

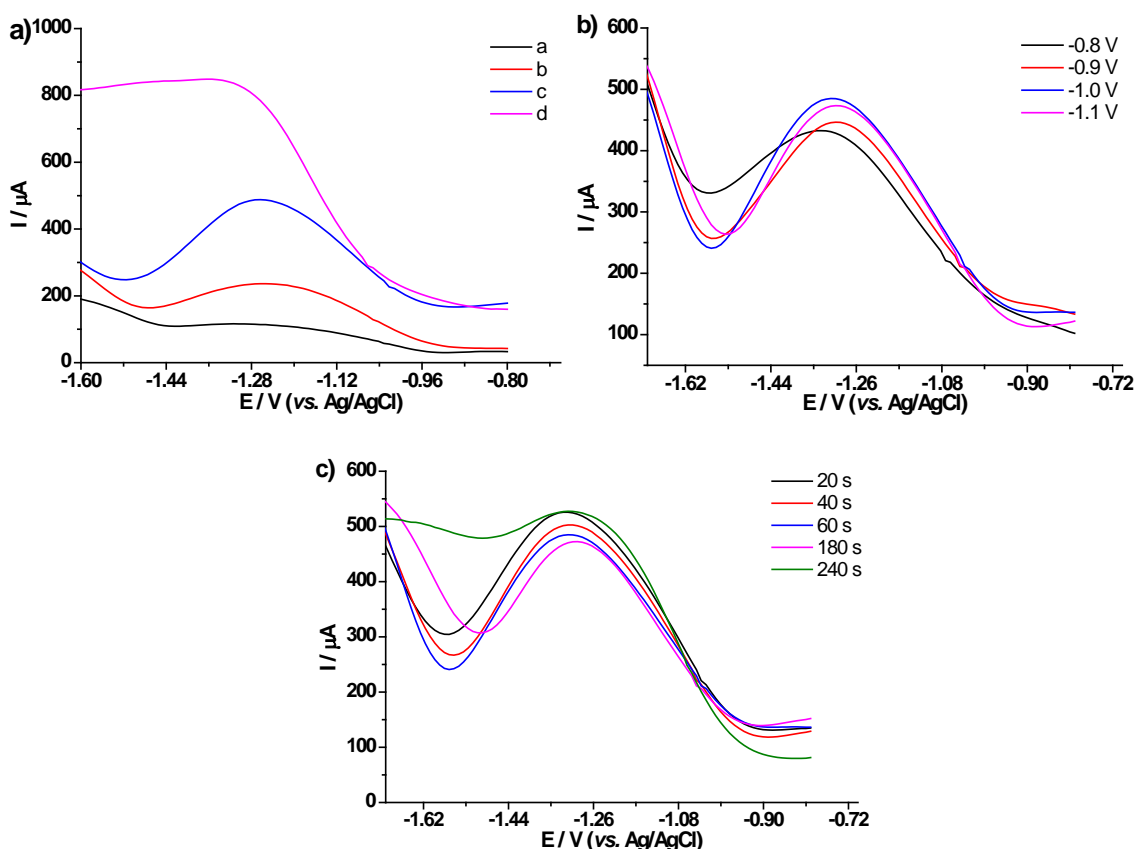
## Soportes

### Tables

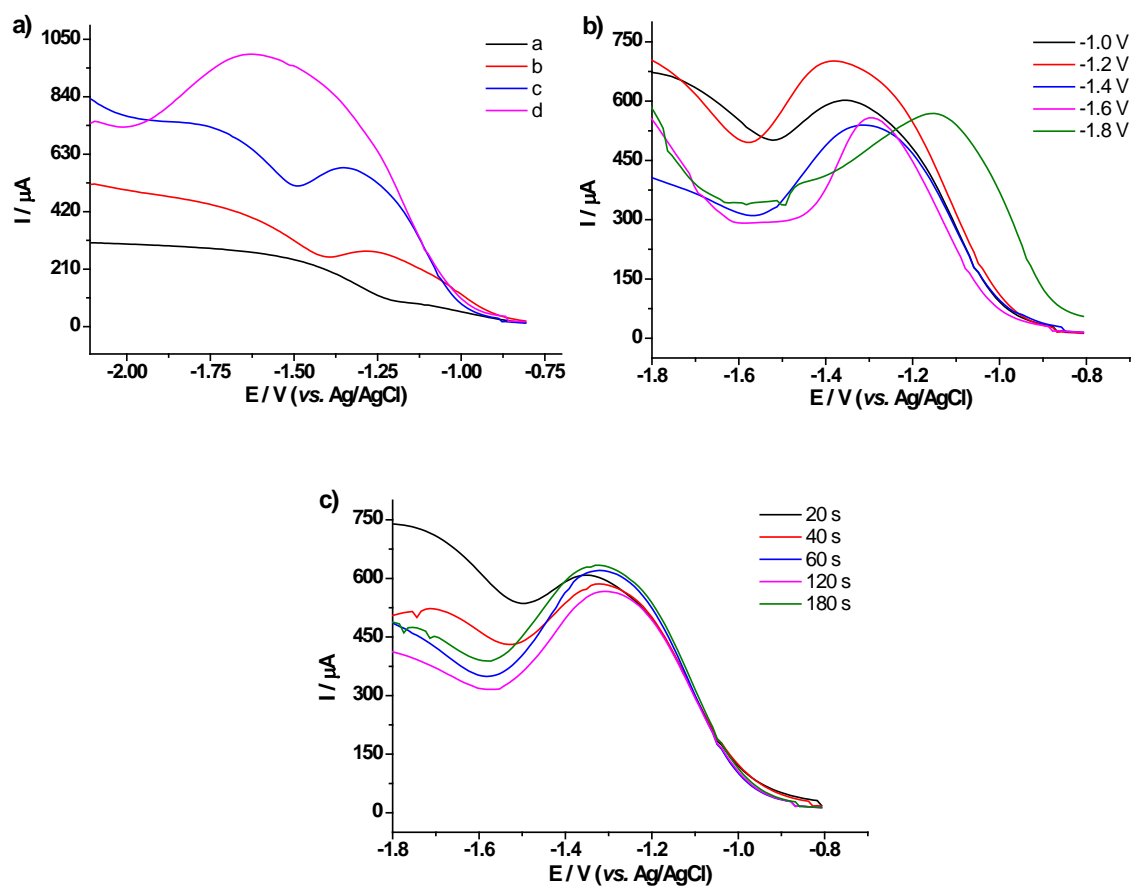
**Table S-1.** Cyclic voltammetry of  $K_4[Fe(CN)_6]$  /  $K_3[Fe(CN)_6]$  on BDD electrode.

$\nu$ ( $V\ s^{-1}$ )	$\Delta E_p$ (mV)	$E_{p1/2}$ (mV)	$I_{p_{ox}}$ ( $\mu A$ )	$I_{p_{red}}$ ( $\mu A$ )	$I_{p_{ox}}/I_{p_{red}}$	$k^o$ ( $cm\ s^{-1}$ )
0.020	74.78	372.90	193.2412	195.8914	0.99	$3.25 \times 10^{-02}$
0.040	87.86	372.77	275.0711	278.4573	0.99	$2.70 \times 10^{-02}$
0.060	96.10	371.78	335.5251	339.4322	0.99	$2.41 \times 10^{-02}$
0.080	101.09	371.84	385.0713	388.0118	0.99	$2.20 \times 10^{-02}$
0.100	107.50	371.63	430.0244	432.3939	0.99	$2.09 \times 10^{-02}$
0.120	113.04	372.27	468.9341	471.5781	0.99	$2.01 \times 10^{-02}$
Average						$2.44 \times 10^{-2} \pm 4.67 \times 10^{-3}$

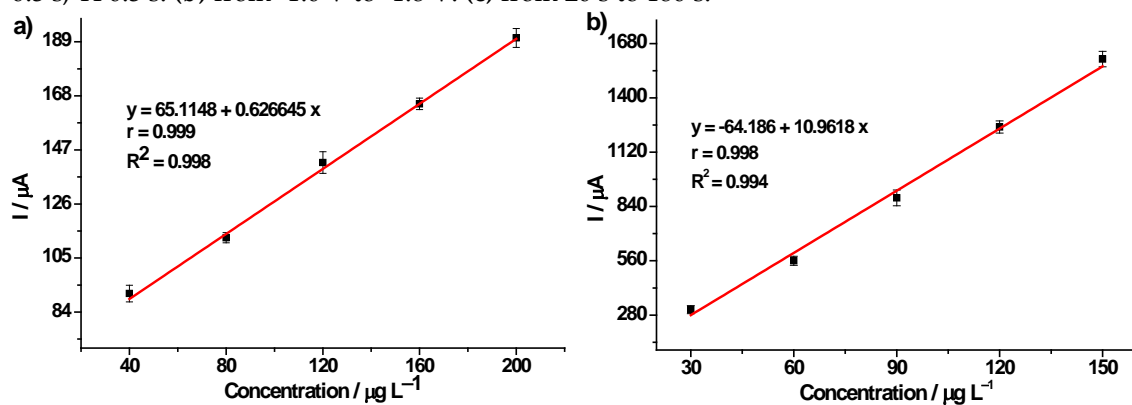
### Figures



**Figure S-1.** Dissolution behavior of  $90\ \mu g\ L^{-1}$  of  $Cu(II)$  in electrolyte  $KNO_3\ 0.1\ mol\ L^{-1}$  /  $HNO_3\ 0.1\ mol\ L^{-1}$  by DPASV at different parameters: **(a)** from MA 0.05 V, MT 0.05 s, TI 0.05 s - MA 0.3 V, MT 0.3 s, TI 0.3 s. **(b)** from -0.8 V to -1.1 V. **(c)** from 20 s to 240 s.



**Figure S-2.** Dissolution behavior of 80  $\mu\text{g L}^{-1}$  of Fe(III) in  $\text{KNO}_3$  0.1 mol  $\text{L}^{-1}$  /  $\text{HNO}_3$  0.01 mol  $\text{L}^{-1}$  electrolyte at different DPASV parameters: (a) from MA 0.05 V, MT 0.05 s, TI 0.05 s - MA 0.3 V, MT 0.3 s, TI 0.3 s. (b) from -1.0 V to -1.8 V. (c) from 20 s to 180 s.



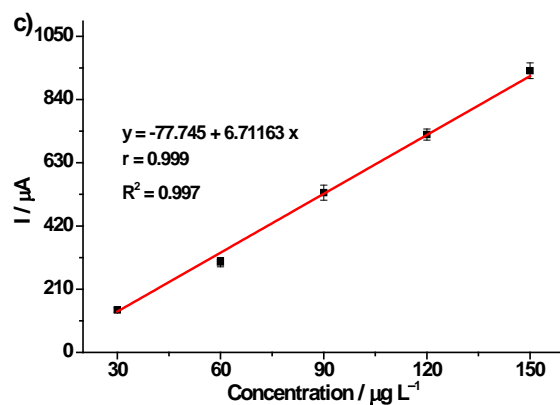
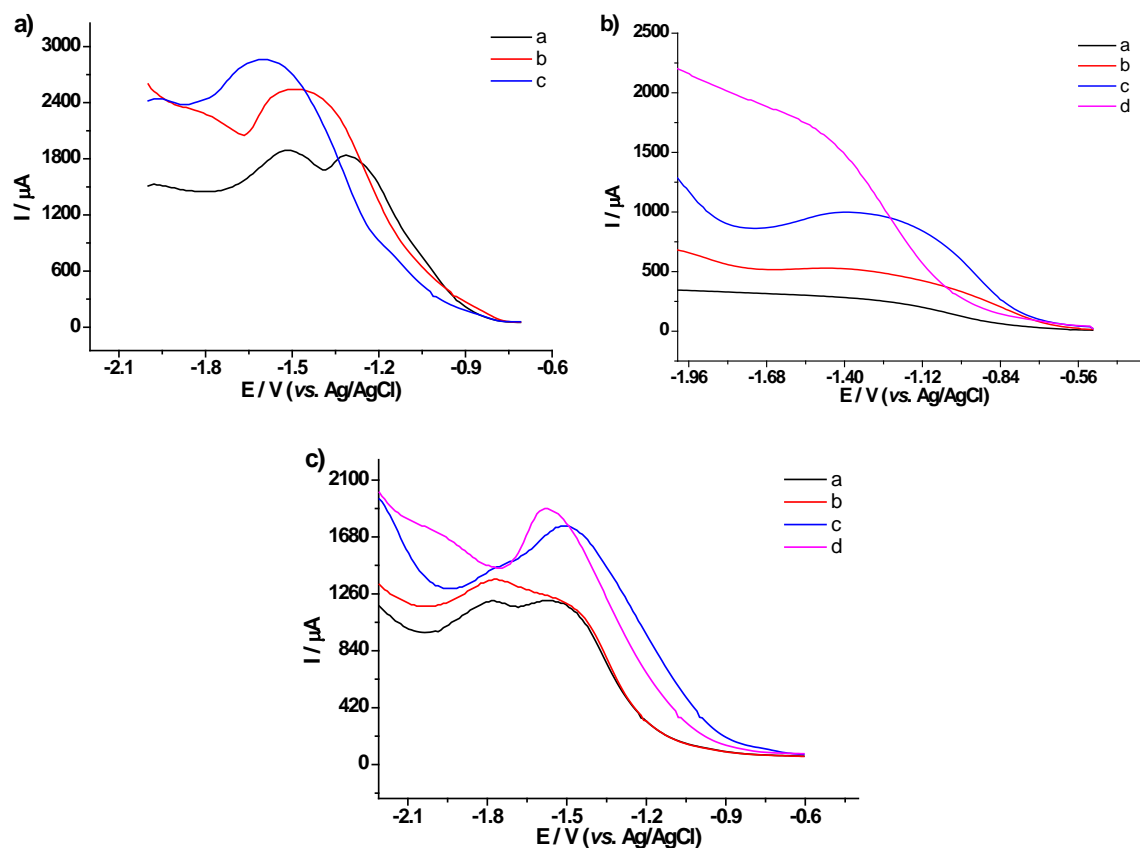


Figure S-3. Calibration plots of: (a) Cd(II); (b) Cu(II); (c) Fe(III).



**Figure S-4.** Beer matrix dissolution behavior with metal standard: (a) 200  $\mu\text{g L}^{-1}$  standard of Cd(II) in 0.1 mol  $\text{L}^{-1}$  acetic acid electrolyte / sodium acetate at pH 4.5, by DPASV at different parameters: a) MA 0.5 V, MT 0.5 s, TI 0.5 s; b) MA 0.6 V, MT 0.6 s, TI 0.6 s; c) MA 0.7 V, MT 0.7 s, TI 0.7 s. (b) standard of 90  $\mu\text{g L}^{-1}$  of Cu(II) in electrolyte  $\text{KNO}_3$  0.1 mol  $\text{L}^{-1}$  /  $\text{HNO}_3$  0.1 mol  $\text{L}^{-1}$  by DPASV at different parameters: a) MA 0.5 V, MT 0.5 s, TI 0.5 s; b) MA 0.6 V, MT 0.6 s, TI 0.6 s; c) MA 0.7 V, MT 0.7 s, TI 0.7 s; d) MA 0.8 V, MT 0.8 s, TI 0.8 s. (c) 80  $\mu\text{g L}^{-1}$  Fe(III) standard in 0.1 mol  $\text{L}^{-1}$   $\text{KNO}_3$  / 0.01 mol  $\text{L}^{-1}$   $\text{HNO}_3$  electrolyte at different DPASV parameters: a) MA 0.4 V, MT 0.4s, TI 0.4s; b) MA 0.5 V, MT 0.5 s, TI 0.5 s; c) MA 0.6 V, MT 0.6 s, TI 0.6 s; d) MA 0.7 V, MT 0.7 s, TI 0.7 s.

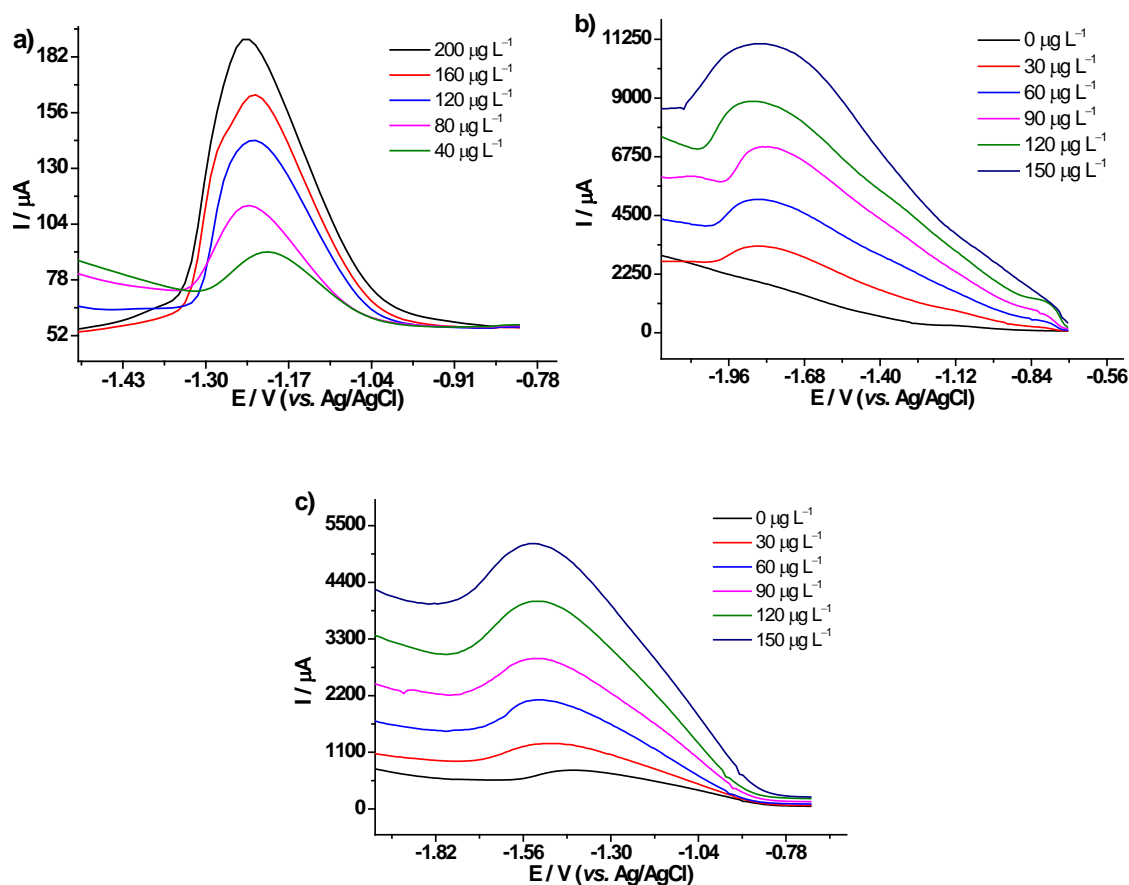


Figure S-5. Voltammogram and standard addition curve by DPASV in craft beer at different concentrations: (a) Cd(II); (b) Cu(II); (c) Fe(III).

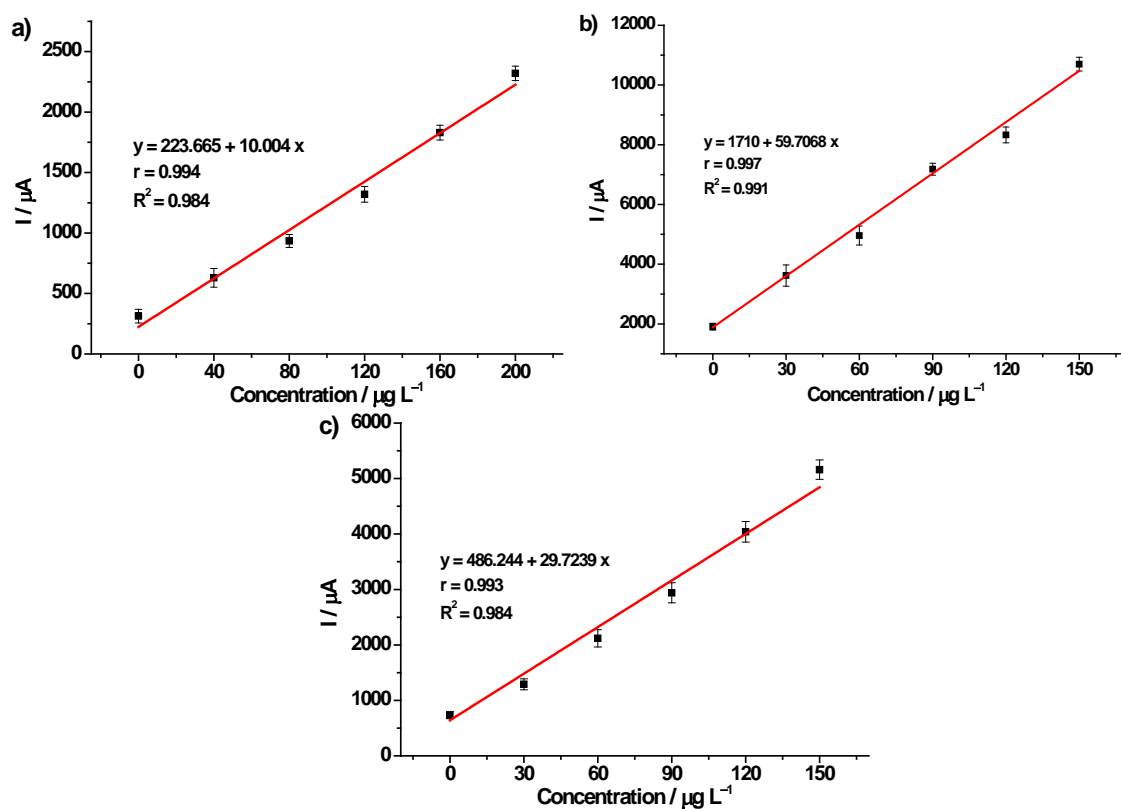


Figure S-6. Standard improvement calibration curves: (a) for Cd(II), (b) for Cu(II) and (c) for Fe(III).