

Supporting Information

A Fluoroponytailed NHC- Silver Complex formed from Vinyl-imidazolium /AgNO₃ under Aqueous-Ammoniacal Conditions

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† Dedicated to the memory of Professor Victor Snieckus.

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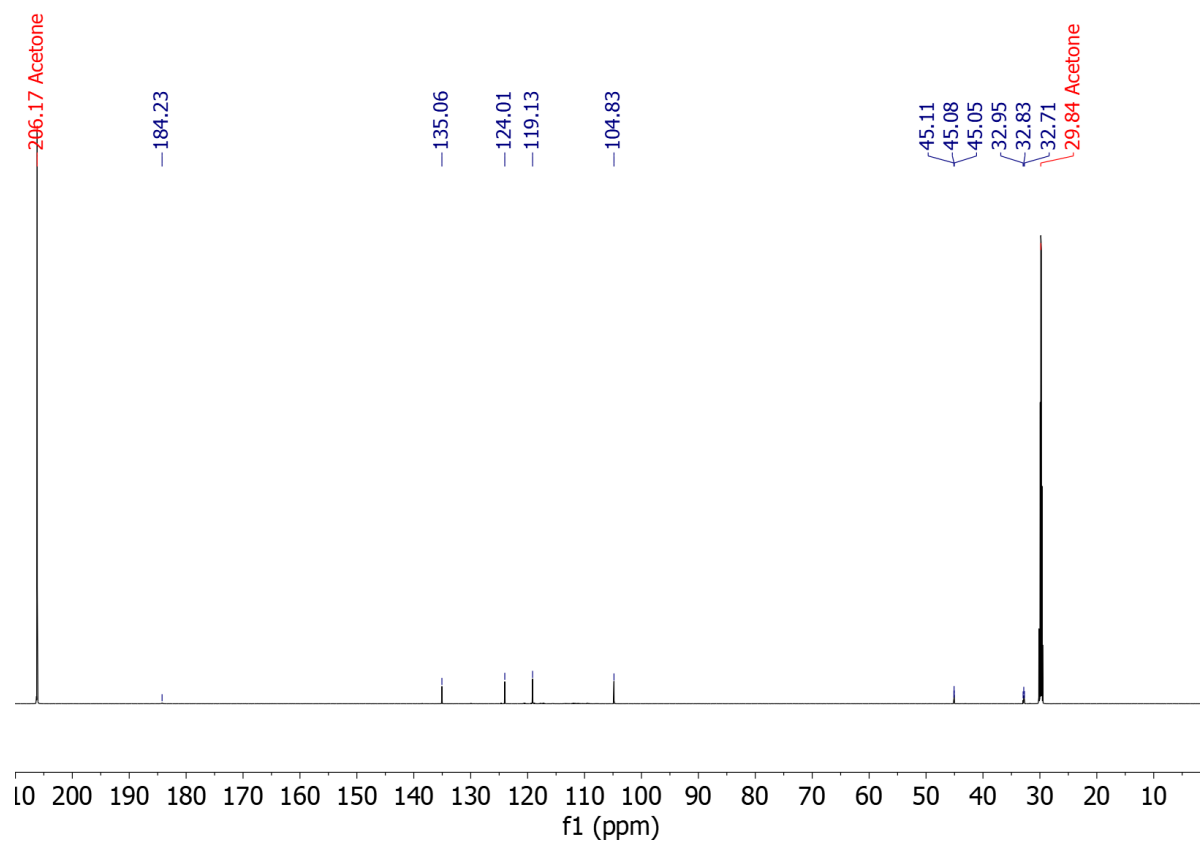


Figure S1: ¹³C NMR spectrum (176 MHz, acetone-*d*₆) of [bis(3-(1*H*,1*H*,2*H*,2*H*-perfluorooctyl)-1-vinylimidazol-2-ylidene)silver(I)] nitrate (**Ag**(FNHC)₂NO₃).

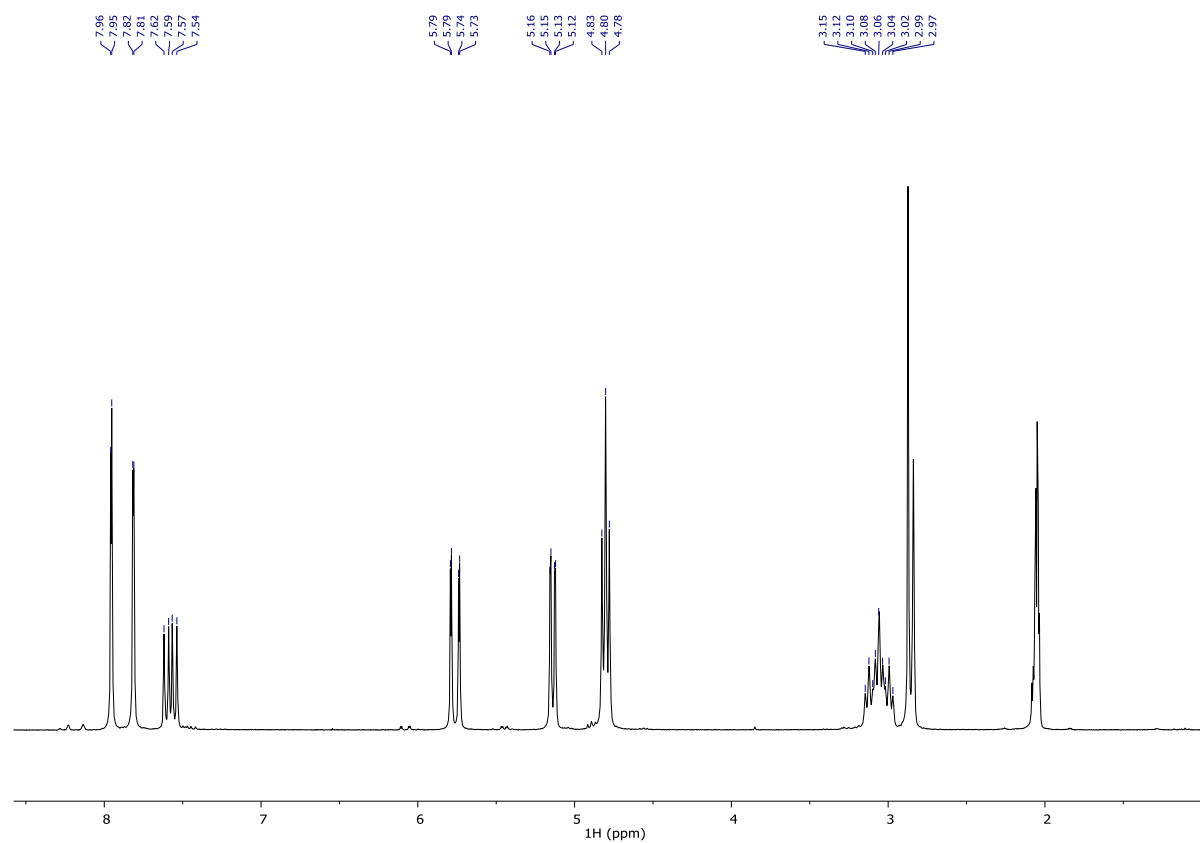


Figure S2: ¹H NMR spectrum (300 MHz, acetone-*d*₆) of **Ag**(FNHC)₂NO₃.

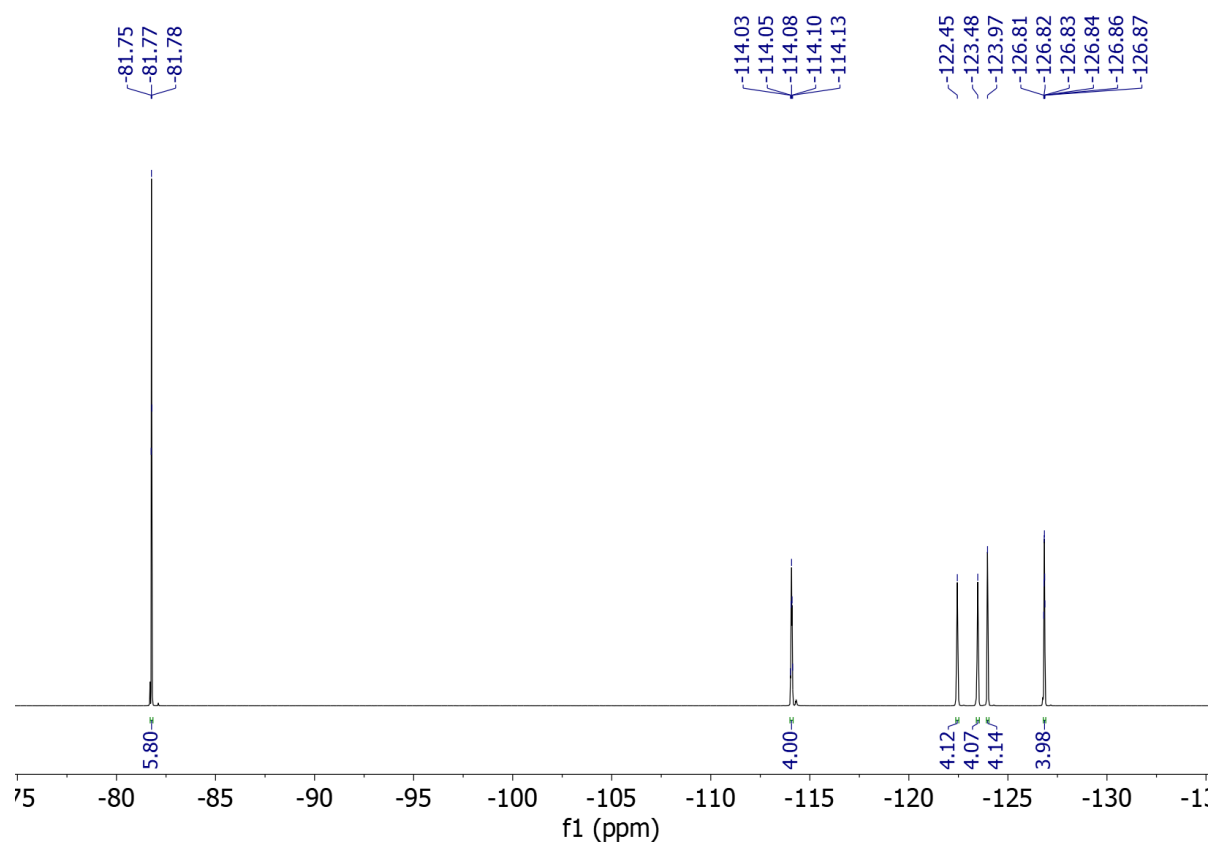


Figure S3: ¹⁹F NMR spectrum (659 MHz, acetone-*d*₆) of Ag(FNHC)₂NO₃.

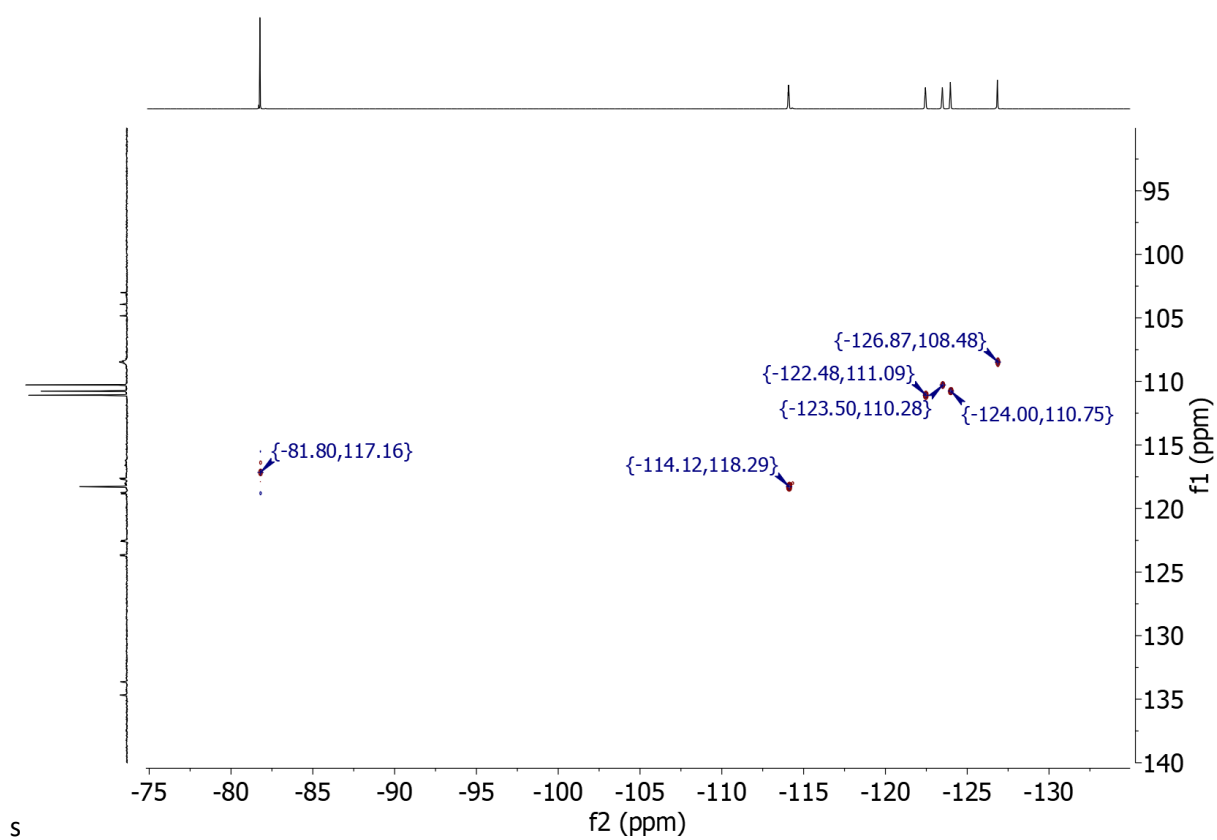


Figure S4: ¹⁹F - ¹³C HSQC NMR spectrum (¹⁹F frequency = 659 MHz, acetone-*d*₆) of Ag(FNHC)₂NO₃. ¹³C (¹⁹F decoupled) trace is shown on the left; ¹⁹F trace is shown on top.

Note: For clarity and better s/n ratio, the ^{13}C trace shown in Figure S4 is ^{19}F decoupled. The ^{13}C trace in Figure S1 is ^1H decoupled and thus looks markedly different.

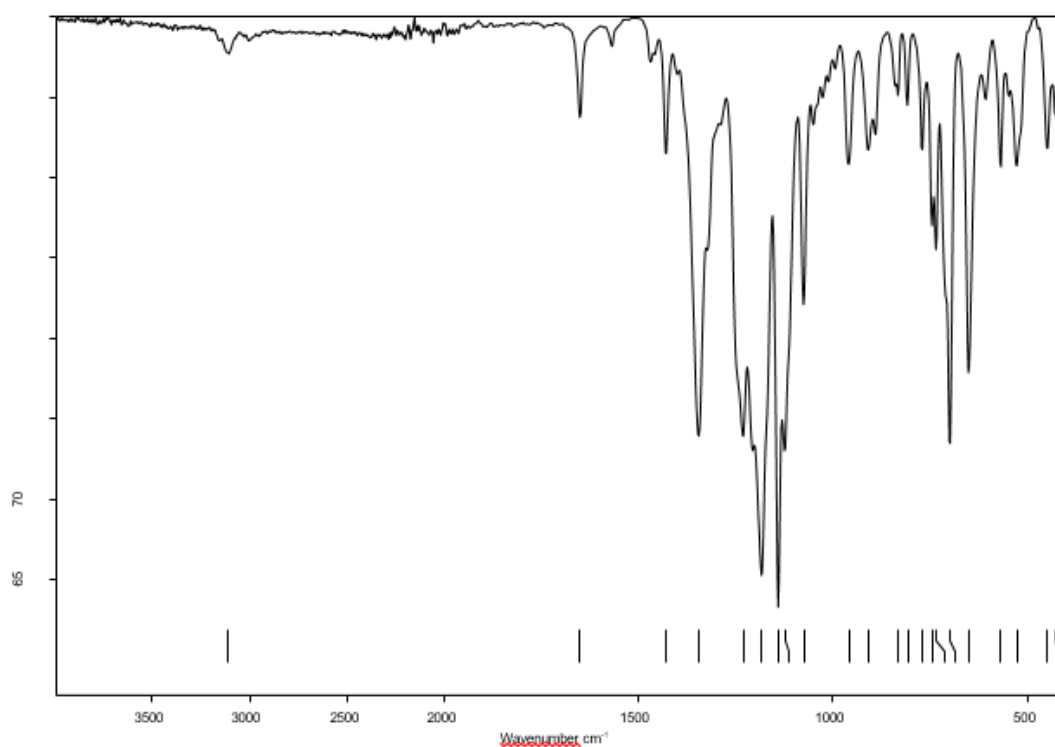


Figure S5: ATR-FT-IR spectrum (neat) of $\text{Ag}(\text{FNHC})_2\text{NO}_3$.

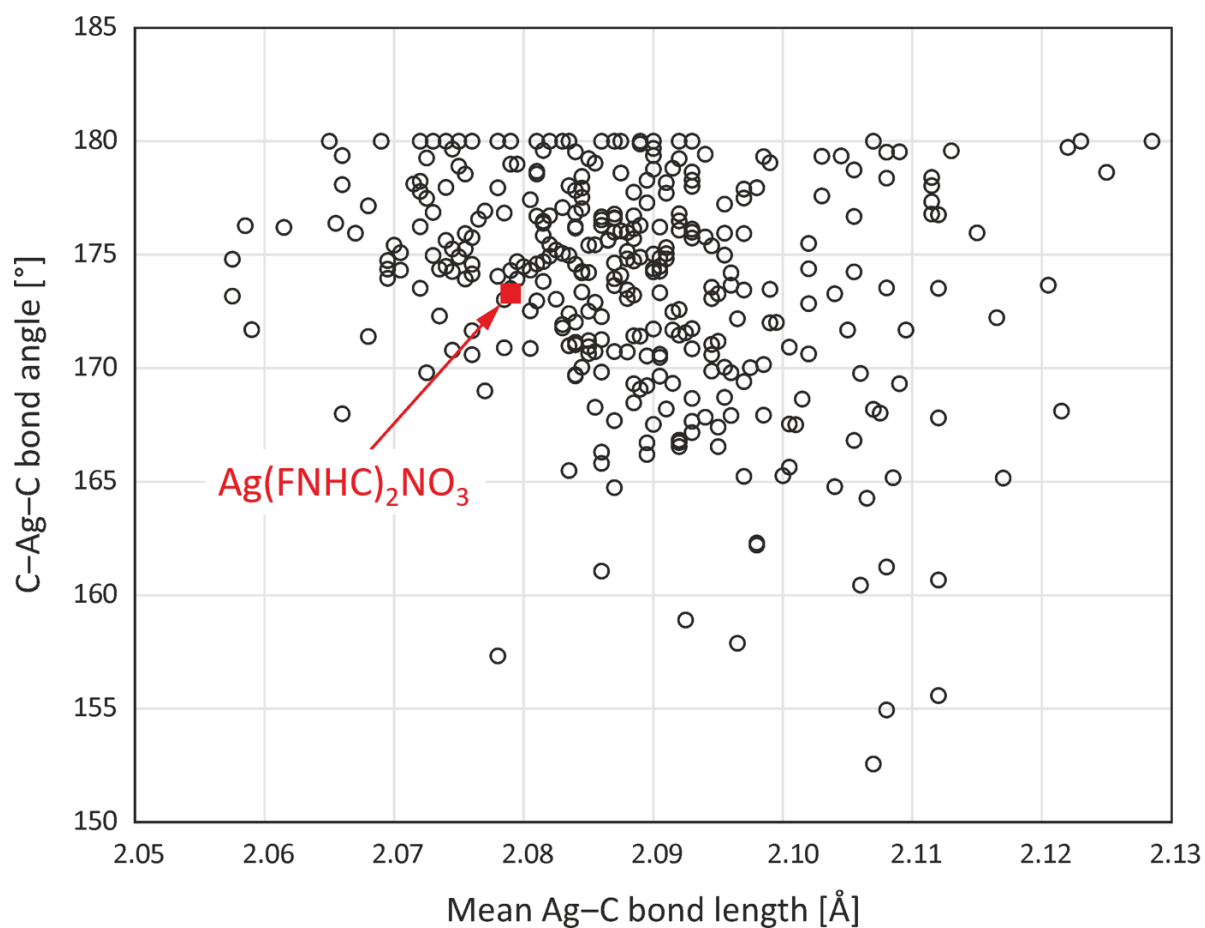


Figure S6: Plot showing Ag-C bond lengths (x-axis) and their corresponding C-Ag-C bond angles (y-axis) in reported bis-carbene-Ag complexes in comparison to those of $\text{Ag}(\text{FNHC})_2\text{NO}_3$.

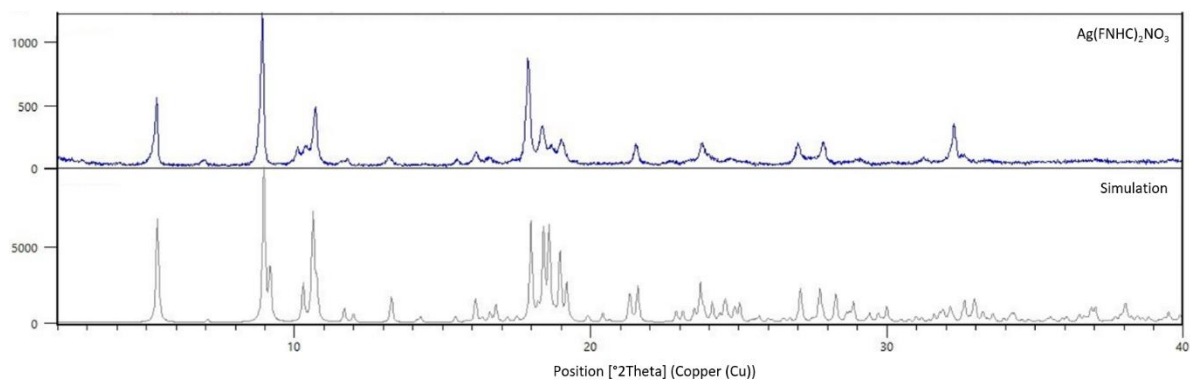


Figure S7: Powder diffractogram of $\text{Ag}(\text{FNHC})_2\text{NO}_3$ (top) and comparison to simulated powder diffraction pattern (bottom).

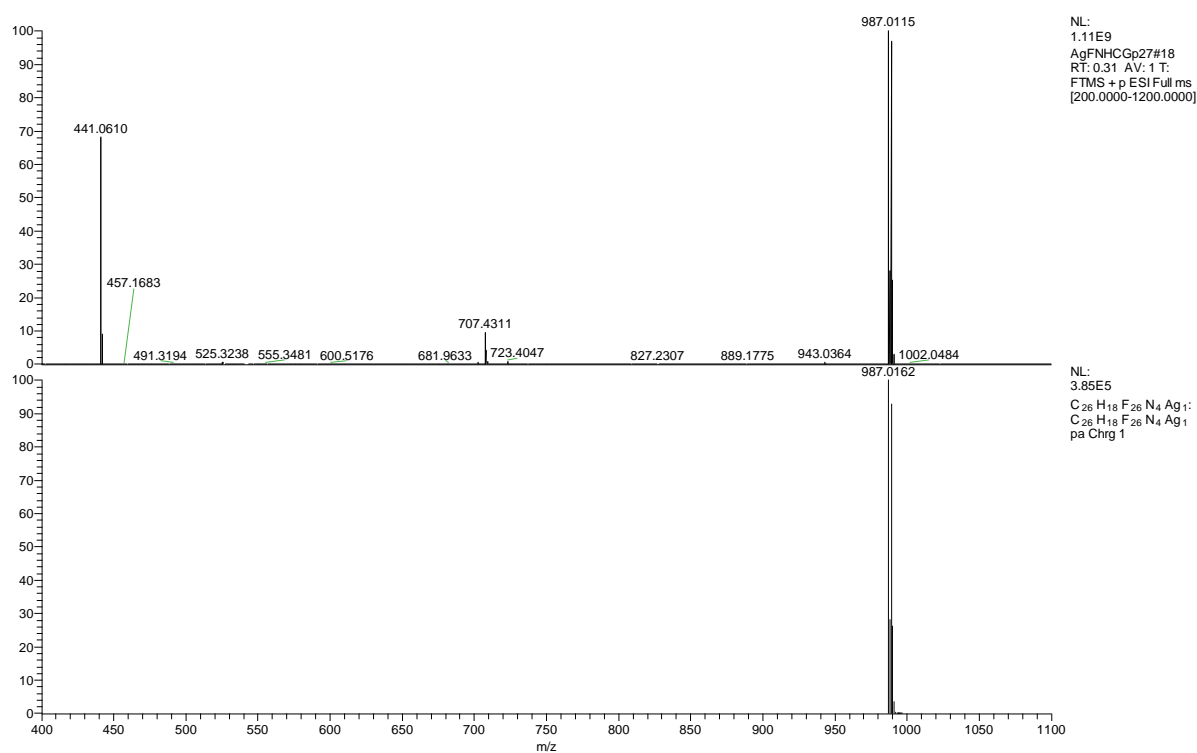


Figure S8: High resolution mass spectrum of $\text{Ag}^+(\text{FNHC})_2$ (top) and simulated spectrum thereof (bottom). The peak at 441.0610 m/z is attributable to the 3-(1*H*,1*H*,2*H*,2*H*-perfluorooctyl)-1-vinylimidazolium ion.

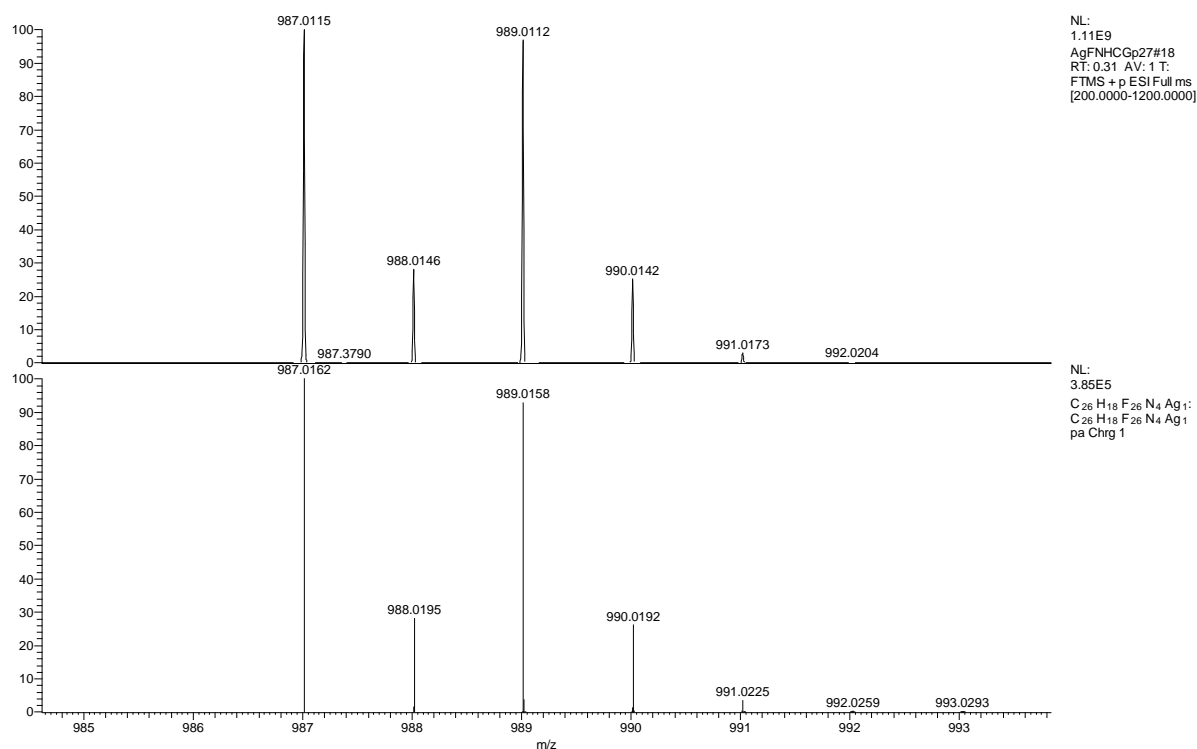


Figure S9: High resolution mass spectrum of the molecule ion peak $\text{Ag}^+(\text{FNHC})_2$ (top) and simulated spectrum thereof (bottom).

Elemental analysis

Elemental composition values for $\text{C}_{26}\text{H}_{18}\text{AgF}_{26}\text{N}_5\text{O}_3$	Calculated	Found
C	29.73 %	28.61%
H	1.73 %	1.55%
N	6.67%	6.36%

Note: Composition values for carbon and hydrogen likely deviate from the expected values due to reluctant combustion behavior of the polyfluorinated compound.