

## Supporting Information

**Table S1.** Mass loading of Pt in the synthesized  $\text{Pt}_n/\text{Ni(OH)}_2$  NSs.

Samples	Pt loading (wt.%)
$\text{Pt}_5/\text{Ni(OH)}_2$	11.48
$\text{Pt}_{10}/\text{Ni(OH)}_2$	22.01
$\text{Pt}_{15}/\text{Ni(OH)}_2$	29.50
$\text{Pt}_{20}/\text{Ni(OH)}_2$	35.83

**Table S2.** BET specific surface area of synthesized  $\text{Ni(OH)}_2$  NSs and  $\text{Pt}_n/\text{Ni(OH)}_2$  NSs.

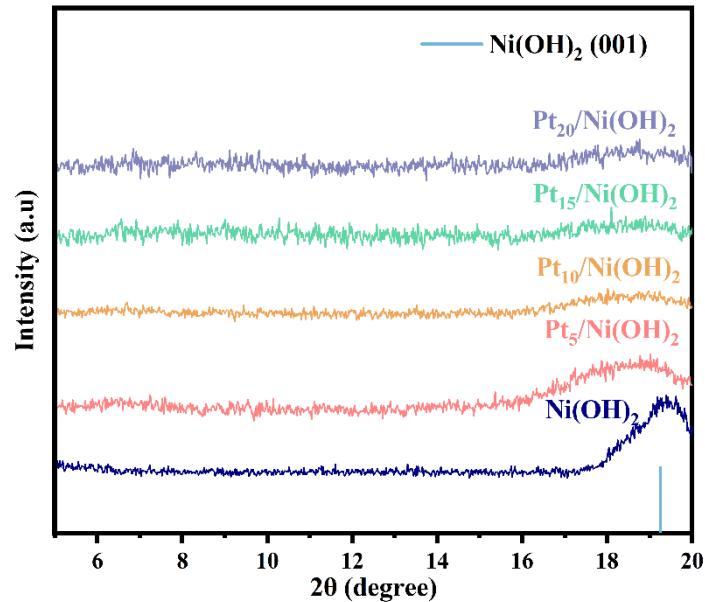
Samples	BET specific surface area( $\text{m}^2/\text{g}$ )
$\text{Ni(OH)}_2$	57.67
$\text{Pt}_5/\text{Ni(OH)}_2$	62.48
$\text{Pt}_{10}/\text{Ni(OH)}_2$	104.47
$\text{Pt}_{15}/\text{Ni(OH)}_2$	87.50
$\text{Pt}_{20}/\text{Ni(OH)}_2$	84.48

**Table S3.** Reaction rate constant  $k$ , mass normalized reaction rate constant  $k_m$  and conversions of 4-NP at 10 min catalyzed by  $\text{Pt}_n/\text{Ni(OH)}_2$  NSs and commercial Pt/C.

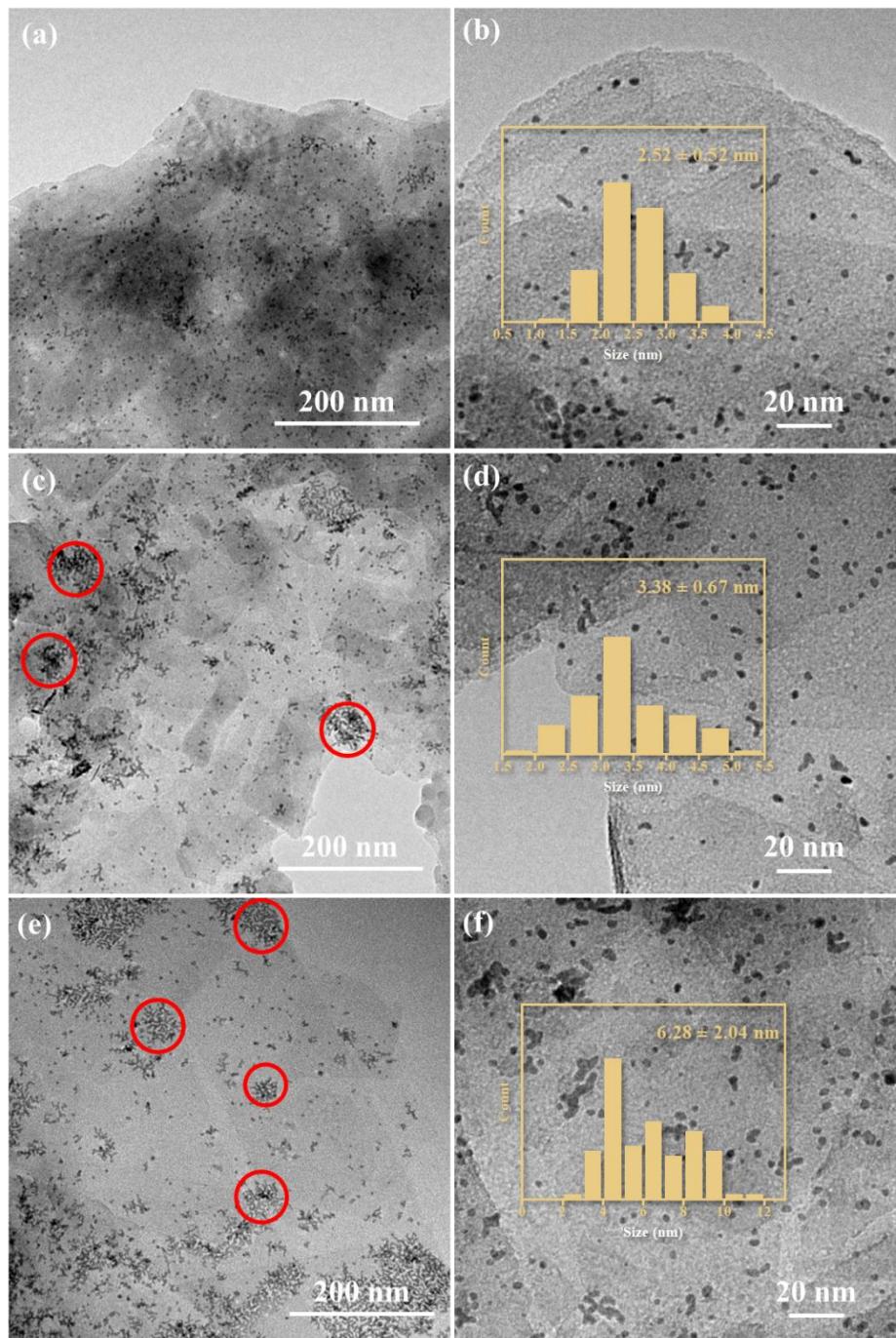
Samples	$k$ ( $\text{s}^{-1}$ )	$k_m$ ( $\text{s}^{-1}\cdot\text{g}^{-1}$ )	Conversion (%)
Pt/C	0.00967	96.70	98.4
$\text{Pt}_5/\text{Ni(OH)}_2$	0.01289	224.56	99.5
$\text{Pt}_{10}/\text{Ni(OH)}_2$	0.02358	214.27	99.8
$\text{Pt}_{15}/\text{Ni(OH)}_2$	0.02294	155.53	99.8
$\text{Pt}_{20}/\text{Ni(OH)}_2$	0.01973	110.13	99.6

**Table S4.** Statistical analysis of conversion of 10 reaction cycles of Pt<sub>n</sub>/Ni(OH)<sub>2</sub> NSs with different Pt loading.

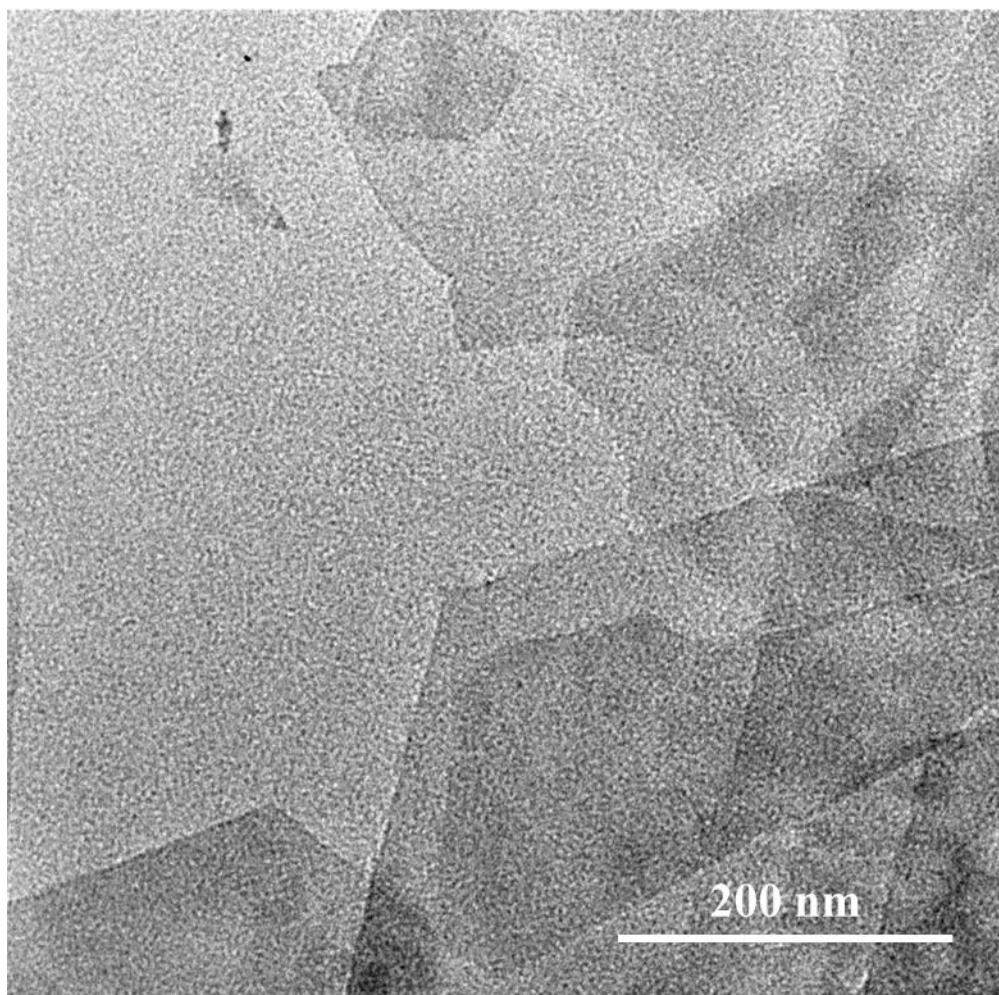
Samples	Mean of conversion	Variance of conversion	Standard deviation of conversion
Pt <sub>5</sub> /Ni(OH) <sub>2</sub>	98.92	0.56	0.75
Pt <sub>10</sub> /Ni(OH) <sub>2</sub>	99.27	0.21	0.46
Pt <sub>15</sub> /Ni(OH) <sub>2</sub>	99.34	0.59	0.77
Pt <sub>20</sub> /Ni(OH) <sub>2</sub>	95.38	7.84	2.80



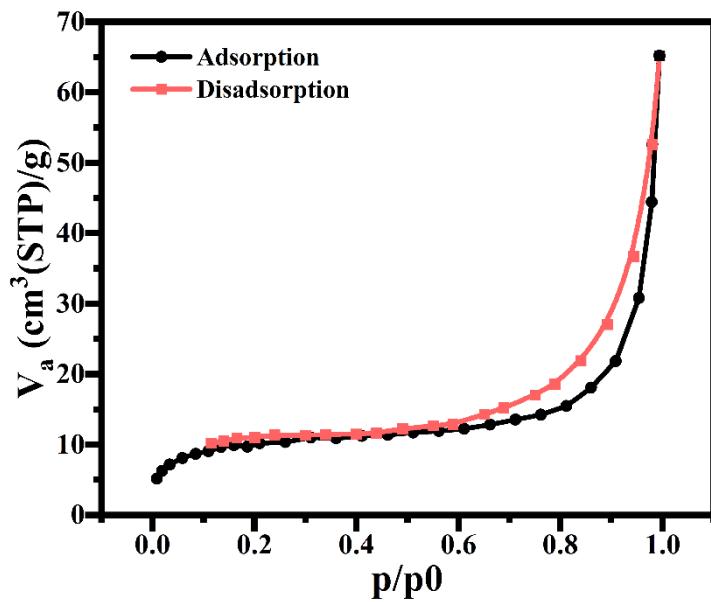
**Figure S1.** Magnified XRD patterns for as-prepared Ni(OH)<sub>2</sub> NSs and Pt<sub>n</sub>/Ni(OH)<sub>2</sub> NSs with different Pt loading.



**Figure S2.** TEM images and particle size distributions of **(a-b)** Pt<sub>5</sub>/Ni(OH)<sub>2</sub> NSs, **(c-d)** Pt<sub>15</sub>/Ni(OH)<sub>2</sub> NSs, and **(e-f)** Pt<sub>20</sub>/Ni(OH)<sub>2</sub> NSs.



**Figure S3.** TEM images of Ni(OH)<sub>2</sub> NSs.



**Figure S4.** Nitrogen adsorption–desorption isotherm of Pt<sub>10</sub>/Ni(OH)<sub>2</sub> NSs.