

Supplementary Materials: Component Engineering of Multiphase Nickel Sulfide-Based Bifunctional Electrocatalysts for Efficient Overall Water Splitting

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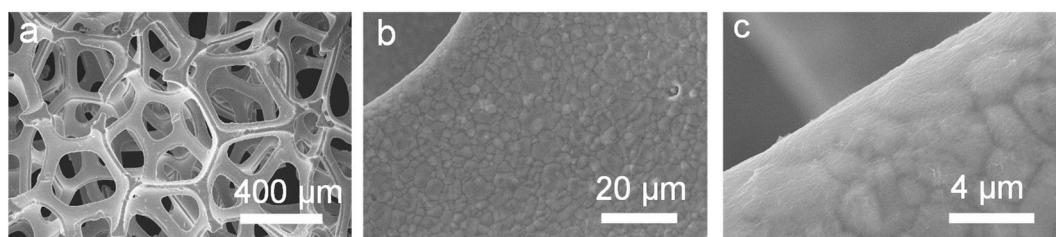


Figure S1. (a-c) The different magnification SEM images of NF surface.

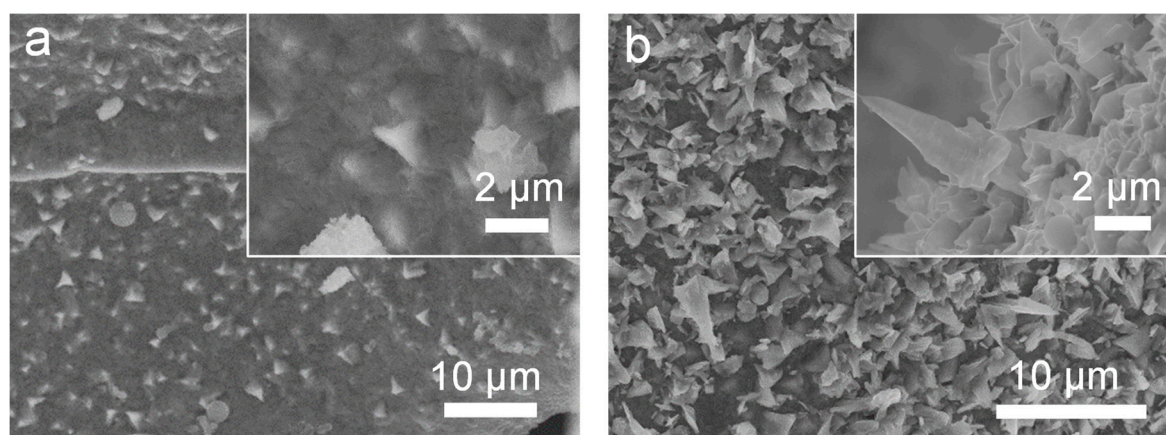


Figure S2. SEM images of Ni_xS_y-2 (a), Ni_xS_y-3 (b). The insets exhibit the corresponding high magnification SEM image, respectively.

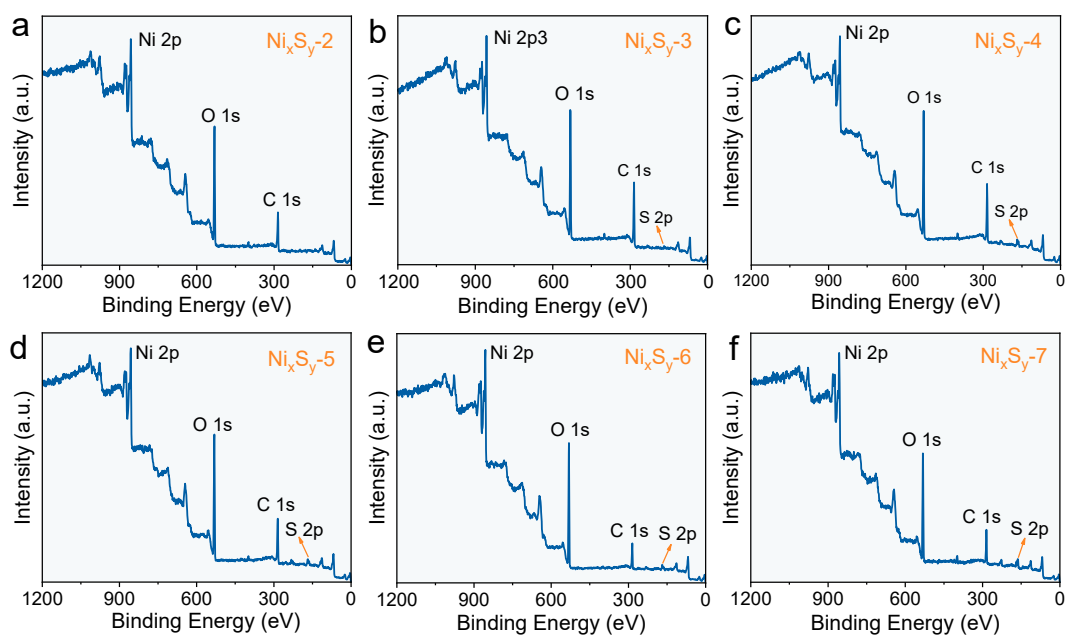


Figure S3. XPS spectra of the survey spectrum for Ni_xS_y-2 (a), Ni_xS_y-3 (b), Ni_xS_y-4 (c), Ni_xS_y-5 (d), Ni_xS_y-6 (e), and Ni_xS_y-7 (f), respectively.

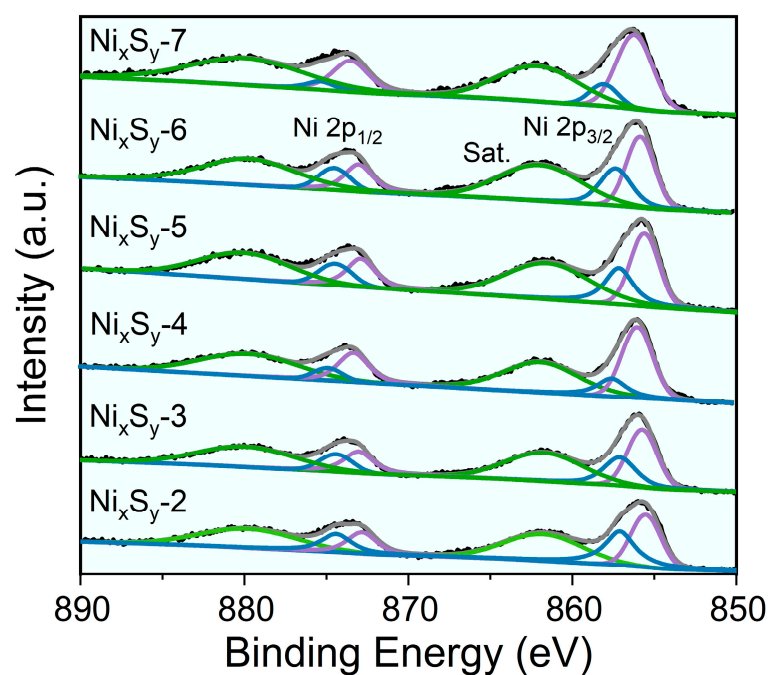


Figure S4. XPS spectra of the Ni 2p spectrum for all the as-prepared samples.

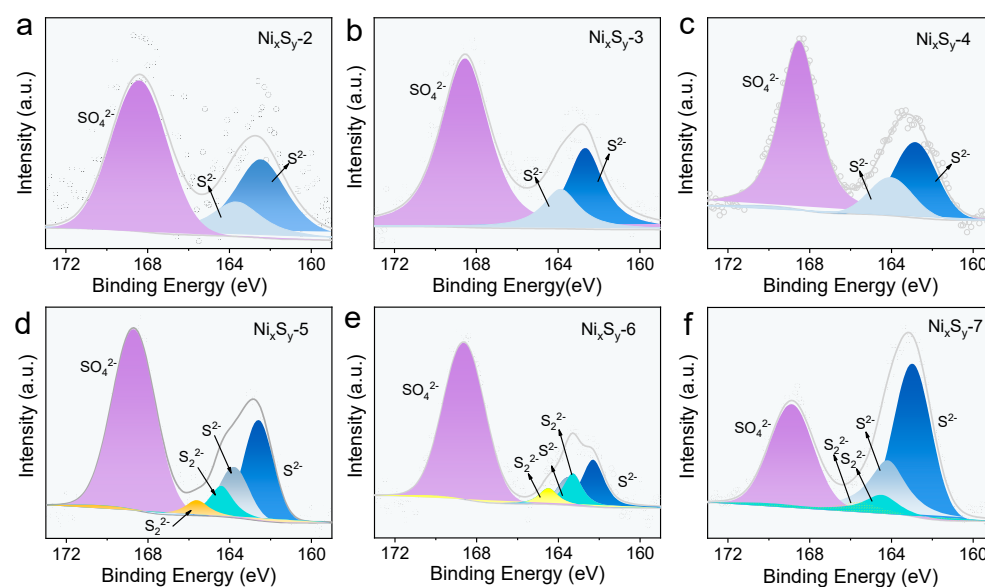


Figure S5. XPS spectra of the S 2p spectrum for Ni_xS_y -2 (a), Ni_xS_y -3 (b), Ni_xS_y -4 (c), Ni_xS_y -5 (d), Ni_xS_y -6 (e), and Ni_xS_y -7 (f), respectively.

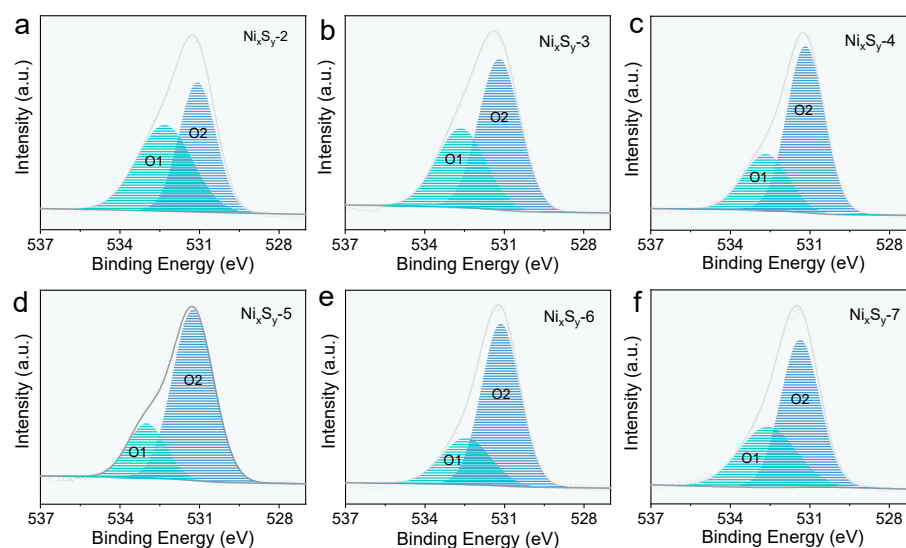


Figure S6. XPS spectra of the O 1s spectrum for Ni_xS_y -2 (a), Ni_xS_y -3 (b), Ni_xS_y -4 (c), Ni_xS_y -5 (d), Ni_xS_y -6 (e), and Ni_xS_y -7 (f), respectively.

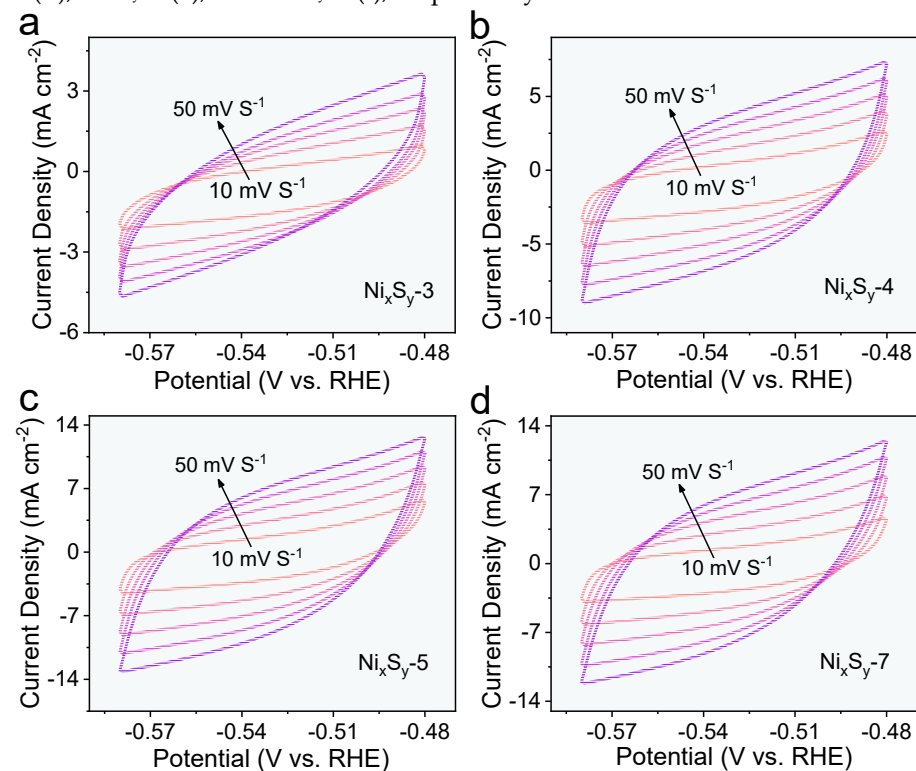


Figure S7. CV curves of the Ni_xS_y -3 (a), Ni_xS_y -4 (b), Ni_xS_y -5 (c), Ni_xS_y -6 (d), respectively, for HER process.

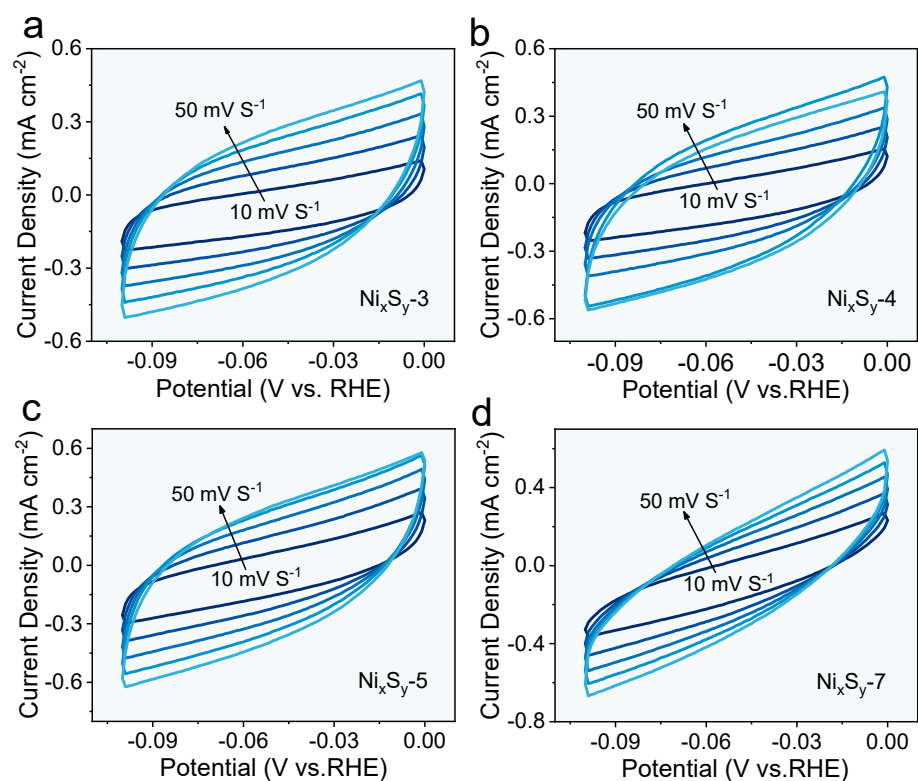


Figure S8. CV curves of the $\text{Ni}_x\text{S}_y\text{-3}$ (a), $\text{Ni}_x\text{S}_y\text{-4}$ (b), $\text{Ni}_x\text{S}_y\text{-5}$ (c), $\text{Ni}_x\text{S}_y\text{-6}$ (d), respectively, for OER process.

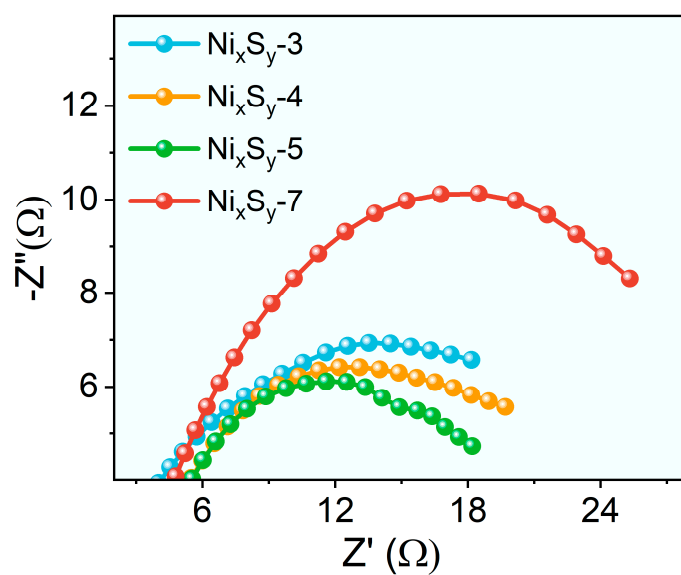


Figure S9. The corresponding EIS plot for $\text{Ni}_x\text{S}_y\text{-3}$ electrode, $\text{Ni}_x\text{S}_y\text{-4}$ electrode, $\text{Ni}_x\text{S}_y\text{-5}$ electrode, and $\text{Ni}_x\text{S}_y\text{-7}$ electrode.

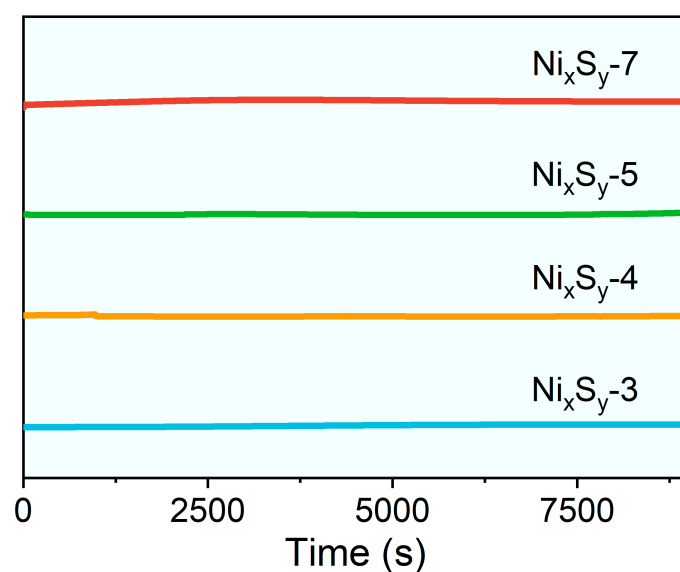


Figure S10. The stability test toward HER at constant current density of 10 mA cm^{-2} for Ni_xS_y -3 electrode, Ni_xS_y -4 electrode, Ni_xS_y -5 electrode, and Ni_xS_y -7 electrode.

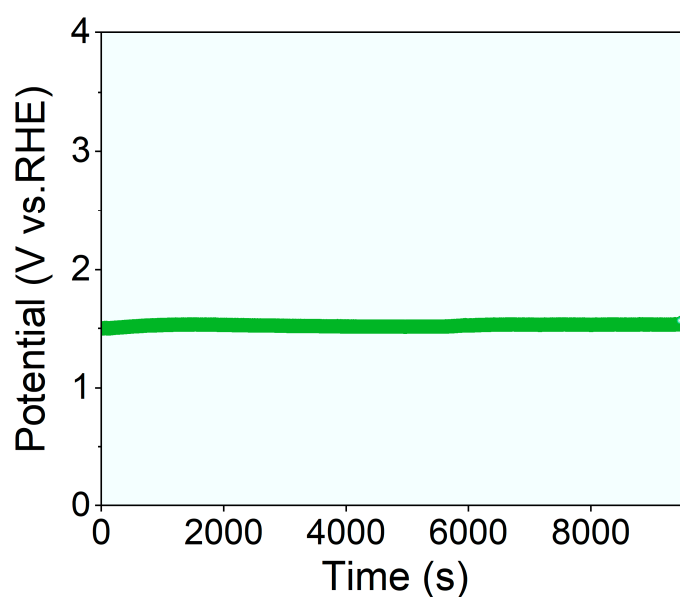


Figure S11. Stability test of overall water splitting at constant current density of 10 mA cm^{-2} for Ni_xS_y -5 electrolyzer.

Table S1. Summary of the HER activities of recently reported non-noble metal-based electrocatalysts in 1 M KOH.

Catalysts	Tafel (mV dec^{-1})	η_{10} (mV)	Reference
NiCoP	64.4	178	<i>Adv. Energy Mater.</i> 2018 , <i>8</i> , 1802615
$\text{Co}_4\text{Mo}_2\text{@NC}$	73.5	218	<i>J. Mater. Chem. A</i> , 2017 , <i>5</i> , 16929
CoP NFs	56.2	136	<i>ACS Catal.</i> 2020 , <i>10</i> , 412
$\text{Co}_3\text{O}_4/\text{MoS}_2$	98	205	<i>Appl. Catal., B</i> 2019 , <i>248</i> , 202
NiS_2 NA	104	149	<i>Electrochim. Acta</i> 2015 , <i>153</i> , 508

NiCo ₂ S ₄ NW	58.9	210	<i>Adv. Funct. Mater.</i> 2016 , <i>26</i> , 4661
Mo ₂ NiB ₂	71	160	<i>Small</i> 2022 , <i>18</i> , 2104303
NiCoP	83	130	<i>Adv. Energy Mater.</i> 2018 , <i>8</i> , 1800555
K ₂ Fe ₄ O ₇ /NF	47.1	109	<i>J. Mater. Chem. A</i> 2021 , <i>9</i> , 7586
MoB	59	225	<i>Angew. Chem. Int. Ed.</i> 2012 , <i>51</i> , 12703
Ni _x S _y -5	140	148	This work

Table S2. Summary of the OER activities of recently reported non-noble metal-based electrocatalysts.

Catalysts	Tafel (mV dec ⁻¹)	η_{10} (mV)	Reference
Mo ₂ C/NCF	55	144	<i>ACS Nano</i> 2016 , <i>10</i> , 11337
CeO _x /CoS	50	269	<i>Angew. Chem. Int., Ed.</i> 2018 , <i>57</i> , 8654
CoNiB	120	302.4	<i>J. Mater. Chem. A</i> 2021 , <i>9</i> , 6469
Mo ₂ NiB ₂	57	280	<i>Small</i> 2022 , <i>18</i> , 2104303
Mo-CoOOH	68	305	<i>Nano Energy</i> , 2018 , <i>48</i> , 73
RuO ₂ /NF	128	257	<i>J. Energy Chem.</i> 2022 , <i>70</i> , 472
K ₂ Fe ₄ O ₇ /NF	59	308	<i>J. Mater. Chem. A</i> , 2021 , <i>9</i> , 7586
h-NiS _x	96	180	<i>Adv. Energy Mater.</i> 2016 , <i>6</i> , 1502333
Fe-Ni ₃ S ₂ /FeNi	54	282	<i>Small</i> , 2017 , <i>13</i> , 1604161
CoNi ₂ Se ₄	72	160	<i>Chem. Commun.</i> 2017 , <i>53</i> , 5412
CoOOH/Co ₉ S ₈	86.4	240	<i>Angew. Chem. Int., Ed.</i> 2022 , <i>61</i> , e2021171
Co/WN-600	65	232	<i>J. Mater. Chem. A</i> , 2020 , <i>8</i> , 22938
Ag/Co(OH) _x	76	250	<i>Angew. Chem., Int. Ed.</i> 2020 , <i>59</i> , 7245
Ni-MoN	98	276	<i>Adv. Energy Mater.</i> 2018 , <i>8</i> , 1802327
Ni _{0.6} Co _{1.4} P	80	300	<i>Adv. Funct. Mater.</i> 2018 , <i>28</i> , 1706008
Ni _x S _y -5	60	111	This work

Table S3. Comparison of electrocatalytic performance of Ni_xS_y-5 with recently reported bifunctional electrocatalysts for overall-water-splitting in alkaline media.

Catalysts	Electrolyte	η_{10} (V)	Reference
Pt/C-RuO ₂	1 M KOH	1.67 V	<i>Appl. Catal., B</i> 2020 , <i>278</i> , 119281
MoS ₂ /Ni ₃ S ₂	1 M KOH	1.56	<i>Angew. Chem., Int. Ed.</i> 2016 , <i>55</i> , 6702

NiCo ₂ S ₄ /NF	1 M KOH	1.61	<i>Adv. Mater. Interfaces</i> , 2018 , <i>5</i> , 1701396
Co ₃ O ₄ /MoS ₂	1 M KOH	205	<i>Appl. Catal. B-Environ.</i> 2019 , <i>248</i> , 202
Mo-CoOOH	1 M KOH	1.56	<i>Nano Energy</i> , 2018 , <i>48</i> , 73
CoP NFs	1 M KOH	1.65	<i>ACS Catal.</i> 2020 , <i>10</i> , 412
Co ₃ O ₄ NCs	1 M KOH	1.91	<i>Chem. Commun.</i> 2015 , <i>51</i> , 8066
K ₂ Fe ₄ O ₇ /NF	1 M KOH	1.59	<i>J. Mater. Chem. A</i> 2021 , <i>9</i> , 7586
Mo ₂ NiB ₂	1 M KOH	1.57	<i>Small</i> 2022 , <i>18</i> , 2104303
NiCo ₂ S ₄	1 M KOH	1.58	<i>Adv. Funct. Mater.</i> 2019 , <i>29</i> , 1807031
Ni _x S _y -5	1 M KOH	1.48	This work