

Supplementary Materials

Electrochemical Investigation of Curcumin–DNA Interaction by Using Hydroxyapatite Nanoparticles–Ionic Liquids Based Composite Electrodes

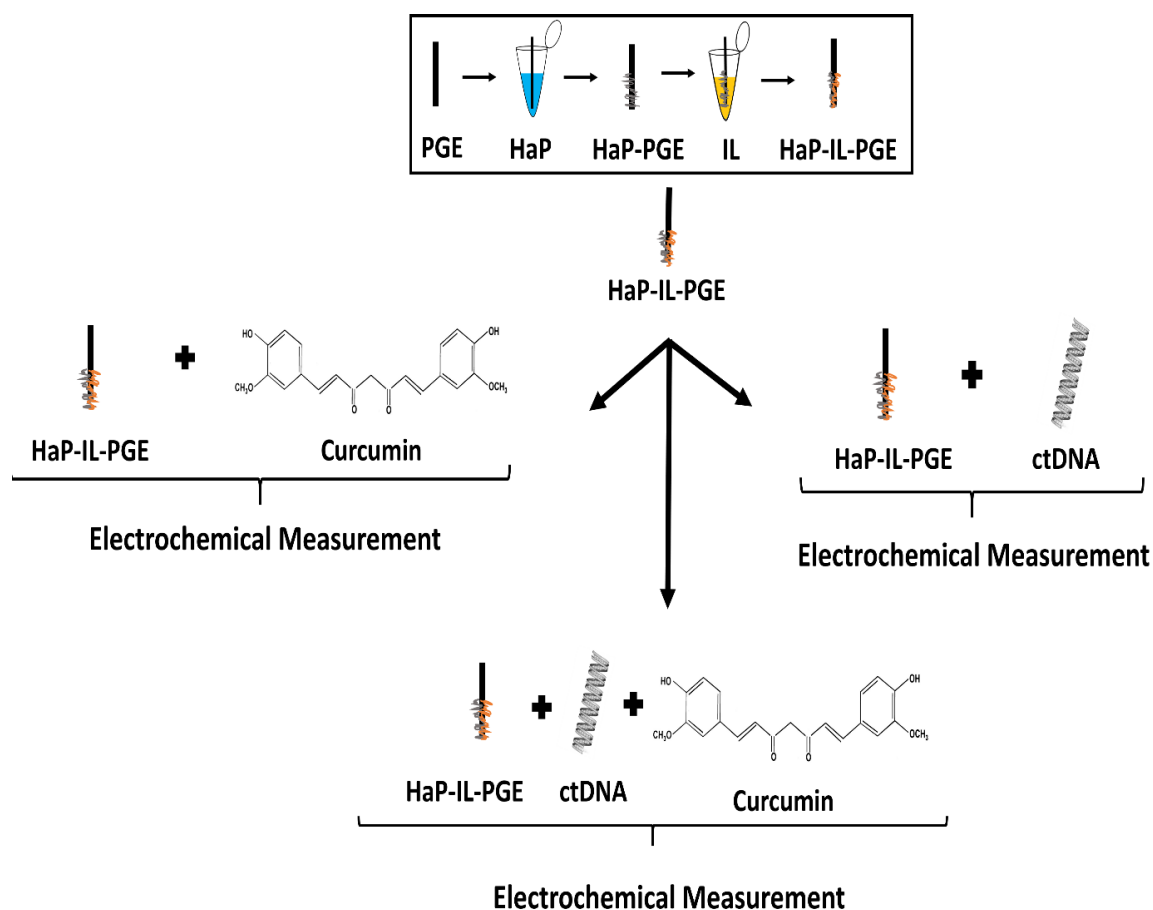
Merve Uca ^{1,2}, Ece Eksin ², Yasemin Erac ³ and Arzum Erdem ^{1,2,*}

¹ Biotechnology Department, Graduate School of Natural and Applied Sciences, Ege University, 35100 Izmir, Turkey; merve.uca@hotmail.com

² Analytical Chemistry Department, Faculty of Pharmacy, Ege University, 35100 Izmir, Turkey; eceksin@hotmail.com

³ Pharmacology Department, Faculty of Pharmacy, Ege University, 35100 Izmir, Turkey; yasemin.erac@ege.edu.tr

* Correspondence: arzum.erdem@ege.edu.tr or arzume@hotmail.com



Scheme S1. The schematic model for the construction of HaP-IL-PGE, immobilization of DNA or Curcumin, and the electrochemical monitoring of Curcumin-ctDNA interaction with HaP-IL-PGEs.

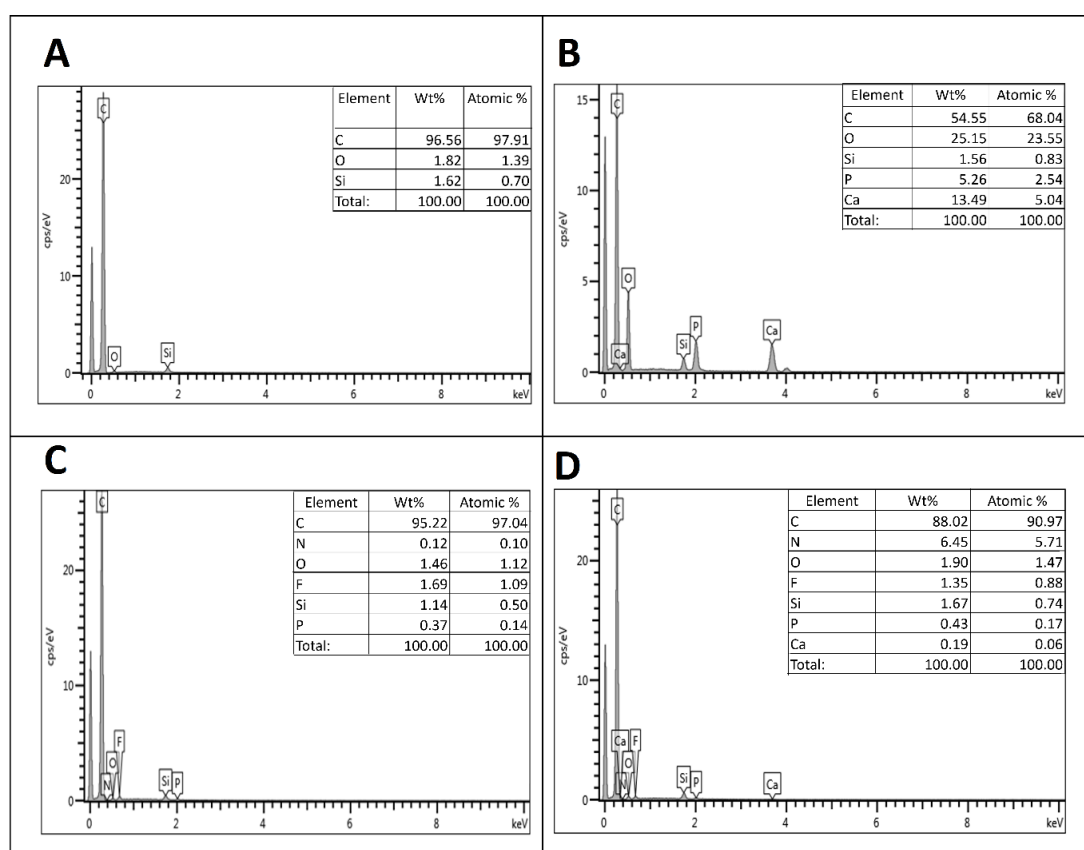


Figure S1. Graphs representing the elemental percentages of (A) PGE, (B) HaP, (C) IL and (D) HaP-IL-PGE obtained by EDX analysis.

Table S1. The average anodic peak currents (I_a) with their calculated surface areas ($n = 3$).

Electrode	I_a (μA)	Surface Area (cm^2)
PGE	87.4 ± 10.3 (RSD % = 11.7 %)	0.27
HaP-PGE	92.2 ± 12.2 (RSD % = 13.21 %)	0.29
IL-PGE	108.7 ± 7 (RSD % = 6.4 %)	0.31
HaP-IL-PGE	115.9 ± 8.7 (RSD % = 7.5 %)	0.35

Table S2. The average R_{ct} values of each of electrodes ($n = 3$) with the decrease % ratio calculated contrast to the one of PGE.

Electrode	R_{ct} (Ohm)	Decrease % at R_{ct}
PGE	116.5 ± 2.1 (RSD % = 1.8 %)	-
HaP-PGE	54.6 ± 14.5 (RSD % = 26.6 %)	58.2%
IL-PGE	33.2 ± 2.6 (RSD % = 8.1 %)	71.5%
HaP-IL-PGE	30.8 ± 2.4 (RSD % = 7.4 %)	73.5%

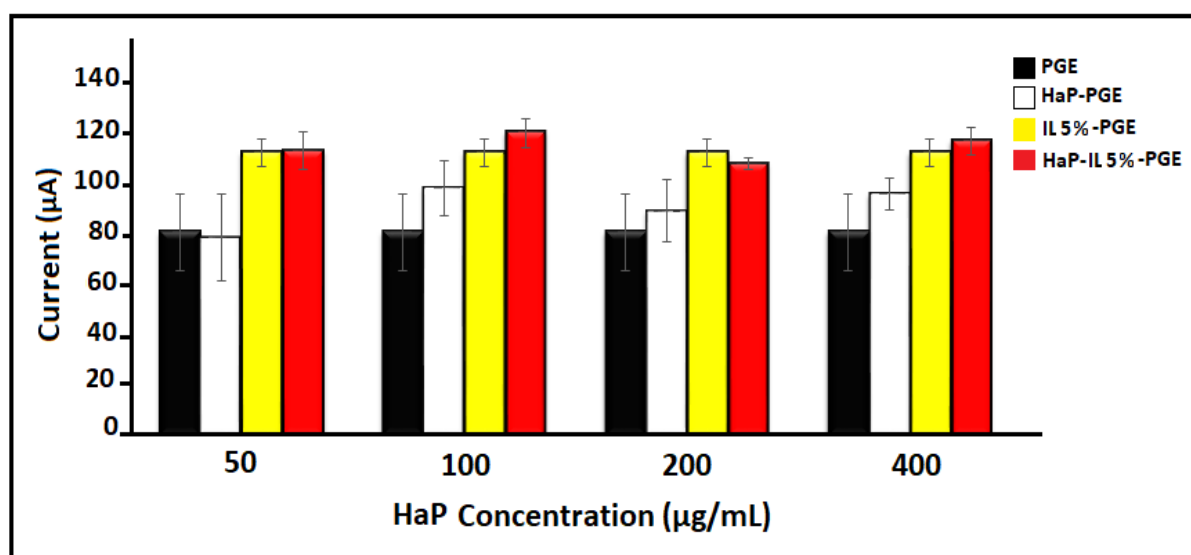


Figure S2. Histograms representing the average oxidation signals measured before and after modification of different HaP concentrations onto the electrode surfaces in the presence of 5 % IL ($n = 13$).

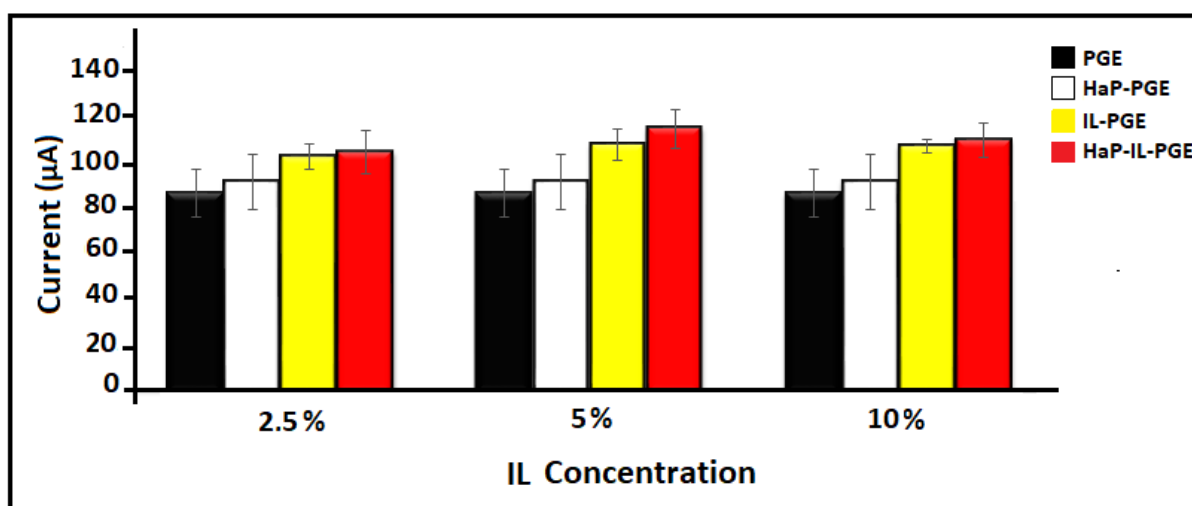


Figure S3. Histograms representing the average oxidation signals measured before and after modification of different IL percentages onto the electrode surfaces in the presence of 100 $\mu\text{g/mL}$ HaP ($n = 9$).

Table S3. The effect of IL % onto the response with the average anodic peak currents (I_a) ($n = 9$).

		IL %	I_a (μA)
Electrode	IL-PGE	2.5 %	103.8 ± 5.7 (RSD % = 5.5 %)
		5 %	108.7 ± 7.0 (RSD % = 6.4 %)
		10 %	108.2 ± 2.9 (RSD % = 2.7 %)
	HaP-IL-PGE	2.5 %	105.6 ± 9.4 (RSD % = 8.9 %)
		5 %	115.9 ± 8.7 (RSD % = 7.5 %)
		10 %	110.9 ± 7.5 (RSD % = 6.8 %)

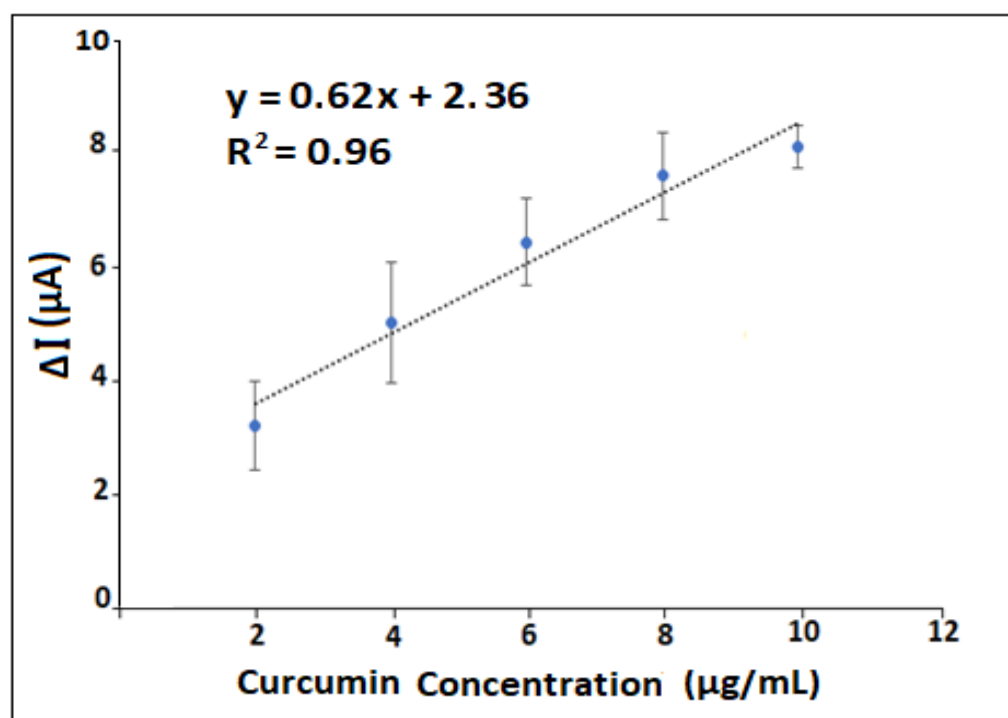


Figure S4. Calibration plot based on the average Curcumin oxidation signals in the presence of various Curcumin concentrations between 2 and 10 μg/mL using HaP-IL-PGEs ($n = 9$).

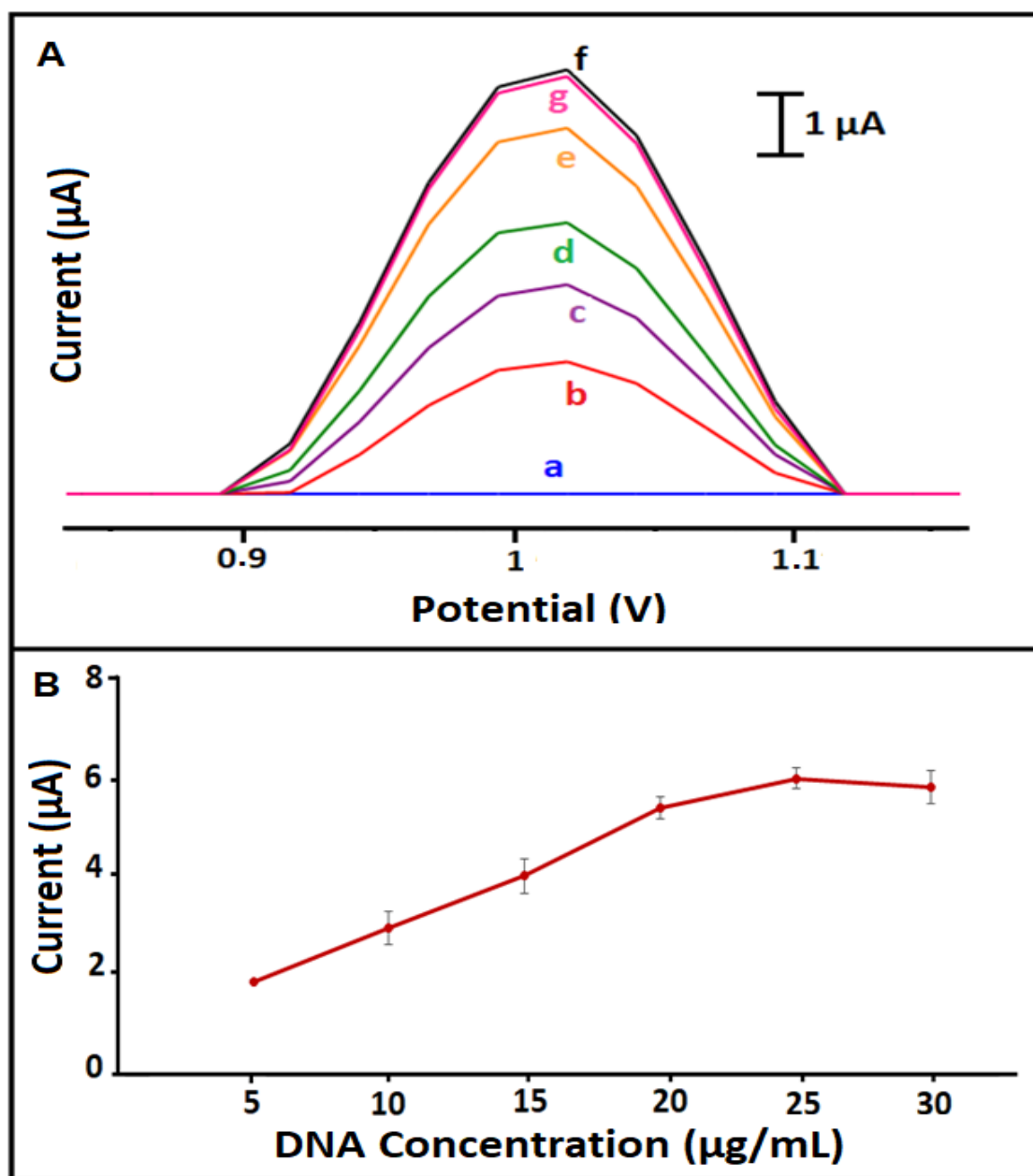


Figure S5. (A) DPVs representing the average guanine signals obtained after immobilization of (a) 5, (b) 10, (c) 15, (d) 20, (e) 25, (f) 30 $\mu\text{g/mL}$ ctDNA onto the surface of HaP-IL-PGEs. (B) The line graph based on the average guanine oxidation signals obtained by HaP-IL-PGE ($n = 3$).

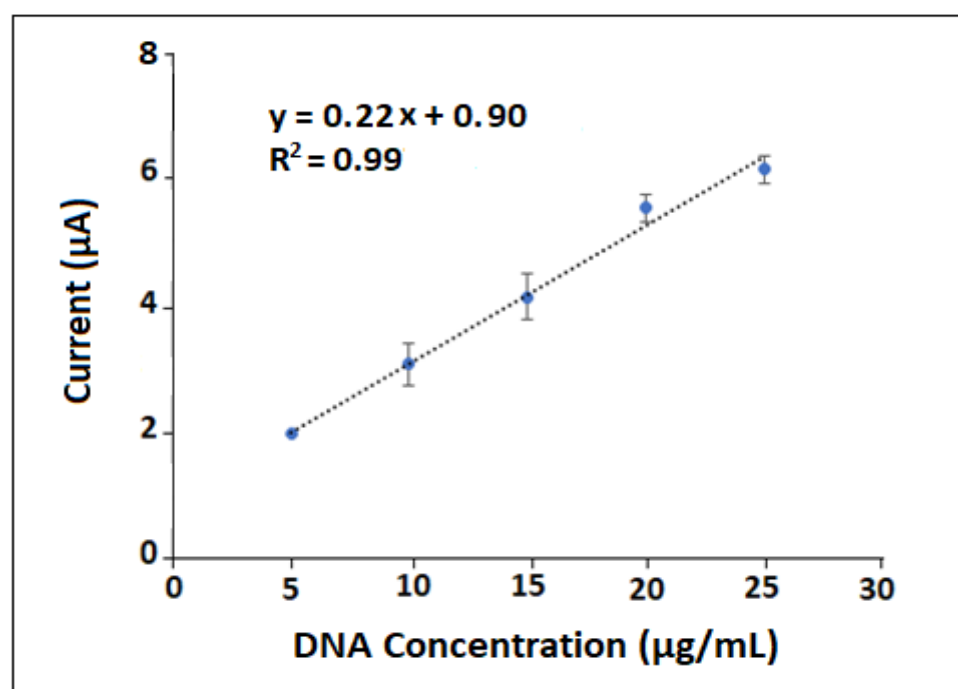


Figure S6. Calibration plot based on the average guanine oxidation signals in the presence of various DNA concentrations between 5 and 25 $\mu\text{g/mL}$ using HaP-IL-PGEs ($n = 3$).

Table S4. The average oxidation signals of Curcumin before and after different interaction times and change ratios after interaction process between Curcumin and ctDNA ($n = 6$).

Interaction time (min)	Curcumin signal before interaction (μA)	Current (I , μA)	Change % at Curcumin signal	Curcumin signal before interaction (μA)	Delta Current (ΔI , μA)*	Change % at Curcumin signal
		Curcumin signal after interaction (μA)			Curcumin signal after interaction (μA)	
1	5.1 ± 0.6	5.2 ± 1.2	3.2 % \uparrow	3.4 ± 0.6	3.6 ± 1.2	4.8 % \uparrow
3	7.3 ± 0.6	6.1 ± 0.7	15.7 % \downarrow	5.6 ± 0.6	4.5 ± 0.7	20.2 % \downarrow
5	8.5 ± 1.4	7.7 ± 1.1	9.1 % \downarrow	6.9 ± 1.4	6.1 ± 1.1	11.3 % \downarrow

*the average signal of control experiment by HaP-IL-PGE was measured at +0.56 V as $1.6 \pm 0.2 \mu\text{A}$ (RSD % = 9.63 %, $n = 6$) that was also overlapping with the oxidation signal of Curcumin measured at +0.56 V. Therefore **Delta Current (ΔI)** was calculated by subtracting the control signal of HaP-IL-PGE from curcumin signal.

Table S5. The average oxidation signals of Curcumin before and after different interaction times and decrease ratios after interaction process between Curcumin and PCR samples ($n = 6$).

Interaction time (min)	Curcumin signal before interaction (μA)	Current (I , μA)	Decrease % at Curcumin signal	Curcumin signal before interaction (μA)	Delta Current (ΔI , μA)*	Decrease % at Curcumin signal
		Curcumin signal after interaction (μA)			Curcumin signal after interaction (μA)	
1	5.5 ± 1.1	5.4 ± 0.4	2.2 %	4 ± 1.1	3.8 ± 0.4	3.1 %
3	8.0 ± 0.8	7.7 ± 0.8	4.7 %	6.5 ± 0.8	6.1 ± 0.8	5.8 %
5	9.3 ± 1.3	9 ± 0.6	3.6 %	7.8 ± 1.3	7.4 ± 0.6	4.3 %

*the average signal of control experiment by HaP-IL-PGE was measured at +0.56 V as $1.6 \pm 0.2 \mu\text{A}$ (RSD % = 9.63 %, $n = 6$) that was also overlapping with the oxidation signal of Curcumin measured at +0.56 V. Therefore **Delta Current (ΔI)** was calculated by subtracting the control signal of HaP-IL-PGE from curcumin signal.

Table S6. The average oxidation signals of guanine before and after different interaction times and change ratios after interaction process between Curcumin and PCR samples ($n = 6$).

Interaction time (min)	Guanine signal before interaction (μA)	Guanine signal after interaction (μA)	Decrease % at Guanine signal
1	4.6 ± 0.3 (RSD % = 7 %)	3.8 ± 0.7 (RSD % = 16.99 %)	16 %
3		3.2 ± 0.2 (RSD % = 4.3 %)	30 %
5		2.6 ± 0.3 (RSD % = 13.5 %)	44.1 %