

# Ni doped Co-MOF-74 synergized with 2D $\text{Ti}_3\text{C}_2\text{T}_x$ MXene as an efficient electrocatalyst for overall water-splitting

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## Supporting information

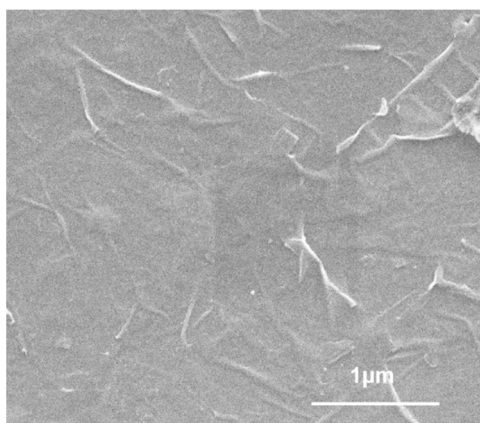


Figure S1. SEM image of as-prepared few-layer  $\text{Ti}_3\text{C}_2\text{T}_x$  MXene.

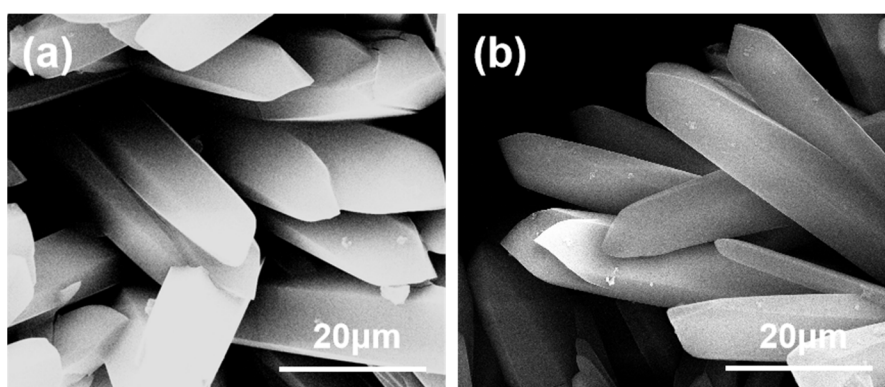


Figure S2. SEM images of a)  $\text{CoNi}_{0.1}$ -MOF-74/MXene/NF, b)  $\text{CoNi}_{0.03}$ -MOF-74/MXene/NF.

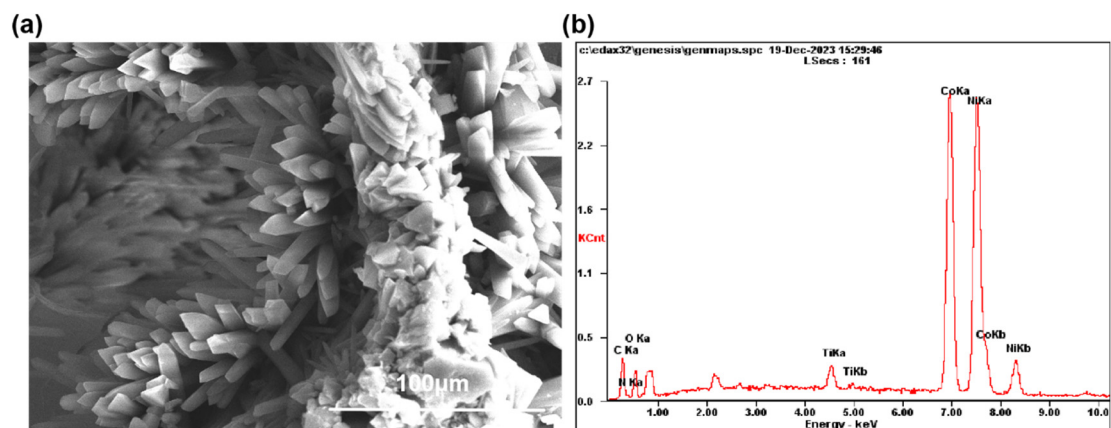


Figure S3. The SEM (a) and corresponding EDS spectra (b) of CoNi<sub>0.04</sub>-MOF-74/MXene.

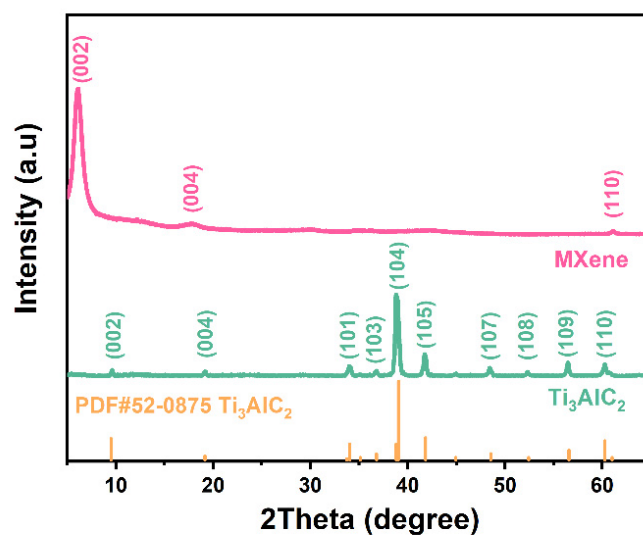


Figure S4. XRD patterns of MXene and Ti<sub>3</sub>AlC<sub>2</sub>

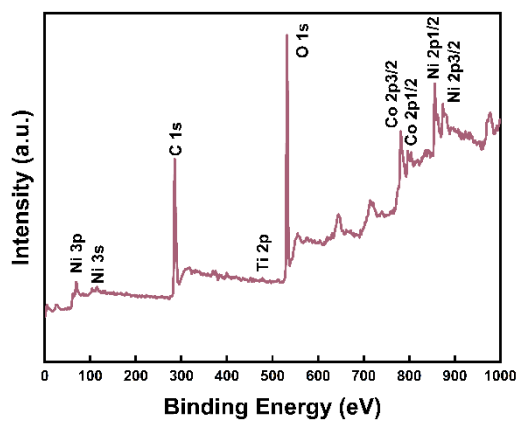
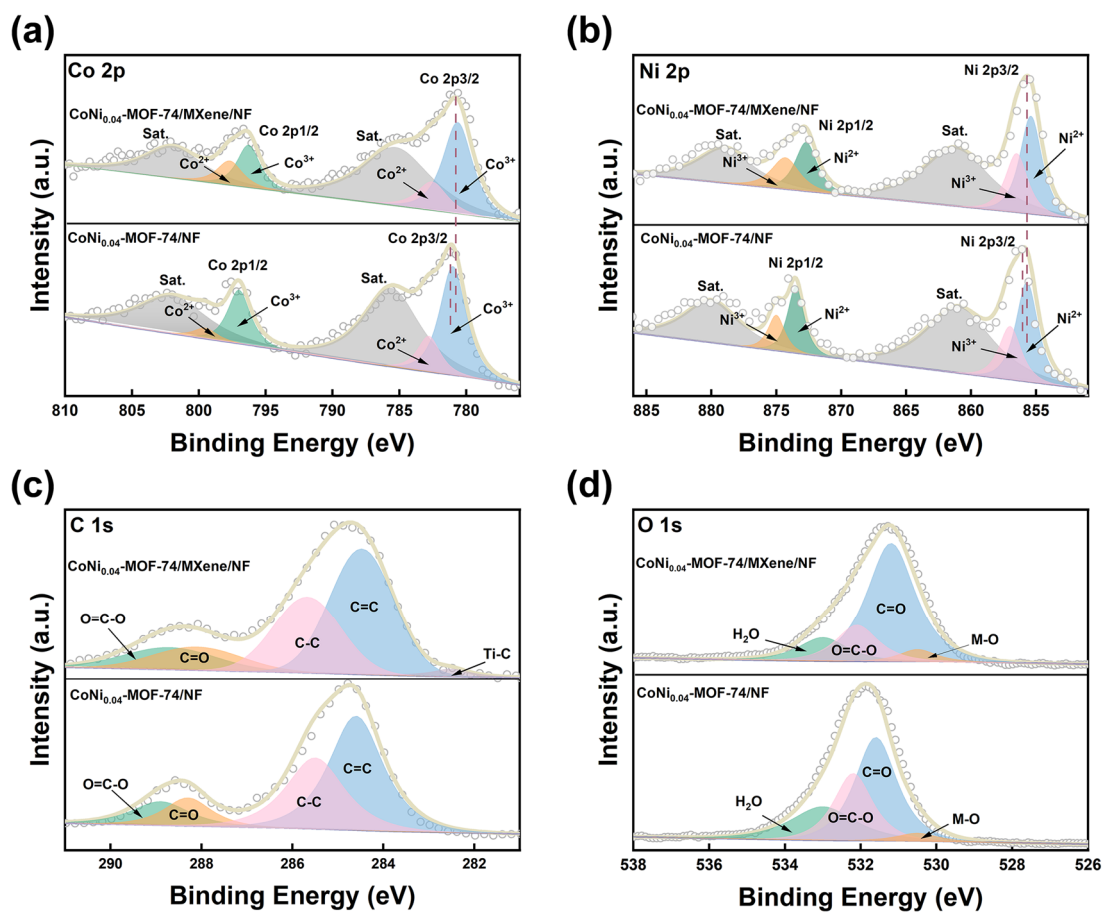


Figure S5. XPS survey spectra of CoNi<sub>0.04</sub>-MOF-74/MXene/NF.



**Figure S6.** XPS (a) Co 2p, (b) Ni 2p, (c) C 1s, and (d) O 1s spectra of the CoNi<sub>0.04</sub>-MOF-74/MXene/NF and CoNi<sub>0.04</sub>-MOF-74/NF.

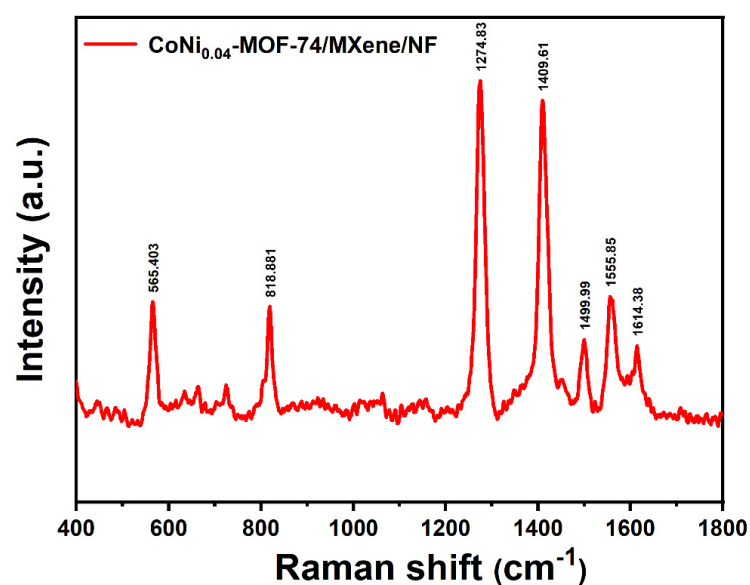


Figure S7. Raman spectra of CoNi<sub>0.04</sub>-MOF-74/MXene/NF

The TOF value was calculated based on the following equation:

$$\text{TOF} = JA / (4Fm)$$

in which TOF is based on the number of redox-active sites,  $J$  is the current density at a specific overpotential,  $A$  is the area of the electrode,  $F$  is the Faraday constant, and  $m$  is the number of active sites.

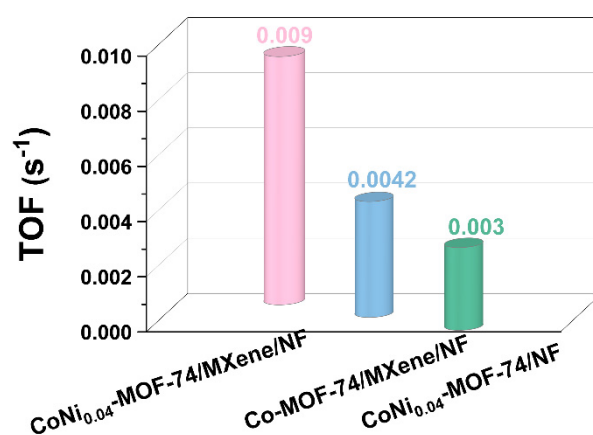
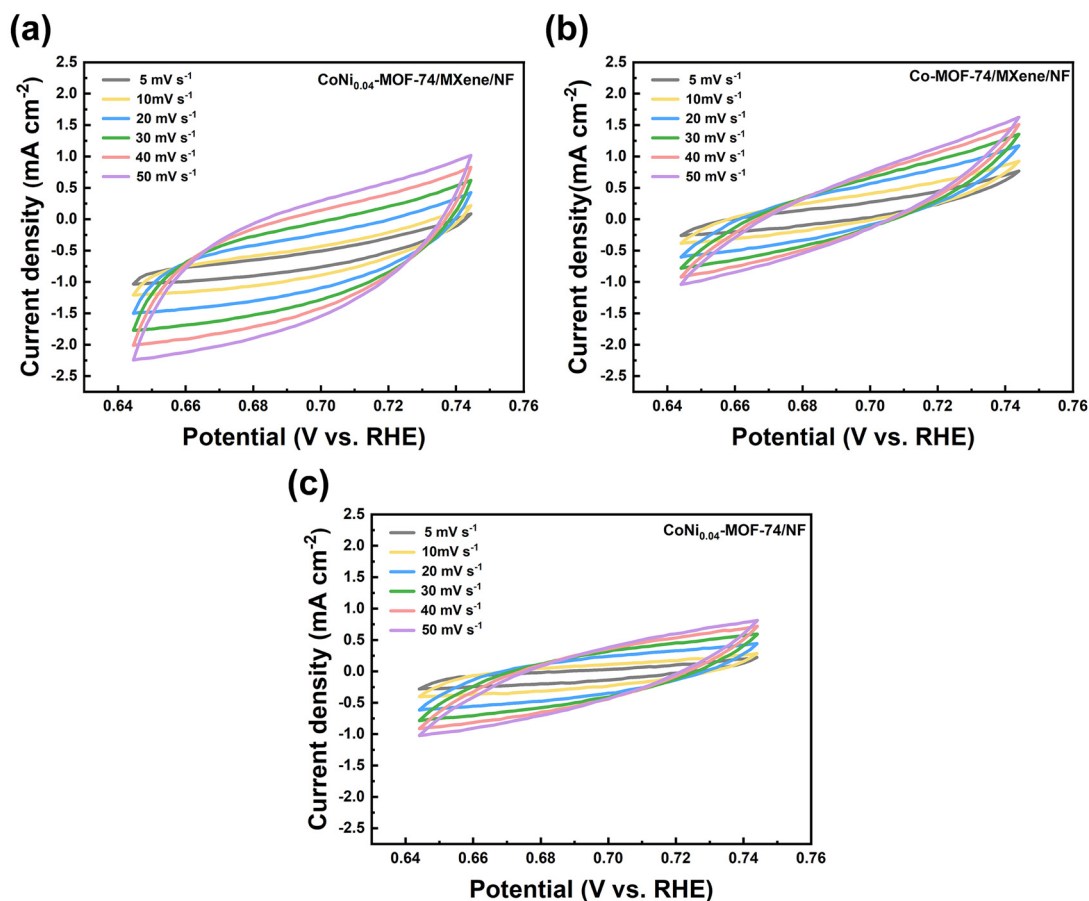
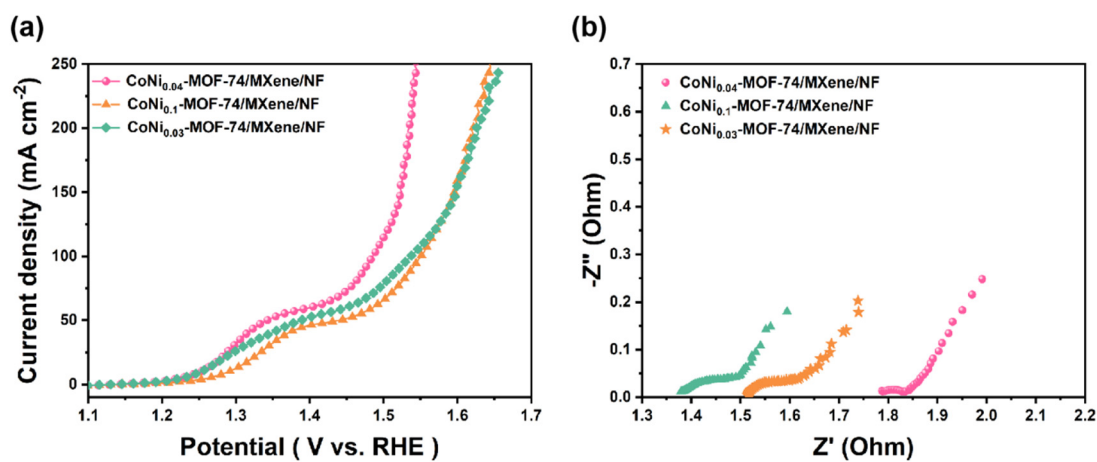


Figure S8. TOF of the CoNi<sub>0.04</sub>-MOF-74/MXene/NF, Co-MOF-74/MXene/NF and CoNi<sub>0.04</sub>-MOF-74/NF for OER.



**Figure S9.** CV curves of the CoNi<sub>0.04</sub>-MOF-74/MXene/NF (a), Co-MOF-74/MXene/NF (b), CoNi<sub>0.04</sub>-MOF-74/NF (c) under different scan rates in the region of 0.644 - 0.744 V vs. RHE for OER process.



**Figure S10.** (a) OER polarization curves and (b) Nyquist plots of CoNi<sub>0.1</sub>-MOF-74/MXene/NF, CoNi<sub>0.04</sub>-MOF-74/MXene/NF, CoNi<sub>0.03</sub>-MOF-74/MXene/NF.

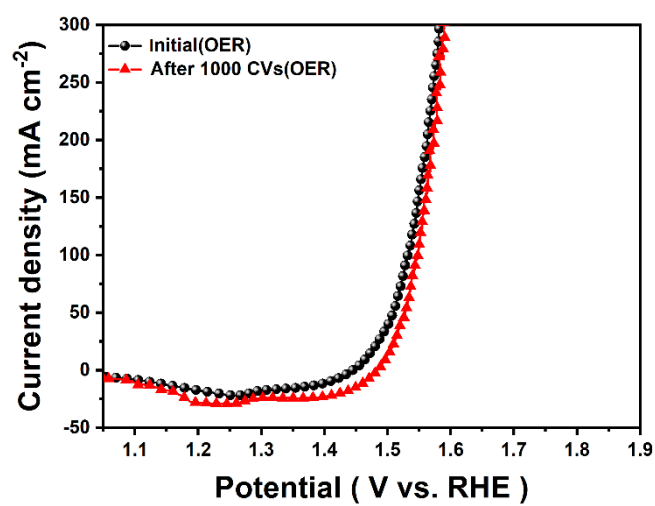


Figure S11. OER polarization curves for CoNi<sub>0.04</sub>-MOF-74/MXene/NF before and after 1,000 cycles.

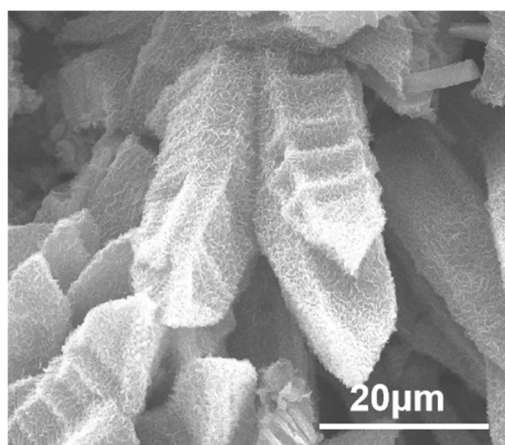
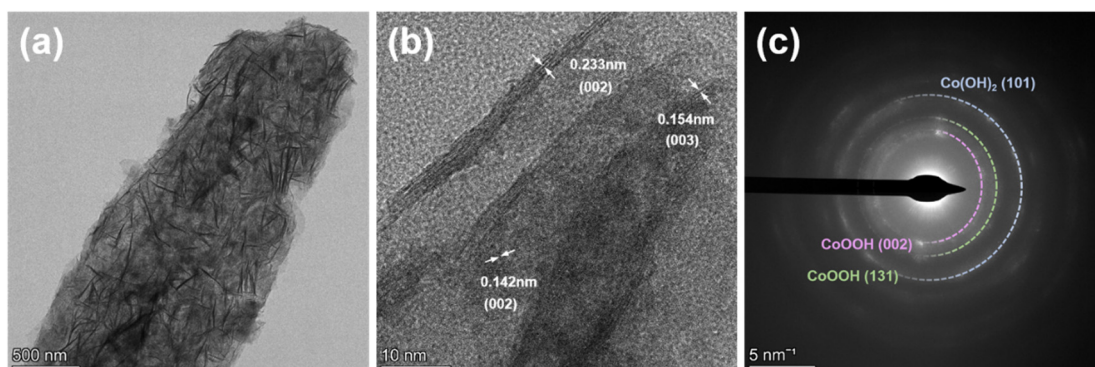
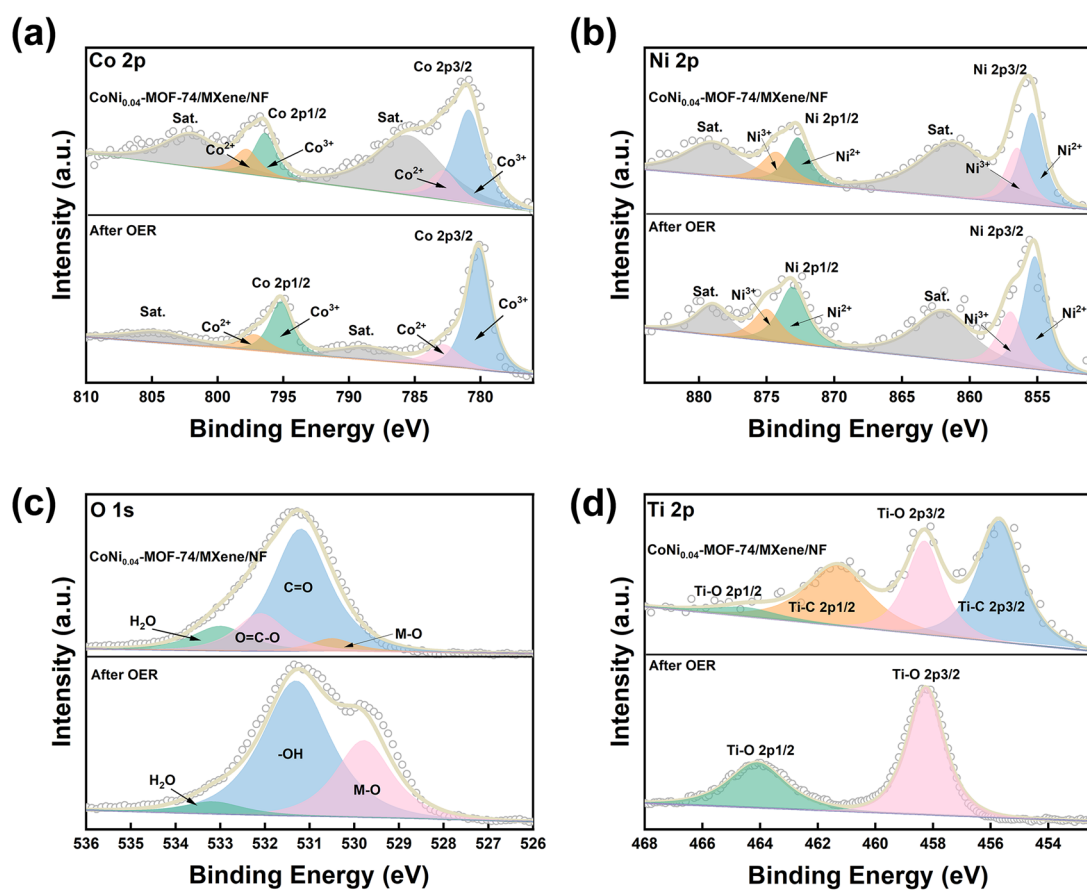


Figure S12. SEM images of CoNi<sub>0.04</sub>-MOF-74/MXene/NF after OER test.

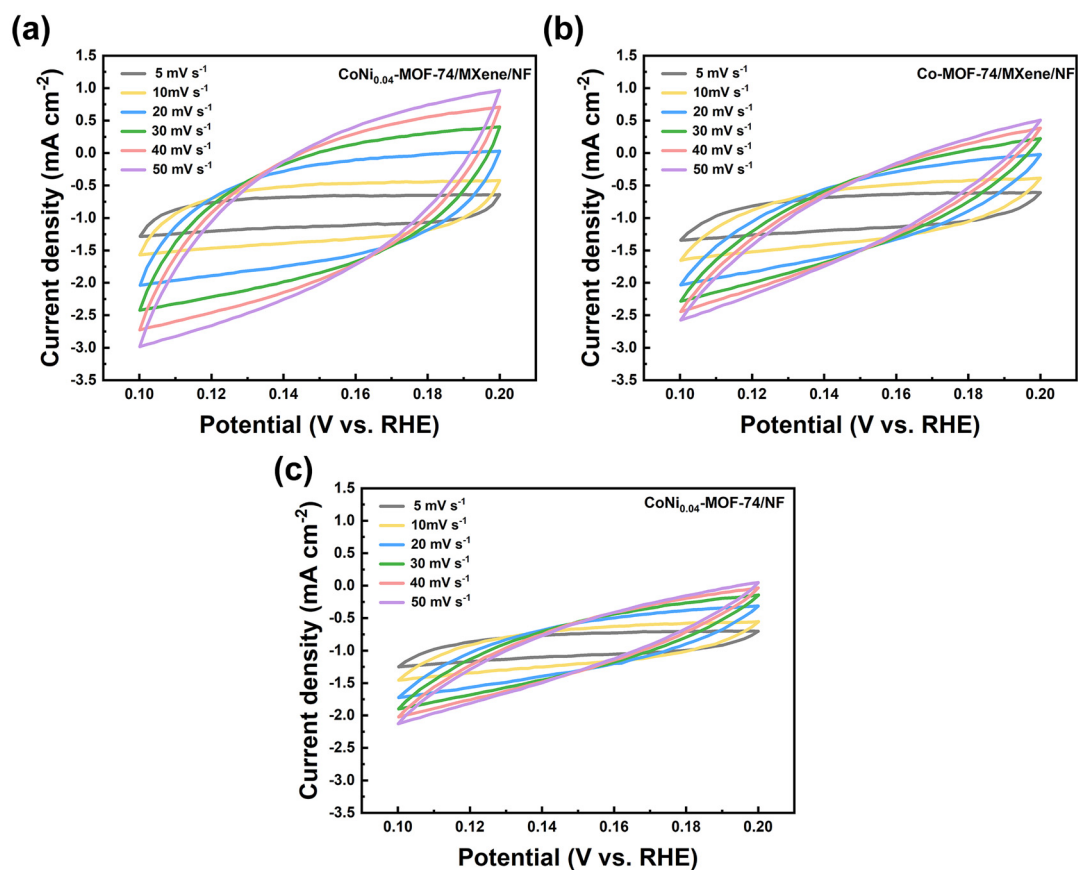


**Figure S13.** TEM images of (a)  $\text{CoNi}_{0.04}\text{-MOF-74/MXene/NF}$  after the OER test; HRTEM image (b) and SAED patterns (c) of  $\text{CoNi}_{0.04}\text{-MOF-74/MXene/NF}$  after the OER test.

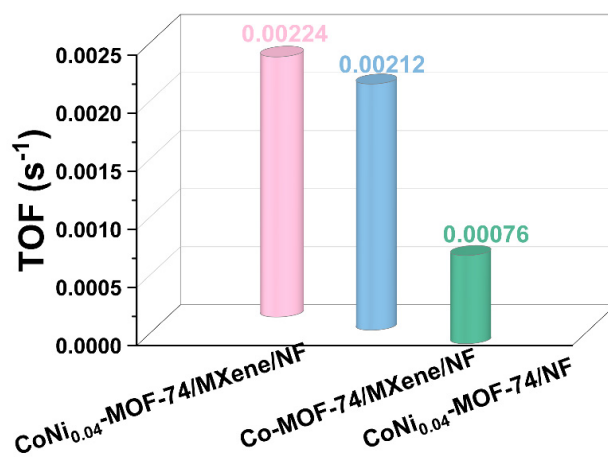


**Figure S14.** XPS (a) Co 2p, (b) Ni 2p, (c) O 1s, and (d) Ti 2p spectra of the CoNi<sub>0.04</sub>-MOF-74/MXene/NF and XPS spectra of CoNi<sub>0.04</sub>-MOF-74 /MXene/NF after the OER test.



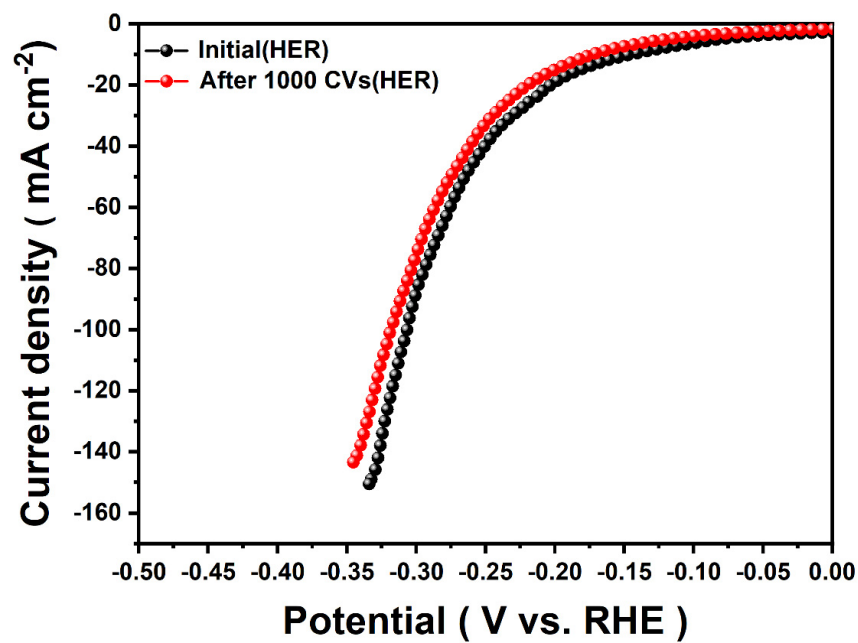


**Figure S15.** CV curves of the CoNi<sub>0.04</sub>-MOF-74/MXene/NF (a), Co-MOF-74/MXene/NF (b), CoNi<sub>0.04</sub>-MOF-

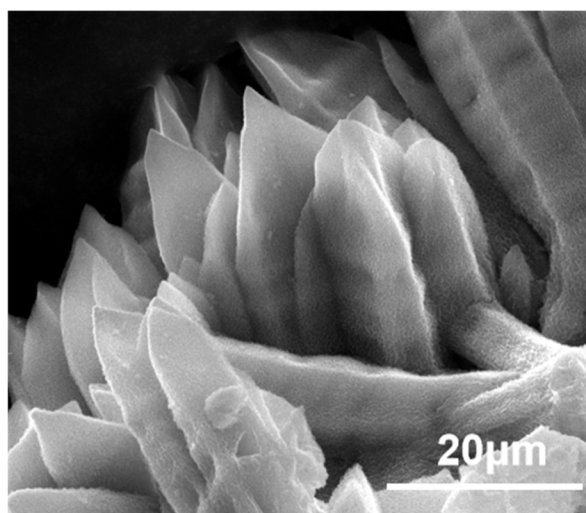


74/NF (c) under different scan rates in the region of 0.1 - 0.2 V vs. RHE for HER process.

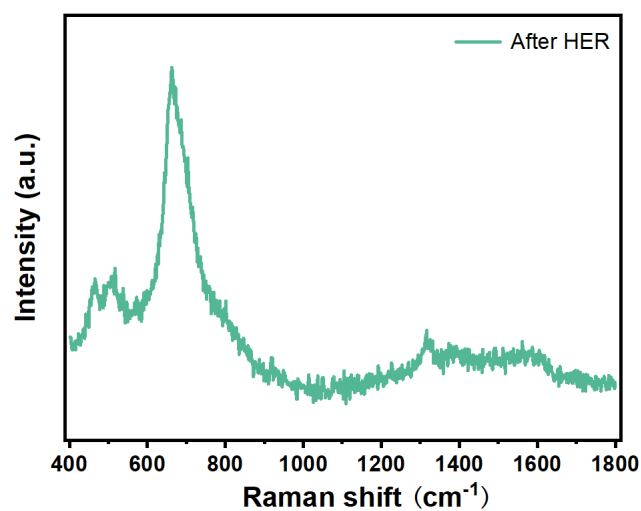
**Figure S16.** TOF of the CoNi<sub>0.04</sub>-MOF-74/MXene/NF, Co-MOF-74/MXene/NF and CoNi<sub>0.04</sub>-MOF-74/NF for HER.



**Figure S17.** HER polarization curves for CoNi<sub>0.04</sub>-MOF-74/MXene/NF before and after 1,000 cycles.



**Figure S18.** SEM images of CoNi<sub>0.04</sub>-MOF-74/MXene/NF after HER test.



**Figure S19.** Raman spectra of CoNi<sub>0.04</sub>-MOF-74/MXene/NF after the HER test.

**Table S1.** Co2p peak area ratio of CoNi<sub>0.04</sub>-MOF-74/MXene/NF and Co-MOF-74/MXene/NF calculated from XPS spectra.

	CoNi <sub>0.04</sub> -MOF-74/MXene/NF	Co-MOF-74/MXene/NF
Co <sup>2+</sup>	12.7%	19.9%
Co <sup>3+</sup>	31.7%	28.7%

**Table S2.** Comparison of Electrochemical performances of OER of CoNi<sub>0.04</sub>-MOF-74/MXene/NF with recently reported OER electrocatalysts in 1 M KOH.

Materials	Overpotential (mV)	Tafel slope (mV dec <sup>-1</sup> )	References
CoNi <sub>0.04</sub> -MOF-74/MXene/NF	256(η100)	46.7	Our Work
Co <sub>2</sub> Ni-MOF@MX-1	265(η10)	51.7	[1]
Ti <sub>2</sub> NT <sub>x</sub> @MOF-CoP	241(η50)	65.8	[2]
MX@MOF-Co <sub>2</sub> P	246(η10)	39.29	[3]
Ti <sub>3</sub> C <sub>2</sub> @mNiCoP	237(η10)	104	[4]
Ni <sub>0.7</sub> Fe <sub>0.3</sub> PS <sub>3</sub> @MXene	282(η10)	36.5	[5]
Mo-NiCoP@MXene/NF	280(η10)	56.91	[6]
Ru@NiCo-MOF HPNs	284(η10)	78.8	[7]
CoNi MOF-CNTs	306 (η10)	42	[8]

**Table S3.** OER catalyst EIS fitting parameters.

Materials	R <sub>s</sub> (Ω)	R <sub>ct</sub> (Ω)	CPE(Ω)
CoNi <sub>0.04</sub> -MOF-74/MXene	1.77	0.109	0.53
Co-MOF-74/MXene	1.48	0.37	0.56
CoNi <sub>0.04</sub> -MOF-74/NF	1.669	1.38	0.51
MXene/NF	1.738	58.61	0.72

**Table S4.** Area ratios of Co2p and Ni2p peaks calculated from XPS spectra for CoNi<sub>0.04</sub>-MOF-74/MXene/NF and after OER testing.

	CoNi <sub>0.04</sub> -MOF-74/MXene/NF	After OER
Co <sup>2+</sup>	12.7%	16.9%
Co <sup>3+</sup>	31.7%	61.2%
Ni <sup>2+</sup>	28.5%	40.7%
Ni <sup>3+</sup>	9.7%	22.0%

**Table S5.** HER catalyst EIS fitting parameters.

Materials	R <sub>s</sub> (Ω)	R <sub>ct</sub> (Ω)	CPE(Ω)
CoNi <sub>0.04</sub> -MOF-74/MXene	1.687	8.518	0.72
Co-MOF-74/MXene	2.299	12.2	0.69
CoNi <sub>0.04</sub> -MOF-74/NF	1.619	33.69	0.72
MXene/NF	1.662	72.43	0.74

**Table S6.** Comparison of Electrochemical performances of OWS of CoNi<sub>0.04</sub>-MOF-74/MXene/NF with recently reported OWS electrocatalysts in 1 M KOH.

Catalyst	Cell voltage [V] at 10 mA cm <sup>-2</sup>	References
CoNi <sub>0.04</sub> -MOF-74/MXene/NF	1.49	Our Work
Ti <sub>2</sub> NT <sub>x</sub> @MOF-CoP	1.61	[2]
CdFe-BDC	1.64	[9]
CoNiBDC/CC	1.625	[10]
Ni-ZIF/Ni-B@NF	1.54	[11]
Fe <sub>2</sub> Ni-MIL-88B/NFF	1.56	[12]
NiFe-MOF-74	1.58	[13]
NiFe-MOF	1.57	[14]
FeMn <sub>6</sub> Ce <sub>0.5</sub> -MOF-74/NF	1.65	[15]

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