

Supporting Information

Microwave Synthesis of Au Nanoparticles in the Presence of Tetrahydrothiophenocucurbituril

Asma S. Atthar , Shreya Saha, Ahmed Abdulrahman and Anthony I. Day *

Chemistry, School of Science, University of New South Wales Canberra, Australian Defence Force Academy, Canberra, ACT 2600, Australia;
asmasamaunnisa@gmail.com (A.S.A.); shreya.saha@adfa.edu.au (S.S.);
a.abdulrahman@adfa.edu.au (A.A.)

* Correspondence: a.day@adfa.edu.au.com

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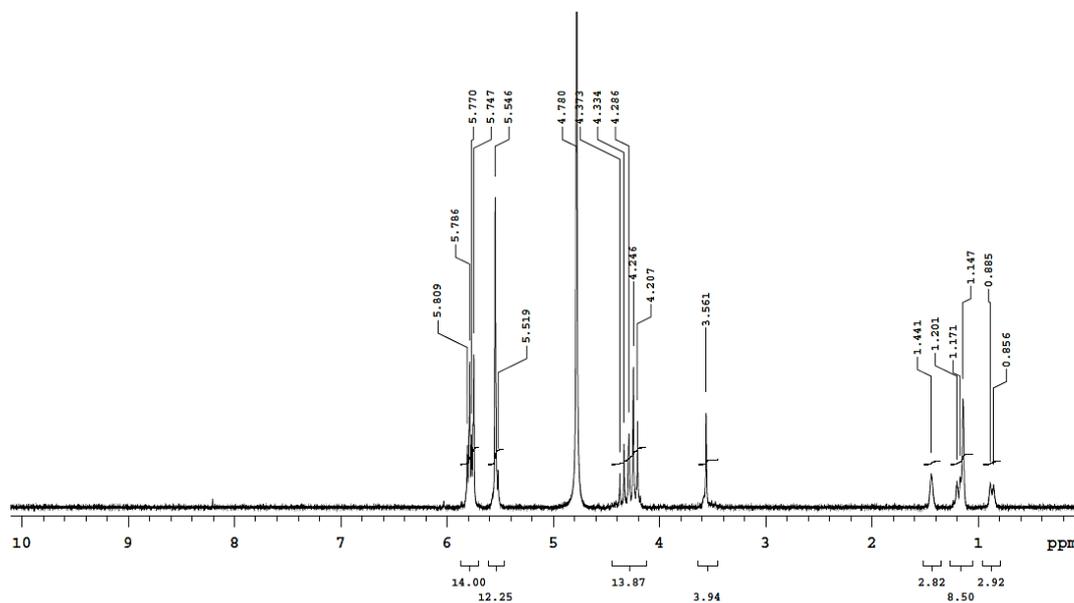
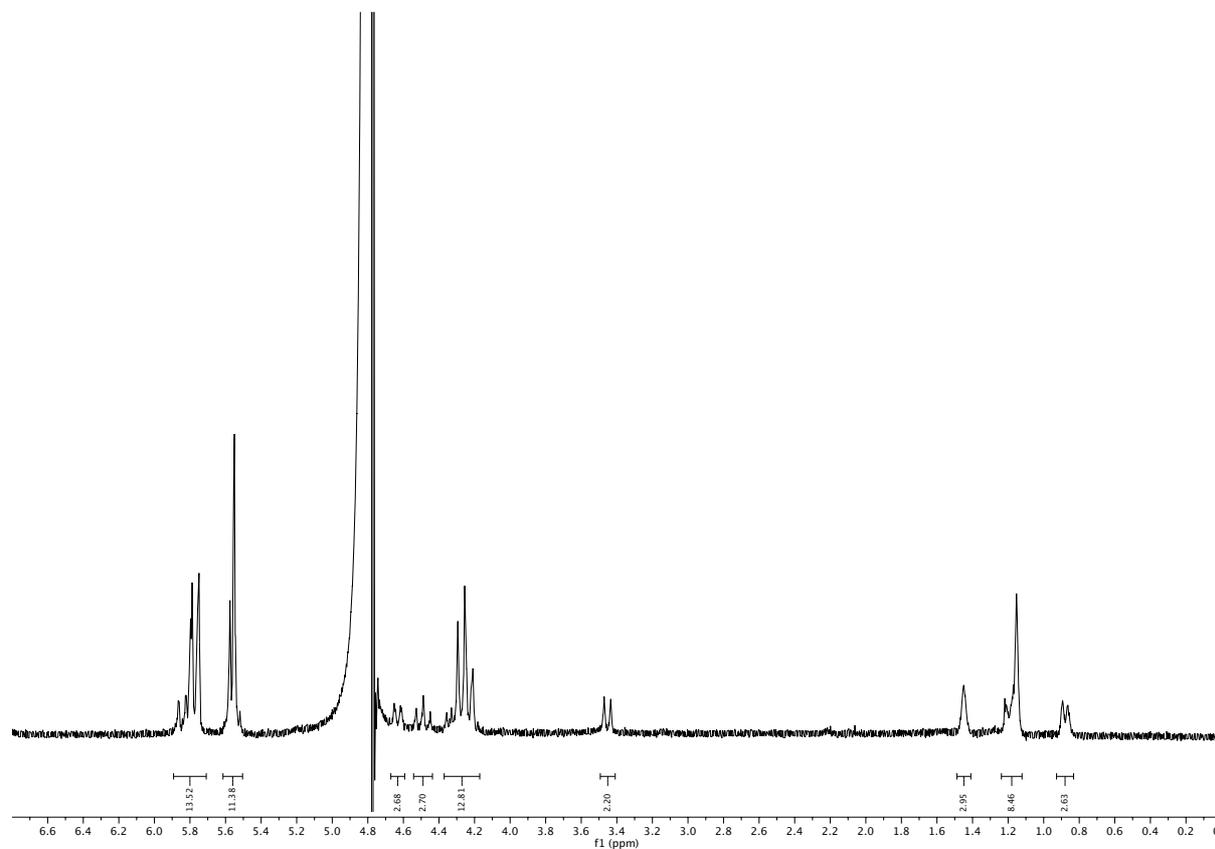


Figure S1. The ^1H NMR spectrum of $\text{ama@THT}_1\text{Q}[7]\text{Cl}$ in D_2O



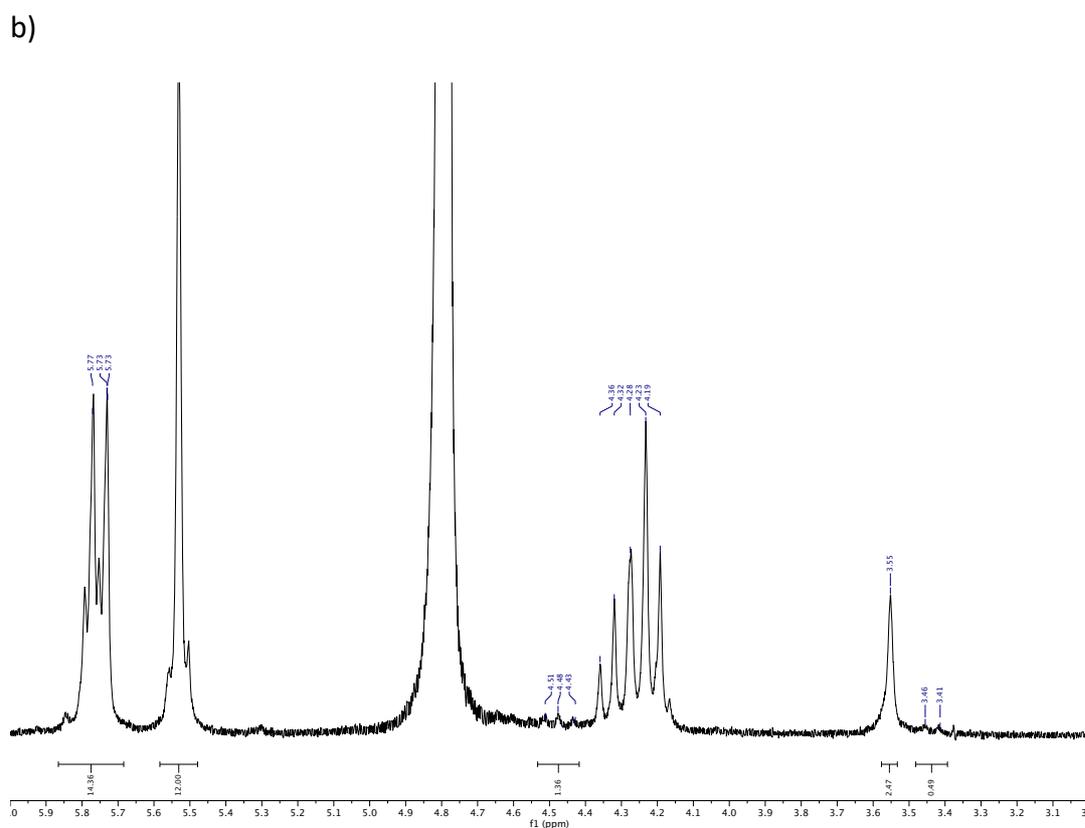
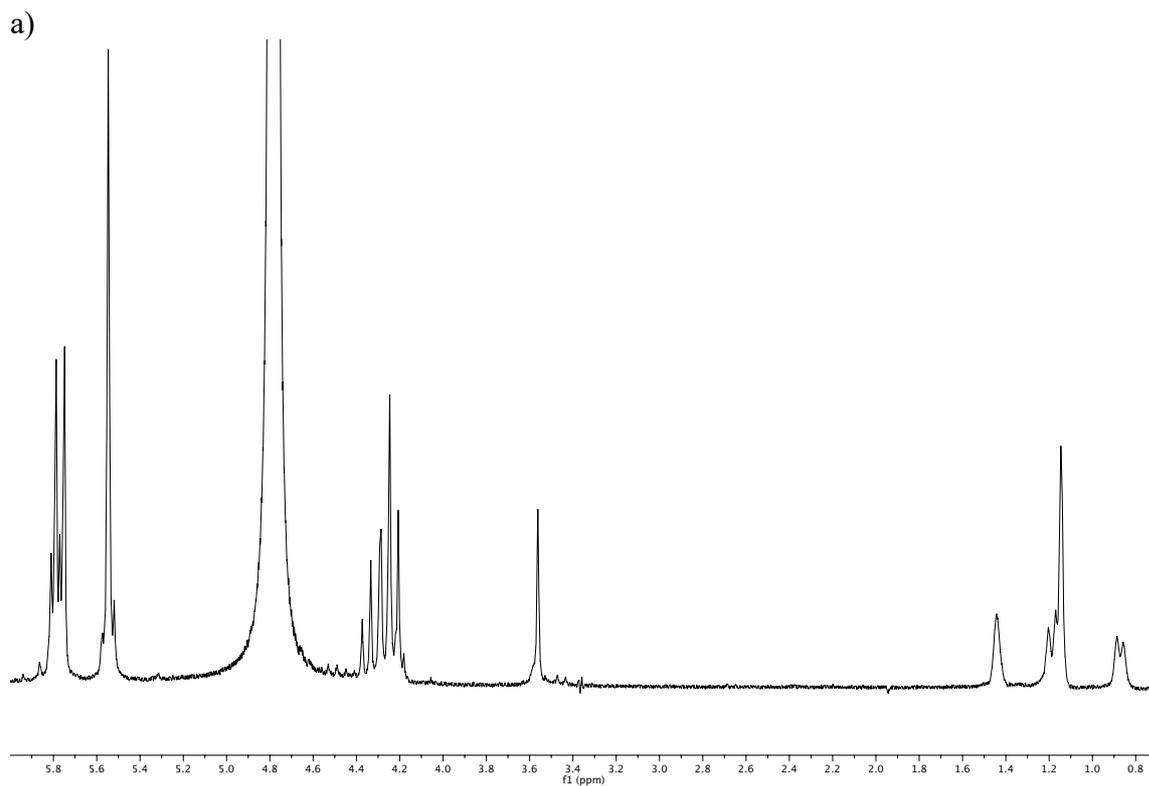


Figure S3. a) The ^1H NMR spectrum of the reaction mixture from the MW reaction of ama@THT₁Q[7] with HAuCl_4 in water heated to 70 °C for 5 min. A freeze-dried sample dissolved in D_2O . b) A zoomed in section of the same sample.

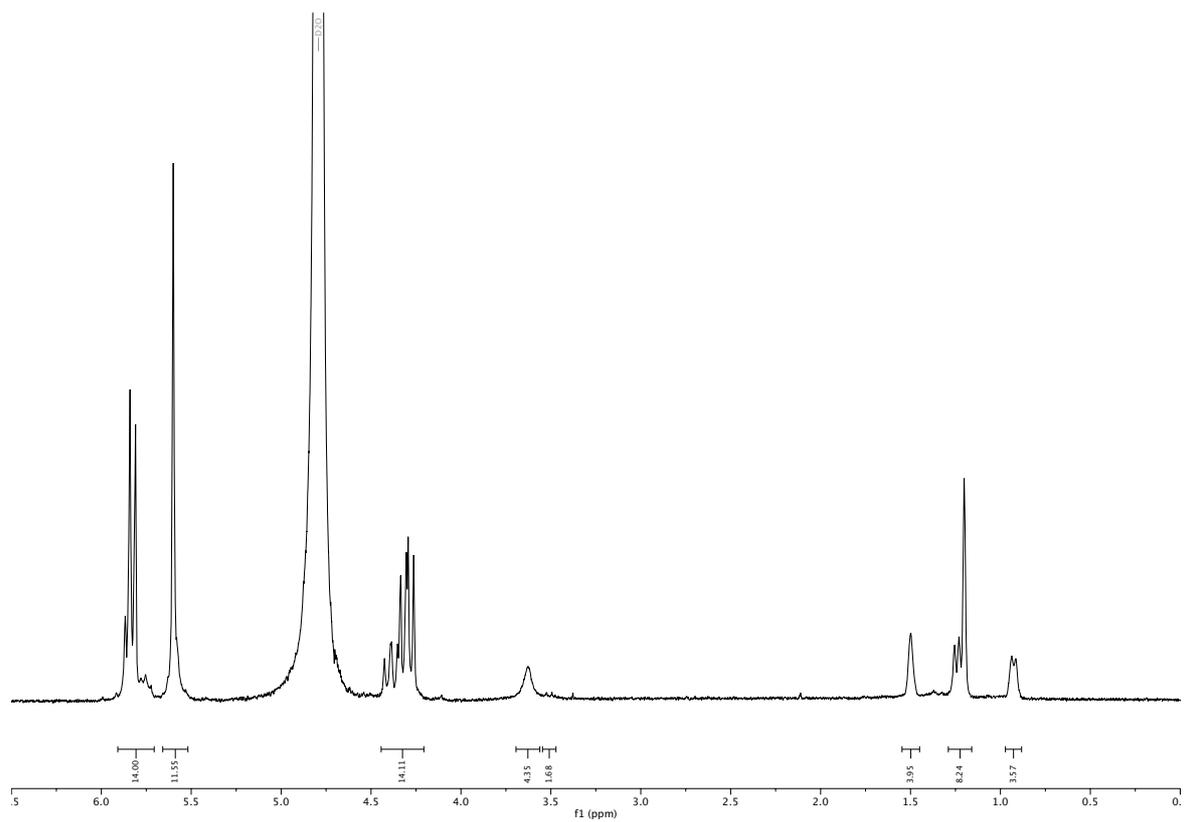


Figure S4. The ^1H NMR spectrum (500 MHz) of the reaction mixture from the MW reaction of $\text{ama@THT}_1\text{Q}[7]\text{PF}_6$ with HAuCl_4 in water heated to $70\text{ }^\circ\text{C}$ for 15 min. At this point the solution is colorless and clear.

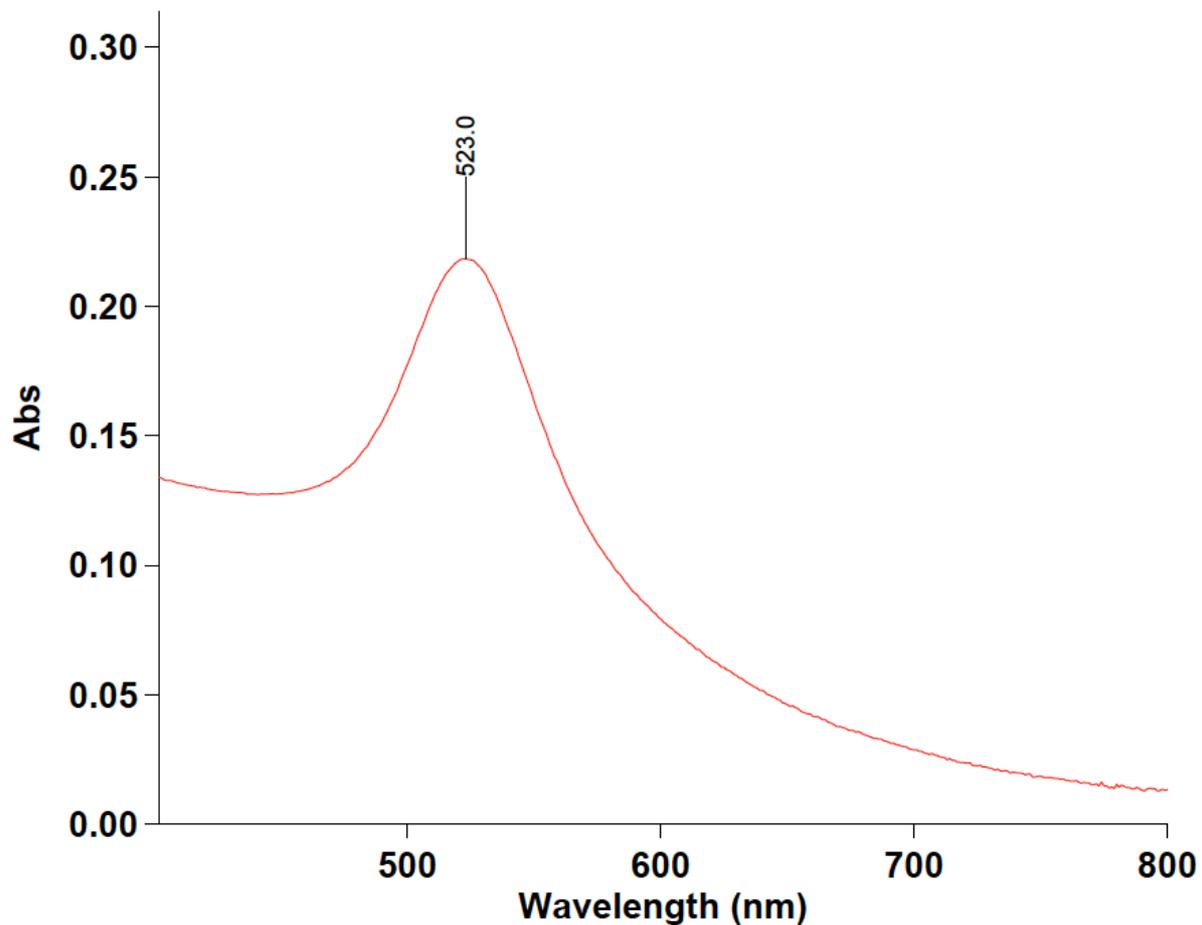


Figure S5. Visible spectra of AuNP-THTglycoluril prepared in a mixture of DMSO/water (2:1) in the MW reactor over 10 min at 70 °C.

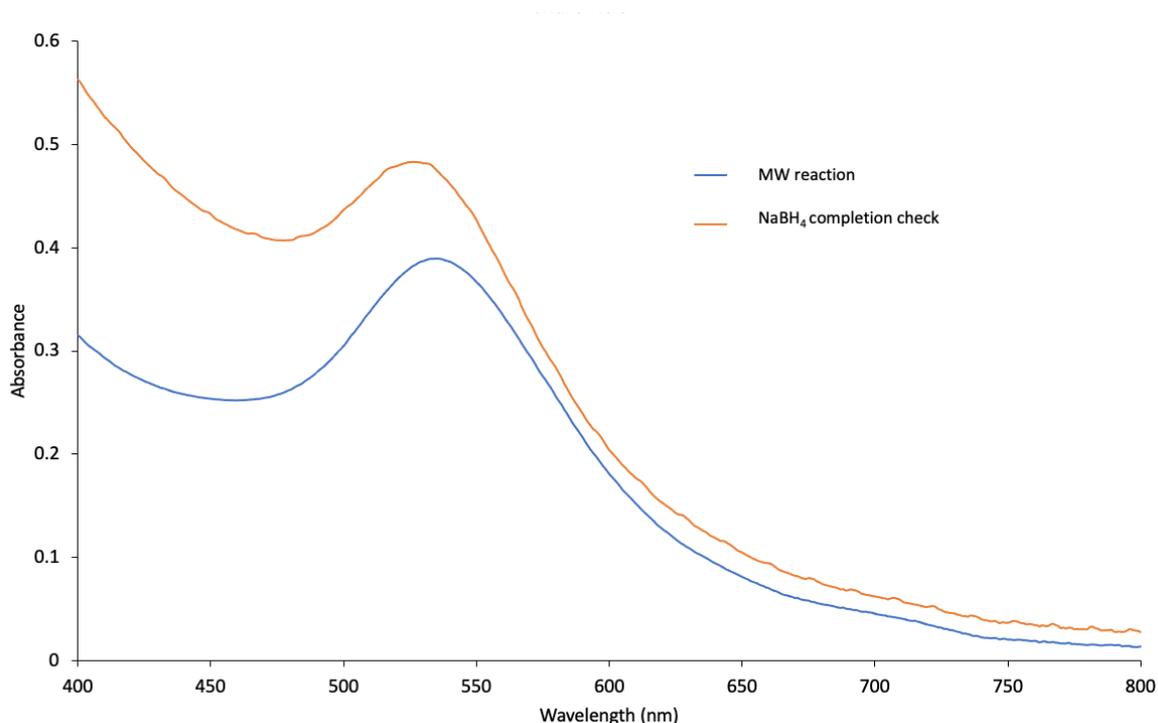


Figure S6. Visible spectra of AuNP-ama@THT₁Q[7] prepared in pure water in the MW reactor over 10 min at 70 °C, with a 1:1 mole ratio of Au(III) to THT₁Q[7]. After the reaction period was complete, a NaBH₄ solution was added revealing the proportion of unreacted Au salts.

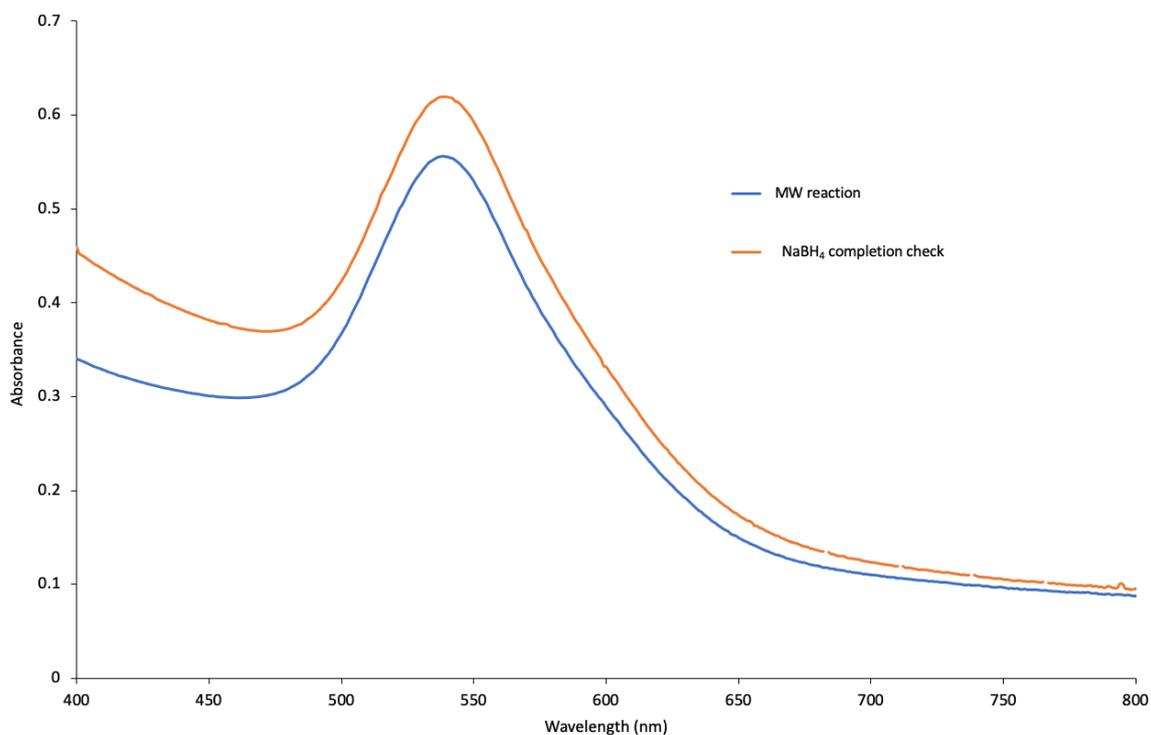


Figure S7. Visible spectra of AuNP-ama@THT₁Q[7] prepared in pure water in the MW reactor over 15 min at 70 °C, with a 1:2 mole ratio of Au(III) to THT₁Q[7]. After the reaction period was complete, a NaBH₄ solution was added revealing the proportion of unreacted Au salts.

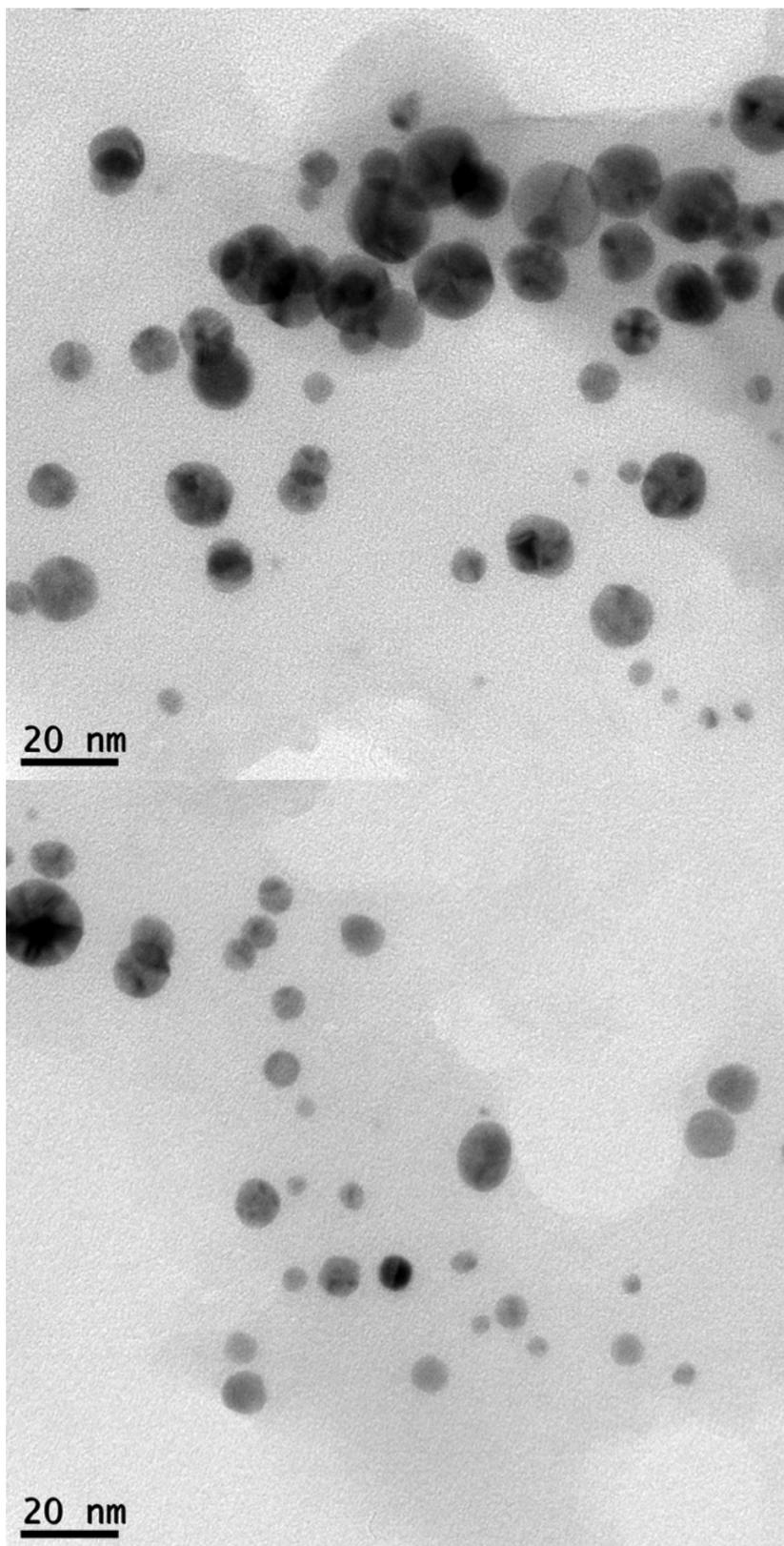


Figure S8. Purified ama@THT₁Q[7] TEM micrographs

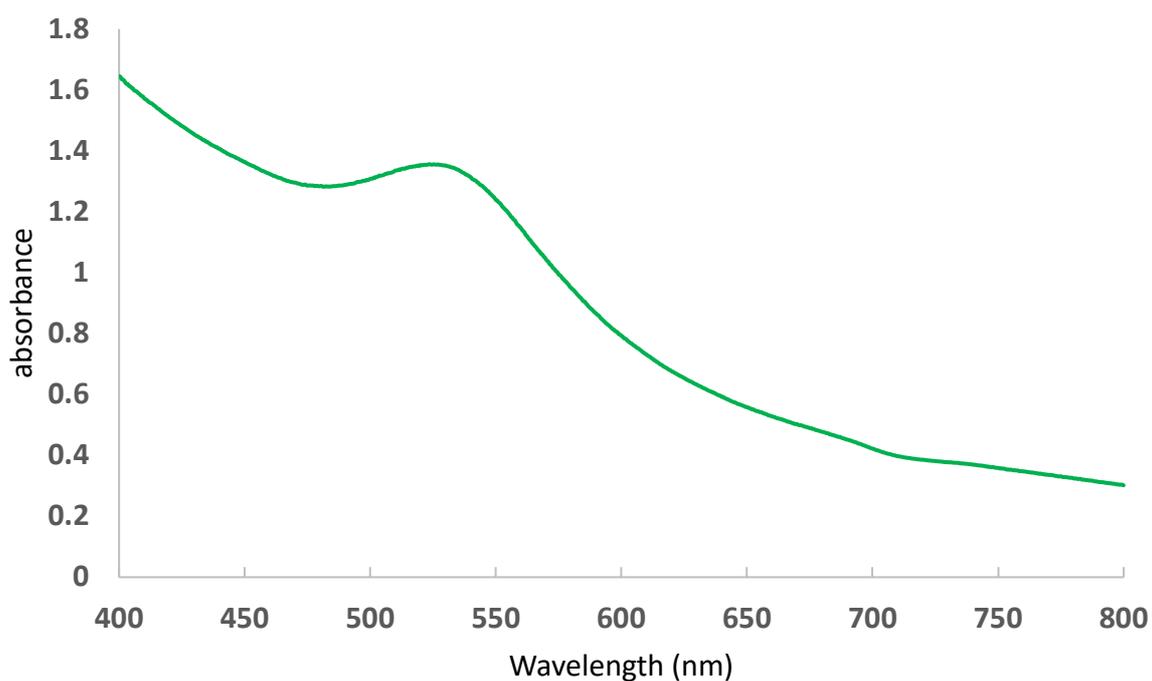


Figure S9. Visible spectrum of AuNP-THT₆Q[6] prepared in an aqueous solution of 50 mM Ca(OAc)₂ in the MW reactor over 15 min at 70 °C. Mole ratio of Au(III) to THT₆Q[6] was 1:1.



AuNP-THT₆Q[6] conjugate in aqueous Ca(OAc)₂.



AuNP-ama@THT₁Q[7] conjugate in H₂O.

Figure S10. Photographs of AuNP-THT_mQ[*n*] conjugates