Adsorption of Chromate Ions by Layered Double Hydroxide–Bentonite Nanocomposite for Groundwater Remediation

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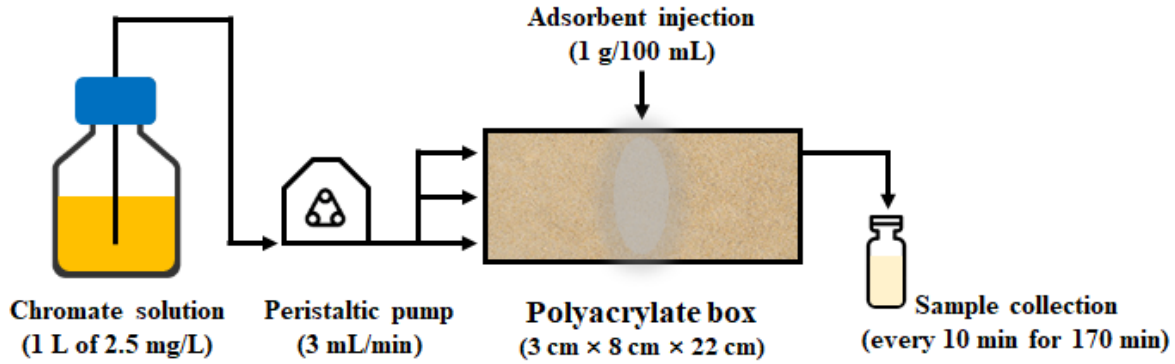

Figure S1. Schematic diagram of chromate adsorption in simulated sub-groundwater condition in box tester.

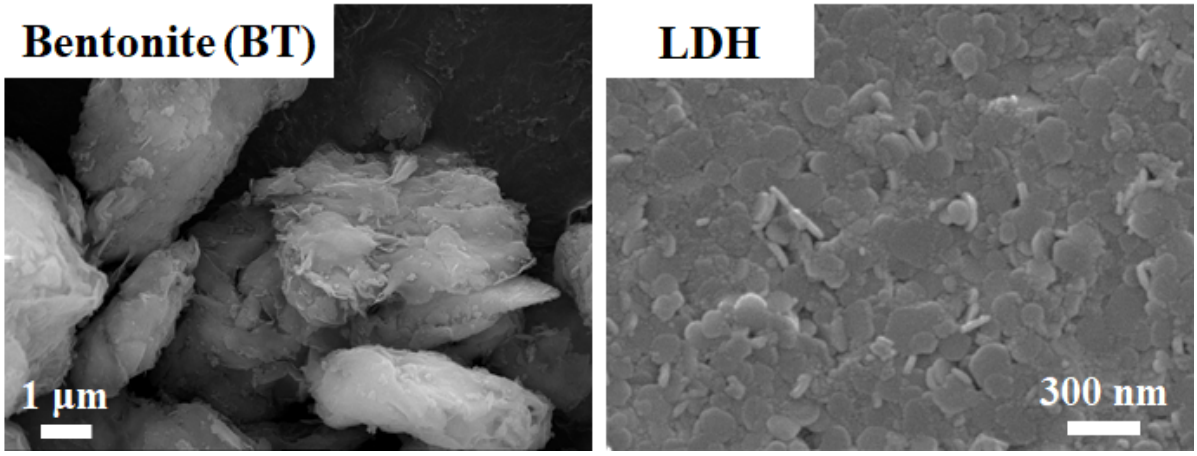

Figure S2. Scanning electron microscopy image of pristine bentonite and LDH.

Table S1. Chromate adsorption capacities of clay-based adsorbents.

Adsorbents	Q_m[a] (mg/g)	Chromate Concentration (mg/L)	Contact Time (min)	Dosing (g/L)	References
Akadama clay	4.29	50	180	5	[1]
OTMAC-Bentonite	12.40	40	60	6	[2]
Shanghai silty clay	1.85	10-250	1440	40	[3]
Natural clay	4.56	10-80	90	5	[4]
Kaolinite	0.571	0.1-16	1440	4	[5]
Illite	0.276	0.1-16	1440	4	[5]
MgAl-Cl (LDH)	58.82	5-200	120	2	[6]
LDH@BT_SP	6.705	2-100	120	1	This study

[a]Obtained from Langmuir isotherm fitting result.

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