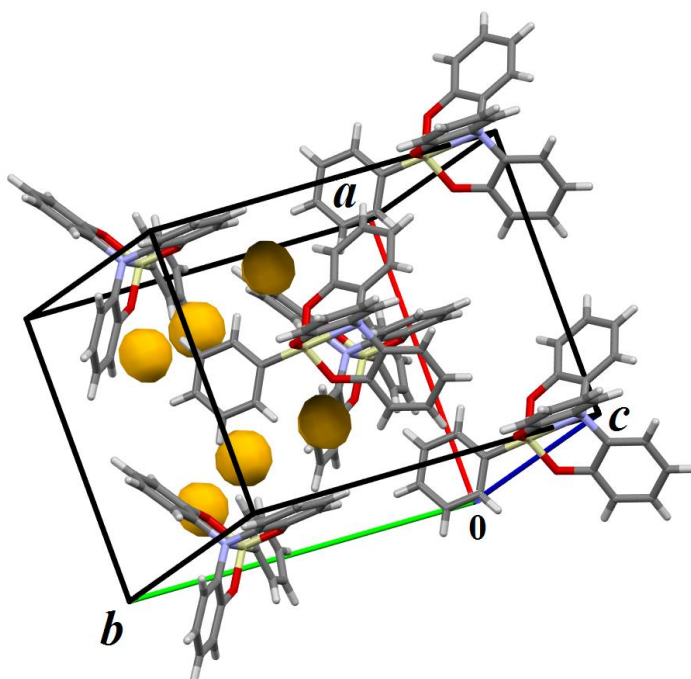


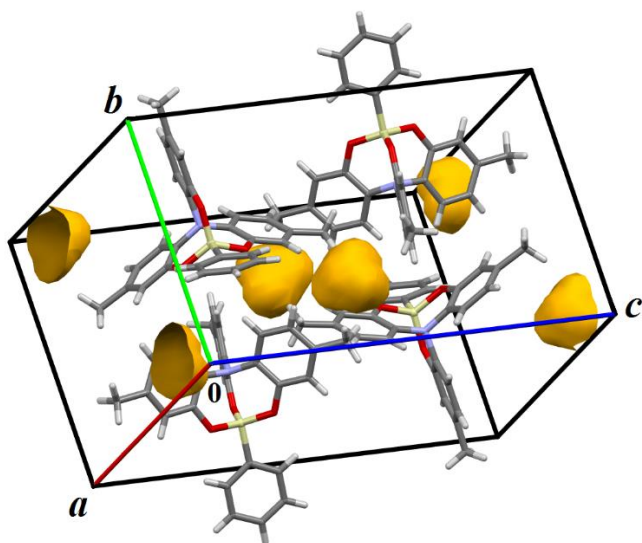
Supplementary Information

## Crystal Structure of New 1-Phenyl Substituted Tribenzsilatrane

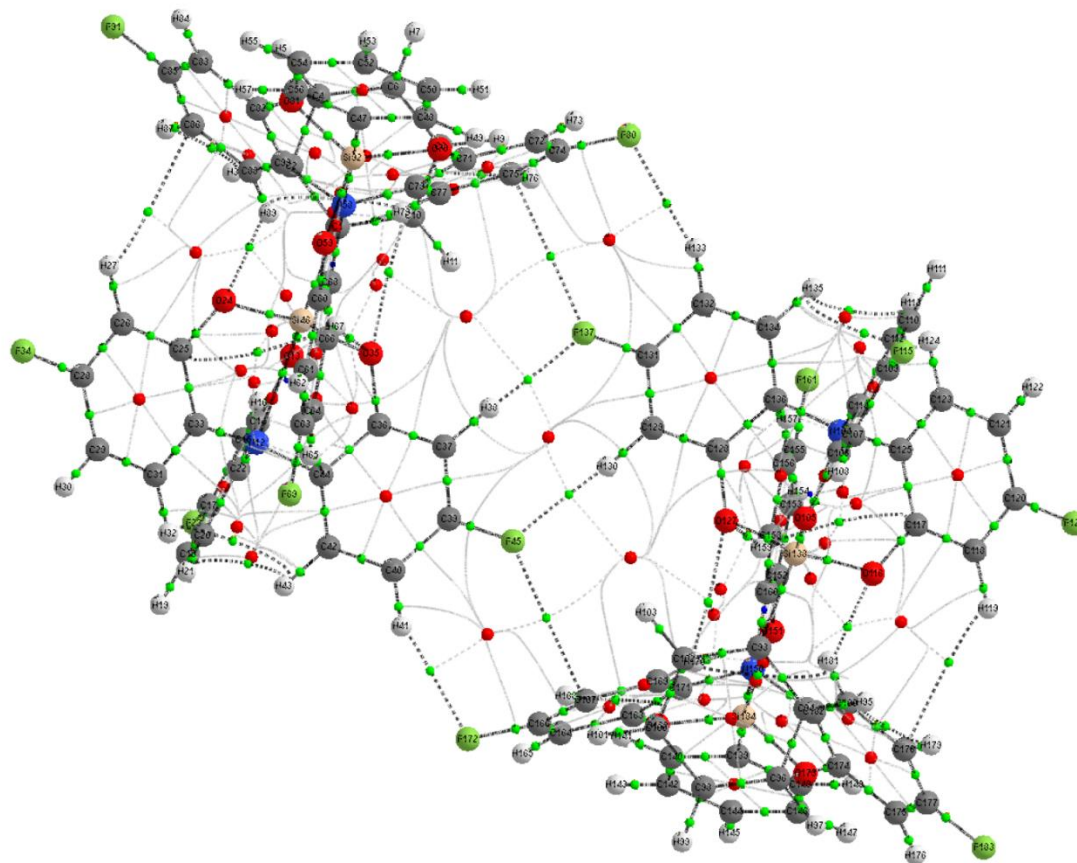
V. Romanovs, S. Belyakov, E. Doronina, V. Sidorkin, T. Roisnel, V. Jouikov



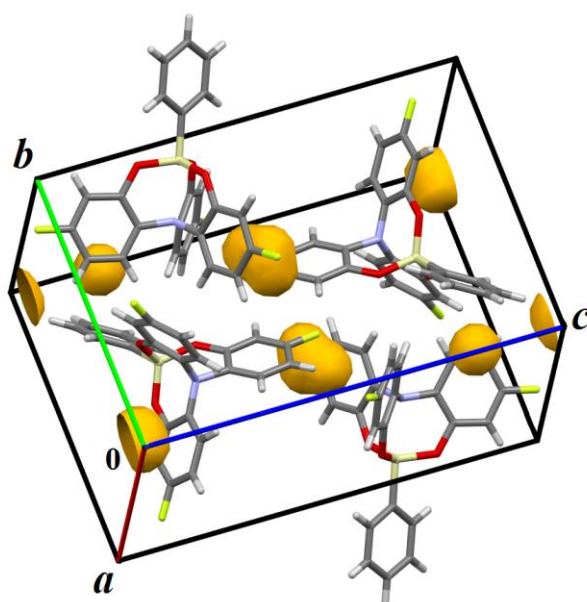
**Figure S1.** The voids in the molecular packing of **2a**.



**Figure S2.** The voids in the molecular packing of **2b**.



**Figure S3.** Crystal cell graph for **2c** from AIM analysis. Small green spheres represent bond critical points  $bcp(3, -1)$ , red spheres are ring critical points  $rcp(3, 1)$ .

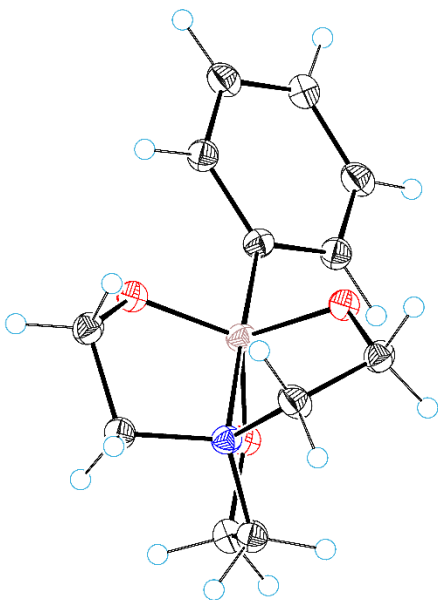


**Figure S4.** The voids in the molecular packing of **2c**.

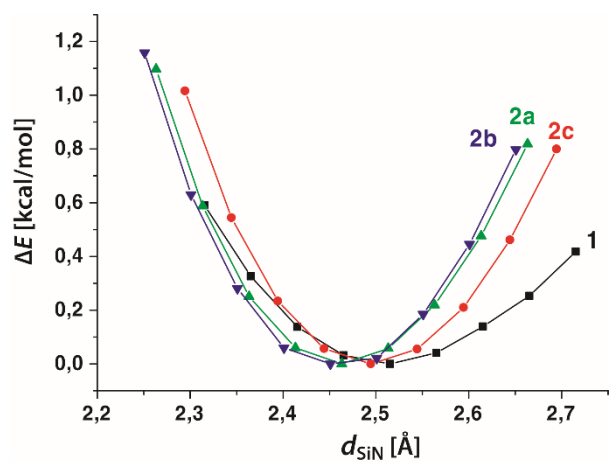
**Table S1.** XRD Experimental and B3PW91/6-311++G(d,p) calculated (*italics*) selected geometric parameters of molecules **2a-c** (bond lengths in Å, angles in °).

Parameter	<b>2a</b>		<b>2b</b>		<b>2c</b>	
	exp.	<i>calc.</i>	exp.	<i>calc.</i>	exp.	<i>calc.</i>
Si1–O2	1.654(8)	<i>1.681</i>	1.657(2)	<i>1.684</i>	1.655(2)	<i>1.683</i>
Si1–O8	1.645(6)	<i>1.683</i>	1.663(2)	<i>1.682</i>	1.658(2)	<i>1.682</i>
Si1–O9	1.645(6)	<i>1.682</i>	1.658(2)	<i>1.682</i>	1.660(2)	<i>1.681</i>
Si1–C12	1.863(11)	<i>1.861</i>	1.850(3)	<i>1.863</i>	1.852(3)	<i>1.857</i>
O2–C3	1.342(14)	<i>1.353</i>	1.370(4)	<i>1.353</i>	1.369(4)	<i>1.349</i>
O8–C7	1.341(8)	<i>1.352</i>	1.372(4)	<i>1.353</i>	1.366(4)	<i>1.349</i>
O9–C10	1.341(8)	<i>1.352</i>	1.374(4)	<i>1.353</i>	1.365(4)	<i>1.349</i>
C3–C4	1.400(15)	<i>1.399</i>	1.387(4)	<i>1.398</i>	1.392(4)	<i>1.400</i>
C6–C7	1.390(11)	<i>1.399</i>	1.376(4)	<i>1.398</i>	1.388(4)	<i>1.400</i>
C10–C11	1.390(11)	<i>1.399</i>	1.381(4)	<i>1.397</i>	1.390(4)	<i>1.400</i>
N5–C4	1.443(13)	<i>1.438</i>	1.446(4)	<i>1.440</i>	1.444(4)	<i>1.437</i>
N5–C6	1.426(8)	<i>1.439</i>	1.451(4)	<i>1.438</i>	1.449(4)	<i>1.436</i>
N5–C11	1.426(8)	<i>1.438</i>	1.458(4)	<i>1.439</i>	1.441(4)	<i>1.436</i>
O2–Si1–O8	117.1(3)	<i>115.0</i>	116.0(1)	<i>115.4</i>	116.2(1)	<i>114.7</i>
O2–Si1–O9	117.1(3)	<i>116.8</i>	117.8(1)	<i>115.2</i>	116.3(1)	<i>114.6</i>
O8–Si1–O9	117.2(5)	<i>115.1</i>	117.5(1)	<i>117.1</i>	116.8(1)	<i>116.5</i>
O2–Si1–C12	100.3(4)	<i>102.1</i>	98.9(1)	<i>102.2</i>	102.5(1)	<i>103.2</i>
O8–Si1–C12	99.7(3)	<i>102.6</i>	101.2(1)	<i>101.7</i>	101.2(1)	<i>102.6</i>
O9–Si1–C12	99.7(3)	<i>102.1</i>	99.7(1)	<i>101.7</i>	99.5(1)	<i>102.5</i>
Si1–O2–C3	126.3(7)	<i>126.4</i>	125.7(2)	<i>126.1</i>	125.9(2)	<i>126.8</i>
Si1–O8–C7	125.3(6)	<i>126.2</i>	124.5(2)	<i>126.2</i>	126.1(2)	<i>127.0</i>
Si1–O9–C10	125.3(6)	<i>126.3</i>	124.9(2)	<i>126.2</i>	126.1(2)	<i>127.0</i>
O2–C3–C4	118.2(11)	<i>119.7</i>	118.1(3)	<i>119.6</i>	119.2(3)	<i>120.0</i>
O8–C7–C6	119.1(11)	<i>119.7</i>	119.5(3)	<i>119.5</i>	119.5(3)	<i>119.8</i>
O9–C10–C11	119.1(11)	<i>119.6</i>	119.9(3)	<i>119.6</i>	119.5(3)	<i>119.9</i>
C3–C4–N5	113.3(10)	<i>114.7</i>	113.9(3)	<i>114.8</i>	114.4(3)	<i>115.0</i>
C7–C6–N5	113.3(9)	<i>114.7</i>	114.1(3)	<i>114.7</i>	113.9(3)	<i>114.9</i>
C10–C11–N5	113.3(9)	<i>114.6</i>	112.8(3)	<i>114.8</i>	113.7(3)	<i>115.0</i>
C4–N5–C6	116.5(7)	<i>117.1</i>	114.7(2)	<i>117.3</i>	116.8(2)	<i>117.4</i>
C4–N5–C11	116.5(7)	<i>117.4</i>	116.1(2)	<i>117.0</i>	116.8(2)	<i>117.4</i>
C6–N5–C11	114.3(9)	<i>117.1</i>	117.4(2)	<i>117.0</i>	116.4(2)	<i>117.7</i>
N5–Si1–C12	179.8(4)	<i>179.6</i>	178.2(1)	<i>179.6</i>	178.0(1)	<i>179.5</i>
Si1–N5	2.329(9)	<i>2.463</i>	2.358(3)	<i>2.451</i>	2.411(3)	<i>2.494</i>
ΔN	0.300(9)	<i>0.357</i>	0.292(4)	<i>0.247</i>	0.267(4)	<i>0.230</i>
ΔSi	0.284(7)	<i>0.243</i>	0.286(3)	<i>0.346</i>	0.318(3)	<i>0.372</i>
η <sub>e</sub>	72.4	<i>58.4</i>	72.2	<i>60.9</i>	65.8	<i>54.9</i>
Σ(O–Si–O)	351.4(3)	<i>346.9</i>	351.3(1)	<i>347.7</i>	349.3(1)	<i>345.8</i>
Σ(C–N–C)	347.4(7)	<i>351.6</i>	348.2(2)	<i>351.3</i>	350.0(2)	<i>352.5</i>
Si–O–C–C	-3.3(3)	<i>11.5</i>	-5.1(5)	<i>10.1</i>	-9.1(2)	<i>11.6</i>
	0.0(0)	<i>11.8</i>	2.0(9)	<i>10.4</i>	-7.1(6)	<i>11.9</i>

	3.3(3)	12.2	-8.0(4)	9.7	-9.7(4)	11.1
O-C-C-N	2.0(8)	1.4	0.2(0)	1.6	-0.7(6)	1.6
	0.0(0)	1.8	-5.7(8)	1.9	-2.8(8)	1.9
	-2.0(8)	2.0	2.1(5)	1.2	0.0(8)	1.3
C-C-N-Si	-0.4(6)	-8.0	2.7(7)	-7.5	6.0(1)	-8.1
	0.0(0)	-8.6	5.4(9)	-7.9	7.3(3)	-8.5
	0.4(6)	-9.0	2.2(3)	-6.8	5.4(0)	-7.5
C-N-Si-O	-0.8(4)	10.5	-4.1(5)	9.6	-8.2(1)	10.6
	0.0(0)	11.1	-3.7(0)	10.1	-8.5(6)	11.0
	0.8(4)	11.6	-4.8(0)	9.0	-7.9(1)	9.9
N-Si-O-C	2.2(2)	-12.1	5.1(4)	-10.9	9.5(7)	-12.2
	0.0(0)	-12.7	1.1(7)	-11.3	8.7(9)	-12.7
	-2.2(2)	-13.2	6.9(2)	-10.4	9.6(6)	-11.6



**Figure S5.** ORTEP drawing of phenylsilatrane (**1**) (first reported in [1]). Thermal ellipsoids are shown with 50% probability; hydrogen atoms are placed at a fixed distance using the riding model and shown as small spheres of arbitrary radii. The molecular structure of phenylsilatrane is practically the same for all three ( $\alpha$ ,  $\beta$ ,  $\gamma$ ) crystal modifications, only their crystal packaging being different.



**Figure S6.** Potential functions of deformation of the N→Si bond in “common” silatrane Ph-Si(OCH<sub>2</sub>CH<sub>2</sub>)<sub>3</sub>N **1** and tribenzsilatranes **2a-c**.

## References

- [1] L. Parkanyi, K. Simon, J. Nagy, *Acta Crystallogr. Sec. B*, **1974**, *30*, 2328.