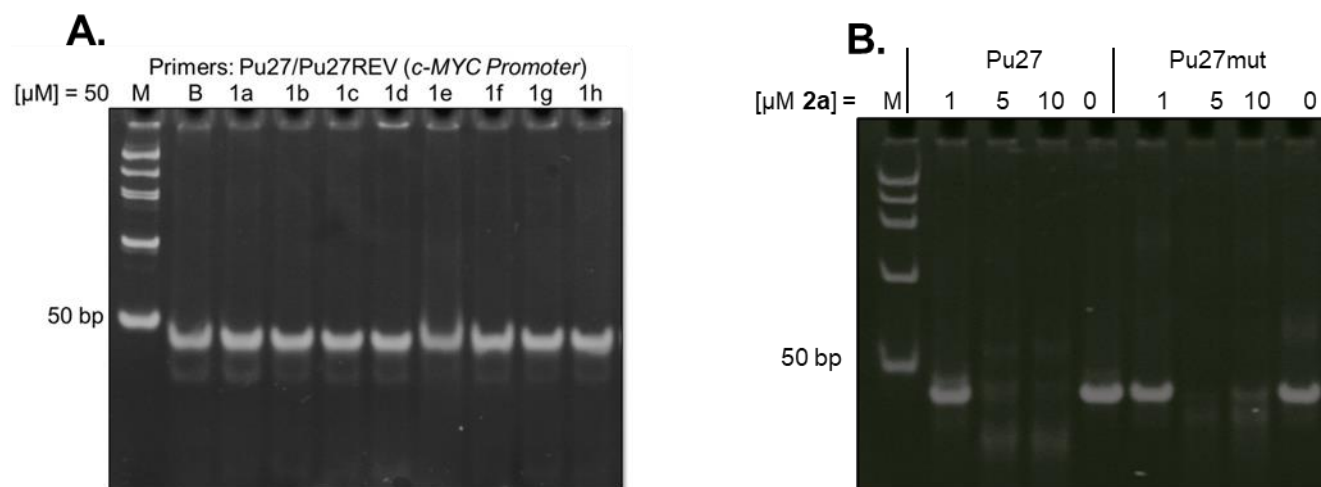


Figure S8. A) Results of PCR-stop assay with compounds **1a-h** at 50 μ M and c-MYC gene promoter Pu27. B) Results of PCR-stop assay with compound **2a** and c-MYC gene promoter Pu27 or with mutated c-MYC gene promoter (Pu27mut). In absence of compound, the 43 bp PCR products from constructions with Pu27 is formed.

5. MM/PBSA calculations

The equation used to determine the binding free energy of the ligands to the quadruplex is expressed as:

$$\Delta G_{bind} = \Delta G_{complex}^{vdW} + \Delta G_{complex}^{ele} + \Delta G^{polar} + \Delta G^{nonpolar} \quad (\text{eq 1})$$

with $V_{complex}^{vdW}$ and $V_{complex}^{ele}$ being the quadruplex–ligand van der Waals and electrostatic interaction energies, respectively. The polar and the nonpolar contributions are expressed as

$$\Delta G^{polar} = G_{complex}^{polar} - (G_{protein}^{polar} + G_{ligand}^{polar}) \quad (\text{eq 2})$$

and

$$\Delta G^{nonpolar} = G_{complex}^{nonpolar} - (G_{protein}^{nonpolar} + G_{ligand}^{nonpolar}) \quad (\text{eq 3})$$

The Poisson–Boltzmann (PB) equation for the polar term is evaluated using the following equation:

$$\nabla \cdot [\epsilon(r) \nabla \cdot \phi(r)] - \epsilon(r) k(r)^2 \sinh[\phi(r)] + \frac{4\pi \rho^f(r)}{k_B T} = 0 \quad (\text{eq 4})$$

where $\phi(r)$ corresponds to the ligand's electrostatic potential, $\epsilon(r)$ the dielectric constant, and $\rho^f(r)$ the fixed charge density. Since the polar solvation energy is known to depend on the chosen value for the dielectric constant ϵ_{solute} of the complex [2], several values (2, 4 and 8) were evaluated to inspect its effect on predicted ΔG_{bind} values. By default, ϵ_{solute} is set to 2, but ultimately, we have used a value of 8 which showed to be the most adequate to use for our system. All the other settings in `g_mmpbsa` were left unaltered (i.e., use of atomic radii as proposed by Bondi [3], linear PB equation solver, 0.05 nm grid resolution, and smoothed van der Waals surface).

To estimate nonpolar solvation energies, we have used a linear dependence of $G_{nonpolar}$ on the SASA, expressed as follows [4]:

$$G^{nonpolar} = \gamma_{surf}SASA + b \quad (\text{eq 5})$$

where γ_{surf} is related to the surface tension of the solvent. A solvent probe radius of 0.14 nm was used to determine the SASA.

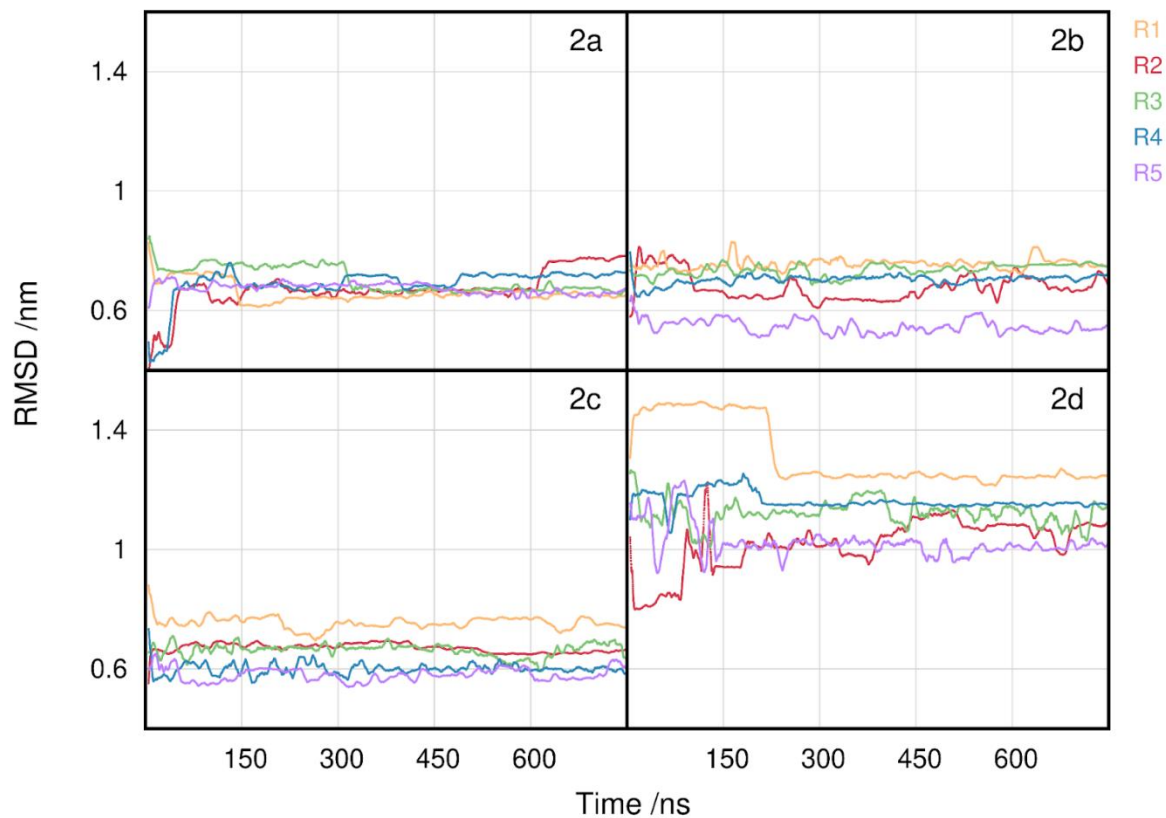


Figure S9. Root mean square deviation (RMSD) of the complex k-RAS + ligand, throughout the simulation time for each replicate simulation.

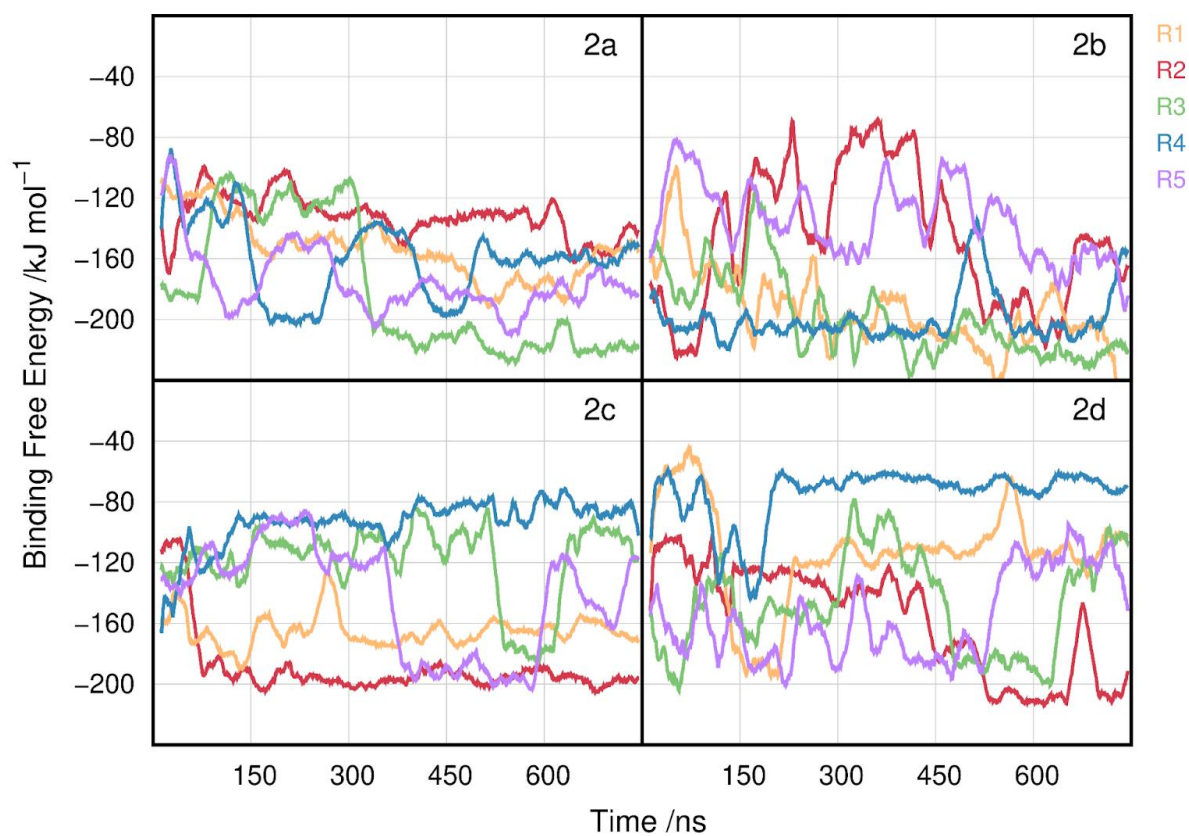


Figure S10. Variation of the MM/PBSA binding free energy between the different ligands and k-RAS for each replicate simulation.

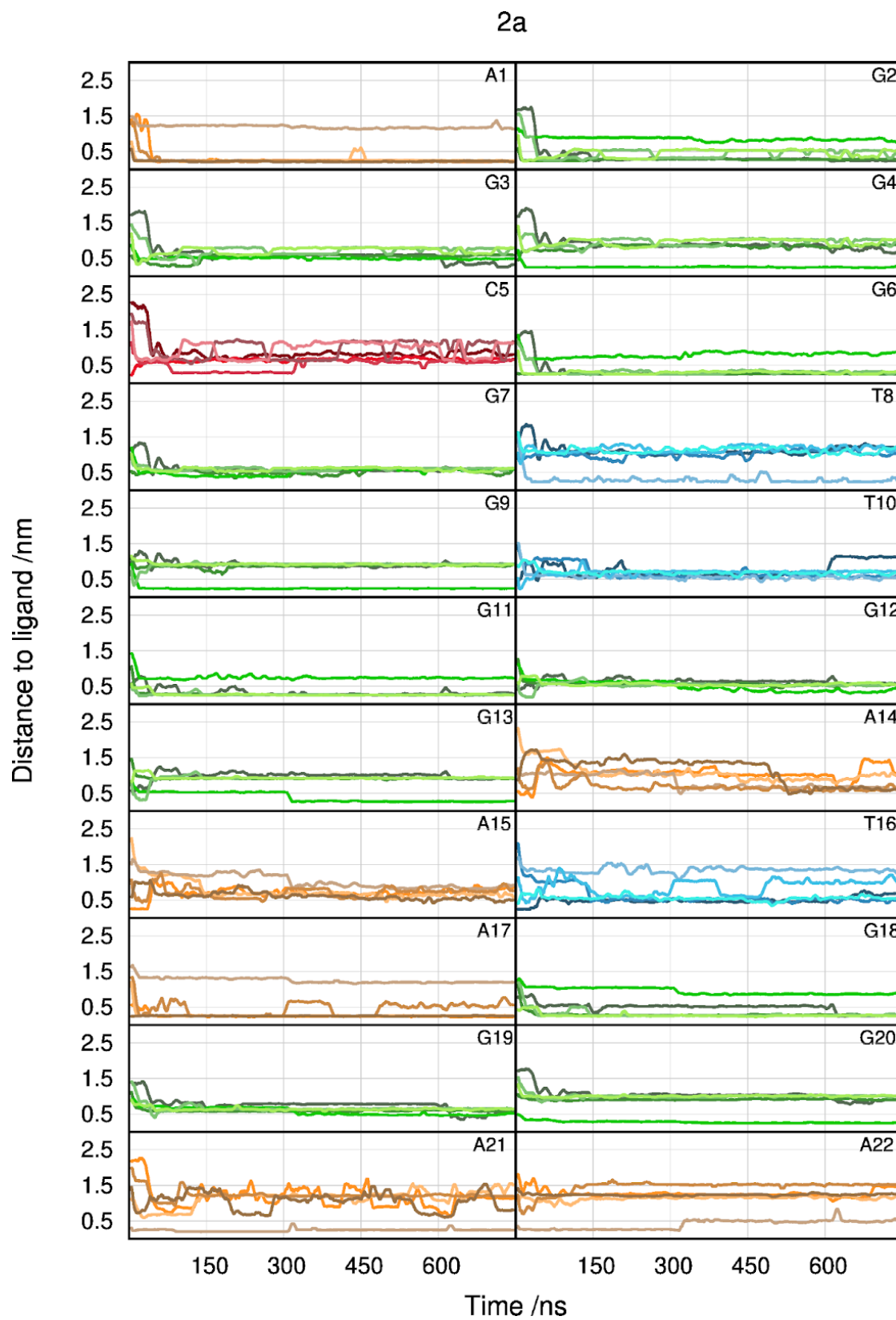
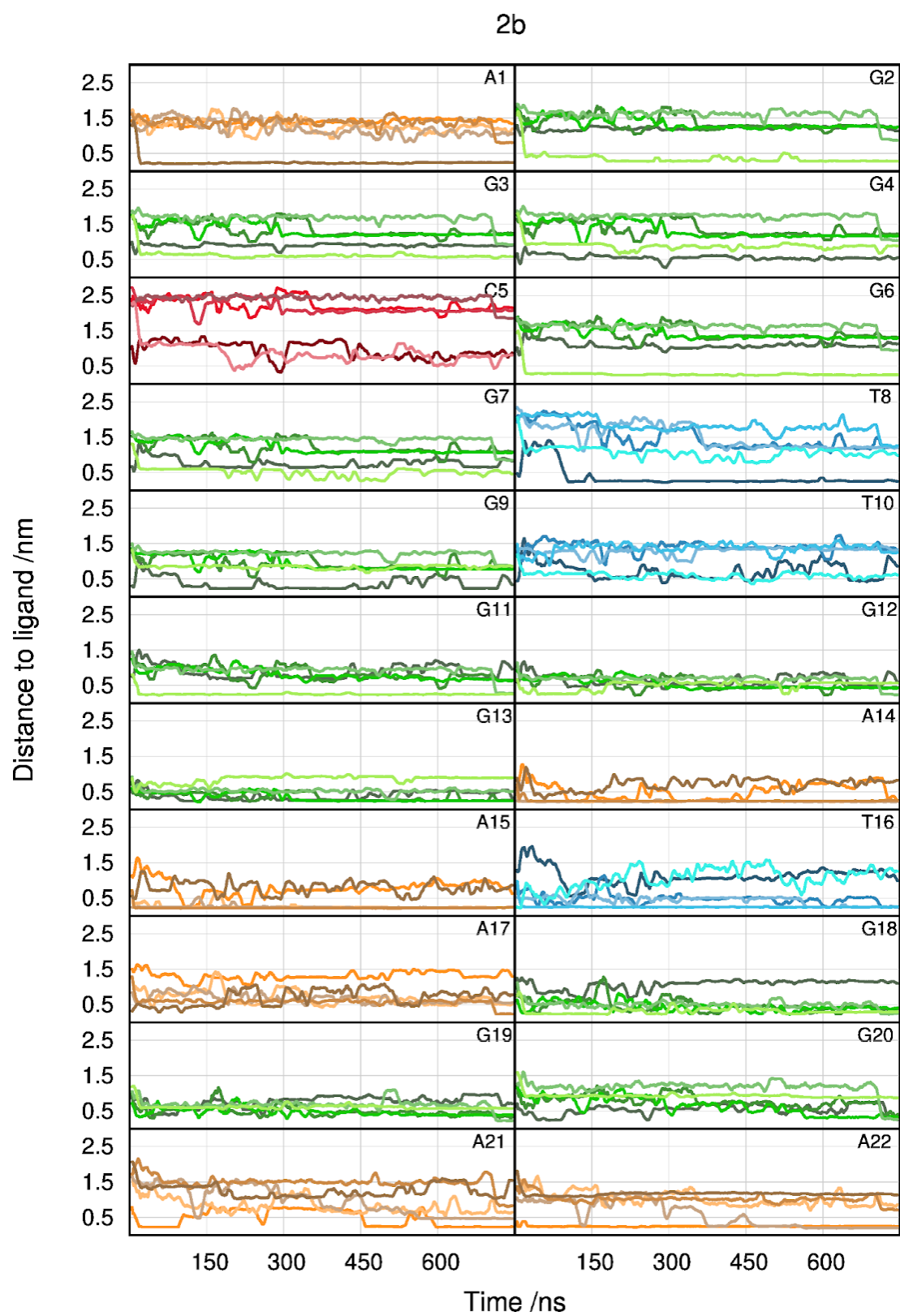


Figure S11. Minimum distance of compound 2a to each pair of bases of k-RAS for all replicate simulations.



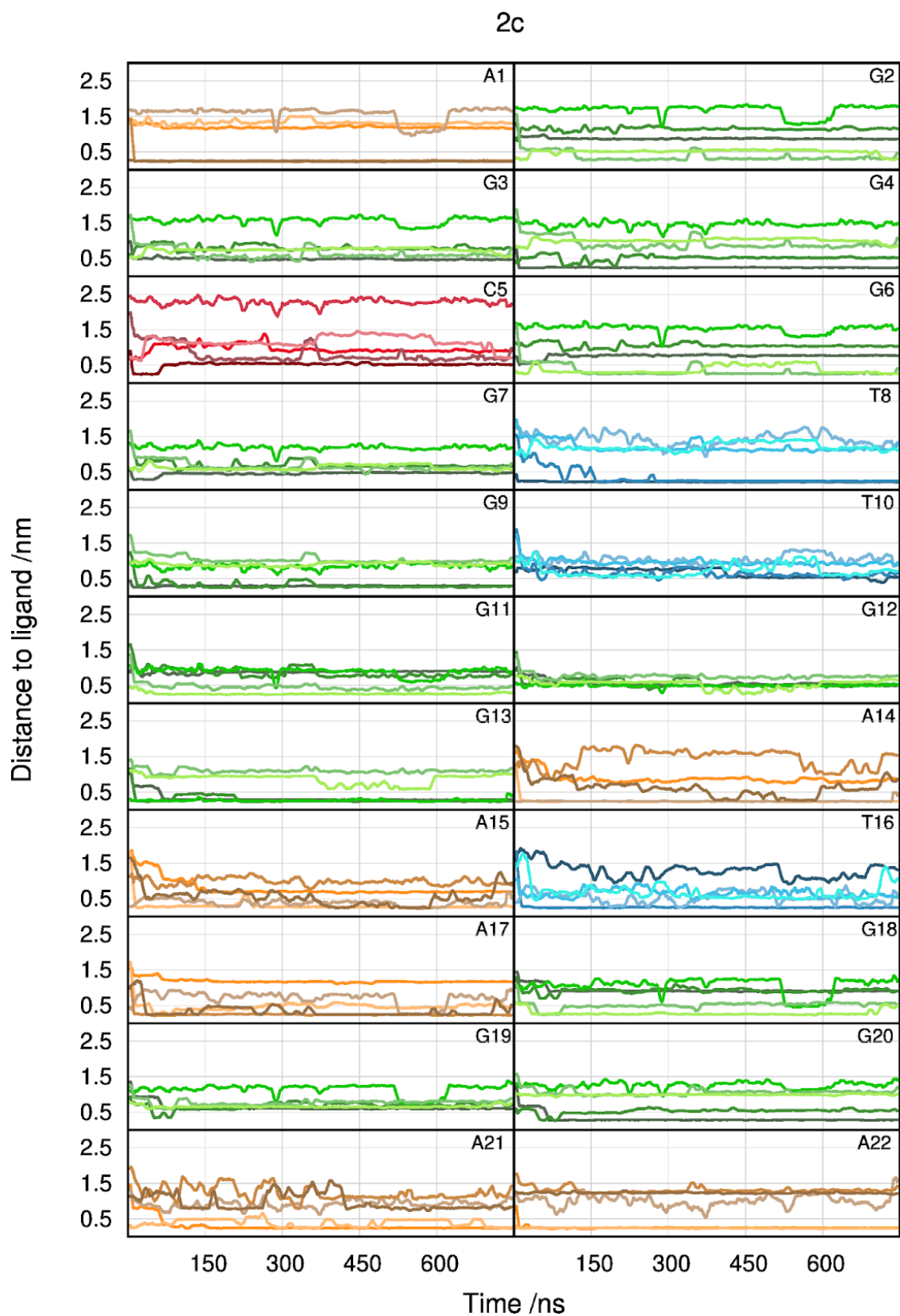


Figure S13. Minimum distance of compound 2c to each pair of bases of k-RAS for all replicate simulations.

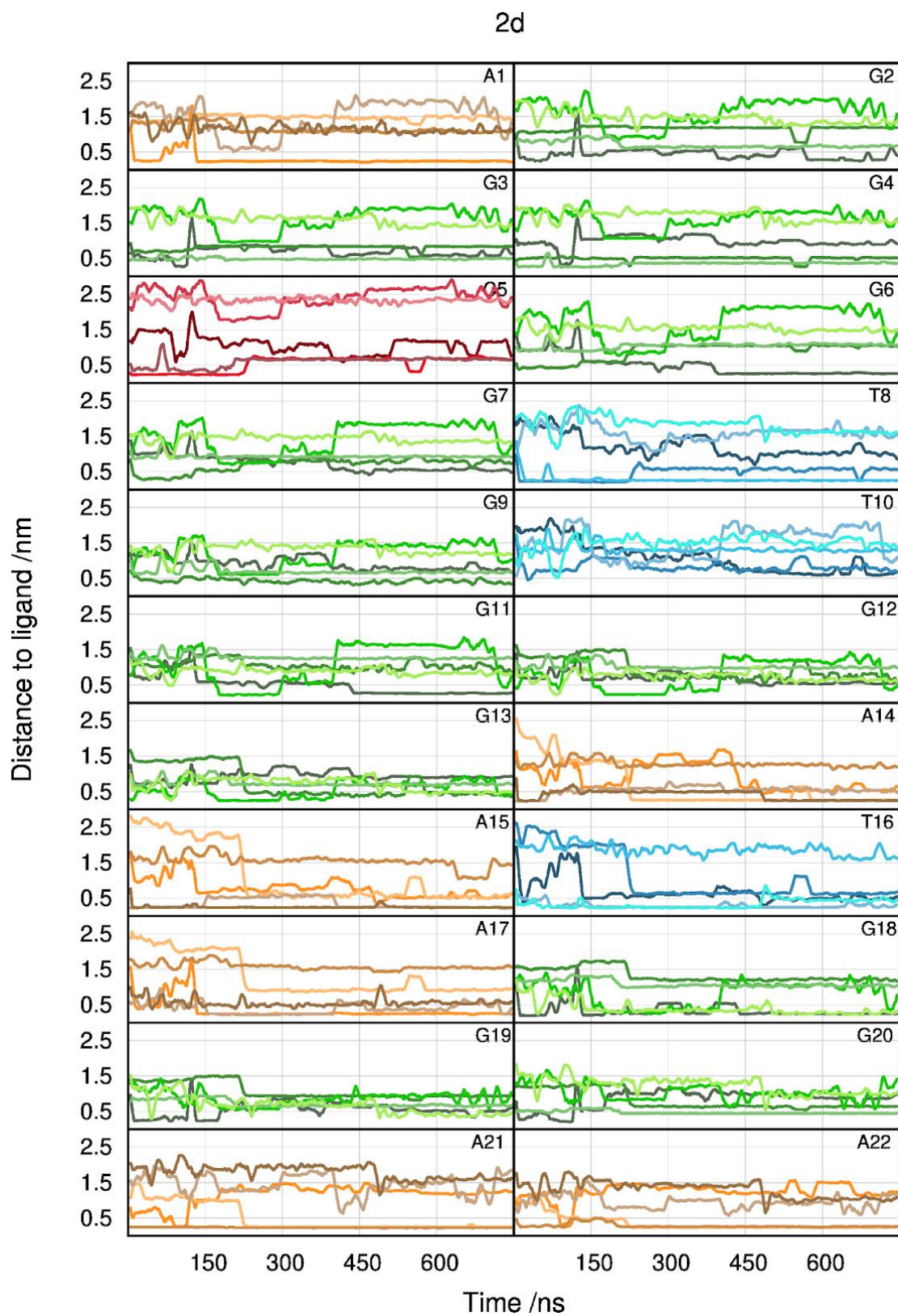


Figure S14. Minimum distance of compound 2d to each pair of bases of k-RAS for all replicate simulations.

6. NMR spectra

NMR-Spectra were recorded on a Bruker 300 Ultrashield 300MHz, using 300MHz scan for ^1H -NMR spectra and 75 MHz for ^{13}C .

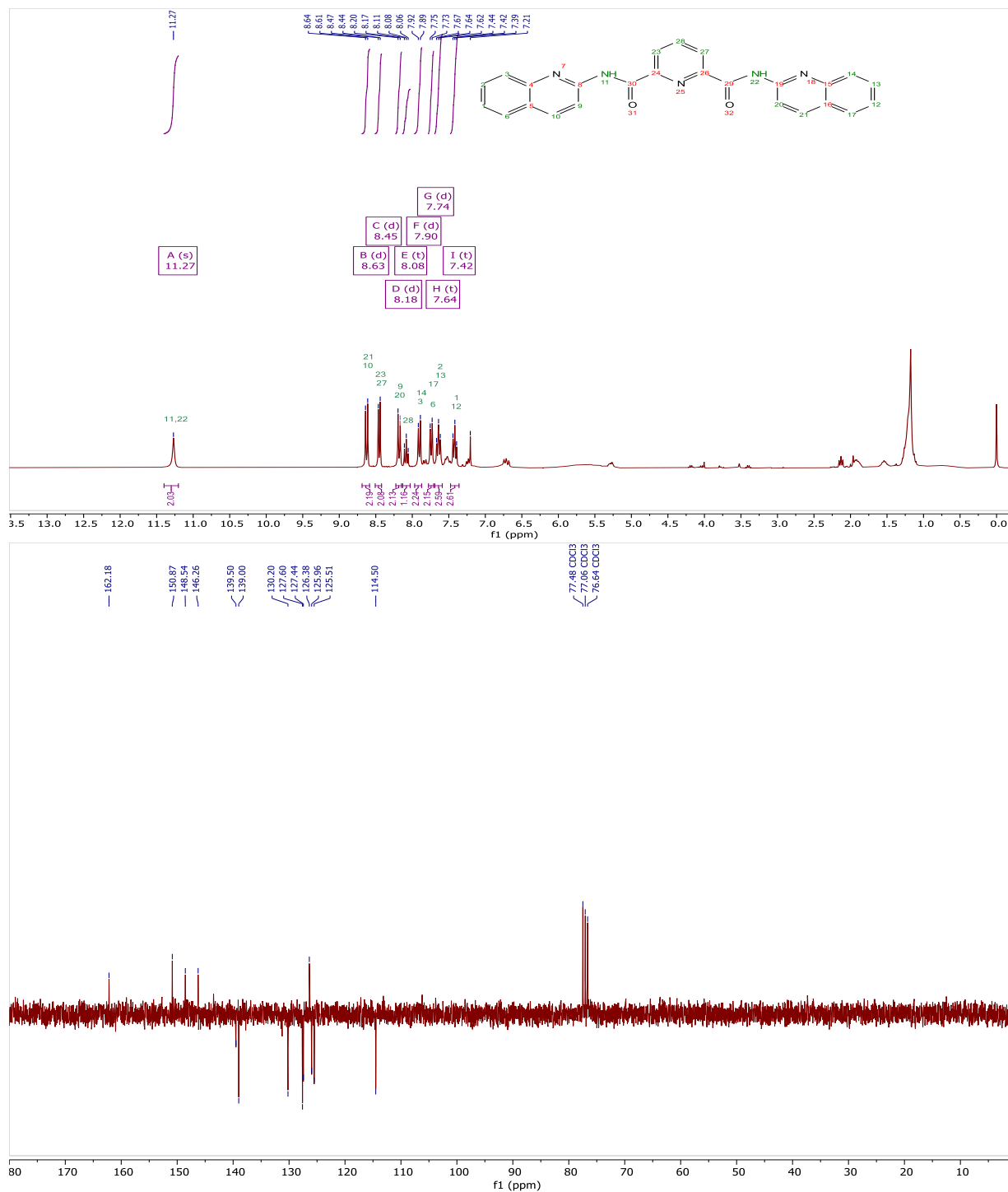


Figure S15. ^1H -NMR (top) and ^{13}C -NMR (bottom) spectra of compound **1a**.

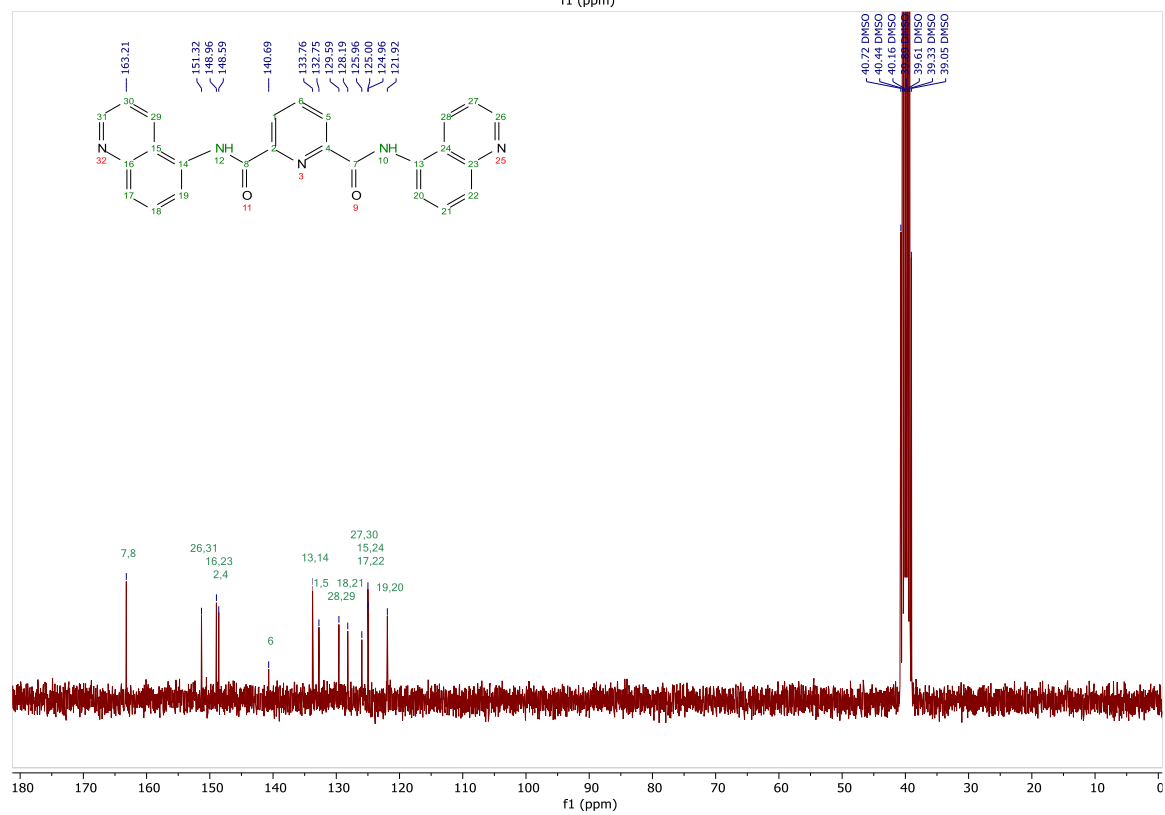
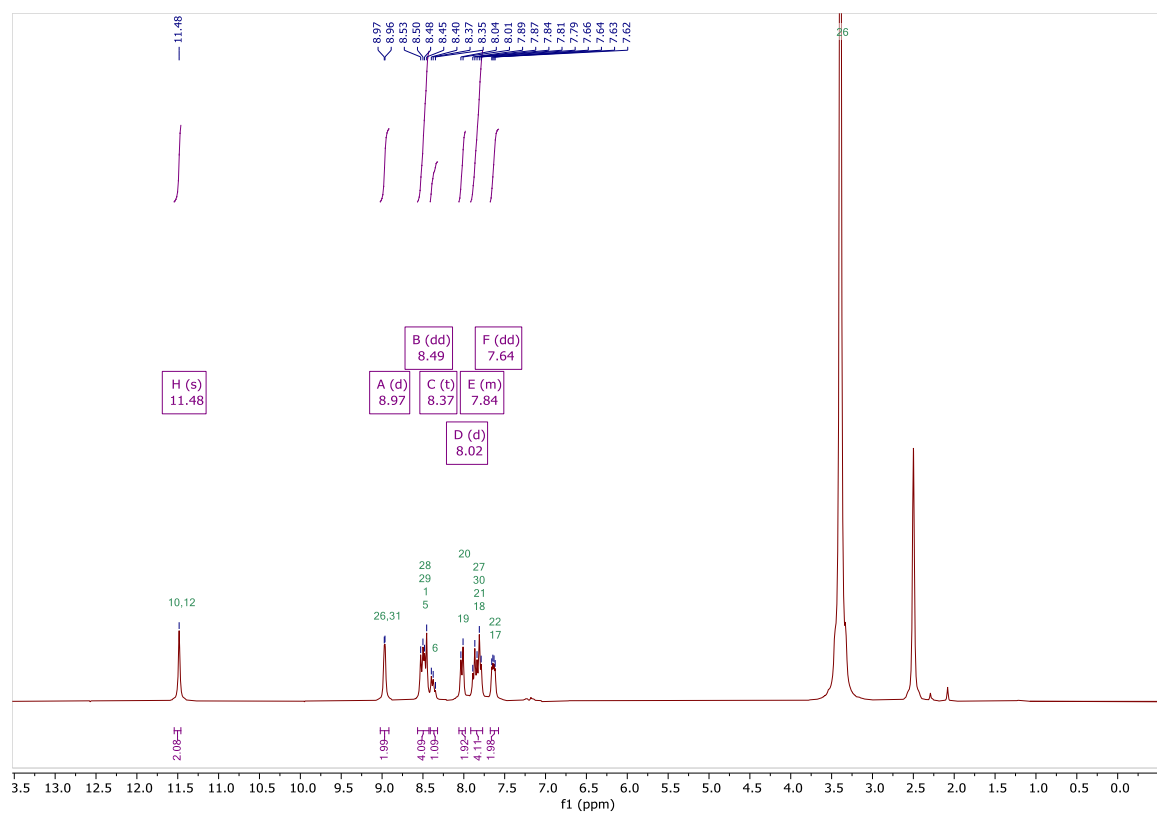


Figure S16. ¹H-NMR (top) and ¹³C-NMR (bottom) spectra of compound **1b**.

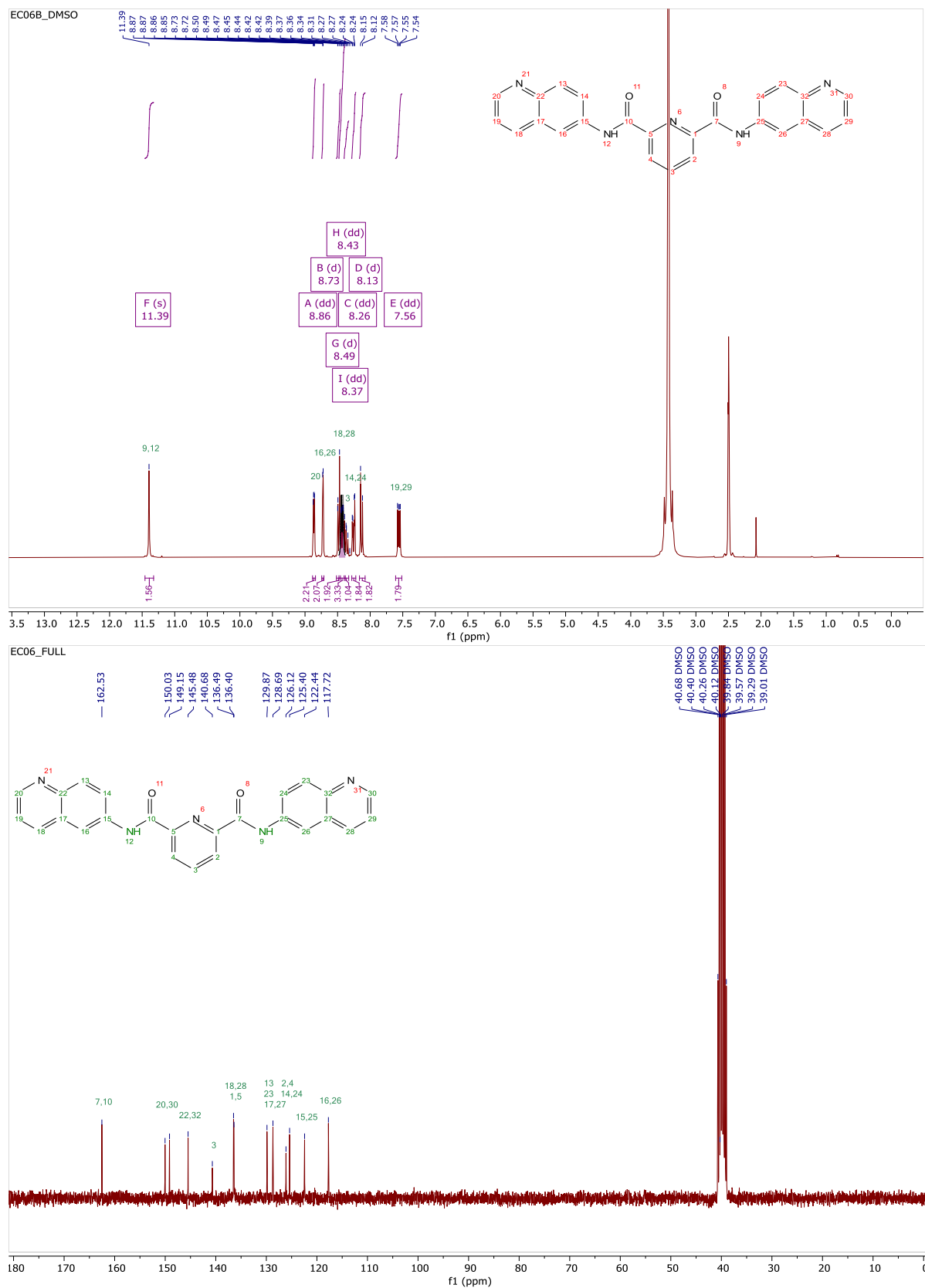


Figure S17. ^1H -NMR (top) and ^{13}C -NMR (bottom) spectra of compound **1d**.

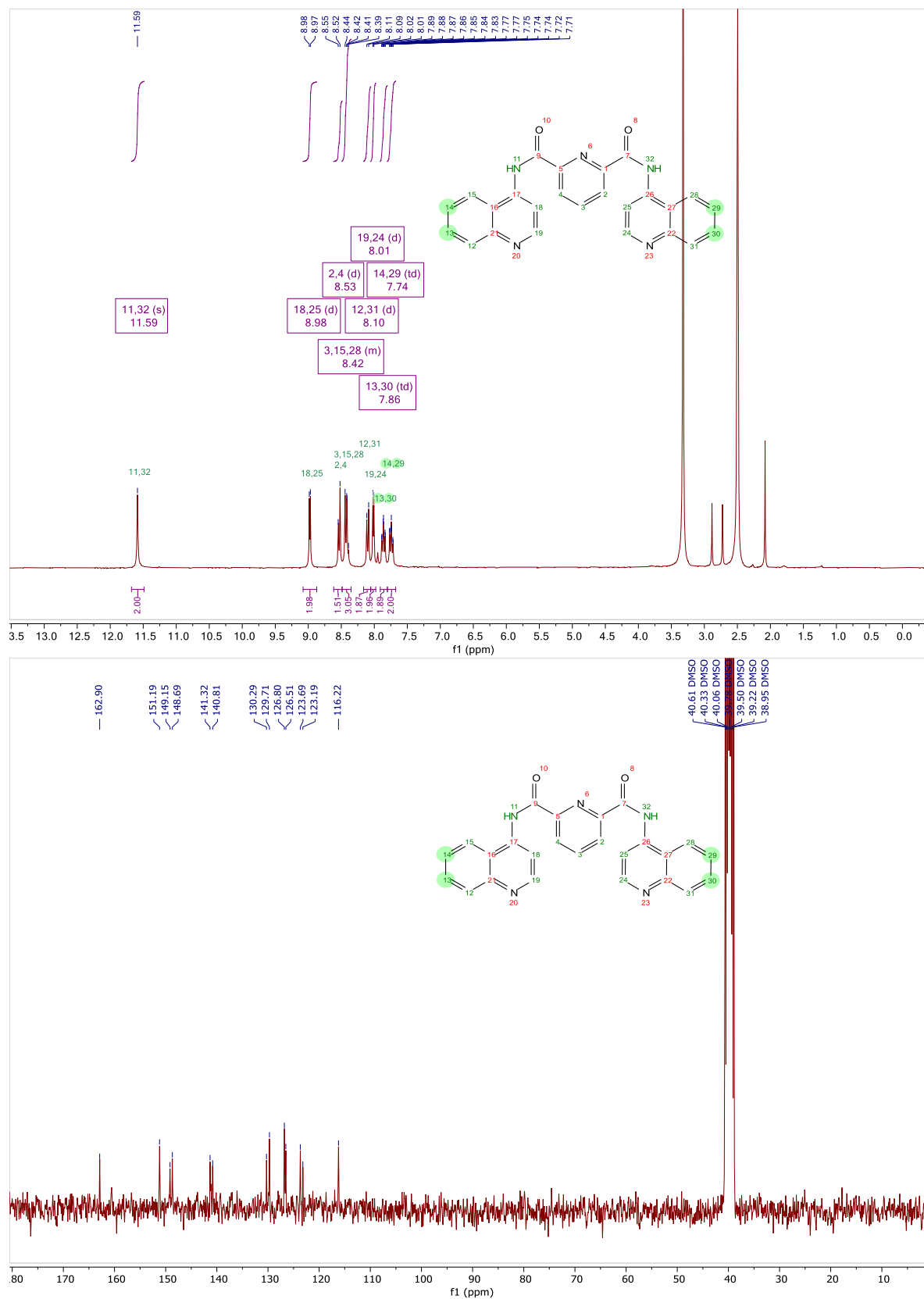


Figure S18. ¹H-NMR (top) and ¹³C-NMR (bottom) spectra of compound **1e**.

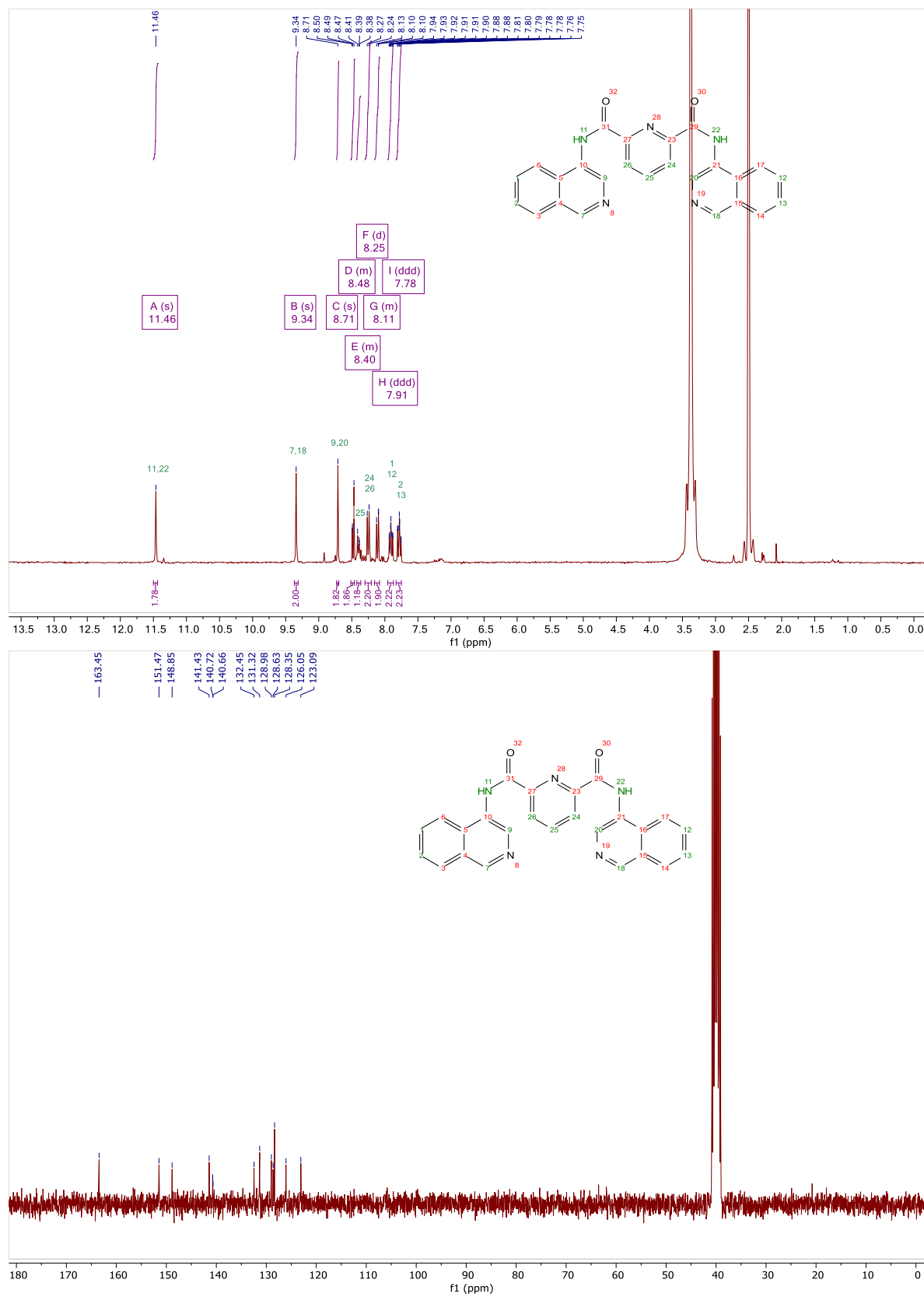


Figure S20. ¹H-NMR (top) and ¹³C-NMR (bottom) spectra of compound **1g**.

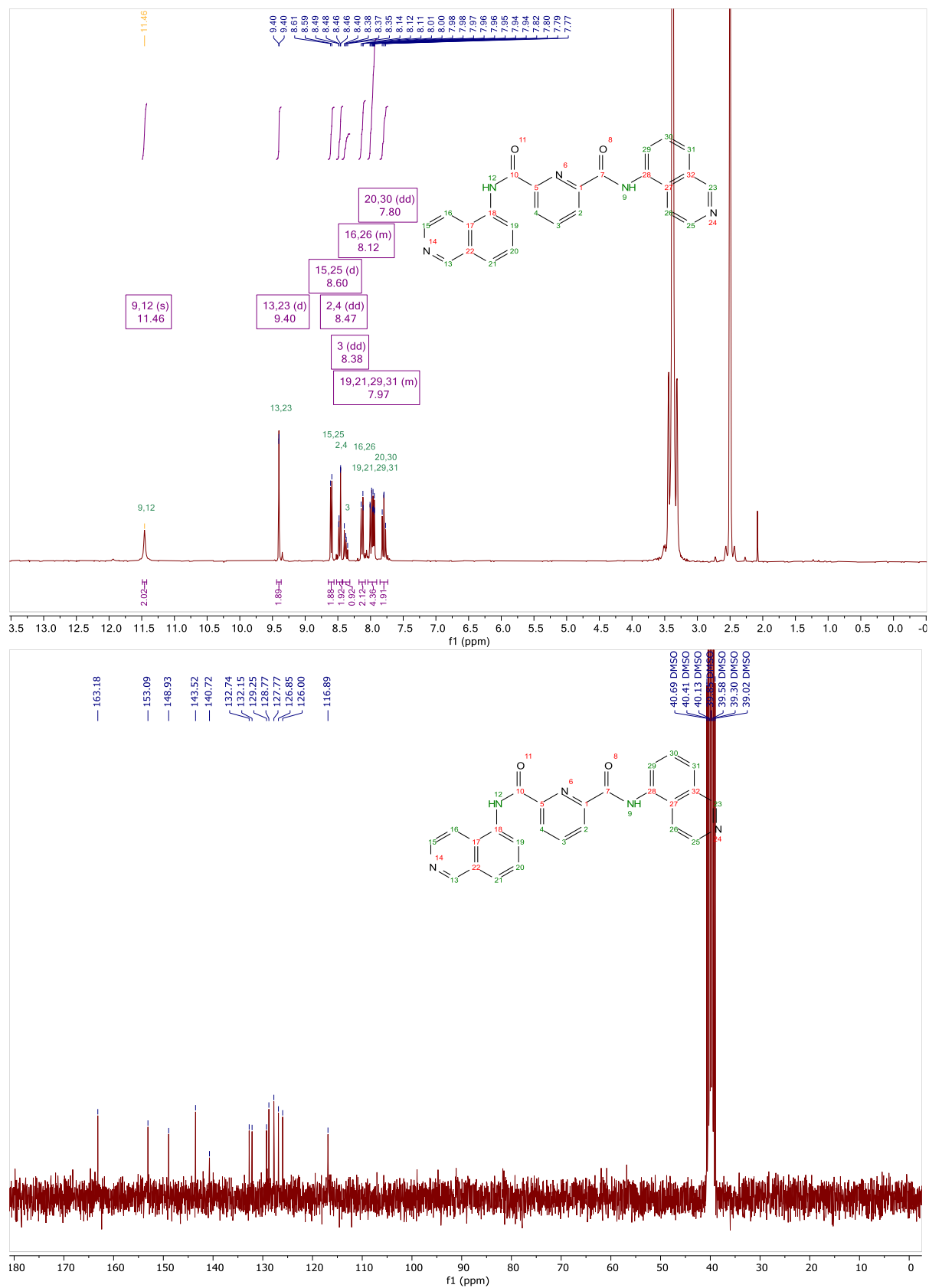


Figure S21. ¹H-NMR (top) and ¹³C-NMR (bottom) spectra of compound **1h**.

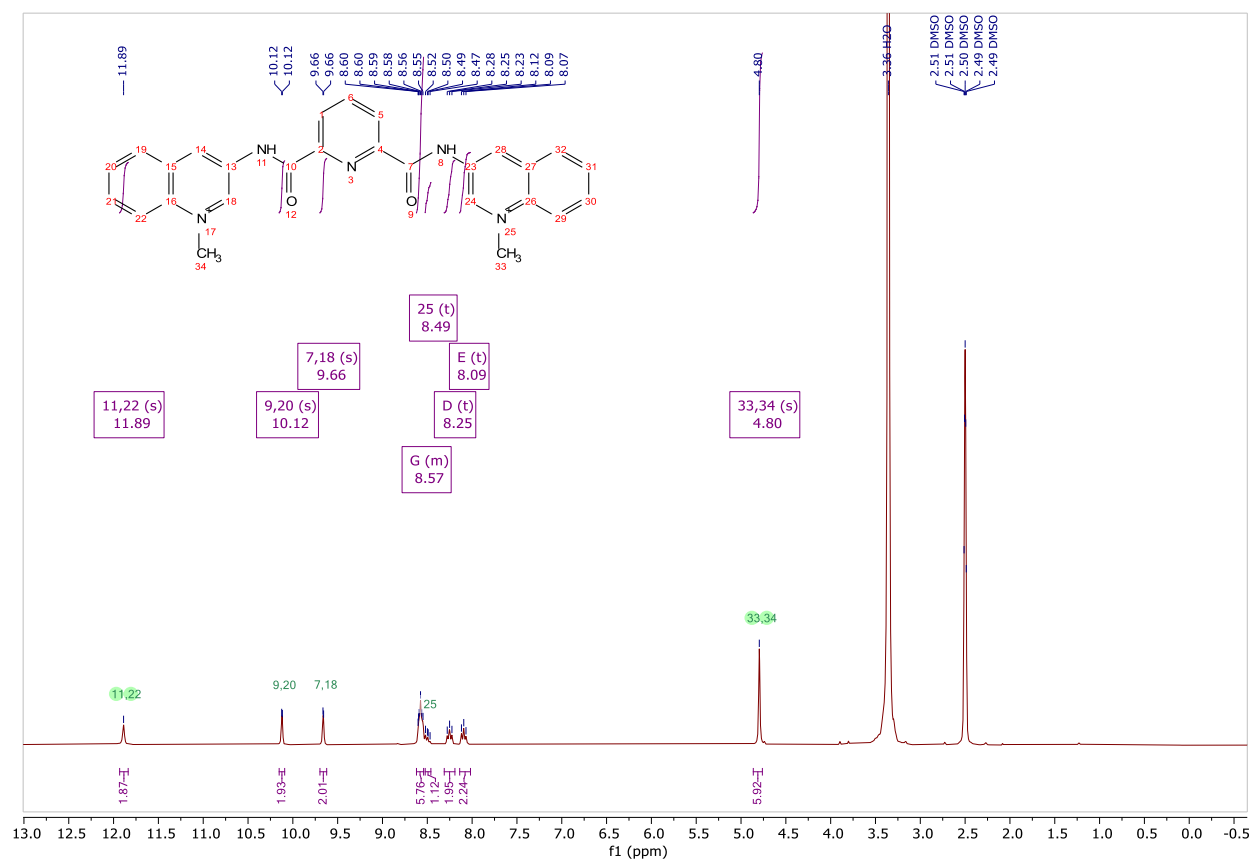


Figure S22. ¹H-NMR Spectrum of **2a**.

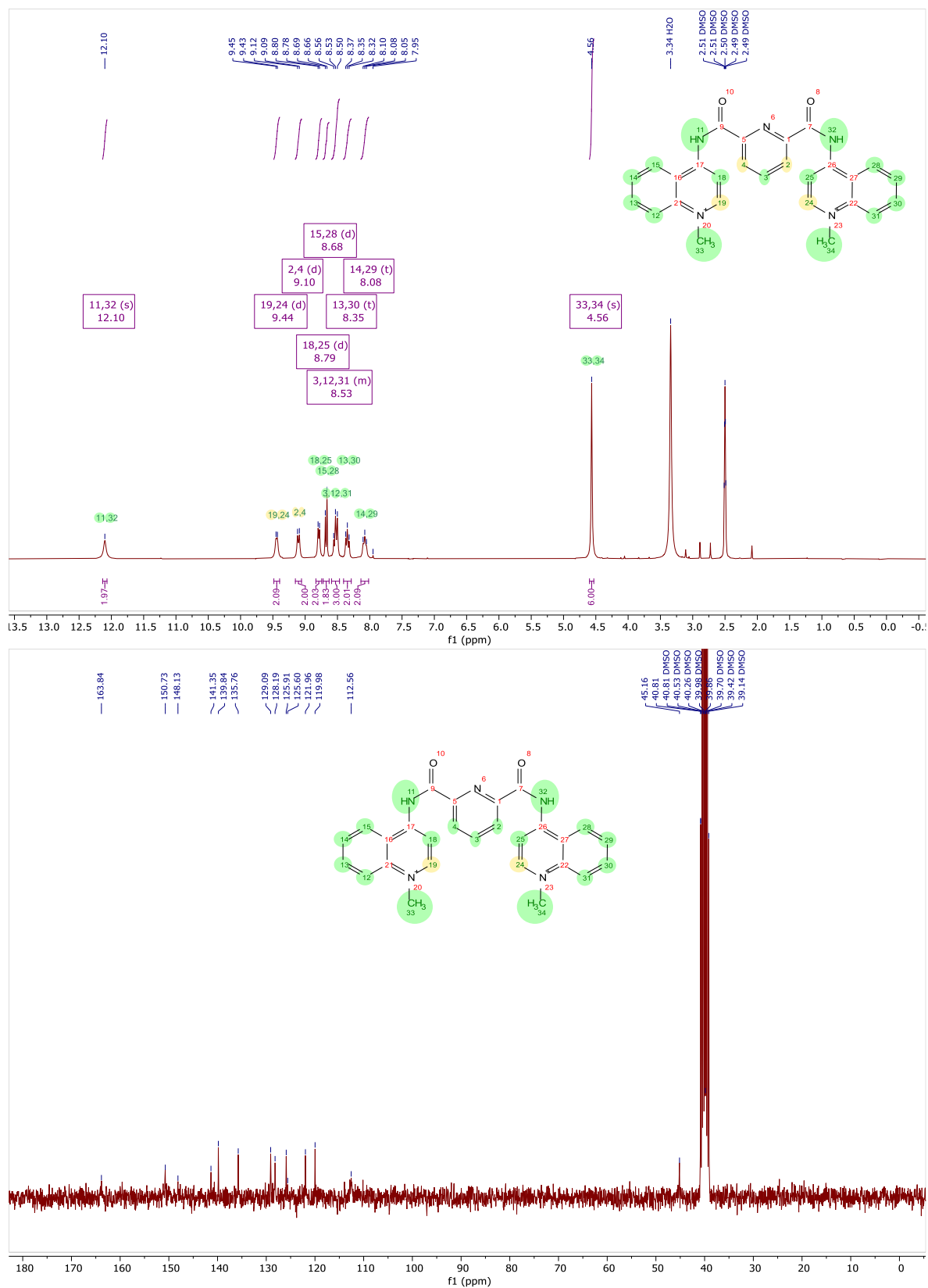


Figure S23. ¹H-NMR (top) and ¹³C-NMR (bottom) spectra of compound **2b**.

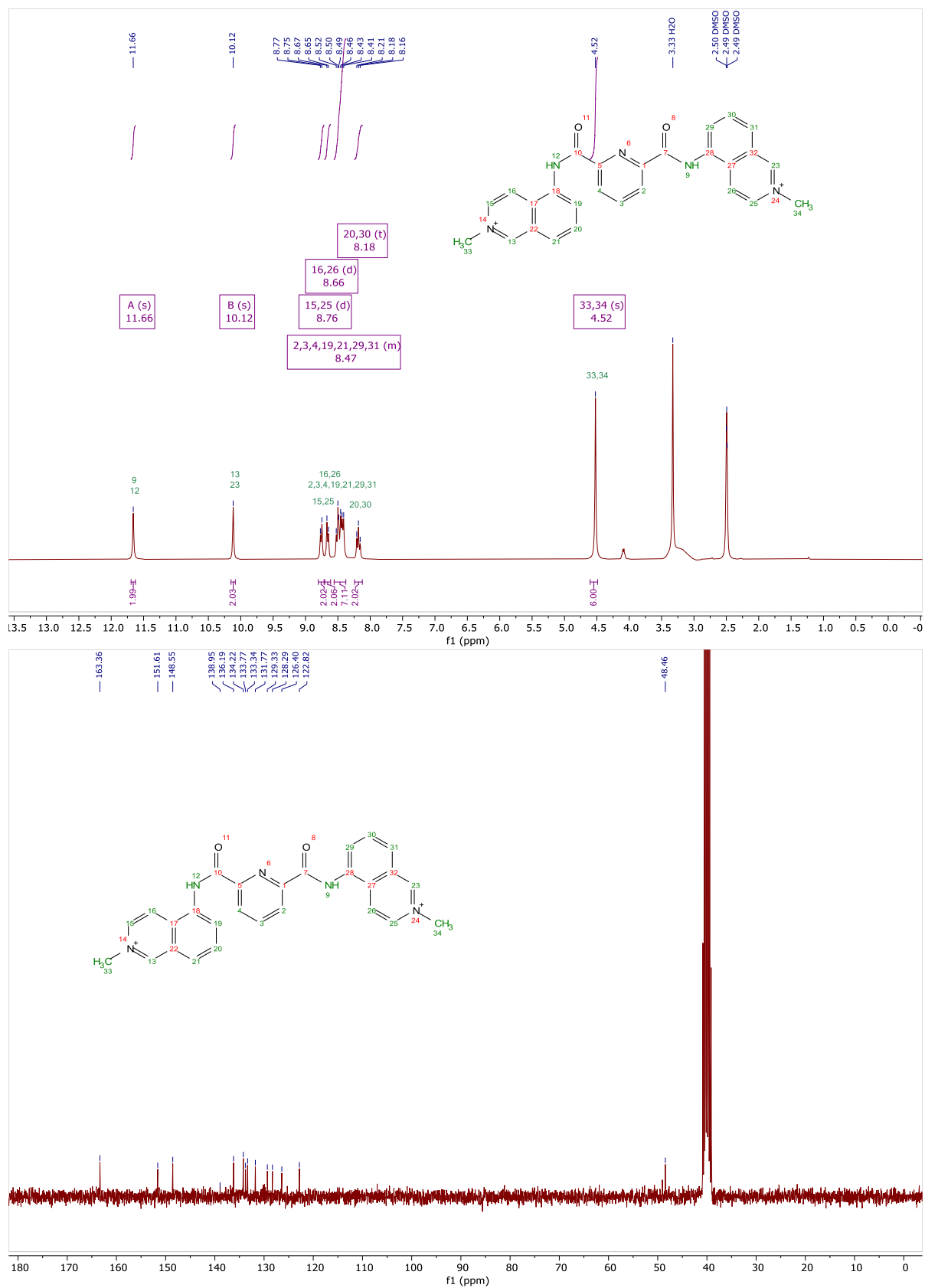


Figure S24. ¹H-NMR (top) and ¹³C-NMR (bottom) spectra of compound 2c.

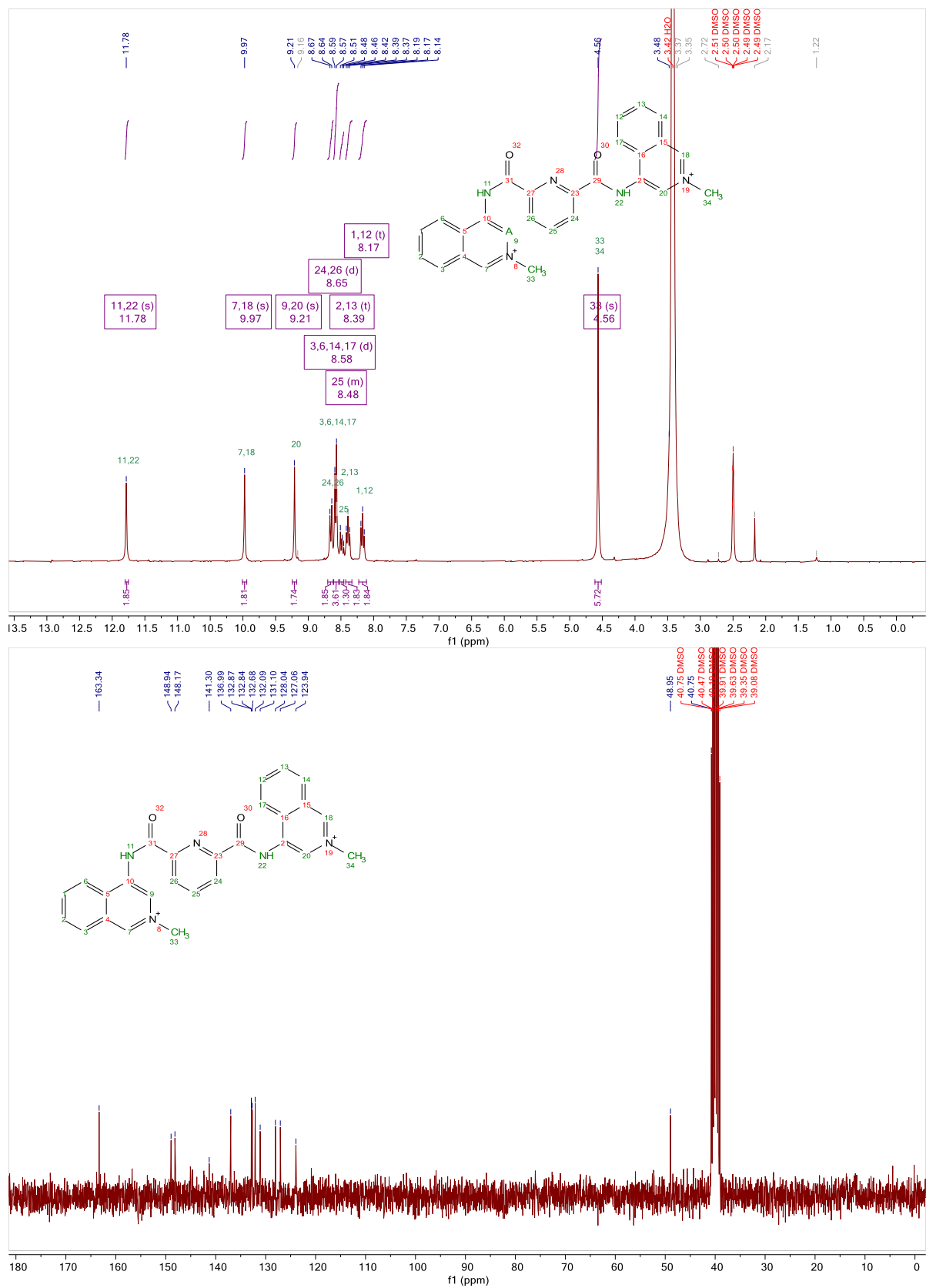


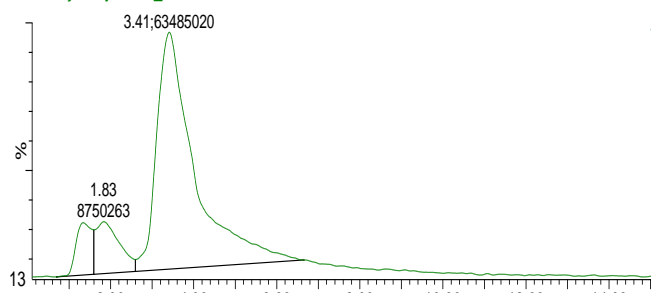
Figure S25. ¹H-NMR (top) and ¹³C-NMR (bottom) spectra of compound 2d.

7. LC-MS Characterization

Compound **1a**

Sample 176_HILIC 2_ 60%H2O20-Jul-201708:52:32

LCMS serviço 20 julho17_1



LCMS serviço 20 julho17_1 194 (3.395) Cm (184:220)

1: Scan ES+ 5.36e6

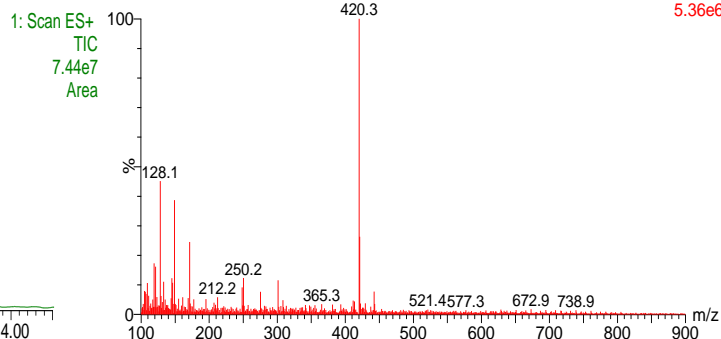
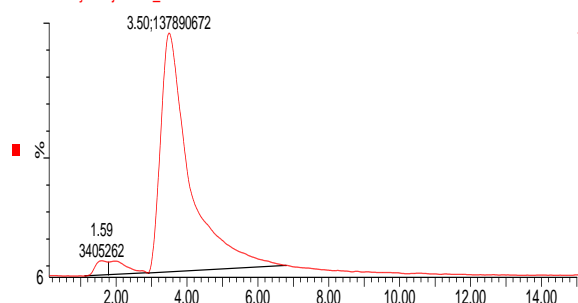


Figure S26. TIC chromatogram of sample **1a** with peak integration. Mass spectrum of compound at RT=3.395 min.

Compound **1b**.

Sample 177_HILIC 2_ 60%H2O19-Jul-201716:05:14

LCMS serviço 19 julho17_9



LCMS serviço 19 julho17_9 201 (3.514) Cm (194:223)

1: Scan ES+ 5.38e7

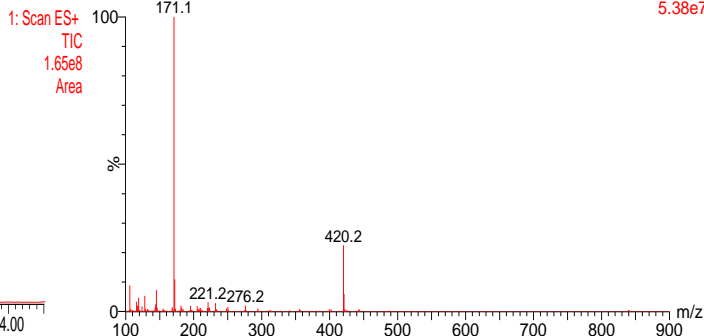
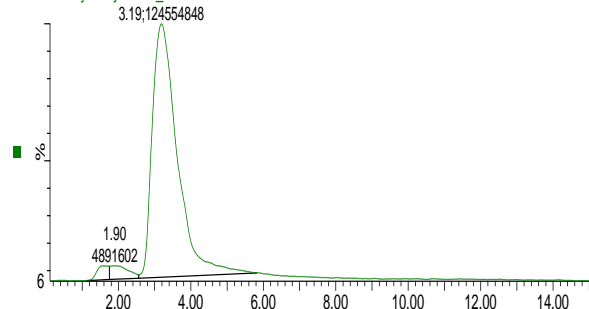


Figure s27. TIC chromatogram of sample **1b** with peak integration. Mass spectrum of compound at RT=3.514 min.

Compound **1c**.

Sample 178_HILIC 2_ 60%H2O19-Jul-201716:22:05

LCMS serviço 19 julho17_10



LCMS serviço 19 julho17_10 183 (3.208) Cm (174:207)

1: Scan ES+ 2.41e7

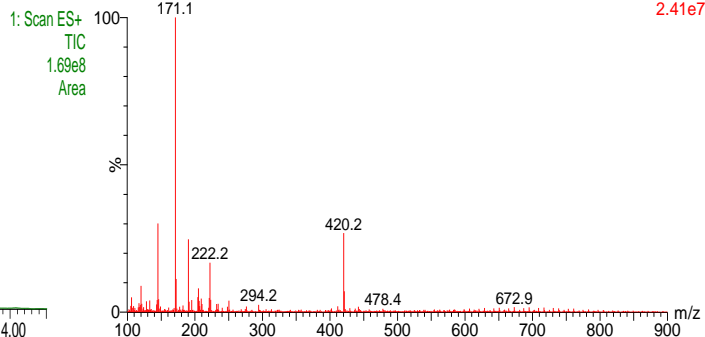
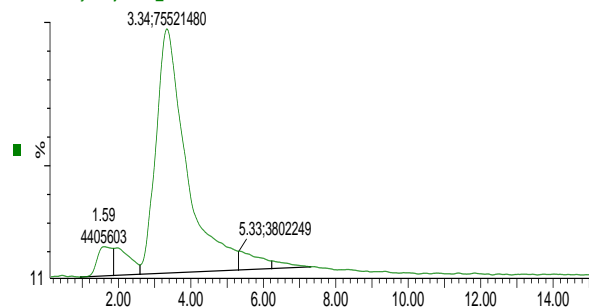


Figure s28. TIC chromatogram of sample **1c** with peak integration. Mass spectrum of compound at RT=3.208 min.

Compound **1d**.

Sample 179_HILIC 2_ 60%H2O19-Jul-201716:38:56

LCMS serviço 19 julho17_11



LCMS serviço 19 julho17_11 190 (3.327) Cm (180:211)

1: Scan ES+
8.68e6

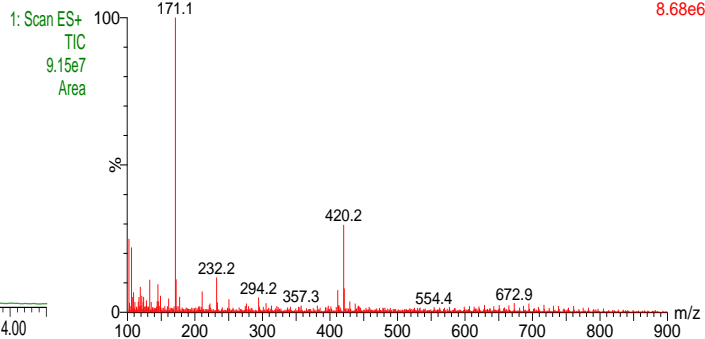
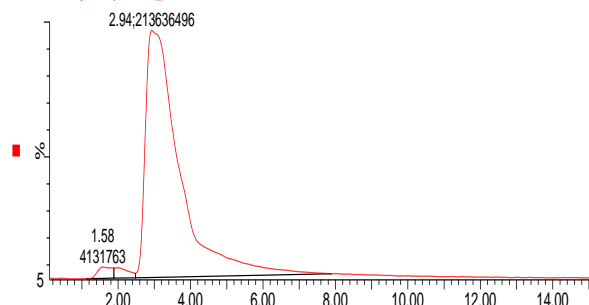


Figure s29. TIC chromatogram of sample **1d** with peak integration. Mass spectrum of compound at RT=3.327 min.

Compound **1e**.

Sample 180_HILIC 2_ 60%H2O19-Jul-201716:55:47

LCMS serviço 19 julho17_12



Sample 180_HILIC 2_ 60%H2O

LCMS serviço 19 julho17_12 167 (2.936) Cm (166:187)

1: Scan ES+
4.43e7

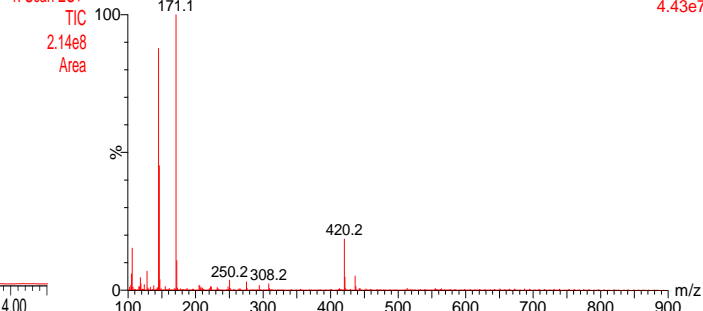
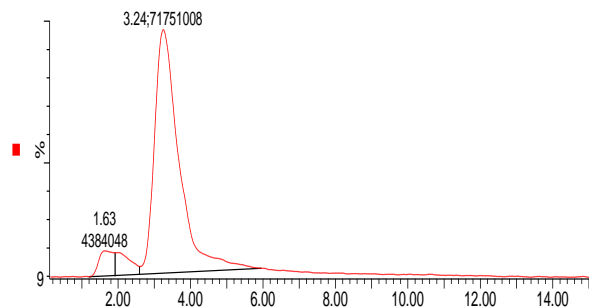


Figure s30. TIC chromatogram of sample **1e** with peak integration. Mass spectrum of compound at RT=2.936 min.

Compound **1f**.

Sample 183_HILIC 2_ 60%H2O19-Jul-201717:46:19

LCMS serviço 19 julho17_15



LCMS serviço 19 julho17_15 185 (3.242) Cm (181:203)

1: Scan ES+
8.04e6

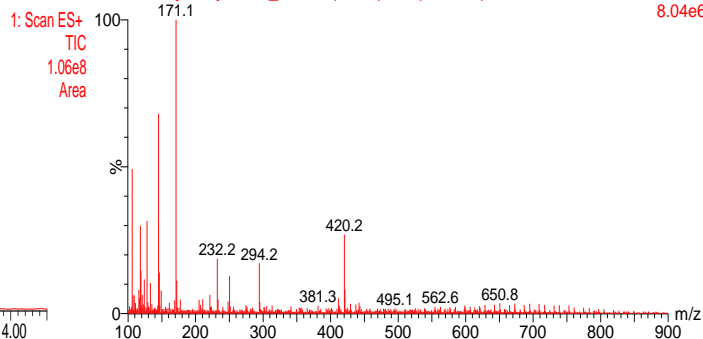
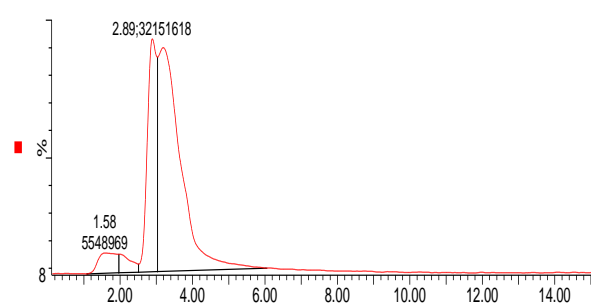


Figure s31. TIC chromatogram of sample **1f** with peak integration. Mass spectrum of compound at RT=3.242 min.

Compound **1g**.

Sample 185_HILIC 2_ 60%H2O19-Jul-201718:20:01

LCMS serviço 19 julho17_17



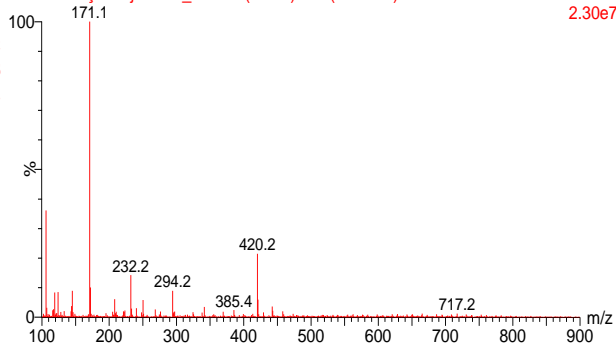
LCMS serviço 19 julho17_17 180 (3.157) Cm (180:184)

1: Scan ES+

TIC

1.26e8

Area



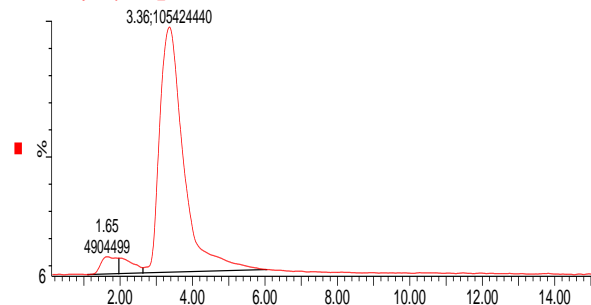
1: Scan ES+
2.30e7

Figure s32. TIC chromatogram of sample **1g** with peak integration. Mass spectrum of compound at RT=3.157 min.

Compound **1h**.

Sample 184_HILIC 2_ 60%H2O19-Jul-201718:03:10

LCMS serviço 19 julho17_16



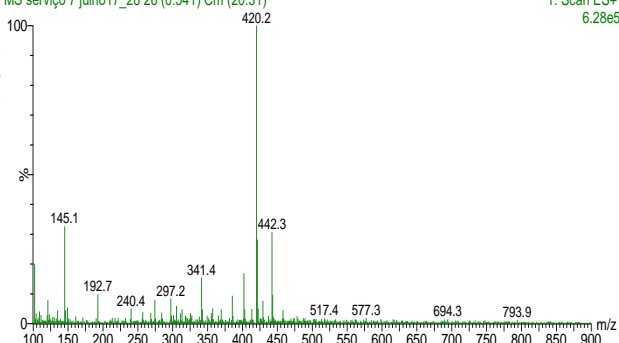
MS serviço 7 julho17_28 26 (0.541) Cm (20:31)

1: Scan ES+

TIC

1.52e8

Area



1: Scan ES+
6.28e5

Figure s33. TIC chromatogram of sample **1h** with peak integration. Mass spectrum of compound at RT=3.365 min.

Compound 2a.

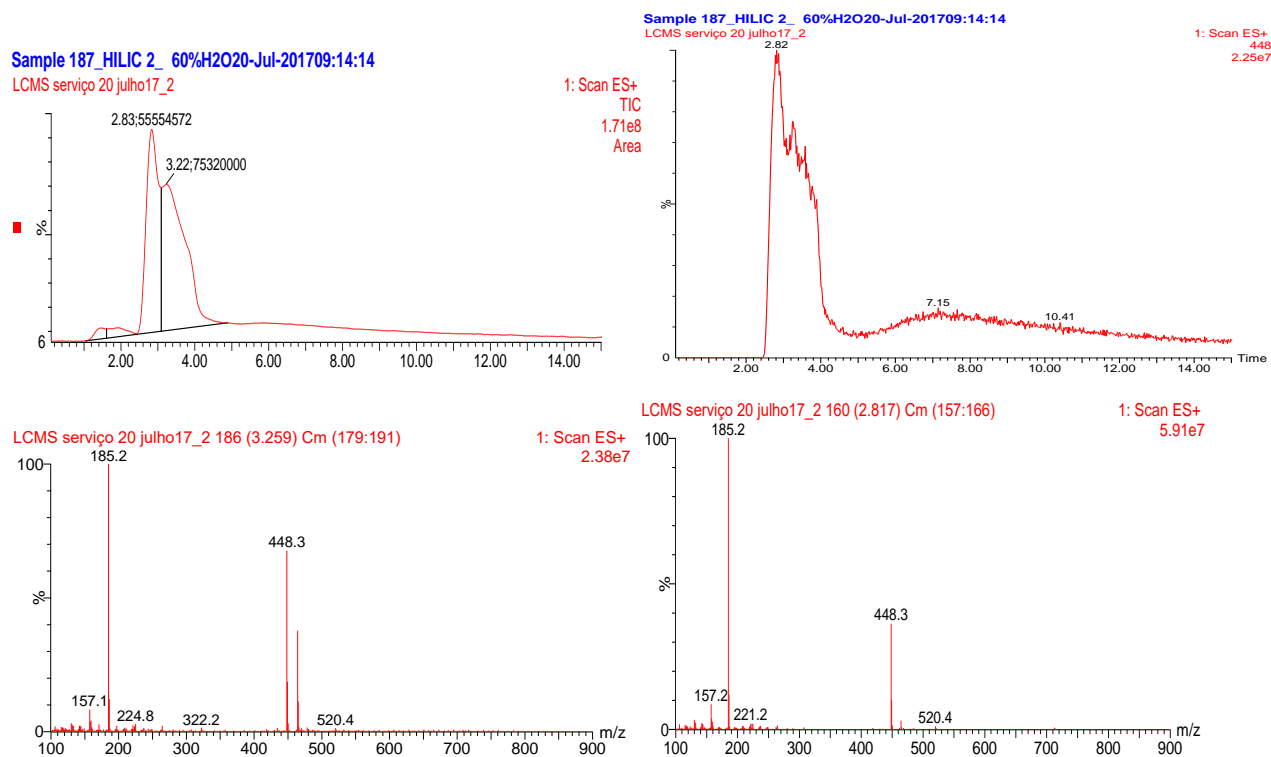


Figure s34. TIC chromatogram of sample **2a** with peak integration (top left). Extracted chromatogram for ion at m/z 448 (top right). Mass spectra of compounds at $RT=3.259$ min (bottom left) and $RT=2.817$ min (bottom right)

Compound 2b.

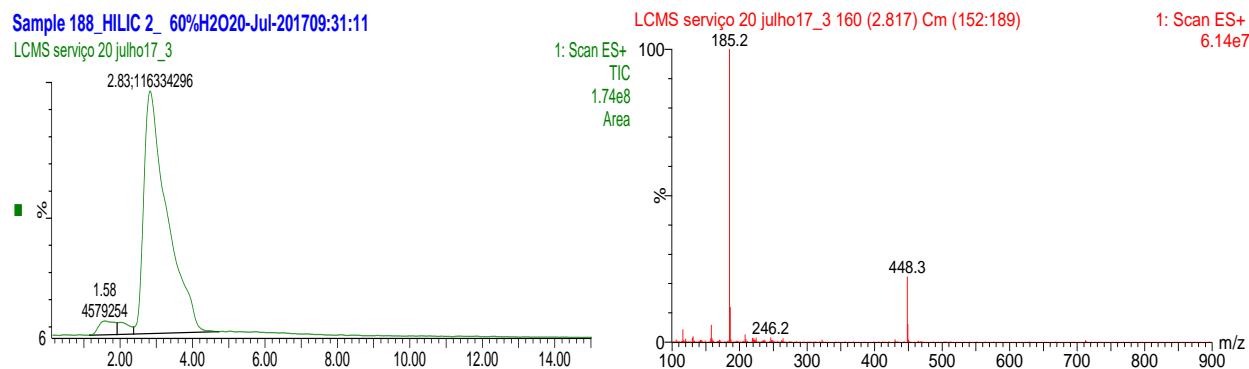
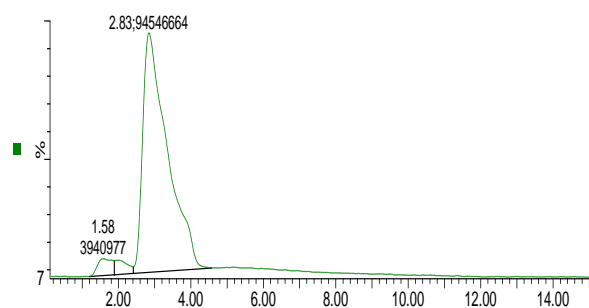


Figure s35. TIC chromatogram of sample **2b** with peak integration. Mass spectrum of compound at $RT=2.817$ min.

Compound 2c.

Sample 189_HILIC 2_ 60%H2O20-Jul-201709:48:02

LCMS serviço 20 julho17_4



LCMS serviço 20 julho17_4 161 (2.834) Cm (156:199)

1: Scan ES+
3.35e7

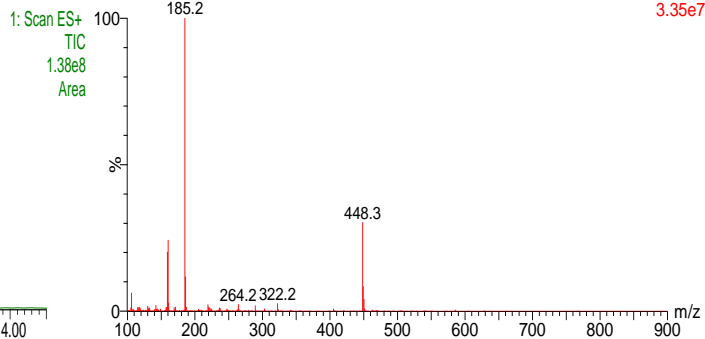
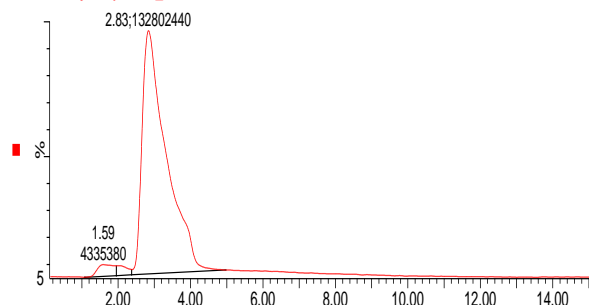


Figure s36. TIC chromatogram of sample 2c with peak integration. Mass spectrum of compound at RT=2.834 min.

Compound 2d.

Sample 190_HILIC 2_ 60%H2O20-Jul-201710:04:53

LCMS serviço 20 julho17_5



LCMS serviço 20 julho17_5 161 (2.834) Cm (157:181)

1: Scan ES+
7.03e7

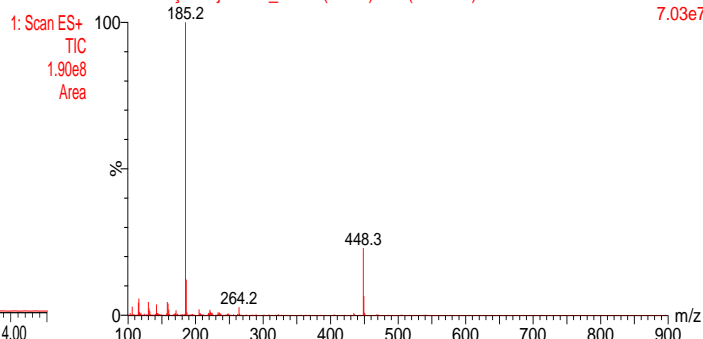


Figure s37. TIC chromatogram of sample 2d with peak integration. Mass spectrum of compound at RT=2.834 min.

8. Supplemental references

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