

# iNovo479: Metabolic Modeling Provides a Roadmap to Optimize Bioproduct Yield from deconstructed Lignin Aromatics by *Novosphingobium aromaticivorans*

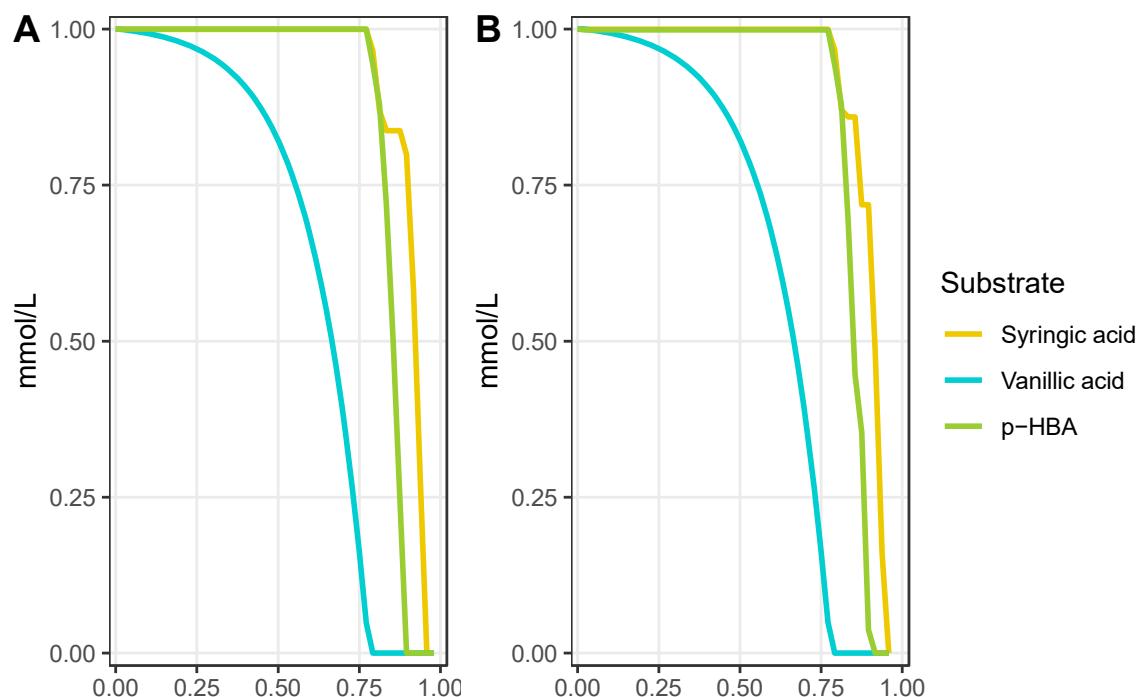
Alexandra M. Linz, Yanjun Ma, Samuel Scholz, Daniel R. Noguera and Timothy J. Donohue

## Supplemental information

**Table S1.** Measured biomass composition of *N. aromaticivorans*

Characteristic	Value <sup>1</sup>
DNA	1.7%
RNA	6.6%
Protein	60.0%
Phospholipids	7.3%

<sup>1</sup> Percent of total biomass



**Figure S1.** Co-metabolism of SA, VA, and p-HBA by versions of iNovo479 with a hypothetical, non-THF dependent demethylation step (Panel A) and with a VanAB-like demethylation step (Panel B). Each organic substrate was provided at 1 mmol/L.

**Table S2. Native vs. hypothetical vs. *P. putida* demethylation reactions in iNovo479**

Model version	Reaction ID	Reversible	Reaction Name	Reaction
native	A007	NO	Vanillic acid demethylation 1	Vanillic acid + THF => 5-Methyltetrahydrofolate + PCA
native	A008	NO	Vanillic acid demethylation 2	5-Methyltetrahydrofolate + NAD+ => NADH + 5,10-Methylenetetrahydrofolate + H+
native	R01220	YES	5,10-methylenetetrahydrofolate:NADP+ oxidoreductase	5,10-Methylenetetrahydrofolate + NADP+ <=> 5,10-Methylenetetrahydrofolate + NADPH
native	R01655	YES	5,10-Methenyltetrahydrofolate 5-hydrolase (decyclizing)	5,10-Methenyltetrahydrofolate + H2O <=> 10-Formyltetrahydrofolate + H+
native	R00944	NO	10-Formyltetrahydrofolate amidohydrolase	10-Formyltetrahydrofolate + H2O => formate + THF
native	A018	NO	Syringic acid demethylation	Syringic acid + THF => 5-Methyltetrahydrofolate + 3-methylgallate
hypothetical	altA007 (replaces A007)	NO	Hypothetical vanillic acid demethylation 1	Vanillic acid + O2 => formate + PCA
hypothetical	altA018 (replaces A018)	NO	Hypothetical syringic acid demethylation	Syringic acid + O2 => formate + 3-methylgallate
<i>P. putida</i>	ppA007 (replaces A007)	NO	<i>P. putida</i> vanillic acid demethylation	Vanillic acid + NADH + O2 + H+ => NAD+ + H2O + formaldehyde + PCA
<i>P. putida</i>	ppA018 (replaces A018)	NO	<i>P. putida</i> syringic acid demethylation	Syringic acid + NADH + O2 + H+ => NAD+ + H2O + formaldehyde + 3-methylgallate

**Table S3.** Reactions added to iNovo479 to simulate engineered strains capable of producing non-native chemicals.

Reaction ID	Reversibility	Name	Bioproduct	Reaction
engR01704	Reversible	Palmitaldehyde:NAD+ oxidoreductase	1-hexadecanol	Hexadecanal + NAD+ + H2O $\leftrightarrow$ Hexadecanoic acid + NADH + H+
engR02462	Reversible	Hexadecanol:NAD+ oxidoreductase	1-hexadecanol	1-Hexadecanol + NAD+ $\leftrightarrow$ Hexadecanal + NADH + H+
engR01976	Reversible	(S)-3-Hydroxybutanoyl-CoA:NADP+ oxidoreductase	Butanoic acid	(S)-3-Hydroxybutanoyl-CoA + NADP+ $\leftrightarrow$ Acetoacetyl-CoA + NADPH + H+
engR03026	Reversible	(S)-3-hydroxybutanoyl-CoA hydro-lyase	Butanoic acid	(S)-3-Hydroxybutanoyl-CoA $\leftrightarrow$ Crotonoyl-CoA + H2O
engR01171	Reversible	butanoyl-CoA:NAD+ trans-2-oxidoreductase	Butanoic acid	Butanoyl-CoA + NAD+ $\leftrightarrow$ Crotonoyl-CoA + NADH + H+
engR01179	Reversible	butanoyl-CoA:acetate CoA-transferase	Butanoic acid	Butanoyl-CoA + Acetate $\leftrightarrow$ Butanoic acid + Acetyl-CoA
engR00449	Irreversible	L-lysine:oxygen 2-oxidoreductase (decarboxylating)	Glutarate	L-Lysine + Oxygen $\Rightarrow$ 5-Aminopentanamide + CO2 + H2O
engR02273	Irreversible	5-aminopentanamide amidohydrolase	Glutarate	5-Aminopentanamide + H2O $\Rightarrow$ 5-Aminopentanoate + Ammonia
engR02274	Reversible	5-aminopentanoate:2-oxoglutarate aminotransferase	Glutarate	5-Aminopentanoate + 2-Oxoglutarate $\leftrightarrow$ 5-Oxopentanoate + L-Glutamate
engR02401	Reversible	glutarate-semialdehyde:NAD+ oxidoreductase	Glutarate	5-Oxopentanoate + NAD+ + H2O $\leftrightarrow$ Glutarate + NADH + H+
engR00841	Irreversible	sn-glycerol-3-phosphate phosphohydrolase	Glycerol	sn-Glycerol 3-phosphate + H2O $\Rightarrow$ Glycerol + Orthophosphate
engR00728	Reversible	L-tyrosine phenol-lyase (deaminating; pyruvate-forming)	Phenol	L-Tyrosine + H2O $\leftrightarrow$ Phenol + Pyruvate + Ammonia

**Table S4.** Kinetic constants used to simulate maximum allowed substrate uptake rates

	Half-saturation constant (K <sub>s</sub> )(mmol/L)	Substrate velocity (V <sub>max</sub> )(mmol/min)	Inhibition constant (K <sub>i</sub> )(mmol/L)
Non-aromatic carbon substrates	0.139	0.500	0.139
S-type aromatics	0.050	0.582	0.050
H-type aromatics	0.050	0.902	0.050
G-type aromatics	0.100	0.569	0.100
Ammonia	0.100	0.500	0.100
Phosphate	0.002	0.060	0.100
Sulfate	0.003	0.002	0.100
Iron	0.003	0.002	0.100