

Supplementary Materials

Effect of alkali metal (Li, Na, K,) on Ni/CaO dual functional materials for integrated CO₂ capture and hydrogenation

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Table S1. Weak, medium, strong and total basicity of the reduced Ni/CaO and M-Ni/CaO (M = Li, Na, K) DFMs.

Materials	Weak basicity ($\mu\text{mol/g}$)	Medium basicity ($\mu\text{mol/g}$)	Strong basicity ($\mu\text{mol/g}$)	Total basicity ($\mu\text{mol/g}$)
Ni/CaO	55.7	462.1	248.9	766.7
Li-Ni/CaO	120.5	321.0	131.2	572.7
Na-Ni/CaO	135.1	547.1	304.9	987.1
K-Ni/CaO	102.8	493.5	283.2	879.5

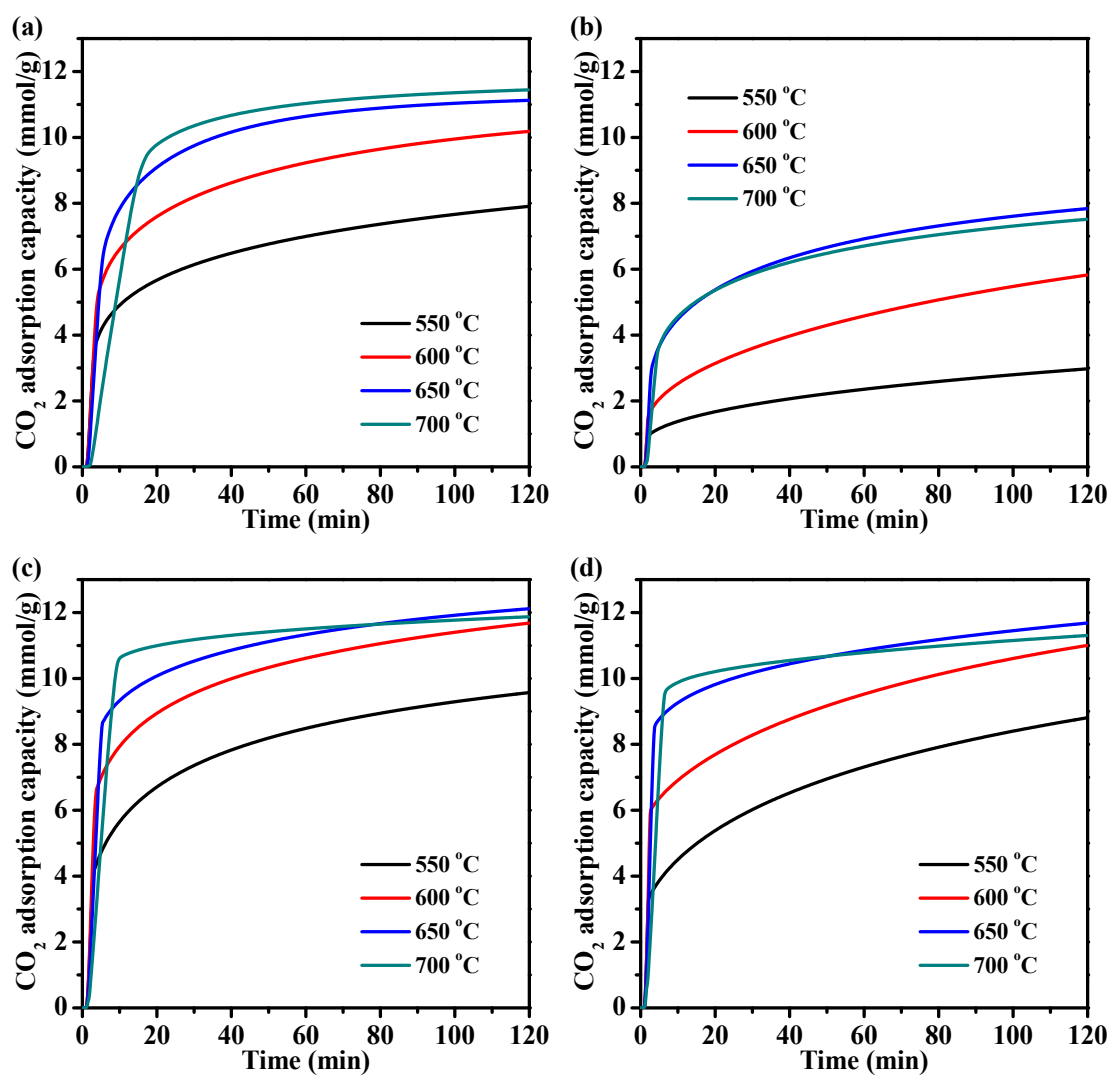


Figure S1. CO₂ capture profiles in the temperature range of 550 to 700 °C for 2 h. (a) Ni/CaO, (b) Li-Ni/CaO, (c) Na-Ni/CaO and (d) K-Ni/CaO.

Table S2. The CO₂ capture capacities of the M-Ni/CaO (M = Li, Na, K) DFMs at the temperature range of 550 – 700 °C for 2 h.

Materials	CO ₂ adsorption capacity (mmol/g)				
	Theoretical value	550 °C	600 °C	650 °C	700 °C
Ni/CaO	17.4	7.9	10.2	11.1	11.4
Li-Ni/CaO	16.5	3.0	5.8	7.8	7.5
Na-Ni/CaO	16.5	9.6	11.7	12.1	11.9
K-Ni/CaO	16.5	8.8	11.0	11.7	11.3

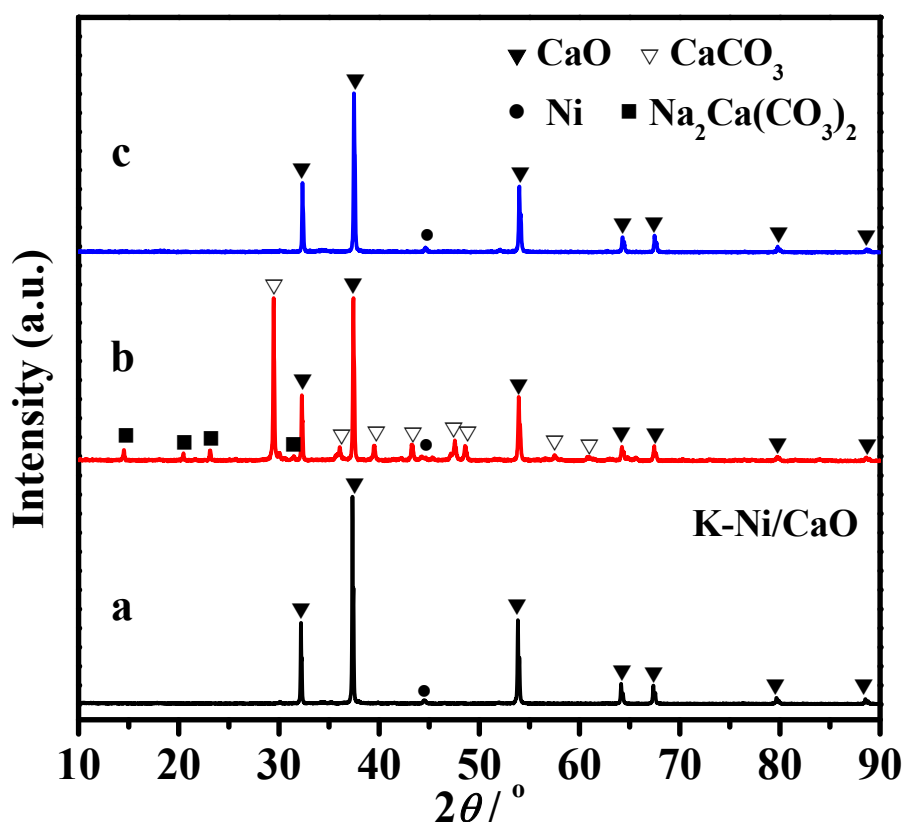


Figure S2. XRD pattern of the Na-Ni/CaO with 15 wt% Na doping after (a) the reduction; (b) the CO₂ capture stage and (c) the integrated CO₂ capture and hydrogenation.

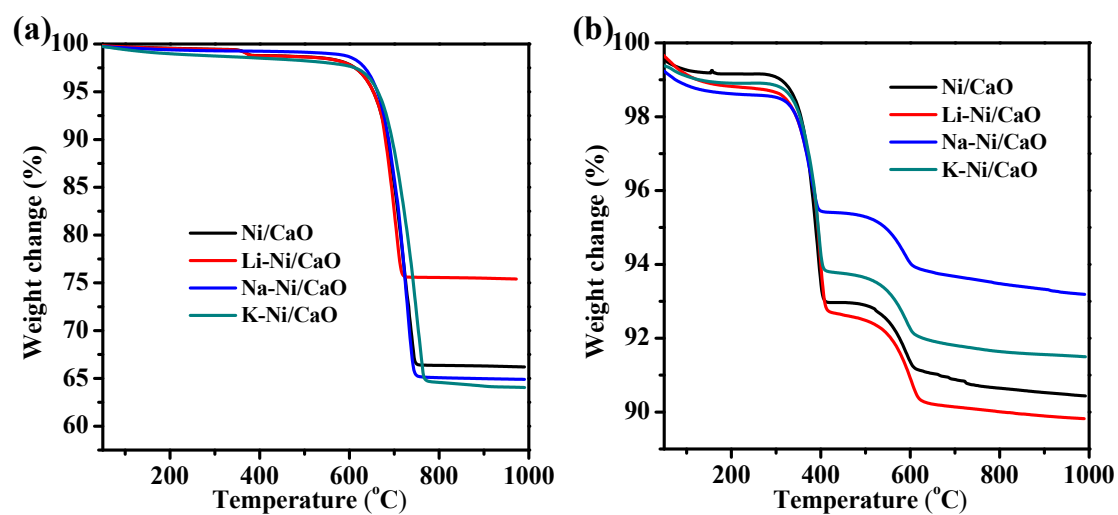


Figure S3. TG profiles of all DFMs (a) after the CO₂ capture stage and (b) the hydrogenation stage.

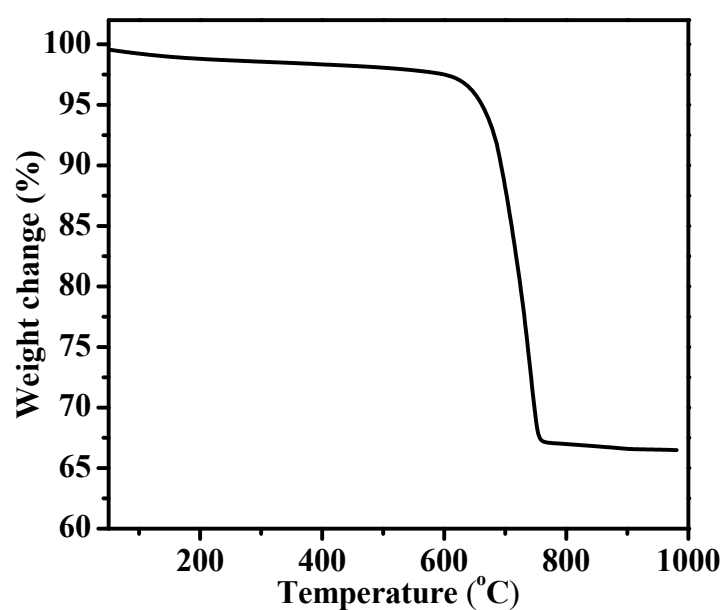


Figure S4. TG profile of the spent Na-Ni/CaO after adsorption in CO₂/Ar for 1 h.

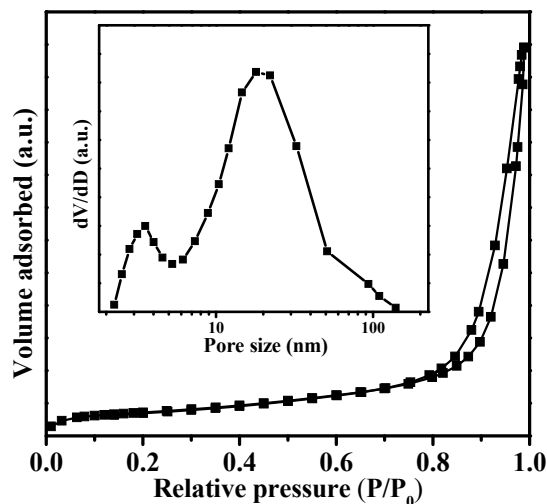


Figure S5. N₂ adsorption-desorption isotherms and pore size distribution (insert part) of spent Na-Ni/CaO DFM after 20 cycles of integrated CO₂ capture and hydrogenation.

Table S3. The combined CO₂ capture and hydrogenation performance over our proposed Na-Ni/CaO DFM and recently documented Ni/CaO based DFMs.

DFMs	Preparation method	Reaction	Temperature (°C)	Feed composition CO ₂ capture + conversion	CO ₂ capacity (mmol/g)	Product yield (mmol/g)	Ref.
Fe ₅ Co ₅ Mg ₁₀ CaO	One-pot sol-gel	RWGS	650	10% CO ₂ + 100% H ₂	9.2	8.28	1
15NiCa	Wet impregnation	Methanation	520	10% CO ₂ + 10% H ₂	-	0.14	2
15NiNa					-	0.19	
Ca ₁ Ni _{0.1}	Sol-gel	RWGS	650	15% CO ₂ + 5% H ₂	15.0	6.9	3
Ca ₁ Ni _{0.1} Ce _{0.033}		RWGS	650	15% CO ₂ + 5% H ₂	14.1	7.3	3
Ni/CaO		Methanation	500	10% CO ₂ /10% H ₂ O + 10% H ₂	8.96	8.34	4
Ni-CaO/γ-Al ₂ O ₃	impregnation	Methanation	320	9.5% CO ₂ + 10% H ₂	0.31	0.14	5
1%NiCaO	One-pot				9.2	2.0	
1%Ni/CeO ₂ -CaO-phy	physical mixing	Methanation	550	15% CO ₂ + 100%	15.3	8.0	6
Ni/CS-P30-C	Sol-gel	RWGS	650	10% CO ₂ + 5% H ₂	13.86	5.52	7
Ni/CS-P30-C-P					8.95	5.33	
Ni/CeO ₂ -CaO	physical mixture	RWGS	650	10% CO ₂ + 5% H ₂	9.6	2.7	8
Ni/CaO	Precipitation-combustion	RWGS	650	10% CO ₂ + 10% H ₂	11.1	6.3	This work
Na-Ni/CaO					12.0	7.0	

1. Shao, B.; Hu, G.; Alkebsi, K.A.M.; Ye, G.; Lin, X.; Du, W.; Hu, J.; Wang, M.; Liu, H.; Qian, F. Heterojunction-redox catalysts of $\text{Fe}_x\text{Co}_y\text{Mg}_{10}\text{CaO}$ for high-temperature CO_2 capture and in situ conversion in the context of green manufacturing. *Energy Environ Sci.* **2021**, *14*, 2291–2301.
2. Bermejo-López, A.; Pereda-Ayo, B.; González-Marcos, J. A.; González-Velasco, J. R. Ni loading effects on dual function materials for capture and in-situ conversion of CO_2 to CH_4 using CaO or Na_2CO_3 . *J. CO₂ Util.* **2019**, *34*, 576–587.
3. Sun, H.; Wang, J.; Zhao, J.; Shen, B.; Shi, J.; Huang, J.; Wu, C. Dual functional catalytic materials of Ni over Ce-modified CaO sorbents for integrated CO_2 capture and conversion. *Appl. Catal., B* **2019**, *244*, 63–75.
4. Jo, S. B.; Woo, J. H.; Lee, J. H.; Kim, T. Y.; Kang, H. I.; Lee, S. C.; Kim, J. C. A novel integrated CO_2 capture and direct methanation process using Ni/ CaO catal-sorbents. *Sustainable Energy Fuels* **2020**, *4*, 4679–4687.
5. Chai, K. H.; Leong, L. K.; Wong, D. S. H.; Tsai, D. H.; Sethupathi, S. Effect of CO_2 adsorbents on the Ni-based dual-function materials for CO_2 capturing and in situ methanation. *J. Chin. Chem. Soc.* **2020**, *67*, 998–1008.
6. Sun, H.; Wang, Y.; Xu, S.; Osman, A. I.; Stenning, G.; Han, J.; Sun, S.; Rooney, D.; Williams, P. T.; Wang, F.; Wu, C. Understanding the interaction between active sites and sorbents during the integrated carbon capture and utilization process. *Fuel* **2021**, *286*, 119308.
7. Wang, G.; Guo, Y.; Yu, J.; Liu, F. H.; Sun, J.; Wang, X.; Wang, T.; Zhao, C. Ni- CaO dual function materials prepared by different synthetic modes for integrated CO_2 capture and conversion. *Chem. Eng. J.* **2022**, *428*, 132110.
8. Sun, S.; Zhang, C.; Guan, S.; Xu, S.; Williams, P. T.; Wu, C. Ni/support- CaO bifunctional combined materials for integrated CO_2 capture and reverse water-gas shift reaction: Influence of different supports. *Sep. Purif. Technol.* **2022**, *298*, 121604.