

Article

Development of Cd (II) Ion Probe Based on Novel Polyaniline-Multiwalled Carbon Nanotube-3-Aminopropyltriethoxysilane Composite

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Abstract: Cadmium belongs to the group of potentially toxic metals that have high health and environmental significance. Due to its adverse effects on the environment, this study develops an effective electrochemical sensor for detecting a polyaniline-multiwalled carbon nanotube-3-aminopropyltriethoxysilane (PANI-MWCNT-APTES) substrate cast on the GCE. The as-prepared PANI-MWCNT-APTES was prepared by a wet chemical method, and its formation was investigated using several techniques. As a result, the prepared material exhibited a limit of detection of 0.015 μM for cadmium ions (Cd^{2+}) in the linear dynamic range of 0.05 μM to 50 μM . Furthermore, the PANI-MWCNT-APTES-modified GCE current response was stable, repeatable, reproducible, and short. In addition, PANI-MWCNT-APTES/GCE was harnessed for the first time for cadmium detection in real water samples, and the result was satisfactory. Therefore, the recorded results suggest that the newly designed PANI-MWCNT-APTES is a promising material for detecting Cd in the near future for human health and environmental protection.

Keywords: linear sweep voltammetry; heavy metal detection; environmental pollution; cadmium ion detection; electrochemical sensors; potential toxic metals

S1. Electrical resistivity.

The electrical conductivity of PANI-MWCNTs/APTES was determined using the following equations. ³⁵

$$\rho = \rho_0 / G^2 \quad (W/S) \quad (1)$$

$$G^2 / (W/S) = (2S / W) \ln 2 \quad (2)$$

$$\rho_0 = V / I \times 2\pi S \quad (3)$$

$$\sigma = \frac{1}{\rho} \quad (4)$$

where ρ is the corrected resistivity (ohm-cm); ρ_0 is the uncorrected resistivity (ohm-cm); G_7 (W/S) is the correction factor used for the case with no conducting bottom surface, which is a function of W , the thickness of the sample under test (cm); S is the probe spacing (cm); I is the current (A); V is the voltage (V); and σ is the dc electrical conductivity ($S\text{ cm}^{-1}$).

Table S1. The results of electrical conductivity measurement.

PANI%	Electrical resistivity ($\Omega\text{ Cm}$)	Electrical conductivity (σ)
3	2.445	0.4089 S/cm
5	2.1268	0.470 S/cm
7	1.63	0.613 S/cm

Electrochemical Impedance Spectroscopy



Figure S1. EIS spectrum for PANI-MWCNT-APTES-modified GCE.

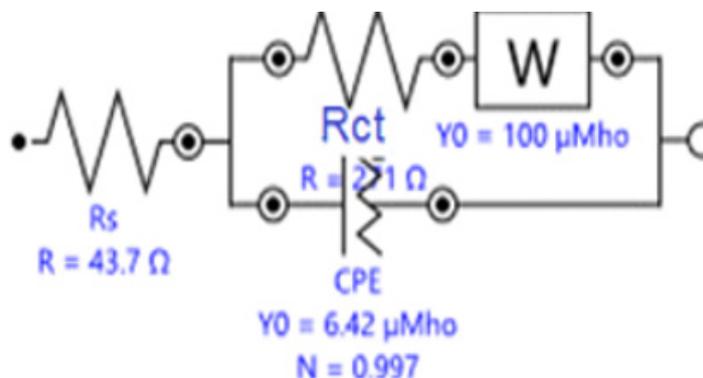


Figure S2. EIS spectrum for bare GCE.

Scheme 2. Analytical performance parameters.

Limit of Detection (LOD)

The LOD was calculated using the following equation:

$$\text{LOD} = 3 \times \text{Sd}/b$$

where S_d is the standard deviation of the blank response and b is the calibration plot's slope.

From our results, $\text{LOD} = 3 \times 0.0015 / (0.29) = 0.0155\ \mu\text{M}$.

Limit of Quantification (LOQ)

The LOQ was determined using the following equation:

$$\text{LOQ} = 10 \times \text{Sd}/\text{b}$$

$$\text{LOQ} = 10 \times 0.0015 / (0.29) = 0.0517 \mu\text{M}$$

Sensitivity

The sensitivity was calculated using the following formula:

$$\text{Sensitivity} = 0.296 \mu\text{A}\mu\text{M}^{-1} / 0.07 \text{ cm}^2 = 4.23 \mu\text{A}\mu\text{M}^{-1}\text{cm}^{-2}$$