

Engineering Cu/NiCu LDH Heterostructure Nanosheet Arrays for Highly-Efficient Water Oxidation

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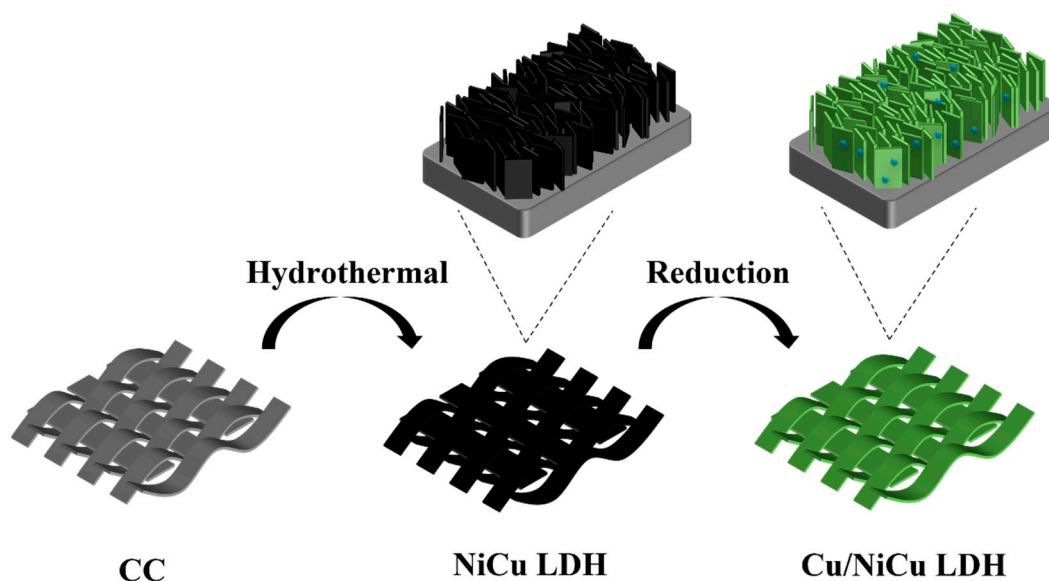


Figure S1. Schematic diagram of the preparation process of Cu/NiCu LDH.

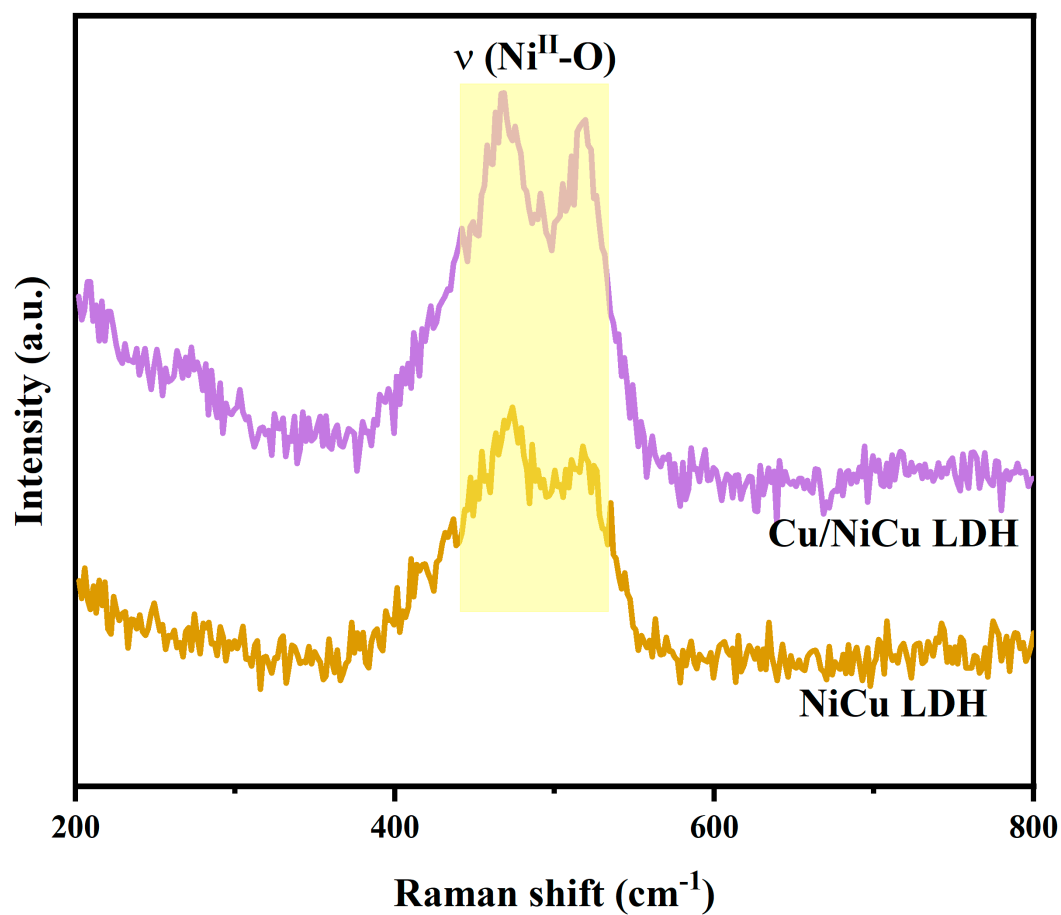


Figure S2. Raman spectrum of the synthesized NiCu LDH and Cu/NiCu LDH.

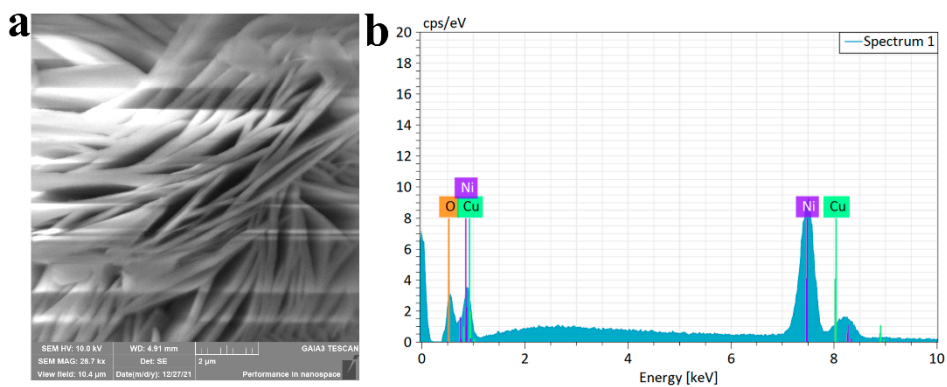


Figure S3. (a) SEM image and (b) EDS spectrum of NiCu LDH.

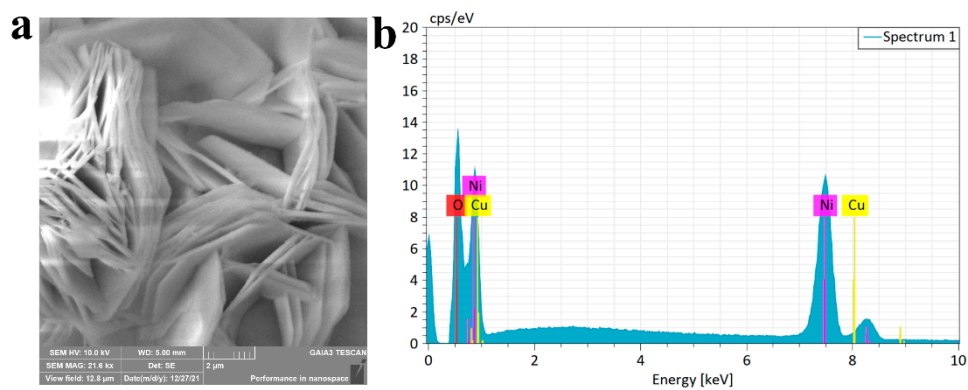


Figure S4. (a) SEM image and (b) EDS spectrum of $\text{Cu}_{1.25}/\text{NiCu}$ LDH.

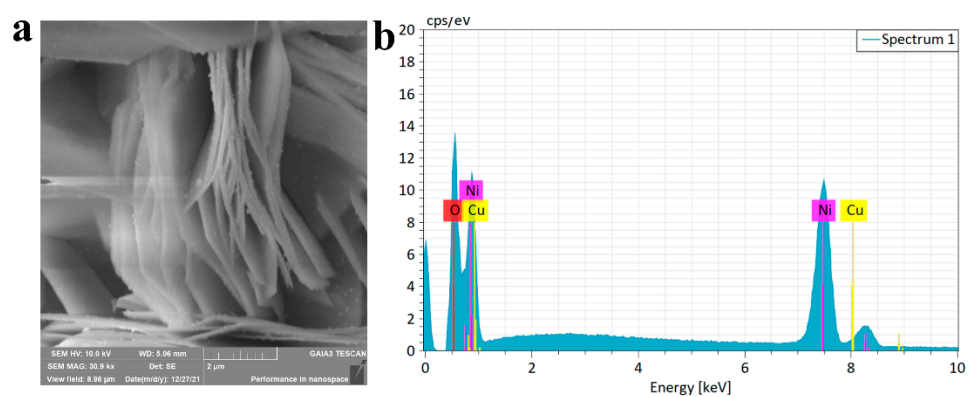


Figure S5. EDS spectrum of $\text{Cu}_{2.5}/\text{NiCu}$ LDH.

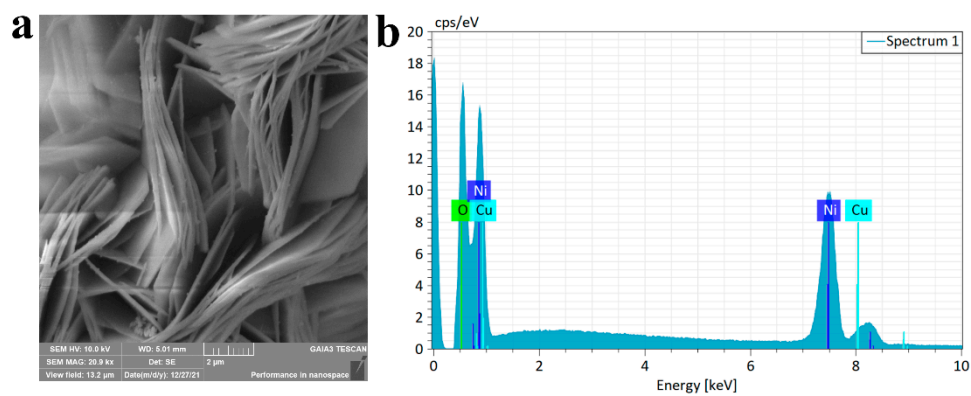


Figure S6. (a) SEM image and (b) EDS spectrum of Cu_5/NiCu LDH.

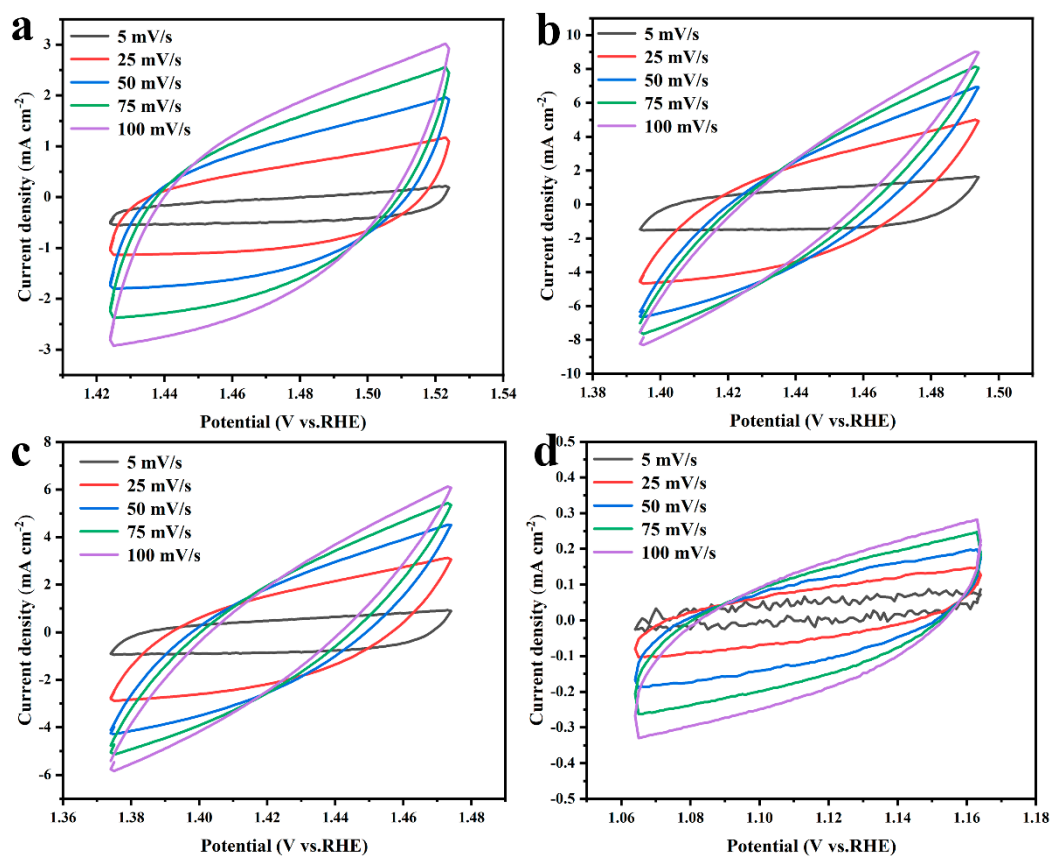


Figure S7. CV curves of (a) NiCu LDH, (b) $\text{Cu}_{1.25}/\text{NiCu}$ LDH, (c) $\text{Cu}_{2.5}/\text{NiCu}$ LDH and (d) Cu_5/NiCu LDH.

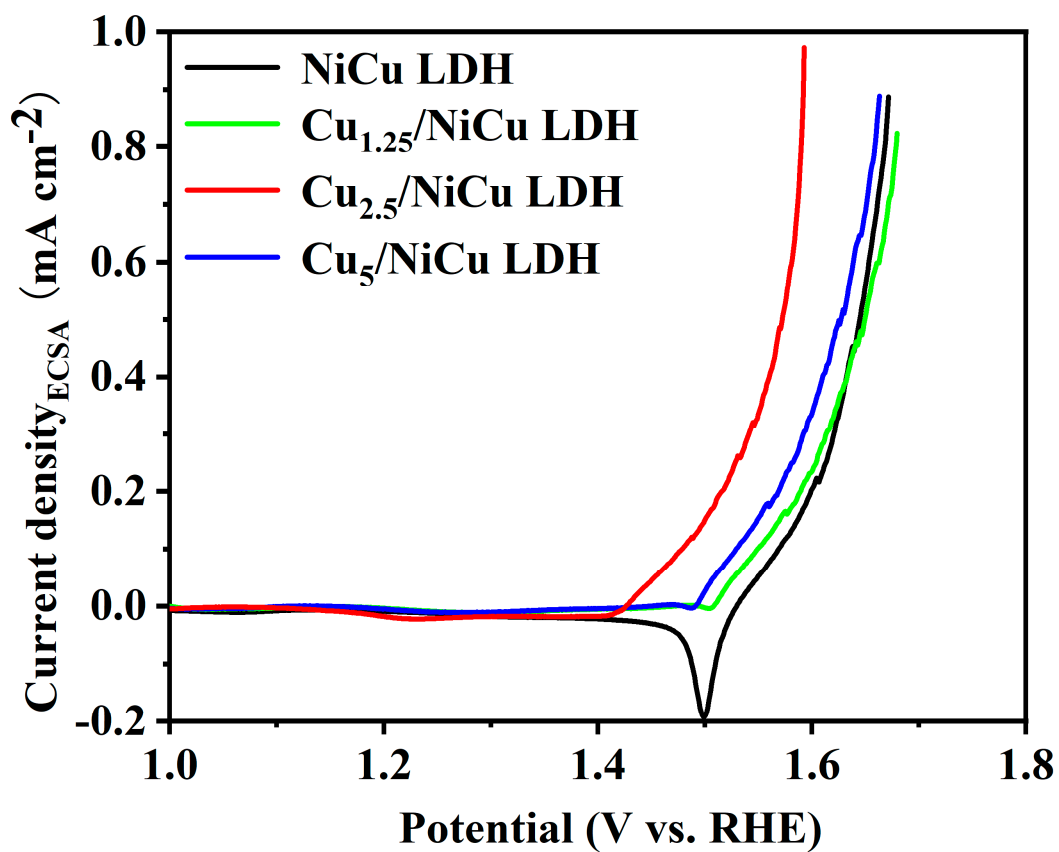


Figure S8. Normalized LSV curves using the ECSA value of Cu/NiCu LDH, Cu_{1.25}/NiCu LDH, Cu_{2.5}/NiCu LDH and Cu₅/NiCu LDH.

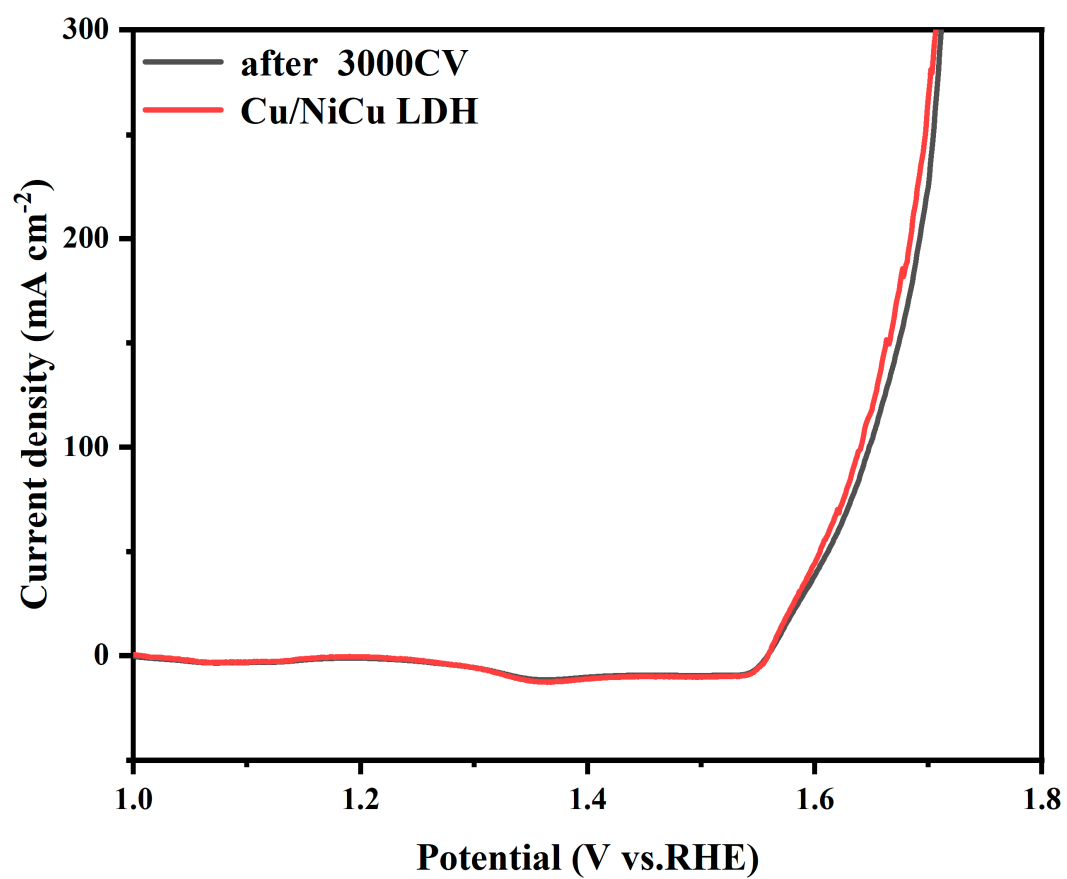


Figure S9. The LSV curves of Cu/NiCu LDH before and after 3000 CV tests.

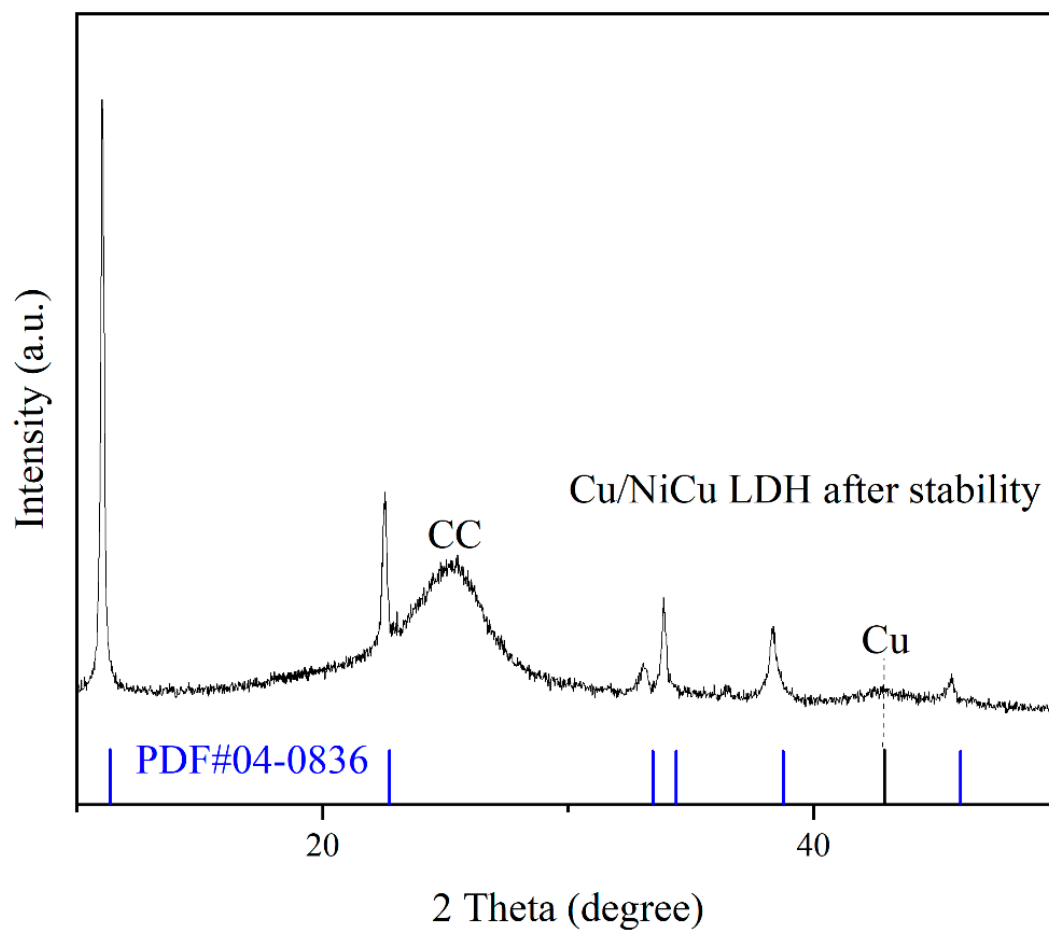


Figure S10. XRD pattern of $\text{Cu}_{2.5}/\text{NiCu}$ LDH after stability test.

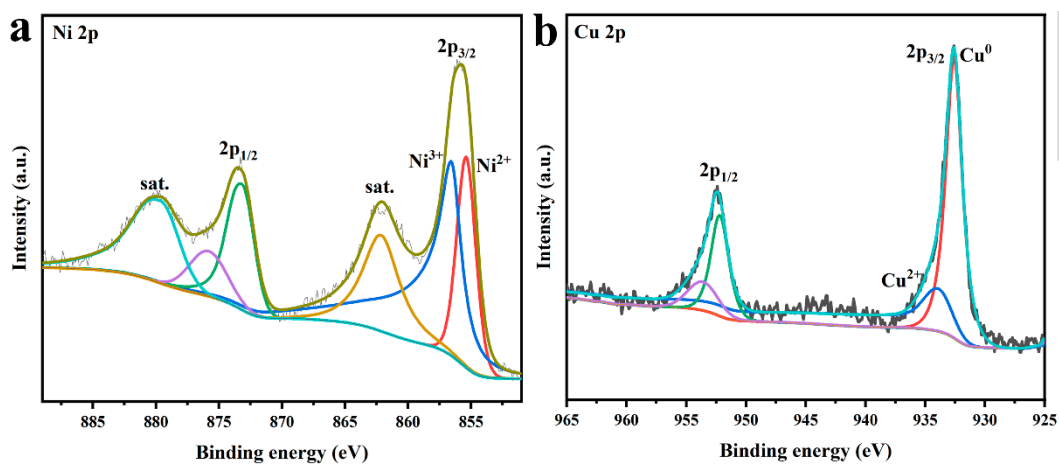


Figure S11. (a) Ni 2p and (b) Cu 2p XPS spectra of $\text{Cu}_{2.5}/\text{NiCu}$ LDH after stability.

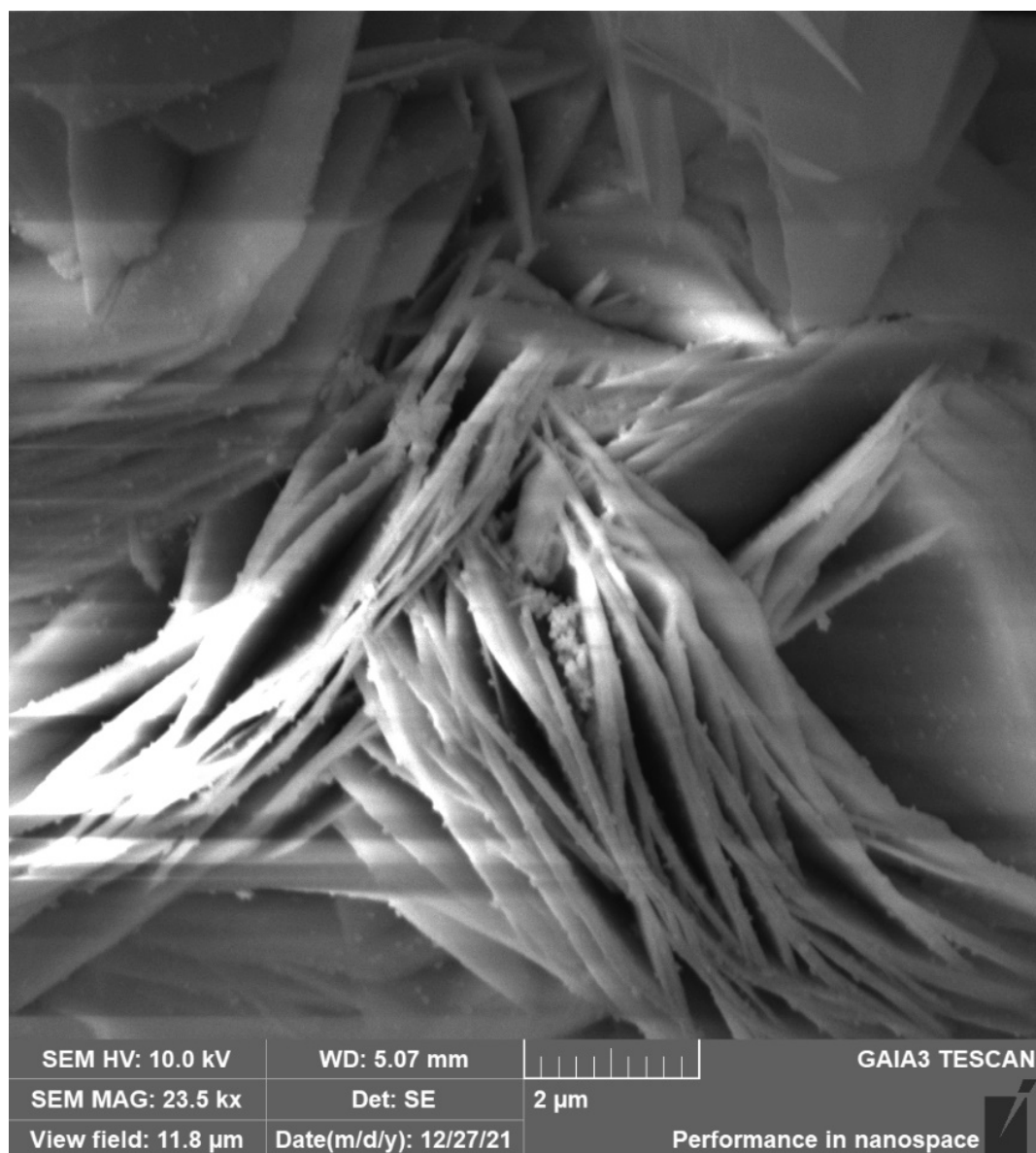


Figure S12. SEM image of $\text{Cu}_{2.5}/\text{NiCu}$ LDH after stability test.

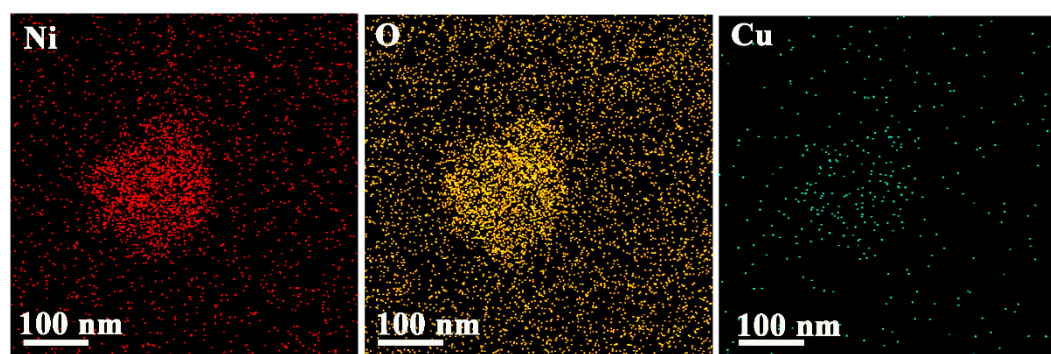


Figure S13. Element mapping of $\text{Cu}_{2.5}/\text{NiCu}$ LDH after stability test.

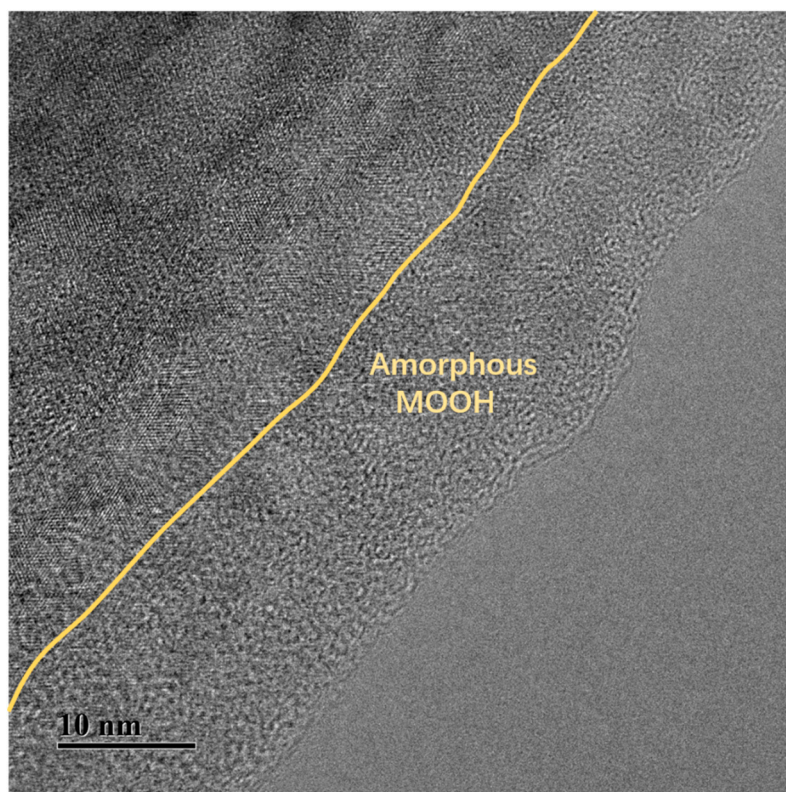


Figure S14. HRTEM image of $\text{Cu}_{2.5}/\text{NiCu}$ LDH after stability test.

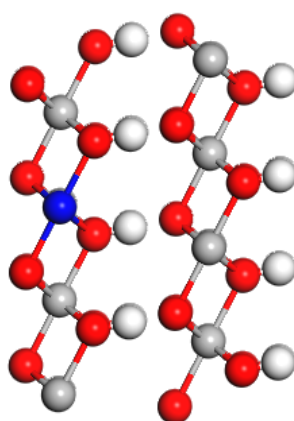


Figure S15. Constructed model of NiCuOOH .

Table S1. EDS results of NiCu LDH, $\text{Cu}_{1.25}/\text{NiCu}$ LDH, $\text{Cu}_{2.5}/\text{NiCu}$ LDH and Cu_5/NiCu LDH.

	Ni (at%)	Cu(at%)	O (at%)
NiCu LDH	34.58	1.08	64.34
$\text{Cu}_{1.25}/\text{NiCu}$ LDH	25.94	1.95	72.10

Cu _{2.5} /NiCu LDH	27.86	2.52	69.62
Cu ₅ /NiCu LDH	30.34	6.14	63.52
Cu _{2.5} /NiCu LDH after stability test	20.7	2.31	76.99

Table S2. Comparison of OER performances of Cu_{2.5}/NiCu LDH with previously reported well-performed OER electrocatalysts.

Catalysts	$\eta@10 \text{ mA cm}^{-2}$ (mV)	Electrolyte	Substrate	Ref.
Cu _{2.5} /NiCu LDH	206	1M KOH	CC	This work
Mxene/NiFe LDH	260	1M KOH	RDE	[1]
GDY/NiFe LDH	260	1M KOH	CF	[2]
Pt/NiFe LDH	230	1M KOH	CC	[3]
Ni ₂ Cr ₁ LDH/NF	319	1M KOH	NF	[4]
S–NiFe LDH/CC	206	1M KOH	CC	[5]
Ni/LDH-ZnO	210	1M KOH	NF	[6]
Exfoliated NiFe LDH	302	1M KOH	GC	[7]
NiO/NiFe LDH	270	1M KOH	GC	[8]
Fe–Ni LDH/MOFs	255	1M KOH	GC	[9]

Glassy carbon (GC); Ni foam (NF); carbon cloth (CC); Cu foam (CF); rotating disk electrode (RDE).

Table S3. ECSA values of NiCu LDH, Cu_{1.25}/NiCu LDH, Cu_{2.5}/NiCu LDH and Cu₅/NiCu LDH.

	NiCu LDH	Cu _{1.25} /NiCu LDH	Cu _{2.5} /NiCu LDH	Cu ₅ /NiCu LDH
ECSA values	324.75	447.75	523.25	415.5

Reference

1. Yu, M.; Zhou, S.; Wang, Z.; Zhao, J.; Qiu, J., Boosting electrocatalytic oxygen evolution by synergistically coupling layered double hydroxide with MXene. *Nano Energy* **2018**, 44, 181-190.
2. Shi, G.; Yu, C.; Fan, Z.; Li, J.; Yuan, M., Graphdiyne-Supported NiFe Layered Double Hydroxide Nanosheets as Functional Electrocatalysts for Oxygen Evolution. *ACS Appl. Mater. Inter.* **2019**, 11, 2662-2669.
3. Anantharaj, S.; Karthick, K.; Venkatesh, M.; Simha, T. V. S. V.; Salunke, A. S.; Ma, L.; Liang, H.; Kundu, S., Enhancing electrocatalytic total water splitting at few layer Pt-NiFe layered double hydroxide interfaces. *Nano Energy* **2017**, 39, 30-43.
4. Ye, W.; Fang, X.; Chen, X.; Yan, D., A three-dimensional nickel-chromium layered double hydroxide micro/nanosheet array as an efficient and stable bifunctional electrocatalyst for overall water splitting. *Nanoscale* **2018**, 10, 19484-19491.
5. Cao, L.-M.; Wang, J.-W.; Zhong, D.-C.; Lu, T.-B., Template-directed synthesis of sulphur doped NiCoFe layered double hydroxide porous nanosheets with enhanced electrocatalytic activity for the oxygen evolution reaction. *J. Mater. Chem. A* **2018**, 6, 3224-3230.
6. Luo, Y.; Wu, Y.; Wu, D.; Huang, C.; Xiao, D.; Chen, H.; Zheng, S.; Chu, P. K., NiFe-Layered Double Hydroxide Synchronously Activated by Heterojunctions and Vacancies for the Oxygen Evolution Reaction. *ACS Appl. Mater. Inter.* **2020**, 12, 42850-42858.
7. Song, F.; Hu, X., Exfoliation of layered double hydroxides for enhanced oxygen evolution catalysis. *Nat. Commun.* **2014**, 5, 4477.
8. Gao, Z.-W.; Liu, J.-Y.; Chen, X.-M.; Zheng, X.-L.; Mao, J.; Liu, H.; Ma, T.; Li, L.; Wang, W.-C.; Du, X.-W., Engineering NiO/NiFe LDH Intersection to Bypass Scaling Relationship for Oxygen Evolution Reaction via Dynamic Tridimensional Adsorption of Intermediates. *Adv. Mater.* **2019**, 31, 1804769.

9. Huo, J.; Wang, Y.; Yan, L.; Xue, Y.; Li, S.; Hu, M.; Jiang, Y.; Zhai, Q.-G., In situsemi-transformation from heterometallic MOFs to Fe-Ni LDH/MOF hierarchical architectures for boosted oxygen evolution reaction. *Nanoscale* **2020**, 12,14514-14523.