

Supplementary information

Effects of substituents on the photophysical/photobiological properties of mono-substituted corroles

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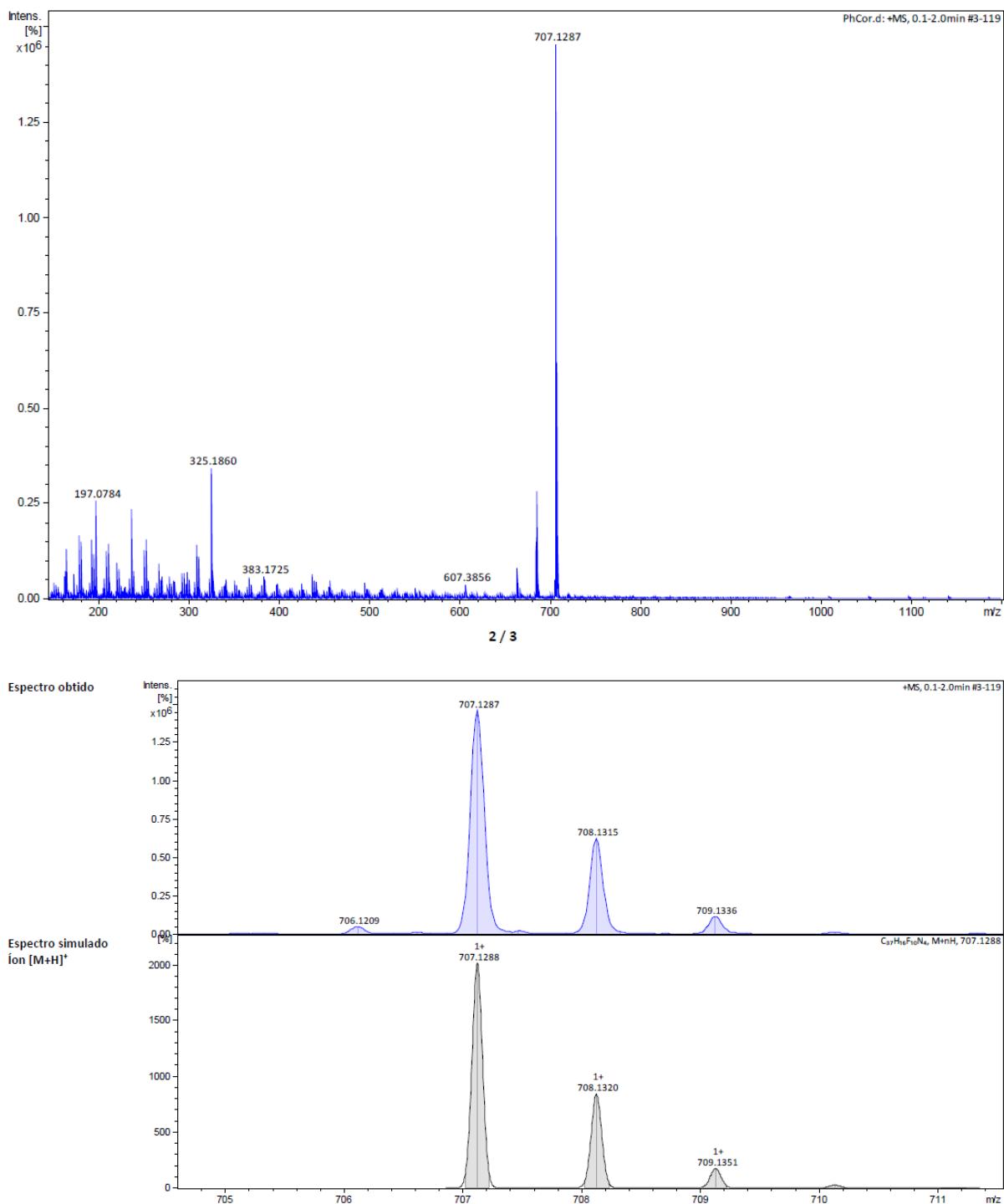


Figure S1. HRMS-ESI(+) mass spectrum of corrole 1.

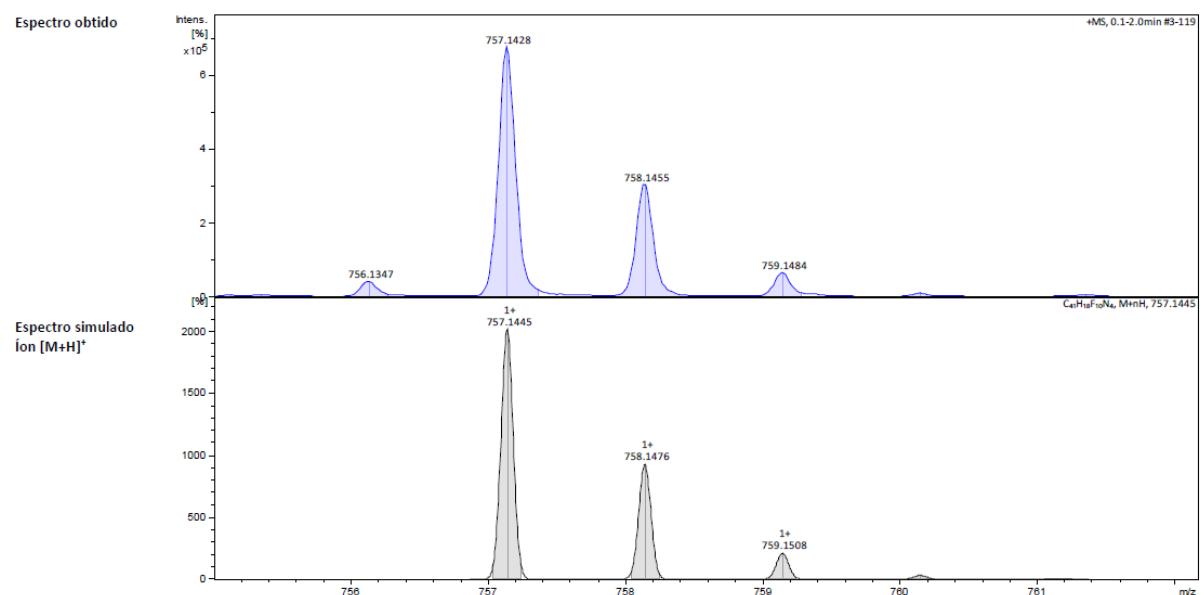
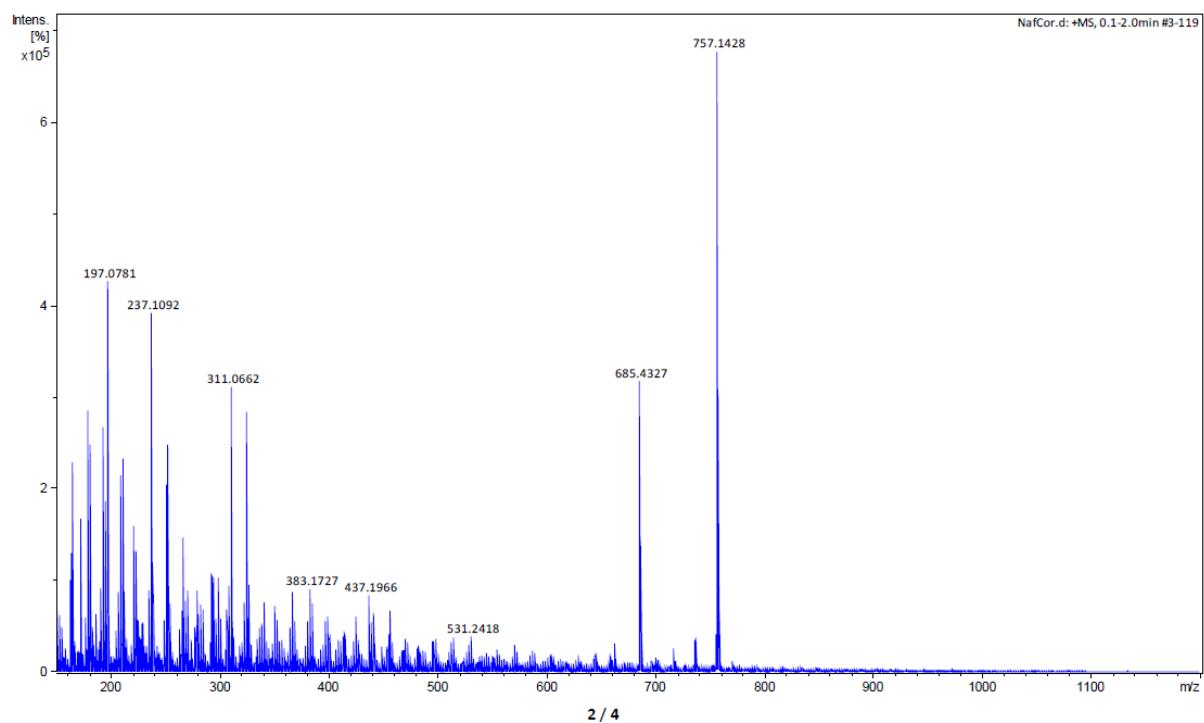


Figure S2. HRMS-ESI(+) mass spectrum of corrole **2**.

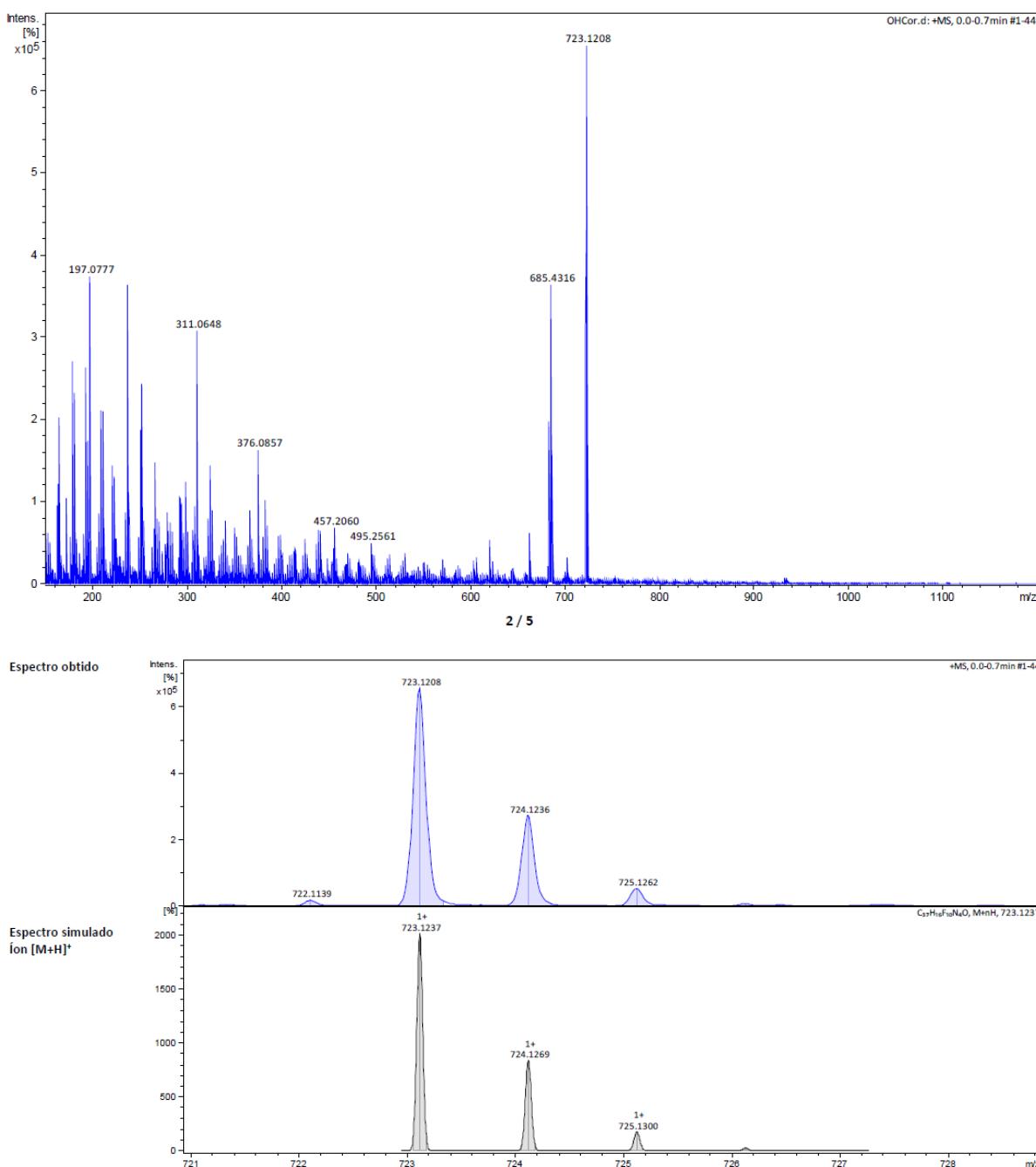


Figure S3. HRMS-ESI(+) mass spectrum of corrole 3.

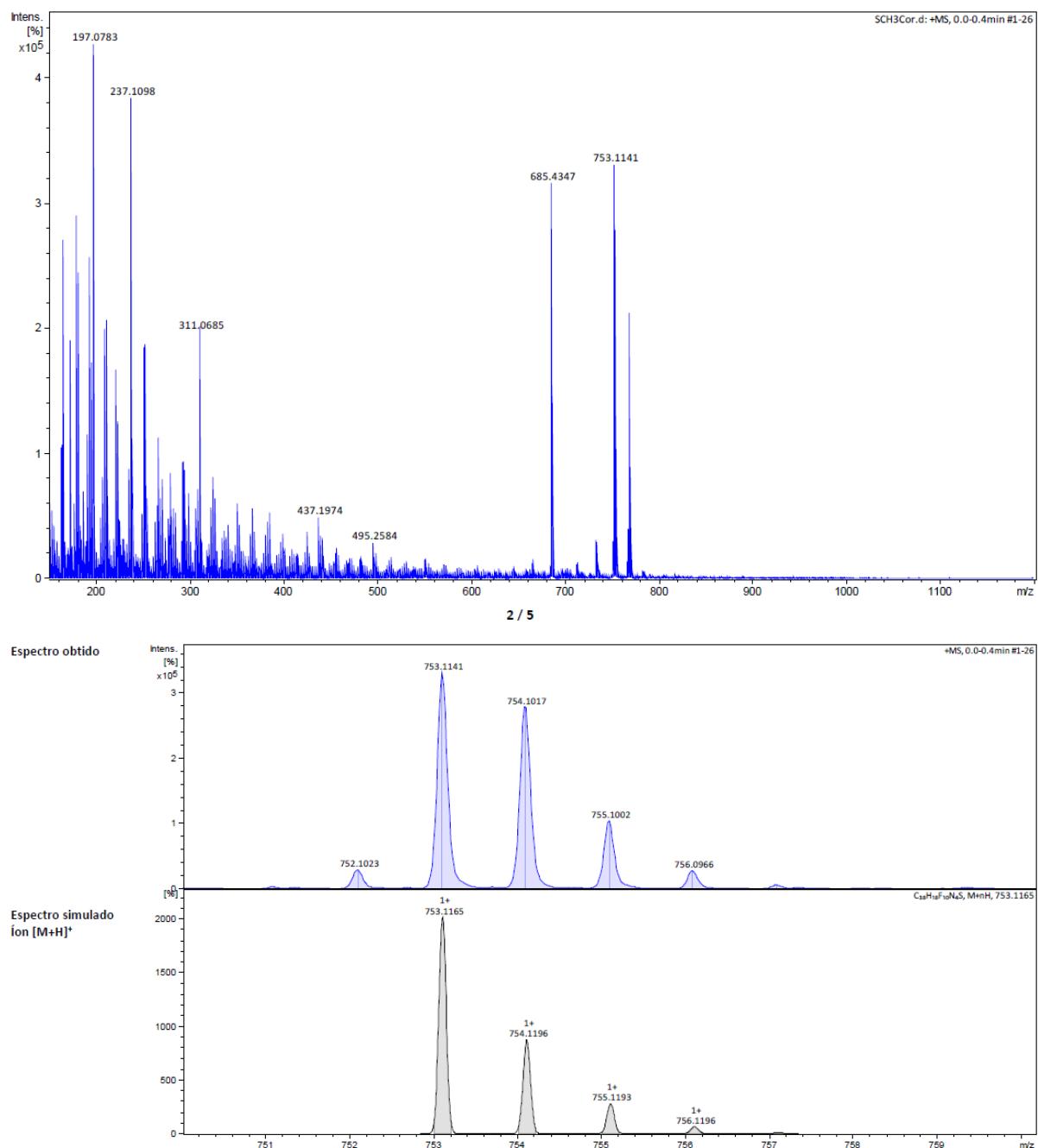


Figure S4. HRMS-ESI(+) mass spectrum of corrole 4.

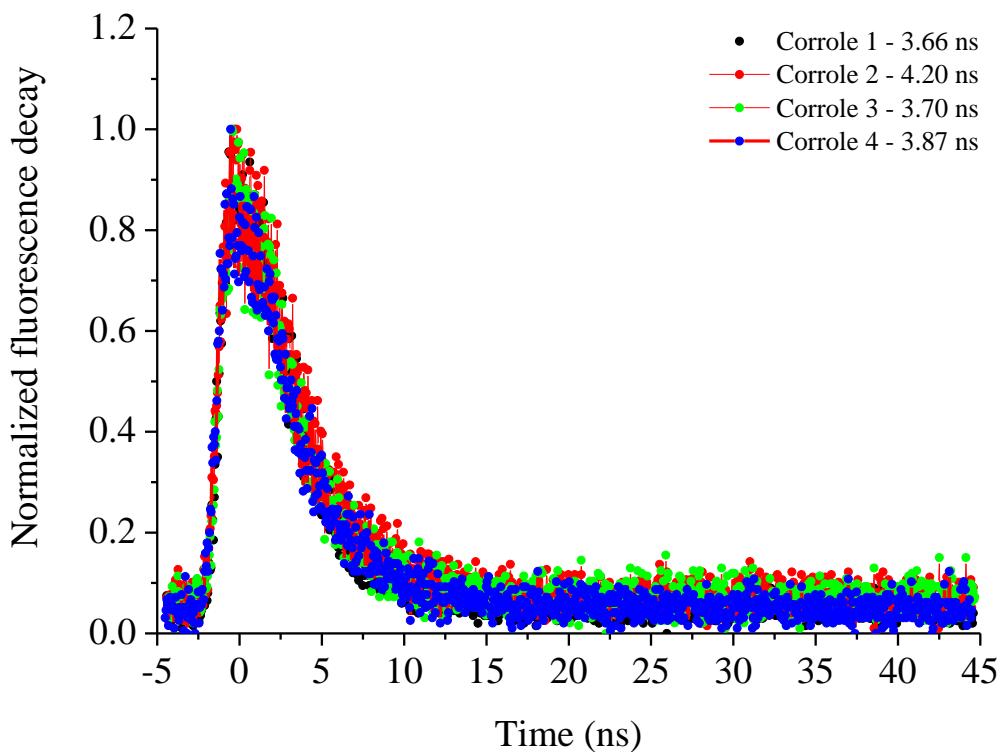


Figure S5. Normalized fluorescence decays of corroles **1-4** in DCM solution.

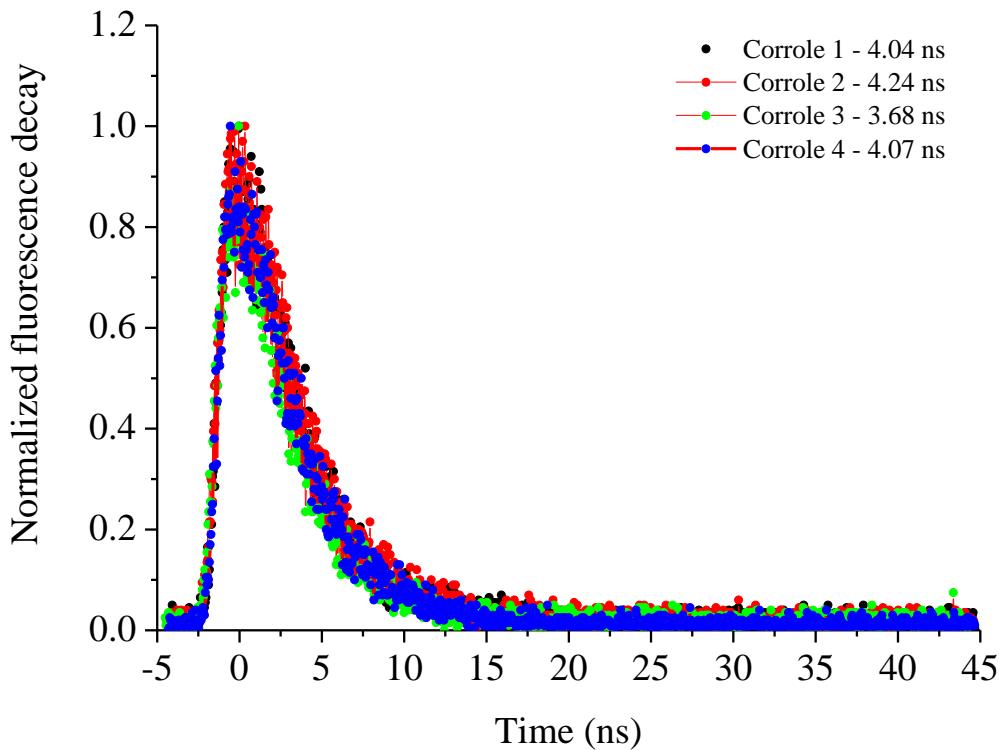


Figure S6. Normalized fluorescence decays of corroles **1-4** in ACN solution.

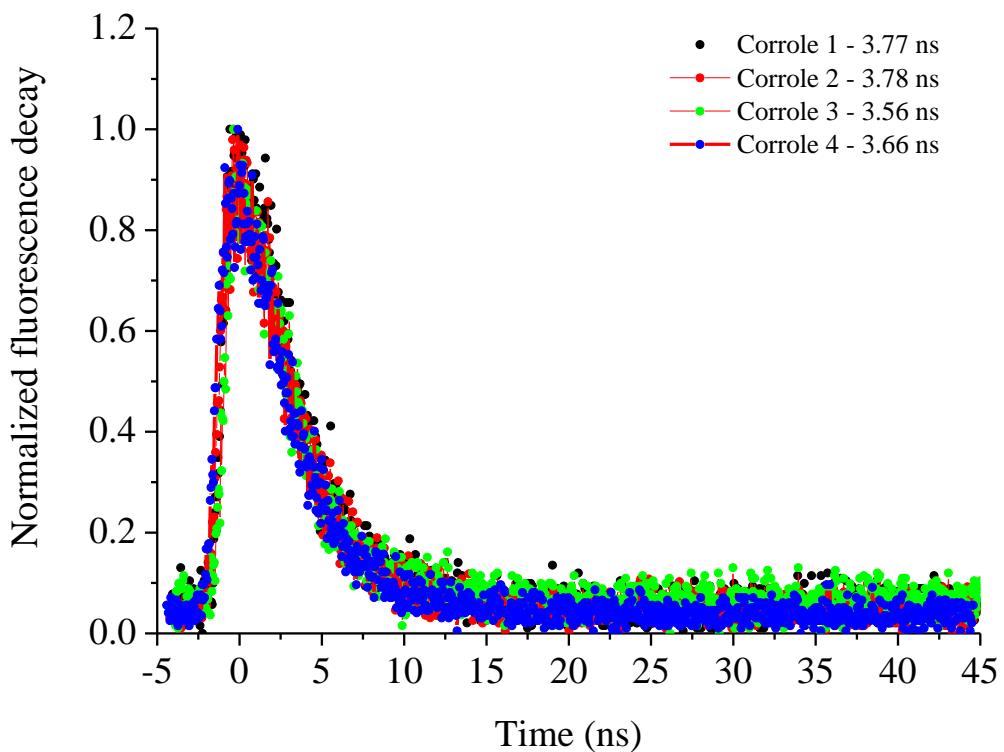


Figure S7. Normalized fluorescence decays of corroles **1-4** in MeOH solution.

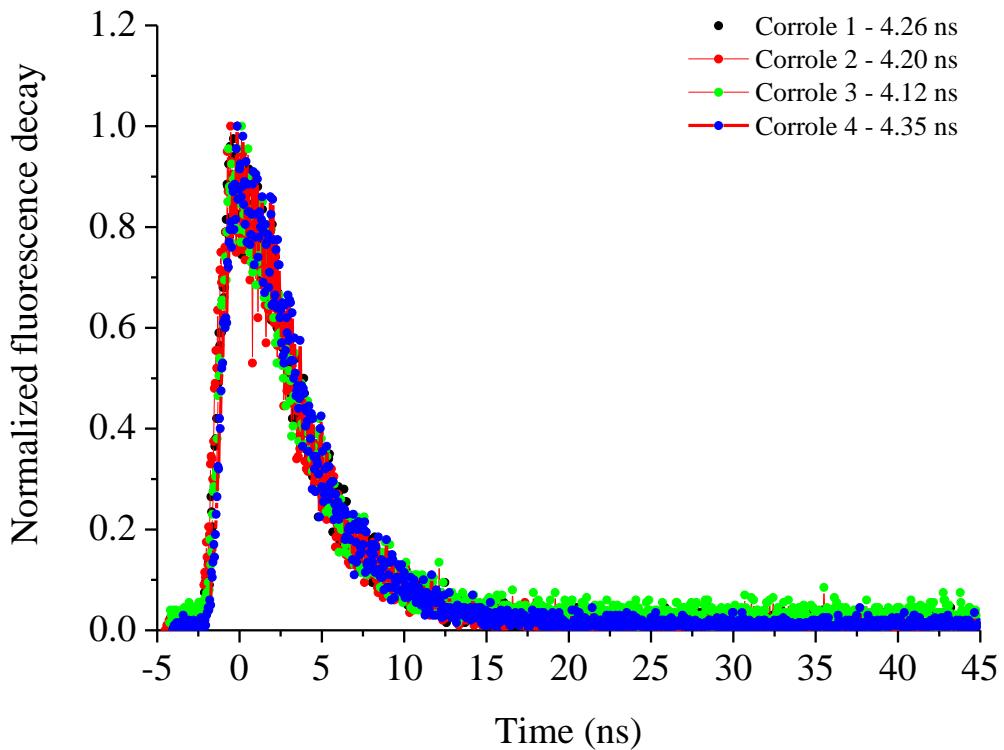


Figure S8. Normalized fluorescence decays of corroles **1-4** in DMSO solution.

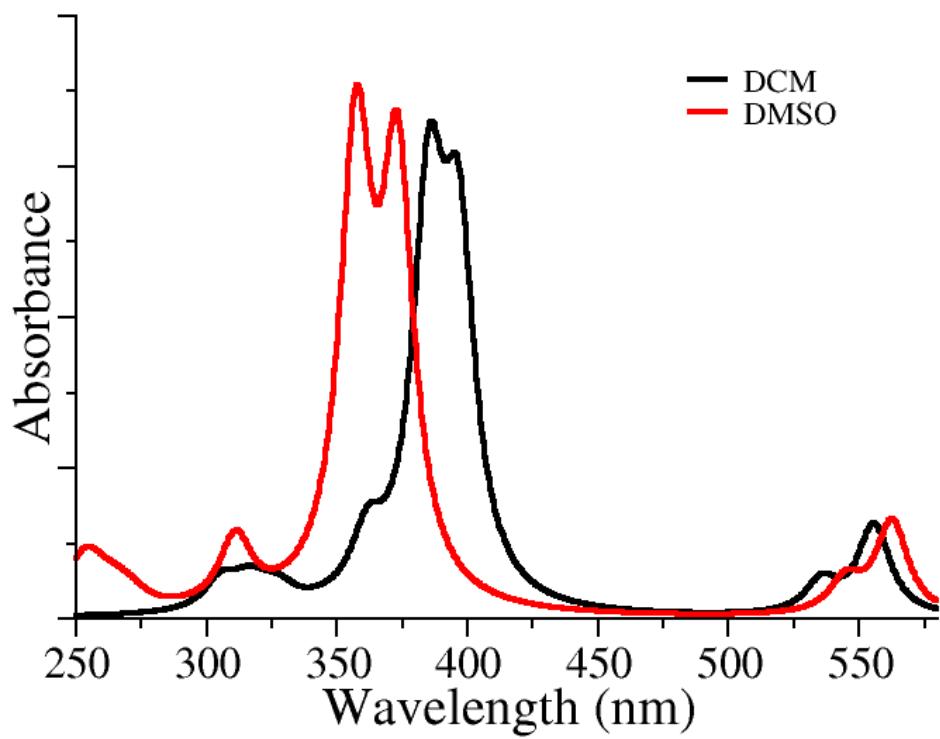


Figure S9. Theoretical optical absorption spectra in DCM and DMSO of corrole **1**.

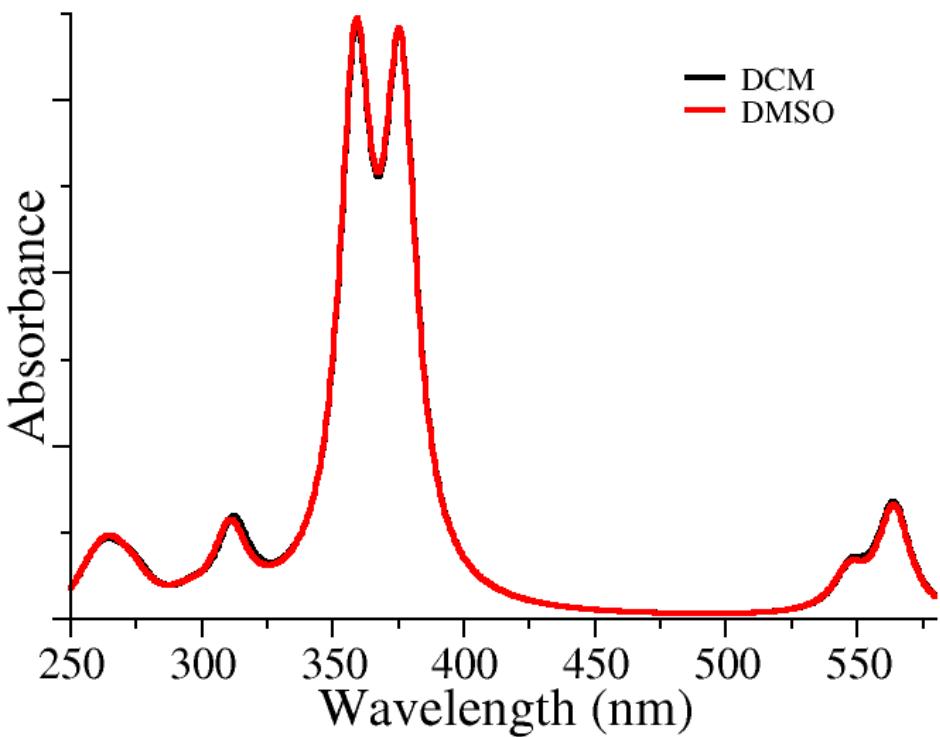


Figure S10. Theoretical optical absorption spectra in DCM and DMSO of corrole **2**.

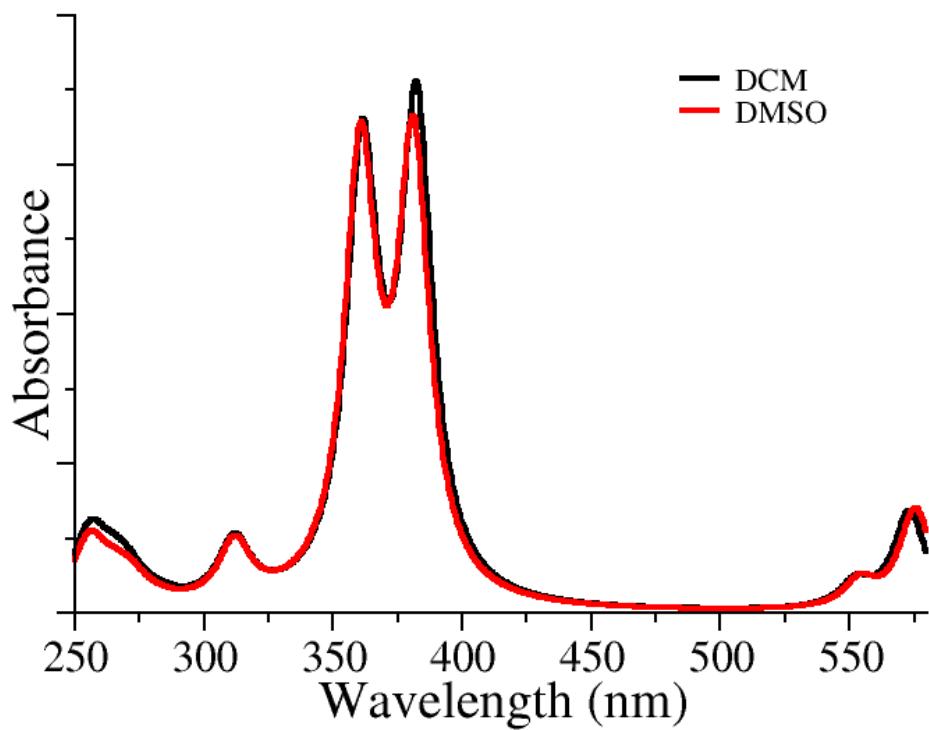


Figure S11. Theoretical optical absorption spectra in DCM and DMSO of corrole **3**.

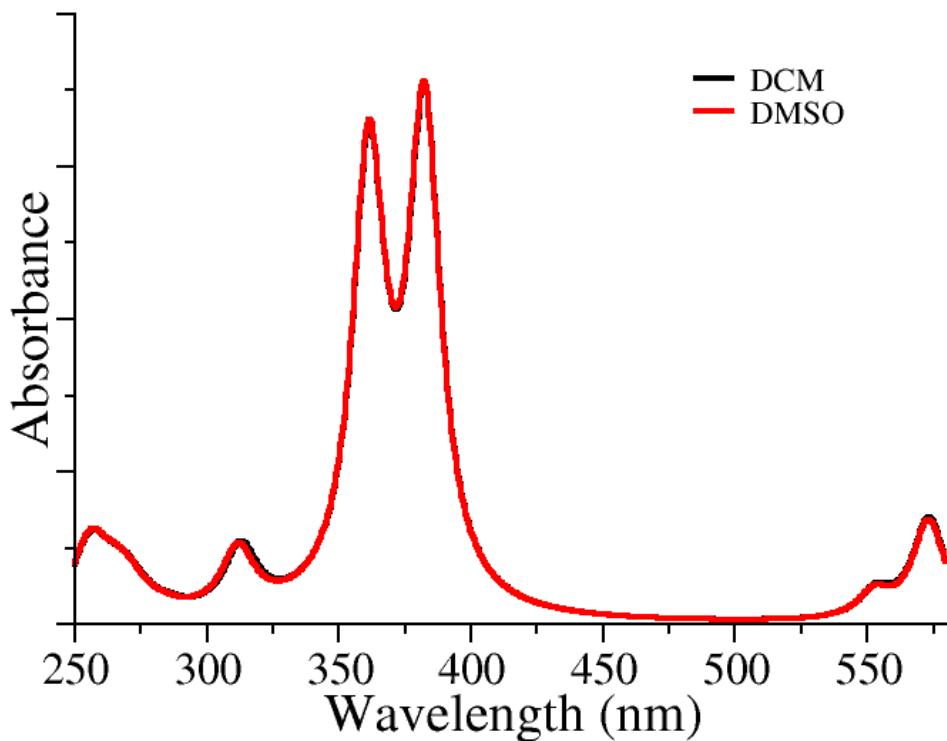


Figure S12. Theoretical optical absorption spectra in DCM and DMSO of corrole **4**.

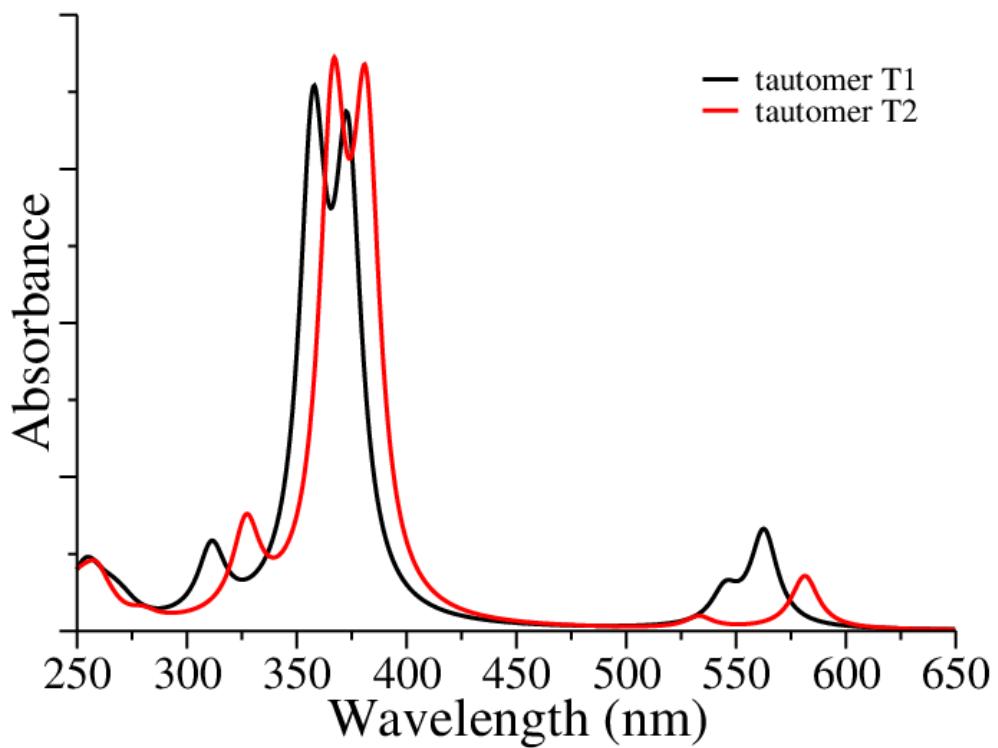


Figure S13. Theoretical optical absorption spectra in DMSO of tautomeric states of corrole 1.

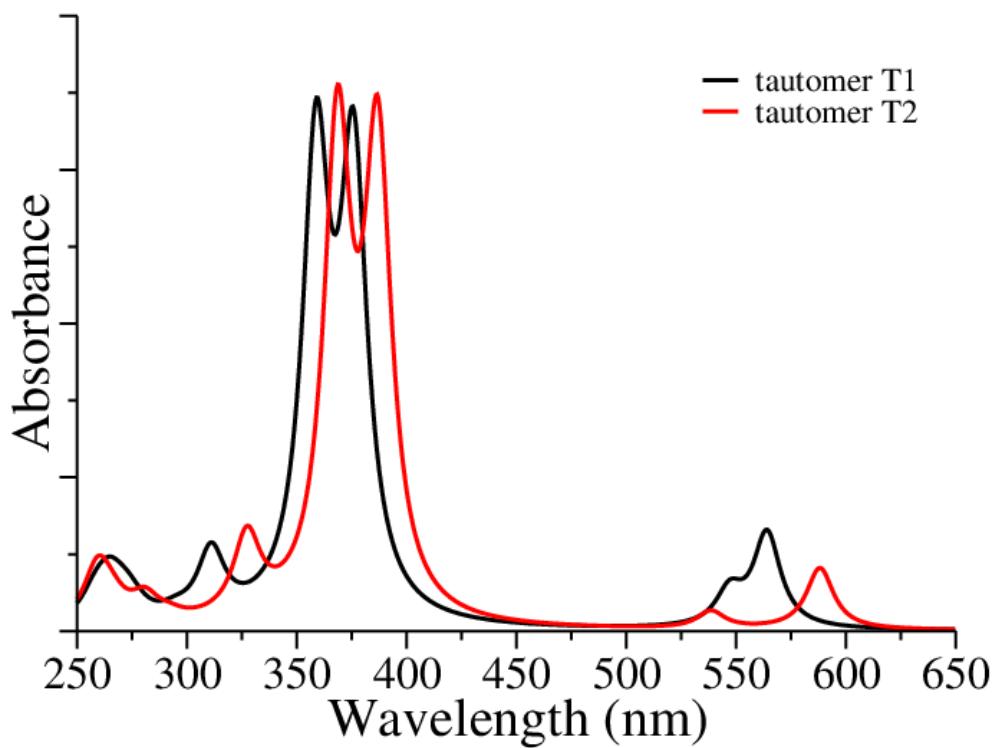


Figure S14. Theoretical optical absorption spectra in DMSO of tautomeric states of corrole 2.

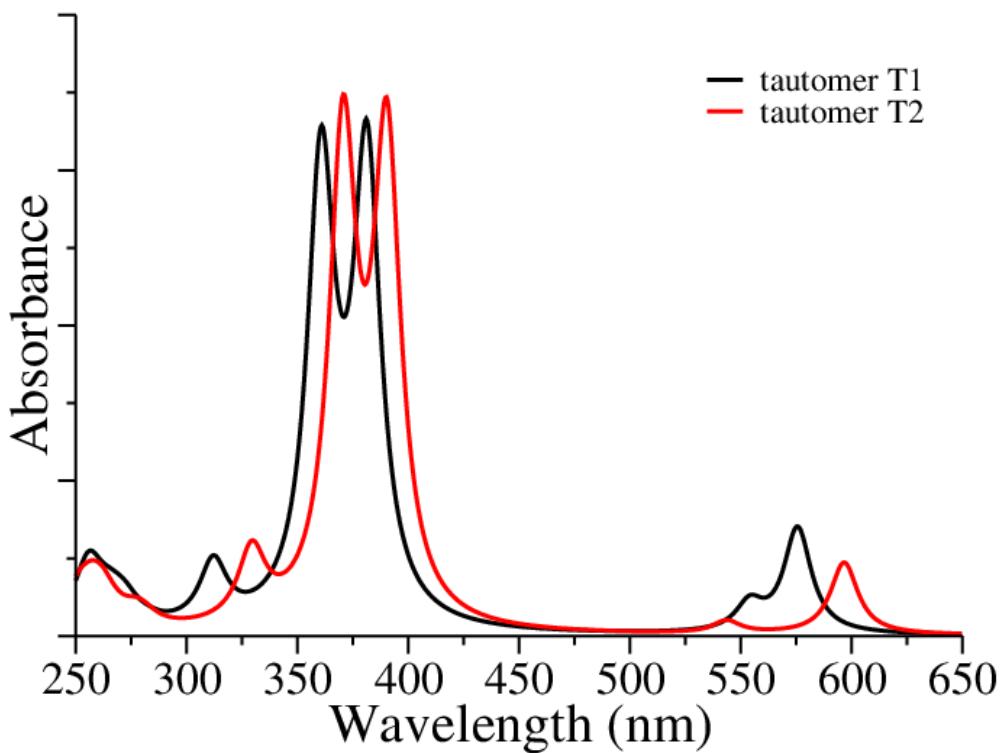


Figure S15. Theoretical optical absorption spectra in DMSO of tautomeric states of corrole 3.

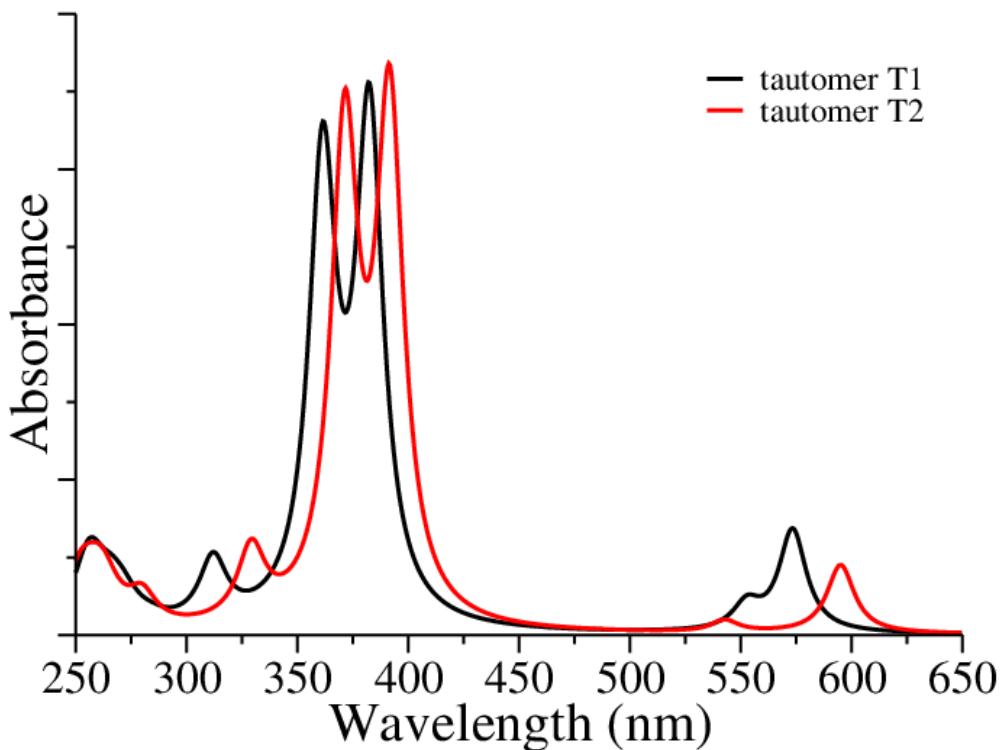


Figure S16. Theoretical optical absorption spectra in DMSO of tautomeric states of corrole 4.

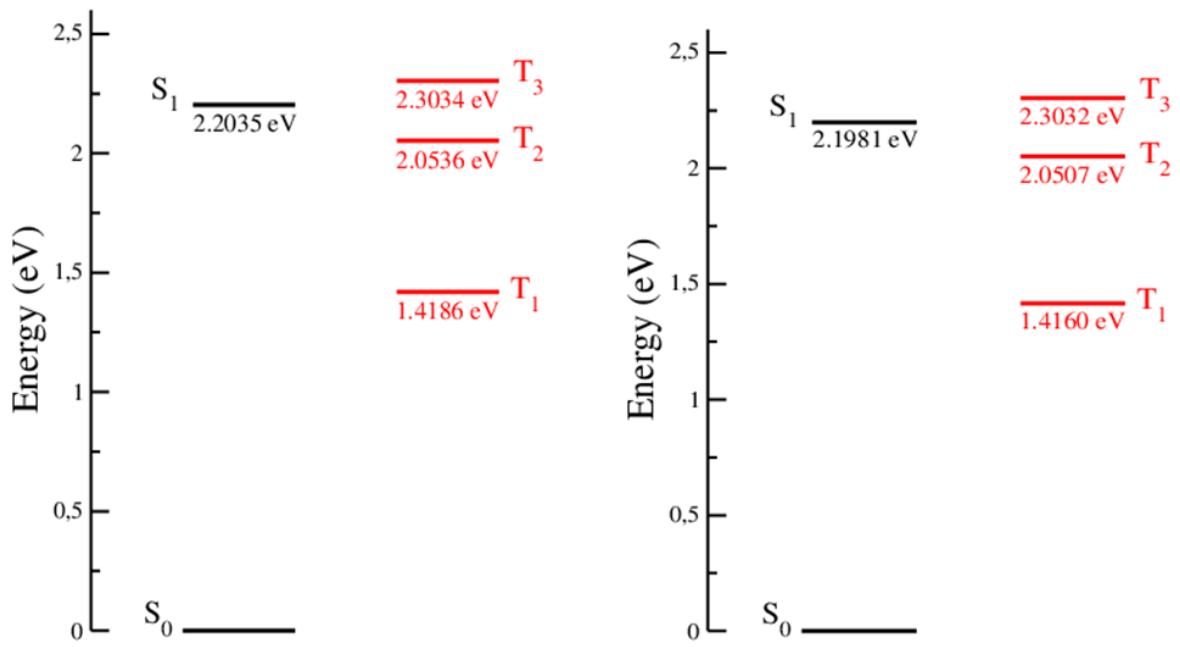


Figure S17. Total energies of the S_1 and T_1 , T_2 and T_3 excited states, relative to the ground state S_0 of corroles **1** (left panel), and **2** (right panel).

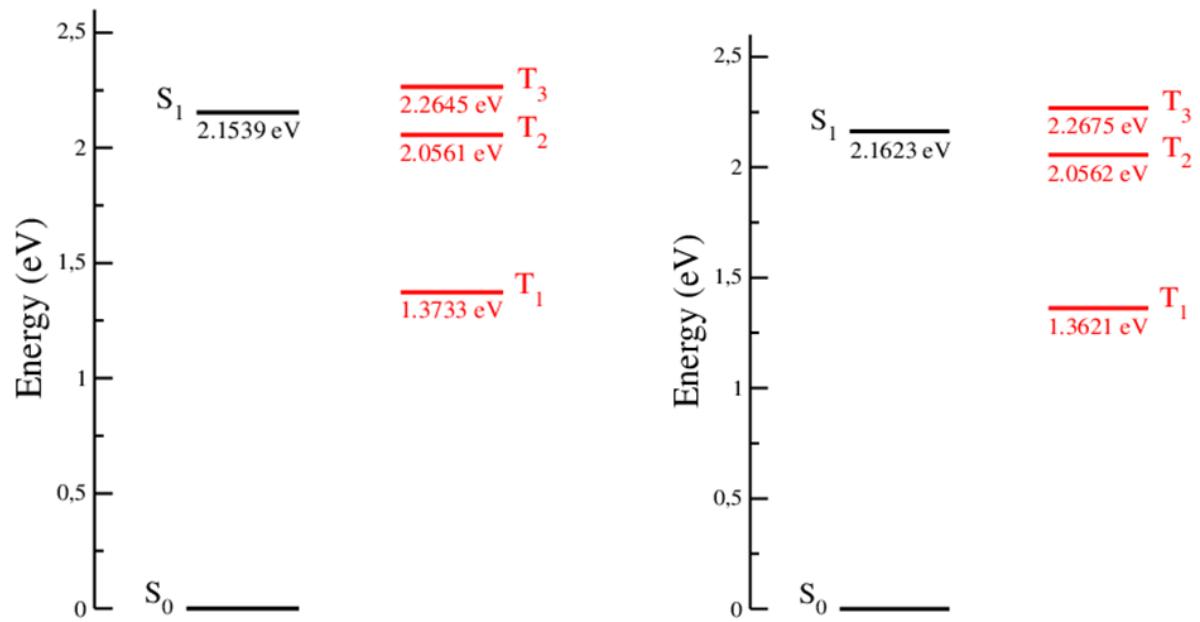


Figure S18. Total energies of the S_1 and T_1 , T_2 and T_3 excited states, relative to the ground state S_0 of corroles **3** (left panel), and **4** (right panel).

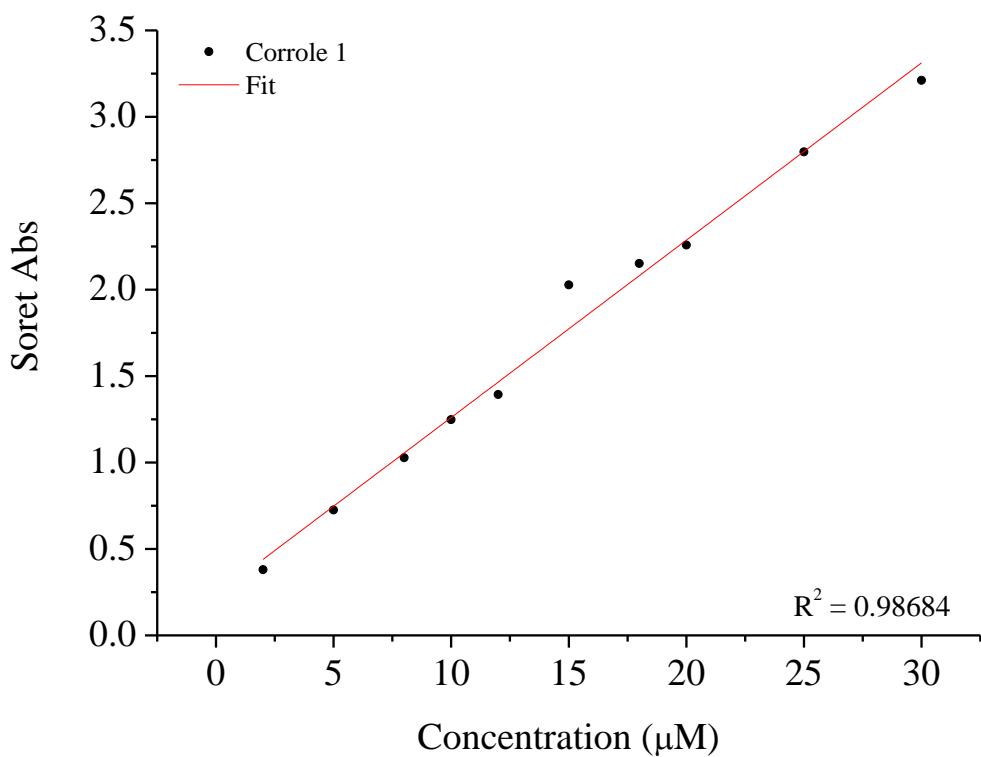
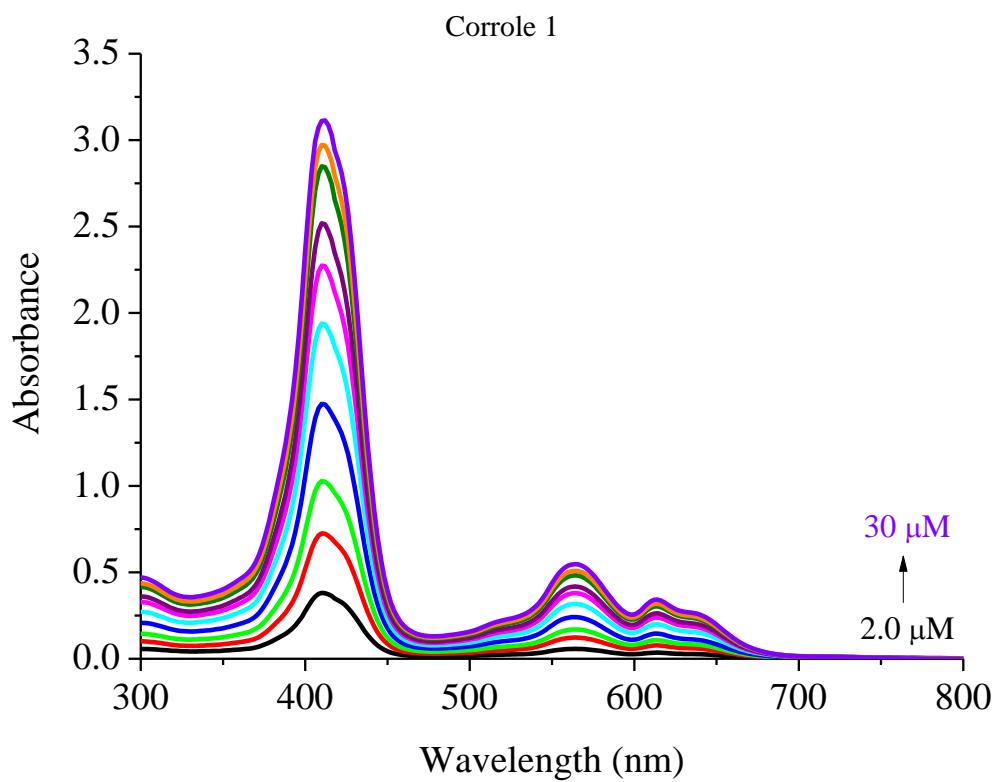


Figure S19. (up) Aggregation behavior of corrole **1** in DCM solution and (down) $\text{Abs}_{\text{Soret}}$ *versus* concentration plot.

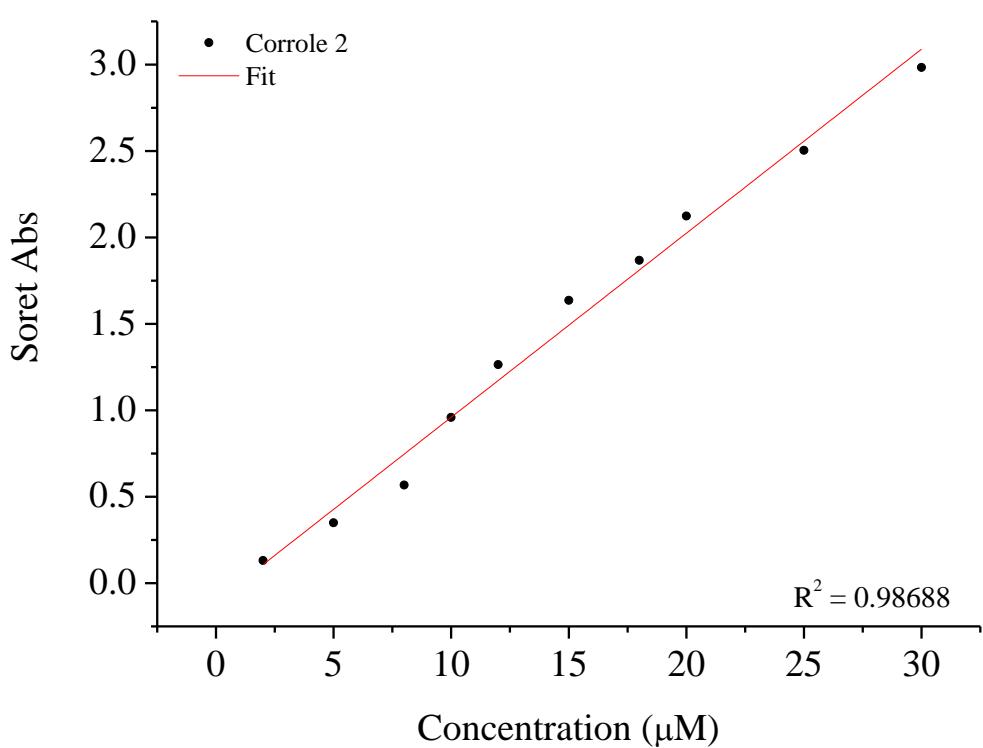
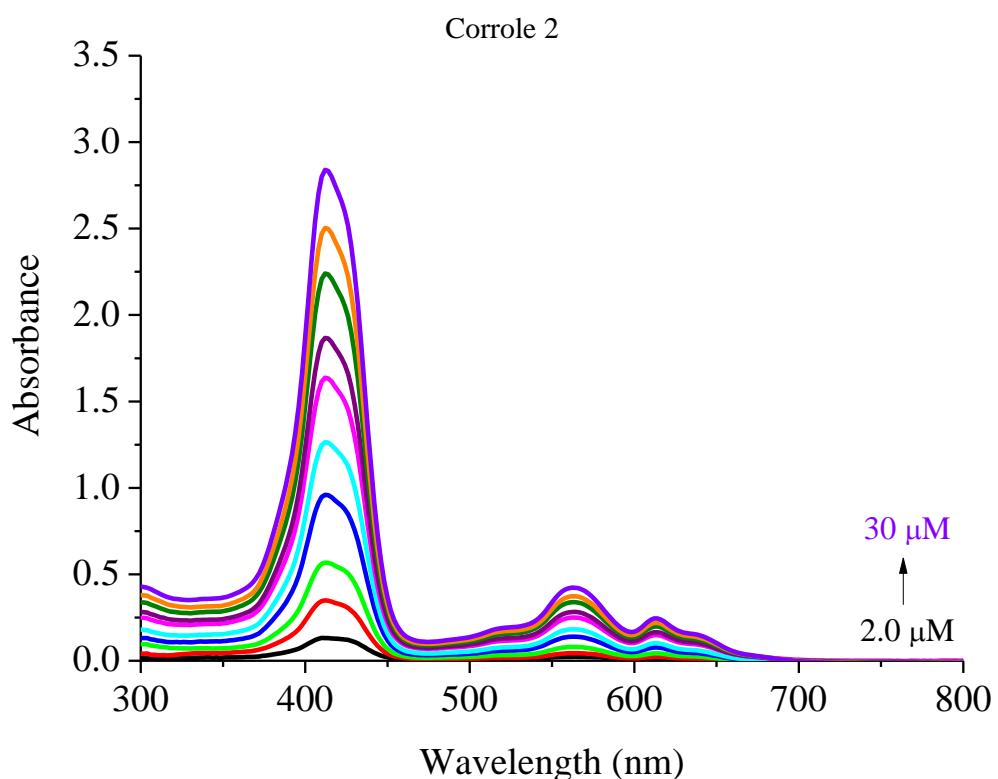


Figure S20. (up) Aggregation behavior of corrole **2** in DCM solution and (down) $\text{Abs}_{\text{Soret}}$ *versus* concentration plot.

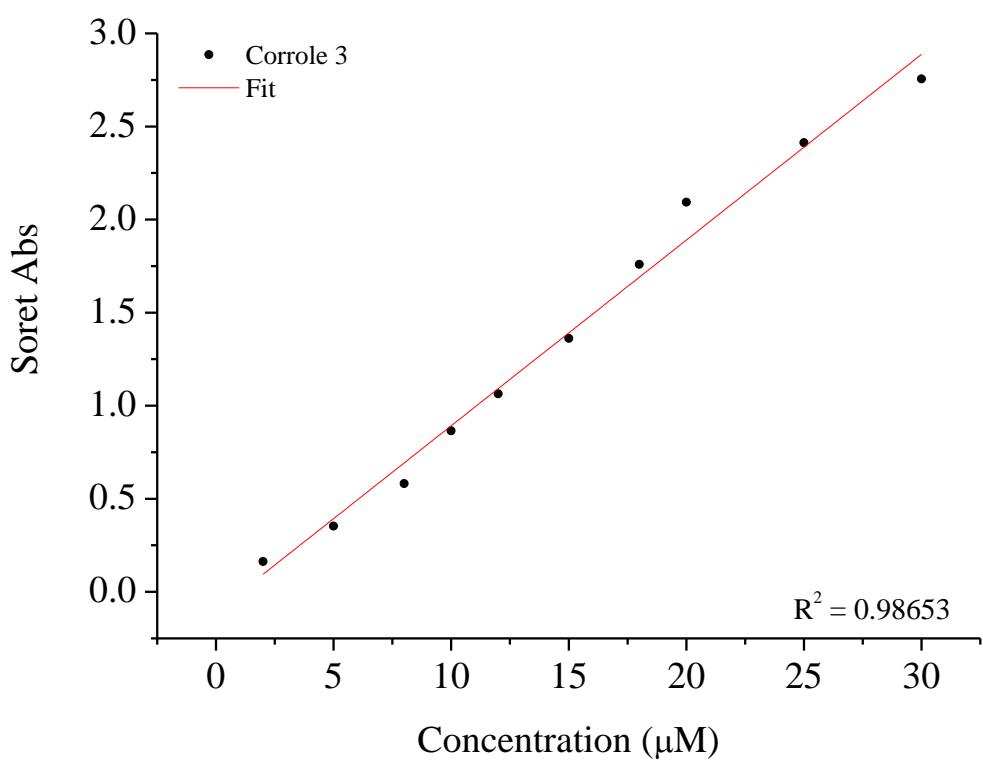
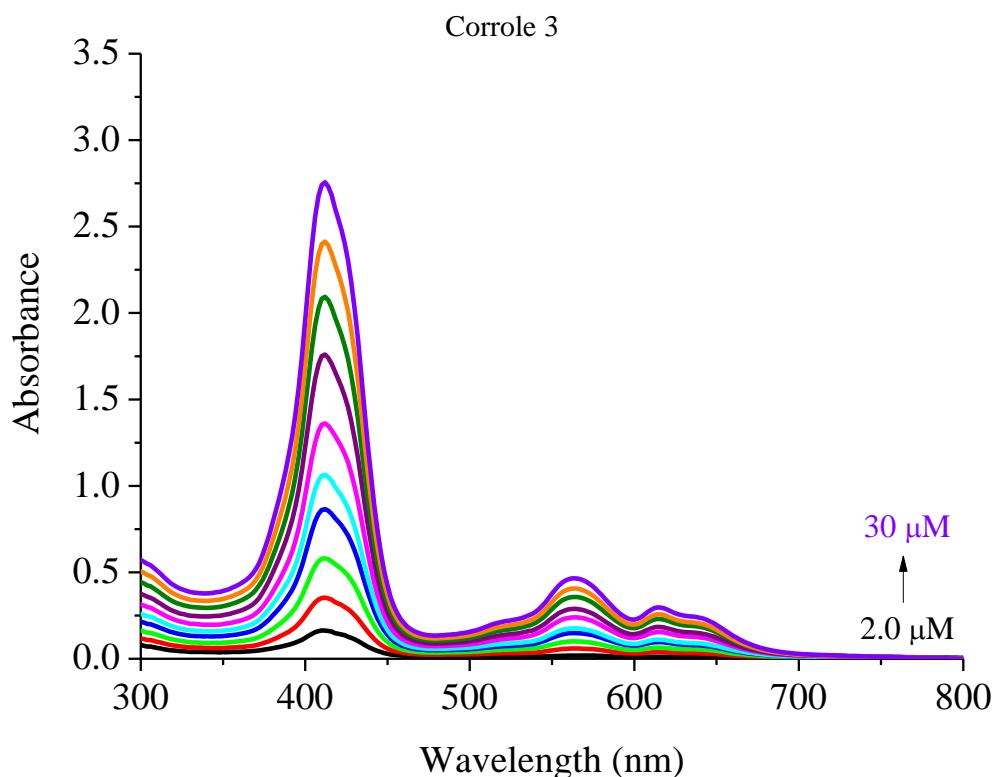


Figure S21. (up) Aggregation behavior of corrole **3** in DCM solution and (down) $\text{Abs}_{\text{Soret}}$ *versus* concentration plot.

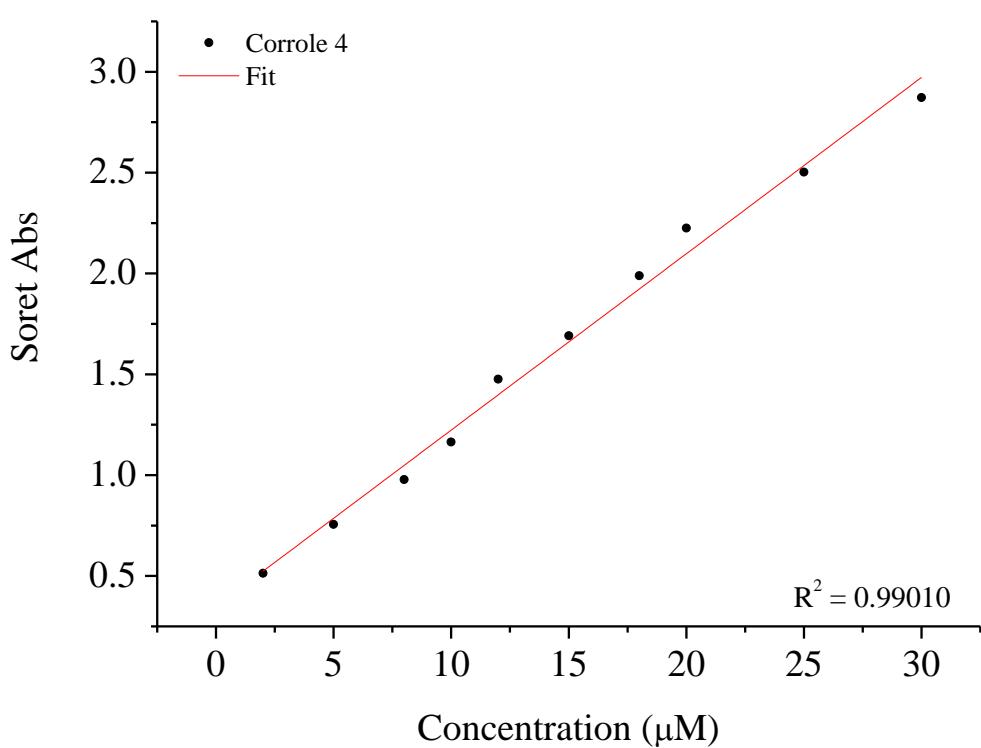
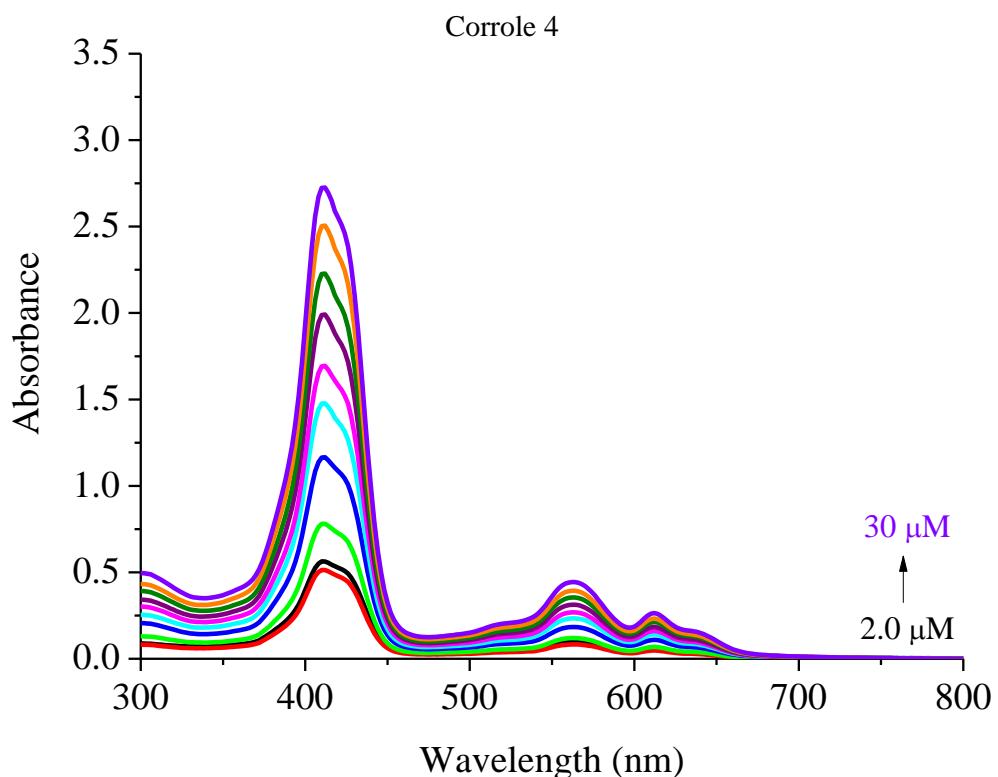


Figure S22. (up) Aggregation behavior of corrole **4** in DCM solution and (down) $\text{Abs}_{\text{Soret}}$ *versus* concentration plot.

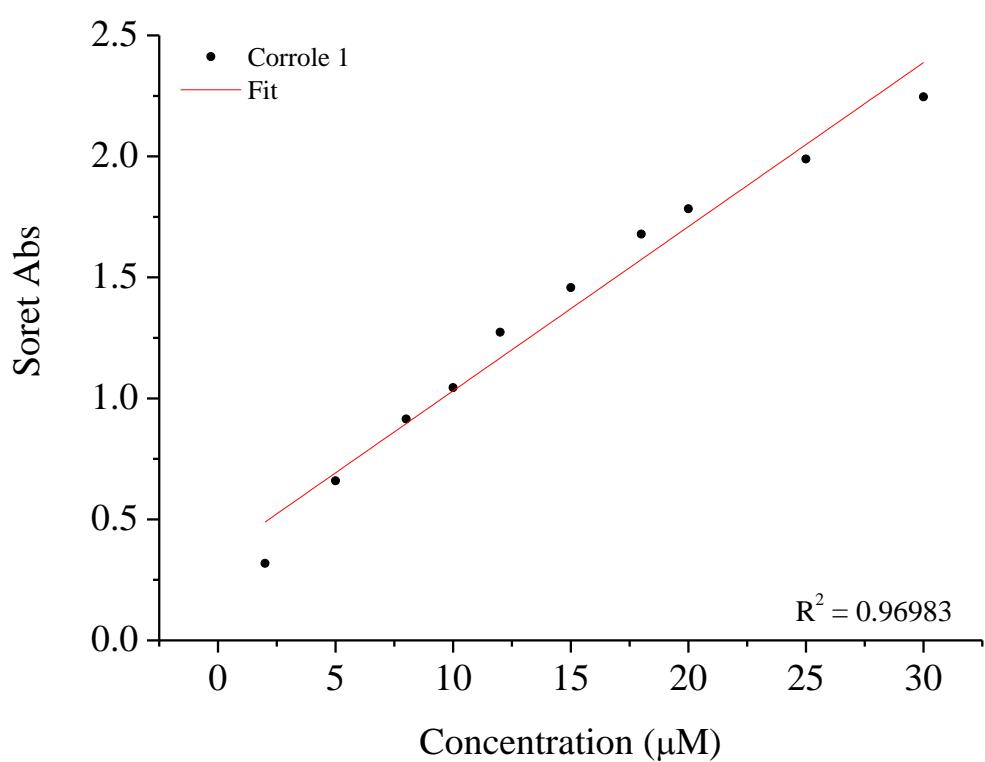
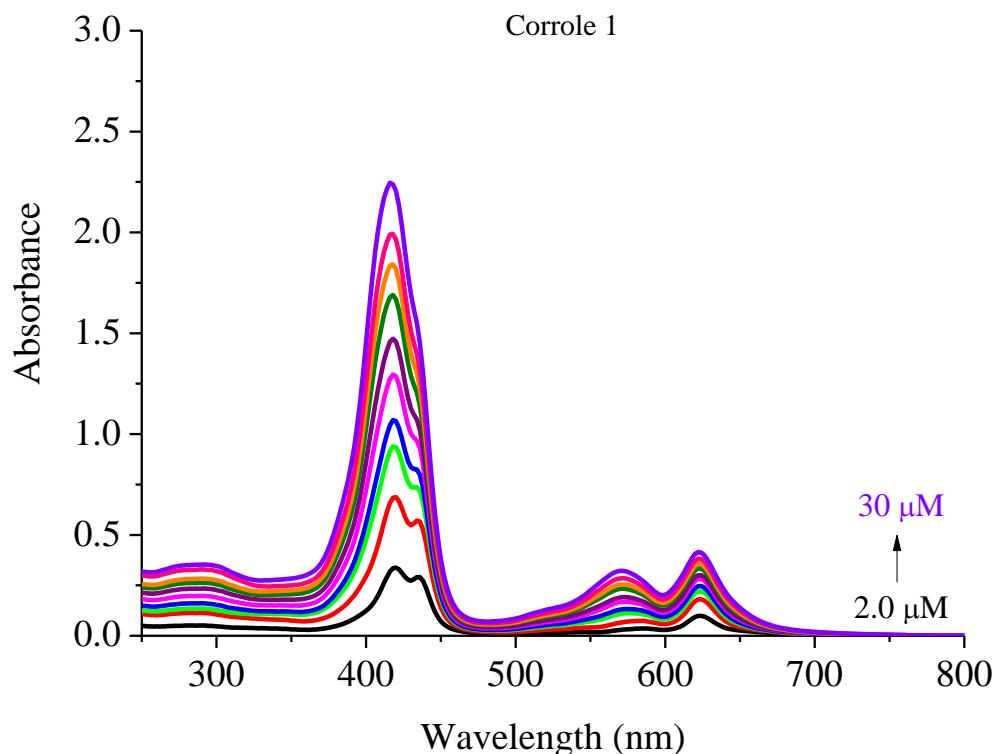


Figure S23. (up) Aggregation behavior of corrole **1** in ACN solution and (down) $\text{Abs}_{\text{Soret}}$ *versus* concentration plot.

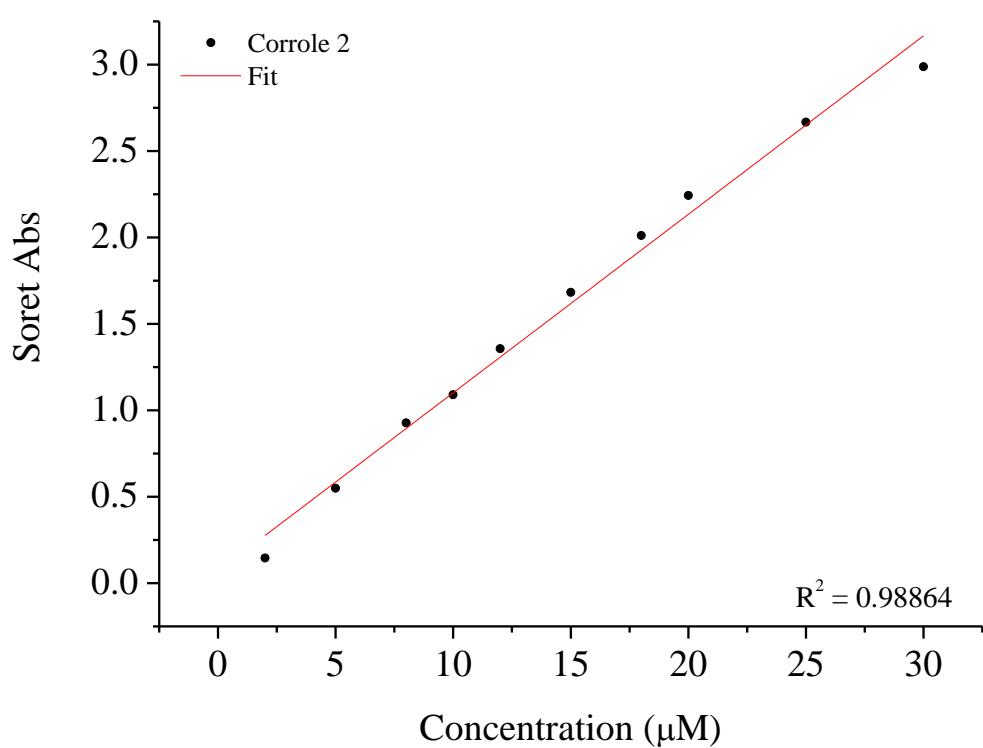
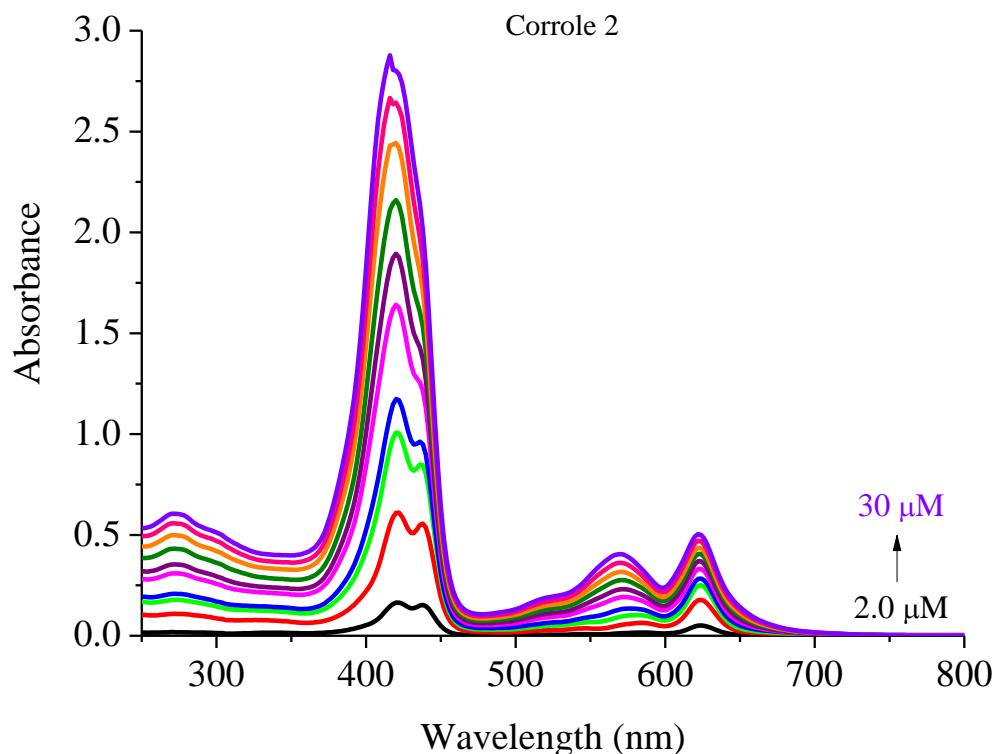


Figure S24. (up) Aggregation behavior of corrole **2** in ACN solution and (down) $\text{Abs}_{\text{Soret}}$ *versus* concentration plot.

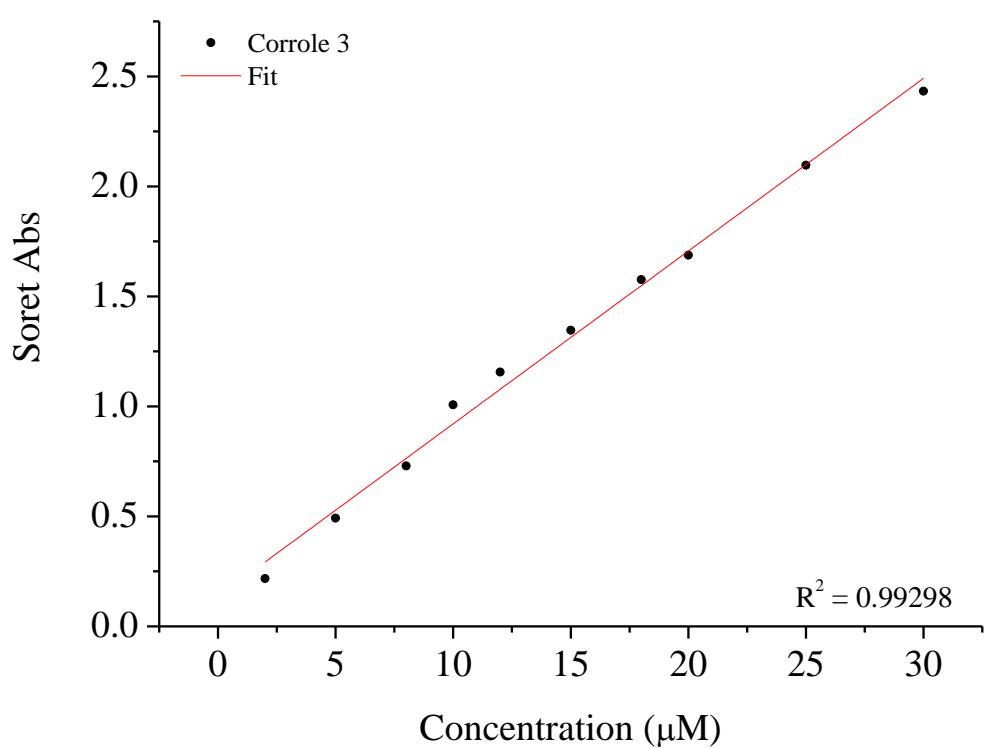
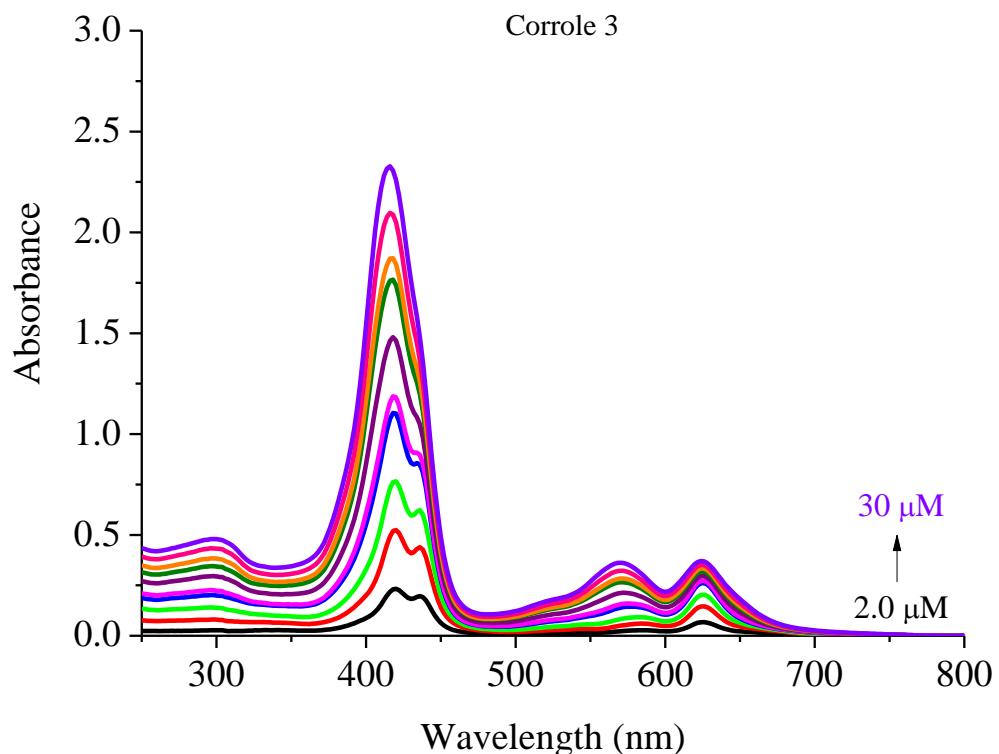


Figure S25. (up) Aggregation behavior of corrole **3** in ACN solution and (down) $\text{Abs}_{\text{Soret}}$ *versus* concentration plot.

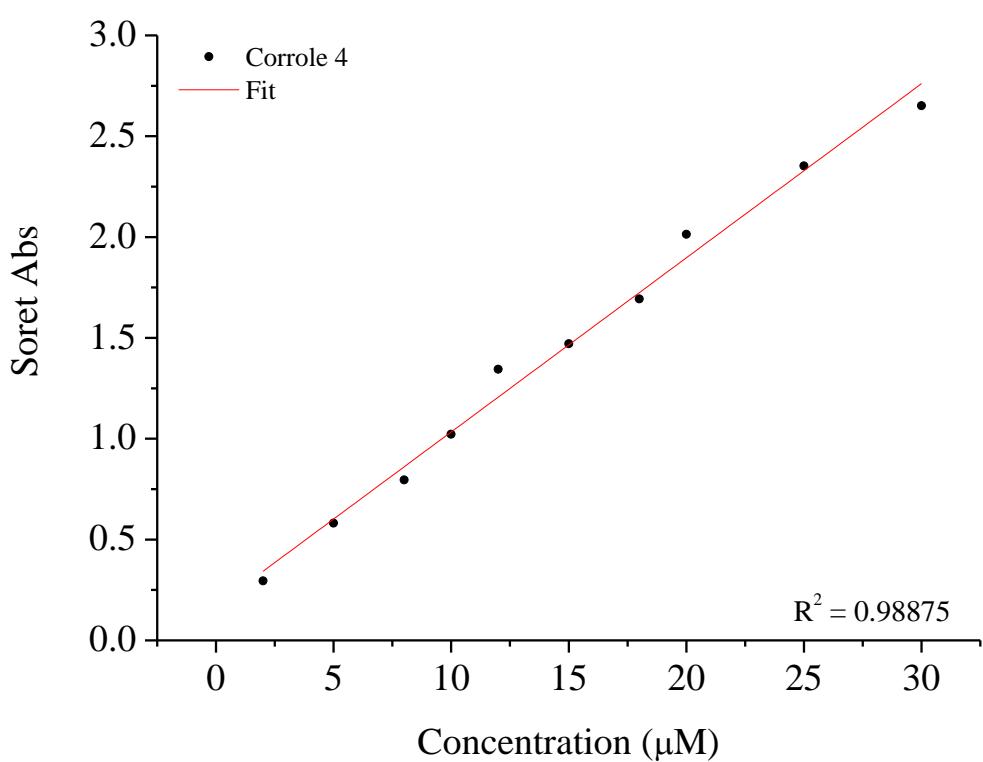
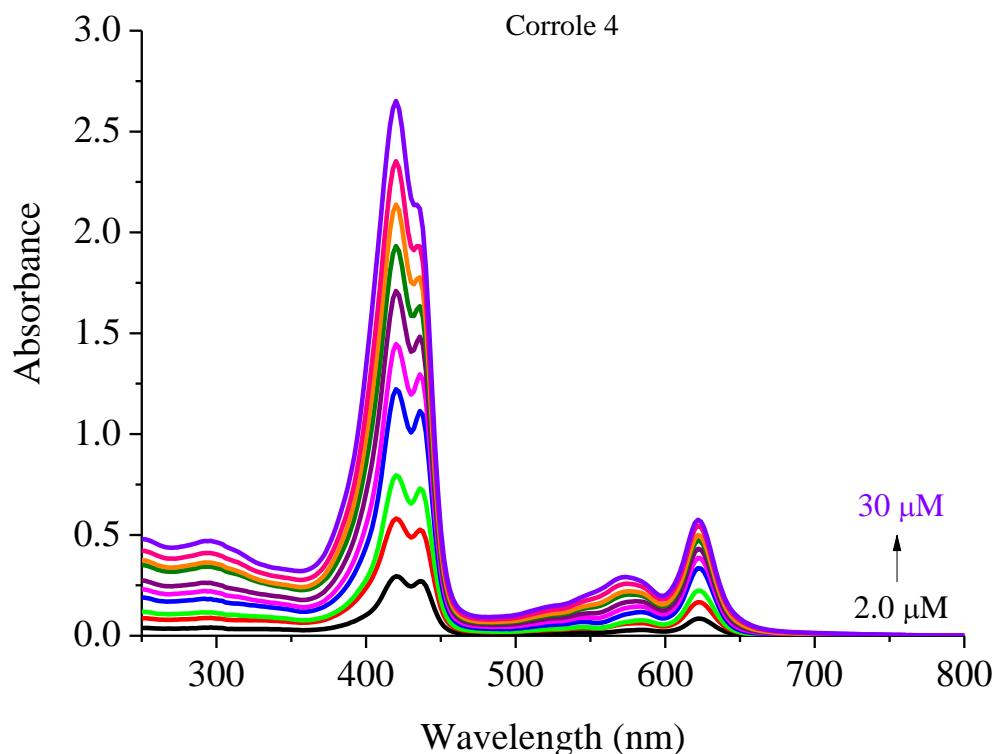


Figure S26. (up) Aggregation behavior of corrole **4** in ACN solution and (down) $\text{Abs}_{\text{Soret}}$ versus concentration plot.

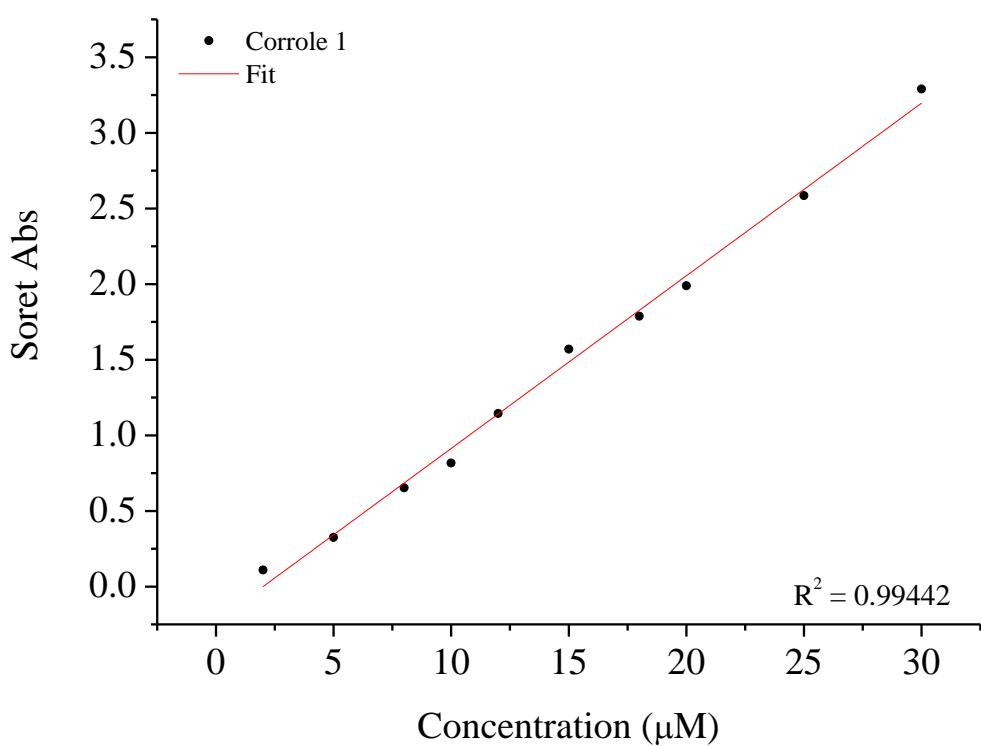
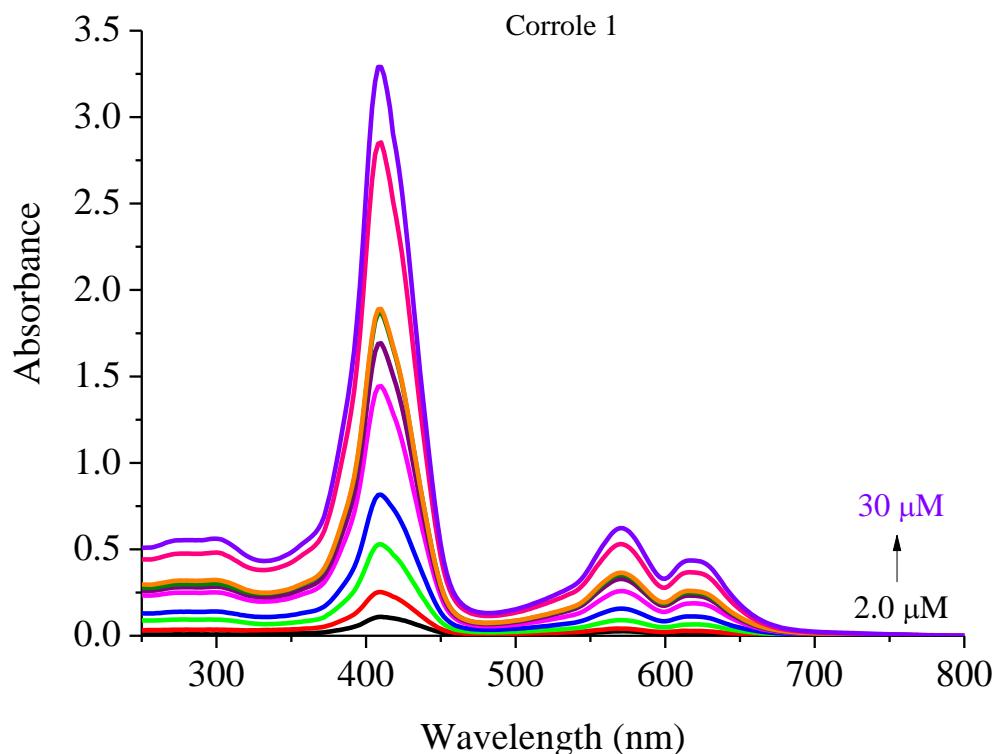


Figure S27. (up) Aggregation behavior of corrole **1** in MeOH solution and (down) $\text{Abs}_{\text{Soret}}$ versus concentration plot.

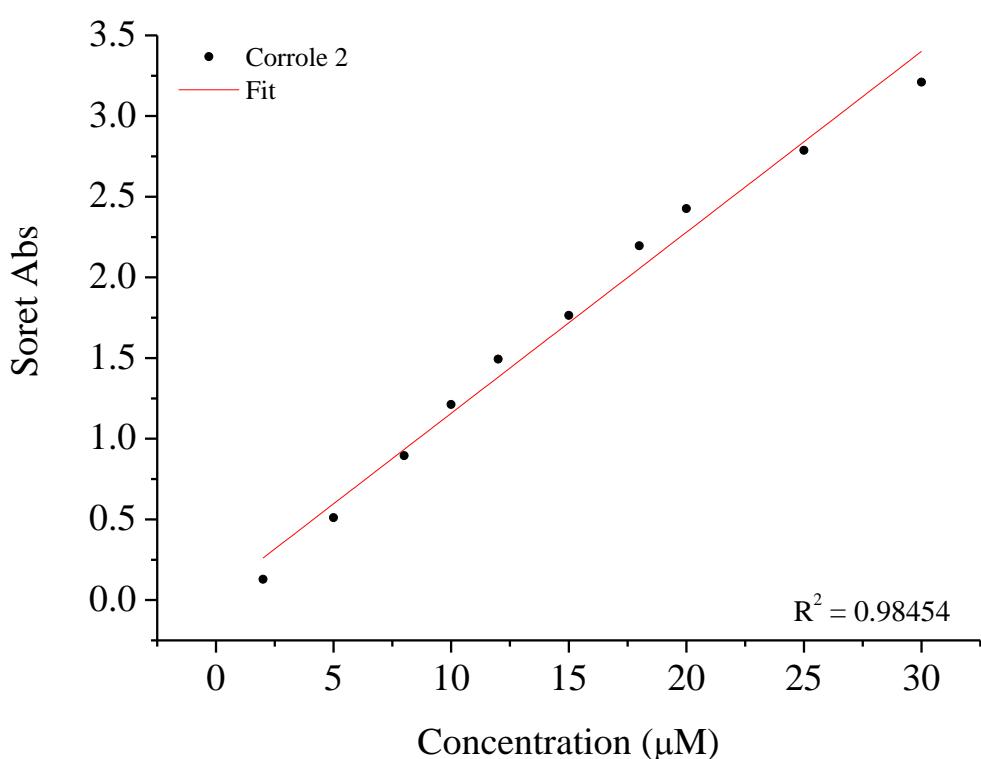
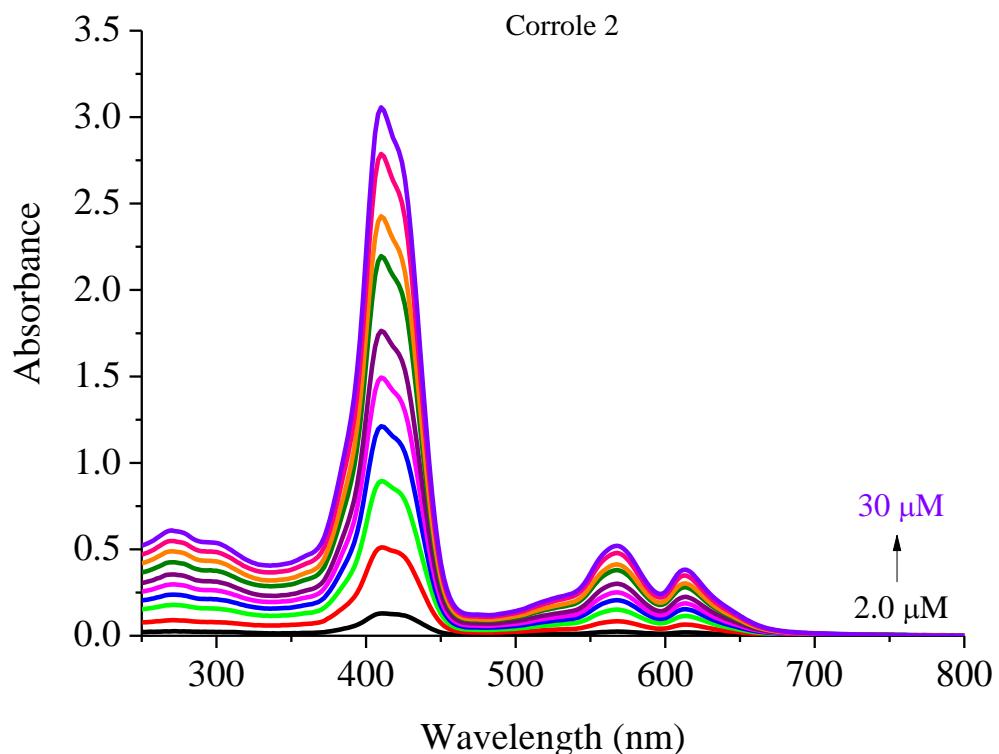


Figure S28. (up) Aggregation behavior of corrole **2** in MeOH solution and (down) $\text{Abs}_{\text{Soret}}$ versus concentration plot.

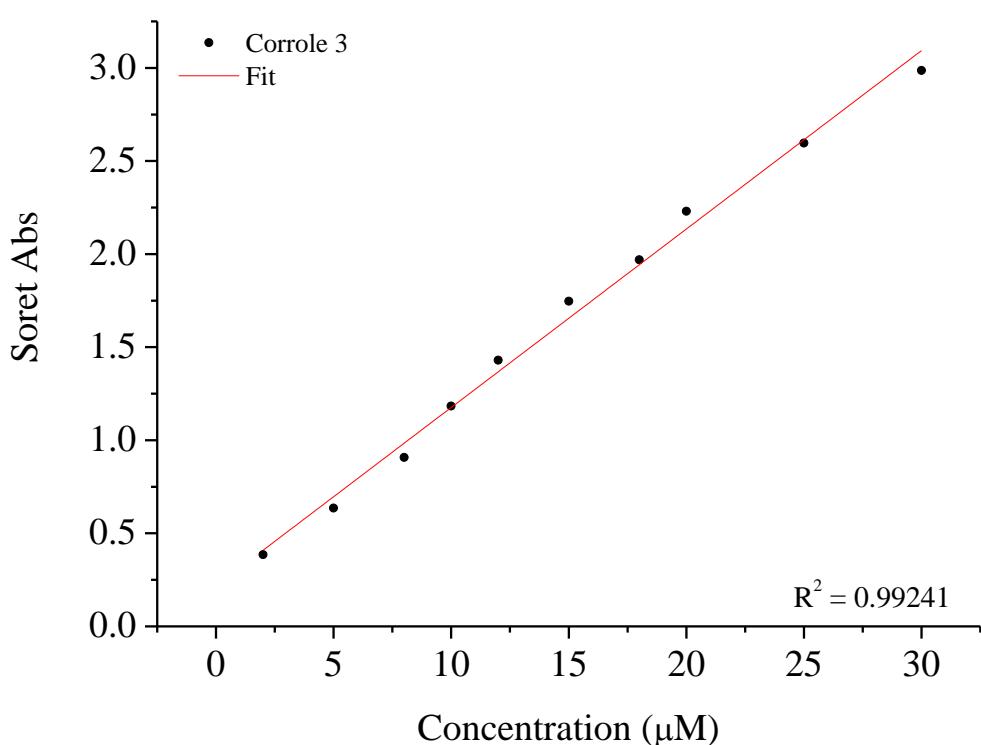
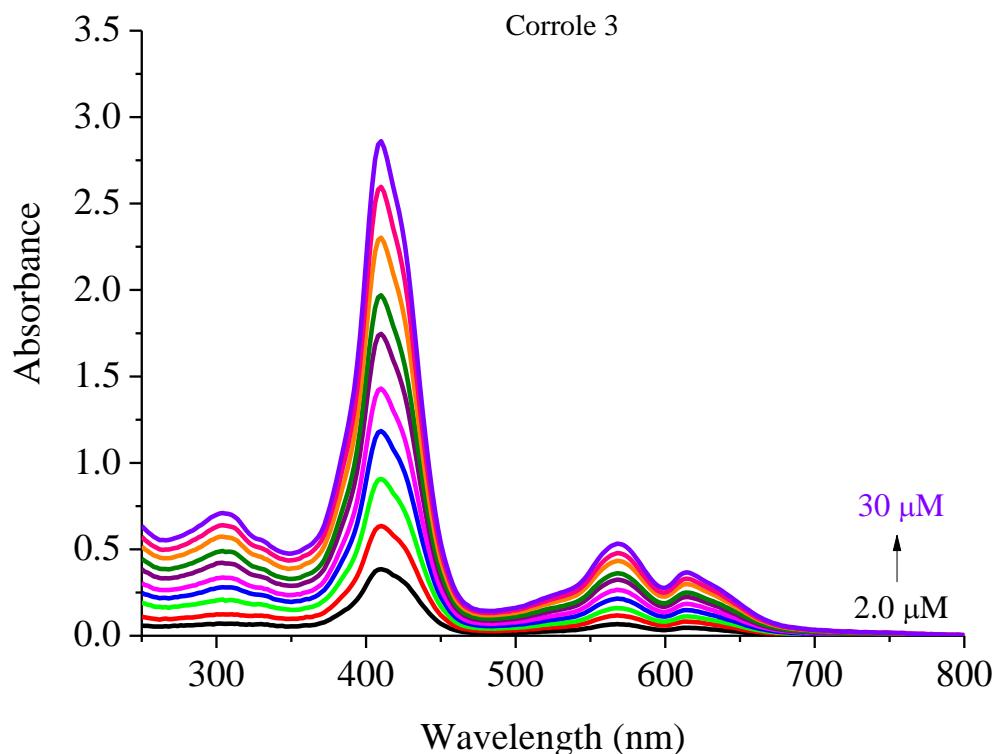


Figure S29. (up) Aggregation behavior of corrole **3** in MeOH solution and (down) $\text{Abs}_{\text{Soret}}$ versus concentration plot.

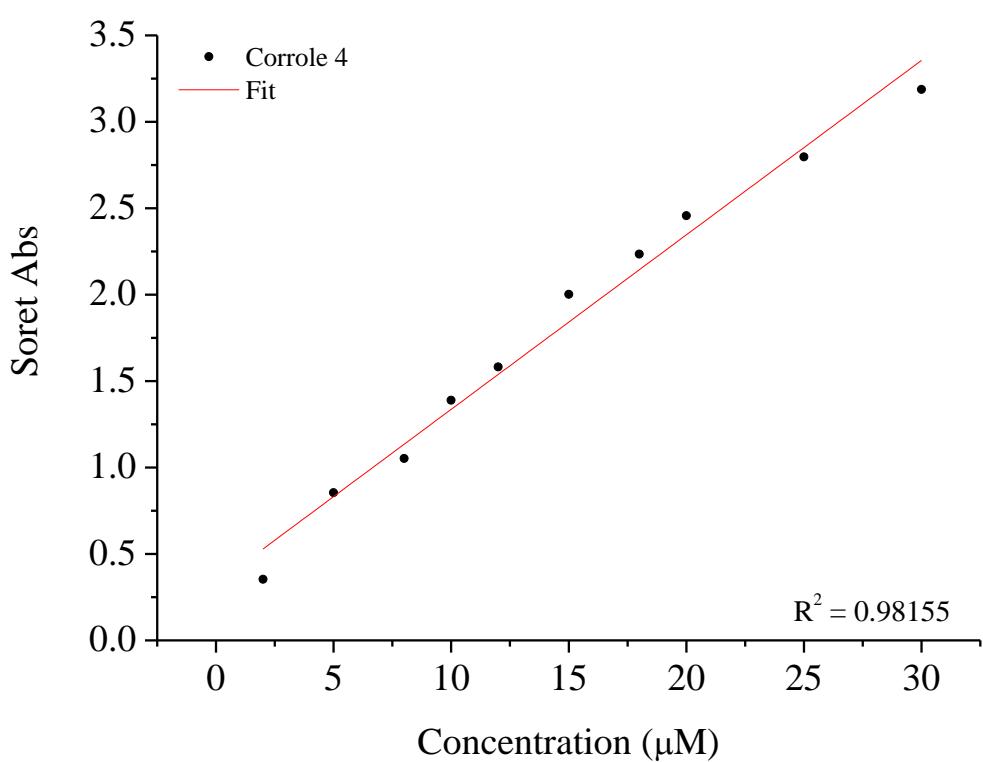
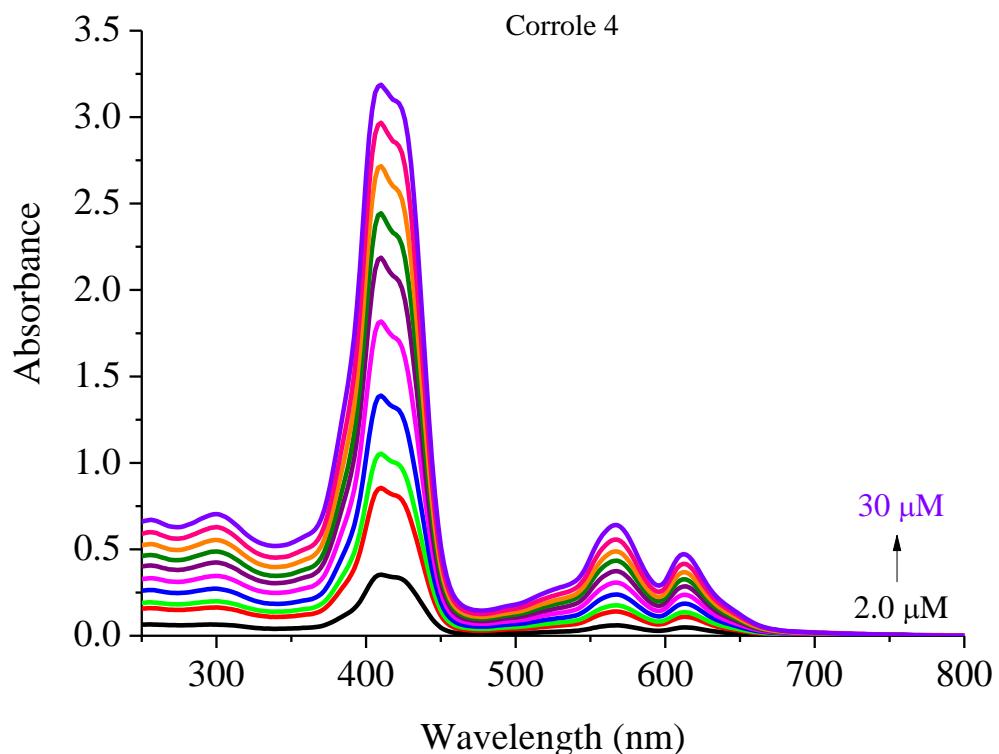


Figure S30. (up) Aggregation behavior of corrole **4** in MeOH solution and (down) $\text{Abs}_{\text{Soret}}$ versus concentration plot.

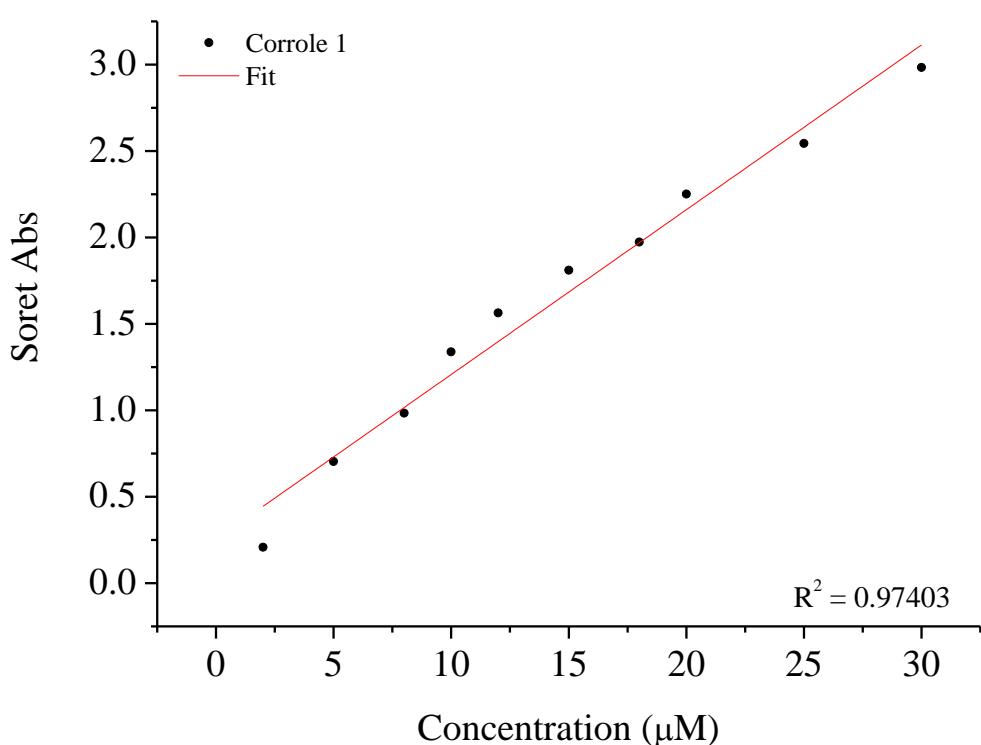
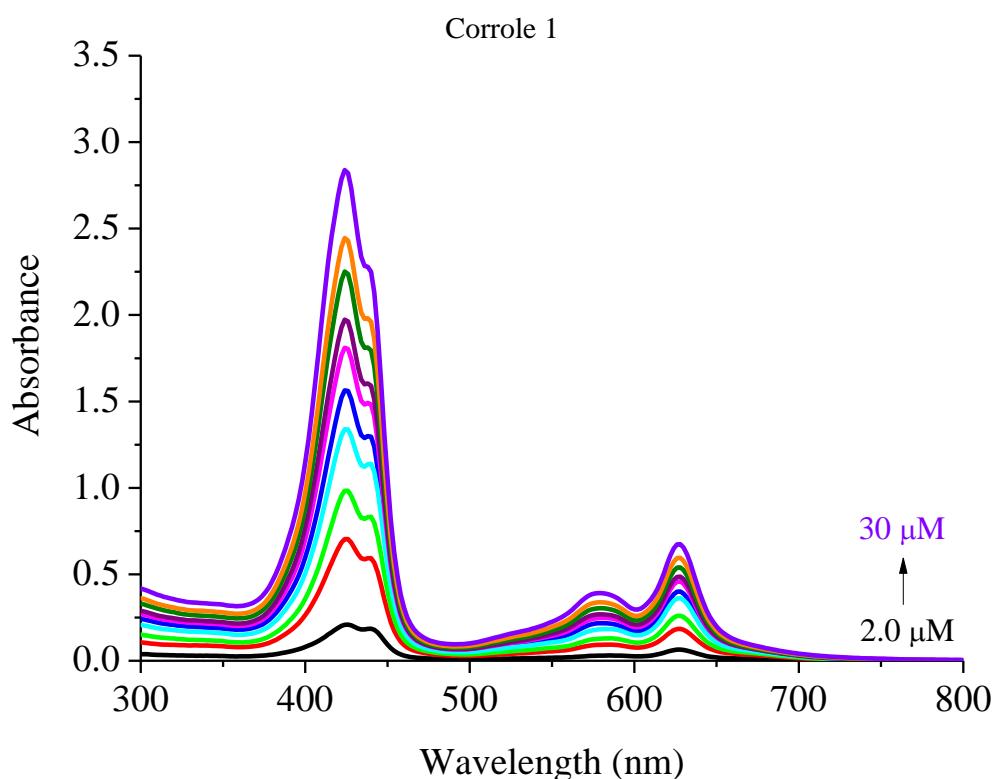


Figure S31. (up) Aggregation behavior of corrole **1** in DMSO solution and (down) $\text{Abs}_{\text{Soret}}$ versus concentration plot.

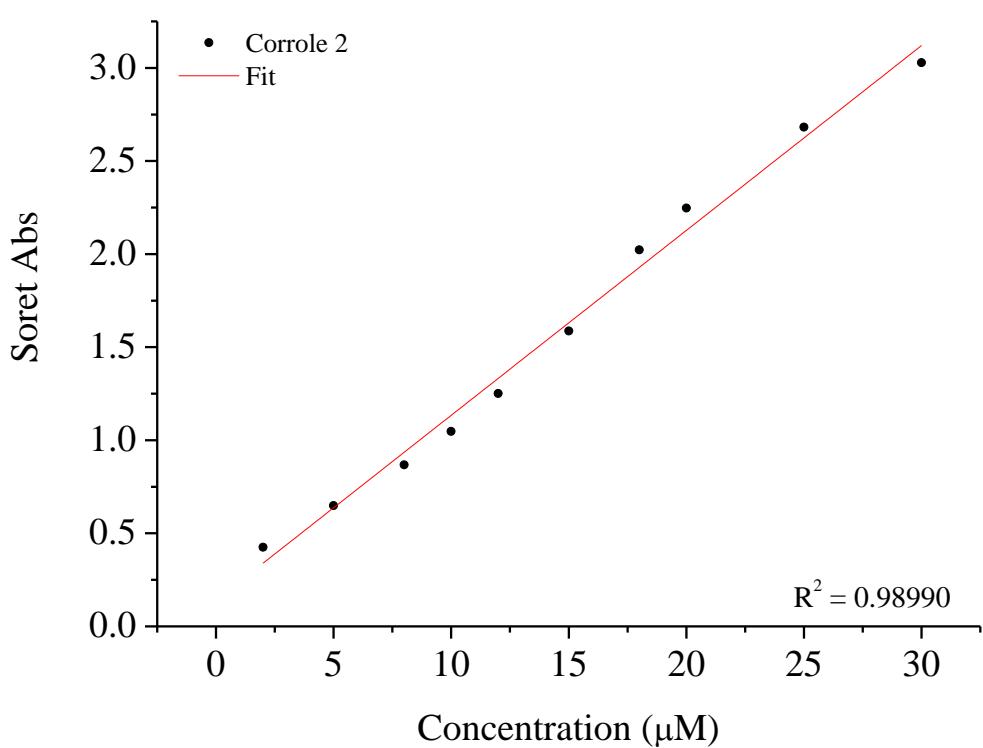
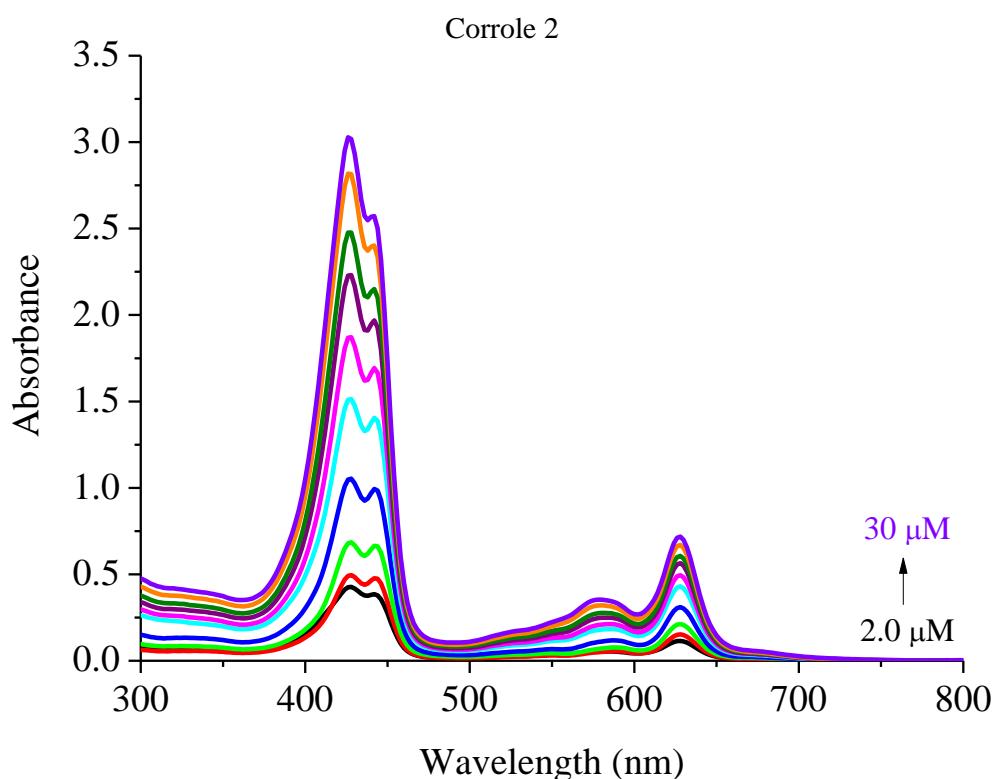


Figure S32. (up) Aggregation behavior of corrole **2** in DMSO solution and (down) $\text{Abs}_{\text{Soret}}$ versus concentration plot.

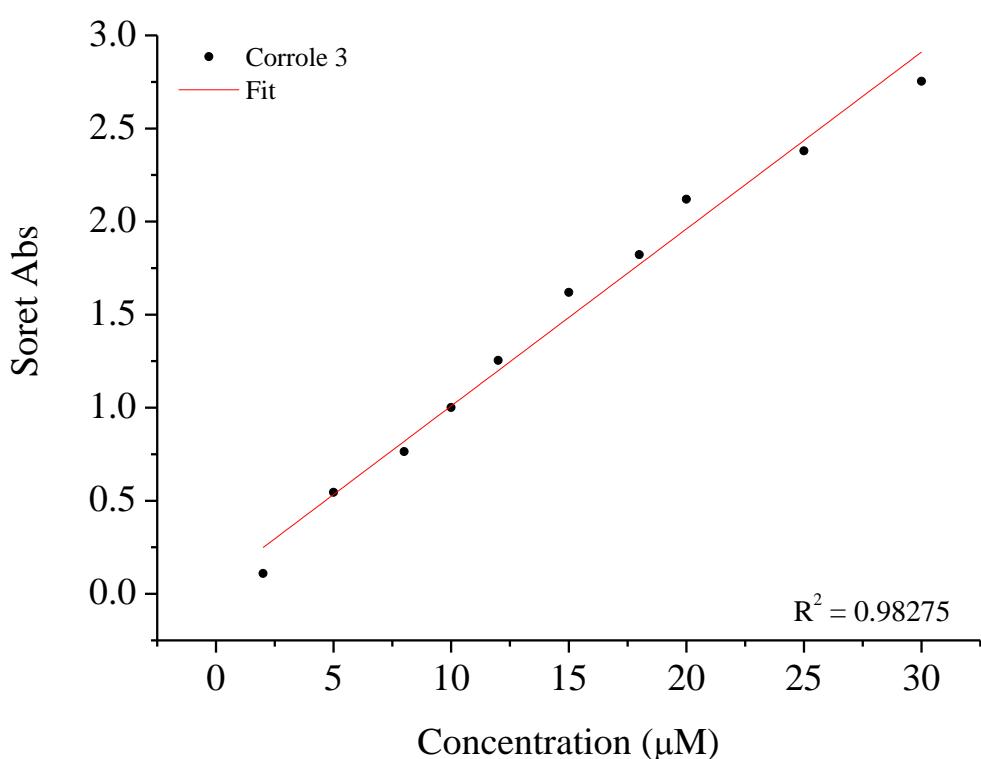
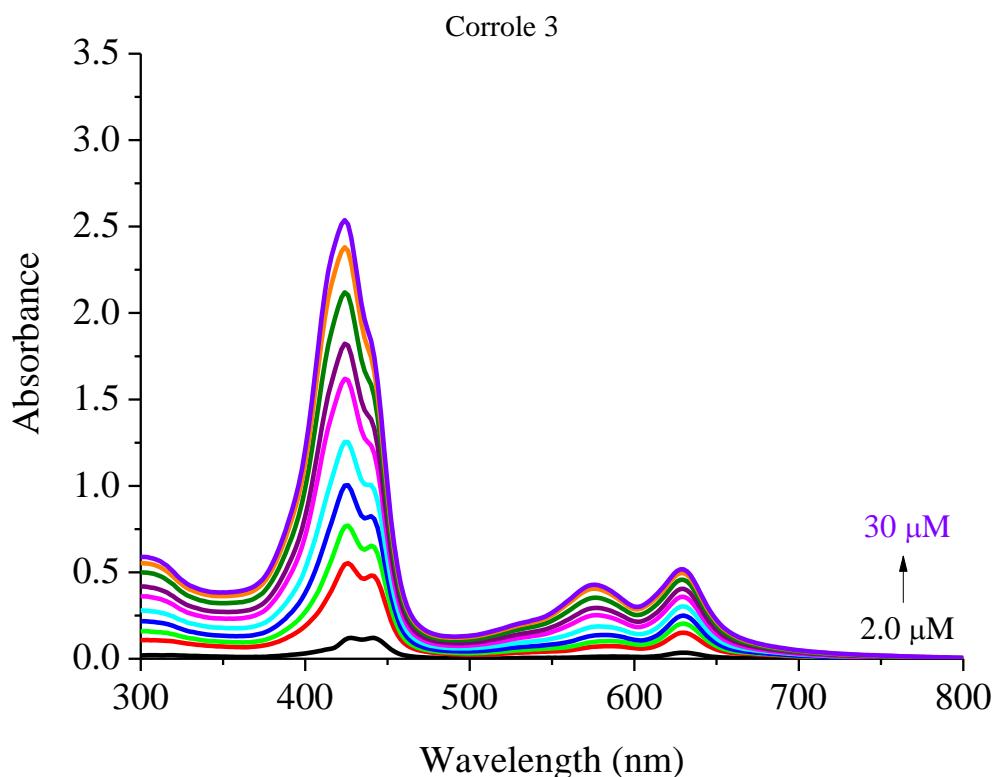


Figure S33. (up) Aggregation behavior of corrole **3** in DMSO solution and (down) $\text{Abs}_{\text{Soret}}$ versus concentration plot.

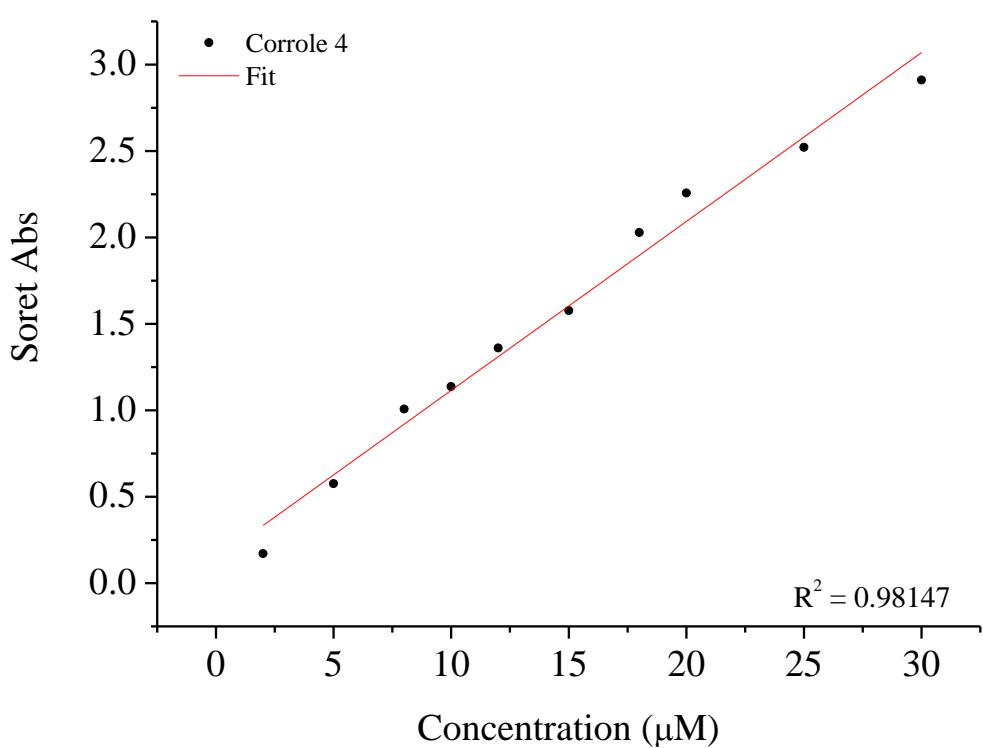
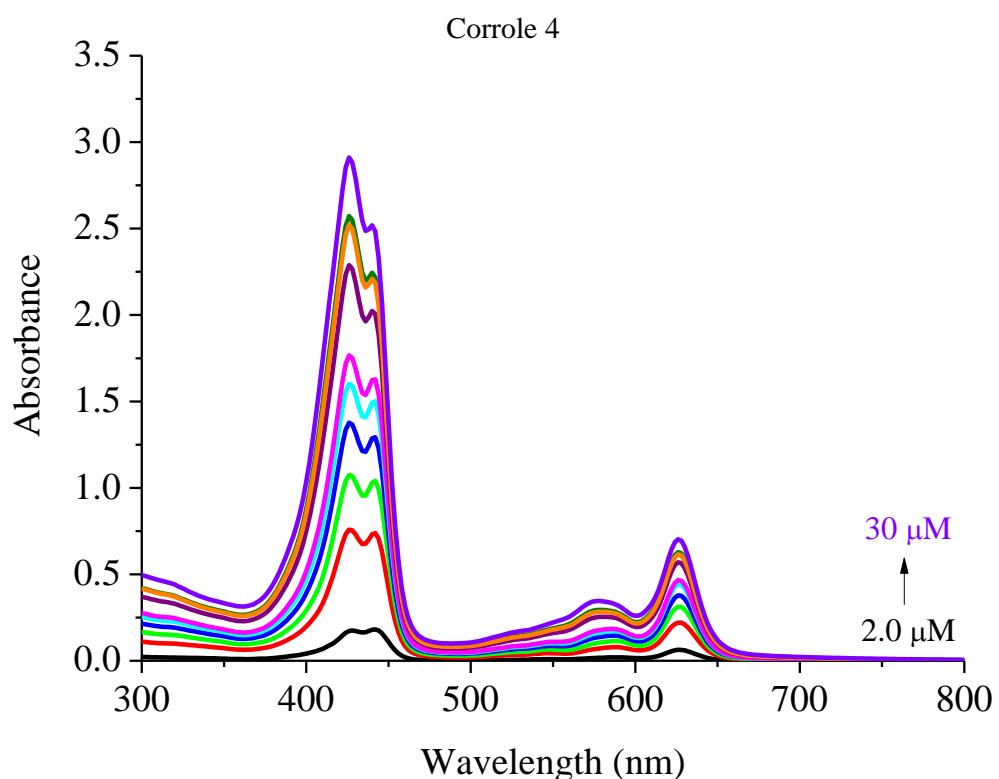


Figure S34. (up) Aggregation behavior of corrole **4** in DMSO solution and (down) $\text{Abs}_{\text{Soret}}$ versus concentration plot.

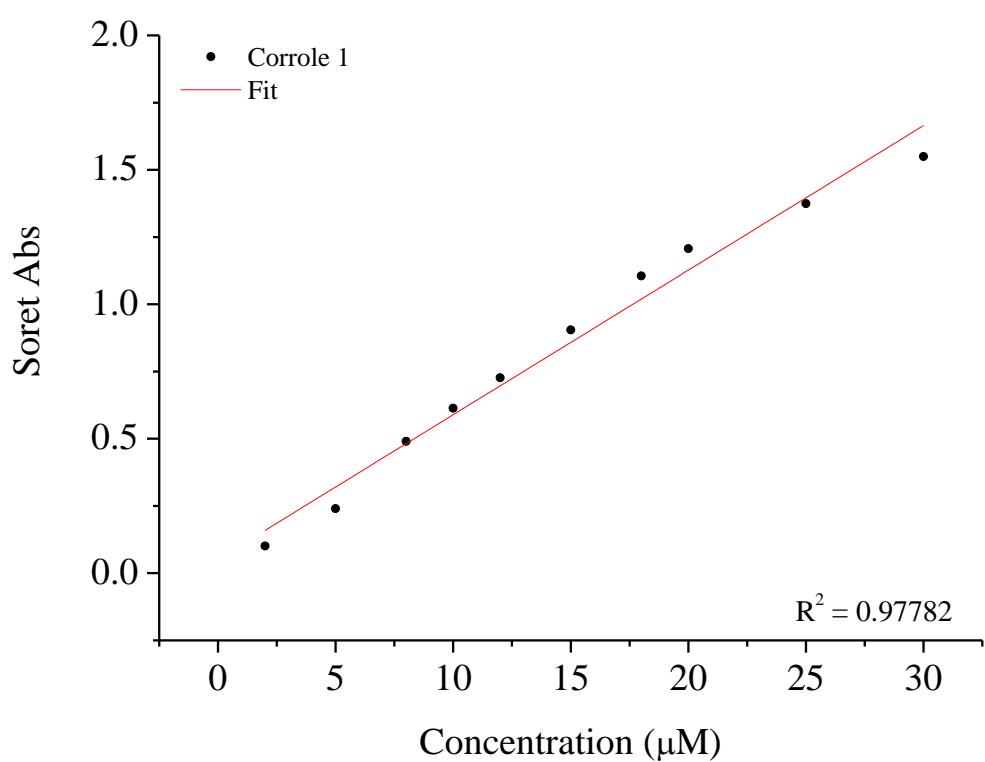
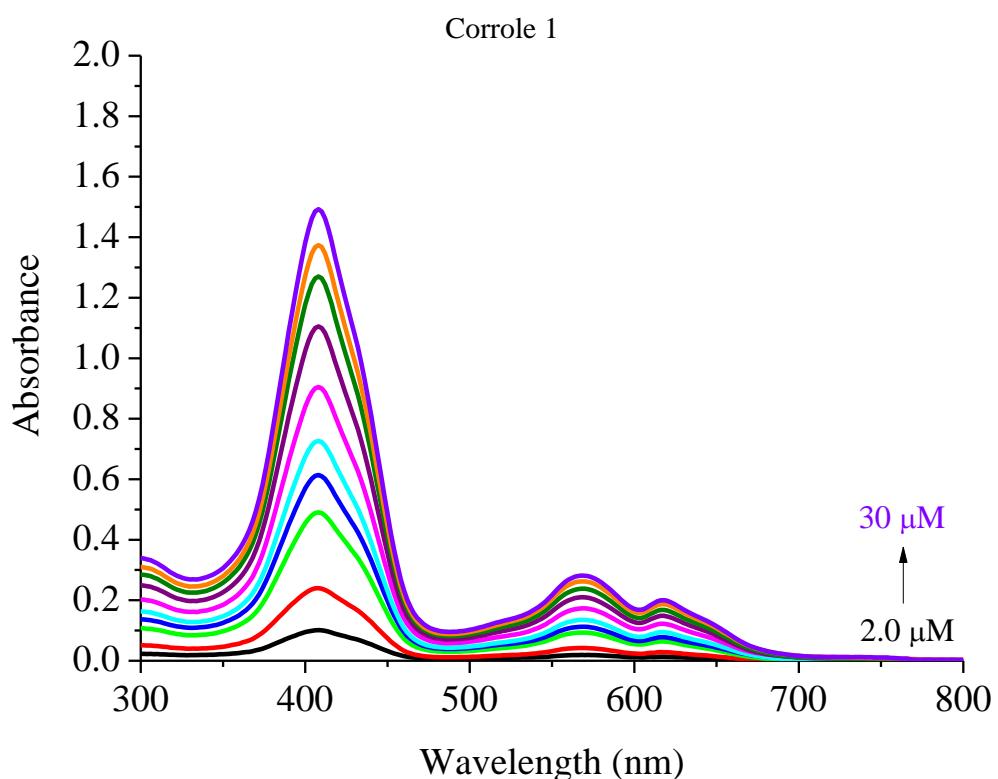


Figure S35. (up) Aggregation behavior of corrole **1** in DMSO(5%)/Tris-HCl pH 7.4 buffered mixture solution and (down) $\text{Abs}_{\text{Soret}}$ *versus* concentration plot.

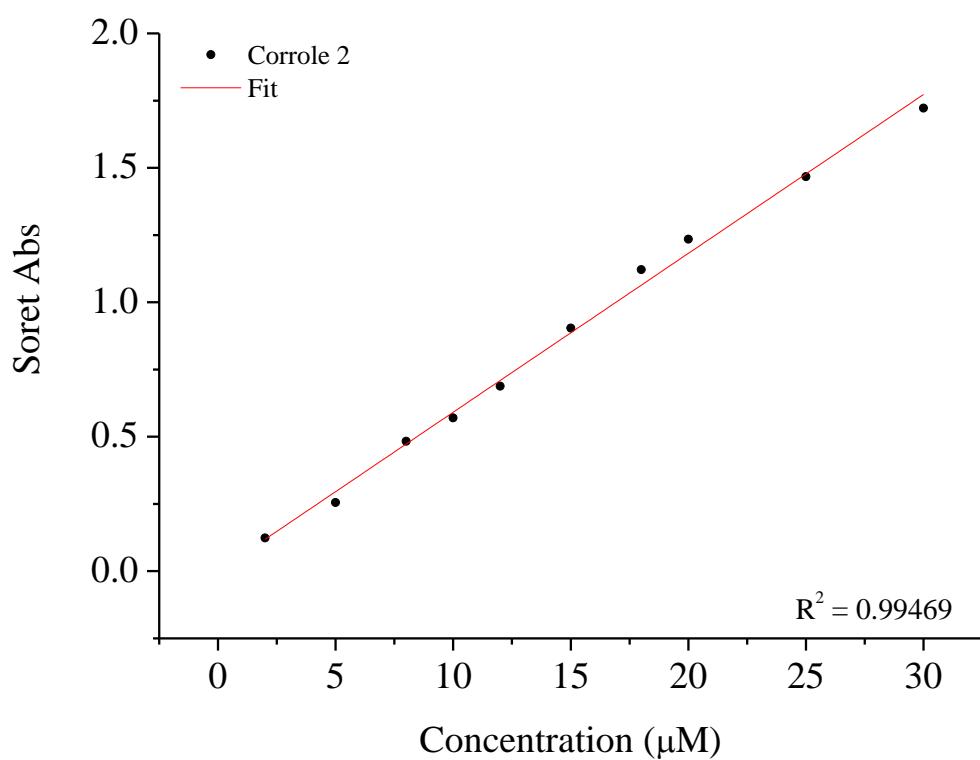
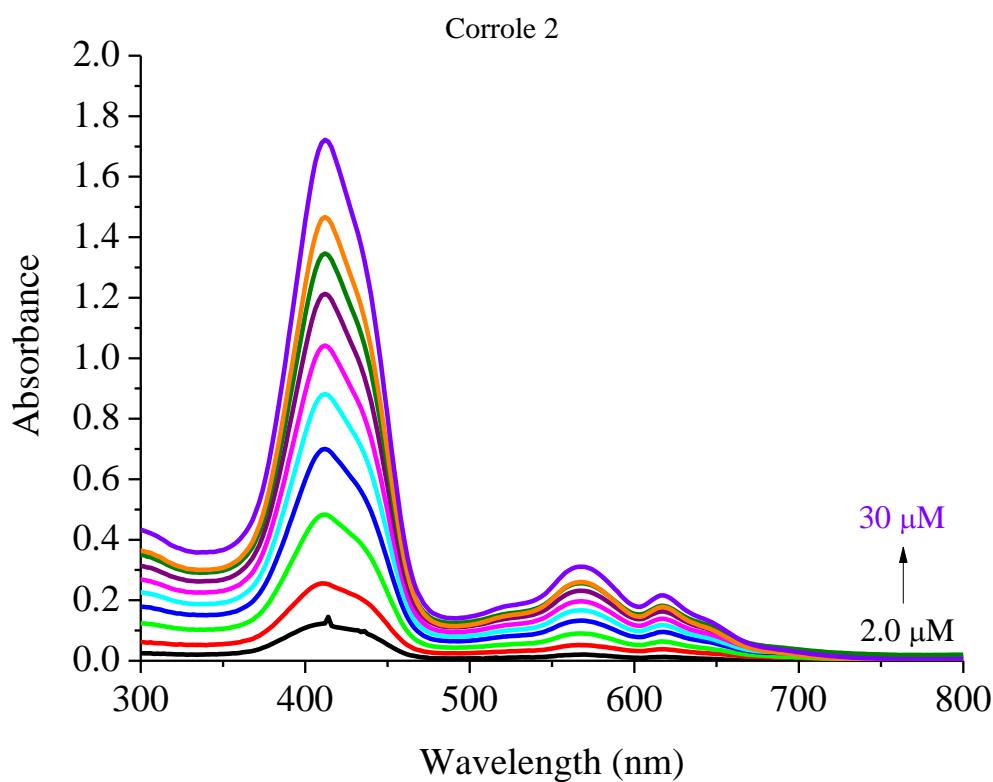


Figure S36. (up) Aggregation behavior of corrole **2** in DMSO(5%)/Tris-HCl pH 7.4 buffered mixture solution and (down) $\text{Abs}_{\text{Soret}}$ *versus* concentration plot.

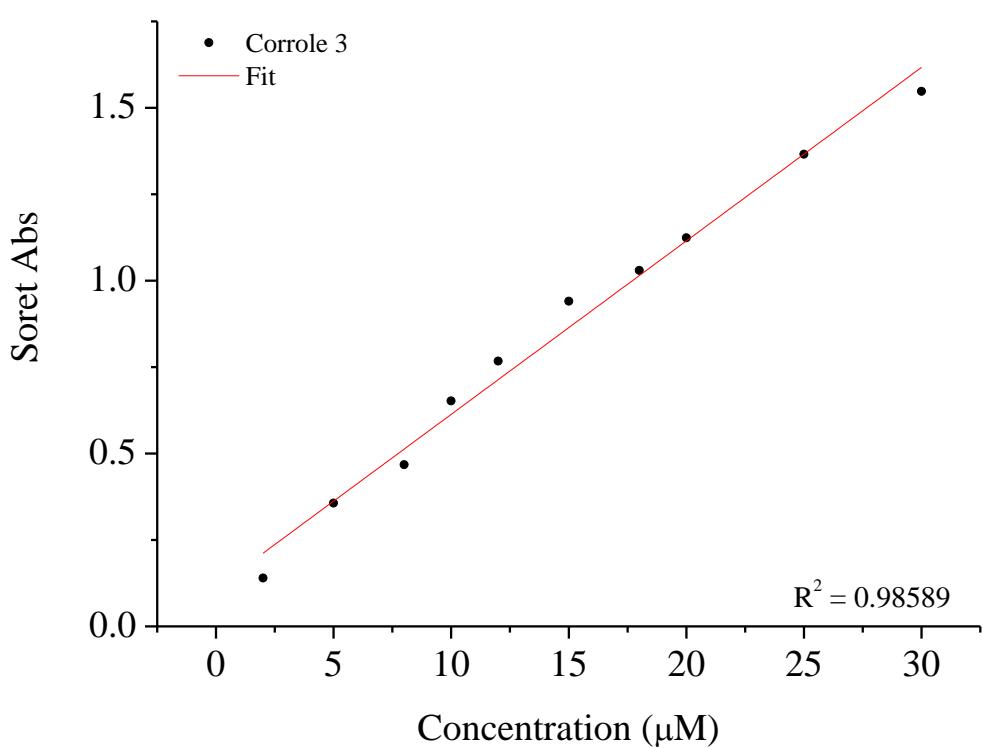
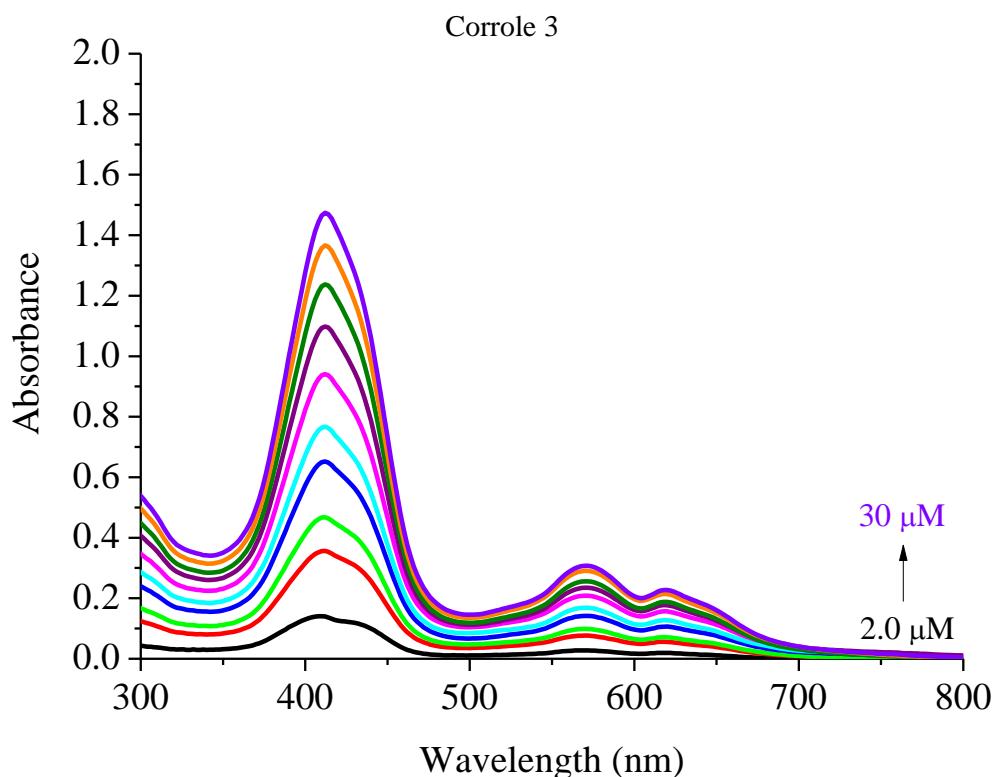


Figure S37. (up) Aggregation behavior of corrole **3** in DMSO(5%)/Tris-HCl pH 7.4 buffered mixture solution and (down) $\text{Abs}_{\text{Soret}}$ *versus* concentration plot.

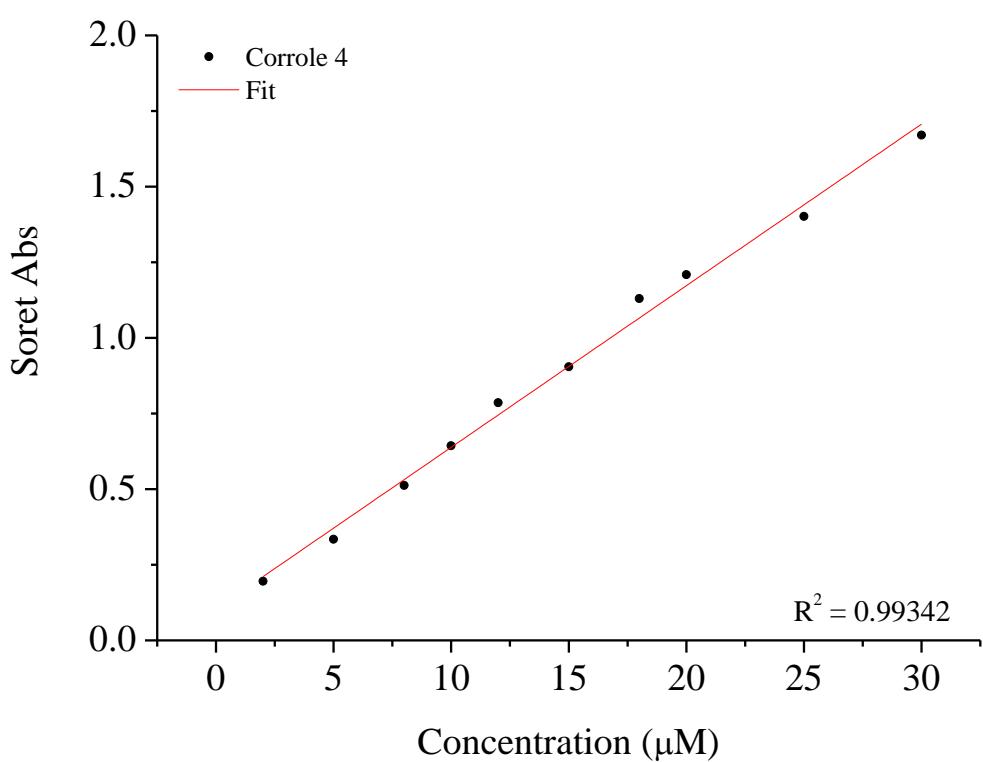
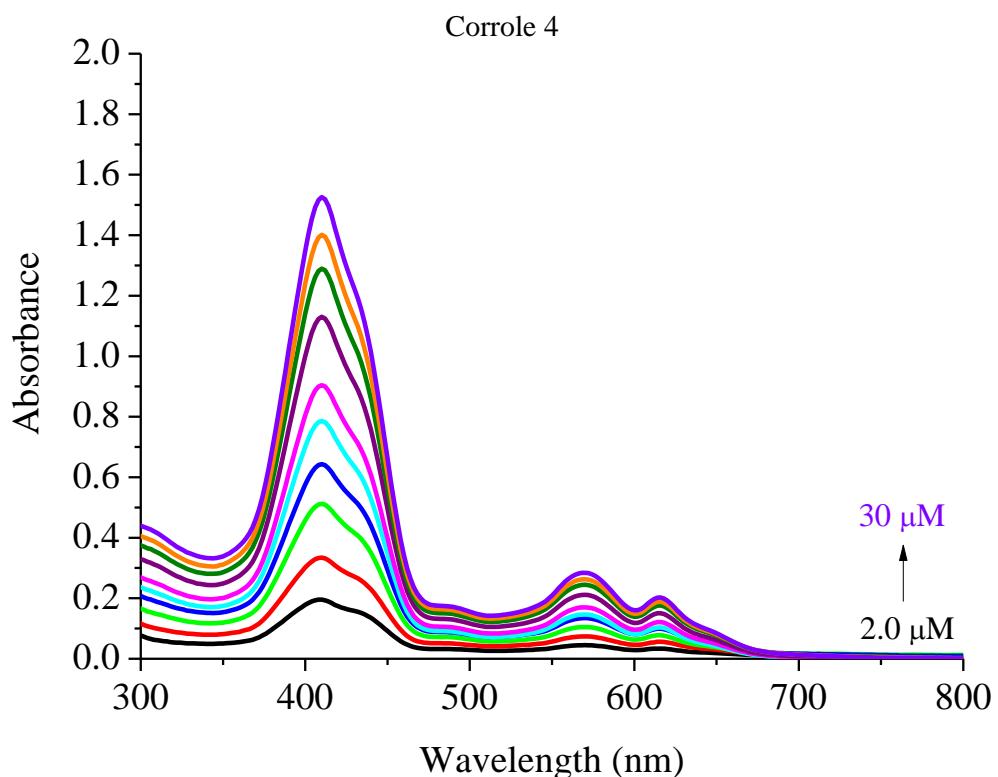


Figure S38. (up) Aggregation behavior of corrole **4** in DMSO(5%)/Tris-HCl pH 7.4 buffered mixture solution and (down) $\text{Abs}_{\text{Soret}}$ *versus* concentration plot.

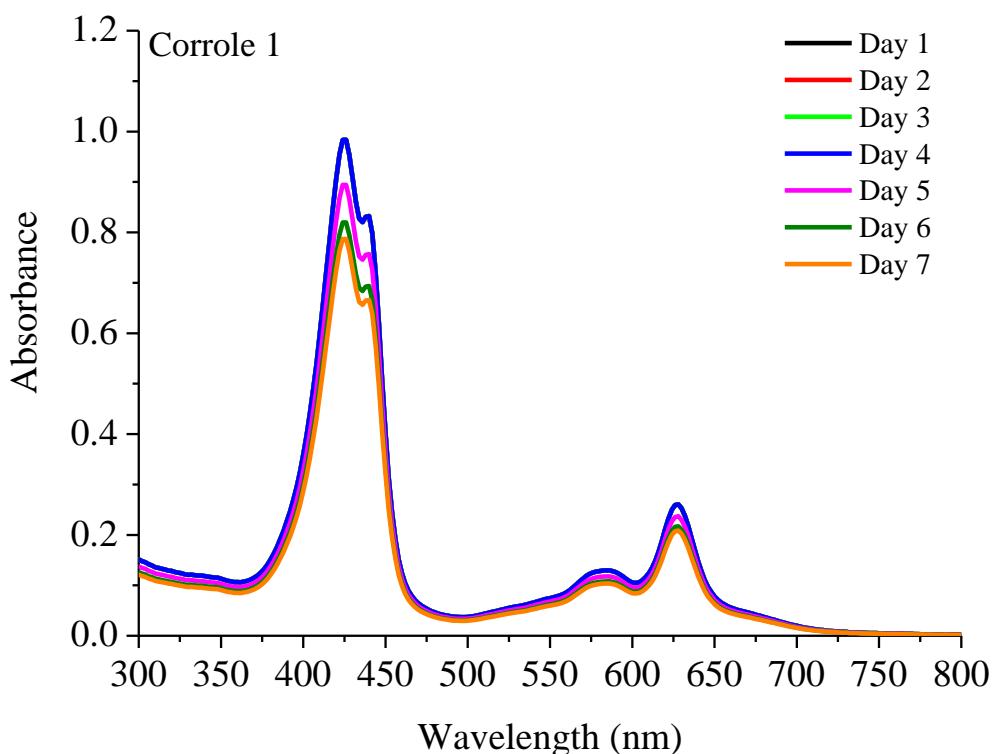


Figure S39. Stability assay of corrole **1** in DMSO solution.

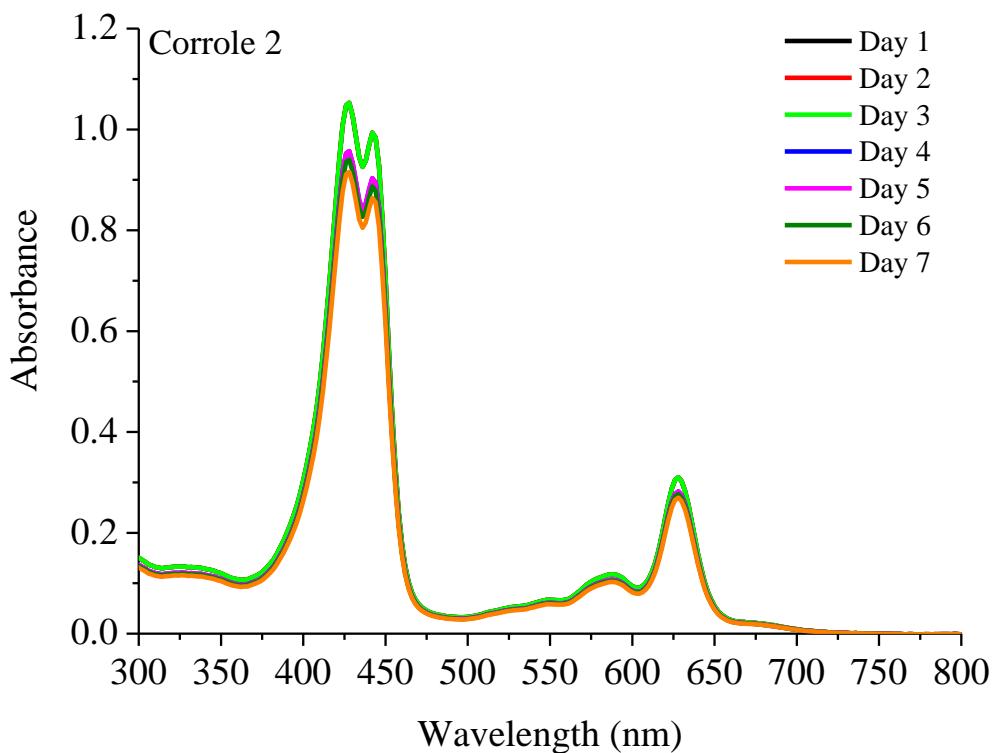


Figure S40. Stability assay of corrole **2** in DMSO solution.

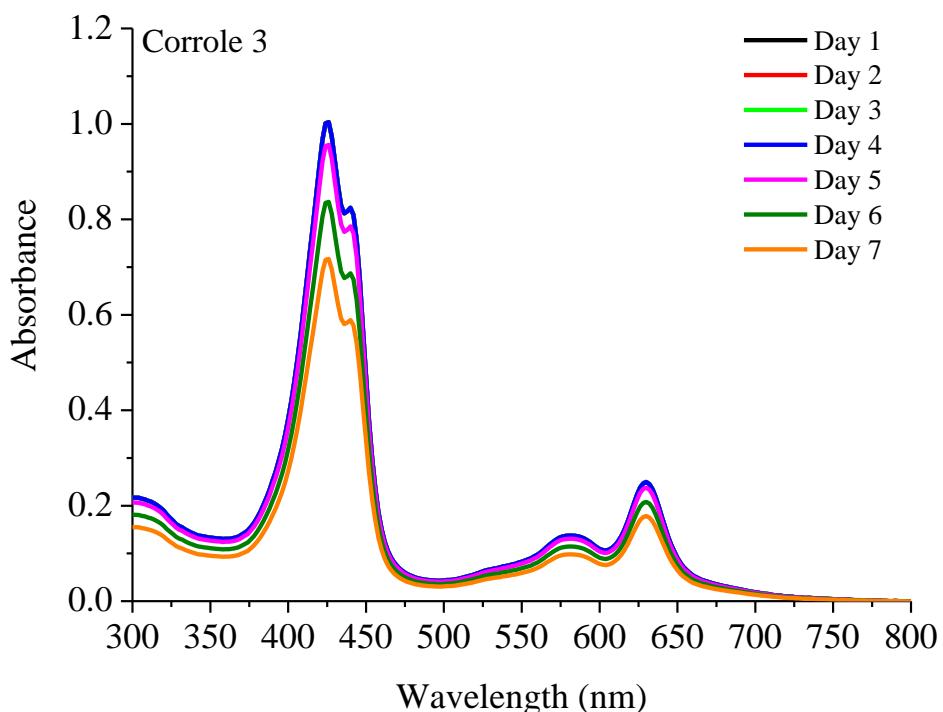


Figure S41. Stability assay of corrole **3** in DMSO solution.

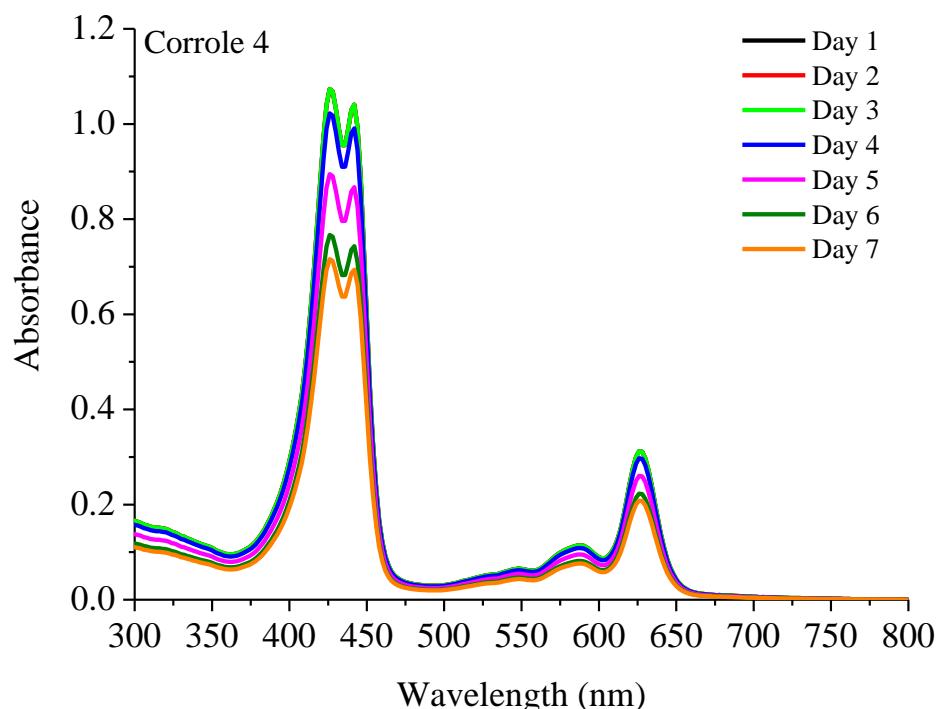


Figure S42. Stability assay of corrole **4** in DMSO solution.

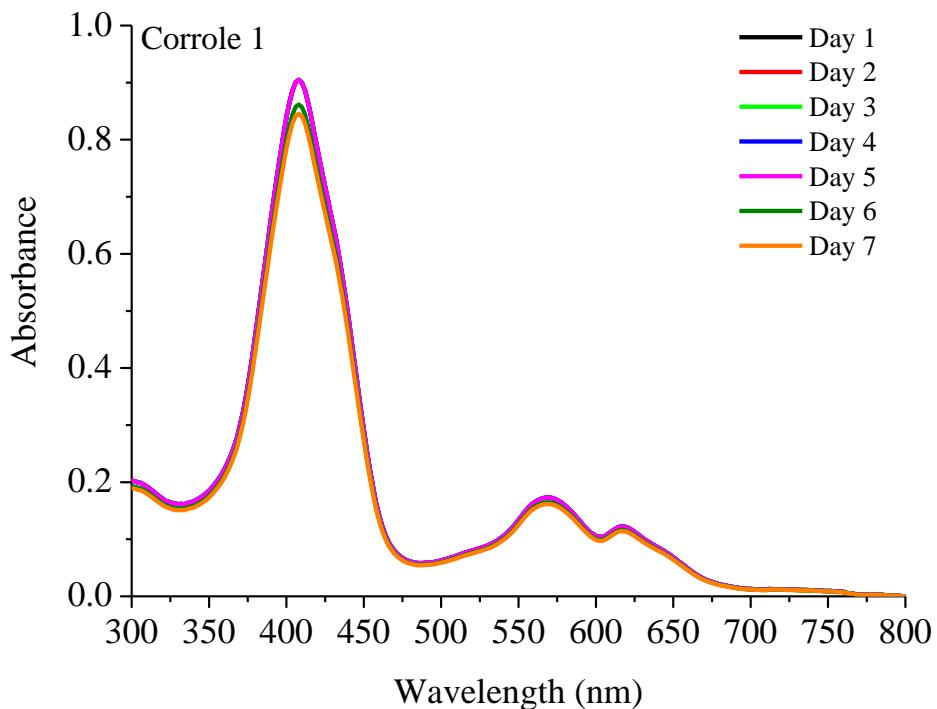


Figure S43. Stability assay of corrole **1** in DMSO(5%)/Tris-HCl pH 7.4 buffered mixture solution.

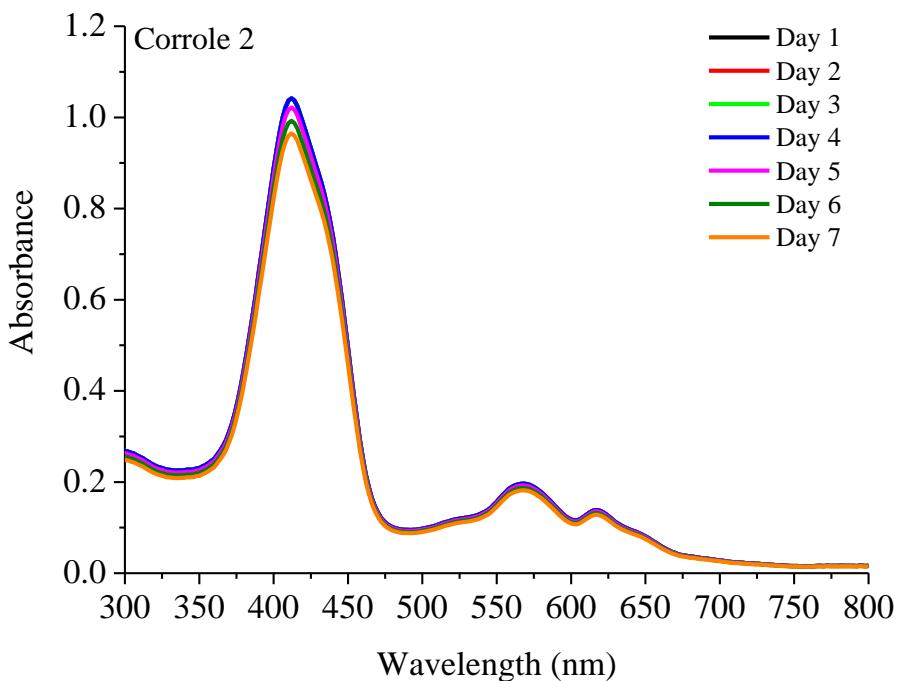


Figure S44. Stability assay of corrole **2** in DMSO(5%)/Tris-HCl pH 7.4 buffered mixture solution.

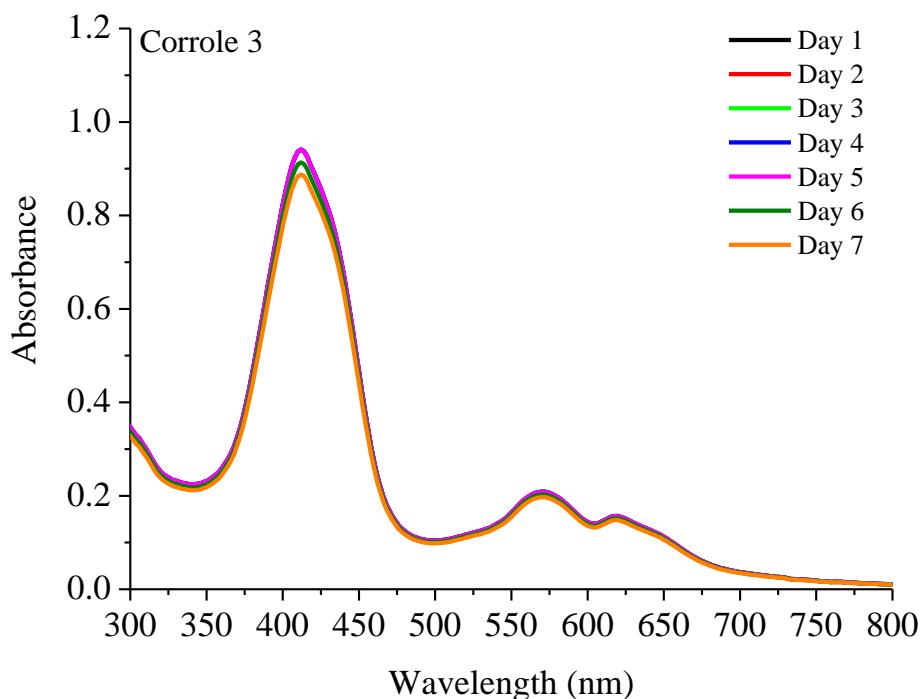


Figure S45. Stability assay of corrole **3** in DMSO(5%)/Tris-HCl pH 7.4 buffered mixture solution.

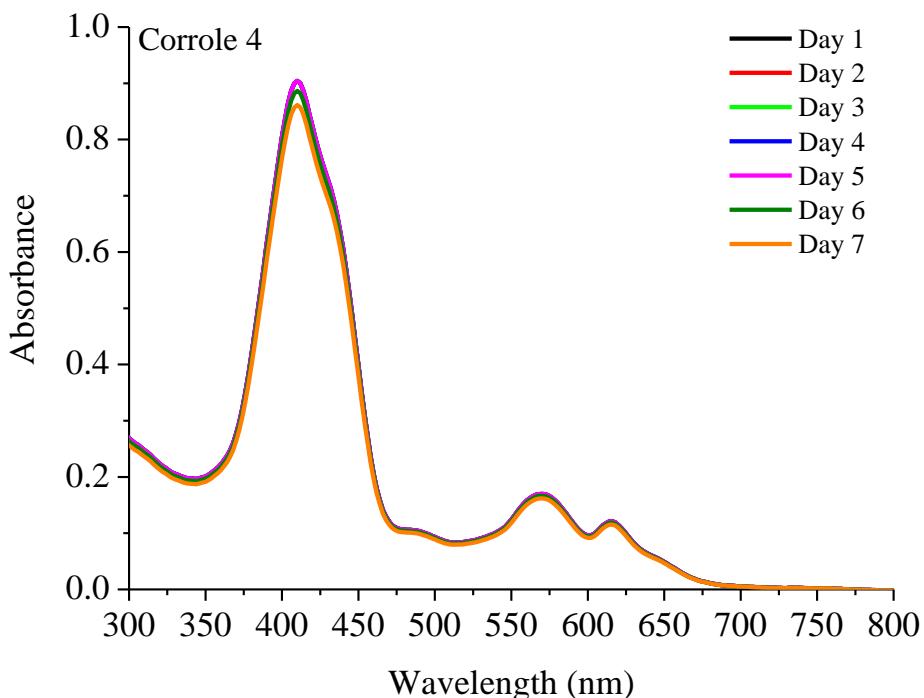


Figure S46. Stability assay of corrole **4** in DMSO(5%)/Tris-HCl pH 7.4 buffered mixture solution.

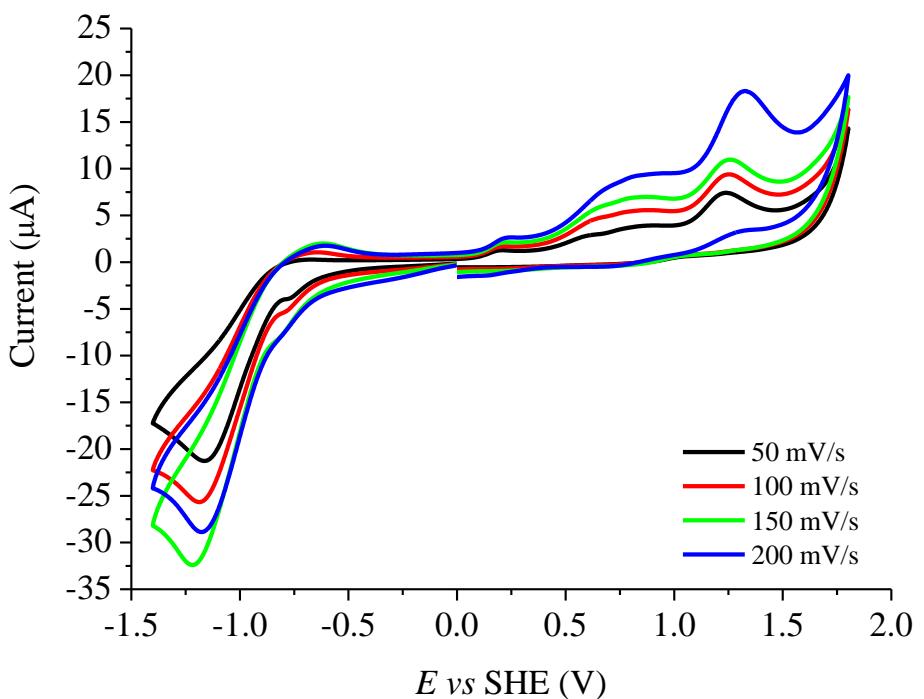


Figure S47. Cyclic voltammetry analysis of corrole **1** in dry DCM solution, using 0.1 M TBAPF₆ as support electrolyte.

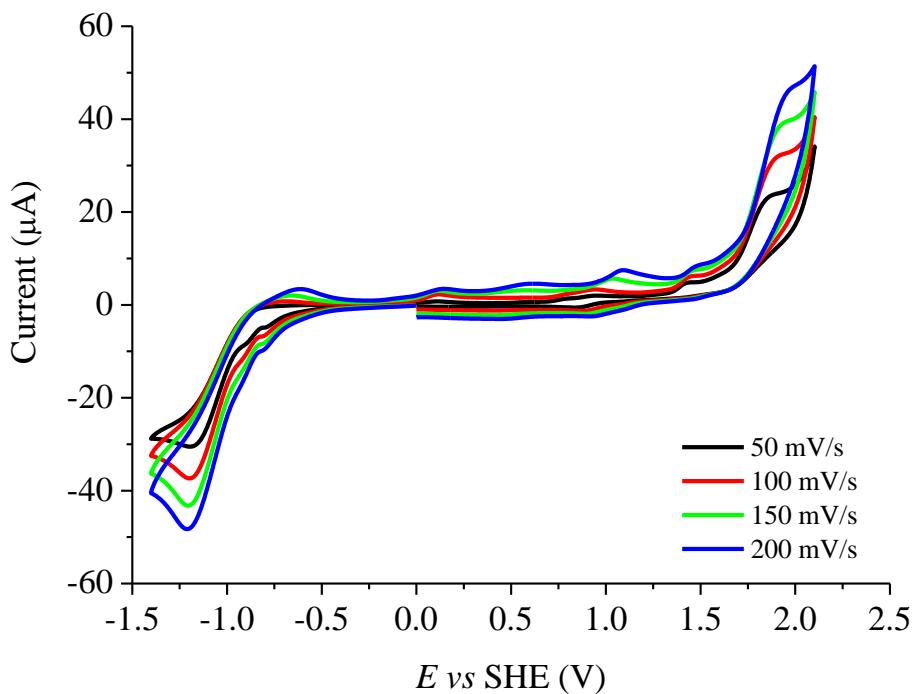


Figure S48. Cyclic voltammetry analysis of corrole **2** in dry DCM solution, using 0.1 M TBAPF₆ as support electrolyte.

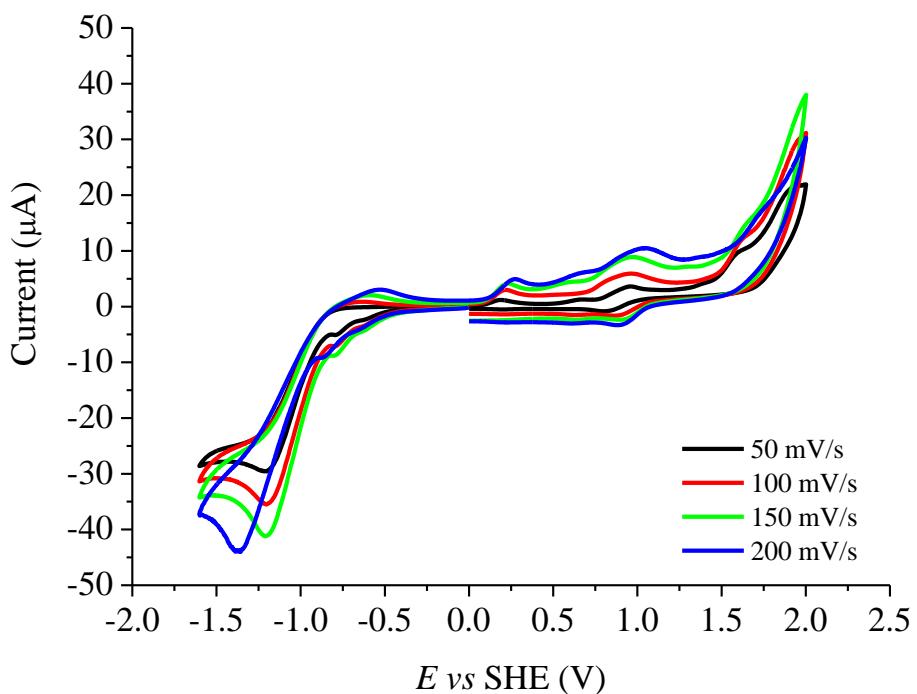


Figure S49. Cyclic voltammetry analysis of corrole **3** in dry DCM solution, using 0.1 M TBAPF₆ as support electrolyte.

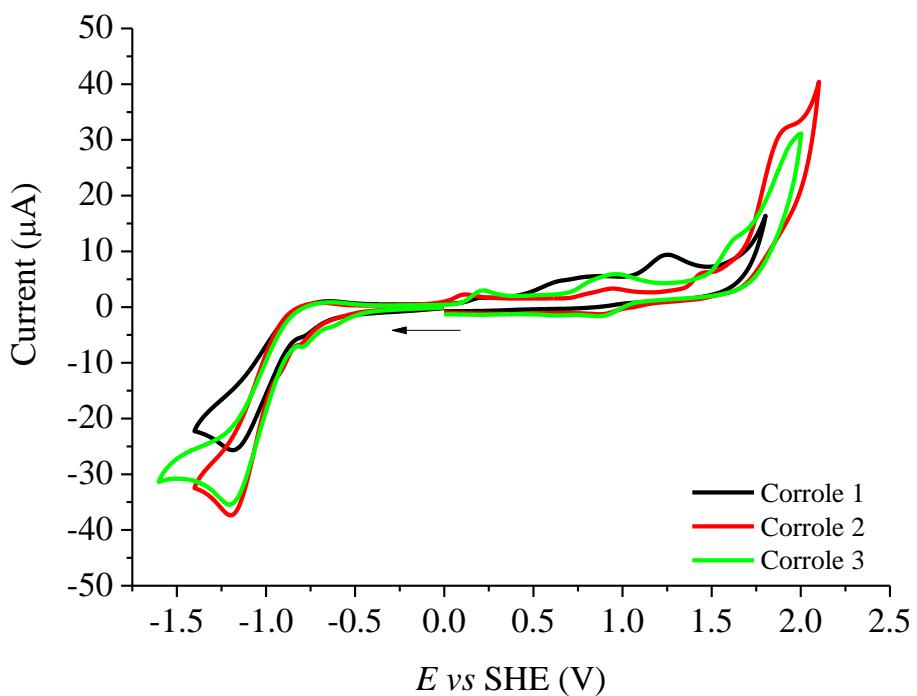


Figure S50. Cyclic voltammetry analysis of corrole **4** in dry DCM solution, using 0.1 M TBAPF₆ as support electrolyte.

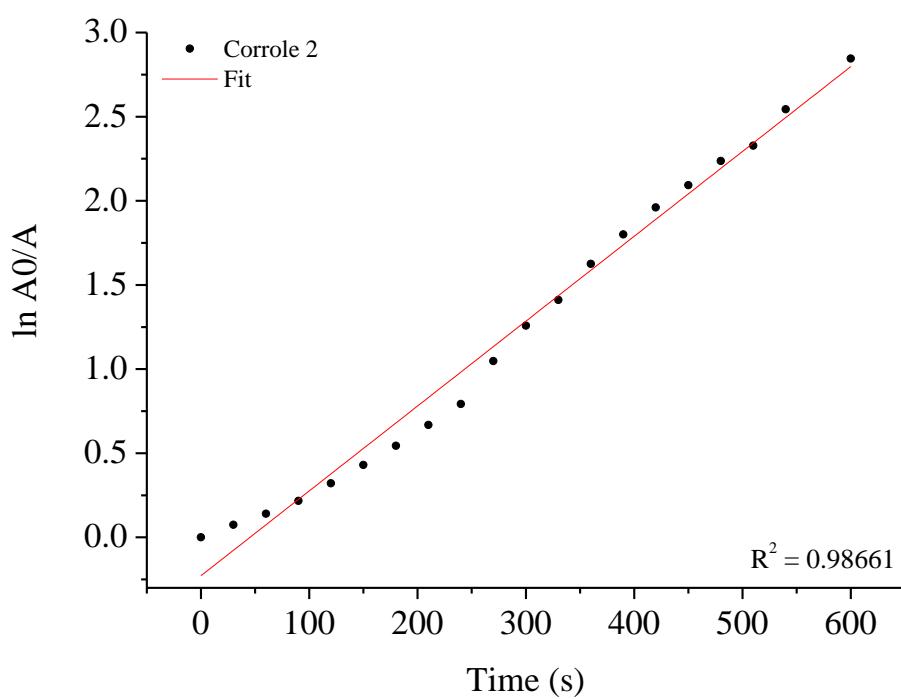
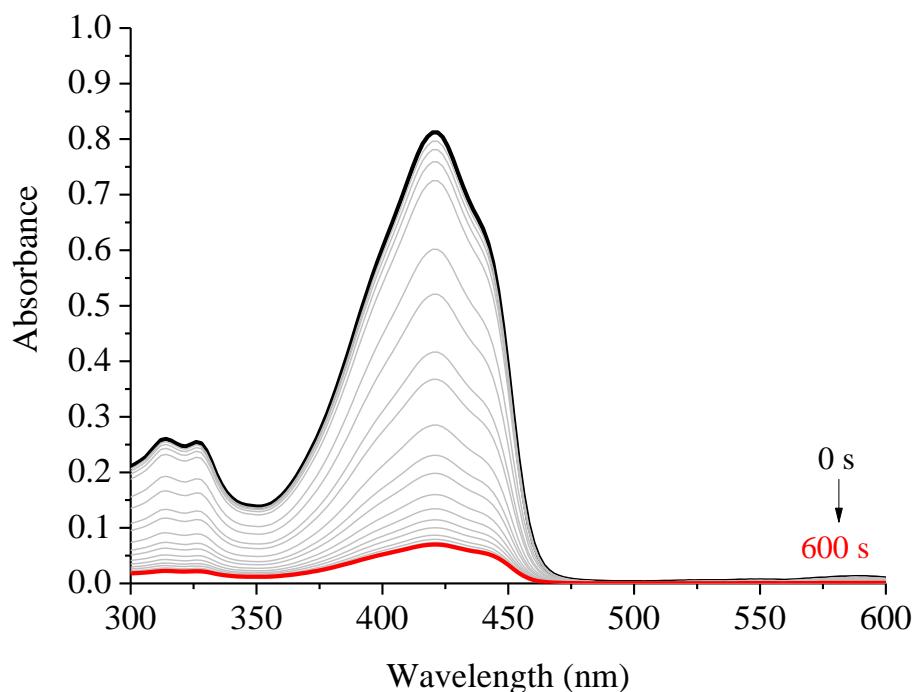


Figure S51. (up) DPBF photo-oxidation assay of corrole **2** in DMSO solution, by red-light irradiation source at 600 s and (down) $\ln A_0/A$ versus time plot.

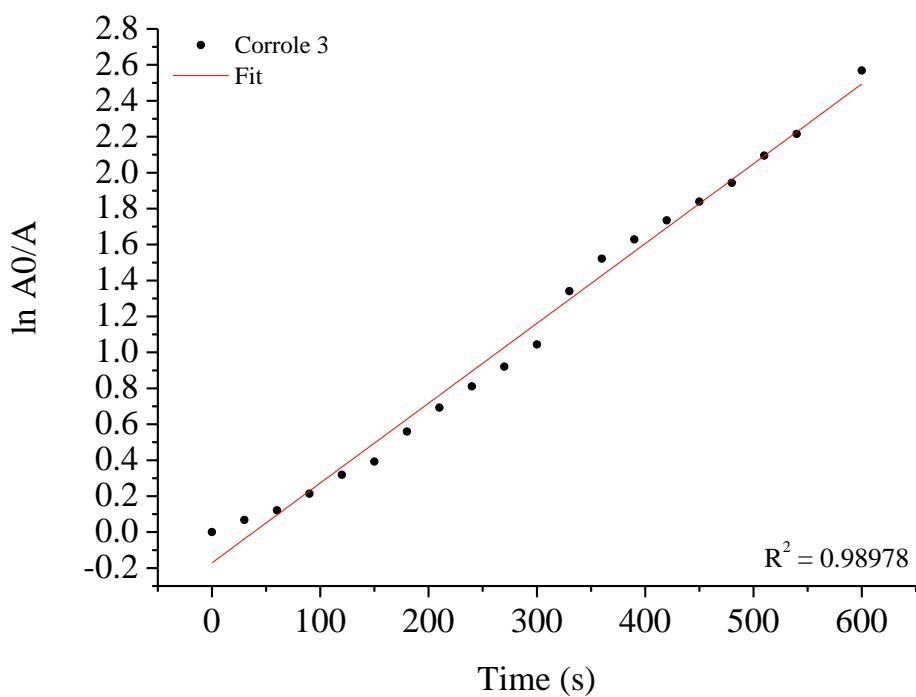
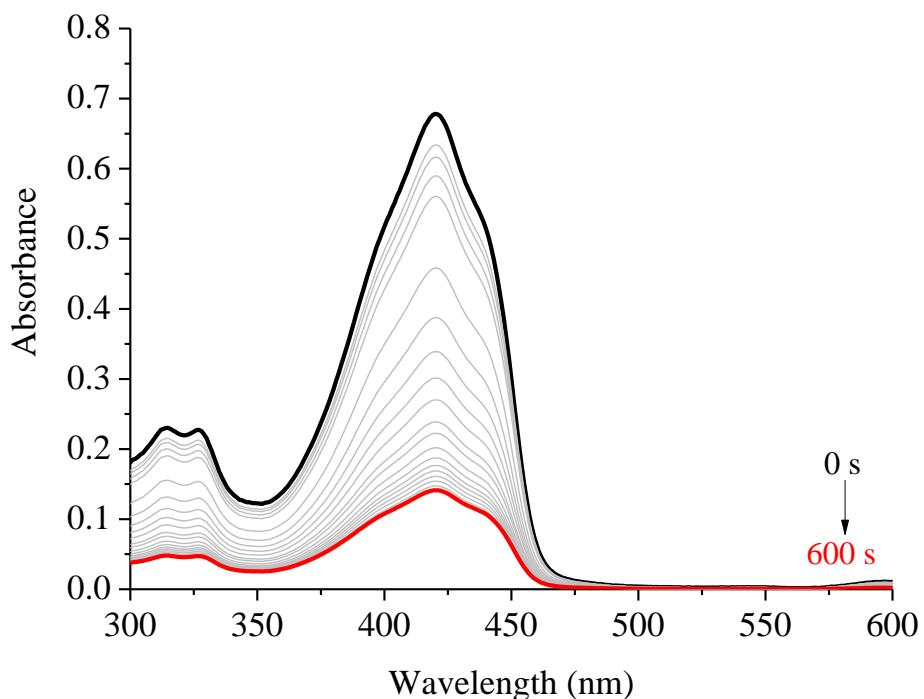


Figure S52. (up) DPBF photo-oxidation assay of corrole **3** in DMSO solution, by red-light irradiation source at 600 s and (down) $\ln A_0/A$ versus time plot.

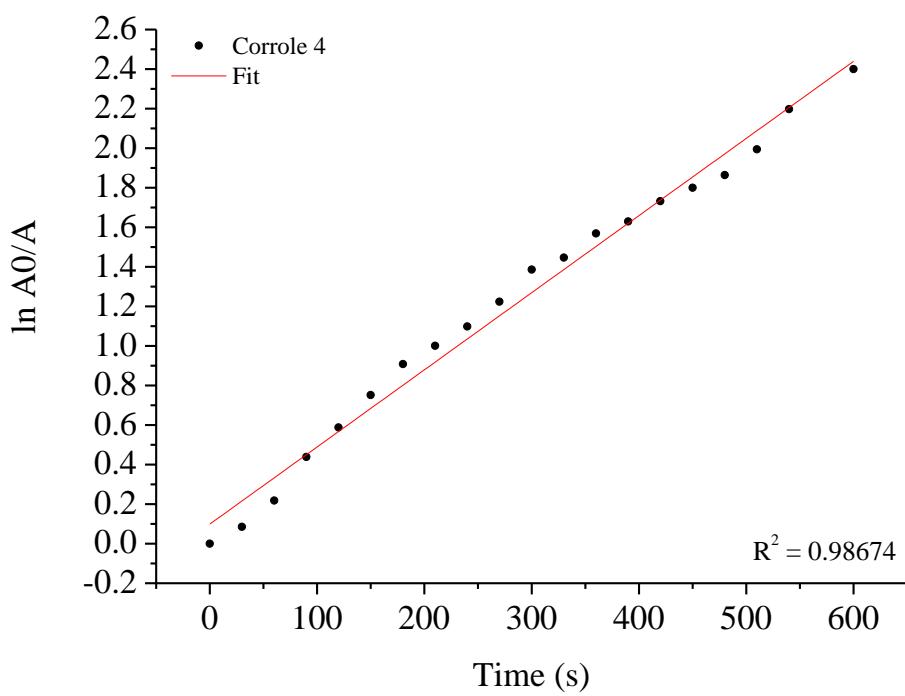
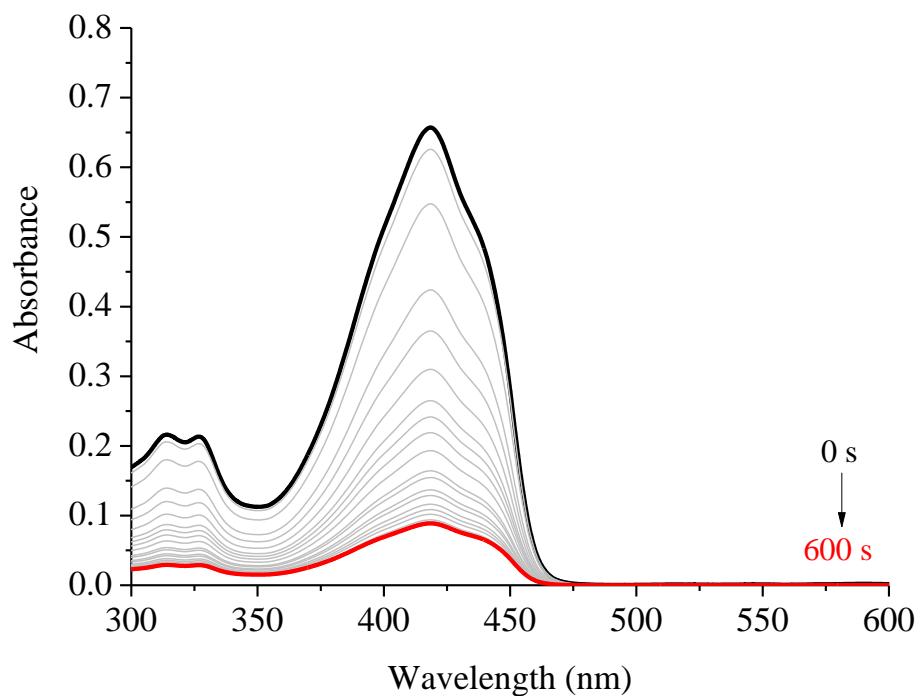


Figure S53. (up) DPBF photo-oxidation assay of corrole **4** in DMSO solution, by red-light irradiation source at 600 s and (down) $\ln A_0/A$ versus time plot.

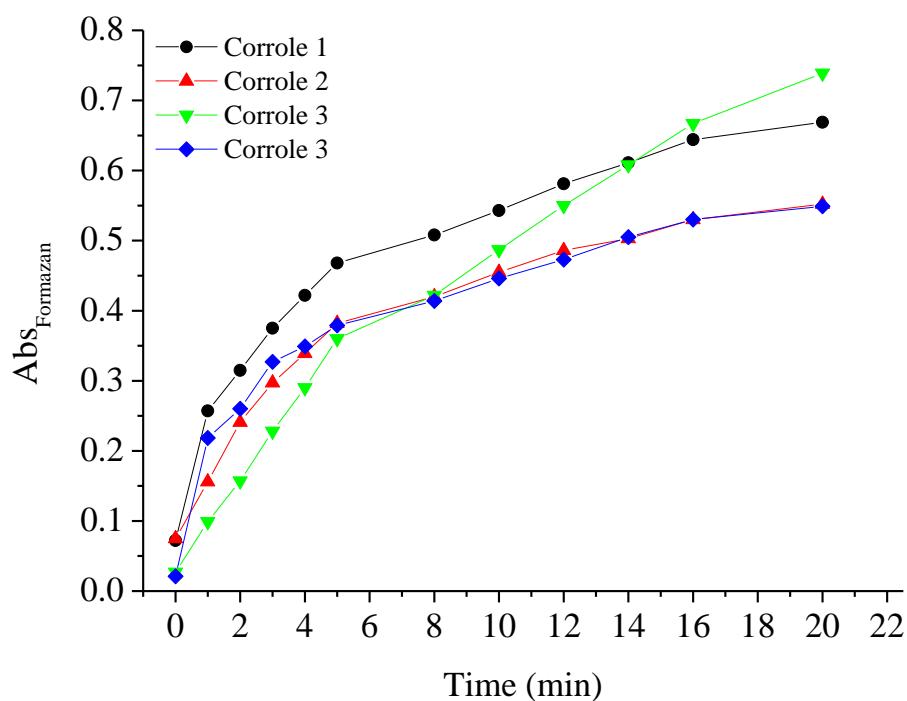
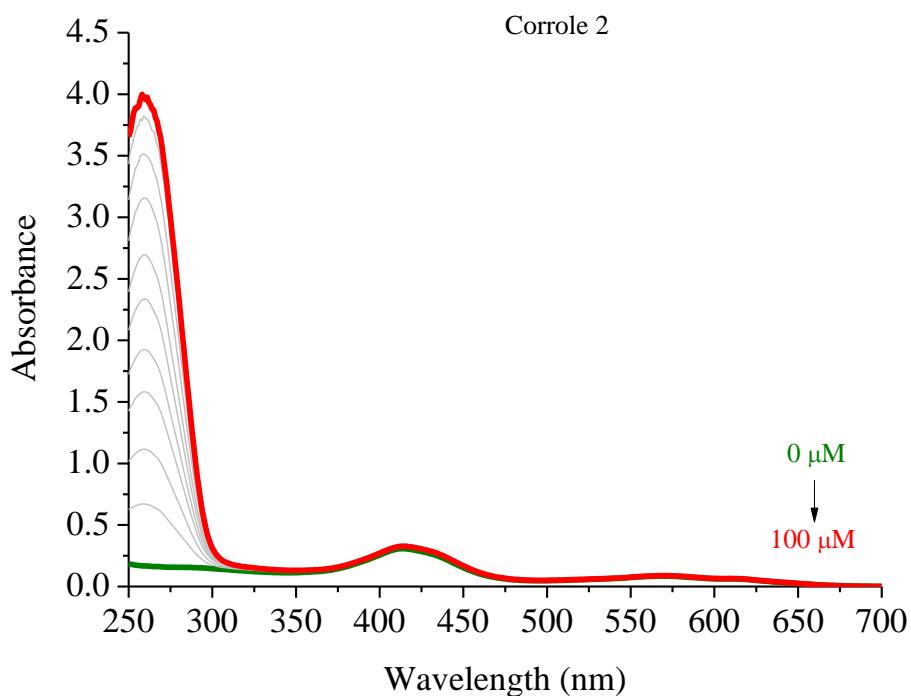


Figure S54. NBT reduction assay of corroles **1-4** in DMSO solution, by white-light irradiation source at 20 min.



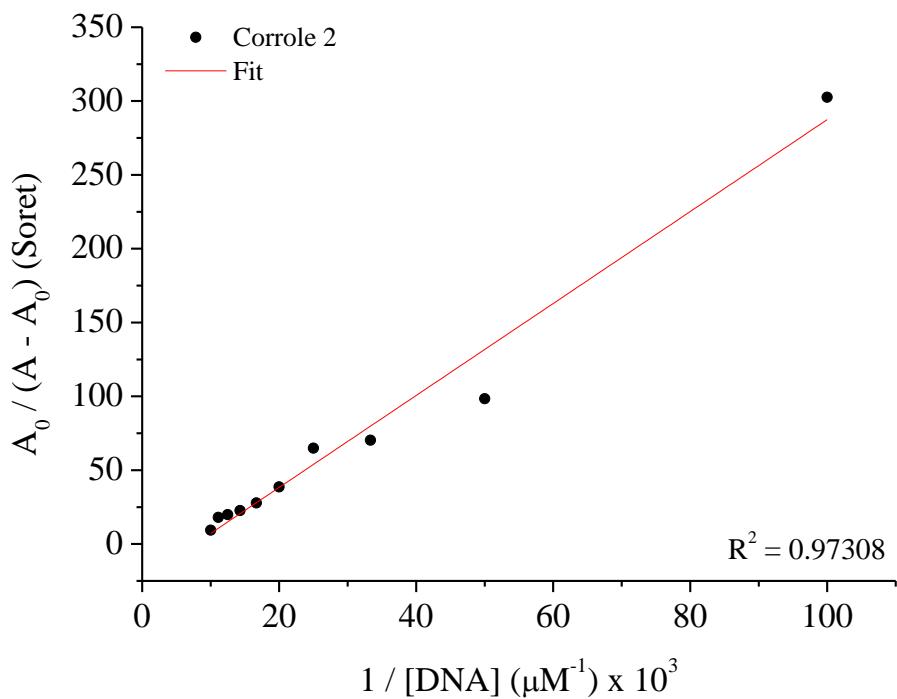
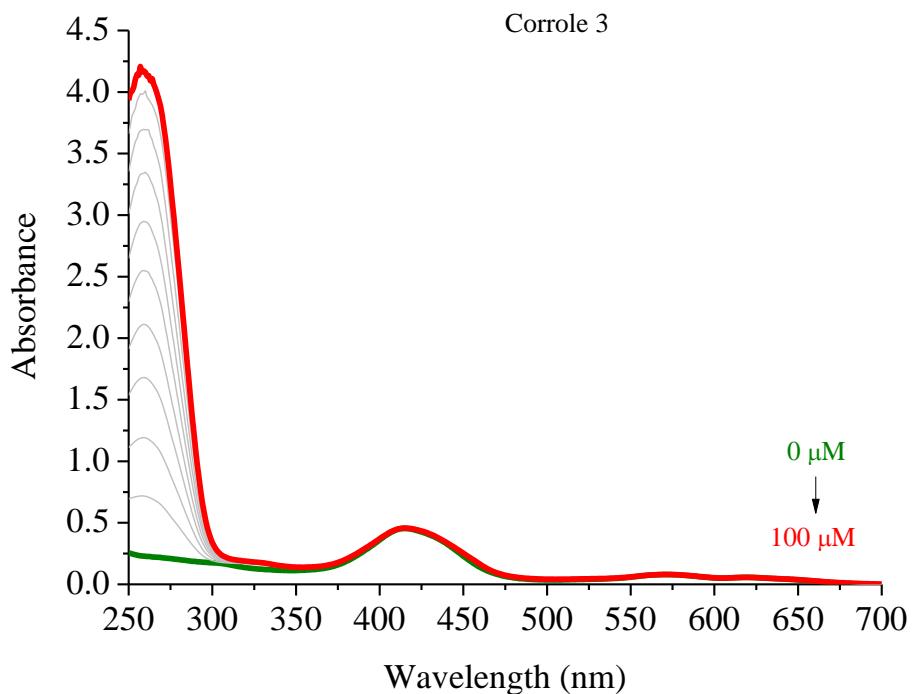


Figure S55. (up) UV-Vis spectra of the corrole **2** upon successive additions of CT-DNA concentrations (0 to 100 μM) in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution. (down) Benesi-Hidelbrandt plots of $A_0 / (A - A_0)$ versus $1 / [\text{CT-DNA}]$.



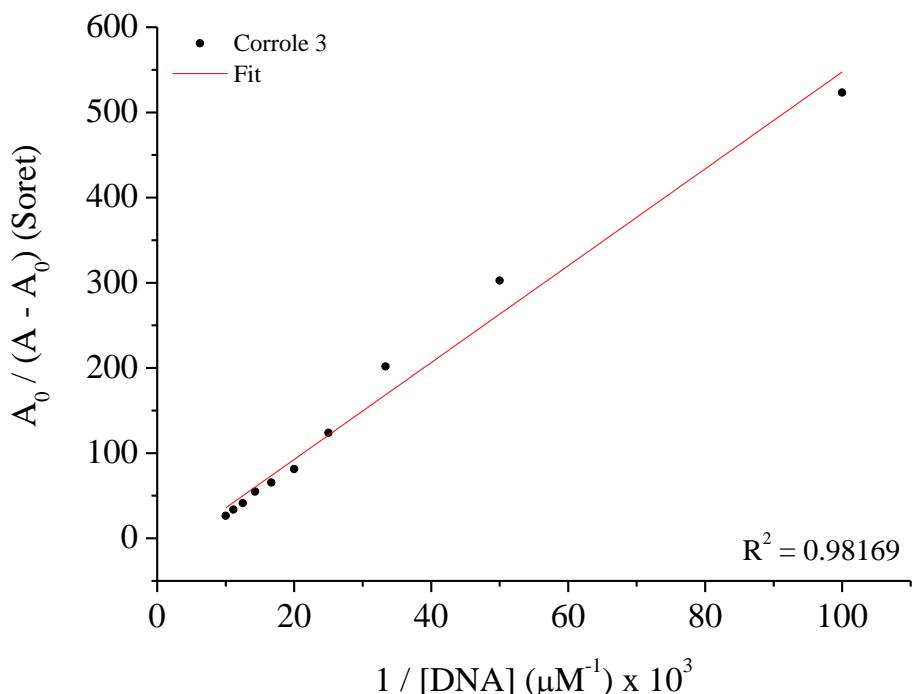
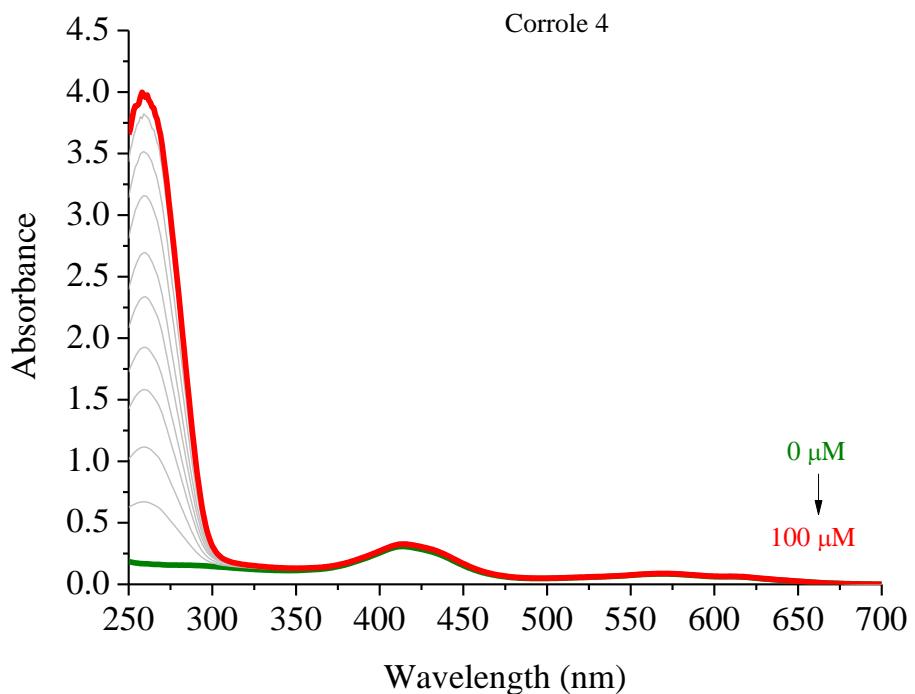


Figure S56. (up) UV-Vis spectra of the corrole **3** upon successive additions of CT-DNA concentrations (0 to 100 μM) in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution. (down) Benesi-Hidelbrandt plots of $A_0 / (A - A_0)$ versus $1 / [\text{CT-DNA}]$.



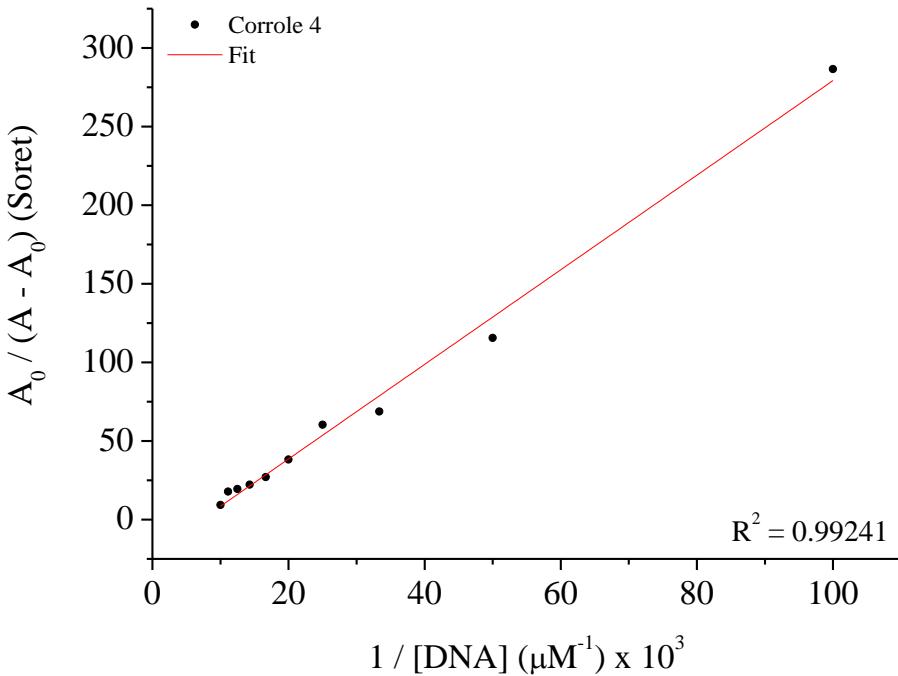


Figure S57. (up) UV-Vis spectra of the corrole **4** upon successive additions of CT-DNA concentrations (0 to 100 μ M) in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution. (down) Benesi-Hidelbrandt plots of $A_0 / (A - A_0)$ versus $1 / [\text{CT-DNA}]$.

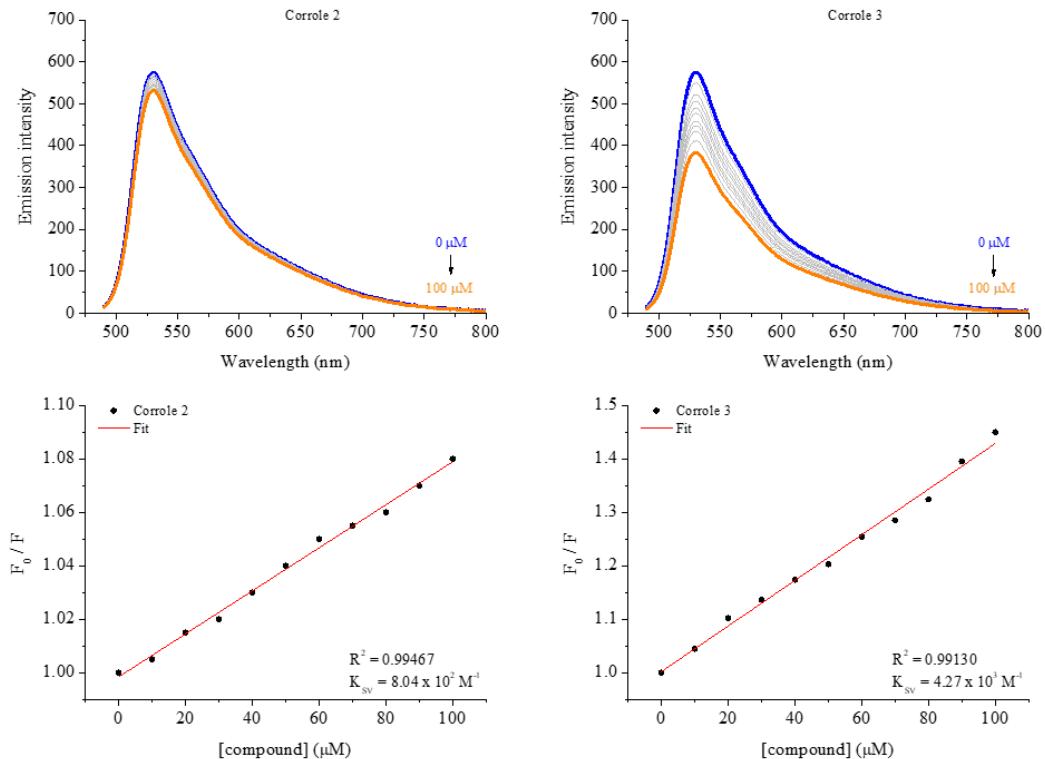


Figure S58. Steady-state fluorescence emission spectra for AO:DNA without and in the presence of corroles **2** and **3**, in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution. Graphs shows the plot F_0 / F versus [corrole]. [corrole] = 0–100 μ M.

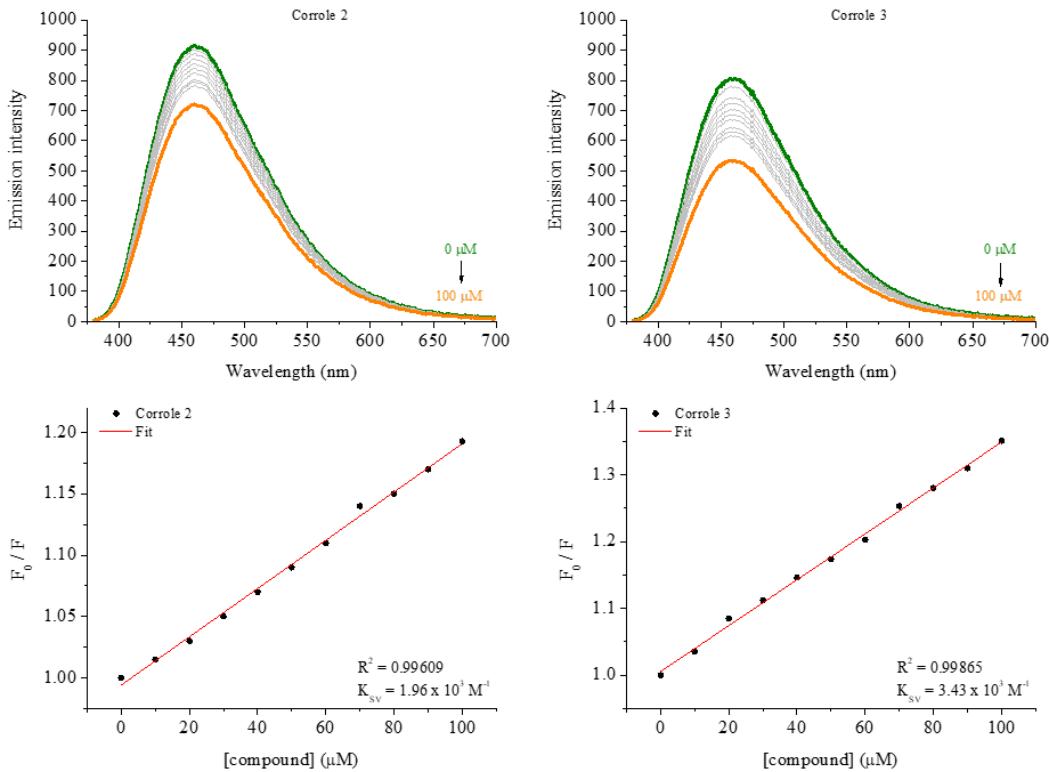


Figure S59. Steady-state fluorescence emission spectra for DAPI:DNA without and in the presence of corroles **2** and **3**, in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution. Graphs shows the plot F_0/F versus [corrole]. [corrole] = 0–100 μM.

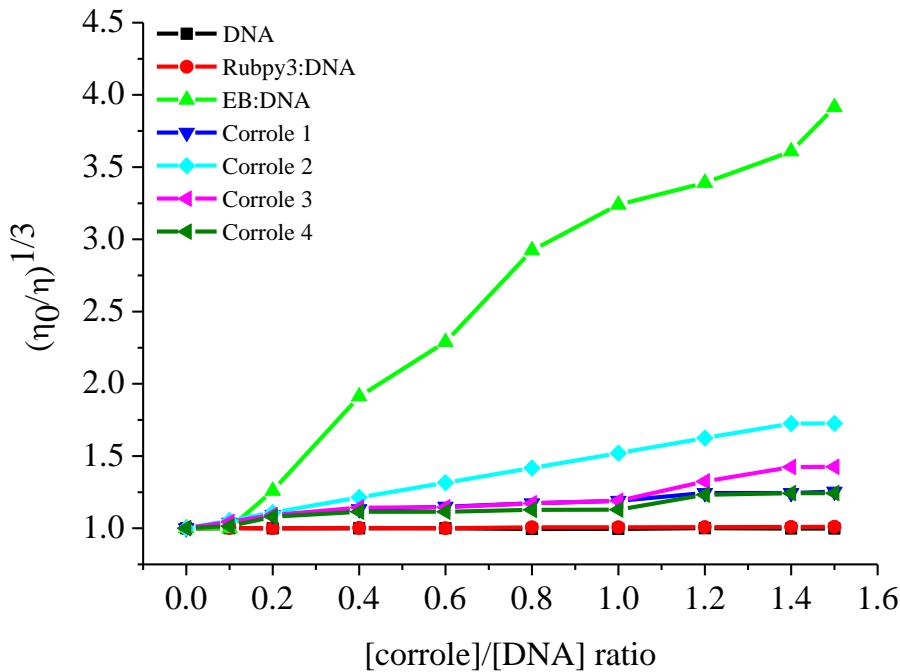


Figure S60. Viscosimetry assay of CT-DNA without and in the presence of corroles **1–4**, in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution.

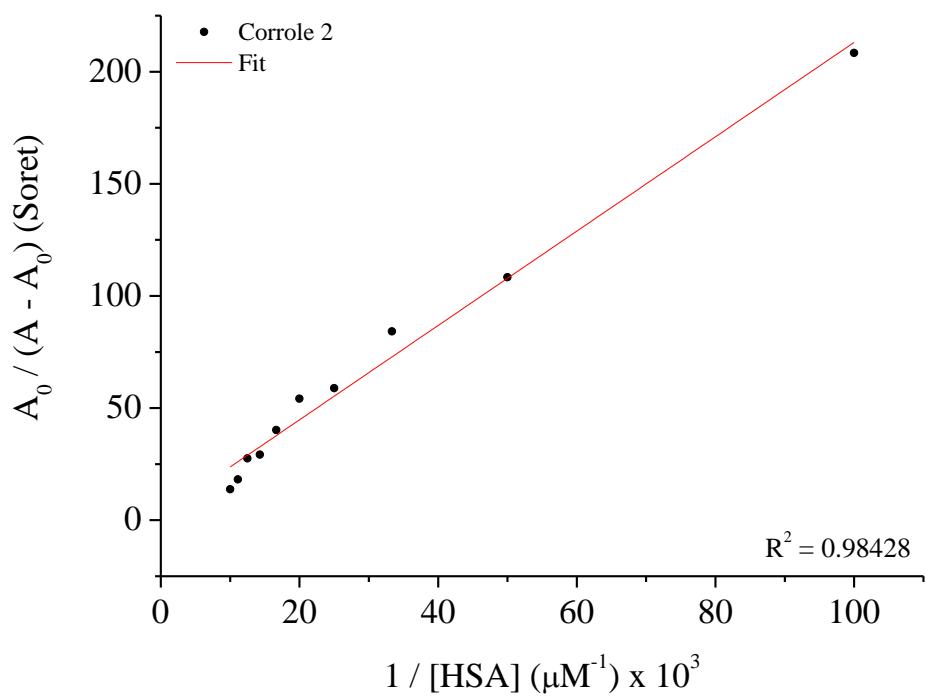
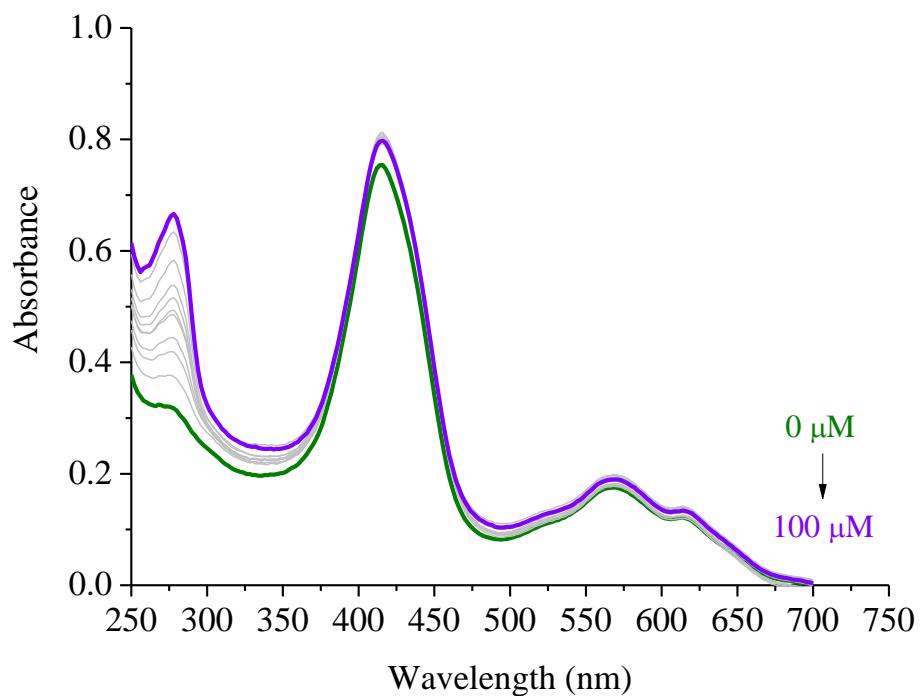


Figure S61. (up) UV-Vis spectra of the corrole **2** upon successive additions of HSA concentrations (0 to 100 μM) in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution. (down) Benesi-Hidelbrandt plots of $A_0 / (A - A_0)$ versus $1 / [HSA]$.

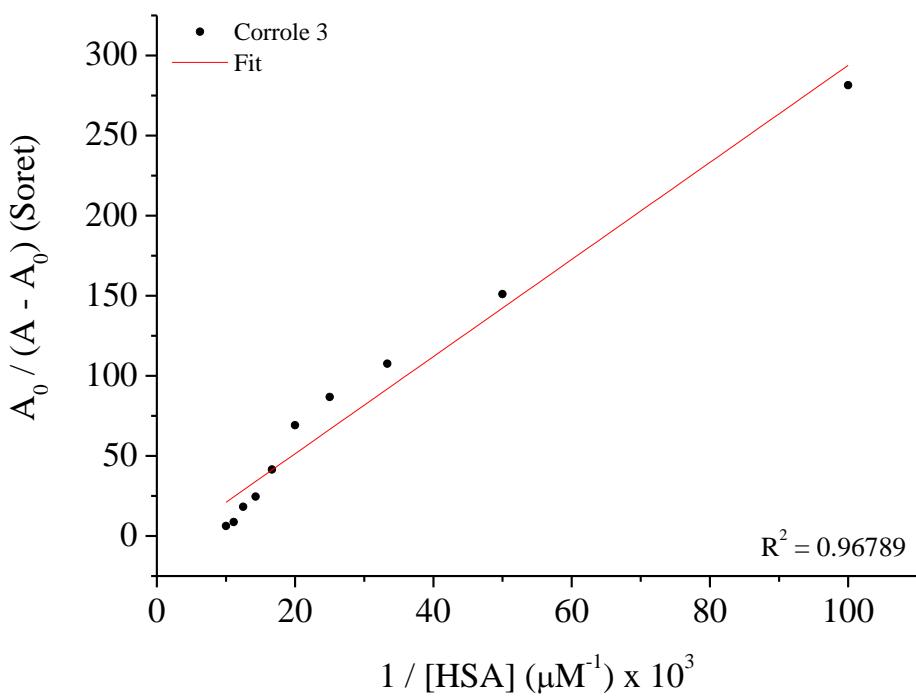
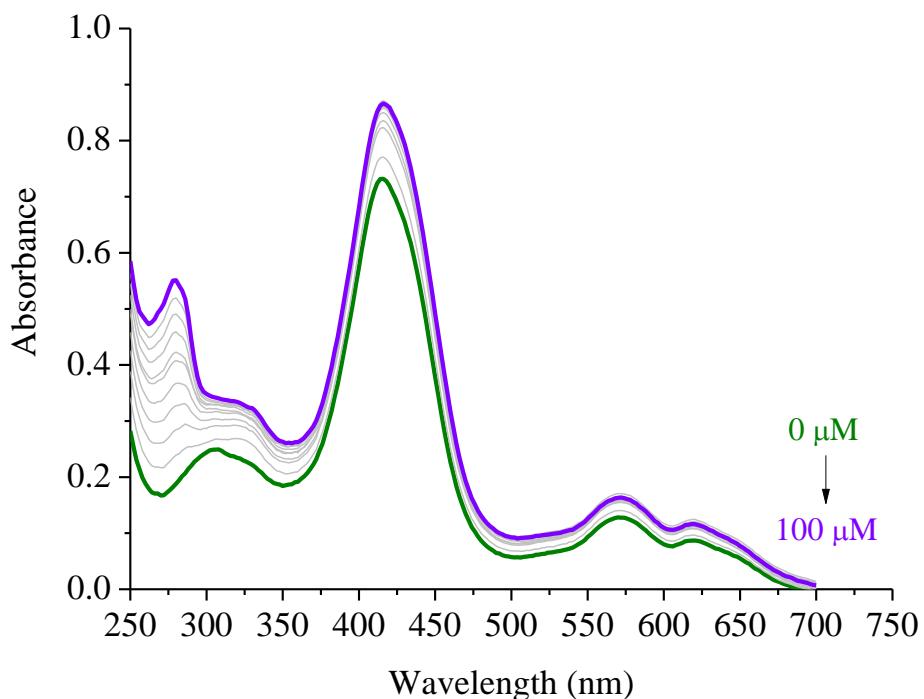


Figure S62. (up) UV-Vis spectra of the corrole **3** upon successive additions of HSA concentrations (0 to 100 μM) in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution. (down) Benesi-Hidelbrandt plots of $A_0 / (A - A_0)$ versus $1 / [\text{HSA}]$.

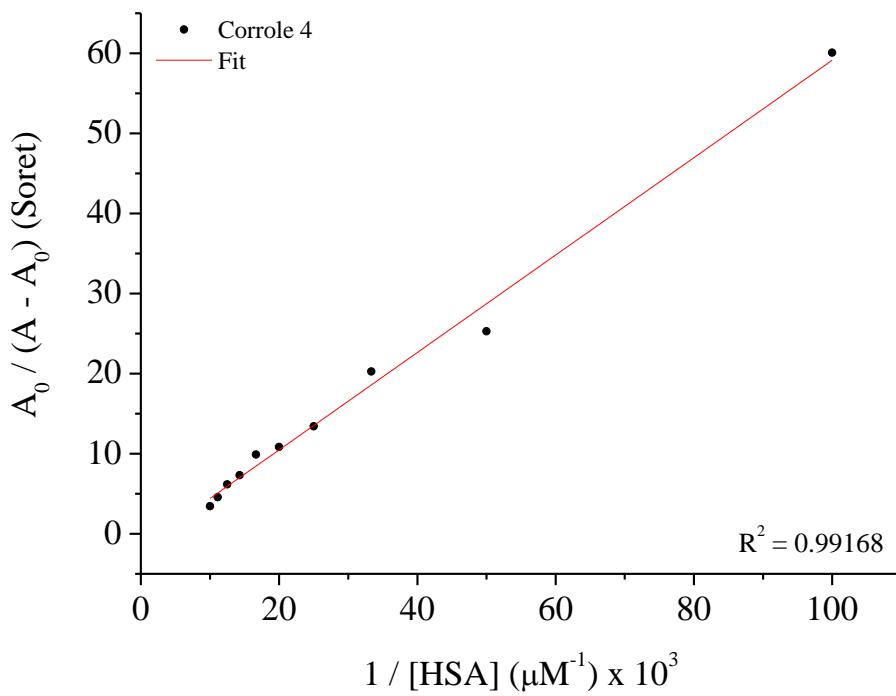
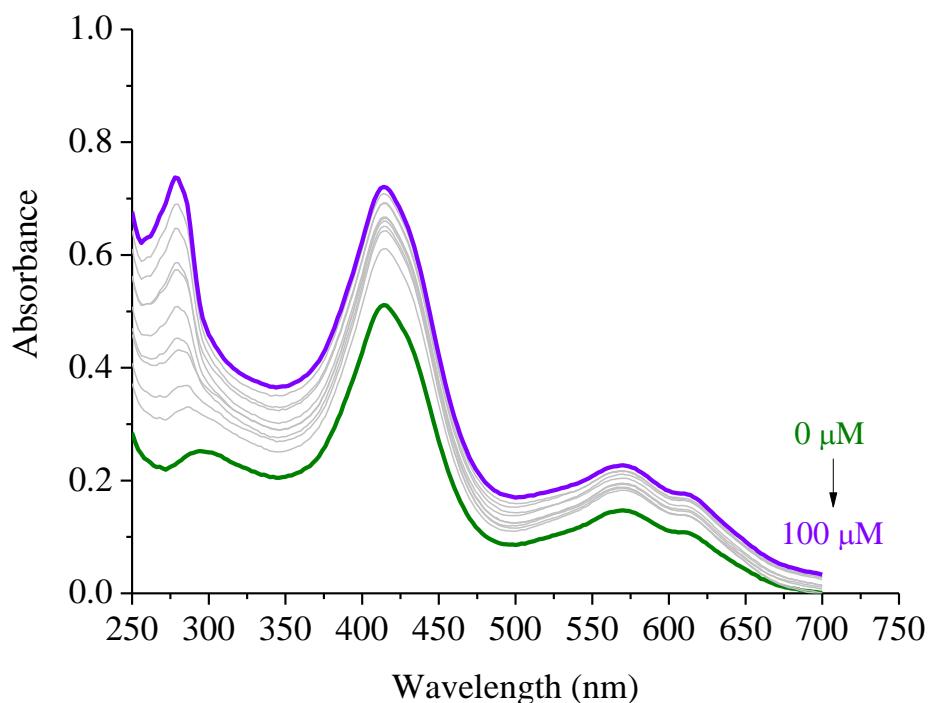


Figure S63. (up) UV-Vis spectra of the corrole **4** upon successive additions of HSA concentrations (0 to 100 μM) in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution. (down) Benesi-Hidelbrandt plots of $A_0 / (A - A_0)$ versus $1 / [\text{HSA}]$.

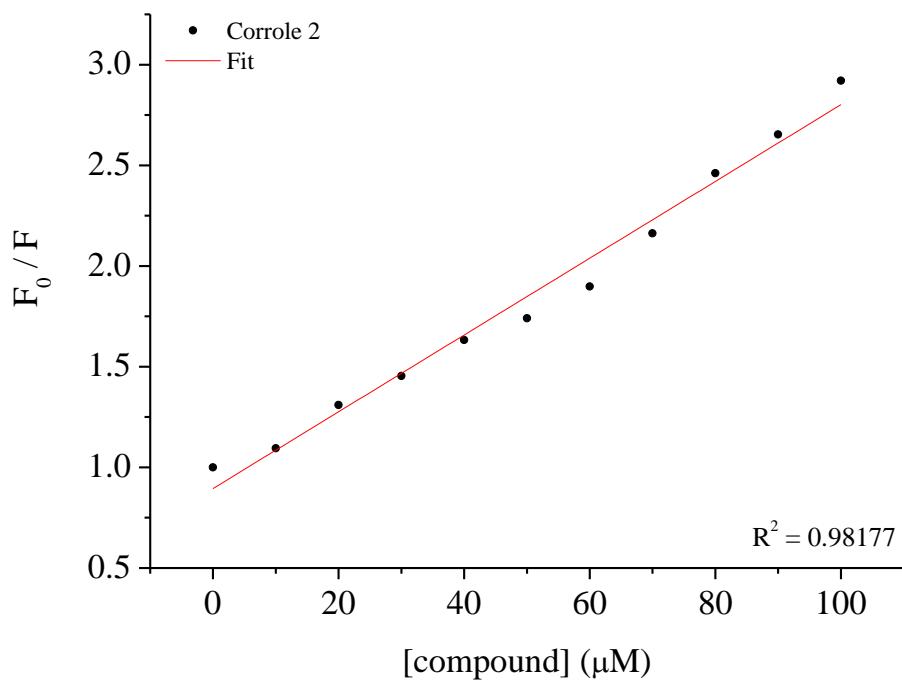
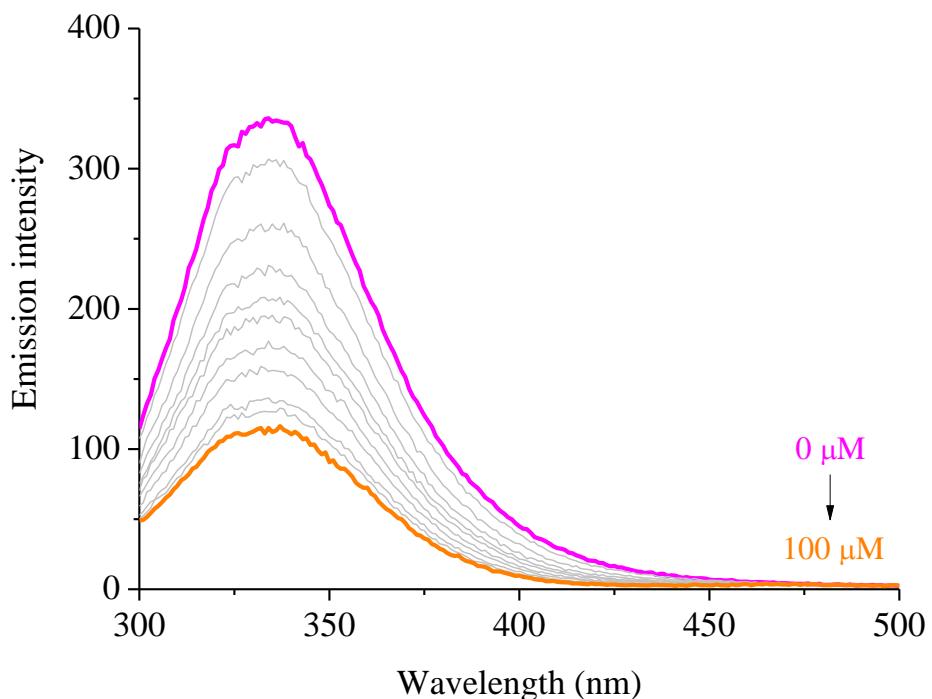


Figure S64. Steady-state fluorescence emission spectra for HSA without and in the presence of corrole **2**, in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution. Graph shows the plot F_0/F versus [corrole]. [corrole] = 0–100 μM .

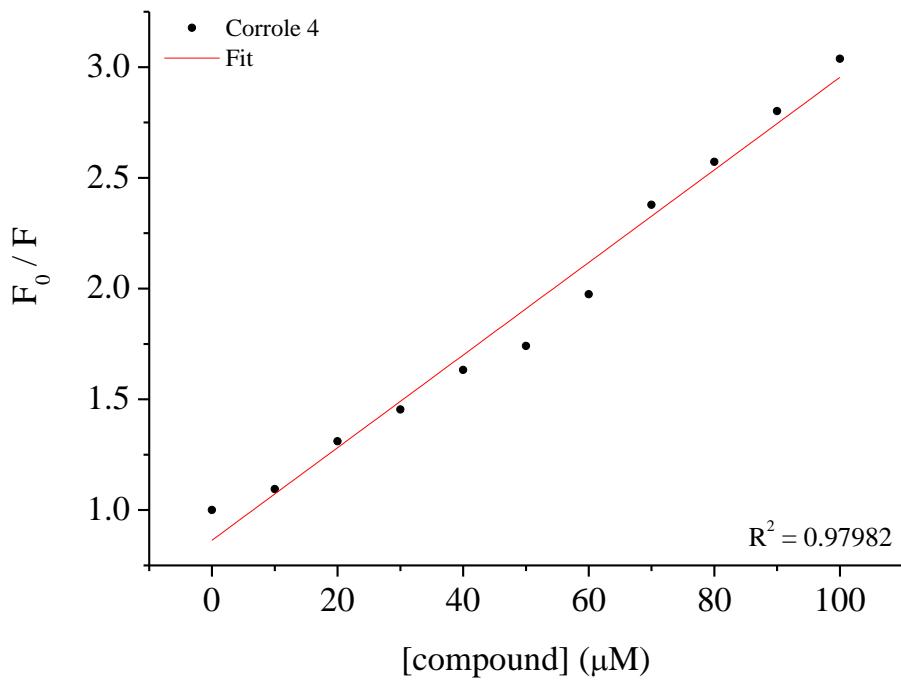
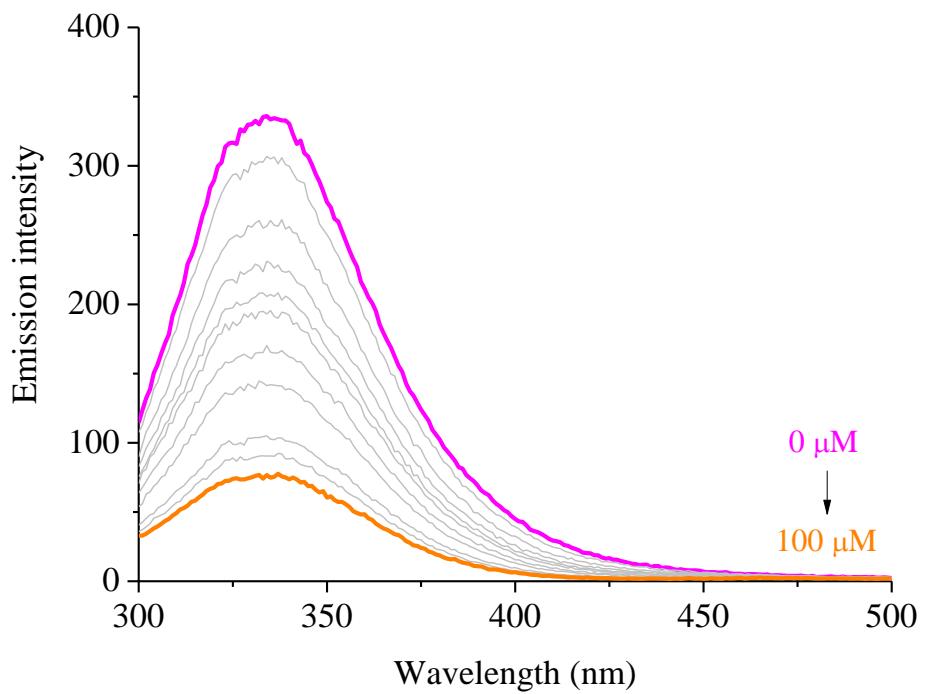


Figure S65. Steady-state fluorescence emission spectra for HSA without and in the presence of corrole 4, in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution. Graph shows the plot F_0/F versus [corrole]. [corrole] = 0–100 μM .

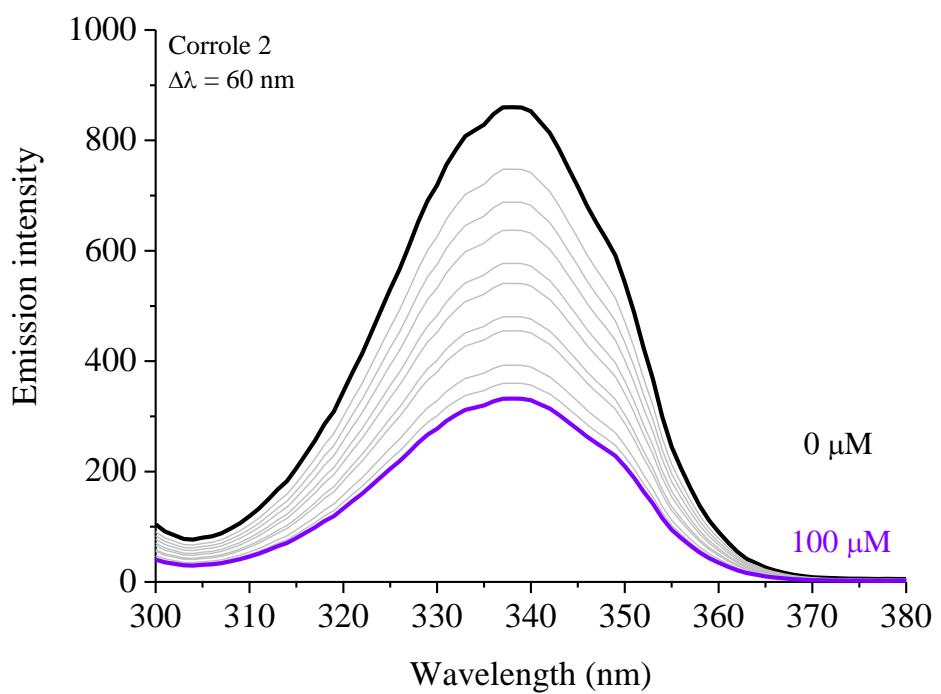
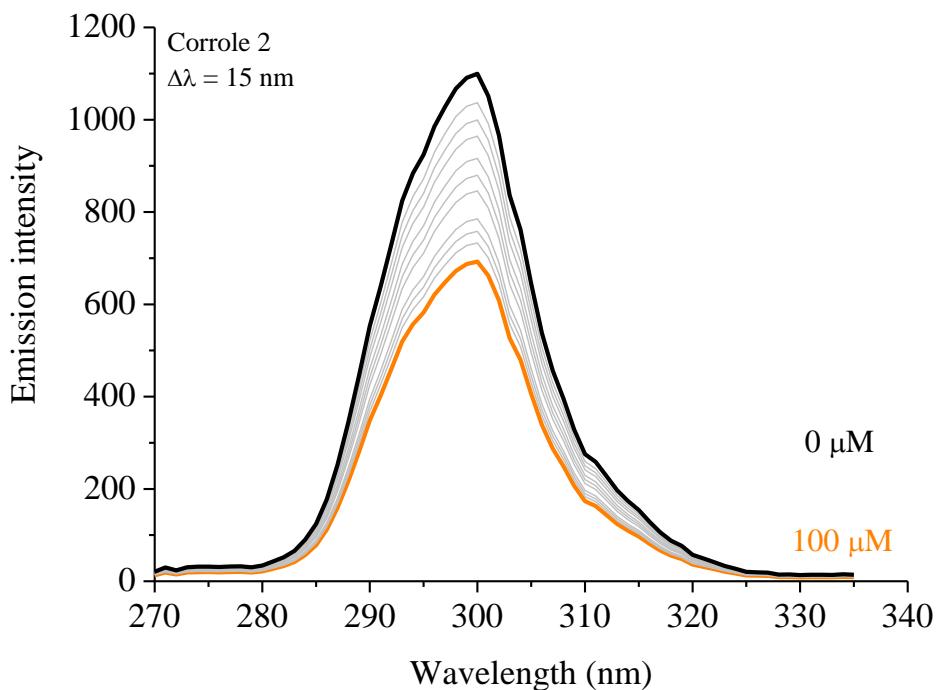


Figure S66. SF fluorescence emission spectra for HSA without and in the presence of corrole **2**, in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution, when (up) $\Delta\lambda = 15 \text{ nm}$ and (down) $\Delta\lambda = 60 \text{ nm}$.

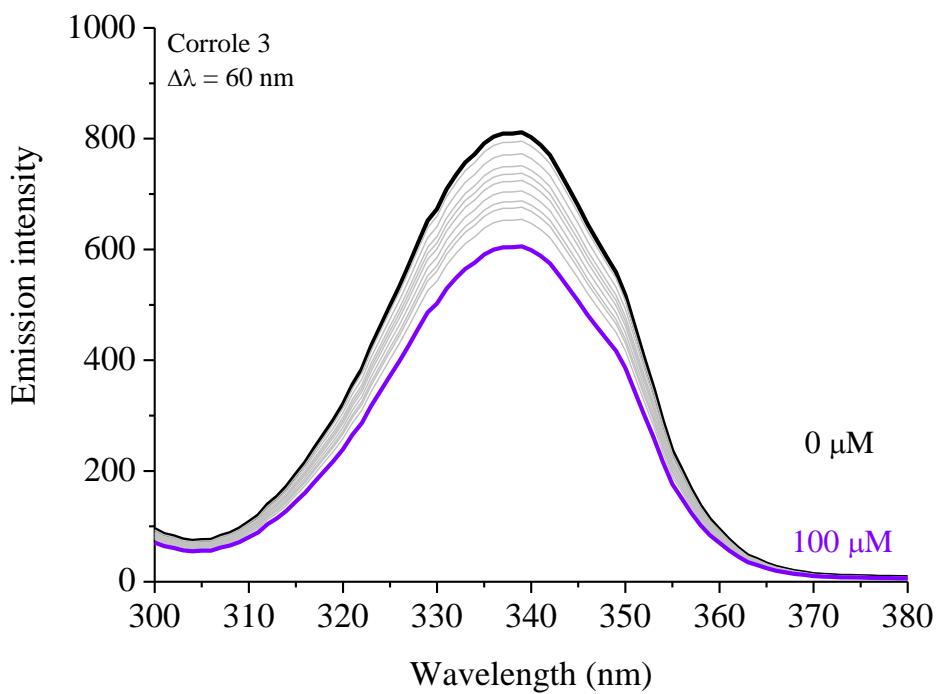
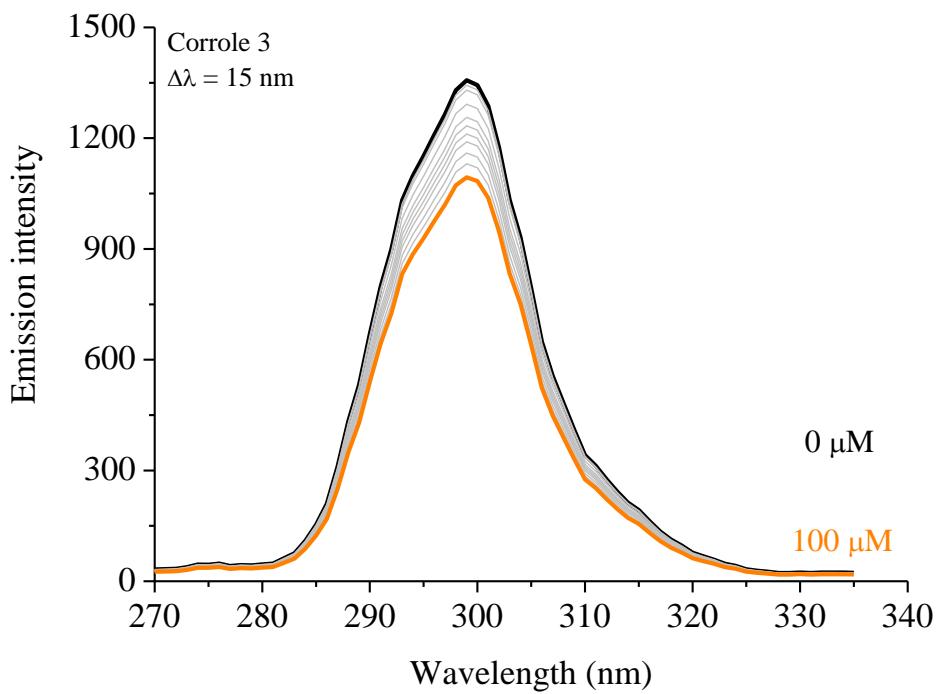


Figure S67. SF fluorescence emission spectra for HSA without and in the presence of corrole 3, in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution, when (up) $\Delta\lambda = 15 \text{ nm}$ and (down) $\Delta\lambda = 60 \text{ nm}$.

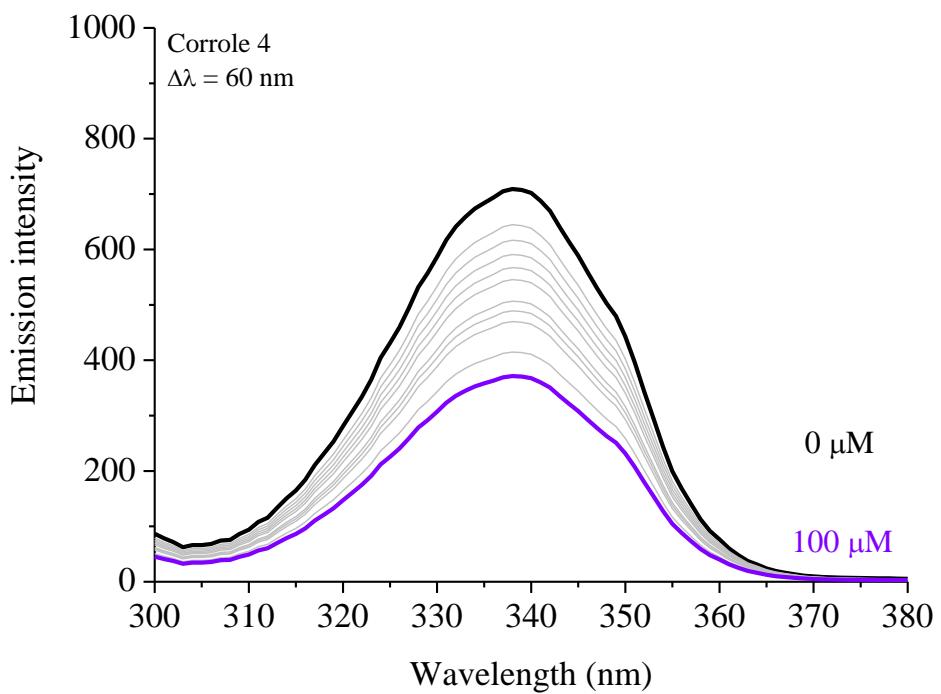
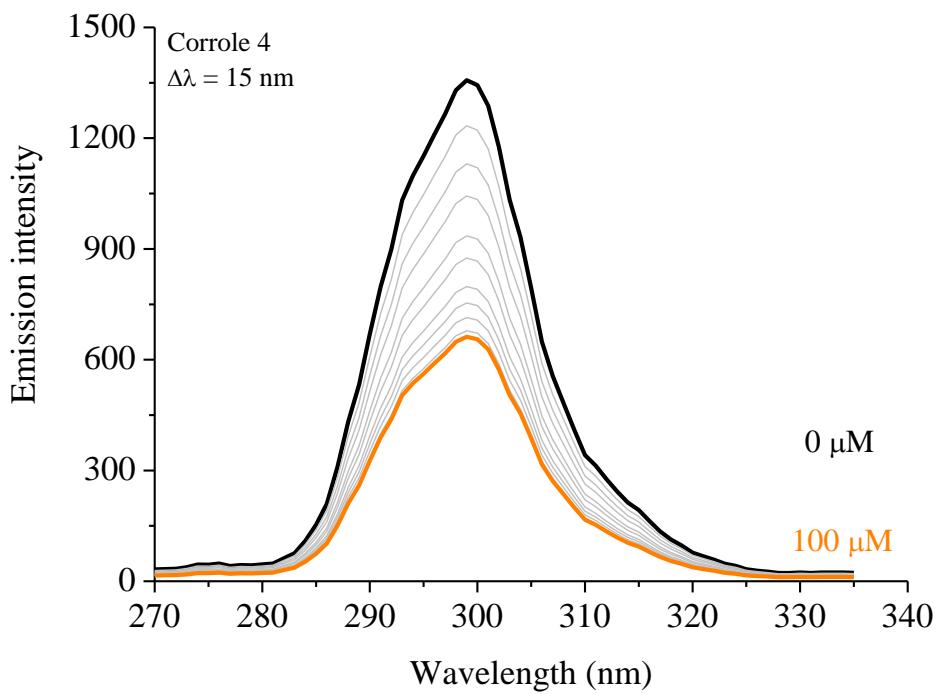


Figure S68. SF fluorescence emission spectra for HSA without and in the presence of corrole 4, in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution, when (up) $\Delta\lambda = 15 \text{ nm}$ and (down) $\Delta\lambda = 60 \text{ nm}$.

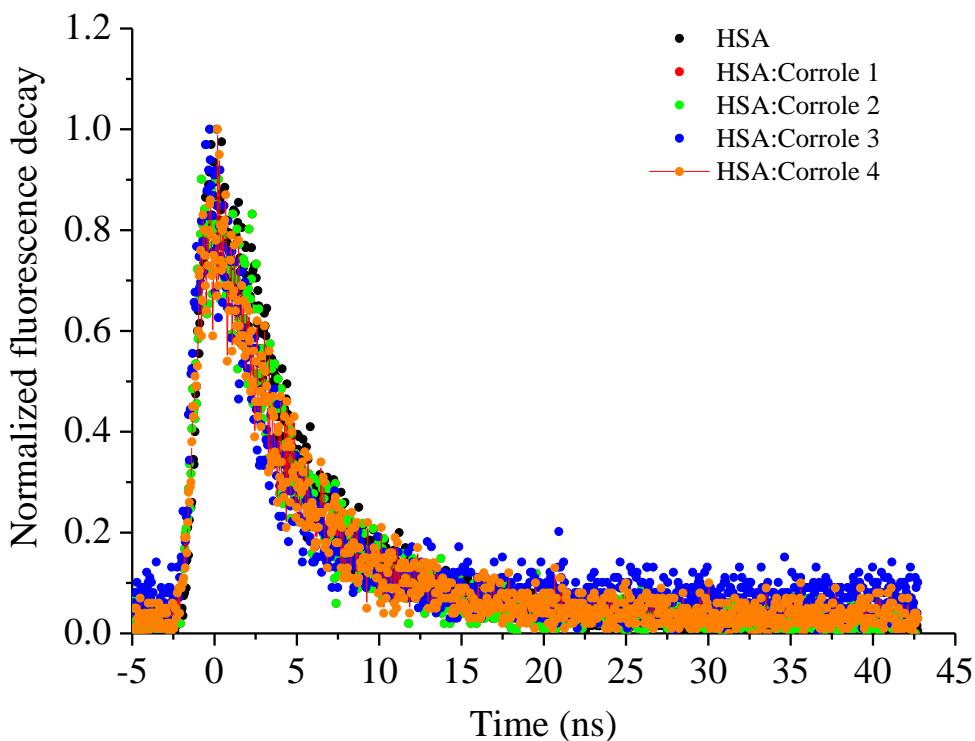


Figure S69. Normalized fluorescence decays of HSA in DMSO(5%)/Tris-HCl pH 7.4 mixture buffered solution, in the absence or in the presence of corroles **1-4**.

Table S1. The absorption wavelengths, λ (nm), and the oscillator strengths, f , of the two main peaks of the Soret and Q-bands of the tautomeric state T1, in DMSO solvent, for the studied corrole compounds.

	Soret				Q			
	λ (nm)	f						
Corrole 1	357.36	1.4144	373.24	1.3183	544.99	0.1032	562.68	0.2926
Corrole 2	358.76	1.4011	375.81	1.3708	547.06	0.1053	564.06	0.2893
Corrole 3	360.69	1.3770	381.29	1.4022	554.05	0.0842	575.62	0.3195
Corrole 4	361.38	1.3800	382.45	1.5170	552.76	0.0793	573.38	0.3105

Table S2. The absorption wavelengths, λ (nm), and the oscillator strengths, f , of the two main peaks of the Soret and Q-bands of the tautomeric state T2, in DMSO solvent, for the studied corrole compounds.

	Soret				Q			
	λ (nm)	f						
Corrole 1	366.45	1.4491	381.47	1.4273	533.03	0.0356	581.29	0.1634
Corrole 2	368.48	1.4580	386.81	1.4265	538.55	0.0527	588.15	0.1885
Corrole 3	370.57	1.4518	390.31	1.4458	543.60	0.0376	596.63	0.2187
Corrole 4	371.36	1.4600	391.58	1.5509	542.74	0.0356	595.04	0.2077

Table S3. Fluorescence decay data of HSA in the absence or in the presence of corroles **1-4**.

Corrole	tf (ns)	χ^2
HSA:1	4.89 ± 0.002	1.03680
HSA:2	4.77 ± 0.002	1.17756
HSA:3	4.60 ± 0.002	0.95752
HSA:4	4.60 ± 0.002	1.07678
HSA	5.25 ± 0.001	1.13768

Table S4. Molecular docking results for the interaction between DNA:corroles in the minor groove.

Corrole	Nitrogenated bases	Interaction	Distance (Å)
Corrole 2	DG-04	Van der Waals	1.50
	DA-05	Van der Waals	2.30
	DT-20	Van der Waals	3.70
	DC-21	Van der Waals	3.40
	DG-22	Hydrogen bonding	3.20
Corrole 3	DG-04	Van der Waals	2.80
	DA-05	Van der Waals	1.70
	DA-06	Hydrogen bonding	1.90
	DT-20	Hydrogen bonding	3.10
	DC-21	Van der Waals	2.00
Corrole 4	DG-22	Hydrogen bonding	3.20
	DG-02	Hydrogen bonding	3.60
	DC-23	Van der Waals	3.40
	DG-04	Van der Waals	1.40
	DC-21	Van der Waals	3.80
	DA-05	Van der Waals	2.40
	DT-20	Van der Waals	3.30

Table S5. Molecular docking results for the interaction between HSA:corroles for the corresponding main binding site.

Corrole	Amino acid residue	Interaction	Distance (Å)
Corrole 2	Lys-195	Hydrogen bonding	2.90
	Lys-199	Van der Waals	1.50
	Trp-214	Van der Waals	2.80
	Arg-222	Hydrogen bonding	3.00
	His-242	Van der Waals	3.10
	Cys-245	Van der Waals	2.80
	Arg-257	Hydrogen bonding	3.70
	Ala-291	Van der Waals	3.40
	Glu-292	Van der Waals	2.20
	Val-343	Van der Waals	2.40

	Arg-114	Hydrogen bonding	3.70
	Leu-115	Van der Waals	3.60
	Arg-117	Van der Waals	1.40
Corrole 3	Ile-142	Van der Waals	1.50
Corrole 4	His-146	Van der Waals	3.40
	Tyr-161	Hydrogen bonding	2.60
	Leu-182	Van der Waals	2.40
	Arg-186	Van der Waals	3.20
	Lys-190	Van der Waals	2.70