

Supplementary Materials

Electrocatalytic Oxygen Reduction to Hydrogen Peroxide on Graphdiyne-Based Single-Atom Catalysts: First-Principles Studies

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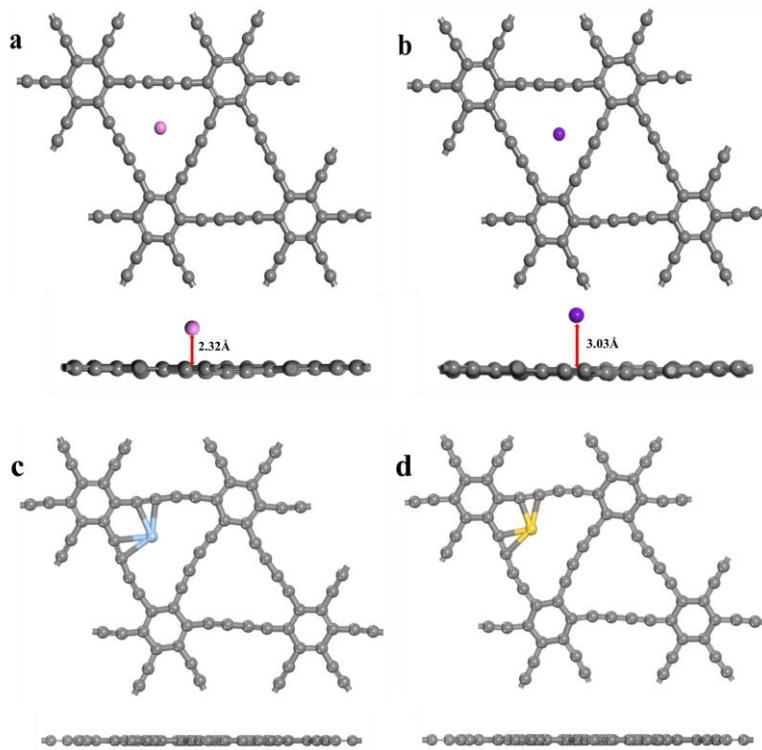


Figure S1. Top and side views of the (a) Zn atom, (b) Cd atom, (c) Ag atom and (d) Au atom supported on GDY. Color scheme: C: Gray; Zn: Pink; Cd: Purple; Ag: Blue; Au: Yellow.

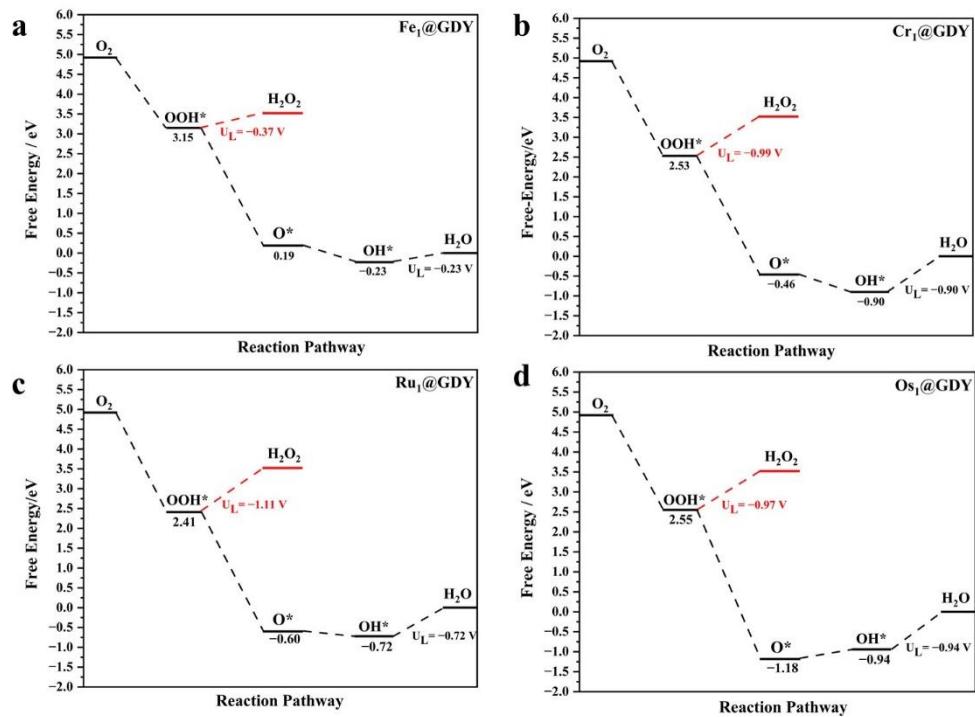


Figure S2. Free energy diagrams of 2e⁻ and 4e⁻ ORR on (a): Fe₁@GDY; (b): Cr₁@GDY; (c): Ru₁@GDY; d: Os₁@GDY.

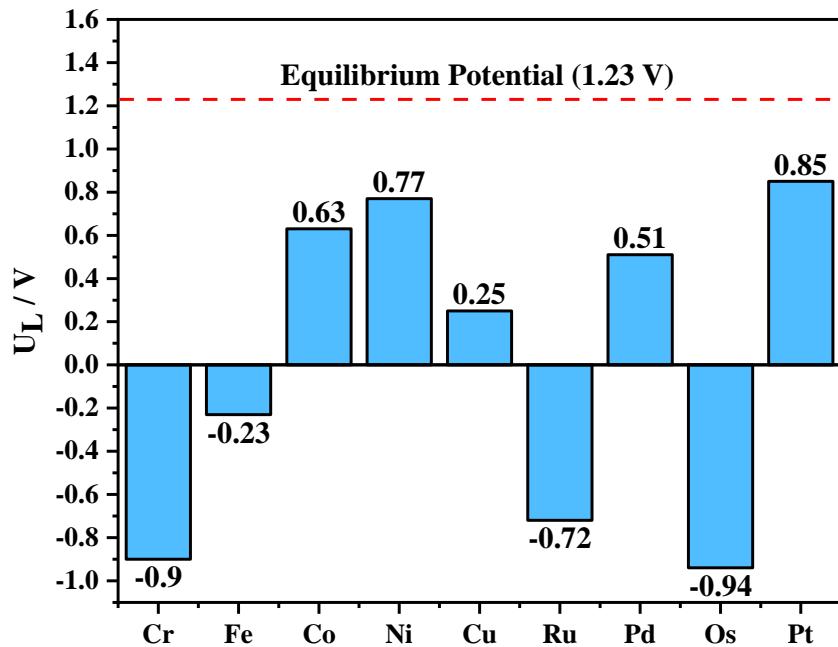


Figure S3. Limiting potentials of 4e⁻ ORR on TM₁@GDY (TM=Cr, Fe, Co, Ni, Cu, Ru, Pd, Os, and Pt) catalysts. The red line represents the equilibrium potential of 4e⁻ ORR.

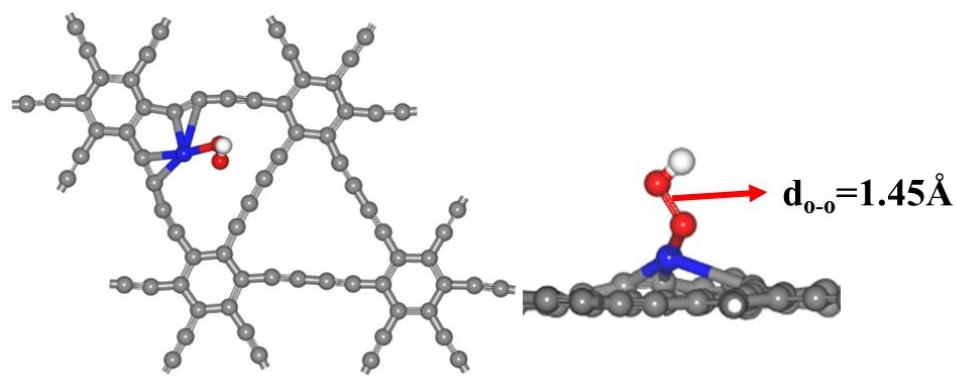


Figure S4. The adsorption structure of OOH^* on $\text{Pt}_1@\text{GDY}$ and the $\text{O}-\text{O}$ bond length.

Table S1. The E_b on $\text{TM}_1@\text{GDY}$.

$\text{TM}_1@\text{GDY}$	E_b (eV)
$\text{Sc}_1@\text{GDY}$	-4.41
$\text{Ti}_1@\text{GDY}$	-4.57
$\text{V}_1@\text{GDY}$	-3.54
$\text{Cr}_1@\text{GDY}$	-3.62
$\text{Mn}_1@\text{GDY}$	-2.39
$\text{Fe}_1@\text{GDY}$	-3.39
$\text{Co}_1@\text{GDY}$	-4.08
$\text{Ni}_1@\text{GDY}$	-4.26
$\text{Cu}_1@\text{GDY}$	-2.00
$\text{Zn}_1@\text{GDY}$	-0.19
$\text{Y}_1@\text{GDY}$	-4.67
$\text{Zr}_1@\text{GDY}$	-5.54
$\text{Nb}_1@\text{GDY}$	-5.58
$\text{Mo}_1@\text{GDY}$	-2.96
$\text{Ru}_1@\text{GDY}$	-4.65
$\text{Rh}_1@\text{GDY}$	-4.01

Pd ₁ @GDY	-2.68
Ag ₁ @GDY	-0.71
Cd ₁ @GDY	-0.19
Hf ₁ @GDY	-5.20
Ta ₁ @GDY	-5.15
W ₁ @GDY	-3.87
Re ₁ @GDY	-3.28
Os ₁ @GDY	-5.13
Ir ₁ @GDY	-4.89
Pt ₁ @GDY	-4.06
Au ₁ @GDY	-0.85

Table S2. The GooH* on TM₁@GDY.

TM ₁ @GDY	GooH* (eV)
Sc ₁ @GDY	2.07
Ti ₁ @GDY	1.61
V ₁ @GDY	1.63
Cr ₁ @GDY	2.53
Mn ₁ @GDY	2.95
Fe ₁ @GDY	3.15
Co ₁ @GDY	3.44
Ni ₁ @GDY	3.76
Cu ₁ @GDY	3.50
Y ₁ @GDY	2.17
Zr ₁ @GDY	1.37
Ru ₁ @GDY	2.41
Rh ₁ @GDY	3.17
Pd ₁ @GDY	4.41

Hf ₁ @GDY	1.57
Os ₁ @GDY	2.55
Ir ₁ @GDY	2.94
Pt ₁ @GDY	4.07