



SUPPORTING INFORMATION

Metal-Free Phosphated Mesoporous SiO₂ as Catalyst for the Low-Temperature Conversion of SO₂ to H₂S in Hydrogen

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Table S1. XPS elemental concentrations for uf-P@m-SiO₂-cal (A: fresh), and uf-P@m-SiO₂ (B: fresh, and C: used).

Element	Atomic %		
	uf-P@m-SiO ₂ -cal	uf-P@m-SiO ₂ (Fresh)	uf-P@m-SiO ₂ (Used)
C	20.3	20.6	16.7
N	0.9	0.9	0.7
O	53.4	53.1	56.0
Si	25.2	25.3	26.5
P	0.1	0.1	0.1

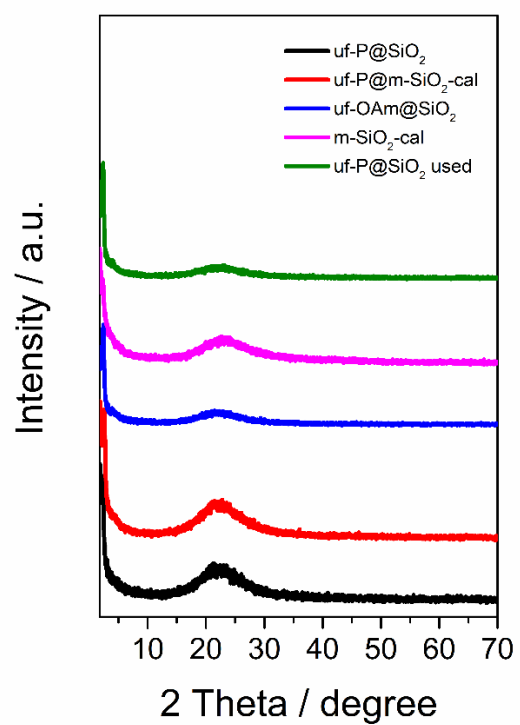


Figure S1. XRD patterns uf-P@m-SiO₂ (fresh and used), uf-P@m-SiO₂-cal, uf-OAm@m-SiO₂, and m-SiO₂-cal.

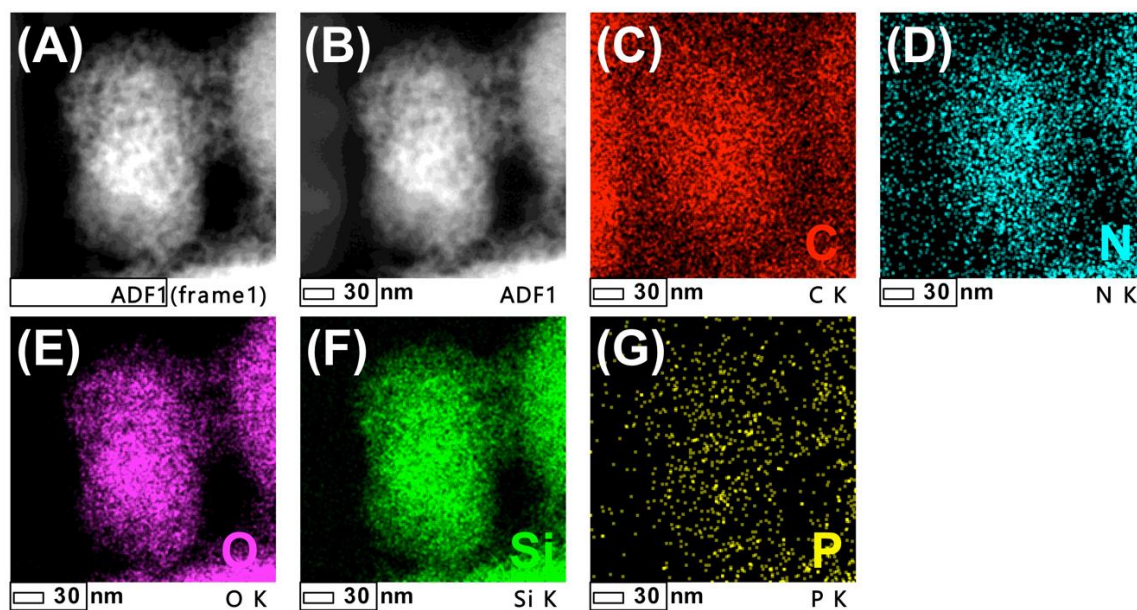


Figure S2. TEM elemental mapping of uf-P@m-SiO₂ (Position 2). (A,B) the TEM image of uf-P@m-SiO₂, (C) the C element mapping, (D) the N element mapping, (E) the O element mapping, (F) the Si element mapping (G) the P element mapping.

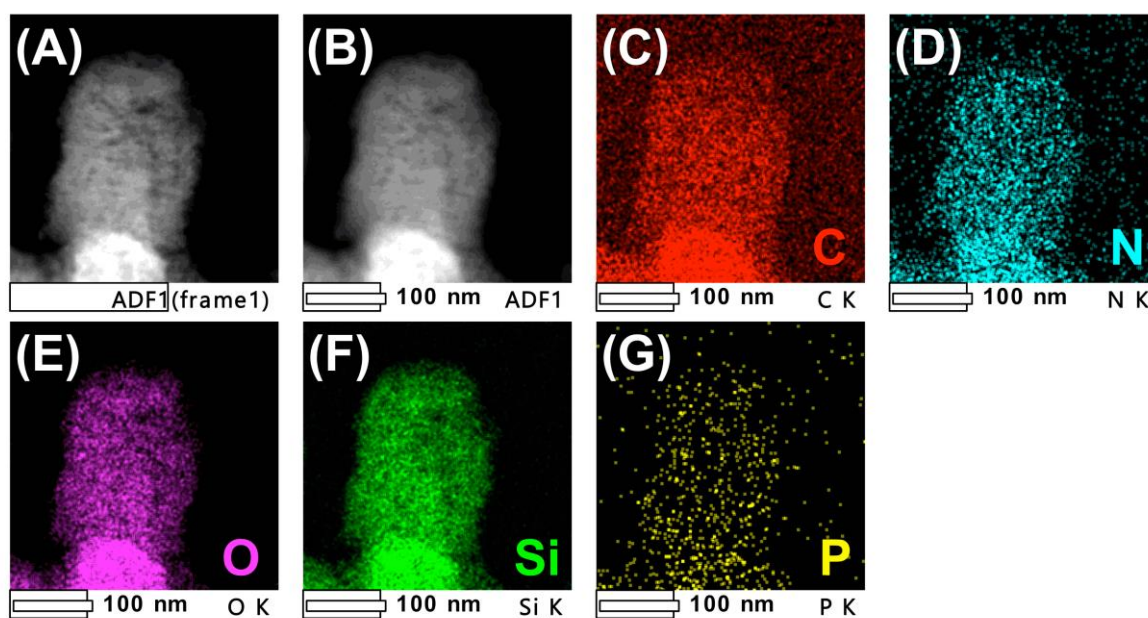
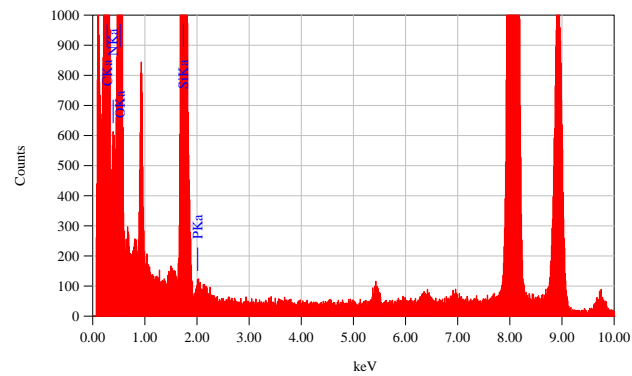
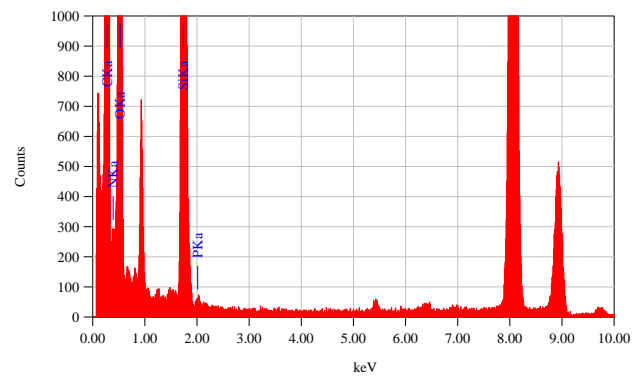


Figure S3. TEM elemental mapping of uf-P@m-SiO₂ (Position 3). (A,B) the TEM image of uf-P@m-SiO₂, (C) the C element mapping, (D) the N element mapping, (E) the O element mapping, (F) the Si element mapping (G) the P element mapping.

Position 1. (corresponding to Figure 5)



Position 2. (corresponding to Figure S2)



Position 3. (corresponding to Figure S3)

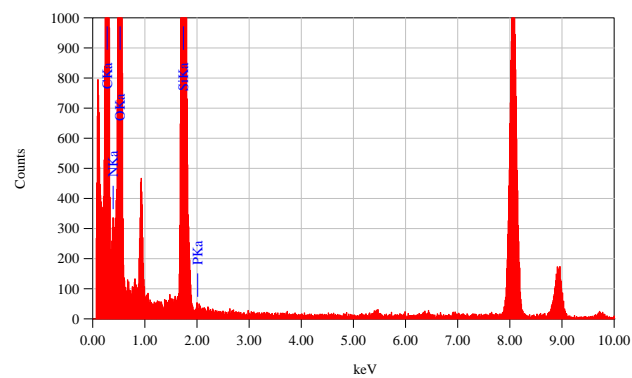


Figure S4. TEM-EDX total area spectrum recorded from the mapped area in Figure 5, Figure S2, and Figure S3 for the sample uf-P@m-SiO₂.

Table S2. TEM-EDX elemental concentrations for uf-P@m-SiO₂.

Element	Atomic %		
	Position 1	Position 2	Position 3
C	74.97	70.53	67.31
N	0.60	0.51	0.68
O	16.83	19.87	22.41
Si	7.57	9.05	9.58
P	0.03	0.03	0.03

Table S3. Weisz-Prater Criterion and Mears Criterion calculation for the catalyst performance test.

Weisz-Prater Criterion	
R (Catalyst particle radius, cm)	2.0E-02
ρ_c (catalyst density, g cm ⁻³)	1.6E-02
C_{As} (Gas concentration at the catalyst surface, mol cm ⁻³)	4.5E-05
D_e (Effective diffusivity cm ² s ⁻¹)	1.0E-05
$r_{A\text{ obs}}$ (mol g ⁻¹ s ⁻¹)	1.0E-07
C_{wp}	1.4E-02
Mears Criterion	
n (reaction order)	1
R (Catalyst particle radius, cm)	2.0E-02
ρ_b (bulk catalyst density, g cm ⁻³)	8.7E-01
k _c (mass transfer coefficient cm s ⁻¹)	5.3
C_{Ab} (bulk gas concentration mol cm ⁻³)	4.5E-05
$r_{A\text{ obs}}$ (mol g ⁻¹ s ⁻¹)	1.0E-07
C_M	7.4E-06

The Weisz-Prater Criterion is far below 1, and Mears Criterion is far below 0.15, so the diffusion of reactants to the catalyst were negligible. [16, 28-29]

Table S4. Comparison of the catalytic performance of different catalysts from literature of the SO₂ conversion to H₂S, selectivity and yield of H₂S.

Catalyst	T (°C)	SO ₂ Conv. (%)	H ₂ S Select. (%)	H ₂ S Yield (%)	Reference
NiO/r-Al ₂ O ₃	320	98	31	30.4	[30]
SnO ₂ -ZrO ₂	550	98	45	44.1	[31]
Fe/γ-Al ₂ O ₃	240	100	100	100	[9]
Fe-Si-Cr	190	100	100	100	[32]
l-Ni ₅ P ₄ @SiO ₂	325	96	99	95	[17]
u-Ni ₅ P ₄ @SiO ₂ -l	240	97	34	33	[17]
uf-P@m-SiO ₂	140	18	6	1	This work
uf-P@m-SiO ₂	160	50	8	4	This work
uf-P@m-SiO ₂	180	76	33	25	This work
uf-P@m-SiO ₂	200	91	39	35	This work
uf-P@m-SiO ₂	220	94	52	49	This work
uf-P@m-SiO ₂	240	96	52	50	This work

References

- (1) Lu, X.; Baker, M. A.; Anjum, D. H.; et al. Ni₂P Nanoparticles Embedded in Mesoporous SiO₂ for Catalytic Hydrogenation of SO₂ to Elemental S. *ACS Appl. Nano Mater.*, **2021**.
- (2) Basina, G.; Polychronopoulou, K.; Zedan, A. F.; Dimos, K.; Katsiotis, M. S.; Fotopoulos, A. P.; Ismail, I.; Tzitzios, V. Ultrasmall Metal-Doped CeO₂ Nanoparticles for Low-Temperature CO Oxidation. *ACS Appl. Nano. Mater.*, **2020**, *3*, 10805–10813.
- (3) Usman, M.; Li, D.; Li, C.; Zhang, S. Highly selective and stable hydrogenation of heavy aromatic-naphthalene over transition metal phosphides. *Sci. China Chem.*, **2015**, *58*, 738–746.