

Article

# Three Novel Thiazole-Arm Containing 1,3,4-Oxadiazole-Based [HS-HS] Fe(II) Dinuclear Complexes

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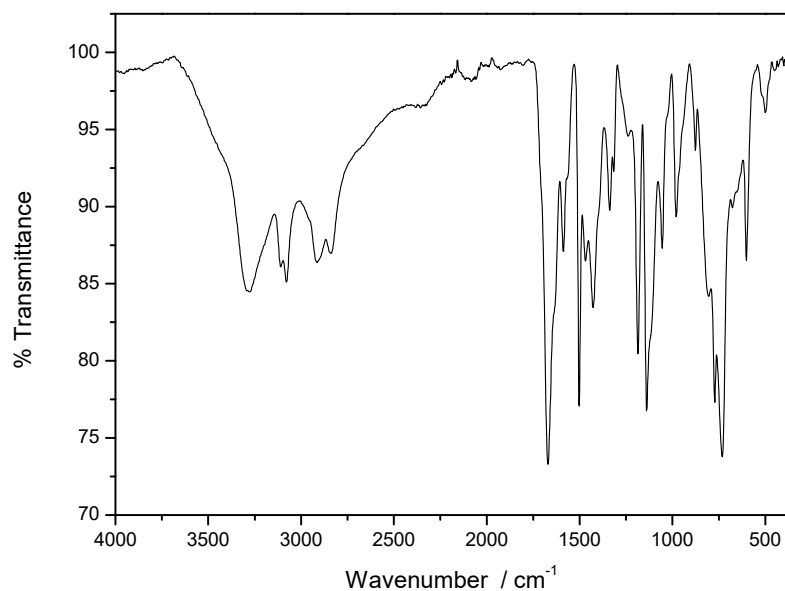
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## Electronic Supplementary Information

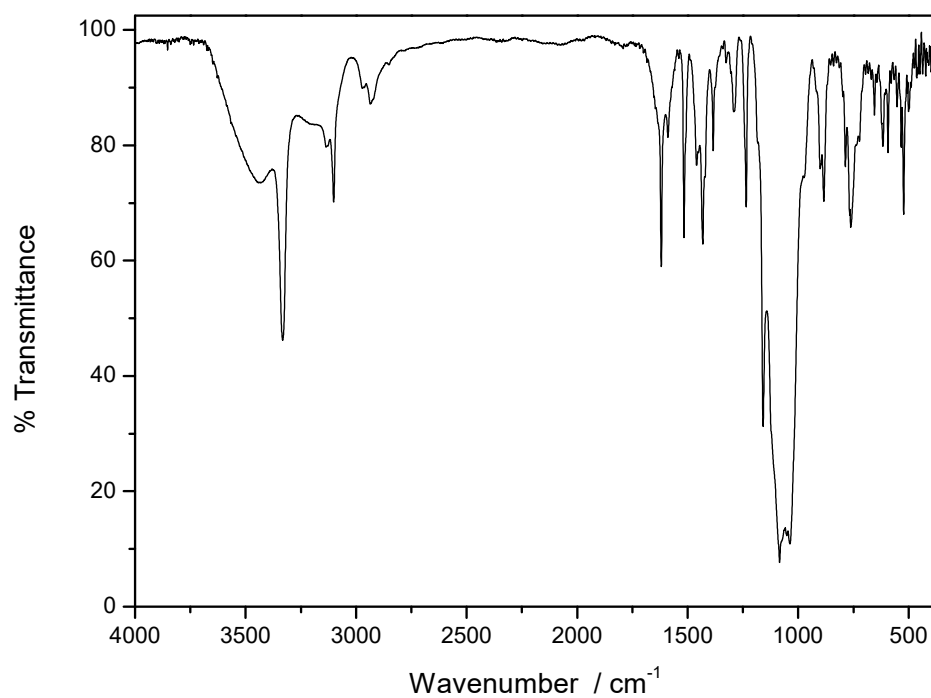
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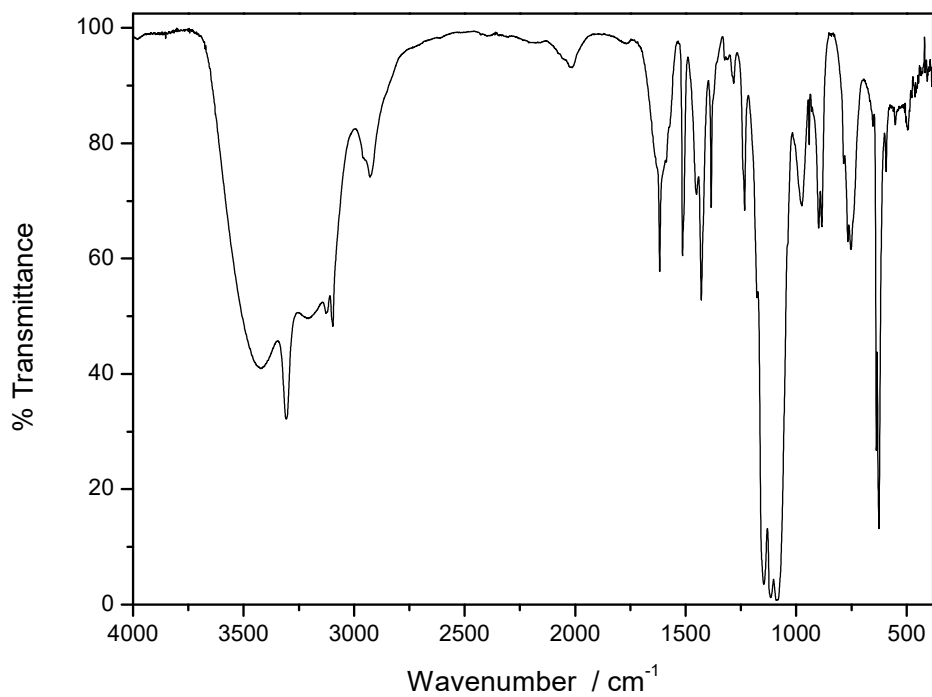
## S1. Infrared Spectra



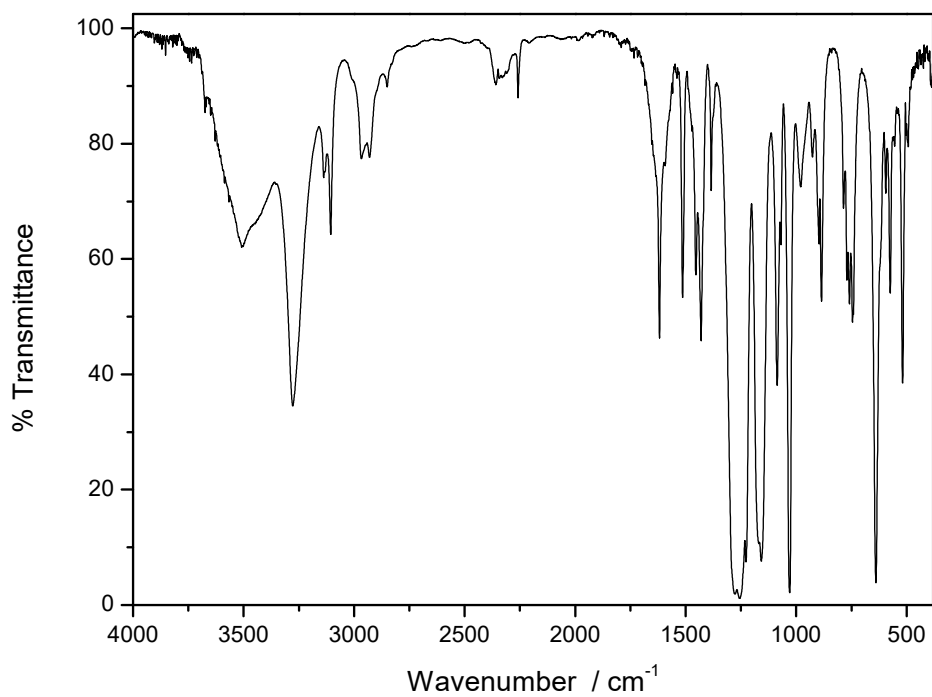
**Figure S1:** IR spectrum of 2,5-Bis[(1,3-thiazol-2-ylmethyl)amino]methyl-1,3,4-oxadiazole ( $L^{\text{Thiazole}}$ ).



**Figure S2:** IR spectrum of  $[\text{Fe}_2(\mu_2-L^{\text{Thiazole}})_2](\text{BF}_4)_4 \cdot 2\text{MeCN}$  (**1**).

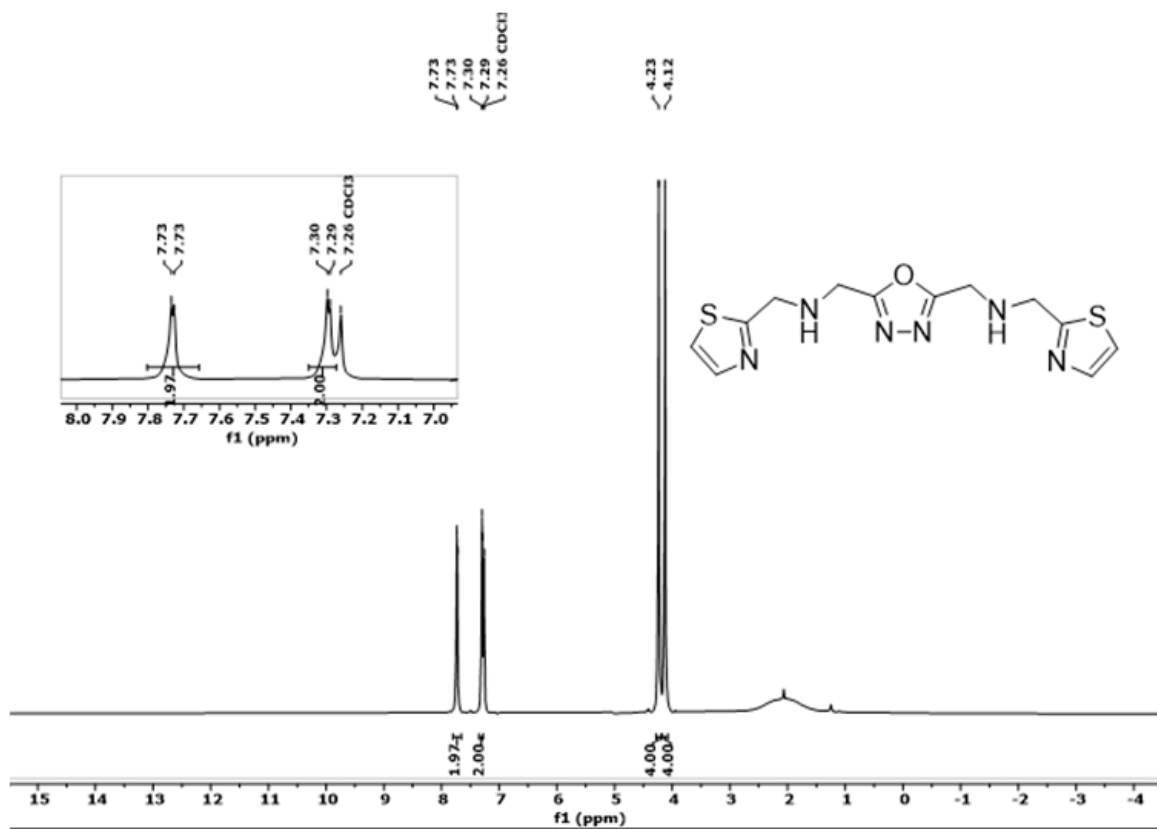


**Figure S3:** IR spectrum of  $[\text{Fe}_2(\mu_2\text{-L}^{\text{Thiazole}})_2](\text{ClO}_4)_4$  (**2**)

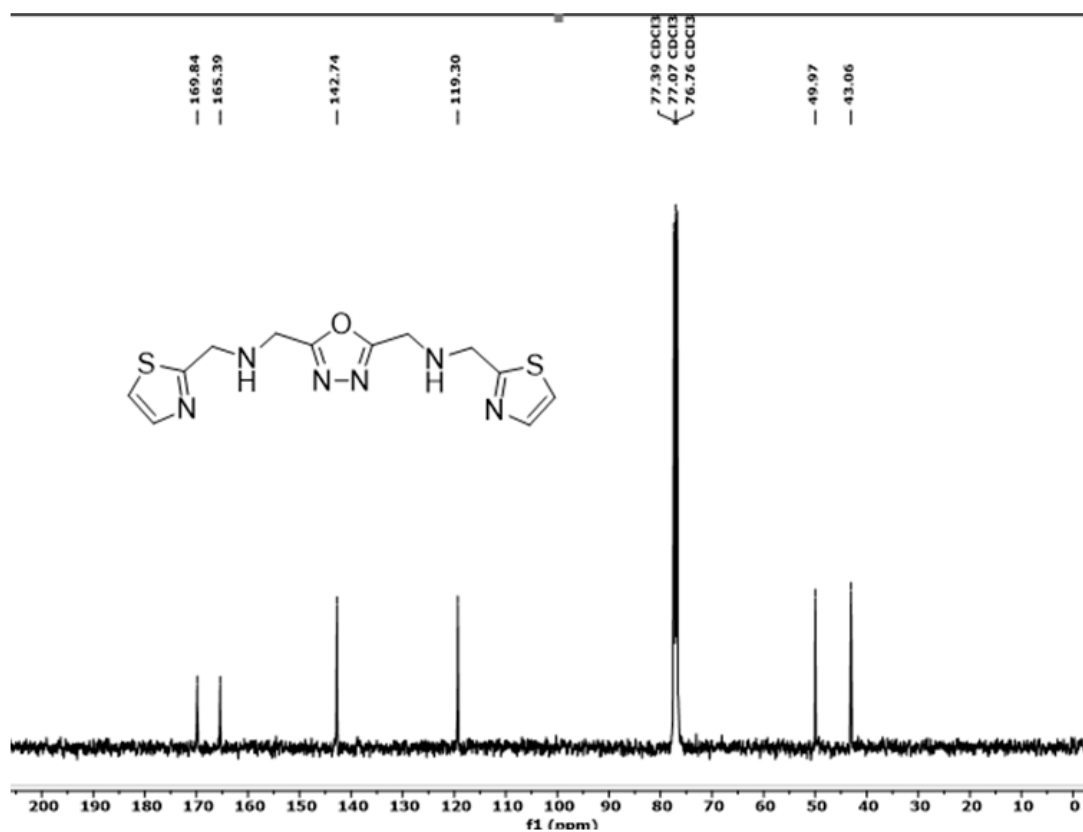


**Figure S4:** IR spectrum of  $[\text{Fe}_2(\mu_2\text{-L}^{\text{Thiazole}})_2](\text{CF}_3\text{SO}_3)_4 \cdot 2\text{MeCN}$  (**3**).

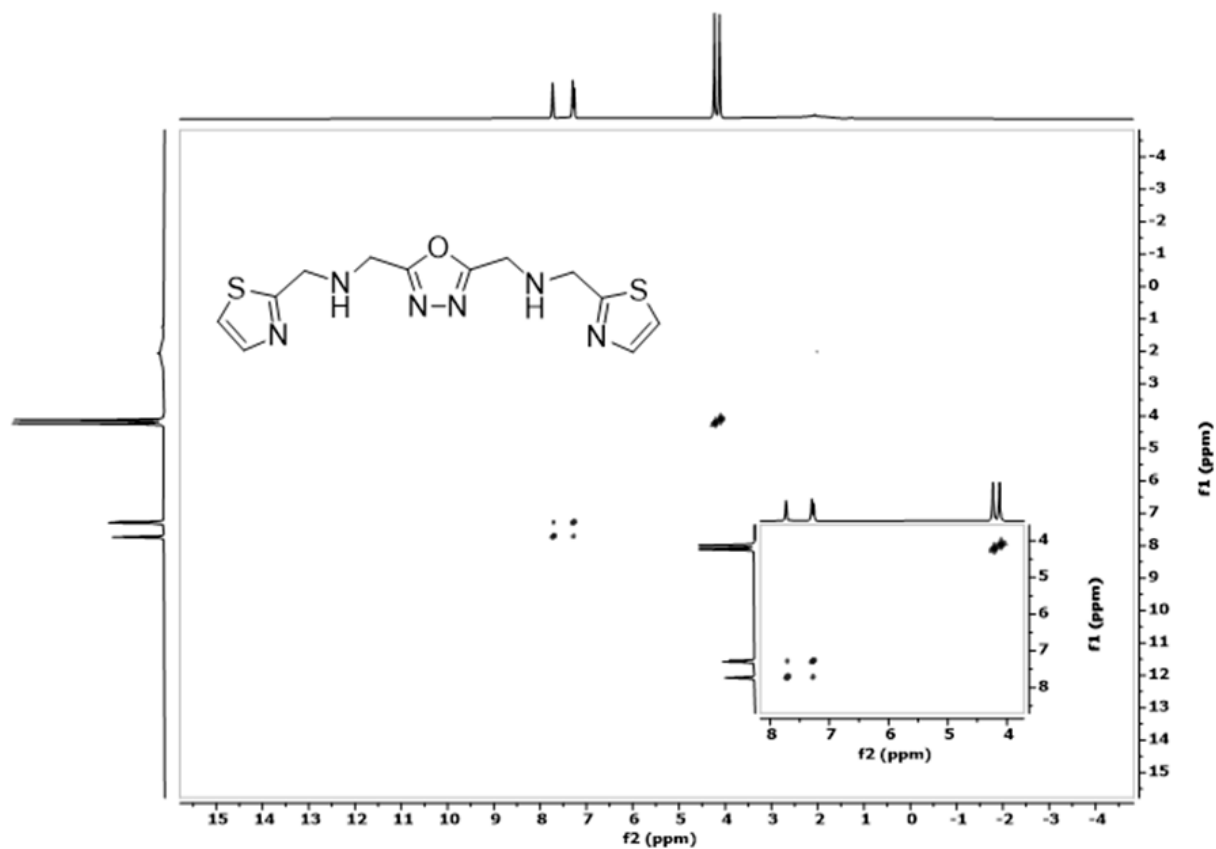
## S2. NMR Spectroscopy



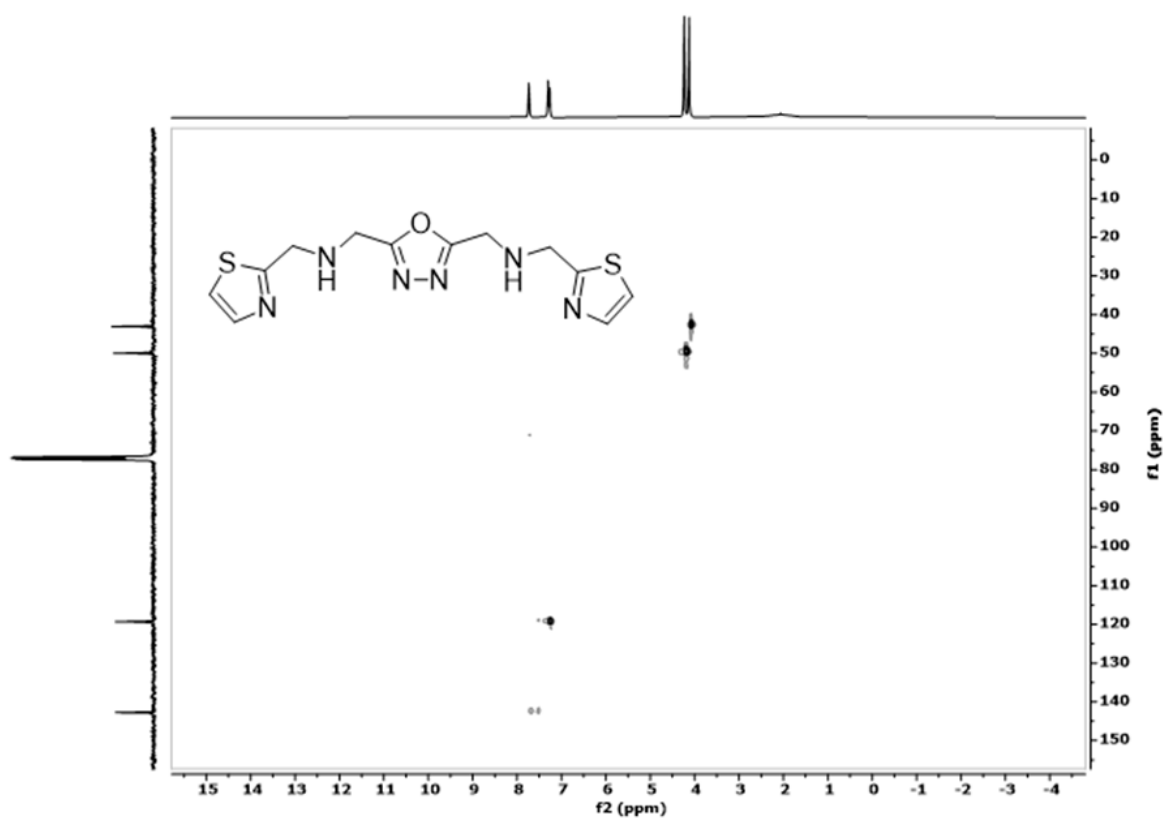
**Figure S5:**  $^1\text{H}$ -NMR spectrum of 2,5-Bis[(1,3-thiazol-2-ylmethyl)amino]methyl-1,3,4-oxadiazole ( $\text{L}^{\text{Thiazole}}$ ) in  $\text{CDCl}_3$ .



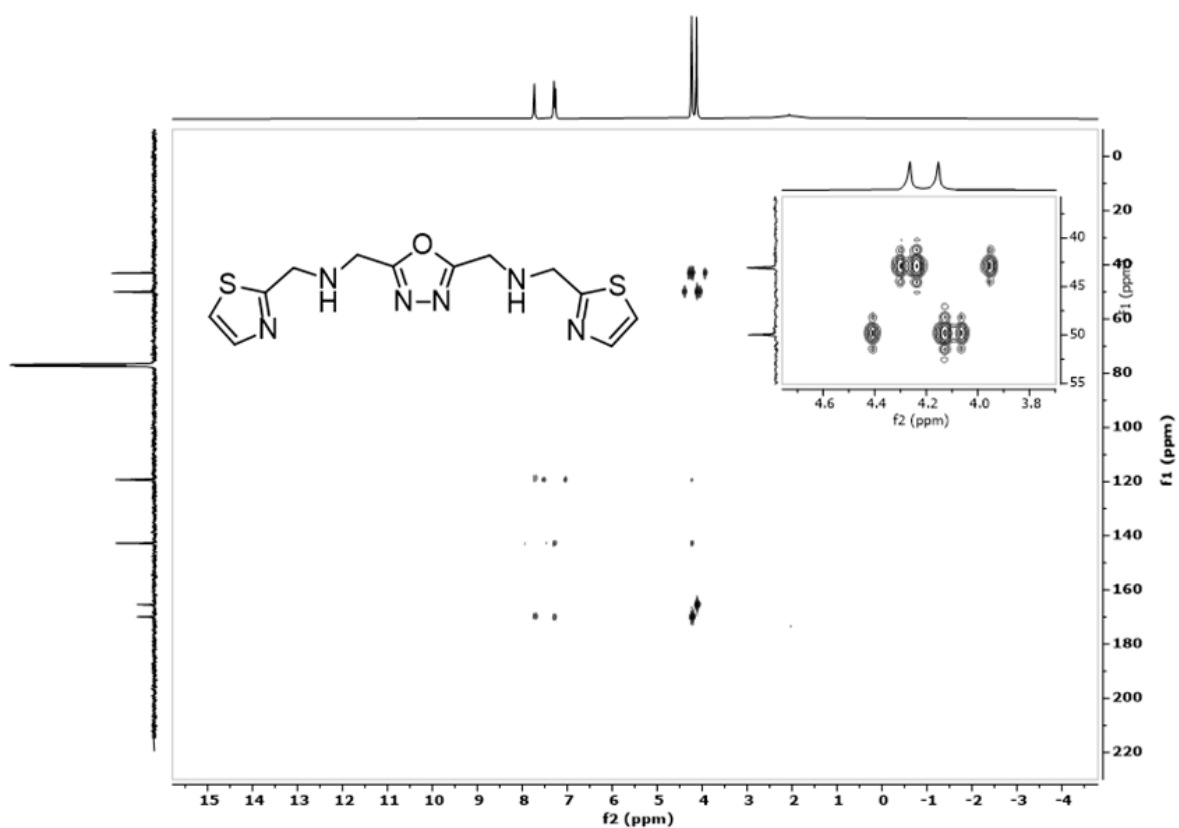
**Figure S6:**  $^{13}C$ -NMR spectrum of 2,5-Bis[(1,3-thiazol-2-ylmethyl)amino]methyl-1,3,4-oxadiazole ( $L^{Thiazole}$ ) in  $CDCl_3$ .



**Figure S7:**  $^1H$ - $^1H$ -COSY spectrum of 2,5-Bis[(1,3-thiazol-2-ylmethyl)amino]methyl-1,3,4-oxadiazole ( $L^{Thiazole}$ ) in  $CDCl_3$ .

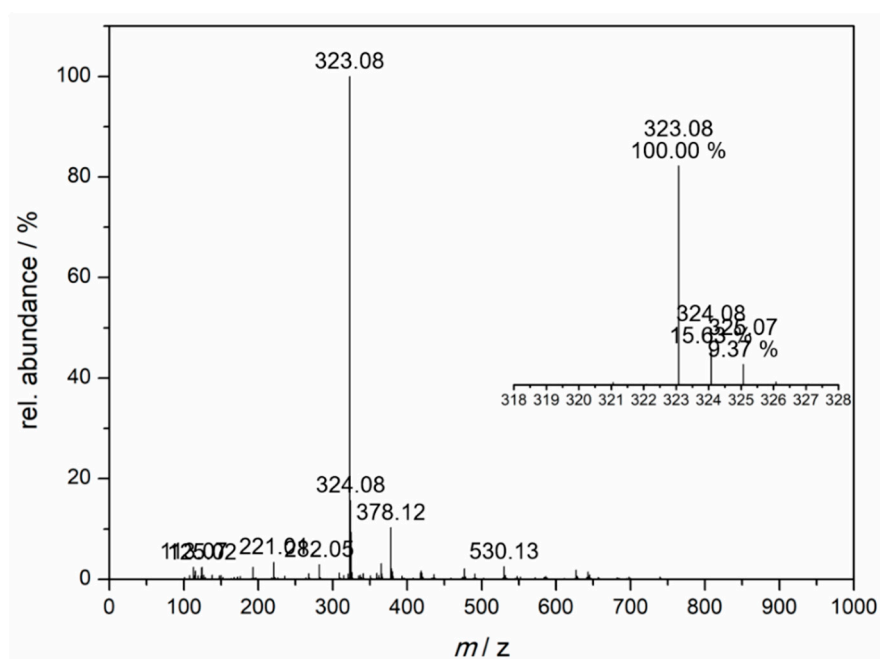


**Figure S8:**  $^1\text{H}$ - $^{13}\text{C}$ -HSQC spectrum of 2,5-Bis[(1,3-thiazol-2-ylmethyl)amino]methyl-1,3,4-oxadiazole ( $\text{L}^{5,\text{TA}}$ ) in  $\text{CDCl}_3$ .



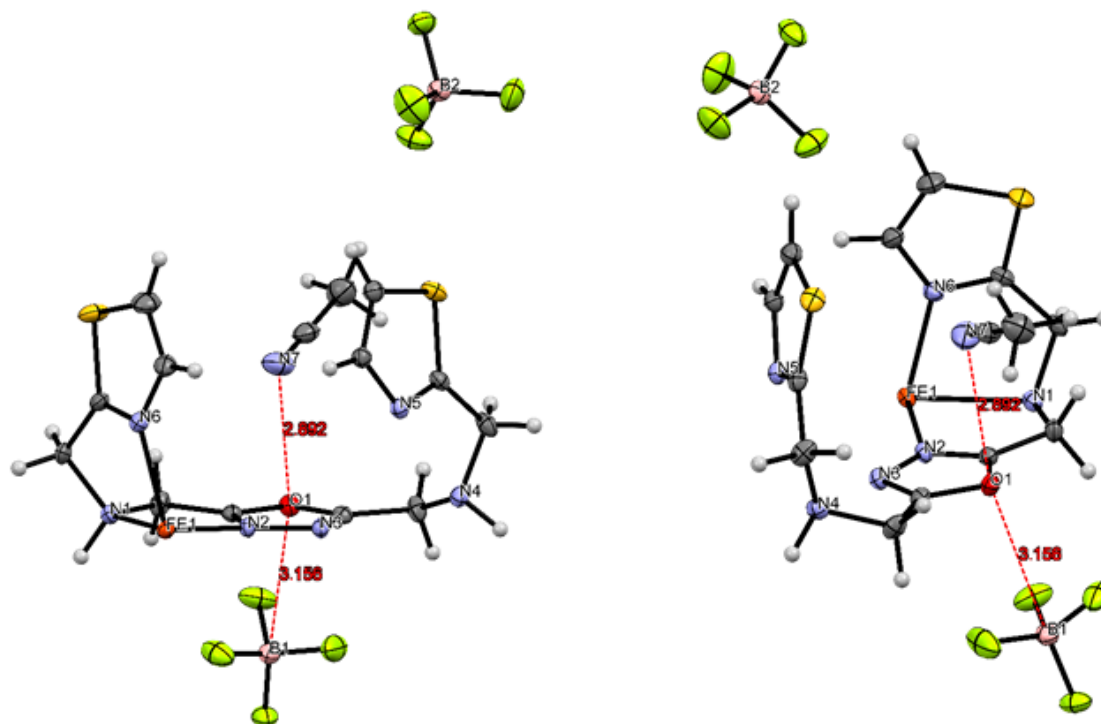
**Figure S9:**  $^1\text{H}$ - $^{13}\text{C}$ -HMBC spectrum of 2,5-Bis[(1,3-thiazol-2-ylmethyl)amino]methyl-1,3,4-oxadiazole ( $L^{5,TA}$ ) in  $\text{CDCl}_3$ .

### S3. FD Mass spectroscopy

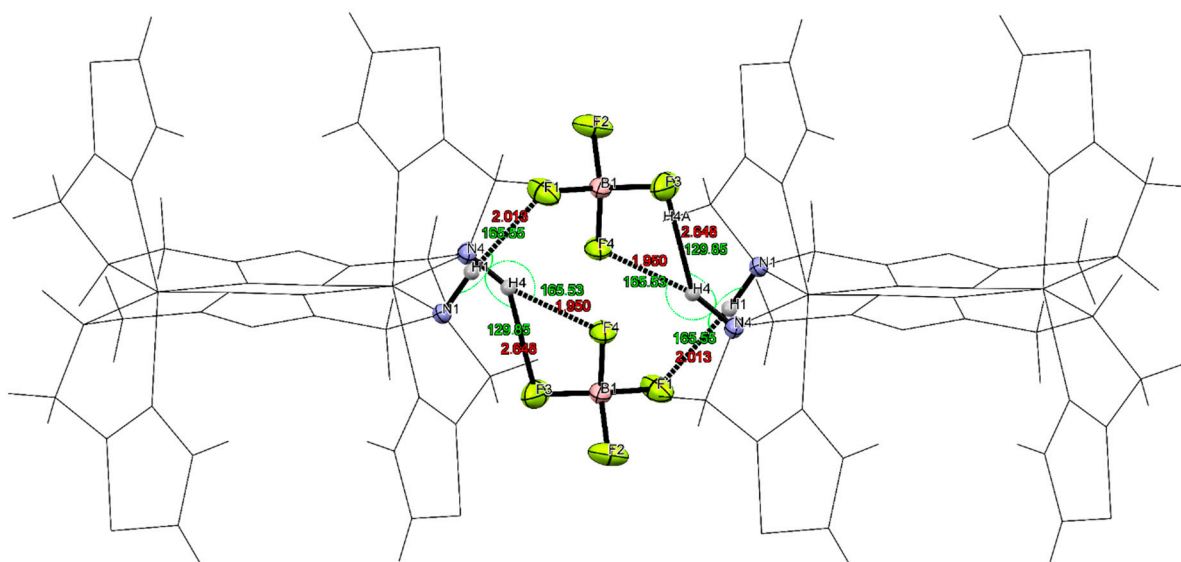


**Figure S10:** FD mass spectrum of 2,5-Bis[(1,3-thiazol-2-ylmethyl)amino]methyl-1,3,4-oxadiazole ( $L^{\text{Thiazole}}$ ).

## S4. Crystallography

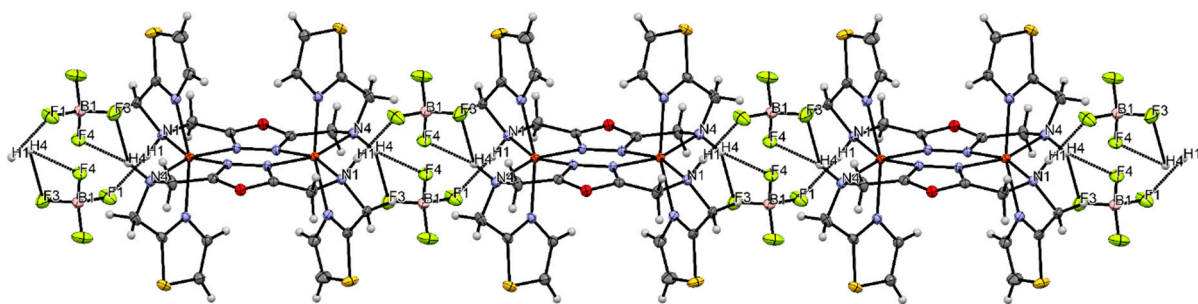


**Figure S11:** Asymmetric Unit of **1** with front view (left) and side view (right). Measured distances are displayed as red dashed lines. ORTEP representation with atomic displacement parameters set to 50 % probability.

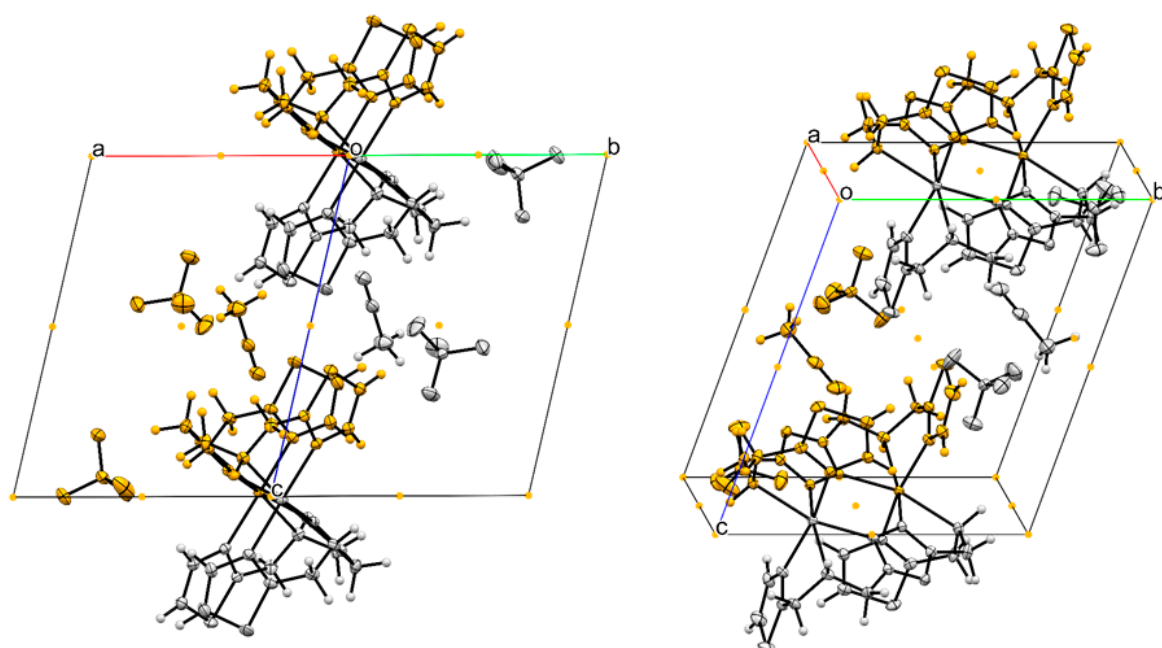


**Figure S12:** Unique hydrogen bonds (dashed black lines) in **1** between complex cations and tetrafluoroborate anions. Acetonitrile molecules and irrelevant tetrafluoroborate anions are omitted for clarity. N-H...F distances are displayed in red. N-H-F angles are shown in green. ORTEP representation with atomic displacement parameters set to 50 % probability.

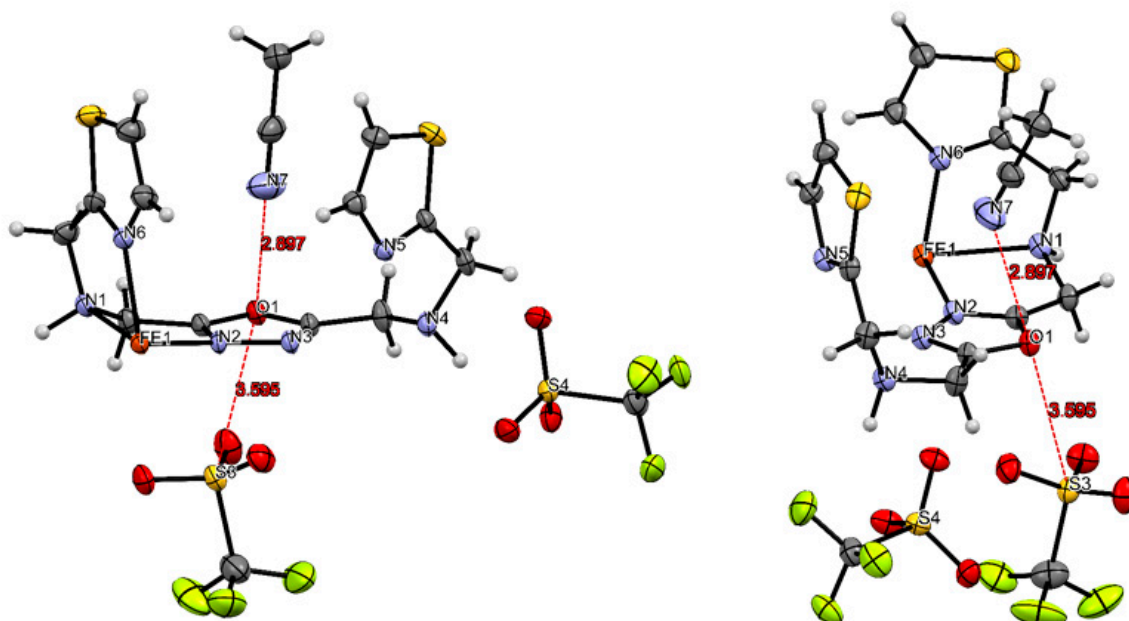




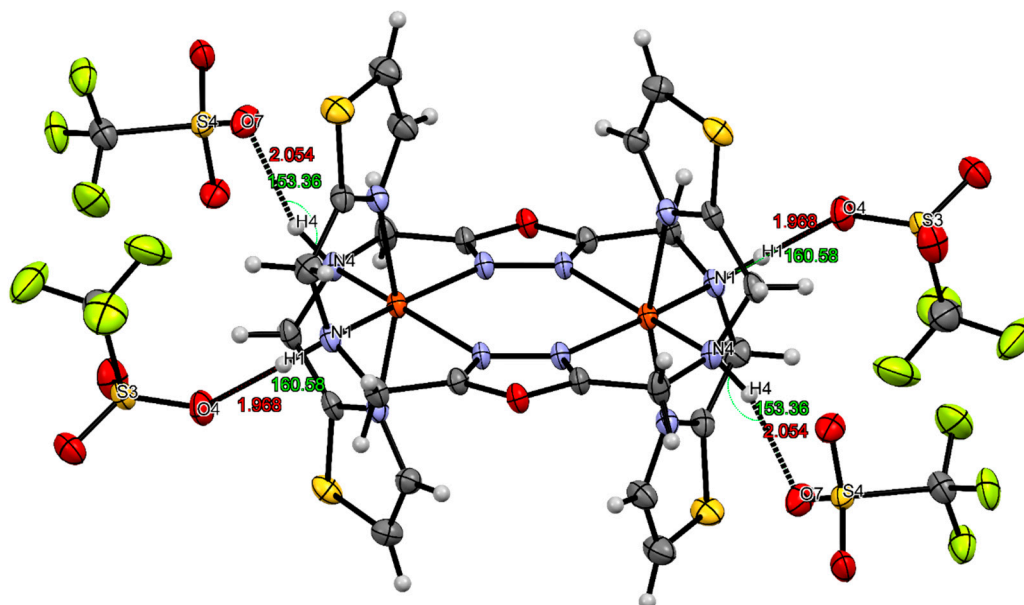
**Figure S13:** One-dimensional network of complex cations via hydrogen bonds (dashed black lines) in **1**. Acetonitrile molecules and irrelevant tetrafluoroborate anions are omitted for clarity. ORTEP representation with atomic displacement parameters set to 50 % probability.



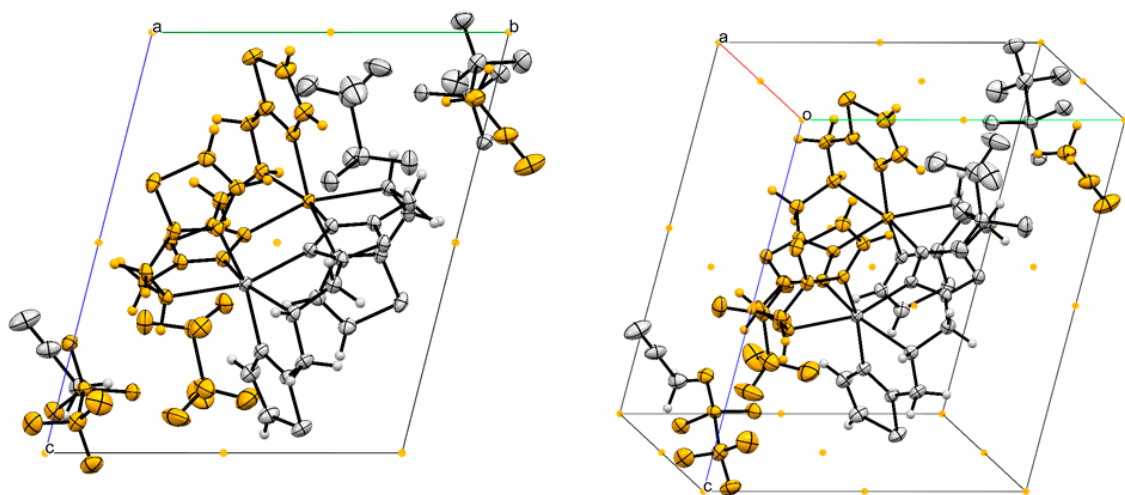
**Figure S14:** Unit cell of **1** with view along the bisector of  $\gamma$  (left) and perspective (right). The colour scheme is set according to symmetry equivalent moieties. Yellow dots represent a center of inversion. ORTEP representation with atomic displacement parameters set to 50 % probability.



**Figure S15:** Asymmetric Unit of **3** with front view (left) and side view (right). Measured distances are displayed as red dashed lines. ORTEP representation with atomic displacement parameters set to 50 % probability.



**Figure S16:** Unique hydrogen bonds (dashed black lines) in **3** between complex cations and tetrafluoroborate anions. Acetonitrile molecules and irrelevant tetrafluoroborate anions are omitted for clarity. N-H...O distances are displayed in red. N-H-O angles are shown in green. ORTEP representation with atomic displacement parameters set to 50 % probability.



**Figure S17:** Unit cell of **3** with view along the a-axis (left) and perspective (right). The colour scheme is set according to symmetry equivalent moieties. Yellow dots represent a center of inversion. ORTEP representation with atomic displacement parameters set to 50 % probability.

**Table S1:** Selected angles of complex 1 and 3. Labeling of the atoms follows the scheme in Figure 3 (left).

	<b>1</b> 120 K	<b>3</b> 120 K
<i>cis</i> -Angles / °		
N1-Fe-N2	74.57(4)	74.69(10)
N2-Fe-N3	90.63(4)	90.67(10)
N3-Fe-N4	75.64(4)	74.47(10)
N4-Fe-N1	120.95(4)	122.24(10)
N1-Fe-N5	97.88(4)	96.63(10)
N2-Fe-N5	94.31(4)	92.48(11)
N3-Fe-N5	94.61(4)	95.11(11)
N4-Fe-N5	76.26(4)	76.07(10)
N1-Fe-N6	75.80(4)	75.84(11)
N2-Fe-N6	98.43(4)	101.61(11)
N3-Fe-N6	95.53(4)	96.75(11)
N4-Fe-N6	93.83(4)	93.41(11)
<i>trans</i> -Angles / °		
N1-Fe-N3	161.24(4)	161.56(10)
N2-Fe-N4	162.43(4)	160.07(11)
N5-Fe-N6	163.60(4)	161.42(11)
Torsions / °		
Fe-N2-N3-Fe1'	9.03(18)	10