

# Exploring Simultaneous Upgrading and Purification of Biomass–Gasified Gases Using Plasma Catalysis

Pages: 9; Tables: 4; Figures: 6

**Figure S1.** XRD analyses of the La–Ni/ $\gamma$ - $\text{Al}_2\text{O}_3$  catalyst calcined at 600 °C.

**Figure S2.** TEM analysis and particle size distributions for fresh catalyst and spent catalysts after 2 h reaction at 330 °C and 450 °C.

**Figure S3.** (a) FTIR spectra, (b–e) C 1 s spectra and relative contents of functional groups for spent catalysts under different conditions, and (f) La 3d spectra and (g) Ni 2p spectra for spent catalyst under MRO at 450 °C.

**Table S1.** Relative contents of the peaks of the spent catalysts.

**Table S2.**  $\text{H}_2$  molar flow from plasma–coupled catalytic processes.

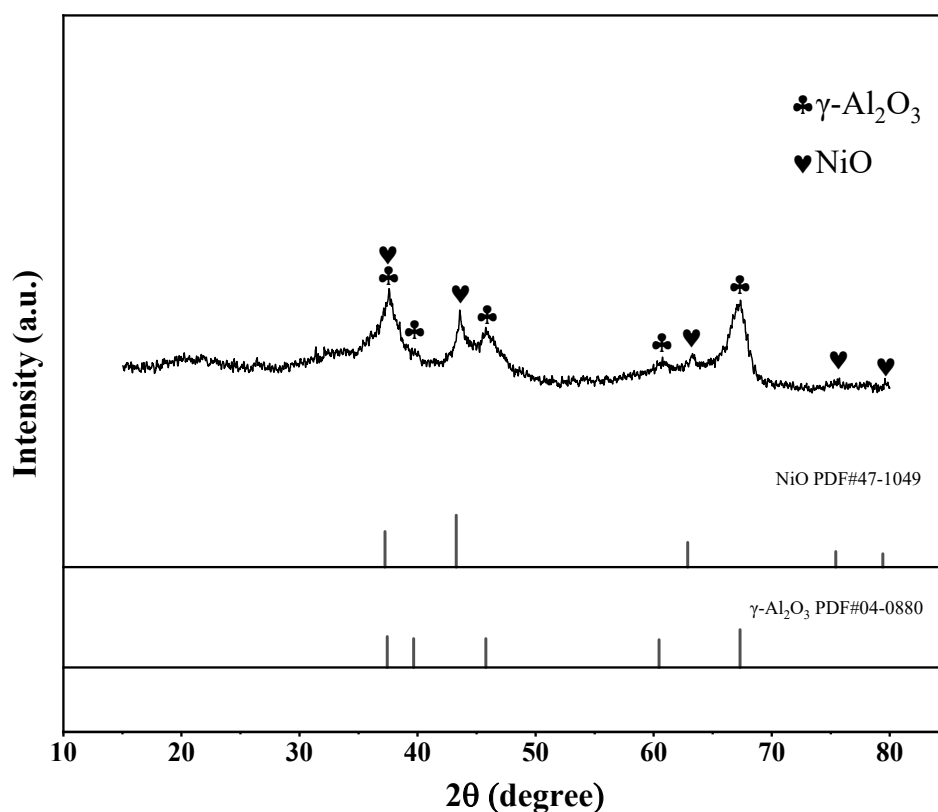
**Table S3.** Catalyst characterization data.

**Figure S4.** (a, b)  $\text{N}_2$  adsorption–desorption isotherms and (c, d) BJH pore width distributions for fresh and spent catalysts.

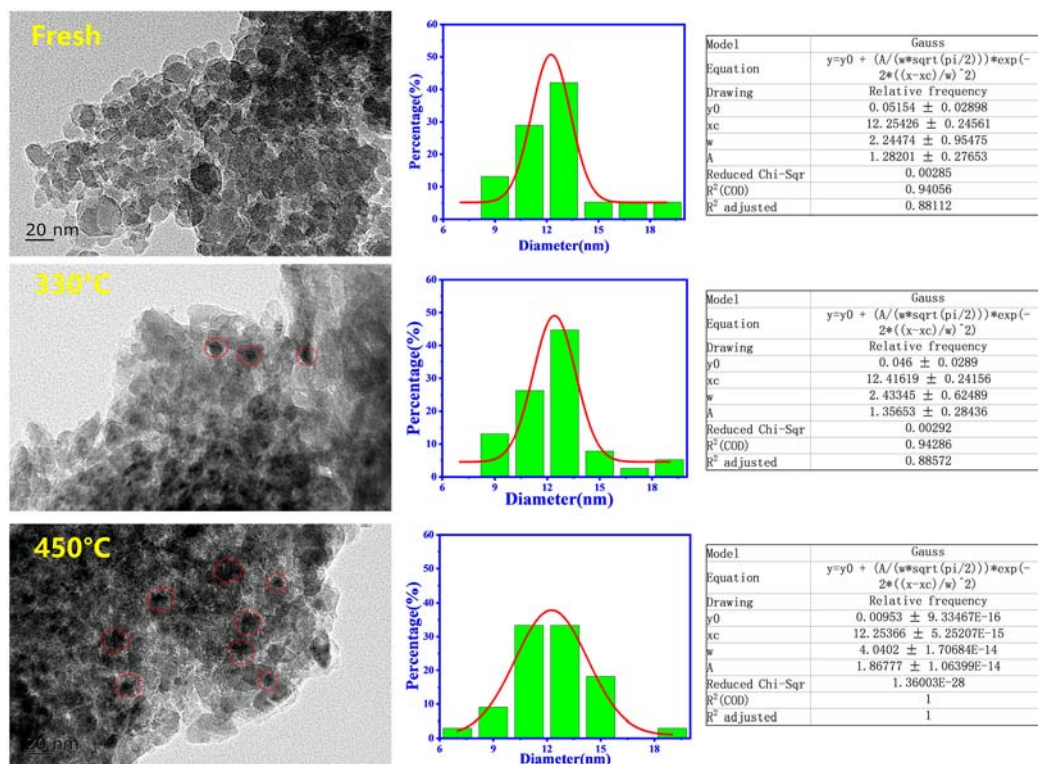
**Figure S5.** GC–MS spectrogram of benzene removal under a gas mixture ( $\text{N}_2$ ,  $\text{CH}_4$  and  $\text{CO}_2$ ) atmosphere and a pure  $\text{N}_2$  atmosphere.

**Table S4.** Main products identified in the liquid sample by GC–MS.

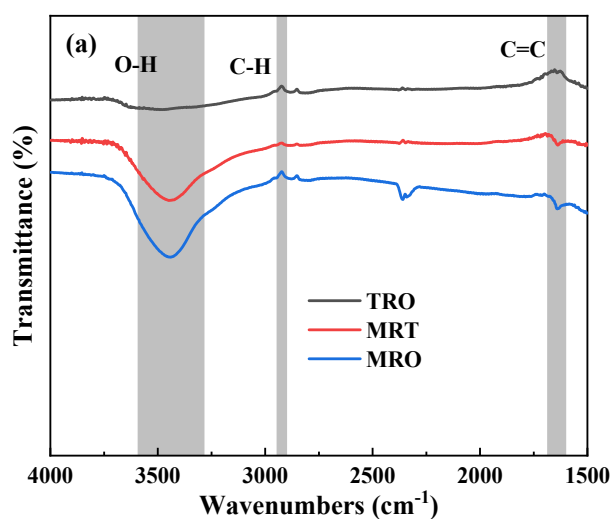
**Figure S6.** Schematic diagram of the experimental setup.

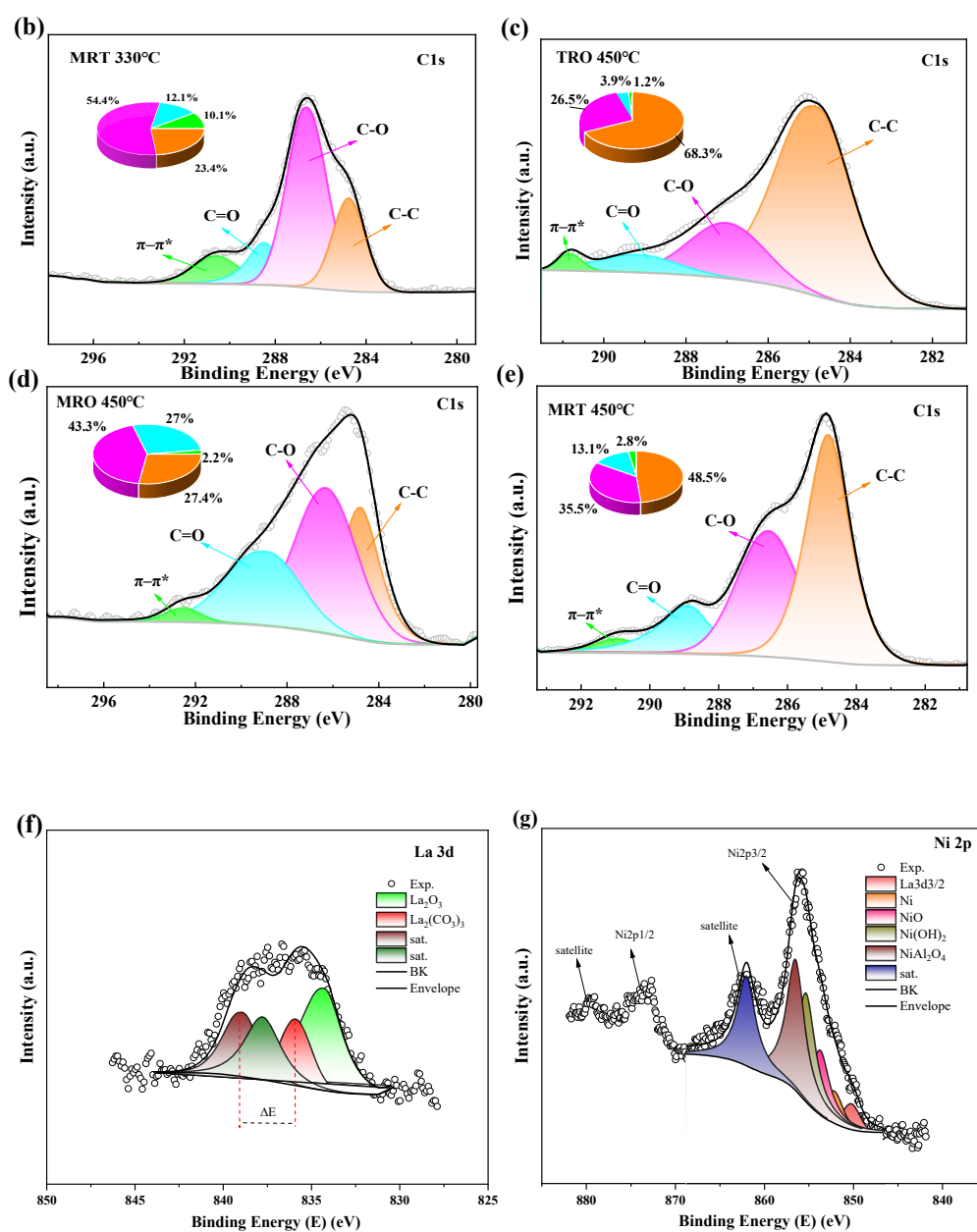


**Figure S1.** XRD analyses of the La–Ni/ $\gamma$ - $\text{Al}_2\text{O}_3$  catalyst calcined at 600 °C.



**Figure S2.** TEM analysis and particle size distributions for fresh catalyst and spent catalysts after 2 h reaction at 330 °C and 450 °C.





**Figure S3.** (a) FTIR spectra, (b–e) C 1 s spectra and relative contents of functional groups for spent catalysts under different conditions, and (f) La 3d spectra and (g) Ni 2p spectra for spent catalyst under MRO at 450 °C.

**Table S1.** Relative contents of the peaks of the spent catalysts.

Samples	C-C (%)	C-O (%)	C=O (%)	O content (%)	$\pi-\pi^*$ (%)
MRT-330	23.4	54.4	12.1	66.5	10.1
MRT-450	48.5	35.5	13.1	48.6	2.8
MRO-450	27.4	43.3	27.0	70.3	2.2
TRO-450	68.3	26.5	3.9	30.4	1.2

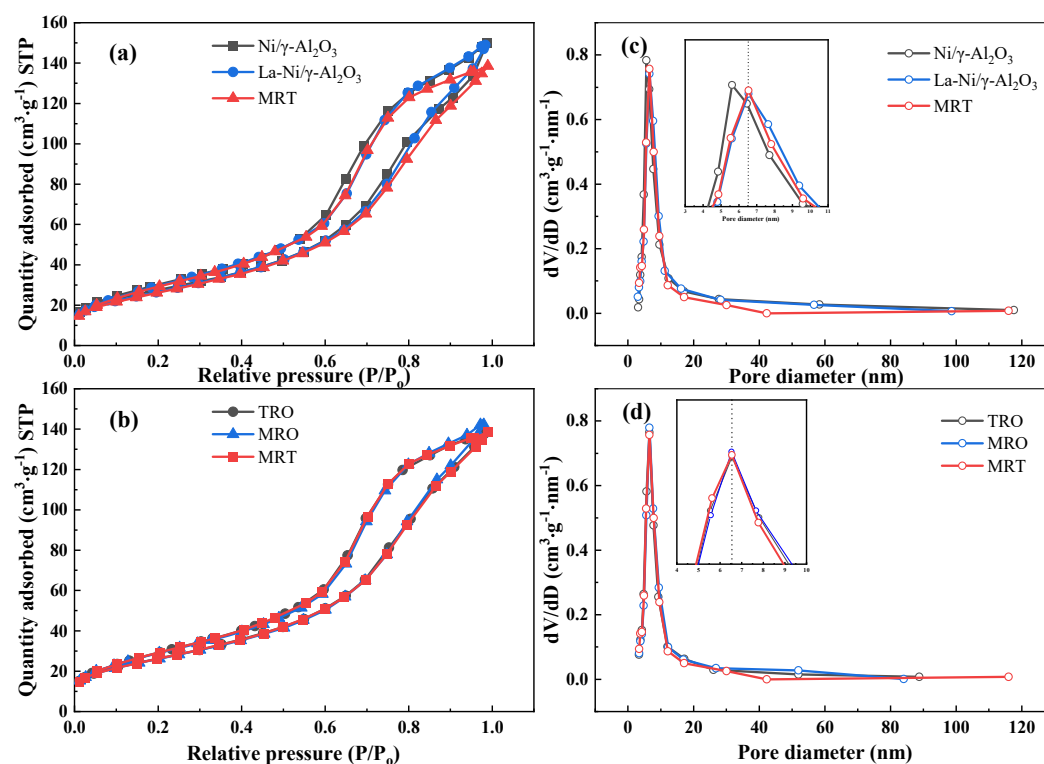
**Table S2.** H<sub>2</sub> molar flow from plasma-catalytic processes.

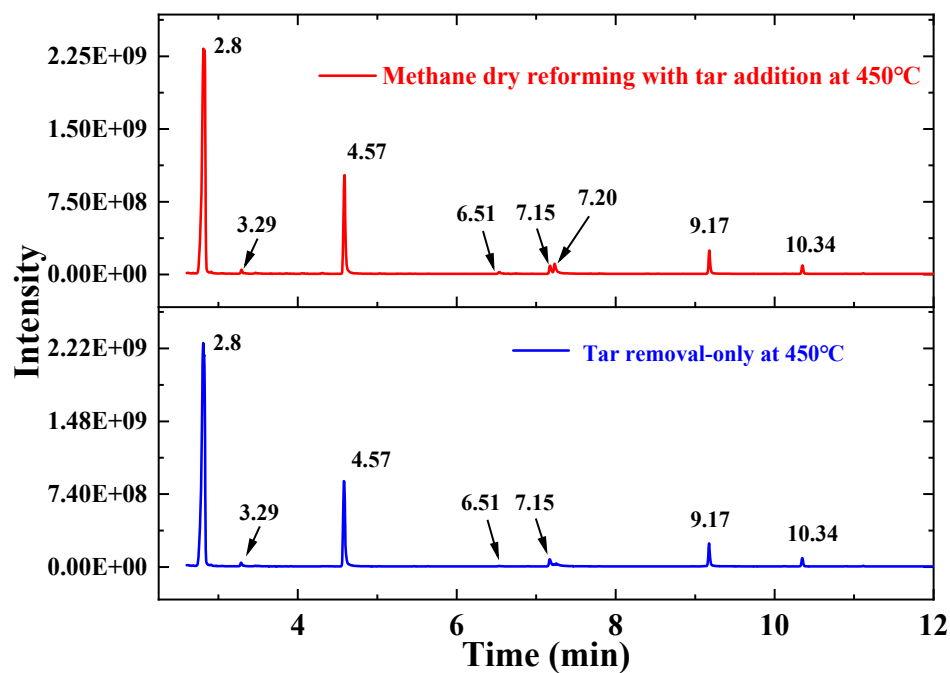
Process	Units	Temperature (°C)		
		250	288	330
MRT	mmol/min	0.56	0.38	0.43
MRO	mmol/min	0.89	0.95	1.05
TRO	mmol/min	0.09	0.11	0.12

**Table S3.** Catalyst characterization data.

Catalyst	BET specific surface area (m <sup>2</sup> ·g <sup>-1</sup> )	Total pore volume (cm <sup>3</sup> ·g <sup>-1</sup> )	Average pore size (nm)
γ-Al <sub>2</sub> O <sub>3</sub>	110.3	0.26	9.35
Ni/γ-Al <sub>2</sub> O <sub>3</sub>	99.1	0.23	9.35
La-Ni/γ-Al <sub>2</sub> O <sub>3</sub>	97.4	0.23	9.45
MRT	96.5	0.21	8.87
TRO	95.9	0.21	8.94
MRO	95.7	0.22	9.18

(MRT = Methane dry reforming with tar addition, MRO = Methane dry reforming-only, TRO = Tar removal-only) (Temperature: 450 °C, discharge power: 35 W, CO<sub>2</sub>/CH<sub>4</sub> ratio: 1, total flow rate: 300 ml/min, tar concentration: 29.6 g/Nm<sup>3</sup>, catalyst: 2%La-10%Ni/γ-Al<sub>2</sub>O<sub>3</sub>).

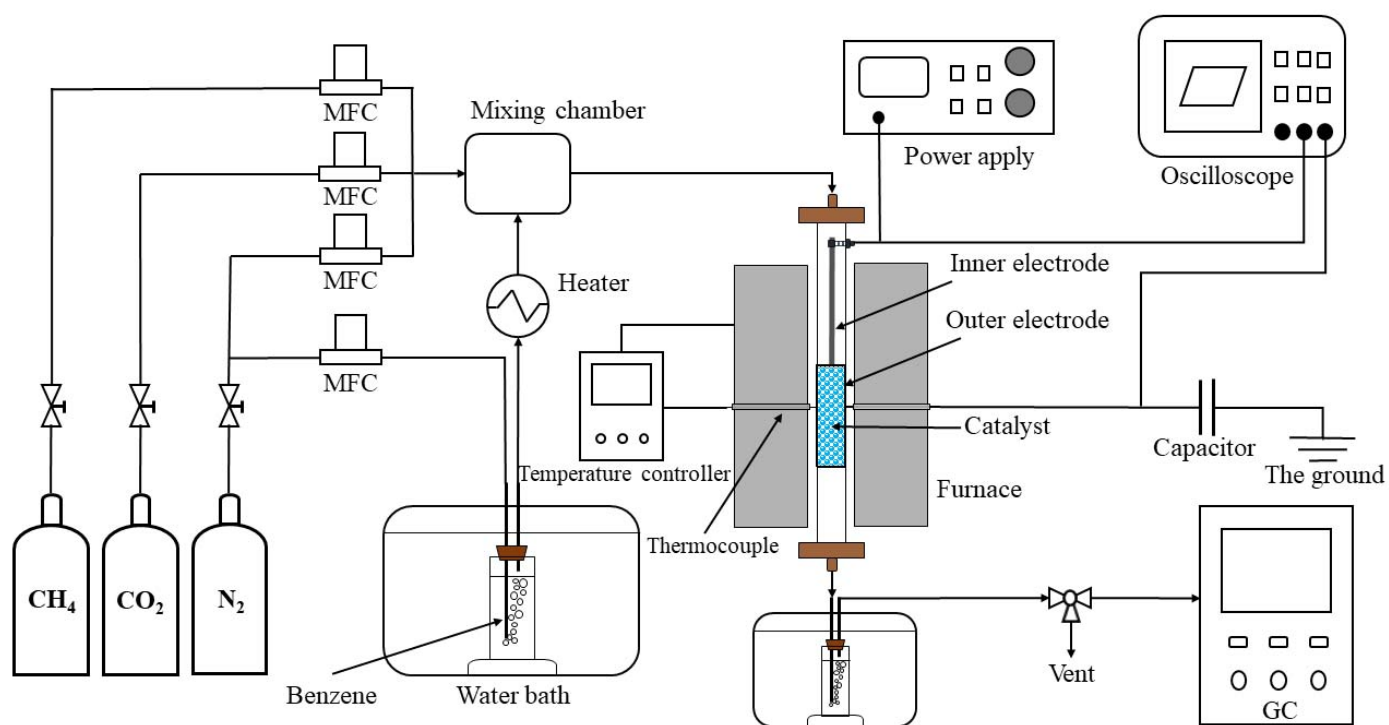
**Figure S4.** (a, b) N<sub>2</sub> adsorption-desorption isotherms and (c, d) BJH pore width distributions for fresh and spent catalysts.



**Figure S5.** GC–MS spectrogram of benzene removal under a gas mixture ( $\text{N}_2$ ,  $\text{CH}_4$  and  $\text{CO}_2$ ) atmosphere and a pure  $\text{N}_2$  atmosphere.

**Table S4.** Main products identified in the liquid sample by GC–MS.

No.	R.T.(min)	M.W.	Formula	Compounds name	Structure	Peak Area (%)	
						MRT	TRO
1	2.8	78	$\text{C}_6\text{H}_6$	Benzene		71.68	73.28
2	3.29	100	$\text{C}_7\text{H}_{16}$	Heptane		0.61	0.62
3	4.57	92	$\text{C}_7\text{H}_8$	Toluene		18.05	16.86
4	6.51	106	$\text{C}_8\text{H}_{10}$	Ethylbenzene		0.44	0.13
5	7.15	104	$\text{C}_8\text{H}_8$	Styrene		1.6	1.47
6	7.20	128	$\text{C}_7\text{H}_{12}\text{O}_2$	2-Propenoic acid, butylester		1.9	/
7	9.17	142	$\text{C}_{10}\text{H}_{22}$	Decane		3.66	3.85
8	10.34	138	$\text{C}_{10}\text{H}_{18}$	trans-Decalin		1.37	1.38



**Figure S6.** Schematic diagram of the experimental setup.