Supplementary Information

A low-cost electrochemical method for the determination of sulfadiazine in aquaculture wastewater

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Supporting Figures:

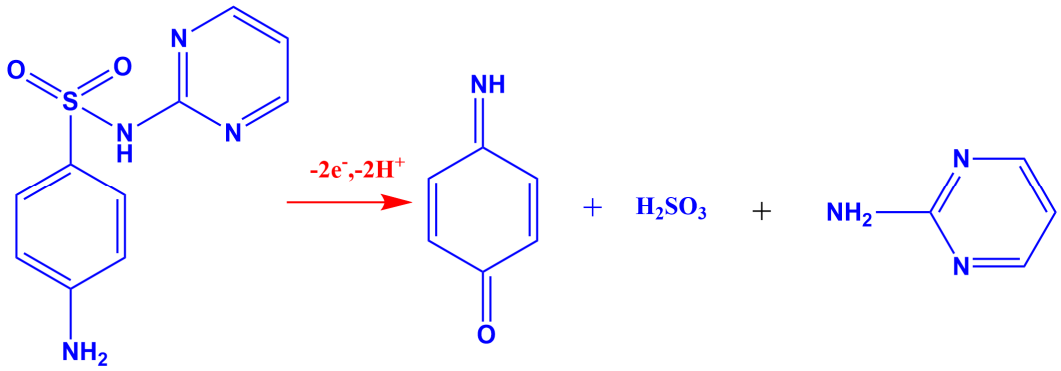

Figure. S1. Electrochemical oxidation mechanism of sulfadiazine.

Tables

Table S1. Analytical data comparison between SDZ detection and previous electrodes.

Electrode	**Electrolyte solution**	**LOD**	**Linear range**	**Ref.**
GCE	0.04 M B-R buffer (pH 6.8)	10.9 μM	62.7-340 μM	[1]
Carboxyl-MWCNTs/GCE	0.04 M B-R buffer (pH 2)	0.07μM	0.50-110 μM	[2]
MIP/GO@COF/GCE	0.2 M PBS (pH 7)	0.16 μM	0.5–200 μM	[3]
MWCNT-GCE	0.04 M B-R buffer (pH 7)	7.1 μM	10-2000 μM	[4]
GCE	0.1M ABS (pH 4)	6.14 μM	20-300 μM	This work

(GCE, glassy carbon electrode; MWCNT, multiwalled carbon nanotube; MIP, molecularly imprinted polymer; GO@COF, grapheneoxide@covalentorganic framework; B-R buffer, Britton–Robinson buffer; PBS, phosphate buffer; ABS, acetic acid-sodium acetate buffer solution).

Supporting references

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