

Supplementary Material

Understanding the Molecular Basis of 5-HT₄ Receptor Partial Agonists through 3D-QSAR Studies

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S1. EC₅₀ experimental

Table S1. Biological activity of all compounds. The highlighted rows show the test set compounds.

Compound	EC ₅₀ nM	pEC ₅₀	Compound	EC ₅₀ nM	pEC ₅₀
1	1.199	8.921	32	56.754	7.246
2	2.600	8.585	33	143.880	6.842
3	39.994	7.398	34	484.172	6.315
4	30.974	7.509	35	53.951	7.268
5	2249.055	5.648	36	114.288	6.942
6	519.996	6.284	37	4.898	8.310
7	466.659	6.331	38	0.300	9.523
8	736.207	6.133	39	8.710	8.060
9	2.000	8.699	40	8.395	8.076
10	72.277	7.141	41	79.068	7.102
11	19.815	7.703	42	598.412	6.223
12	857.038	6.067	43	1940.886	5.712
13	5.702	8.244	44	10.399	7.983
14	10.000	8.000	45	1940.886	5.712
15	9.795	8.009	46	1.500	8.824
16	1.500	8.824	47	0.700	9.155
17	3.396	8.469	48	61.944	7.208
18	0.500	9.301	49	97.949	7.009
19	500.035	6.301	50	758.578	6.120
20	1358.313	5.867	51	0.600	9.222
21	21.979	7.658	52	13.614	7.866
22	57.943	7.237	53	0.100	10.000
23	33.963	7.469	54	8.995	8.046
24	17.989	7.745	55	796.159	6.099
25	18.281	7.738	56	8.790	8.056
26	38.994	7.409	57	1.300	8.886
27	50.003	7.301	58	0.300	9.523
28	10.990	7.959	59	0.400	9.398
29	839.460	6.076	60	1088.930	5.963
30	4.797	8.319	61	20.989	7.678
31	5.200	8.284	62	41.976	7.377

S2. Details of the Compounds

S2.1. Dataset Collection

A total of 62 partial agonists of the 5-HT₄ receptor which showed promissory potency were collected from the literature [1–3]. All the compounds with pEC₅₀ values ranging from 5.64 to 10.0 were used in this study. The geometry for all these molecules was converted into a 3D structure using OCHEM. The 3D structure of the molecules was processed with OMEGA [4] module using the following parameters: (i) AM1_BCC Force field, (ii) FixpKa from the QUAPAC package for all possible

ionisation states at a given biological pH, (iii) one low energy conformation per ligand. Force- and Gaussian-field 3D-QSAR calculations were performed for all the molecules. All the training and test set molecules with experimental and predicted EC₅₀ values were listed in Table 1.

S2.2. Alignment

Alignment of molecules is the most crucial input for the generation of 3D-QSAR models. The compound with the highest activity (53) was used as the template molecule. A shape-based alignment was used for all conformers of each ligand. These alignments were carried out with ROCS suite [5]. Finally, each ligand's best conformer was filtered considering electrostatic field compound 53, as is shown in Figure S1.

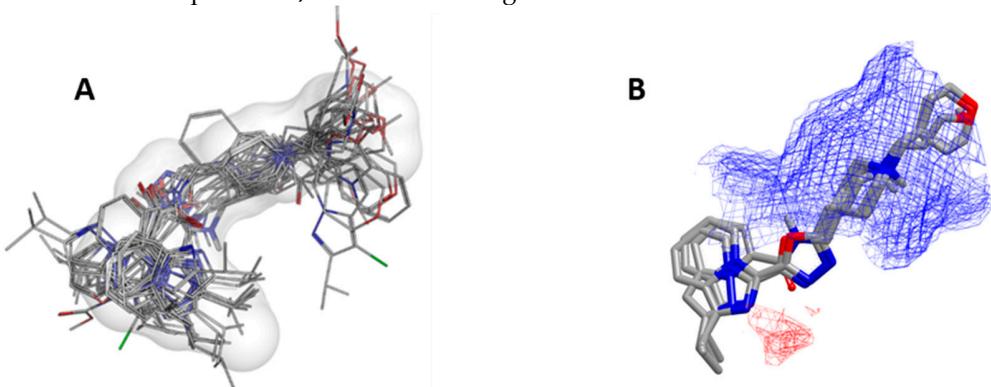


Figure S1. (A) structure aligned with ROCS. (B) structures aligned considering the electrostatic field.

S2.3. Field-Based QSAR Model

3D-QSAR analysis using Field-based methods was performed by QSAR tool of Schrodinger Suite. The 3D-QSAR method constructs the model by relating the known activities and molecular elements of a set of aligned compounds. The steric and electrostatic field around the ligand in a 3D-grid was calculated using field-based 3D-QSAR. Force-field based QSAR model (henceforward FFQSAR) is an alignment-dependent method in which molecular field interaction energy terms are correlated with biological activities/responses using multivariate statistical analyses. In Gaussian-field, 3D-QSAR model interaction energy calculations were performed using steric, electrostatic, hydrogen bond donor (HBD), and hydrogen bond acceptor (HBA) potential fields and it uses Gaussian equations for field calculations (henceforward GFQSAR).

The lattice and probe step sizes were adjusted automatically. The partial least squares (PLS) analysis is applied to construct the best model through the linear correlation of FFQSAR and GFQSAR concerning pEC₅₀ [1–3]. A cross-validation analysis was performed using the leave-one-out method. Finally, the optimum number of components was identified by the cross-validation method. Correlation and cross-validation coefficients (Q² and R² respectively) were calculated according to the formula:

$$Q^2 = 1 - \frac{\sum_{i=1}^{\infty} (\hat{y}_i - y_i)^2}{\sum_{i=1}^{\infty} (y_i - \bar{y})^2} \quad (1)$$

$$R^2 = \frac{[\sum (y_i - \bar{y})(\hat{y}_i - \bar{y})]^2}{\sum (y_i - \bar{y})^2 \times \sum (\hat{y}_i - \bar{y})^2} \quad (2)$$

where \hat{y}_i and y_i are predicted, observed activity values, and \bar{y} and \hat{y} are observed and predicted mean activity values of the training set, respectively. The $\sum_{i=1}^{\infty} (y_i - \bar{y})^2$ is the predictive residual sum of squares (PRESS).

High Q² and R² (Q² > 0.6, R² > 0.8) values are regarded as proof of the built model's high predictive ability.

S2.4. Validation of the 3D-QSAR Model

A good internal validation showed only a high Q² in the training set of compounds, but it did not indicate the established models' high predictive ability. Therefore, external validation was indispensable. The predictive power of 3D-

QSAR models was validated by calculating biological activities of the compounds which were not included in the training set and used as a test set.

The predictive correlation coefficient R^2_{test} ($R^2_{test} > 0.6$) [6], based on the test set was calculated using Equation (3):

$$R^2_{test} = \left(\frac{SD - PRESS}{SD} \right) \quad (3)$$

The sum of squared deviation (SD) between the biological activities of the test set molecules and the mean activity of the training set molecules. $PRESS$ is the sum of squared derivations between the predicted and actual activities of the test set molecules.

The performance of the regression models constructed here was evaluated using the root mean squared error ($RMSE$), mean absolute error (MAE) ($RMSE$ and MAE close to zero), residual sum of squares (RSS) and concordance correlation coefficient (CCC ; where $CCC \geq 0.85$) of the training and validation sets [7]. The $RMSE$ and the MAE are calculated for the data set as Equations (4) to (7):

$$RMSE = \sqrt{\frac{\sum_{i=1}^n (y_i - \hat{y}_i)^2}{n}} \quad (4)$$

$$MAE = \frac{\sum_{i=1}^n |y_i - \hat{y}_i|}{n} \quad (5)$$

$$RSS = \sum_{i=1}^n (y_i - \hat{y}_i)^2 \quad (6)$$

$$CCC = \frac{2 \sum_{i=1}^n (y_i - \bar{y})(\hat{y}_i - \bar{\hat{y}})}{\sum_{i=1}^n (y_i - \bar{y})^2 + \sum_{i=1}^n (\hat{y}_i - \bar{\hat{y}})^2 + n(\bar{y} - \bar{\hat{y}})^2} \quad (7)$$

To obtain the best predictive model for the test set, additional validation of model, the following:

$$r_0^2 = 1 - \frac{\sum(y_i - k \times \hat{y}_i)^2}{\sum(y_i - \bar{y})^2} \quad (8)$$

$$r_0'^2 = 1 - \frac{\sum(\hat{y}_i - k \times y_i)^2}{\sum(\hat{y}_i - \bar{\hat{y}})^2} \quad (9)$$

$$k = \frac{\sum(y_i \times \hat{y}_i)}{\sum(\hat{y}_i)^2} \quad (10)$$

$$k' = \frac{\sum(y_i \times \hat{y}_i)}{\sum(y_i)^2} \quad (11)$$

$$\frac{(r^2 - r_0^2)}{r^2} < 0.1 \quad or \quad \frac{(r^2 - r_0'^2)}{r^2} < 0.1 \quad (12)$$

$0.85 \leq k \leq 1.15$ or $0.85 \leq k' \leq 1.15$ r_0^2 and $r_0'^2$ are squared correlation coefficients of determination for regression lines through the origin between predicted (y) and observed (x) activities and vice versa. The values of k and k_0 are the slopes of their models, respectively.

To further assess the models, another statistical validation parameter r_m^2 and Δr_m^2 were determined by the following Equations (13) and (14):

$$r_m^2 = r^2 \left(1 - \sqrt{|r^2 - r_0^2|} \right) \quad (13)$$

$$\Delta r_m^2 = |r_m^2 - r_m'^2| \quad (14)$$

r_m^2 value of more than 0.5 ($r_m^2 > 0.5$) and $\Delta r_m^2 < 0.2$ show good external predictability of the models.

The internal and external validation for GFQSAR meet threshold values which demonstrates reliability of the model with the SEAD fields. In contrast, FFQSAR has the concordance correlation coefficient (CCC) value below the tolerated

threshold value (FFQSAR is 0.815 and the threshold > 0.85). Therefore, the design of new analogues will be based on the GFQSAR model.

Table S2. Summary of results obtained using Force- and Gaussian-field.

Internal validation parameters			
Parameters	Threshold Value	Force-field QSAR	Gaussian-field QSAR
R ² training	> 0.8	0.821	0.898
Q ² LOO	> 0.6	0.804	0.886
F		188.056	360.131
P		3.31E-13	2.92E-14

External validation parameters			
Parameters	Threshold Value	Force-field QSAR	Gaussian-field QSAR
R ² test	> 0.6	0.667	0.695
r ² ₀	Close to value of R ² test	0.574	0.678
r' ² ₀	Close to value of R ² test	0.661	0.656
k	0.85 < k < 1.15	0.996	0.992
k'	0.85 < k' < 1.15	0.997	1.003
(r ² - r ² ₀)/r ²	< 0.1	0.139	0.025
(r ² - r' ² ₀)/r ²	< 0.1	0.008	0.056
r ² ₀ - r' ² ₀	< 0.3	0.087	0.021
r ² _m	> 0.5	0.558	0.841
Q ² F ₁	> 0.6	0.785	0.819
Q ² F ₂	> 0.6	0.614	0.674
CCC	> 0.85	0.815	0.851
Δr ² _m	< 0.2	0.065	0.077

Error-based metrics:			
Parameters	Force-field QSAR	Gaussian-field QSAR	
RMSEP	0.632	0.581	
SD	0.431	0.377	
MAE	0.473	0.450	

Q^2 = the square of the LOO cross-validation (CV) coefficient; R^2_{test} is the regression coefficient for the test set exclusively; " r^2_0 " and " k " are the correlation coefficient between the actual and predicted activities for test set and the respective slope of regression; and " r'^2_0 " and " k' " are the correlation coefficient between the predicted and actual activities for test set and the respective slope of regression. " r^2_m " was defined in equation 11. Parameters are defined in the section "Validation of the QSAR model".

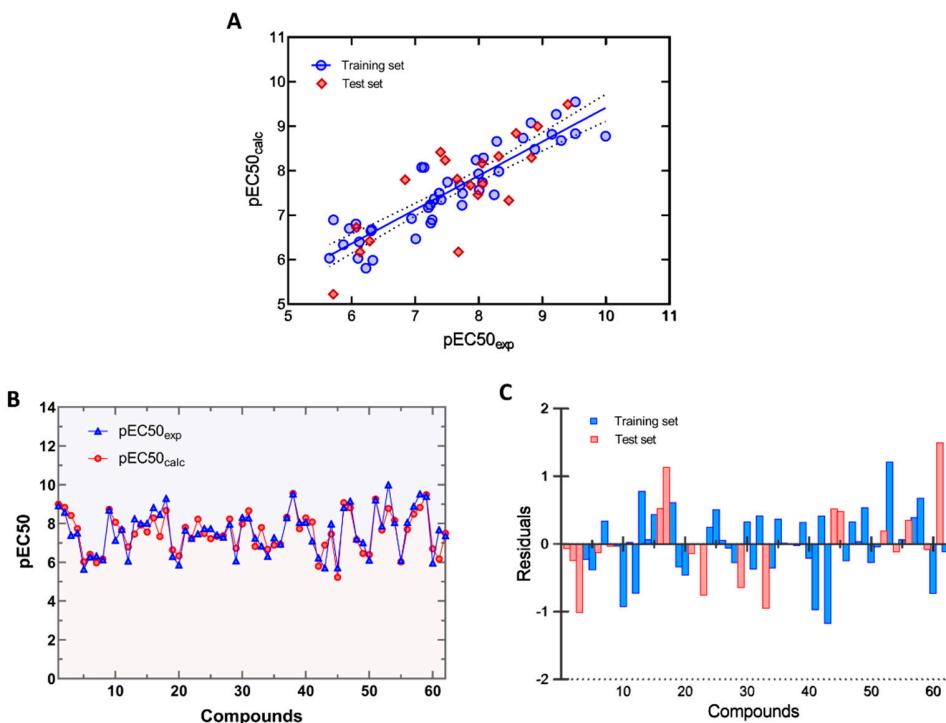


Figure S2. (A) Plots of experimental versus calculated pEC_{50} values for the training and test set molecules for Force field QSAR (FFQSAR) model. (B) Residual plots between predicted and observed values for FFQSAR. (C) FFQSAR predictions for every molecule in the complete set

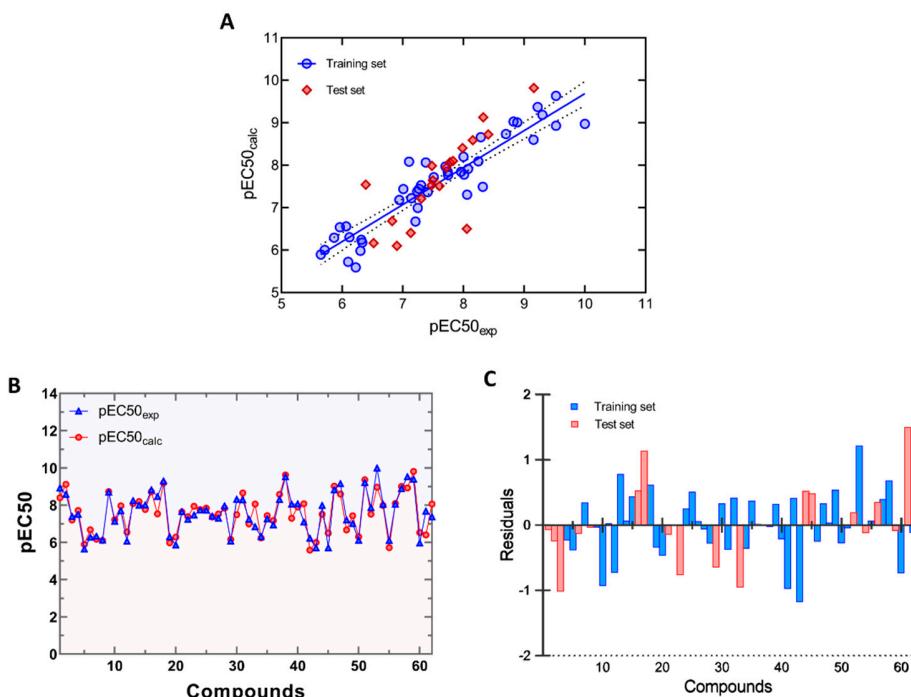


Figure S3. (A) Plots of experimental versus calculated pEC_{50} values for the training and test set molecules for Gaussian field 3D-QSAR (GFQSAR) model. (B) Residual plots between predicted and observed values for GFQSAR. (C) GFQSAR predictions for every molecule in the complete set

S2.5. Prediction ADMET Properties

Drug candidates need to have good ADMET (Absorption, Distribution, Metabolism, Excretion and Toxicity) and drug-likeness profiles to initially estimate pharmacokinetic and drug-likeness parameters in the drug discovery process [8].

In this work, new candidates with ADMET properties include human intestinal absorption, steady-state volume of distribution (VDss), hepatic metabolism, total clearance, AMES toxicity, and hepatotoxicity and skin sensitisation properties. ADMET can be predicted using pkCSM [9].

The prediction of drug similarity of new molecules is estimated using parameters based on Lipinski, Ghose, Veber and Egan rules, and their synthetic accessibility by applying the SwissADME web tool [10] (<http://www.swissadme.ch>). The SwissADME synthetic accessibility score is mainly based on the assumption of the molecular fragments in the "actually" obtainable molecules which correlates with the ease of synthesis. The score is normalised to range from 1 (very easy) to 10 (very difficult to synthesise).

S3. Novel derivatives

Table S3. 39 new compounds with predicted pEC_{50} .

Num	SMILES	Selected	GFQSAR	FFQSAR
1	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)CC4CCOCC4		8.985	8.714
2	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)[C@H]4CCC[C@H]([C@H]45)COCC5	var1	9.221	8.283
3	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)[C@H](CCC4)C[C@H]4COC	var2	9.375	8.209
4	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)[C@H](CCC4)C[C@H]4CO	var3	9.432	8.285
5	CC(C)c(c1)oc(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)CC4CCOCC4		8.963	8.411
6	CC(C)c(c1)oc(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)CCCCOC		9.085	8.403
7	CC(C)c(c1)oc(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)C(C)(C)CCCC		9.061	8.017
8	CC(C)c(c1)oc(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)C(C)(C)C4CCOCC4		8.993	8.049
9	CC(C)c(c1)oc(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)CCCCO		9.031	8.379
10	CC(C)c(c1)oc(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)C/C=C/CO		9.041	8.418
11	CC(C)c(c1)sc(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)CC4CCOCC4		8.533	8.272
12	CC(C)(C)c(c1)oc(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)CC4CCOCC4		8.794	7.620
13	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)[C@H](C4)CC[C@H]([C@@H]45)CCC[C@H]5OC		9.179	8.077
14	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)C/C=C/CO	var4	9.114	8.719
15	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)C[C@H](O)CC		9.090	8.659
16	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)[C@H](C4=O)CCCC4		8.821	8.363
17	CCC(=O)C[N@H+](CC1)CC[C@H]1CNC(=O)c(cc2)n(c23)nc(c3)C(C)C		8.779	8.882
18	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)C[C@H]4CCCCO4		8.998	8.698
19	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)C[C@H]4CCCO4		8.992	8.710
20	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)Cc4ccco4		8.802	8.853
21	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)C[C@H](O)CC4CCCCC4		9.044	8.800
22	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)C[C@H](O)C4CCCCC4		9.009	8.138
33	CC(C)c(c1)nn(c12)c(cc2)C(=O)NCCCC[N@@H+](CCO3)C[C@H]3CCC=O	var5	9.513	8.775
23	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)C[C@H]4CCOCO4	var6	9.111	8.698
24	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)[C@H](N4)CCC[C@H]4C N		9.001	8.475
25	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)[C@H](N4)CCC[C@H]4C NC		9.104	8.529
26	CC(C)c(c1)nn(c12)c(cc2)-c3ncn(o3)[C@H]4CC[N@@H+](CC4)[C@H](CCC5)C[C@H]5CN		6.995	6.582
27	CC(C)c(c1)nn(c12)c(cc2)-c3ncn(o3)[C@H]4CC[N@@H+](CC4)[C@H](CCC5)C[C@H]5CO		7.223	6.966
28	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)[C@H](N4)CCC[C@H]4C N		9.022	8.705
29	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H]3CC[N@@H+](CC3)[C@H](CCC4)C[C@H]4C(=O) OC		9.139	7.681
30	C1CCCCC1C(=O)[C@@H]2C[C@H](CCC2)[N@H+](CC3)CC[C@H]3CNC(=O)c(cc4)n(c 45)nc(e5)C(C)C		9.004	7.819
31	COC(=O)CCC[N@H+](CC1)CC[C@H]1CNC(=O)c(cc2)n(c23)nc(c3)C(C)C		8.820	9.203
32	CC(C)c(c1)nn(c12)c(cc2)C(=O)NCCCC[N@@H+](CCO3)C[C@H]3C		9.203	8.012
34	CC(C)c(c1)nn(c12)c(cc2)C(=O)NCCCCC3CCOCC3		9.347	6.535
35	CC(C)c(c1)nn(c12)c(cc2)C(=O)NCCCC[N@@H+](CCO3)C[C@H]3C4CCOCC4	var7	9.547	8.690
36	CC(=O)C[NH2+]CCCCNC(=O)c(cc1)n(c12)nc(c2)C(C)C	var8	9.259	9.417
37	CC(C)c(c1)nn(c12)c(cc2)C(=O)NC[C@H](C[C@H]3C=O)CC[N@@H+]3[C@H](CCC4)C[C @H]4CO	var9	9.700	8.818
38	CC(=O)[C@H]1C[C@H](CC[NH2+]1)CNC(=O)c(cc2)n(c23)nc(c3)C(C)C	var10	9.849	8.641
39	CC(C)c(c1)nn(c12)c(cc2)-c3cn(nn3)[C@H]4CC[N@@H+](CC4)CC5CCOCC5		7.488	8.289

Table S4. Structural characteristics of the 10 compounds selected with SwissADME.

Molecule	SMILES	Formula	MW	Heavy atoms	Aromatic heavy atoms	Fraction Csp3	Rotatable bonds	H-bond acceptors	H-bond donors	MR	TPSA
var1	O=C(c1cccc2n1nc(c2)C(C)C)NC[C@@H]1CCN(CC1)[C@@H]1CCC[C@H]2[C@H]1CCOC2	C26H38N4O2	438.61	32	9	0.69	6	4	1	131.62	58.87
var2	CO[C@H]1CCCC[C@H](C1)N1CC[C@H](CC1)CN C(=O)c1cccc2n1nc(c2)C(C)C	C25H38N4O2	426.59	31	9	0.68	8	4	1	128.93	58.87
var3	OC[C@H]1CCCC[C@H](C1)N1CC[C@H](CC1)CNC(=O)c1cccc2n1nc(c2)C(C)C	C24H36N4O2	412.57	30	9	0.67	7	4	2	124.2	69.87
var4	OC/C=C/CN1CC[C@H](CC1)CNC(=O)c1cccc2n1nc(c2)C(C)C	C21H30N4O2	370.49	27	9	0.52	8	4	2	111.42	69.87
var5	O=CCCC[C@H]1OCCN(C1)CCCCNC(=O)c1cccc2n1nc(c2)C(C)C	C22H32N4O3	400.51	29	9	0.59	11	5	1	116.82	75.94
var6	O=C(c1cccc2n1nc(c2)C(C)C)NC[C@@H]1CCN(CC1)C[C@H]1CCOCO1	C22H32N4O3	400.51	29	9	0.64	7	5	1	115.59	68.1
var7	O=C(c1cccc2n1nc(c2)C(C)C)NCCCCN1CCO[C@H](C1)C1CCOCC1	C24H36N4O3	428.57	31	9	0.67	9	5	1	125.21	68.1
var8	CC(=O)CNCCCCNC(=O)c1cccc2n1nc(c2)C(C)C	C18H26N4O2	330.42	24	9	0.5	10	4	2	94.61	75.5
var9	CO[C@H]1CCCC[C@H](C1)N1CC[C@H](CC1)CNC(=O)c1cccc2n1nc(c2)C(C)C	C27H40N4O3	468.63	34	9	0.67	9	5	1	138.74	75.94
var10	CC(=O)[C@@H]1NCC[C@H](C1)CNC(=O)c1cccc2n1nc(c2)C(C)C	C19H26N4O2	342.44	25	9	0.53	6	4	2	101.22	75.5

Table S5. Lipophilicity calculated for the 10 compounds selected with SwissADME.

Molecule	Canonical SMILES	iLOGP	XLOGP3	WLOGP	MLOGP	Silicos-IT Log P	Consensus Log P
var1	O=C(c1cccc2n1nc(c2)C(C)C)NC[C@@H]1CCN(CC1)[C@@H]1CCCC[C@H]2[C@H]1CCOC2	4.74	4.03	3.72	3.4	3.15	3.81
var2	CO[C@H]1CCC[C@H](C1)N1CC[C@H](CC1)CNC(=O)c1ccc2n1nc(c2)C(C)	4.55	4	3.72	3.19	3.3	3.75
var3	OC[C@H]1CCC[C@H](C1)N1CC[C@H](CC1)CNC(=O)c1cc	3.85	3.46	3.07	2.98	2.75	3.22
var4	OC/C=C/CN1CC[C@H](CC1)CNC(=O)c1cccc2n1nc(c2)C(C)	3.8	2.24	2.07	2.26	2.46	2.57
var5	C(C)C	3.75	2.07	2.27	1.68	3.12	2.58
var6	O=C(c1cccc2n1nc(c2)C(C)C)NC[C@@H]1CCN(CC1)C[C@H]1CCOCO1	4	2.62	2.28	2.16	2.36	2.69
var7	CCOCC1	4.63	2.88	2.71	2.18	3.15	3.11
var8	CC(=O)CNCCCCNC(=O)c1cccc2n1nc(c2)C(C)C	2.92	1.87	2.15	1.58	2.57	2.22
var9	COC[C@H]1CCC[C@H](C1)N1CC[C@H](C[C@H]1C(=O)C)CNC(=O)c1cccc2n1nc(c2)	4.17	3.78	3.68	2.72	3.37	3.54
var10	C(C)C	2.94	1.94	1.76	1.81	2.16	2.12

Table S6. Solubility calculated for the 10 compounds selected with SwissADME.

Molecule	Canonical SMILES	ESOL Log S	ESOL Solubility (mg/ml)	ESOL Solubility (mol/l)	ESOL Class	Ali Log S	Ali Solubility (mg/ml)	Ali Solubility (mol/l)	Ali Class	Silicos-IT LogSw	Silicos-IT Solubility (mg/ml)	Silicos-IT Solubility (mol/l)	Silicos-IT class
var1	O=C(c1cccc2n1nc(c2)C(C)C)NC[C@@H]1CCN(C1)C[C@H]1CCCC[C@H]2[C@H]1CCOC2	-4.91	5.39E-03	1.23E-05	Moderately soluble	-4.97	4.71E-03	1.07E-05	Moderately soluble	-5.24	2.52E-03	5.75E-06	Moderately soluble
var2	CO[C@H]1CCC[C@H](C1)N1CC[C@H](CC1)CNC(=O)c1cccc2n1nc(c2)C(C)	-4.69	8.68E-03	2.03E-05	Moderately soluble	-4.94	4.92E-03	1.15E-05	Moderately soluble	-5.44	1.53E-03	3.60E-06	Moderately soluble
var3	OC[C@H]1CCC[C@H](C1)N1CC[C@H](CC1)CNC(=O)c1cccc2n1nc(c2)C(C)	-4.34	1.90E-02	4.59E-05	Moderately soluble	-4.61	1.02E-02	2.46E-05	Moderately soluble	-4.75	7.27E-03	1.76E-05	Moderately soluble
var4	OC/C=C/CN1CC[C@H](CC1)CNC(=O)c1cccc2n1nc(c2)C(C)	-3.27	2.00E-01	5.41E-04	Soluble	-3.34	1.68E-01	4.54E-04	Soluble	-4.08	3.10E-02	8.36E-05	Moderately soluble
var5	O=C(c1cccc2n1nc(c2)C(C)C)NC[C@@H]1CCN(C1)C[C@H]1CCCC[C@H]2[C@H]1CCOC2	-3.13	2.96E-01	7.40E-04	Soluble	-3.29	2.04E-01	5.08E-04	Soluble	-5.16	2.75E-03	6.87E-06	Moderately soluble
var6	CC(=O)CNCCCCNC(=O)c1cccc2n1nc(c2)C(C)C	-3.74	7.26E-02	1.81E-04	Soluble	-3.7	7.99E-02	1.99E-04	Soluble	-4.49	1.29E-02	3.21E-05	Moderately soluble
var7	CC(=O)CNCCCCNC(=O)c1cccc2n1nc(c2)C(C)C	-3.93	5.01E-02	1.17E-04	Soluble	-3.97	4.59E-02	1.07E-04	Soluble	-5.28	2.26E-03	5.27E-06	Moderately soluble
var8	CC(=O)CNCCCCNC(=O)c1cccc2n1nc(c2)C(C)C	-2.68	6.84E-01	2.07E-03	Soluble	-3.08	2.77E-01	8.37E-04	Soluble	-5.29	1.68E-03	5.08E-06	Moderately soluble
var9	CC(=O)CNCCCCNC(=O)c1cccc2n1nc(c2)C(C)C	-4.73	8.75E-03	1.87E-05	Moderately soluble	-5.07	4.00E-03	8.54E-06	Moderately soluble	-5.51	1.46E-03	3.11E-06	Moderately soluble
var10	CC(=O)CNCCCCNC(=O)c1cccc2n1nc(c2)C(C)C	-3.06	3.01E-01	8.80E-04	Soluble	-3.15	2.43E-01	7.08E-04	Soluble	-4.59	8.72E-03	2.55E-05	Moderately soluble

S4. References

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