

Supporting Information

for

Coupling of Thiazole-2-Amines with Isocyanide Ligands in *bis*-(Isocyanide) Platinum

Complex: A New Type of Reactivity

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X-ray Diffraction Details

Table S1. Crystal data and structure refinement for **4a**, **4b**, and **5a**

Identification code	4a (yug102)	4b (yug102-2)	5a (yor019)
Empirical formula	C ₃₉ H ₃₈ Cl ₂ N ₆ Pt ₂ S	C ₄₀ H ₃₉ Cl ₅ N ₆ Pt ₂ S	C ₄₁ H ₄₁ Cl ₅ N ₆ Pt ₂ S
Formula weight	1083.89	1203.26	1217.29
Temperature/K	100(2)	100(2)	100(2)
Crystal system	triclinic	monoclinic	monoclinic
Space group	P-1	P2 ₁ /c	P2 ₁ /c
a/Å	7.6428(2)	8.13876(12)	14.4775(2)
b/Å	8.2736(4)	18.8975(3)	18.7218(3)
c/Å	30.3466(11)	27.7978(4)	15.7573(3)
α/°	86.354(3)	90	90
β/°	88.298(3)	96.0606(13)	94.769(2)
γ/°	84.719(3)	90	90
Volume/Å ³	1906.35(13)	4251.47(11)	4256.15(12)
Z	2	4	4
ρ _{calc} /cm ³	1.888	1.880	1.900
μ/mm ⁻¹	7.562	6.974	15.763
F(000)	1040.0	2312.0	2344.0
Crystal size/mm ³	0.15 × 0.1 × 0.1	0.15 × 0.12 × 0.1	0.2 × 0.15 × 0.15
Radiation	Mo Kα (λ = 0.71073)	MoKα (λ = 0.71073)	CuKα (λ = 1.54184)
2θ range for data collection/°	5.054 to 49.996	5.222 to 54.998	6.126 to 134.996
Index ranges	-9 ≤ h ≤ 9, -8 ≤ k ≤ 9, -36 ≤ l ≤ 31	-10 ≤ h ≤ 10, -24 ≤ k ≤ 24, -36 ≤ l ≤ 36	-17 ≤ h ≤ 17, -22 ≤ k ≤ 21, -18 ≤ l ≤ 18
Reflections collected	12848	86216	28850
Independent reflections	6675 [R _{int} = 0.0338, R _{sigma} = 0.0495]	9760 [R _{int} = 0.0363, R _{sigma} = 0.0188]	7670 [R _{int} = 0.0664, R _{sigma} = 0.0484]
Data/restraints/parameters	6675/0/459	9760/0/495	7670/0/505
Goodness-of-fit on F ²	1.135	1.287	1.038

Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.0337$, $wR_2 = 0.0619$	$R_1 = 0.0278$, $wR_2 = 0.0595$	$R_1 = 0.0425$, $wR_2 = 0.1131$
Final R indexes [all data]	$R_1 = 0.0416$, $wR_2 = 0.0643$	$R_1 = 0.0302$, $wR_2 = 0.0603$	$R_1 = 0.0455$, $wR_2 = 0.1158$
Largest diff. peak/hole / $e \text{ \AA}^{-3}$	2.09/-2.28	1.48/-2.02	1.97/-1.97
Flack parameter	n/a	n/a	n/a
CCDC number	2214271	2212279	2212276

Table S2. Crystal data and structure refinement for **5b**, **6**, and **7**

Identification code	5b (yor21)	6 (asm232)	7 (yor024)
Empirical formula	$C_{41}H_{41}Cl_5N_6Pt_2S$	$C_{25}H_{29}Cl_3N_6OPtS_2$	$C_{30}H_{32}Cl_{13}N_6PtS_2$
Formula weight	1217.29	795.10	1196.67
Temperature/K	100(2)	100(2)	100.01(10)
Crystal system	orthorhombic	monoclinic	orthorhombic
Space group	$P2_12_12_1$	$P2_1/n$	Pbca
$a/\text{\AA}$	8.11247(5)	11.9674(3)	14.2834(3)
$b/\text{\AA}$	20.40385(10)	16.2090(5)	23.4916(4)
$c/\text{\AA}$	26.31527(14)	16.0375(5)	26.8793(4)
$\alpha/^\circ$	90	90	90
$\beta/^\circ$	90	108.421(3)	90
$\gamma/^\circ$	90	90	90
Volume/ \AA^3	4355.85(4)	2951.53(15)	9019.1(3)
Z	4	4	8
$\rho_{\text{calc}}/\text{g cm}^{-3}$	1.856	1.789	1.763
μ/mm^{-1}	15.402	5.198	14.034
$F(000)$	2344.0	1560.0	4680.0
Crystal size/ mm^3	$0.6 \times 0.5 \times 0.3$	$0.1 \times 0.1 \times 0.1$	$0.12 \times 0.12 \times 0.1$
Radiation	$\text{CuK}\alpha$ ($\lambda = 1.54184$)	$\text{MoK}\alpha$ ($\lambda = 0.71073$)	$\text{CuK}\alpha$ ($\lambda = 1.54184$)
2Θ range for data collection/ $^\circ$	5.48 to 139.99	5.11 to 54.996	6.576 to 155.088
Index ranges	$-9 \leq h \leq 9$, $-24 \leq k \leq 24$, $-32 \leq l \leq 32$	$-15 \leq h \leq 12$, $-19 \leq k \leq 21$, $-20 \leq l \leq 20$	$-18 \leq h \leq 17$, $-29 \leq k \leq 29$, $-33 \leq l \leq 18$
Reflections collected	110650	13267	35479
Independent reflections	8247 [$R_{\text{int}} = 0.0653$, $R_{\text{sigma}} = 0.0230$]	6781 [$R_{\text{int}} = 0.0399$, $R_{\text{sigma}} = 0.0737$]	9208 [$R_{\text{int}} = 0.0641$, $R_{\text{sigma}} = 0.0543$]
Data/restraints/parameters	8247/0/505	6781/0/342	9208/0/475
Goodness-of-fit on F^2	1.044	1.027	1.085
Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.0241$, $wR_2 = 0.0580$	$R_1 = 0.0425$, $wR_2 = 0.0783$	$R_1 = 0.0441$, $wR_2 = 0.1161$

Final R indexes [all data]	$R_1 = 0.0248$, $wR_2 = 0.0584$	$R_1 = 0.0694$, $wR_2 = 0.0899$	$R_1 = 0.0526$, $wR_2 = 0.1216$
Largest diff. peak/hole / $e \text{ \AA}^{-3}$	0.84/-1.17	2.53/-1.12	2.02/-2.02
Flack parameter	-0.034(5)	n/a	n/a
CCDC number	2212278	2212275	2212277

NMR Spectra

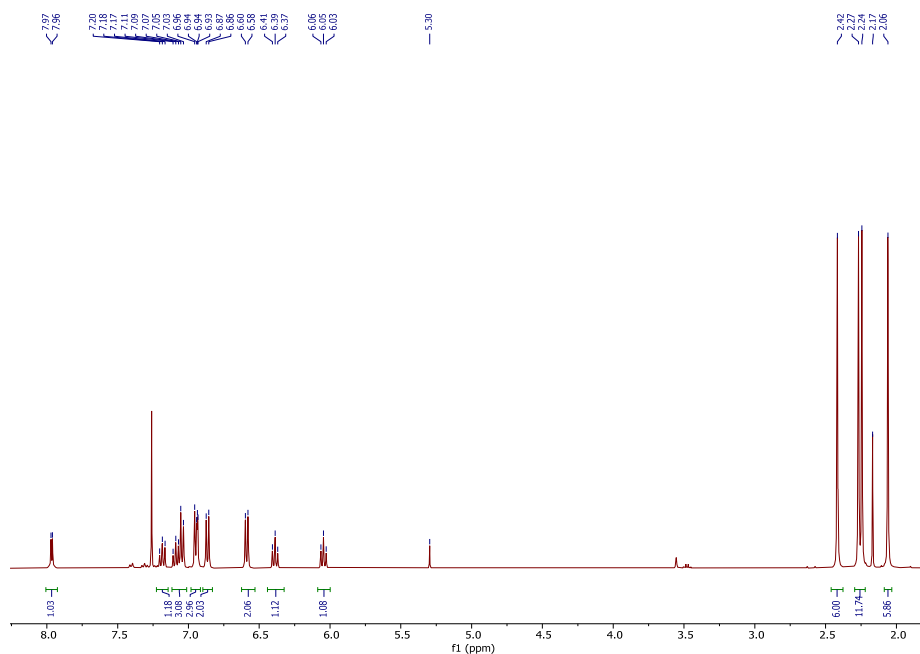


Figure S1. The ^1H NMR spectrum of **4a** in CDCl_3 .

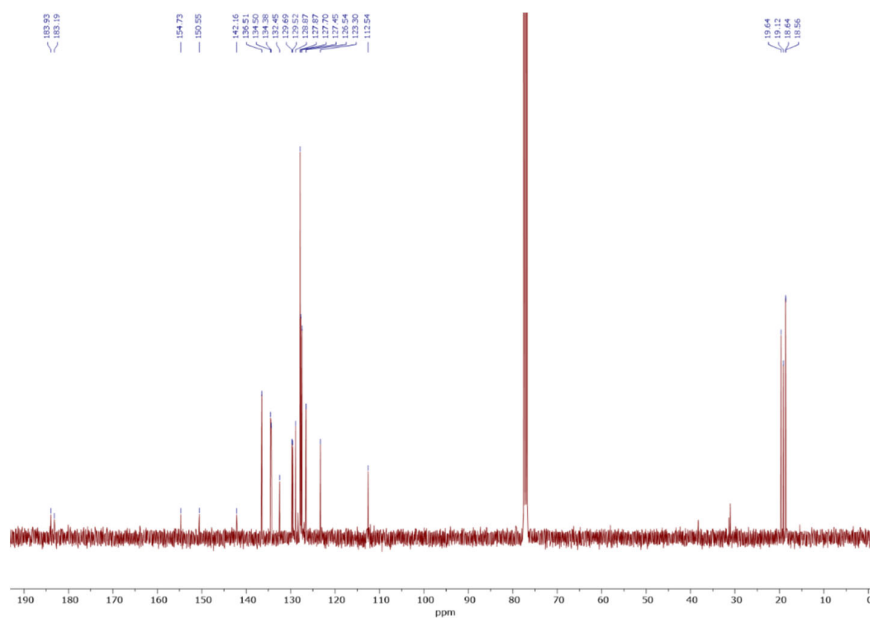


Figure S2. The ^{13}C NMR spectrum of **4a** in CDCl_3 .

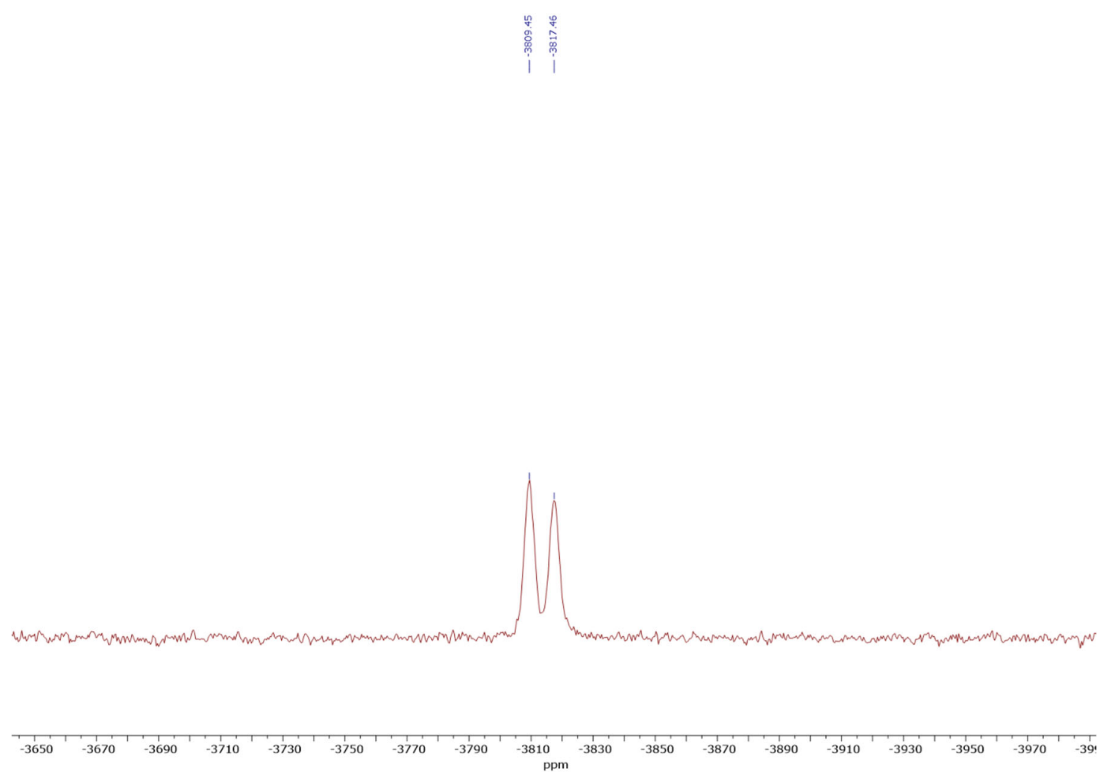


Figure S3. The ^{195}Pt spectrum of **4a** in CDCl_3 .

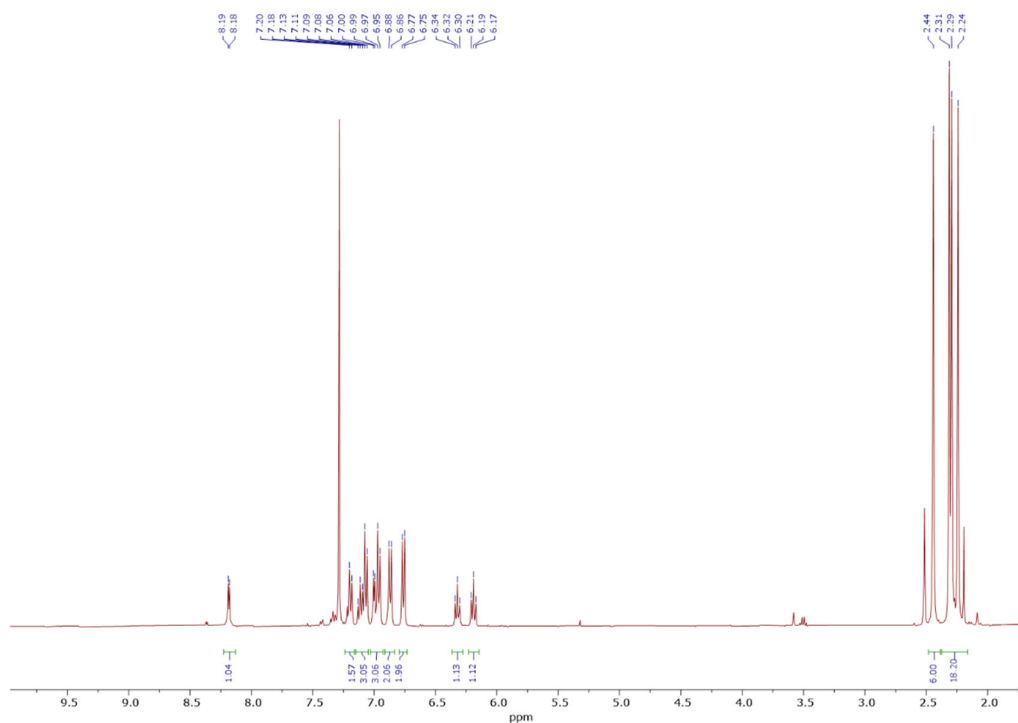


Figure S4. The ^1H NMR spectrum of **4b** in CDCl_3 .

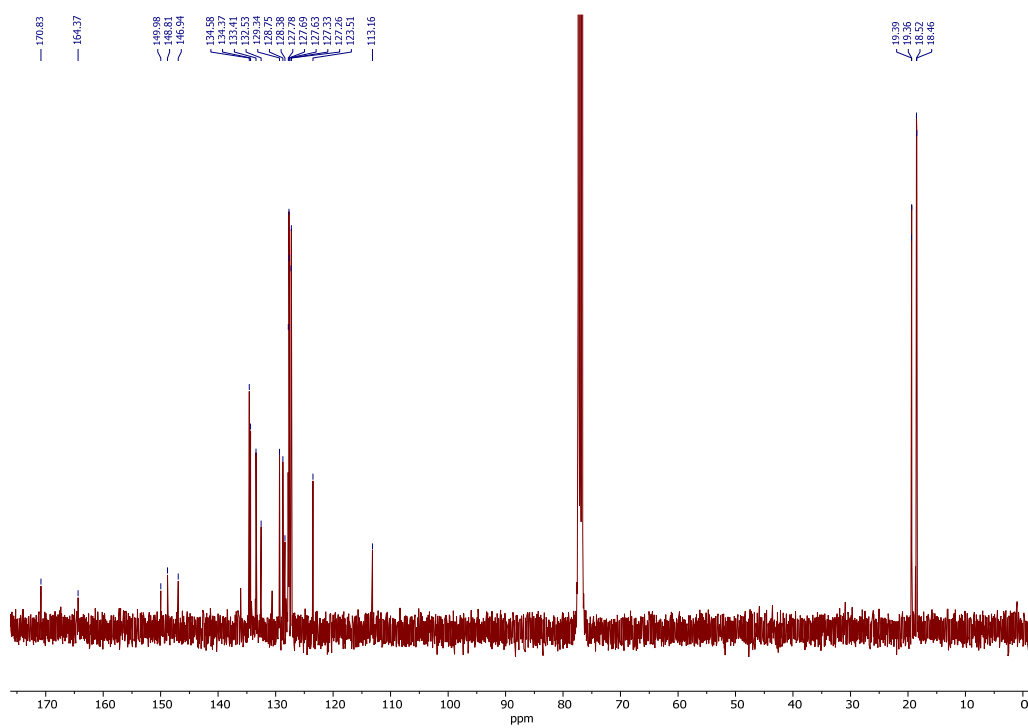


Figure S5. The ^{13}C NMR spectrum of **4b** in CDCl_3 .

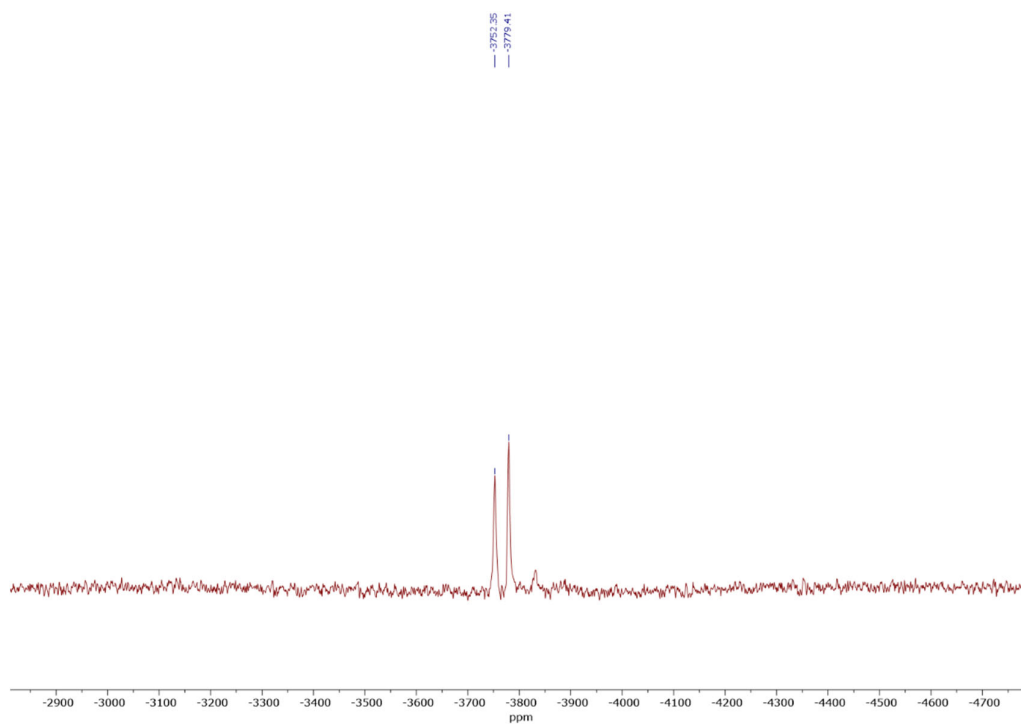


Figure S6. The ^{195}Pt NMR spectrum of **4b** in CDCl_3 .



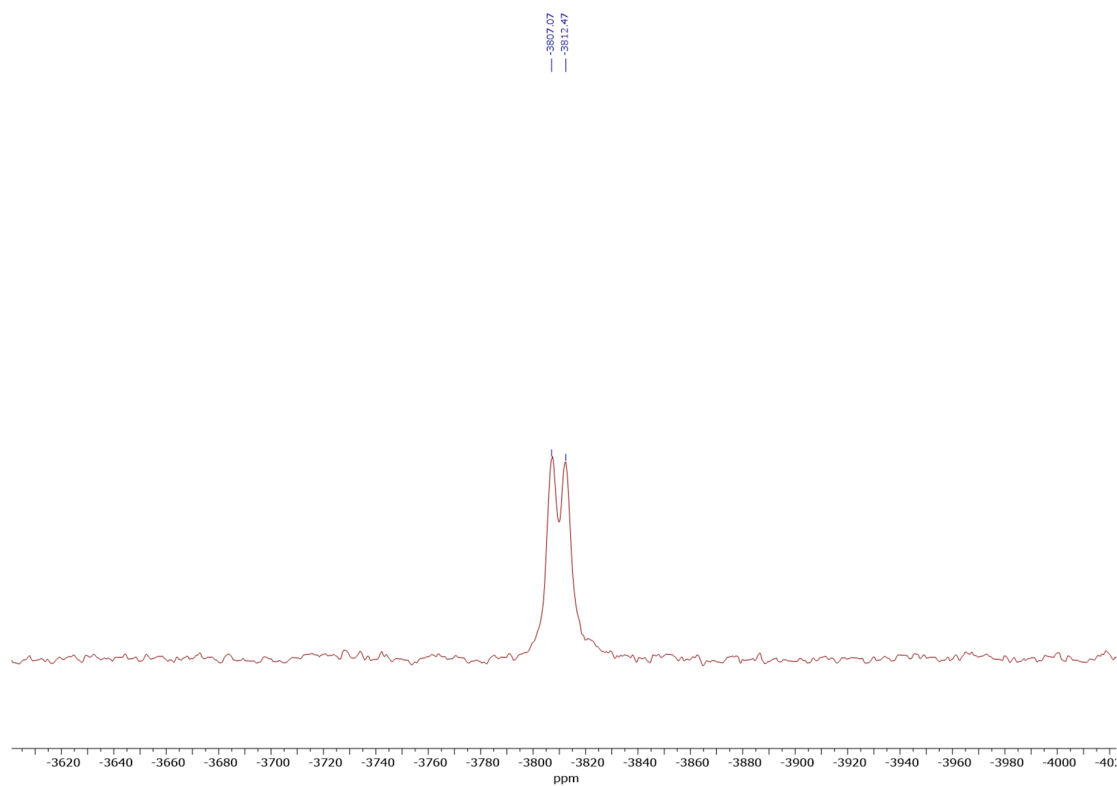


Figure S9. The ^{195}Pt NMR spectrum of **5a** in CDCl_3 .

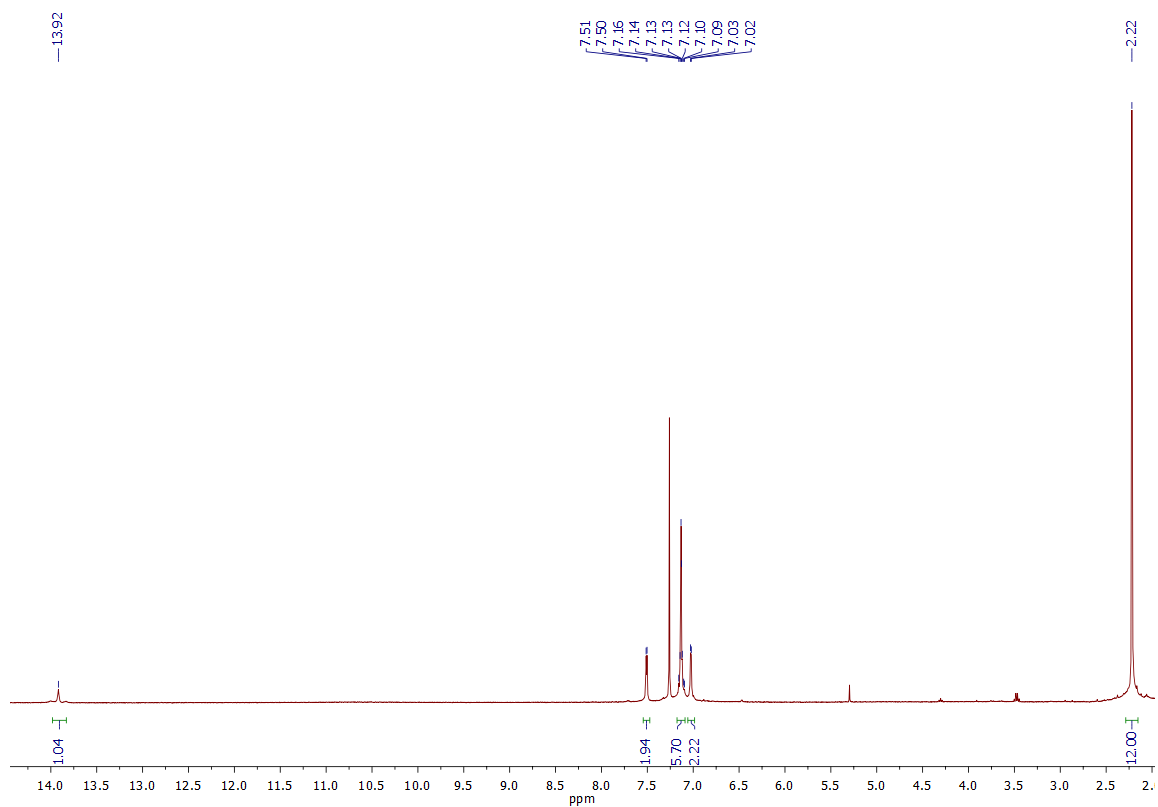


Figure S10. The ^1H NMR spectrum of **6** in CDCl_3 .

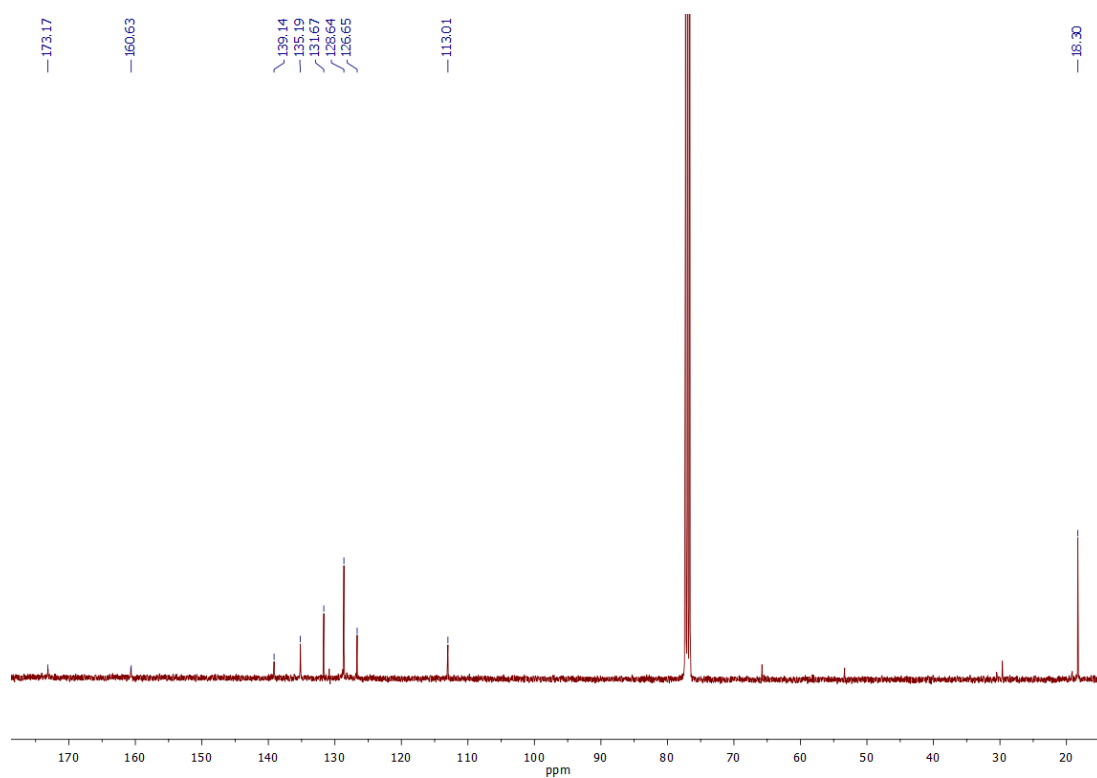


Figure S11. The ¹³C NMR spectrum of **6** in CDCl₃.

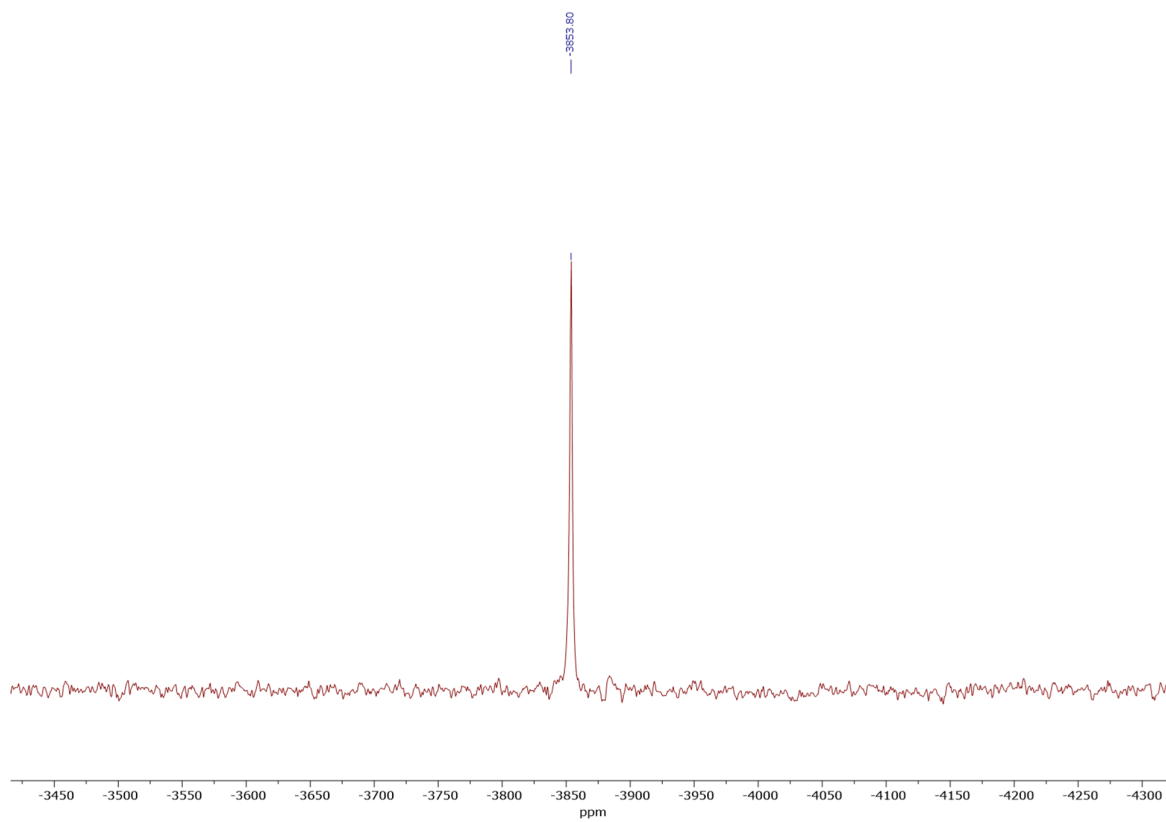


Figure S12. The ¹⁹⁵Pt NMR spectrum of **6** in CDCl₃.

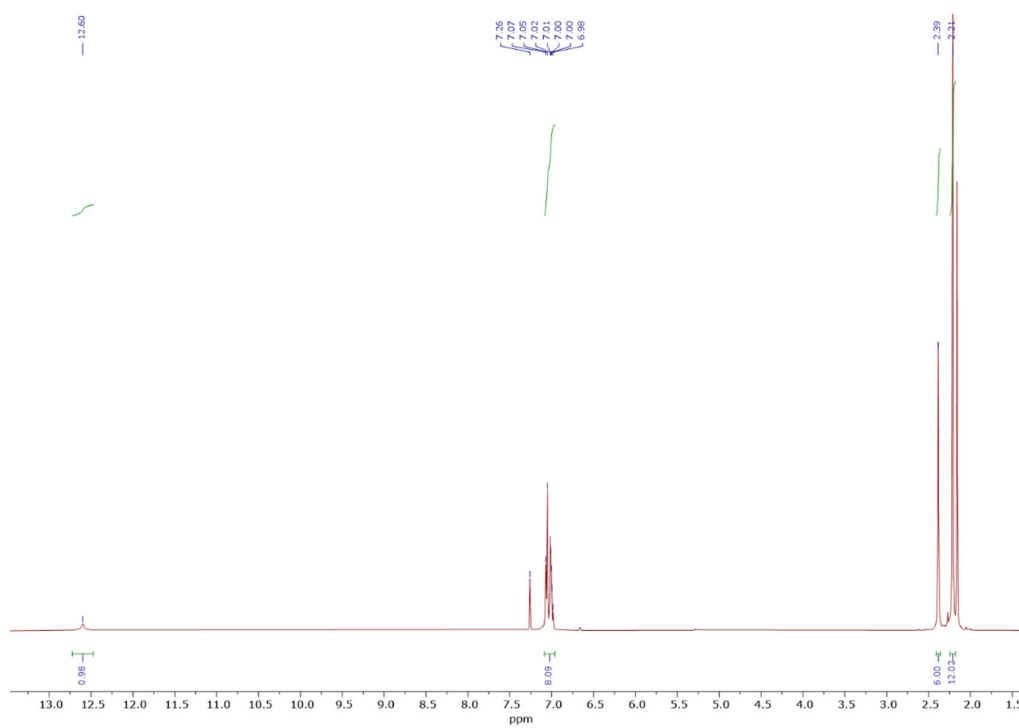


Figure S13. The ^1H NMR spectrum of **7**•Acetone in CDCl_3 .

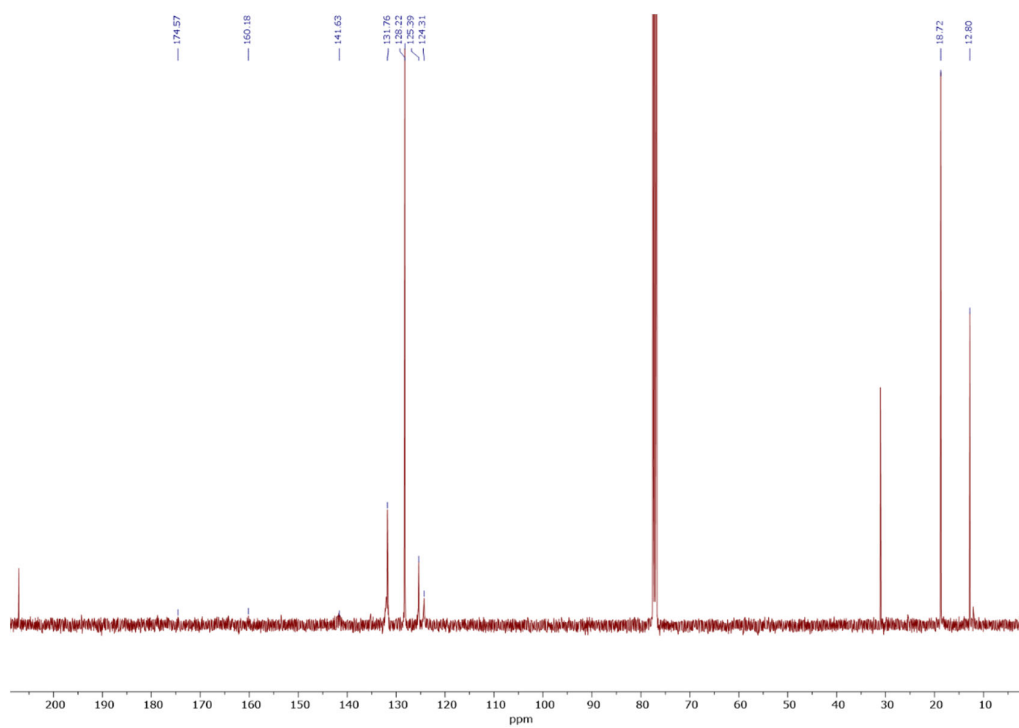


Figure S14. The ^{13}C NMR spectrum of **7**•Acetone in CDCl_3 .

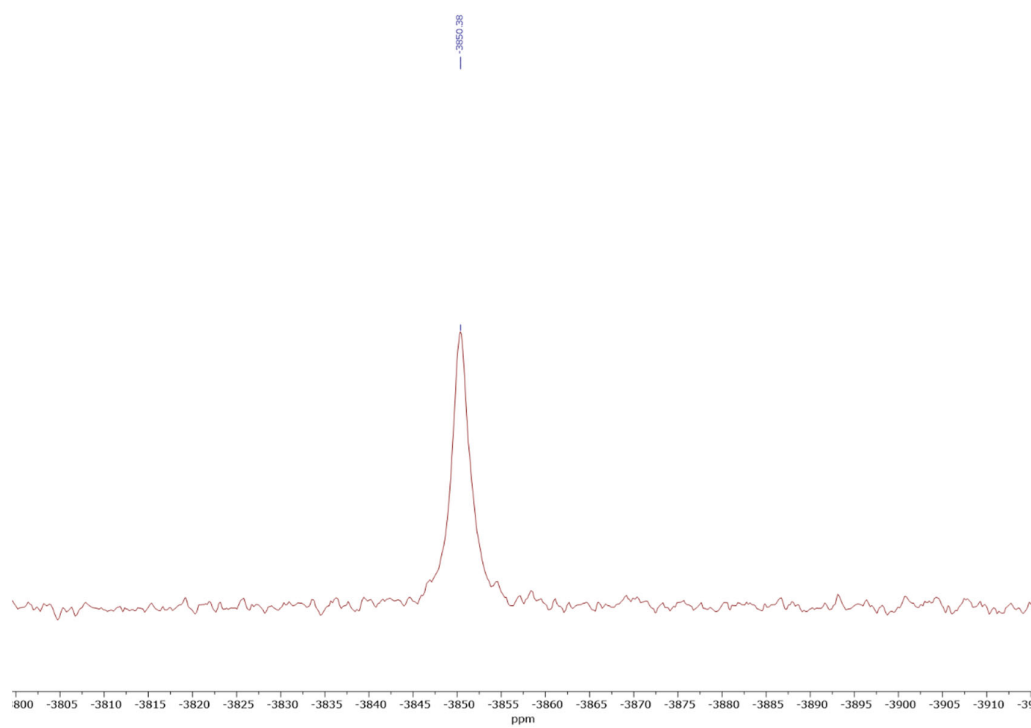


Figure S15. The ^{195}Pt NMR spectrum of **7**•Acetone in CDCl_3 .