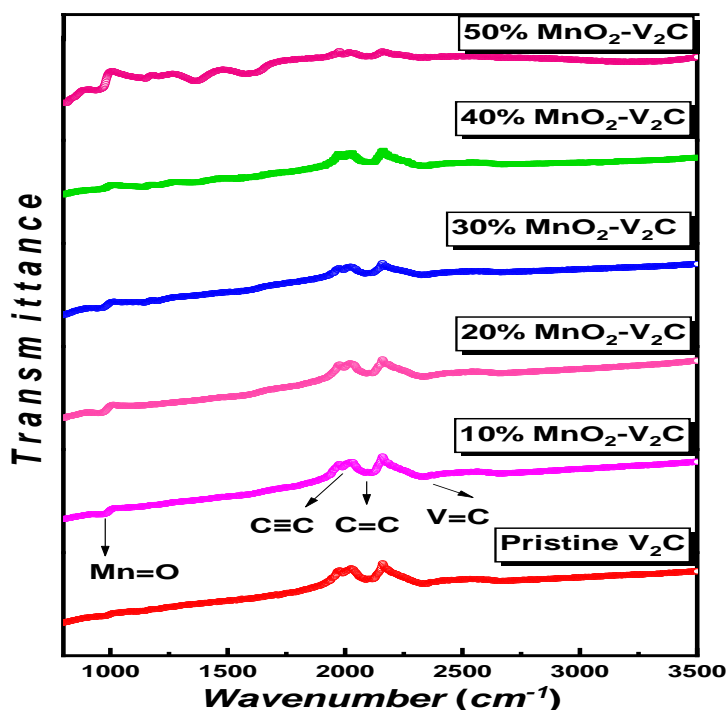


## Supplementary Materials:

### Experimental and Computational Analysis of $\text{MnO}_2@\text{V}_2\text{C}$ -MXene for Enhanced Energy Storage

Figure S1 show FTIR plots for pristine  $\text{V}_2\text{C}$  MXene and (10%, 20%, 30% 40%, 50% weight)  $\text{MnO}_2/\text{V}_2\text{C}$  nanocomposites. The peak at  $2337\text{ cm}^{-1}$  is typically seen for the MXene sheets [1]. An intense peak is also observed at  $2074\text{ cm}^{-1}$  represent to stretching of  $\text{C} \equiv \text{C}$  bond while a small bump is also observed around  $1975\text{ cm}^{-1}$  which contributes to  $\text{C}=\text{C}$  and  $\text{C}=\text{O}$  vibrations [2]. The peaks in the region between  $500\text{ cm}^{-1}$  to  $550\text{ cm}^{-1}$  are the characteristic peak of V-O bands [3]. A peak is majorly observed in  $\text{MnO}_2\text{-V}_2\text{C}$  nanocomposite samples around  $936\text{-}1031\text{ cm}^{-1}$ . This peak does not exist in Pristine MXene and is only existing in  $\text{MnO}_2\text{-V}_2\text{C}$  composite which is an evidence contributing to  $\text{Mn}=\text{O}$  bonds presence and shows the presence of  $\text{MnO}_2$  with  $\text{V}_2\text{C}$  as M Mylarappa et al. shows transmittance peaks of  $\text{MnO}_2$  [4]. The layered structure of  $\text{V}_2\text{C}$  samples exists after  $\text{MnO}_2$  adsorbed at variated percentage intercalation in  $\text{V}_2\text{C}$  till 40% however, the destruction caused in  $\text{V}_2\text{C}$  sheets is comparably higher when we increase  $\text{MnO}_2$  percentage beyond 40% in the structure with the destruction of MXene sheets. At equal ratios, i.e. 50% of  $\text{MnO}_2\text{-V}_2\text{C}$  ratio, the  $\text{V}=\text{C}$  bonds are not observable and different peaks at  $1601\text{ cm}^{-1}$  and  $1343\text{ cm}^{-1}$  arise which contribute to the presence of  $\text{Mn}=\text{O}$  bonds [5]. Several small transmission peaks around  $1000\text{--}1500\text{ cm}^{-1}$  are assigned to bending vibrations of bonds of O-H which are connected with Mn-atoms [6].



**Figure S1.** FTIR plots for Pristine MXene  $\text{V}_2\text{C}$  and  $\text{MnO}_2\text{-V}_2\text{C}$  nanocomposite at varying percentages.

Raman Spectroscopy was initiated at room temperature for analysis of the acoustic and optical modes. Argon ion laser radiation wavelength  $514\text{ nm}$  was focused to a spot size of  $\sim 1\text{ }\mu\text{m}$  with an incident power kept below  $1\text{ mW}$ . The knowledge about phonon spectrum of  $\text{V}_2\text{AlC}$ , Pristine  $\text{V}_2\text{C}$  and  $\text{MnO}_2@\text{V}_2\text{C}$  composite can be obtained which has several broad peaks ranging from  $200$  to  $2100\text{ cm}^{-1}$  and is assigned as non-stoichiometric. The crystallographic structure of  $\text{V}_2\text{C}$  has 3 atoms of the primitive unit cell in  $\text{V}_2\text{C}$  monosheet which give rise to 6 optical modes and 3 acoustic modes at the  $\Gamma$  point of the Brillouin

zone [7]. The information regarding the phonon spectrum is essential for understanding macroscopic properties such as electrical and thermal properties, so the phonon spectrum of Pristine  $V_2C$  was studied from literature, also Gao et al. reports the phonon dispersions for  $V_2C$  [8]. Without terminations,  $V_2C$  consist of three acoustic modes from which two acoustic modes exhibit linear dispersion near  $\Gamma$  and thus correspond to an in-plane vibrations. The G band is a result of in-plane vibrations of  $sp^2$  bonded Carbon atoms. On the contrary, third acoustic mode corresponds to an out-of-plane vibration of V-atoms and has a quadratic dispersion close to  $\Gamma$ . This out of plane vibration of V-atoms results in D band and is attributed to the presence of structural defects. The above-mentioned quadratic dependence has an analogy to the one observed in graphene, which is also considered as parental family to MXenes system [9].  $V_2C$  with termination groups, such as -F and -OH results in different optical and acoustic modes as in the case of  $V_2CF_2$ , the 5 atoms of primitive unit cell led to 12 optical modes and 3 acoustic modes at the  $\Gamma$  point of the Brillouin zone. In the case of  $V_2C(OH)_2$ , the 7 atoms of each primitive cell led to 18 optical modes and 3 acoustic modes at the  $\Gamma$  point of the Brillouin zone.

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