

Supporting Information

First-Principles Study of the Structure-Performance Relation of Pristine $W_{n+1}C_n$ and Oxygen-Functionalized $W_{n+1}C_nO_2$ MXenes as Cathode Catalysts for Li-O₂ Batteries

Liwei Zhu, Jiajun Wang, Jie Liu, Ruxin Wang, Meixin Lin, Tao Wang, Yuchao Zhen, Jing Xu*, Lianming Zhao*

School of Materials Science and Engineering, China University of Petroleum (East China), Qingdao, Shandong 266580, PR China

* Corresponding authors.

E-mail addresses: lmzhao@upc.edu.cn (L. Zhao), xujing@upc.edu.cn (J. Xu).

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Fig. S1. The top and side views of the adsorption configuration of LiO_2 on (a1) W_2C , (a2) W_3C_2 , (a3) W_4C_3 , (b1) W_2CO_2 , (b2) $\text{W}_3\text{C}_2\text{O}_2$, and (b3) $\text{W}_4\text{C}_3\text{O}_2$ MXenes.

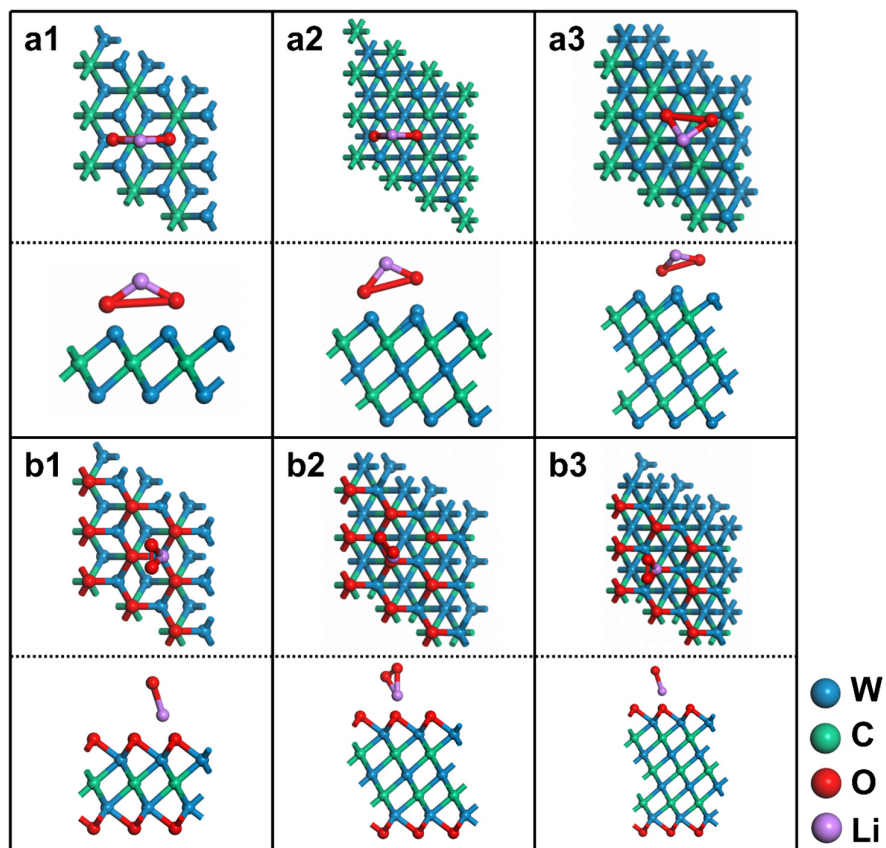


Fig. S2. The top and side views of the adsorption configuration of Li_2O_2 on (a1) W_2C , (a2) W_3C_2 , (a3) W_4C_3 , (b1) W_2CO_2 , (b2) $\text{W}_3\text{C}_2\text{O}_2$, and (b3) $\text{W}_4\text{C}_3\text{O}_2$ MXenes.

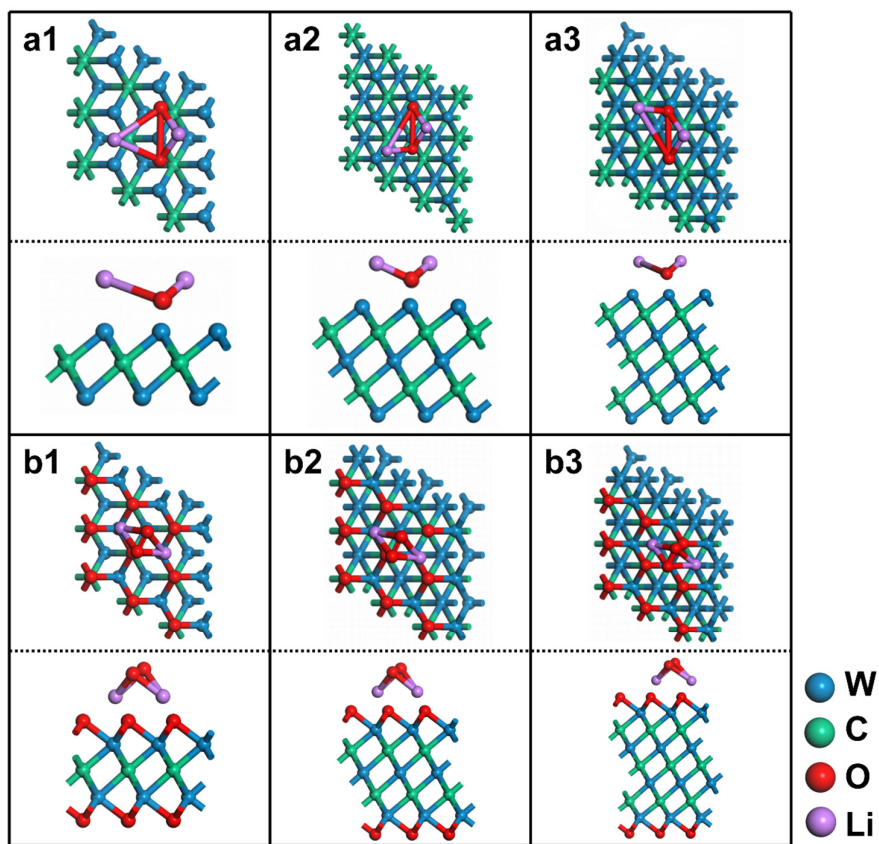


Table S1. Formation energy E_f (in eV) of W_2C , W_3C_2 , and W_4C_3 .

Materials	E_f/eV
W_2C	-3.48
W_3C_2	-3.35
W_4C_3	-3.33

Table S2. The adsorption energy E_{ads} (in eV) of Li_xO_2 ($x = 1, 2$, and 4) on W_{n+1}C_n and $\text{W}_{n+1}\text{C}_n\text{O}_2$.

Materials	$E_{\text{ads}}(\text{O}_2)$	$E_{\text{ads}}(\text{LiO}_2)$	$E_{\text{ads}}(\text{Li}_2\text{O}_2)$	$E_{\text{ads}}((\text{Li}_2\text{O})_2)$
W_2C	−6.36	−15.52	−23.28	−36.05
W_3C_2	−5.98	−13.73	−20.84	−33.33
W_4C_3	−5.77	−12.96	−19.63	−31.44
W_2CO_2	−4.98	−11.84	−18.59	−30.27
$\text{W}_3\text{C}_2\text{O}_2$	−4.74	−11.27	−17.70	−29.06
$\text{W}_4\text{C}_3\text{O}_2$	−4.43	−10.61	−16.51	−27.62

Table S3. The average length (in Å) of the Li–O ($d_{\text{Li-O}}$) and O–O bonds ($d_{\text{O-O}}$) in LiO₂/Li₂O₂ and the adsorbed distance of LiO₂/Li₂O₂ on W_{n+1}C_n ($d_{\text{Li-sub}}/d_{\text{O-sub}}$).

LiO ₂					Li ₂ O ₂			
Distance	Free	W ₂ C	W ₃ C ₂	W ₄ C ₃	Free	W ₂ C	W ₃ C ₂	W ₄ C ₃
$d_{\text{O-O}}$	1.372	3.153	3.233	2.703	1.597	2.849	2.876	2.911
$d_{\text{Li-O}}$	1.774	1.845	2.006	1.946	1.718	2.491	2.344	2.396
$d_{\text{Li-sub}}$	\	2.512	2.582	2.461	\	2.244	2.204	2.220
$d_{\text{O-sub}}$	\	1.628	1.416	1.581	\	1.357	1.338	1.409

Table S4. The average length (in Å) of the Li–O ($d_{\text{Li-O}}$) and O–O bonds ($d_{\text{O-O}}$) in $\text{LiO}_2/\text{Li}_2\text{O}_2$ and the adsorbed distance of $\text{LiO}_2/\text{Li}_2\text{O}_2$ on $\text{W}_{n+1}\text{C}_n\text{O}_2$ ($d_{\text{Li-sub}}/d_{\text{O-sub}}$).

LiO ₂					Li ₂ O ₂			
Distance	Free	W ₂ CO	W ₃ C ₂ O	W ₄ C ₃ O	Free	W ₂ CO	W ₃ C ₂ O	W ₄ C ₃ O
e		2	2	2		2	2	2
$d_{\text{O-O}}$	1.37 2	1.274	1.286	1.304	1.59 7	1.336	1.312	1.303
$d_{\text{Li-O}}$	1.77 4	2.126	2.053	1.977	1.71 8	2.112	2.172	2.203
$d_{\text{Li-sub}}$	\	1.435	1.446	1.499	\	1.322	1.262	1.284
$d_{\text{O-sub}}$	\	3.373	3.246	3.322	\	2.672	2.674	2.722

Table S5. U_{Dc} , U_0 , U_c , η_{ORR} , η_{OER} , and η_{TOT} (in V) for $W_{n+1}C_n$ and $W_{n+1}C_nO_2$.

Materials	U_{Dc}	U_0	U_c	η_{ORR}	η_{OER}	η_{TOT}
W_2C	3.67	4.70	6.44	1.04	1.74	2.77
W_3C_2	3.52	4.12	5.05	0.60	0.93	1.53
W_4C_3	3.18	3.69	4.47	0.51	0.78	1.29
W_2CO_2 ,	3.12	3.60	4.14	0.48	0.54	1.02
$W_3C_2O_2$	2.96	3.36	3.81	0.39	0.45	0.84
$W_4C_3O_2$	2.83	3.08	3.46	0.25	0.38	0.63