



Supplementary Material

Visible-Light-Driven Ag-Doped BiOBr Nanoplates with an Enhanced Photocatalytic Performance for the Degradation of Bisphenol A

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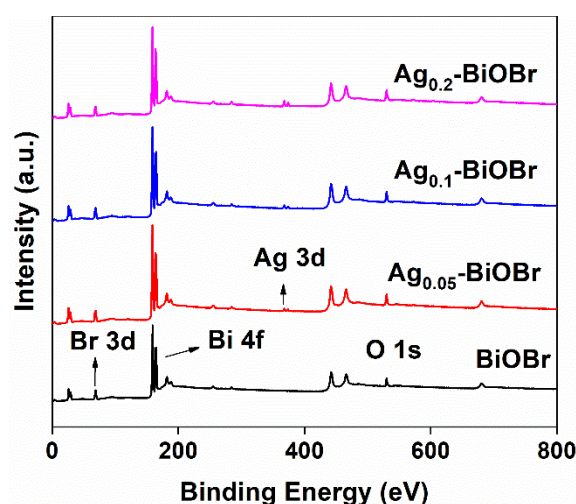


Figure S1. XPS survey spectra of BiOBr and Ag-BiOBr.

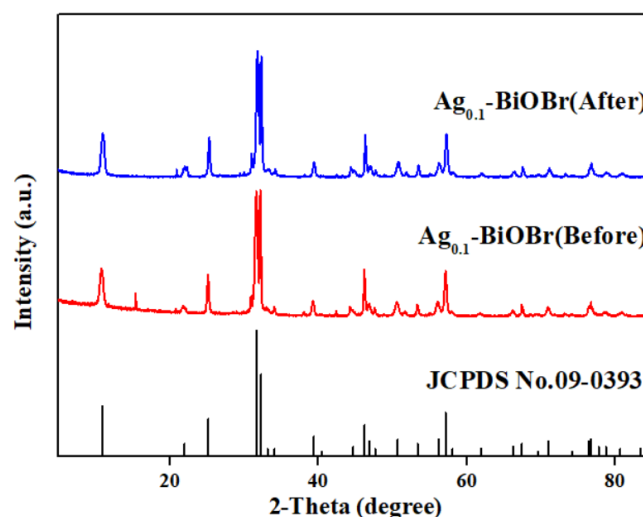


Figure S2. XRD spectra of Ag_{0.1}-BiOBr before and after reaction.

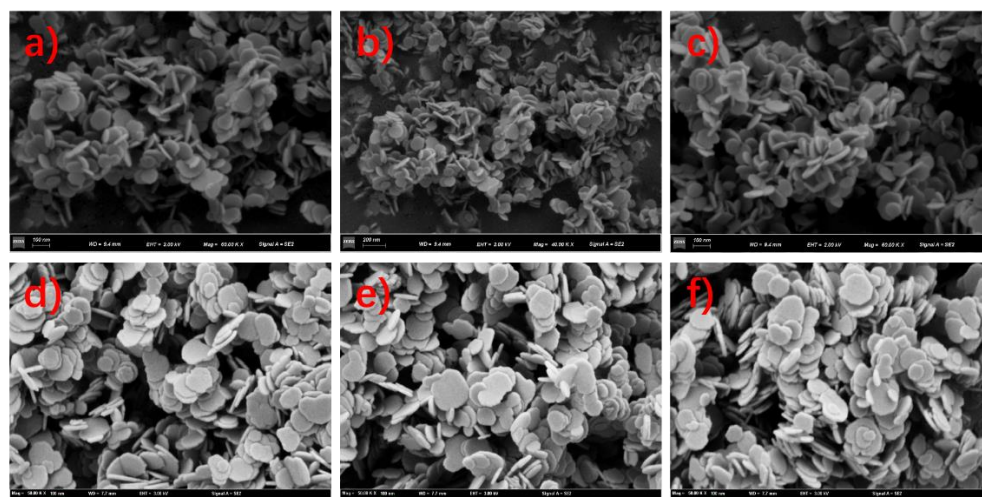


Figure S3. SEM images of Ag_{0.1}-BiOBr (a-c) before and (d-f) after reaction.

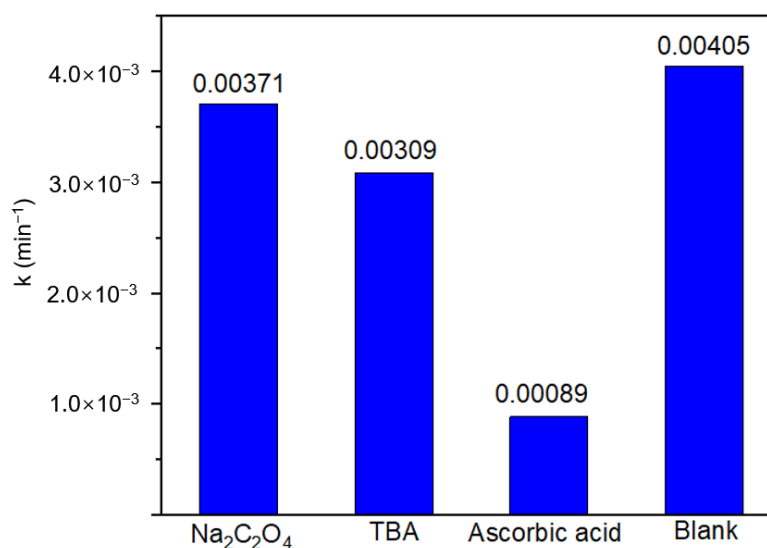


Figure S4. Kinetics constants of photocatalytic degradation of BPA over Ag_{0.1}-BiOBr with and without added free radical scavengers.

Table S1. Comparison with the already published work.

Previous work	In our work	Ref.
<ul style="list-style-type: none"> The combination of AgBr-Ag-BiOBr possessed a p-metal-n structure. The combination of Ag/AgBr/BiOBr was a Z-scheme heterojunction. The target pollutants were methyl orange, which can induce dye sensitization. The target pollutants were RhB and acid orange 7 with color which can induce dye sensitization. 	<ul style="list-style-type: none"> The pollutant used herein was BPA which is a kind of colorless endocrine disrupting compound (EDC) without dye sensitization. The characterizations of products demonstrated that BiOBr possessed a smooth surface, indicating that there was no heterojunction or Schottky junction. Ag elements displayed a high dispersion in BiOBr crystal with the form of Ag(I) instead of Ag(0). 	[1-2]

- The light source is UV light.
- The target pollutant was methyl orange (MO), a colored dye, could induce dye sensitization.
- In this work, we used 420 nm cutoff filter to simulate the visible light illumination condition.
- The pollutant used herein was BPA which is a kind of colorless edcs without dye sensitization

[3]

Table S2. BET surface areas, pore volume and pore size of the samples.

Samples	BET surface area (m ² /g)	Pore volume (cm ³ /g)	Pore size (nm)
BiOBr	12.2749	0.043885	14.30080
Ag _{0.05} -BiOBr	15.4384	0.107037	27.73252
Ag _{0.1} -BiOBr	15.2018	0.102173	26.88460
Ag _{0.2} -BiOBr	20.5491	0.185371	36.08350

Table S3. The comparison of photodegradation efficiency of BPA over different photocatalysts.

Photocatalysts	Source of light	Amount of catalyst (mg)	Initial concentration of BPA (mg/L)	Amount of BPA solution (mL)	Reaction time (min)	Degradation rate	Ref.
Gd ³⁺ doped BiVO ₄	LED lamp (36 W, 730 lm)	1000	10	<50	180	80%	[4]
TiO ₂ @MIL-101(Cr)	125 W mercury lamp Visible light (high pressure 300W Xenon lamp)	40	50	80	240	92%	[5]
α-Fe ₂ O ₃	Visible light (>420 nm, 500W Xenon lamp)	50	—	50	360	93%	[6]
Bi ₁₂ O ₁₇ Cl ₂	Visible light (>420 nm, 500W Xenon lamp)	20	10	40	120	55%	[7]
Ag _{0.1} -BiOBr	Visible light (>420 nm, 500W Xenon lamp)	10	10	50	140	82%	This work

References

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