

Supporting information for

Old Donors for New Molecular Conductors: Combining TMTSF and BEDT-TTF with Anionic $(\text{TaF}_6)_{1-x}/(\text{PF}_6)_x$ Alloys

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Single crystal X-ray crystallography

Table S1. Crystal Data and Structure Refinement for (TMTSF)₂(TaF₆)_{0.12}(PF₆)_{0.88}, (TMTSF)₂(TaF₆)_{0.44}(PF₆)_{0.56}, (TMTSF)₂(TaF₆)_{0.56}(PF₆)_{0.44} and (TMTSF)₂(TaF₆)_{0.84}(PF₆)_{0.16}.

Compound	(TMTSF) ₂ (TaF ₆) _{0.12} (PF ₆) _{0.88}	(TMTSF) ₂ (TaF ₆) _{0.44} (PF ₆) _{0.56}	(TMTSF) ₂ (TaF ₆) _{0.56} (PF ₆) _{0.44}	(TMTSF) ₂ (TaF ₆) _{0.84} (PF ₆) _{0.16}
Empirical formula	C ₂₀ H ₂₄ F ₆ P _{0.88} Se ₈ Ta _{0.12}	C ₂₀ H ₂₄ F ₆ P _{0.56} Se ₈ Ta _{0.44}	C ₂₀ H ₂₄ F ₆ P _{0.44} Se ₈ Ta _{0.56}	C ₂₀ H ₂₄ F ₆ P _{0.16} Se ₈ Ta _{0.84}
Molecular weight	1059.49	1107.78	1125.63	1167.02
T (K)	296(2)	293(2)	293(2)	293(2)
Wavelength (Å)	1.54184	1.54184	1.54184	1.54184
Crystal system	Triclinic	Triclinic	Triclinic	Triclinic
Space group	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
<i>a</i> (Å)	7.2868(8)	7.2916(4)	7.2939(7)	7.2911(8)
<i>b</i> (Å)	7.708(1)	7.7185(4)	7.7194(6)	7.7323(9)
<i>c</i> (Å)	13.5513(8)	13.6966(5)	13.748(1)	13.8709(8)
α(deg)	83.285(8)	83.021(4)	83.080(7)	82.821(7)
θ(deg)	86.195(6)	85.826(4)	85.783(8)	85.416(7)
γ(deg)	71.09(1)	71.339(5)	71.365(8)	71.66(1)
<i>V</i> (Å ³)	714.8(1)	724.44(6)	727.6(1)	735.8(1)
<i>Z</i>	1	1	1	1
<i>D_c</i> (g cm ⁻³)	2.461	2.539	2.569	2.634
<i>F</i> (000)	492	511	518	534
Abs coeff (mm ⁻¹)	13.734	15.534	16.196	17.691
Crystal size (mm ³)	0.188 × 0.028 × 0.016	0.141 × 0.032 × 0.024	0.13 × 0.021 × 0.012	0.092 × 0.017 × 0.016
θ (min / max)	3.285 / 73.994	3.253 / 75.812	3.240 / 76.297	3.214 / 72.067
Transmission (min/max)	0.934 / 1.000	0.285 / 0.698	0.642 / 1.000	0.973 / 1.000
Data collected/unique	5167 / 2769	6386 / 2954	5263 / 2932	5154 / 2811
Data observed	1968	2733	2402	2382
<i>R</i> (int)	0.0334	0.0195	0.0344	0.0312
GOF on <i>F</i> ²	1.024	1.038	1.021	1.028
final <i>R</i> indices ^a [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.0380, w <i>R</i> ₂ = 0.0889	<i>R</i> ₁ = 0.0274, w <i>R</i> ₂ = 0.0705	<i>R</i> ₁ = 0.0412, w <i>R</i> ₂ = 0.0988	<i>R</i> ₁ = 0.0354, w <i>R</i> ₂ = 0.0869
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.0597, w <i>R</i> ₂ = 0.0998	<i>R</i> ₁ = 0.0297, w <i>R</i> ₂ = 0.0725	<i>R</i> ₁ = 0.0522, w <i>R</i> ₂ = 0.1060	<i>R</i> ₁ = 0.0444, w <i>R</i> ₂ = 0.0929
Largest peak in final: difference (e Å ⁻³)	0.479 / -0.751	0.670 / -0.722	0.752 / -0.935	0.865 / -0.659
CCDC number	2071481	2071482	2071483	2071484

^a $R(F) = \sum ||F_o| - |F_c|| / \sum |F_o|$; $wR(F^2) = [\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]]^{1/2}$

Table S2. Crystal Data and Structure Refinement for (TMTSF)₂TaF₆ (293 K and 100 K), δ_o -(BEDT-TTF)₂(TaF₆)_{0.43}(PF₆)_{0.57}, δ_m -(BEDT-TTF)₂(TaF₆)_{0.94}(PF₆)_{0.06}, and (BEDT-TTF)₂(TaF₆)₂·CH₂Cl₂.

Compound	(TMTSF) ₂ TaF ₆	(TMTSF) ₂ TaF ₆	δ_o -(BEDT-TTF) ₂ (TaF ₆) _{0.43} (PF ₆) _{0.57}	δ_m -(BEDT-TTF) ₂ (TaF ₆) _{0.94} (PF ₆) _{0.06}	(BEDT-TTF) ₂ (TaF ₆) ₂ , CH ₂ Cl ₂
Empirical formula	C ₂₀ H ₂₄ F ₆ Se ₈ Ta	C ₂₀ H ₂₄ F ₆ Se ₈ Ta	C ₂₀ H ₁₆ F ₆ P _{0.57} S ₁₆ Ta _{0.43}	C ₂₀ H ₁₆ F ₆ P _{0.06} S ₁₆ Ta _{0.94}	C ₂₁ H ₁₈ Cl ₂ F ₁₂ S ₁₆ Ta ₂
Molecular weight	1191.02	1191.02	979.35	1055.54	1444.11
T (K)	293(2)	100(1)	295(2)	298(2)	295(2)
Wavelength (Å)	1.54184	1.54184	1.54184	1.54184	1.54184
Crystal system	Triclinic	Triclinic	Orthorhombic	Monoclinic	Triclinic
Space group	<i>P</i> -1	<i>P</i> -1	<i>Pnna</i>	<i>I</i> 2/ <i>a</i> ^b	<i>P</i> -1
<i>a</i> (Å)	7.3130(8)	7.1344(4)	14.9866(4)	14.939(1)	8.2558(2)
<i>b</i> (Å)	7.7519(7)	7.7255(5)	33.0248(7)	6.6793(4)	9.8754(4)
<i>c</i> (Å)	13.898(1)	13.7204(9)	6.6706(1)	33.584(1)	12.7497(5)
α (deg)	82.854(7)	83.378(5)	90	90	80.069(4)
β (deg)	85.340(8)	87.118(5)	90	93.914(5)	85.341(3)
γ (deg)	71.728(9)	70.200(6)	90	90	79.823(3)
<i>V</i> (Å ³)	741.6(1)	706.72(8)	3301.5(1)	3343.1(3)	1006.41(6)
<i>Z</i>	1	1	4	4	1
<i>D</i> _c (g cm ⁻³)	2.667	2.798	1.970	2.097	2.383
<i>F</i> (000)	543	543	1945	2063	688
Abs coeff (mm ⁻¹)	18.515	19.428	13.032	15.582	19.562
Crystal size (mm ³)	0.098×0.061 × 0.022	0.098×0.061 × 0.022	0.215×0.149 × 0.061	0.118×0.049 × 0.019	0.329×0.062 × 0.031
θ (min / max)	3.208 / 73.734	3.243 / 73.023	2.676 / 76.021	2.638 / 71.997	3.524 / 76.193
Transmission (min/max)	0.444 / 1.000	0.536 / 1.000	0.440 / 1.000	0.310 / 0.767	0.090 / 0.735
Data collected/unique	4659 / 2820	4463 / 2680	10452 / 3435	7179 / 3214	7612 / 4063
Data observed	2063	2524	3187	2534	3952
<i>R</i> (int)	0.0585	0.0268	0.0191	0.0338	0.0206
GOF on <i>F</i> ²	0.948	1.030	1.044	1.090	1.023
final <i>R</i> indices ^a [<i>I</i> > 2 σ (<i>I</i>)]	<i>R</i> ₁ = 0.0476, <i>wR</i> ₂ = 0.11404	<i>R</i> ₁ = 0.0317, <i>wR</i> ₂ = 0.0843	<i>R</i> ₁ = 0.0359, <i>wR</i> ₂ = 0.0854	<i>R</i> ₁ = 0.0450, <i>wR</i> ₂ = 0.1098	<i>R</i> ₁ = 0.0316, <i>wR</i> ₂ = 0.0860
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.0652, <i>wR</i> ₂ = 0.1271	<i>R</i> ₁ = 0.0341, <i>wR</i> ₂ = 0.0876	<i>R</i> ₁ = 0.0386, <i>wR</i> ₂ = 0.0875	<i>R</i> ₁ = 0.0593, <i>wR</i> ₂ = 0.1186	<i>R</i> ₁ = 0.0321, <i>wR</i> ₂ = 0.0866
Largest peak in final: difference (e Å ⁻³)	1.030 / -0.852	1.171 / -1.617	0.771 / -0.783	1.016 / -0.538	1.874 / -1.436
CCDC number	2071485	2071486	2071487	2071488	2071489

^a $R(F) = \sum ||F_o| - |F_c|| / \sum |F_o|$; $wR(F^2) = [\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]]^{1/2}$

^b Reduced cell in *P*-1 space group: *a* = 6.6738(4) Å, *b* = 14.947(1) Å, *c* = 18.213(1) Å, α = 110.307(7)°, β = 100.540(6)°, γ = 90.006(6)°, *V* = 1671.2(2) Å³

Crystal structures of TMTSF salts

Compound $(\text{TMTSF})_2(\text{TaF}_6)_{0.12}(\text{PF}_6)_{0.88}$

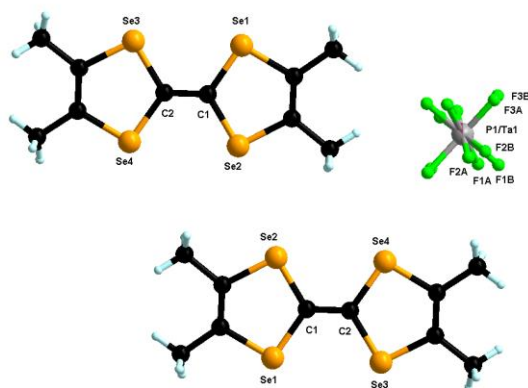


Figure S1. Molecular structure of $(\text{TMTSF})_2(\text{TaF}_6)_{0.12}(\text{PF}_6)_{0.88}$. The three independent fluorine atoms of the anion are disordered over two positions with partial refined occupancy factors of 0.78/0.22.



Figure S2. Picture of $(\text{TMTSF})_2(\text{TaF}_6)_{0.12}(\text{PF}_6)_{0.88}$ needles with nested black platelets of $(\text{TMTSF})_3\text{Ta}_2\text{F}_{10}\text{O}$.

Compounds $(\text{TMTSF})_2(\text{TaF}_6)_{0.44}(\text{PF}_6)_{0.56}$

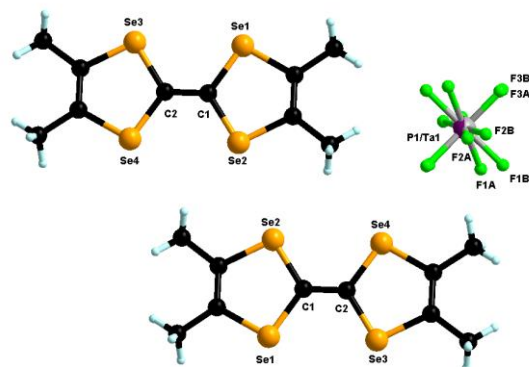


Figure S3. Molecular structure of $(\text{TMTSF})_2(\text{TaF}_6)_{0.44}(\text{PF}_6)_{0.56}$. The three independent fluoride atoms of the anion are disordered over two positions with partial refined occupancy factors of 0.76/0.24.

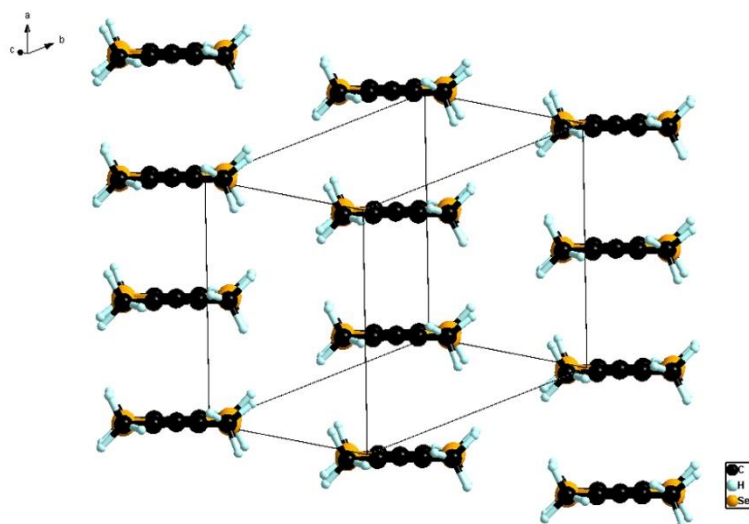


Figure S4. Packing of donors in the structure of $(\text{TMTSF})_2(\text{TaF}_6)_{0.44}(\text{PF}_6)_{0.56}$.

Compounds $(\text{TMTSF})_2(\text{TaF}_6)_{0.56}(\text{PF}_6)_{0.44}$

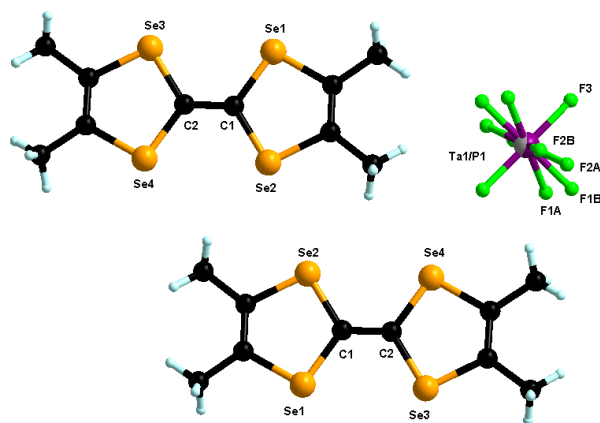


Figure S5. Molecular structure of $(\text{TMTSF})_2(\text{TaF}_6)_{0.56}(\text{PF}_6)_{0.44}$. Two of the three independent fluoride atoms of the anion are disordered over two positions with partial refined occupancy factors of 0.71/0.29.

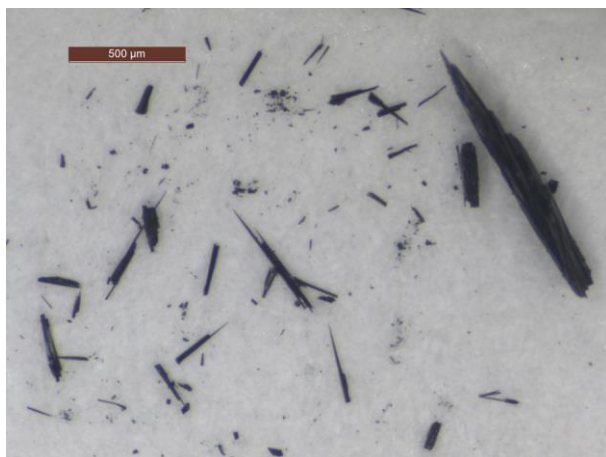


Figure S6. Picture of $(\text{TMTSF})_2(\text{TaF}_6)_{0.56}(\text{PF}_6)_{0.44}$ thin needles.

Compound $(\text{TMTSF})_2(\text{TaF}_6)_{0.84}(\text{PF}_6)_{0.16}$

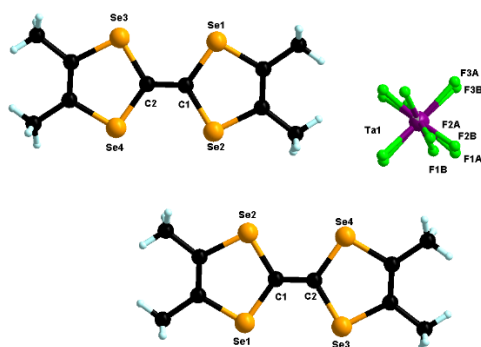


Figure S7. Molecular structure of $(\text{TMTSF})_2(\text{TaF}_6)_{0.84}(\text{PF}_6)_{0.16}$. The three independent fluoride atoms of the anion are disordered over two positions with partial refined occupancy factors of 0.61/0.39.

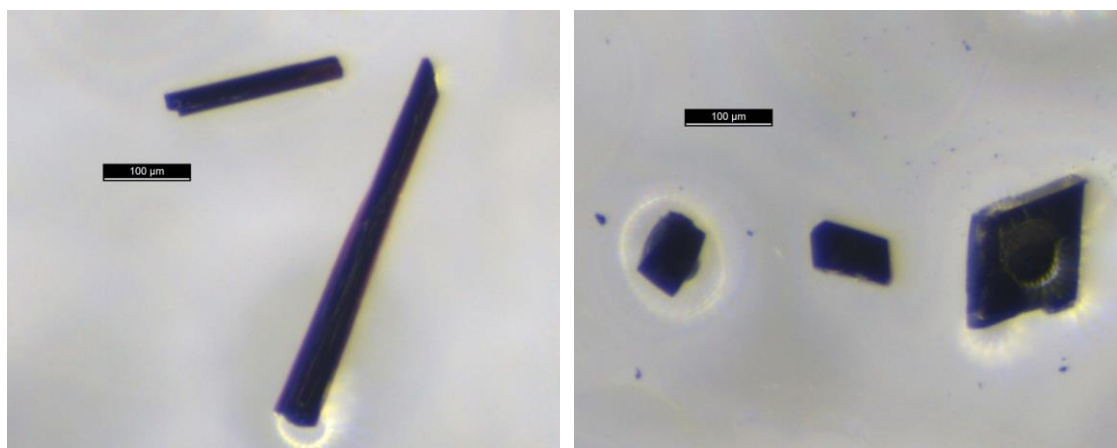


Figure S8. Picture of $(\text{TMTSF})_2(\text{TaF}_6)_{0.84}(\text{PF}_6)_{0.16}$ needles (left) and picture of $(\text{TMTSF})_3\text{Ta}_2\text{F}_{10}\text{O}$ prisms.

^{19}F NMR spectroscopy of TMTSF salts

Nuclear magnetic resonance spectra were recorded on a Bruker Avance DRX 500 spectrometer operating at 470 MHz for ^{19}F . Chemical shifts are expressed in parts per million (ppm). The following abbreviations are used: b, broad; s, singlet; d, doublet; quint, quintet; m, multiplet.



^{19}F NMR (DMSO- d_6 , 500 MHz, δ in ppm): -70.90/-69.39 (d, $J = 755$ Hz) for PF_6^- / -4.92, 18.34 (d), 33.68 (d, $\text{Ta}_2\text{F}_{10}\text{O}$, $8F_{eq}$) and 41.48 (bs, TaF_6) for TaF_6^- and $\text{Ta}_2\text{F}_{10}\text{O}$. The integration gives $\approx 68\%$ PF_6^- .

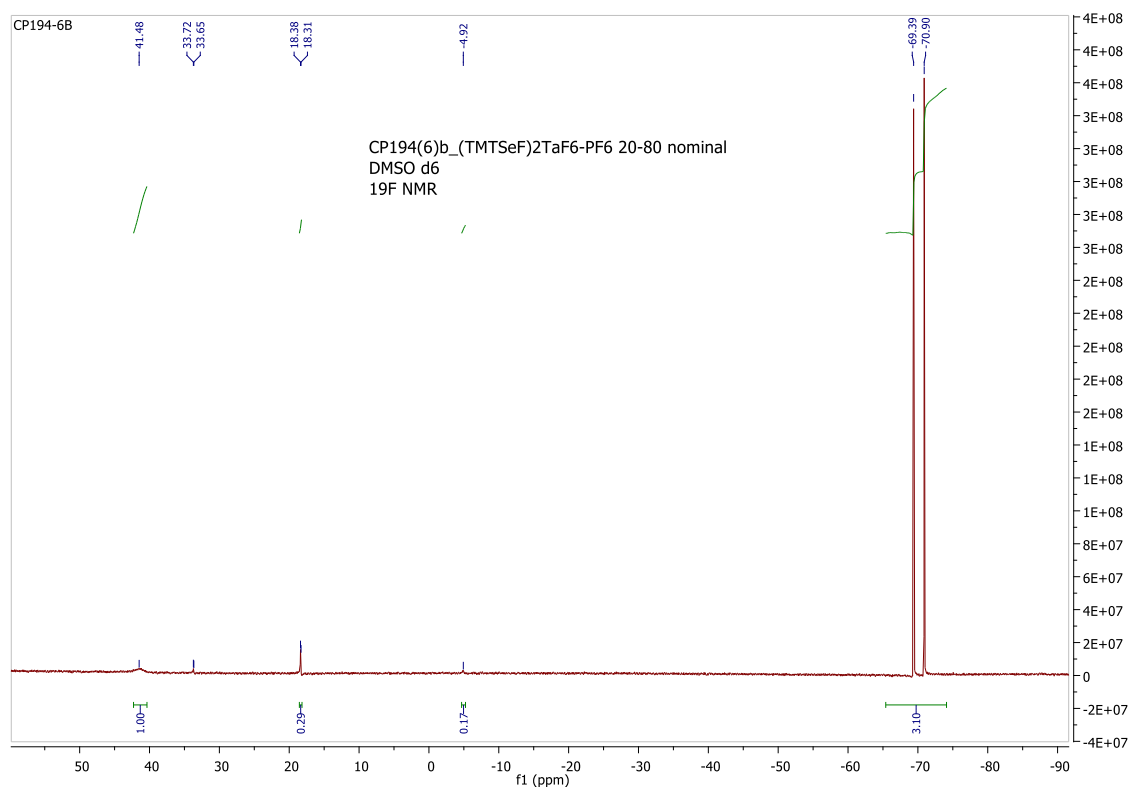


Figure S9. ^{19}F NMR (DMSO- d_6) spectrum of the crystalline material obtained by electrocrystallization of TMTSF with a mixture $(n\text{-Bu}_4\text{N})\text{TaF}_6 / (n\text{-Bu}_4\text{N})\text{PF}_6$ of nominal composition Ta/P 0.2/0.8.



^{19}F NMR (DMSO- d_6 , 500 MHz, δ in ppm): -70.90/-69.39 (d, $J = 755$ Hz) for PF_6^- / -4.90 (quint), 18.35 (d) and 41.32 (bs, TaF_6) for TaF_6^- and $\text{Ta}_2\text{F}_{10}\text{O}$. Integration gives $\approx 52\%$ of PF_6^- .

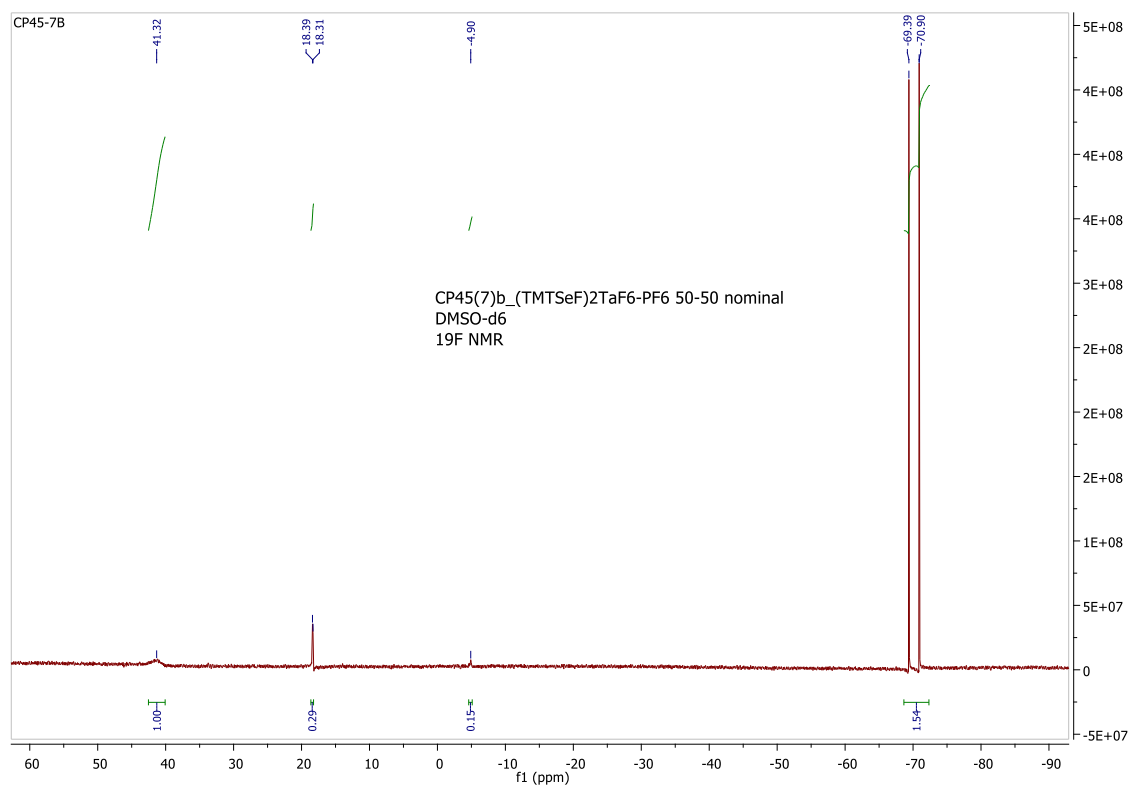


Figure S10. ^{19}F NMR (DMSO- d_6) spectrum of the crystalline material obtained by electrocrystallization of TMTSF with a mixture $(n\text{-Bu}_4\text{N})\text{TaF}_6 / (n\text{-Bu}_4\text{N})\text{PF}_6$ of nominal composition Ta/P 0.5/0.5.



^{19}F NMR (DMSO- d_6 , 500 MHz, δ in ppm): -70.90/-69.38 (d, $J = 760$ Hz) for PF_6^- / -5.01 (quint), 18.32 (d) and 41.12 (bs, TaF_6) for TaF_6^- and $\text{Ta}_2\text{F}_{10}\text{O}$. Integration gives $\approx 60\%$ of PF_6^- .

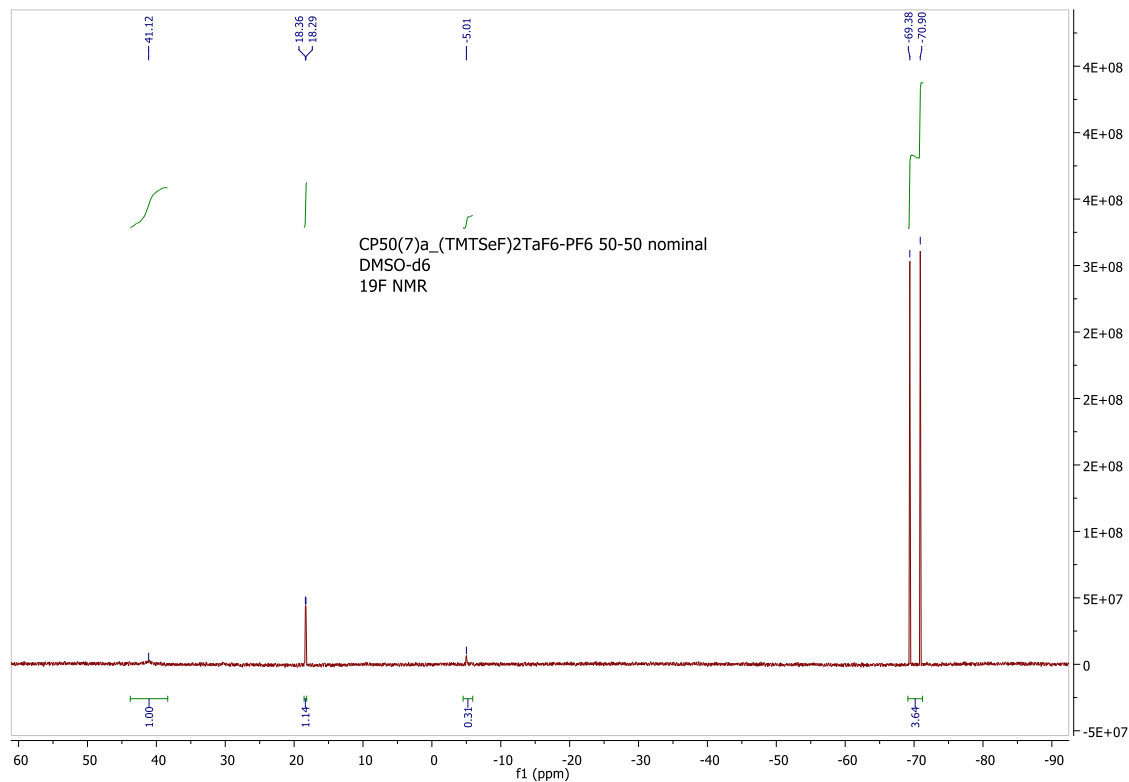


Figure S11. ^{19}F NMR (DMSO- d_6) spectrum of the crystalline material obtained by electrocrystallization of TMTSF with a mixture $(n\text{-Bu}_4\text{N})\text{TaF}_6 / (n\text{-Bu}_4\text{N})\text{PF}_6$ of nominal composition Ta/P 0.5/0.5.



^{19}F NMR (DMSO- d_6 , 500 MHz, δ in ppm): -70.90/ -69.39 (d, $J = 755$ Hz) for PF_6^- / -5.26 (quint), 18.28 (d), 33.73 (d, $\text{Ta}_2\text{F}_{10}\text{O}$, $8F_{eq}$) and 41.48 (bs, TaF_6) for TaF_6^- and $\text{Ta}_2\text{F}_{10}\text{O}$. Integration gives $\approx 16\%$ of PF_6^- .

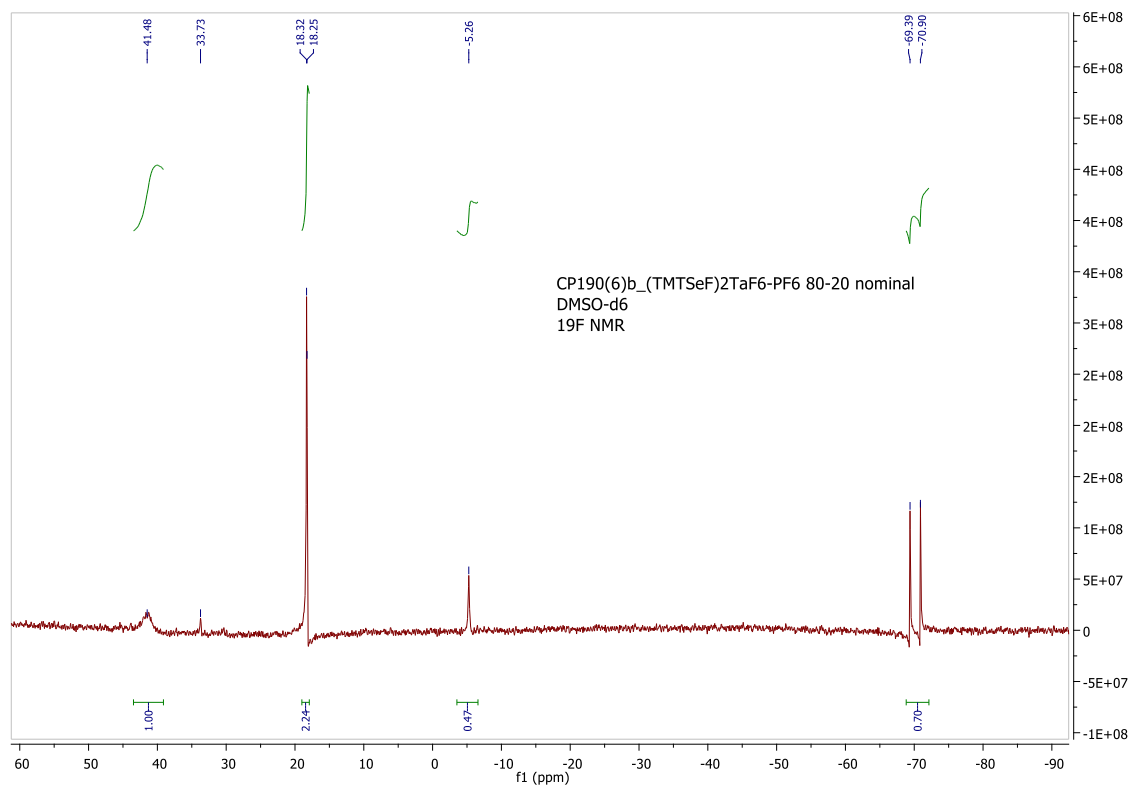


Figure S12. ^{19}F NMR (DMSO- d_6) spectrum of the crystalline material obtained by electrocrystallization of TMTSF with a mixture ($n\text{-Bu}_4\text{N}$) TaF_6 / ($n\text{-Bu}_4\text{N}$) PF_6 of nominal composition Ta/P 0.8/0.2.

Crystal structures of BEDT-TTF salts

Compound $\delta_o\text{-(BEDT-TTF)}_2(\text{TaF}_6)_{0.43}(\text{PF}_6)_{0.57}$

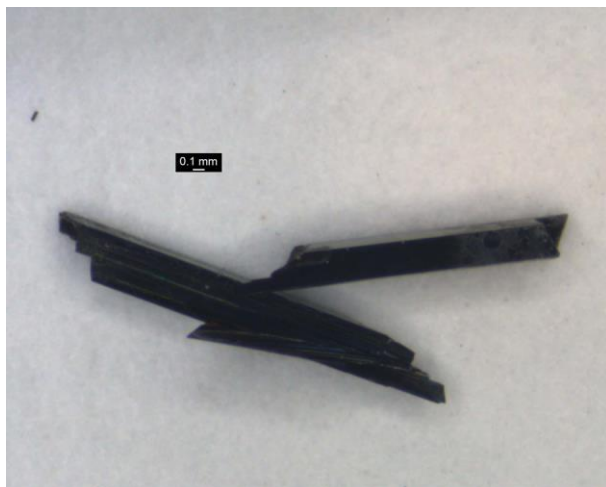


Figure S13. Picture of $\delta_o\text{-(BEDT-TTF)}_2(\text{TaF}_6)_{0.43}(\text{PF}_6)_{0.57}$ needles.

Compound $\delta_m\text{-(BEDT-TTF)}_2(\text{TaF}_6)_{0.94}(\text{PF}_6)_{0.06}$

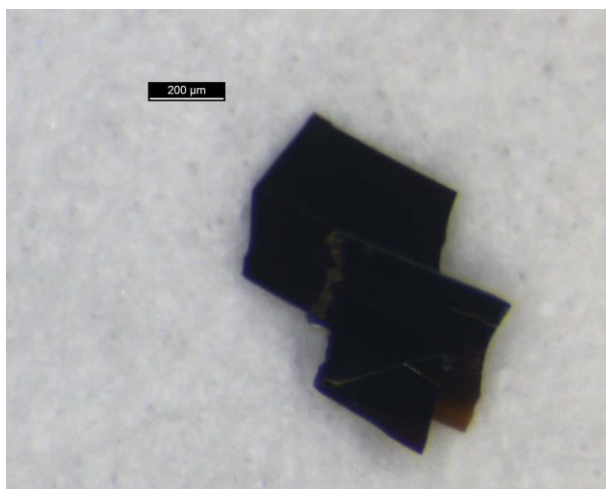


Figure S14. Picture of $\delta_m\text{-(BEDT-TTF)}_2(\text{TaF}_6)_{0.94}(\text{PF}_6)_{0.06}$ platelets.

¹⁹F NMR spectroscopy

The crystalline material was dissolved in DMSO-d₆.

¹⁹F NMR (DMSO-d₆, 500 MHz, δ in ppm): -70.90/-69.39 (d, $J = 755$ Hz) for PF₆ / 18.32 (d), 33.66 (d, Ta₂F₁₀O, 8F_{eq}) and 41.39 (m) for TaF₆. Integration gives $\approx 6\%$ of PF₆.

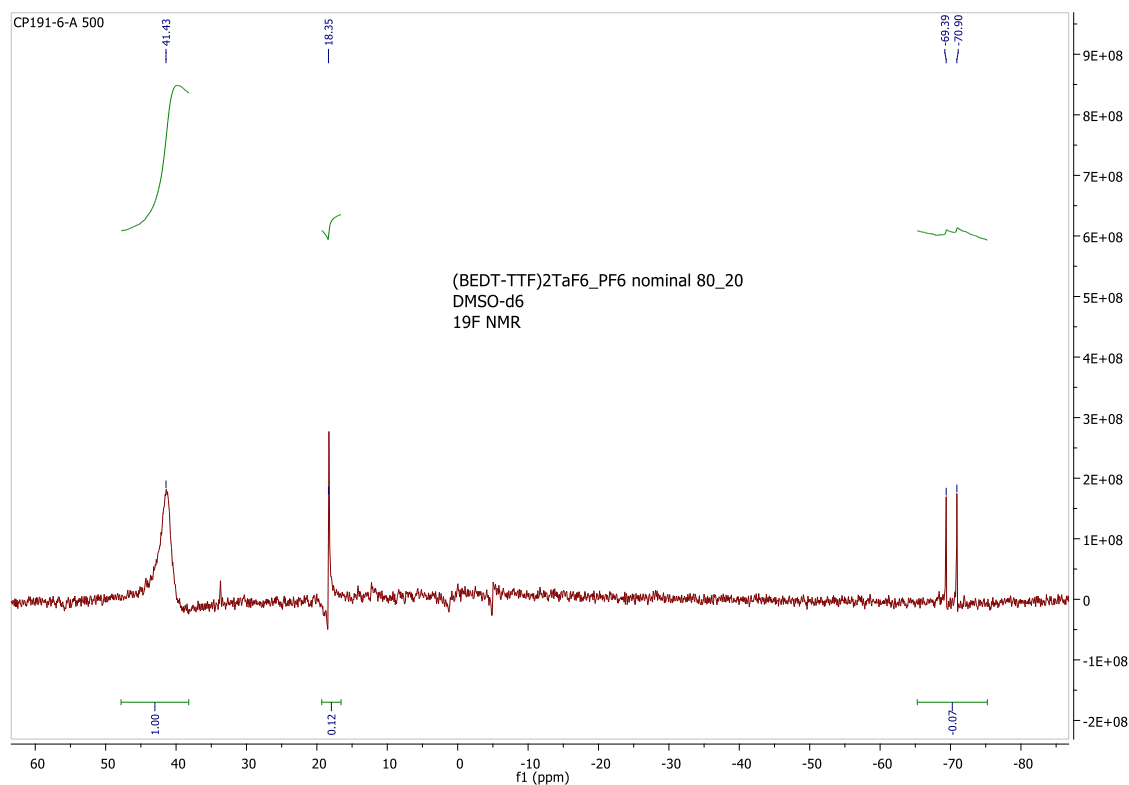


Figure S15. ¹⁹F NMR (DMSO-d₆) spectrum of the crystalline material obtained by electrocrystallization of BEDT-TTF with a mixture (*n*-Bu₄N)TaF₆ / (*n*-Bu₄N)PF₆ of nominal composition Ta/P 0.8/0.2.

Compound (BEDT-TTF)₂(TaF₆)₂·CH₂Cl₂

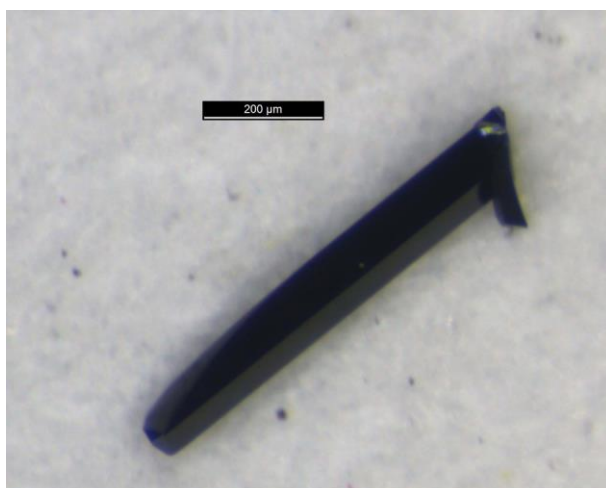


Figure S16. Picture of (BEDT-TTF)₂(TaF₆)₂·CH₂Cl₂ thick needles.

Table S3. Comparison of the proportion of phosphorus and tantalum between the inserted precursor salts used for the electrocrystallization syntheses (nominal ratio), the refined ratio in the corresponding crystallographic structures (refined ratio) and the ^{19}F NMR integration in the TMTSF and BEDT-TTF salts.

Donor	TMTSF	TMTSF	TMTSF	TMTSF	BEDT-TTF	BEDT-TTF
Nominal rate	20%-(<i>n</i> -Bu ₄ N)TaF ₆ 80%-(<i>n</i> -Bu ₄ N)PF ₆	50%-(<i>n</i> -Bu ₄ N)TaF ₆ 50%-(<i>n</i> -Bu ₄ N)PF ₆	50%-(<i>n</i> -Bu ₄ N)TaF ₆ 50%-(<i>n</i> -Bu ₄ N)PF ₆	80%-(<i>n</i> -Bu ₄ N)TaF ₆ 20%-(<i>n</i> -Bu ₄ N)PF ₆	20%-(<i>n</i> -Bu ₄ N)TaF ₆ 80%-(<i>n</i> -Bu ₄ N)PF ₆	80%-(<i>n</i> -Bu ₄ N)TaF ₆ 20%-(<i>n</i> -Bu ₄ N)PF ₆
Crystal refined rate	Ta: 0.12 P: 0.88	Ta: 0.44 P: 0.56	Ta: 0.56 P: 0.44	Ta: 0.84 P: 0.16	Ta: 0.43 P: 0.57	Ta: 0.94 P: 0.06
Crystal Formula	(TMTSF) ₂ (TaF ₆) _{0.12} (PF ₆) _{0.88}	(TMTSF) ₂ (TaF ₆) _{0.44} (PF ₆) _{0.56}	(TMTSF) ₂ (TaF ₆) _{0.56} (PF ₆) _{0.44}	(TMTSF) ₂ (TaF ₆) _{0.84} (PF ₆) _{0.16}	δ _o -(BEDT-TTF) ₂ (TaF ₆) _{0.43} (PF ₆) _{0.57}	δ _m -(BEDT-TTF) ₂ (TaF ₆) _{0.94} (PF ₆) _{0.06}
^{19}F NMR analysis	Ta: 0.32 P: 0.68	Ta: 0.48 P: 0.52	Ta: 0.40 P: 0.60	Ta: 0.84 P: 0.16	-	Ta: 0.94 P: 0.06

Table S4. Central C=C and C–S internal bond distances (Å) of (TMTSF)₂(TaF₆)_{0.12}(PF₆)_{0.88}, (TMTSF)₂(TaF₆)_{0.44}(PF₆)_{0.56}, (TMTSF)₂(TaF₆)_{0.56}(PF₆)_{0.44}, (TMTSF)₂(TaF₆)_{0.84}(PF₆)_{0.16} and (TMTSF)₂TaF₆ (293 K and 100 K).

	(TMTSF) ₂ (TaF ₆) _{0.12} (PF ₆) _{0.88}	(TMTSF) ₂ (TaF ₆) _{0.44} (PF ₆) _{0.56}	(TMTSF) ₂ (TaF ₆) _{0.56} (PF ₆) _{0.44}	(TMTSF) ₂ (TaF ₆) _{0.84} (PF ₆) _{0.16}	(TMTSF) ₂ TaF ₆ (293 K)	(TMTSF) ₂ TaF ₆ (100 K)
C1—C2	1.355(8)	1.355(4)	1.359(9)	1.347(9)	1.356(12)	1.352(7)
Se1—C1	1.861(6)	1.876(3)	1.874(6)	1.874(5)	1.875(8)	1.882(5)
Se2—C1	1.879(7)	1.877(3)	1.874(5)	1.889(6)	1.886(8)	1.884(5)
Se3—C2	1.884(6)	1.881(3)	1.876(5)	1.884(6)	1.878(8)	1.878(5)
Se4—C2	1.885(6)	1.877(3)	1.881(5)	1.876(5)	1.875(8)	1.880(5)

Table S5. Central C=C and C–S internal bond distances (Å) of δ_o-(BEDT-TTF)₂(TaF₆)_{0.43}(PF₆)_{0.57}, δ_m-(BEDT-TTF)₂(TaF₆)_{0.94}(PF₆)_{0.06} and (BEDT-TTF)₂(TaF₆)₂·CH₂Cl₂.

	δ _o -(BEDT-TTF) ₂ (TaF ₆) _{0.43} (PF ₆) _{0.57}	δ _m -(BEDT-TTF) ₂ (TaF ₆) _{0.94} (PF ₆) _{0.06}	(BEDT-TTF) ₂ (TaF ₆) ₂ , CH ₂ Cl ₂
C1—C2	1.362(4)	1.356(7)	1.386(5)
S1—C1	1.732(3)	1.735(5)	1.721(4)
S2—C1	1.737(3)	1.745(6)	1.716(4)
S3—C2	1.738(3)	1.739(5)	1.721(4)
S4—C2	1.738(3)	1.737(5)	1.720(4)

Single crystal resistivity measurements

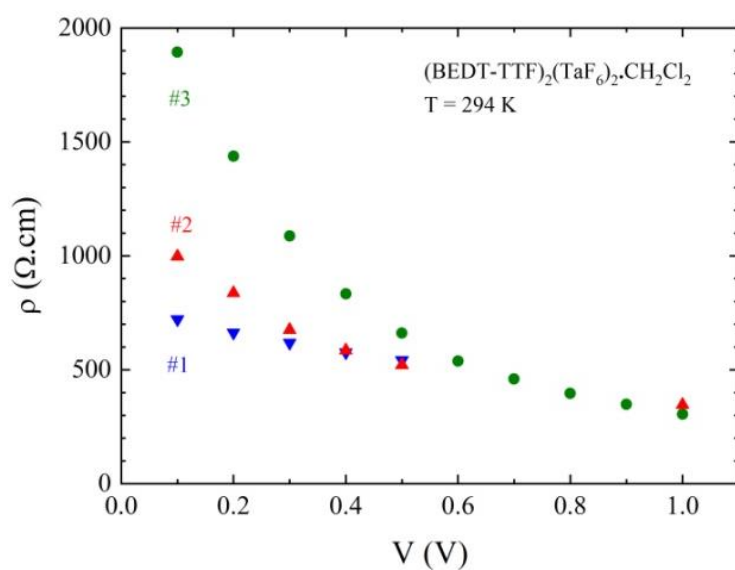


Figure S17. Electrical resistivity of three single crystals of $(\text{BEDT-TTF})_2(\text{TaF}_6)_2 \cdot \text{CH}_2\text{Cl}_2$ at room temperature plotted as a function of the applied voltage used for a two contact measurement.

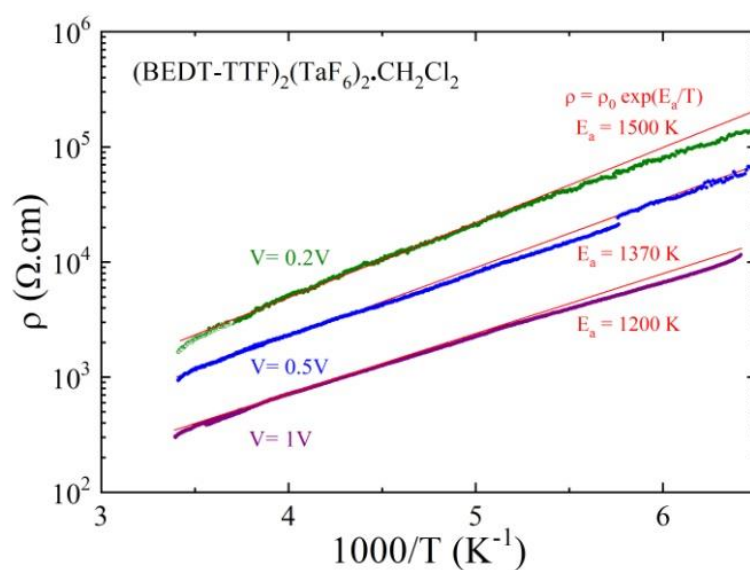


Figure S18. Temperature dependence of the electrical resistivity of a single crystal (#3) of $(\text{BEDT-TTF})_2(\text{TaF}_6)_2 \cdot \text{CH}_2\text{Cl}_2$ plotted as $\log \rho$ versus $1/T$ measured with different applied voltages: 0.2, 0.5 and 1 V. The red line is the linear fit to the data giving the activation energy E_a .