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## Supporting Information

Article

# Mitigating Co Metal Particle Agglomeration and Enhancing ORR Catalytic Activity through Nitrogen-Enriched Porous Carbon Derived from Biomass

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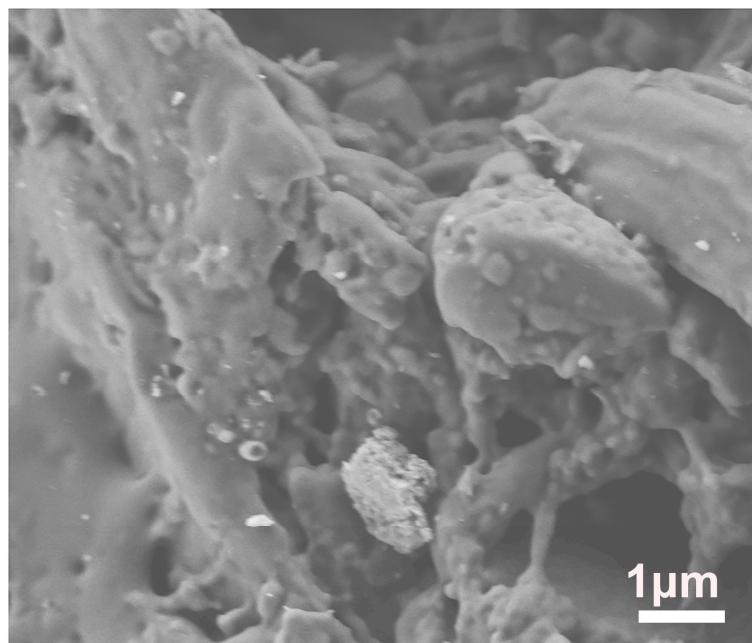
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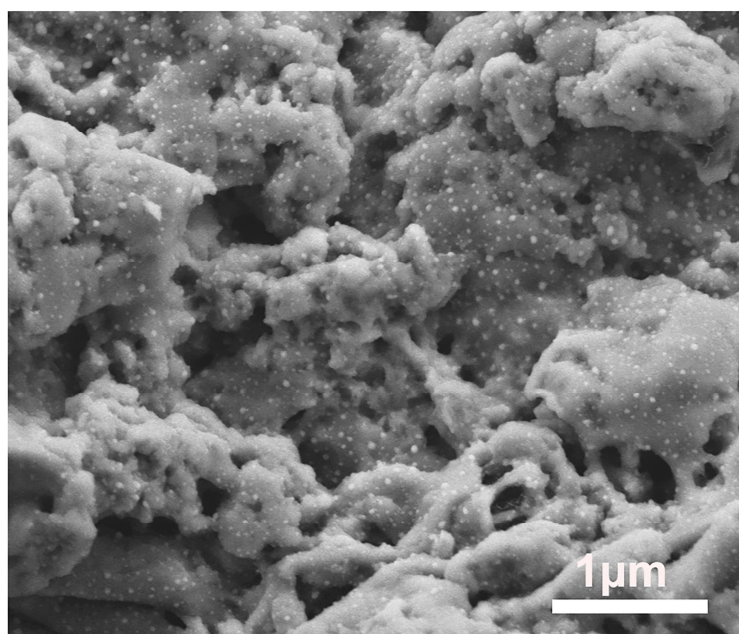
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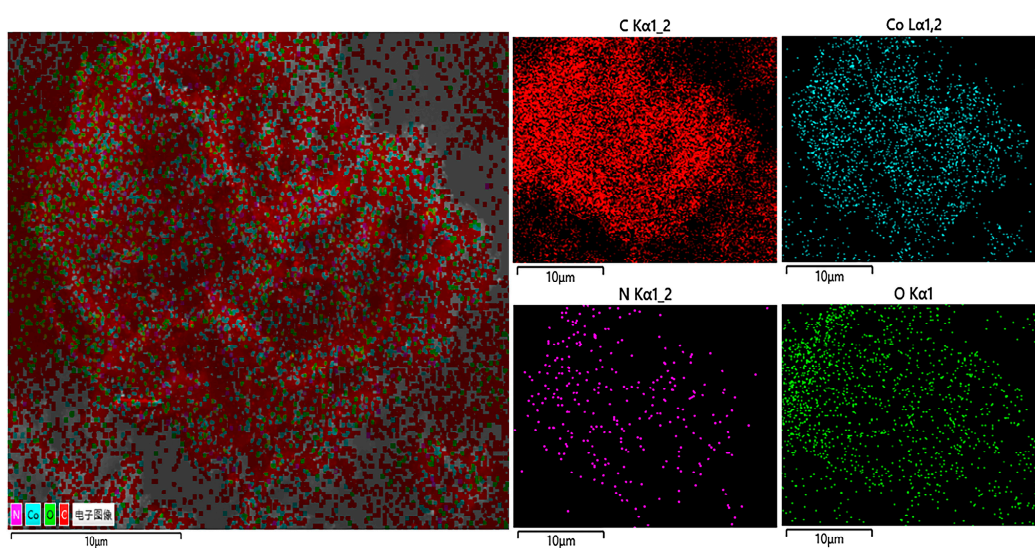
## Supplementary figures



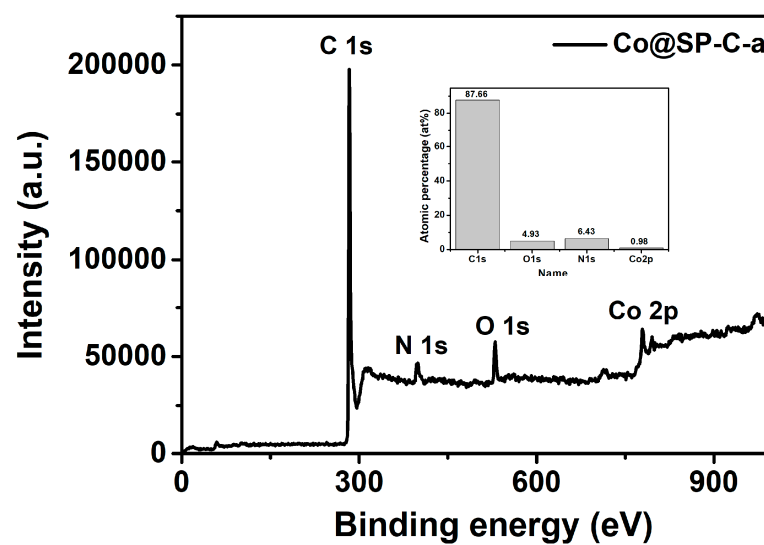
**Figure S1.** The SEM image of SP-C.



**Figure S2.** The SEM image of SP-C-a.



**Figure S3.** SEM image of Co@SP-C-a used in the EDS mapping area revealing the elemental distribution of Co, C, N, and O.



**Figure S4.** XPS survey spectra of Co@SP-C-a.

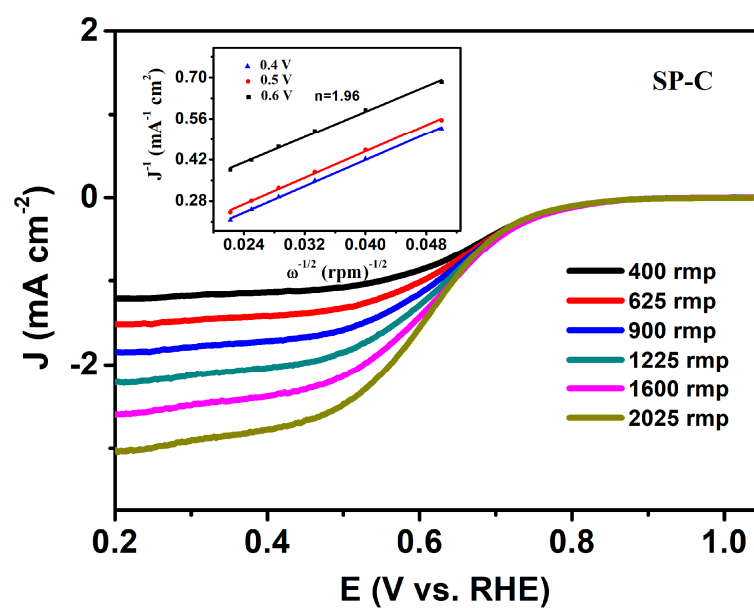


Figure S5. LSV curves of SP-C at various rotating speeds, respectively. (Inset: K-L plots of SP-C at various potentials.).

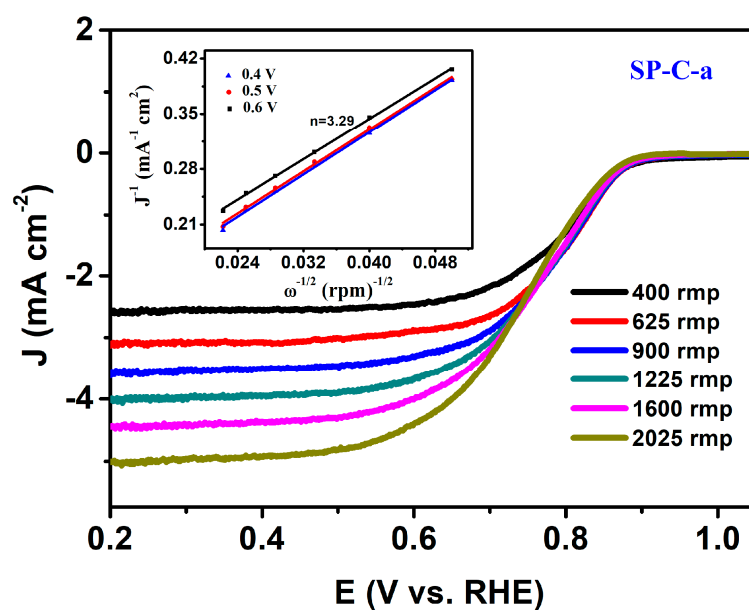
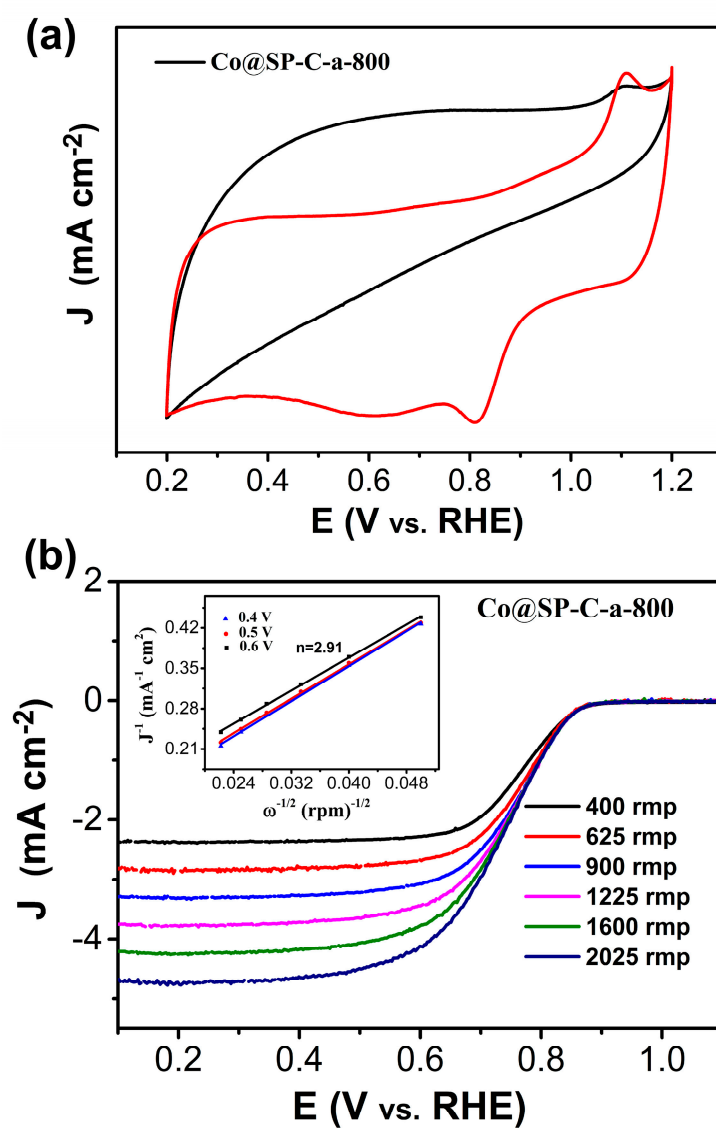


Figure S6. LSV curves of SP-C-a at various rotating speeds, respectively. (Inset: K-L plots of SP-C-a at various potentials.).





**Figure S7.** CV and LSV curves of SP-C-a-800 at various rotating speeds, respectively. (Inset: K–L plots of SP-C-a-800 at various potentials.).

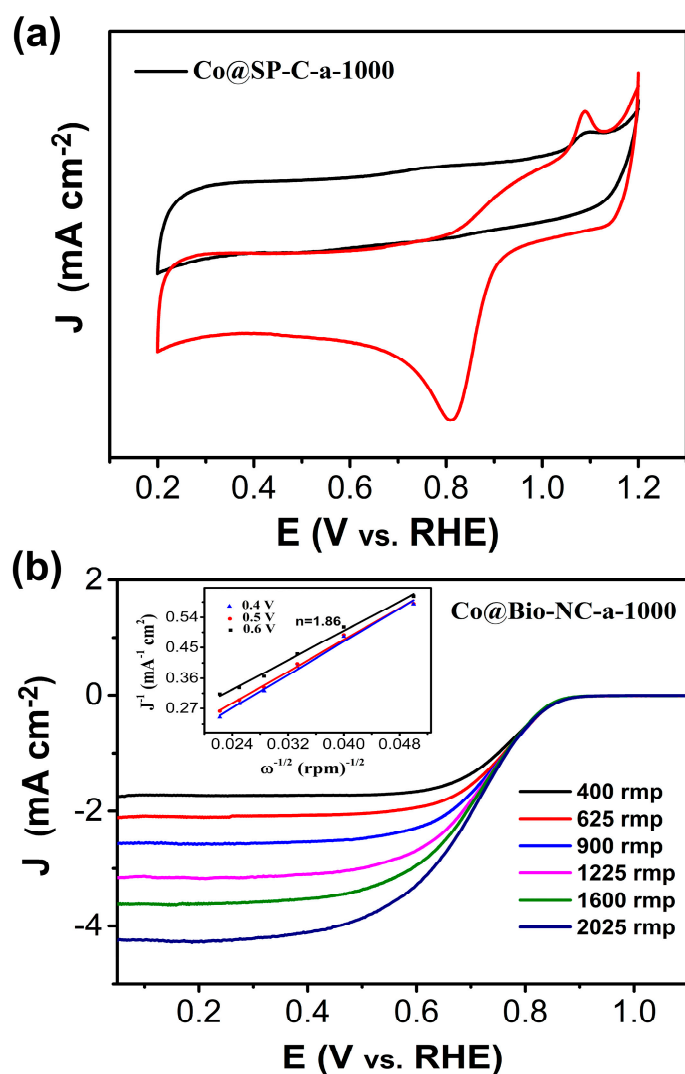


Figure S8. CV and LSV curves of SP-C-a-1000 at various rotating speeds, respectively. (Inset: K-L plots of SP-C-a-1000 at various potentials.).

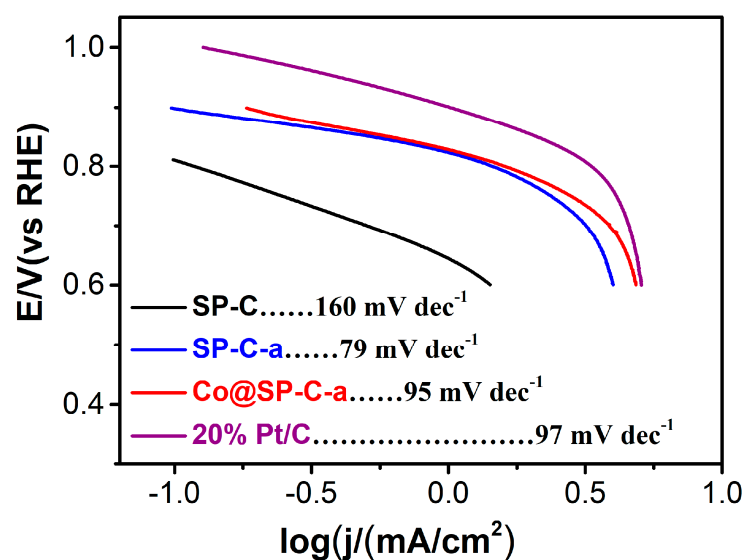
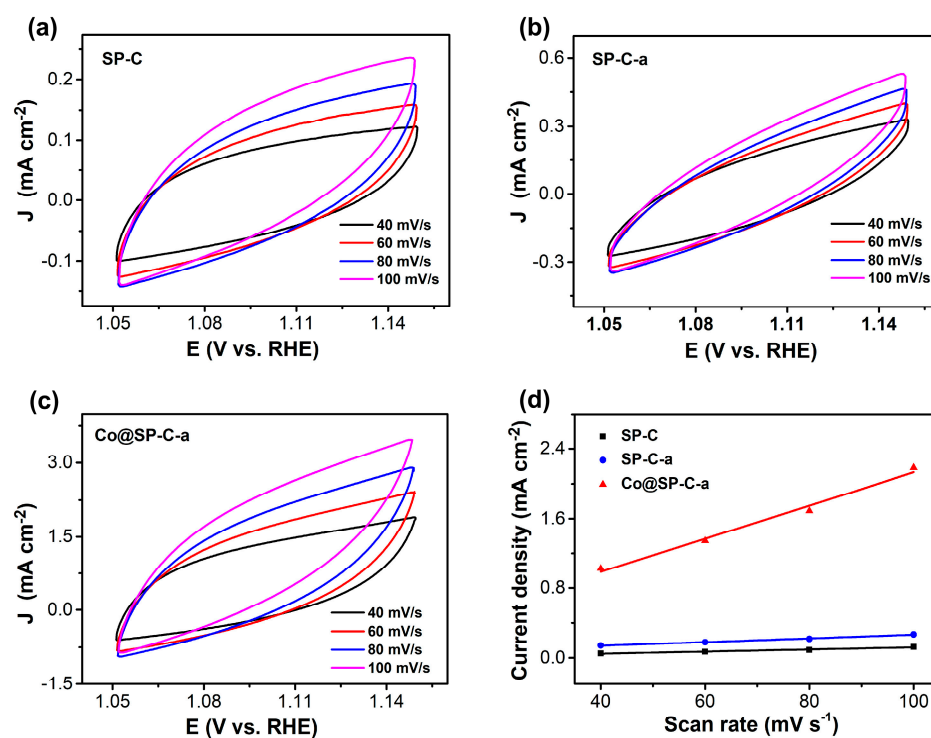
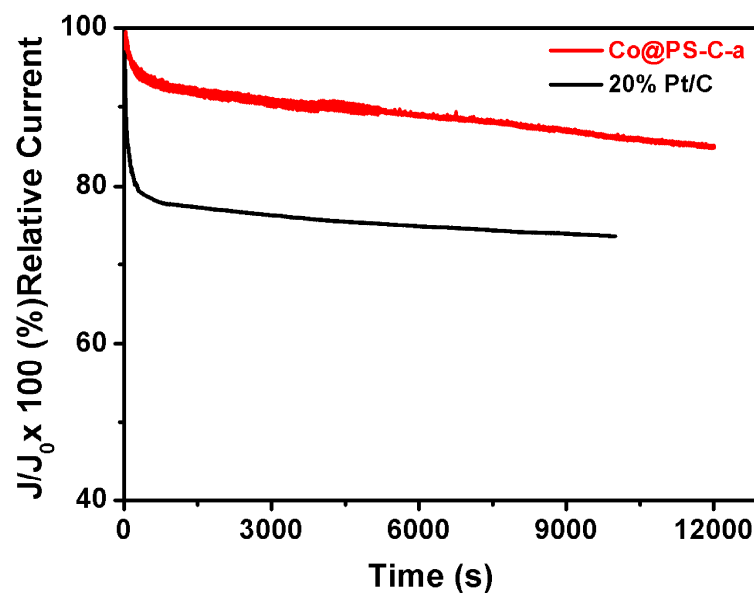


Figure S9. Tafel plots of SP-C, SP-C-a, Co@SP-C-a, and 20% Pt/C were obtained via LSV data.



**Figure S10.** Cyclic voltammograms (CV) at various scan rates of (a) SP-C, (b) SP-C-a, and (c) Co@SP-C-a in 0.1 M KOH solution. (d) The electrochemical double-layer capacitance ( $C_{dl}$ ) of SP-C, SP-C-a, and Co@SP-C-a in 0.1 M KOH electrolyte.



**Figure S11.** Amperometric  $i-t$  curves of Co@SP-C-a and 20 wt% Pt/C in  $O_2$ -saturated 0.1 M KOH solution with the rotation speed of 1600 rpm.

**Table S1.** The ORR performance of the SP-C, SP-C-a, Co@SP-C-a-800, Co@SP-C-a-1000, and Co@SP-C-a, in alkaline media at 1600 rpm, respectively.

Sample	$E_{\text{onset}}$ (V)	$E_{1/2}$ (V)	$J_L$ (mA cm <sup>-2</sup> )	$n$	Tafel (mV dec <sup>-1</sup> )
SP-C	0.76	0.57	2.58	1.96	160
SP-C-a	0.86	0.78	4.45	3.29	75
Co@SP-C-a-800	0.87	0.81	4.25	2.91	/
Co@SP-C-a-1000	0.85	0.83	3.62	1.86	/
Co@SP-C-a	0.89	0.81	6.13	3.96	95
20 wt% Pt/C	0.96	0.86	5.61	4.00	97

**Table S2.** Comparison of the ORR performance for Co@SP-C-a catalysts at 1600 rpm in 0.1 M KOH.

Catalysts	$E_{1/2}$ (V)	$J_L$ (mA cm <sup>-2</sup> )	$E_{\text{onset}}$ (V)	Tafel slope (mV dec <sup>-1</sup> )	$n$	Reference
Co@SP-C-a	0.83	6.13	0.89	--	3.96	This work
Co@Co <sub>3</sub> O <sub>4</sub> @C	0.78	4.65	0.90	--	--	<i>Energy Environ. Sci.</i> , 2015 [1]
Co@NG-acid	0.83	4.00	0.90	--	3.90	<i>Adv. Funct. Mater.</i> , 2016 [2]
Co-NC@CoP-NC	0.78	4.70	0.89	--	3.97	<i>J. Mater. Chem. A</i> , 2016 [3]
Co@Co <sub>3</sub> O <sub>4</sub> @PPD	0.78	4.20	0.90	--	3.78~3.96	<i>Small</i> , 2016 [4]
Co@NCNT	0.83	6.20	1.03	--	--	<i>J. Mater. Chem. A</i> , 2016 [5]
Co <sub>2</sub> N-CNF	0.85	5.71	0.92	60	4.00	<i>Adv. Mater.</i> , 2016 [6]
Cal-CoZIF-VXC72	0.84	5.92	--	35	4.00	<i>Adv. Mater.</i> , 2017 [7]
Co <sub>3</sub> O <sub>4</sub> /Co-N-C	0.91	5.10	0.98	68.5	3.62	<i>Journal of Power Sources</i> , 2017 [8]
Co/NPC	0.79	5.46	0.91	--	3.85~4.00	<i>J Mater Sci</i> , 2018 [9]
Co <sub>3</sub> O <sub>4</sub> /Co@N-G	0.81	5.20	0.96	--	3.91~3.96	<i>J. Mater. Chem. A</i> , 2019 [10]
Co@NPC/C-MWCNTs)	0.79	4.39	0.87	40.3	3.51	<i>Chemical Engineering Journal</i> , 2022 [11]
Co@NCNTs-700-2	0.81	7.4	0.89	--	3.92~4.13	<i>Journal of Alloys and Compounds</i> , 2022 [12]
Co-ZIF <sub>x</sub> /CNF <sub>y</sub>	0.85	2.96	0.93	--	4.00	<i>Journal of Colloid and Interface Science</i> , 2023 [13]
Co@NC	0.895	4.6	--	61	3.48	<i>Journal of Electroanalytical Chemistry</i> , 2023 [14]

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