

Supporting Information for

Complexation of Gold(III) with Pyridoxal 5'-Phosphate-Derived Hydrazones in Aqueous Solution

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Table S1. Preliminary virtual screening results using PASS Online software [1].

Compound	Strain	Disease	Confidence
PLP-INH	<i>RESISTANT Mycobacterium tuberculosis H37Rv</i>	Tuberculosis	0.1909
	<i>Enterococcus faecalis ATCC 29212</i>	Bacterial infection	0.1762
Complex Au ³⁺ - PLP-INH	<i>RESISTANT Mycobacterium tuberculosis H37Rv</i>	Tuberculosis	0.2552
	<i>Enterococcus faecalis ATCC 29212</i>	Bacterial infection	0.1930
Complex Au ³⁺ - PLP-F2H	<i>RESISTANT Mycobacterium tuberculosis H37Rv</i>	Tuberculosis	0.1781
PLP-T2H	<i>Dialister micraerophilus</i>	Periodontitis	0.3580
	<i>Dialister pneumosintes</i>		0.3580
	<i>Dialister propionicifaciens</i>		0.2577
	<i>Dialister invisus</i>		0.2190
	<i>Mycobacterium mageritense</i>	Tuberculosis	0.1647
Complex Au ³⁺ - PLP-T2H	<i>Dialister pneumosintes</i>	Periodontitis	0.3812
	<i>Dialister micraerophilus</i>		0.3812
	<i>Dialister propionicifaciens</i>		0.2850
	<i>Dialister invisus</i>		0.2349
	<i>Mycobacterium mageritense</i>	Tuberculosis	0.1582
PLP-T3H	<i>Dialister invisus</i>	Periodontitis	0.5628
Complex Au ³⁺ - PLP-T3H	<i>Dialister invisus</i>	Periodontitis	0.5836

"AntiBac-Pred functional of PASS Online software allows user to predict the fact that chemical compound can inhibit the growth of one or more of 353 bacteria in concentration below the 10000 nM. The score for each compound is expressed as confidence in its activity, which is a difference between probabilities for chemical compound to inhibit and to do not inhibit the growth of the particular bacteria. The higher confidence means the higher chance of the positive prediction to be true.

Only activities with $P_a > P_i$ (confidence > 0) are considered as possible for a particular compound". The description is taken from [2].

Reference

1. Biomed. Chem.: Res. Methods, 2018, 1(1), e00004, doi: 10.18097/bmcrm00004
2. <http://www.way2drug.com/antibac/>

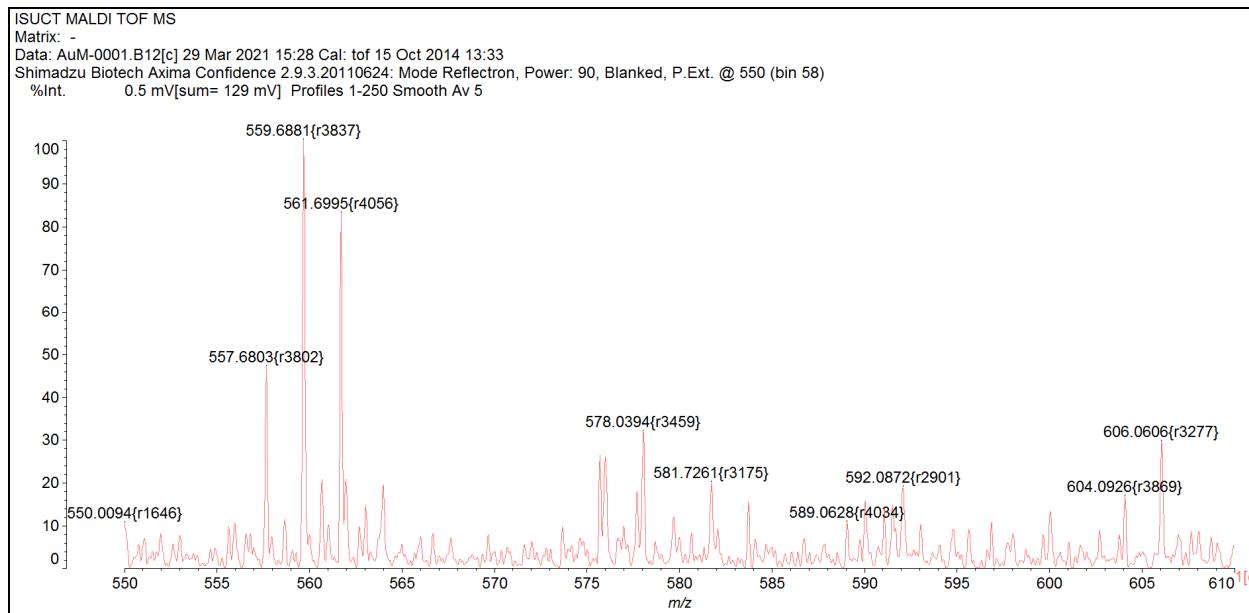


Figure S1. Mass-spectrum of Au-PLP-INH complex precipitated from an aqueous solution.

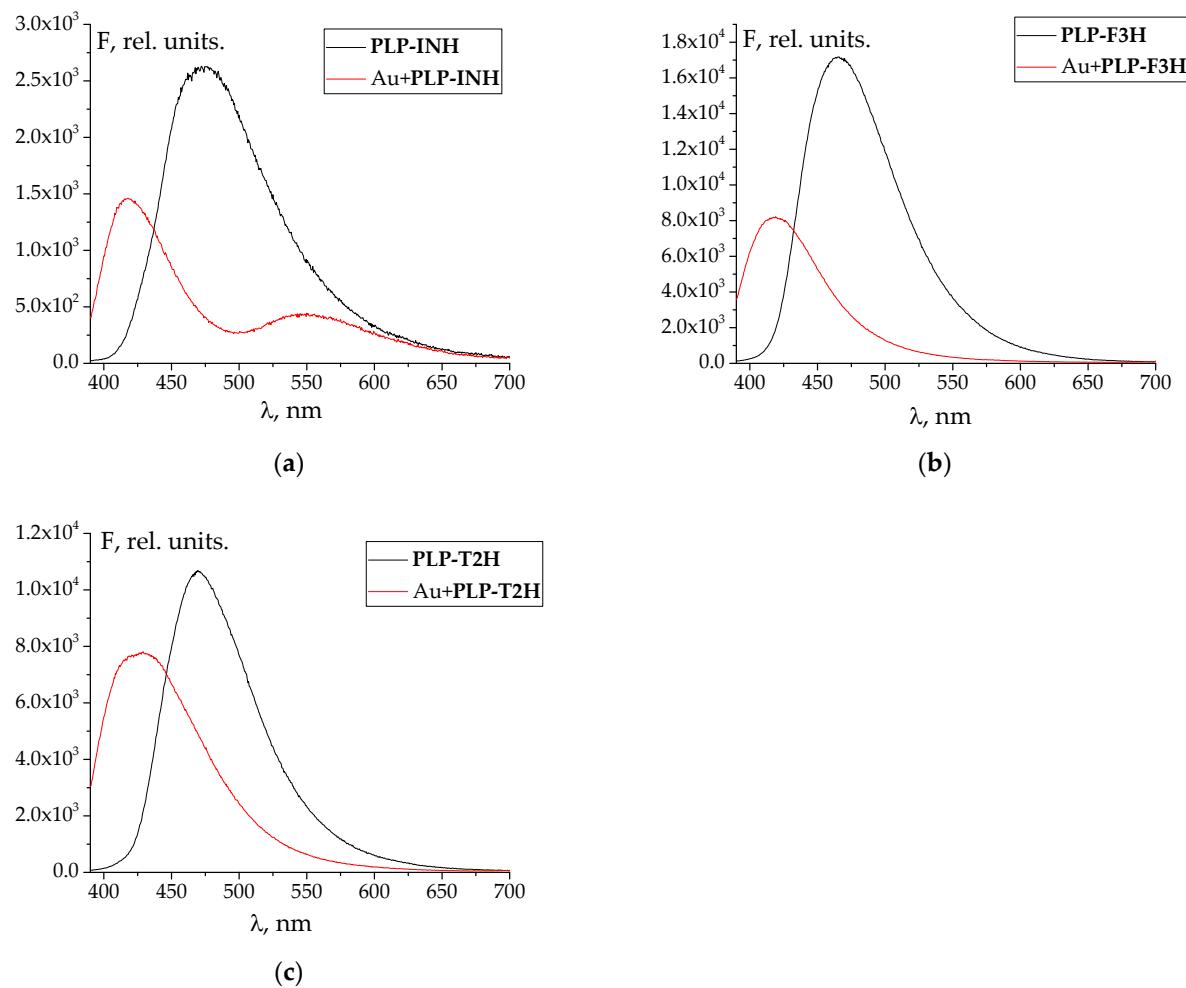


Figure S2. Emission spectra of free hydrazones and their mixtures with Au^{3+} : (a) PLP-INH; (b) PLP-F3H; (c) PLP-T2H. $\lambda_{\text{ex}} = 365 \text{ nm}$.

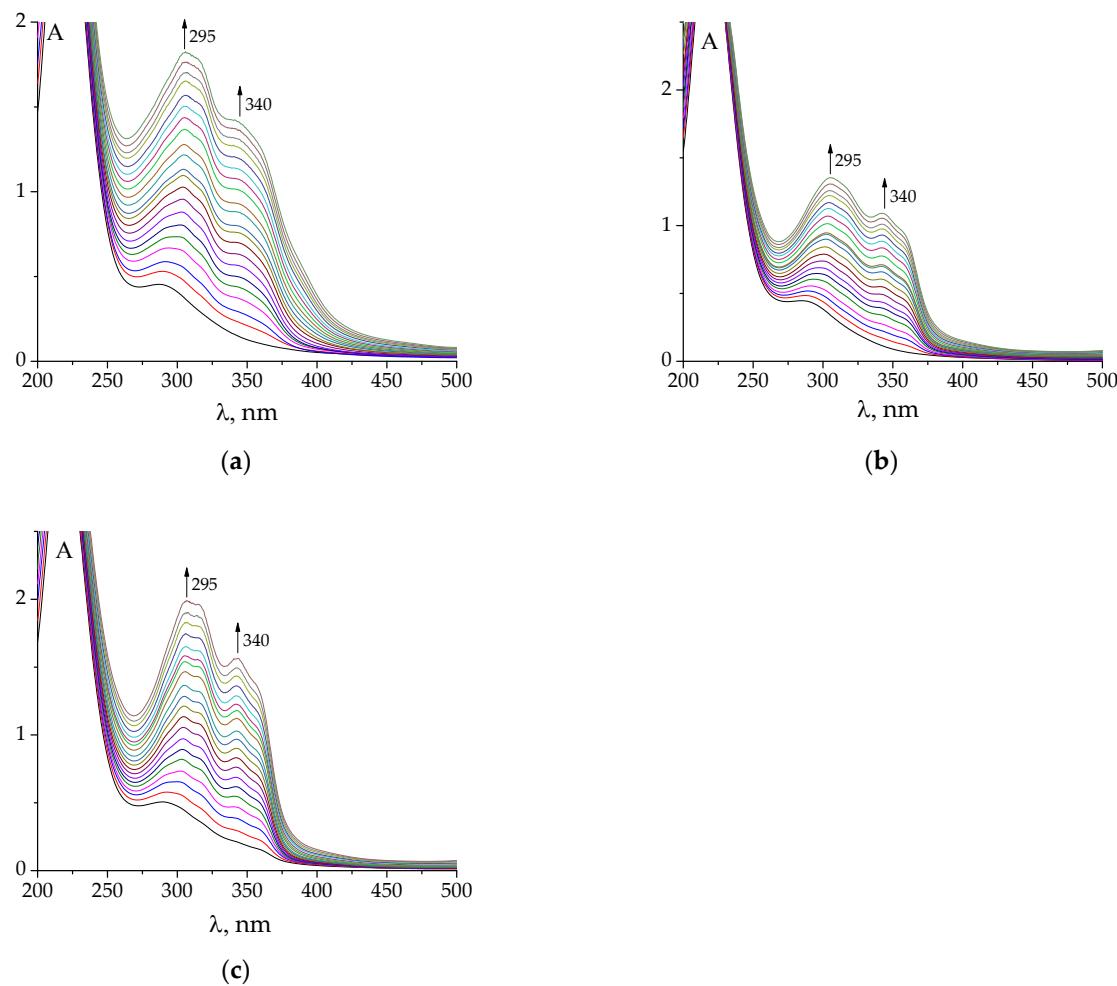


Figure S3. Examples of changes in UV-Vis spectra of HAuCl_4 upon addition of **PLP-INH** (a), **PLP-F3H** (b), **PLP-T3H** (c). Titrand: $C(\text{AuCl}_4) = 2 \cdot 10^{-4} \text{ mol L}^{-1}$; $C(\text{H}^+) = 5 \cdot 10^{-4} \text{ mol L}^{-1}$. Titrant: $C(\text{PLP-INH}) = 1.5 \cdot 10^{-3} \text{ mol L}^{-1}$; $C(\text{OH}^-) = 0.01074 \text{ mol L}^{-1}$ (a); Titrant: $C(\text{PLP-F3H}) = 1.5 \cdot 10^{-3} \text{ mol L}^{-1}$; $C(\text{OH}^-) = 0.01074 \text{ mol L}^{-1}$ (b); Titrant: $C(\text{PLP-T3H}) = 1.5 \cdot 10^{-3} \text{ mol L}^{-1}$; $C(\text{OH}^-) = 0.01074 \text{ mol L}^{-1}$ (b). Twenty additions of $10 \mu\text{L}$ volume.

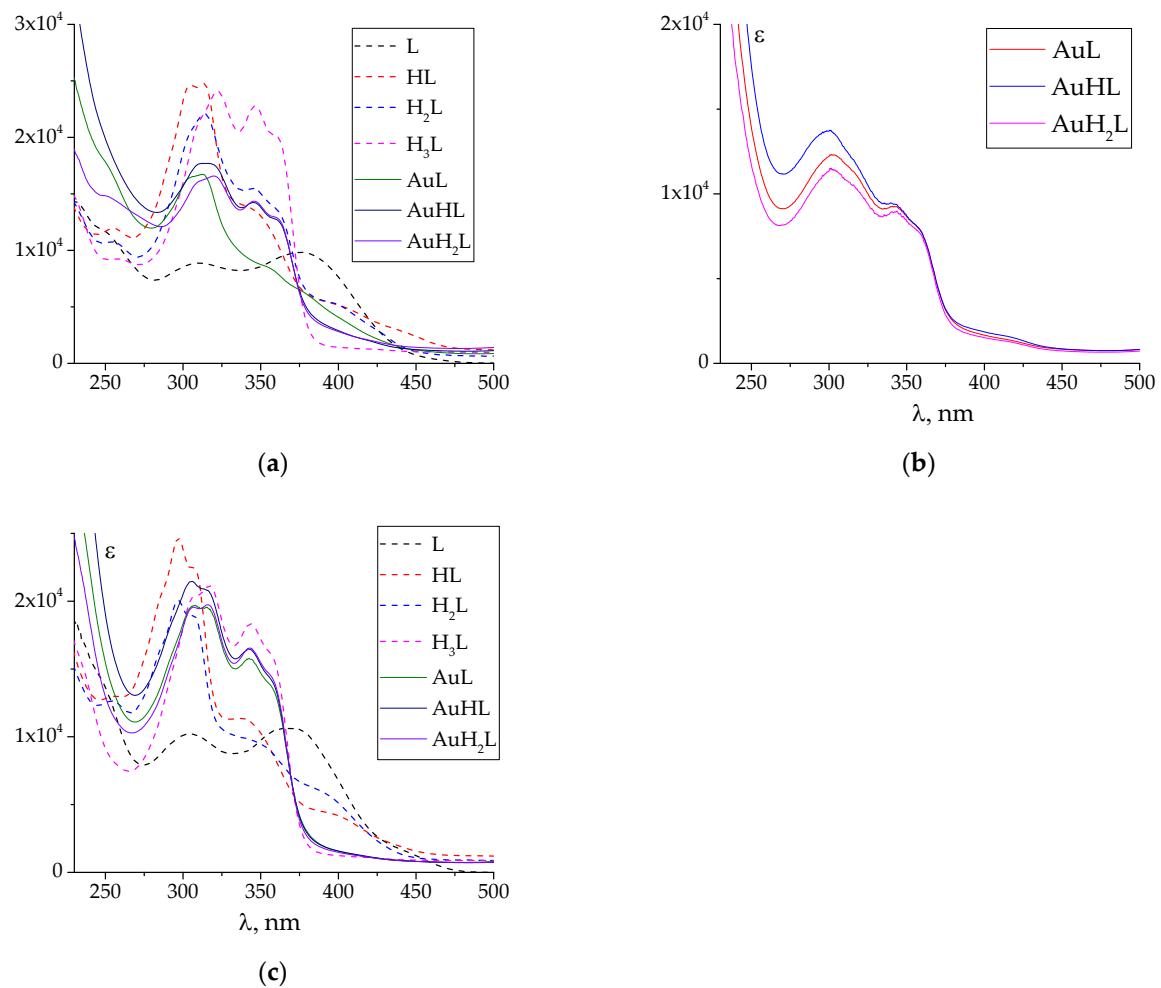


Figure S4. Calculated UV-Vis spectra of individual protonated and complex species of **PLP-T2H** (a), **PLP-F3H** (b), **PLP-T3H** (c). Spectra of protonated species are adopted from paper 10.1016/j.molliq.2020.112822.