

# Catalytic Effect of CO<sub>2</sub> and H<sub>2</sub>O Molecules on $\cdot\text{CH}_3 + {}^3\text{O}_2$ Reaction

Mohamad Akbar Ali <sup>1,\*</sup>, Manas Ranjan Dash <sup>2</sup> and Latifah Mohammed Al Maieli <sup>1</sup>

<sup>1</sup> Department of Chemistry, College of Science, King Faisal University, Al-Ahsa 31982, Saudi Arabia; almaielilatifah@gmail.com

<sup>2</sup> Department of Chemistry, DIT University, Uttarakhand 248009, India; manas.iitm@gmail.com

\* Correspondence: aamohamad@kfu.edu.sa

**Abstract:** The methyl ( $\cdot\text{CH}_3$ ) +  ${}^3\text{O}_2$  radical is an important reaction in both atmospheric and combustion processes. We investigated potential energy surfaces for the effect of CO<sub>2</sub> and H<sub>2</sub>O molecules on a  $\cdot\text{CH}_3 + \text{O}_2$  system. The mechanism for three reaction systems, i.e., for  $\cdot\text{CH}_3 + {}^3\text{O}_2$ ,  $\cdot\text{CH}_3 + {}^3\text{O}_2$  (+CO<sub>2</sub>) and  $\cdot\text{CH}_3 + {}^3\text{O}_2$  (+H<sub>2</sub>O), were explored using *ab initio*/DFT methods [CCSD(T)//M062X/6-311++G(3df,3pd)] in combination with a Rice–Ramsperger–Kassel–Marcus (RRKM)/master-equation (ME) simulation between a temperature range of 500 to 1500 K and a pressure range of 0.0001 to 10 atm. When a CO<sub>2</sub> and H<sub>2</sub>O molecule is introduced in a  $\cdot\text{CH}_3 + {}^3\text{O}_2$  reaction, the reactive complexes, intermediates, transition states and post complexes become thermodynamically more favorable. The calculated rate constant for the  $\cdot\text{CH}_3 + {}^3\text{O}_2$  ( $3 \times 10^{-15} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$  at 1000 K) is in good agreement with the previously reported experimentally measured values ( $\sim 1 \times 10^{-15} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$  at 1000 K). The rate constant for the effect of CO<sub>2</sub> ( $3 \times 10^{-16} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$  at 1000 K) and H<sub>2</sub>O ( $2 \times 10^{-17} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$  at 1000 K) is at least one–two-order magnitude smaller than the free reaction ( $3 \times 10^{-15} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$  at 1000 K). The effect of CO<sub>2</sub> and H<sub>2</sub>O on  $\cdot\text{CH}_3 + {}^3\text{O}_2$  shows non-RRKM behavior, however, the effect on  $\cdot\text{CH}_3 + {}^3\text{O}_2$  shows RRKM behavior. Our results also demonstrate that a single CO<sub>2</sub> and H<sub>2</sub>O molecule has the potential to accelerate a gas-phase reaction at temperature higher than >1300 K and slow the reaction at a lower temperature. The result is unique and observed for the first time.

**Keywords:**  $\cdot\text{CH}_3$  radical;  ${}^3\text{O}_2$  radical; water and CO<sub>2</sub> catalysis; *ab initio*/DFT; RRKM/ME

**Table S1.** Redundant internal coordinates of reactants, complexes, products and transition states obtained using M06-2X/6-311++G(3df,3pd) taken from gaussian checkpoint file.

(a) CH <sub>3</sub> + O <sub>2</sub> ==> CH <sub>2</sub> O + HO								
CH <sub>3</sub>	O <sub>2</sub>	RC	TS1	Int-1	TS2	Int-2	TS3	PC
Charge = 0 Multiplic- ity = 2	Charge = 0 Multi- plicity = 3	Charge = 0 Multiplicity = 2	Charge = 0 Multiplicity = 2	Charge = 0 Multiplicity = 2	Charge = 0 Multiplicity = 2	Charge = 0 Multiplicity = 2	Charge = 0 Multiplic- ity = 2	Charge = 0 Multiplicity = 2
C,0,0.,0.000 0001111,0.	O,0,0.,0.,0.5 938390313 O,0,0.,0.,- 0.59383903 13	C,0,1.89977 14092,- 0.066472136 8,0.0062984 681	C,0,1.43396 38582,- 0.204606976 1,- 0.041036864 1	C,0,1.08346 6449,- 0.180447232 2,- 0.000108254 4	C,0,0.93885 90949,- 0.254499968 2,0.0000010 222 H,0,- 0.064670196 4,- 1.117653894 6,- 0.000008890 6	1.118275367 9,0.2495177 688,0.10092 69949 O,0,- 0.061659221 9,- 0.572175002 7,- 0.104347832 3 H,0,- 2.048259569 3,- 0.298388570 6,0.0873675 153 H,0,- 1.028952177 6,1.2622953 239,- 0.266287484 2 O,0,1.12859 15628,0.189 7888208,- 0.098638149 2 H,0,1.40243 97739,0.124 9366599,0.8 254409554	1.14591277 75,0.26689 63639,0.15 15185491 O,0,0.1140 577578,- 0.56567448 01,- 0.06913651 58 C,0,1.1500 525939,0.2 192667703, - 0.02462498 7 H,0,- 1.70689488 91,- 0.13339096 24,- 0.52411213 22 H,0,1.9409 175785,- 0.03860070 85,0.66804 10743 H,0,1.1244 157364,1.1 542780168, - 0.56873798 85	C,0,1.57777 80365,0.282 2699021,- 0.001587569 5 H,0,2.64784 61523,0.025 6072139,0.0 282689007 H,0,1.30142 3077,1.3485 055912,- 0.023589027 6 O,0,- 1.319591142 2,1.3914153 616,- 0.082323831 8 O,0,0.73591 14379,- 0.572702514 ,- 0.009945152 4 H,0,- 0.894074801 5,0.5112897 552,- 0.061304729 4
H,0,0.,1.077 7366481,0.		0.066472136 8,0.0062984 681	0.204606976 1,- 0.041036864 1	0.180447232 2,- 0.000108254 4	0.254499968 2,0.0000010 222 H,0,- 0.064670196 4,- 1.117653894 6,- 0.000008890 6	688,0.10092 69949 O,0,- 0.061659221 9,- 0.572175002 7,- 0.104347832 3 H,0,- 2.048259569 3,- 0.298388570 6,0.0873675 153 H,0,- 1.028952177 6,1.2622953 239,- 0.266287484 2 O,0,1.12859 15628,0.189 7888208,- 0.098638149 2 H,0,1.40243 97739,0.124 9366599,0.8 254409554	63639,0.15 15185491 O,0,0.1140 577578,- 0.56567448 01,- 0.06913651 58 C,0,1.1500 525939,0.2 192667703, - 0.02462498 7 H,0,- 1.70689488 91,- 0.13339096 24,- 0.52411213 22 H,0,1.9409 175785,- 0.03860070 85,0.66804 10743 H,0,1.1244 157364,1.1 542780168, - 0.56873798 85	2699021,- 0.001587569 5 H,0,2.64784 61523,0.025 6072139,0.0 282689007 H,0,1.30142 3077,1.3485 055912,- 0.023589027 6 O,0,- 1.319591142 2,1.3914153 616,- 0.082323831 8 O,0,0.73591 14379,- 0.572702514 ,- 0.009945152 4 H,0,- 0.894074801 5,0.5112897 552,- 0.061304729 4
H,0,0.93334 72196,- 0.538868157 4,0. H,0,- 0.933347219 6,- 0.538868157 4,0.		H,0,2.85065 84729,0.428 5997927,- 0.106729961 3 H,0,1.46189 40515,- 0.179038467 6,0.9847565 221 H,0,1.42999 04275,- 0.534728805 5,- 0.843081135 9 O,0,- 1.196362315 9,- 0.534021670 5,0.0181992 062 O,0,- 0.934033045 2,0.6235922 877,- 0.031720099 2	H,0,1.22142 13268,- 0.859057023 7,- 0.869575347 6 H,0,1.98309 81651,0.703 6117787,- 0.225334869 7 H,0,1.42633 02762,- 0.613203815 2,0.9553642 996 O,0,- 1.448780223 ,- 0.223233346 4,0.0680200 964 O,0,- 0.596625403 4,0.6278073 826,0.05549 66854	H,0,1.86568 7945,0.5739 675178,- 0.001363984 5 H,0,1.12899 55479,- 0.798440088 5,- 0.894701196 4 H,0,1.12995 48664,- 0.796439147 0.89582239 44 O,0,- 0.158612522 3,0.5374362 06,- 0.000246285 6 O,0,- 1.163903286 ,- 0.275924256 0.00041832 65	H,0,1.47522 22341,- 0.371242254 0.93846064 3 H,0,1.47522 92384,- 0.371228082 3,- 0.938456372 3 O,0,- 0.022663454 3,0.7318364 354,0.00000 48771 O,0,- 1.043283916 7,- 0.307583236 3,- 0.000005199 4	1.118275367 9,0.2495177 688,0.10092 69949 O,0,- 0.061659221 9,- 0.572175002 7,- 0.104347832 3 H,0,- 2.048259569 3,- 0.298388570 6,0.0873675 153 H,0,- 1.028952177 6,1.2622953 239,- 0.266287484 2 O,0,1.12859 15628,0.189 7888208,- 0.098638149 2 H,0,1.40243 97739,0.124 9366599,0.8 254409554	1.14591277 75,0.26689 63639,0.15 15185491 O,0,0.1140 577578,- 0.56567448 01,- 0.06913651 58 C,0,1.1500 525939,0.2 192667703, - 0.02462498 7 H,0,- 1.70689488 91,- 0.13339096 24,- 0.52411213 22 H,0,1.9409 175785,- 0.03860070 85,0.66804 10743 H,0,1.1244 157364,1.1 542780168, - 0.56873798 85	C,0,1.57777 80365,0.282 2699021,- 0.001587569 5 H,0,2.64784 61523,0.025 6072139,0.0 282689007 H,0,1.30142 3077,1.3485 055912,- 0.023589027 6 O,0,- 1.319591142 2,1.3914153 616,- 0.082323831 8 O,0,0.73591 14379,- 0.572702514 ,- 0.009945152 4 H,0,- 0.894074801 5,0.5112897 552,- 0.061304729 4
(b) CH <sub>3</sub> + O <sub>2</sub> (+CO <sub>2</sub> ) ==> CH <sub>2</sub> O + HO + (CO <sub>2</sub> )								
RC-c	TS-1c	Int-1c	TS2-c	Int-2c	TS3-c	PC-c		
Redundant internal coordinates	Charge = 0	Charge = 0 Multiplic- ity = 2	Charge = 0 Multiplic- ity = 2	Charge = 0	Charge = 0 Multiplic- ity = 2	Charge = 0 Multiplicity = 2 Redundant internal coordinates found in file.		

found in file. C,0,- 0.403090903 2,2.1732845 739,- 0.039004478 H,0,- 1.033316849 5,1.8447244 599,- 0.847934667 6 H,0,0.50949 72709,2.702 5562777,- 0.256707599 4 H,0,- 0.764925361 1,2.1310466 408,0.97390 71184 O,0,- 2.053520643 6,- 0.898983533 5,- 0.293291493 O,0,- 1.345426574 9,- 0.655631704 8,0.6289091 873 O,0,1.74166 44438,- 0.115754946 9,0.9595284 092 C,0,1.40808 98453,- 0.307337627 3,- 0.128229577 O,0,1.08629 37724,- 0.510644139 8,- 1.218102899 8	Multiplicit y = 2 Redundant internal coordinate s found in file. C,0,- 2.18757416 35,1.111516 921,- 0.11279309 15 H,0,- 3.24313279 32,0.908076 2668,- 0.18019295 81 H,0,- 1.68848850 65,1.587249 2361,- 0.93964384 74 H,0,- 1.72767241 97,1.160195 513,0.86020 16983 O,0,- 1.49923736 9,- 0.97506453 77,- 0.53223066 12 O,0,- 0.54845292 29,- 1.32159400 59,0.119404 0508 O,0,0.97203 02583,1.186 5989513,- 0.38349344 49 C,0,1.67914 7191,0.3711 023634,0.02 7917306 O,0,2.39607 47255,-	C,0,1.89724 99291,0.867 3673169,0.0 003322581 H,0,2.92755 40525,1.211 9081956,0.0 004787045 H,0,1.36544 31143,1.191 2976433,0.8 922755081 H,0,1.36520 10269,1.192 6737843,- 0.890956510 6 O,0,1.96841 08625,- 0.567093369 9,- 0.000791527 5 O,0,0.79168 78482,- 1.100220545 1,- 0.000739896 4 O,0,- 1.140864063 1,1.2541227 984,- 0.001995216 2 C,0,- 1.631824767 7,0.2070525 461,- 0.000073120 3 O,0,- 2.149945002 7,- 0.821434369 7,0.0018638 002	C,0,- 1.821863996 9,0.6102913 195,- 0.193757727 6 H,0,- 1.583865675 1,0.1672093 957,1.03056 90043 H,0,- 2.891808084 2,0.5359797 015,- 0.369119021 8 H,0,- 1.317235815 ,1.56756516 73,- 0.291540875 7 O,0,- 1.090426801 2,- 0.474518635 3,- 0.634087993 8 O,0,- 0.921329492 8,- 0.884697738 4,0.7552183 638 O,0,1.08787 11434,1.279 9329222,0.2 120133097 C,0,1.49123 8922,0.2198 283069,- 0.015660184 3 O,0,1.92249 27998,- 0.823149439 4,- 0.242341874 6	Multiplicity = 2 C,0,- 1.088745861 8,1.1574292 376,0.54208 65617 O,0,- 1.145686627 2,0.3542998 278,- 0.548139739 O,0,- 1.751710890 6,- 0.871169485 1,- 0.195224747 2 C,0,1.50339 62303,- 0.196138694 6,0.0287415 513 O,0,1.93196 17688,0.820 6219602,- 0.296615253 8 O,0,1.09580 19026,- 1.228845913 7,0.3576874 373 H,0,- 0.985643768 2,- 1.393561411 9,0.0852516 491 H,0,- 0.650188154 1,2.1147658 58,0.303790 1544 H,0,- 1.857373599 9,1.0249206 217,1.28944 23862	Redun- dant inter- nal coordi- nates found in file. C,0,1.97474 06962,- 1.018170174 7,0.1950561 68 O,0,1.01625 94944,- 0.261693452 5,- 0.239177259 1 O,0,1.48693 66434,1.205 982271,- 0.243844578 5 C,0,- 1.672365743 5,- 0.027729563 8,0.0684029 918 O,0,- 1.911543386 2,- 1.140325518 7,- 0.101479072 8 O,0,- 1.460913075 8,1.0961900 654,0.24553 05054 H,0,0.65430 41902,1.616 6283116,0.0 286598833 H,0,2.29021 78716,- 1.838963724 1,- 0.435185292 1 H,0,2.50770 63096,- 0.731575214	C,0,-1.9136499046,- 1.2889220091,0.0434040673 O,0,-0.7884085292,-0.9502256698,- 0.1923419416 O,0,- 1.7958978164,1.602610716,0.0063607139 C,0,1.6847928571,0.1520348181,0.024231 3927 O,0,2.1763938999,- 0.8620026965,0.2528363251 O,0,1.2220663057,1.1887000368,- 0.2018108232 H,0,-0.8335671716,1.4705206777,- 0.0600651658 H,0,-2.3247115401,-2.2243122859,- 0.366464402 H,0,-2.5763443219,- 0.6788132984,0.6745011392
--	--	---	---	--	--	---

	0.43257270 8,0.4383049 48				2,1.0914506 54	
(c) CH <sub>3</sub> + O <sub>2</sub> (+H <sub>2</sub> O) ==> CH <sub>2</sub> O + HO + (H <sub>2</sub> O)						
RC-h	TS1-h	Int-1h	TS2-h	Int-2h	TS3-h	PC-h
Charge = 0 Multiplicity = 2 C,0,- 2.963651346 4,- 0.383513479 6,0.0666842 607 H,0,- 2.844620190 1,- 0.008445291 4,1.0703247 H,0,- 3.040900352 8,- 1.444490263 4,- 0.106457842 7 H,0,- 3.071917365 9,0.3059045 698,- 0.754674328 4 O,0,- 0.212102531 2,- 0.083067885 6,- 0.216687471 7 O,0,0.07095 24791,1.028 5247424,0.0 974197551 O,0,2.85630 03458,- 0.437687968 7,0.0193938 134 H,0,2.49225 48209,0.445	Charge = 0 Multiplic y = 2 Redundant internal coordinate s found in file. C,0,- 2.15840728 57,1.089089 2868,0.1434 12266 H,0,- 2.98357888 12,0.556833 7124,0.5868 219063 H,0,- 2.32102532 46,1.588798 0037,- 0.79696258 29 H,0,- 1.32628971 22,1.377955 3428,0.7633 617593 O,0,- 1.13819598 71,- 0.77121732 85,- 0.62317454 41 O,0,- 0.88234905 35,- 1.56011364 09,0.249469 1609 O,0,- 3.73517712 1,- 0.21768732	Charge = 0 Multiplicity = 2 C,0,1.98978 9814,0.7875 366116,- 0.081157475 3 H,0,3.05043 02885,0.962 8979976,- 0.234411247 H,0,1.62692 5035,1.2549 502159,0.83 18943371 H,0,1.39445 98435,1.123 3628274,- 0.927674082 O,0,1.84997 40534,- 0.640320379 7,0.0440870 018 O,0,0.62115 65482,- 0.987321383 2,0.2324612 046 O,0,- 0.853909349 8,1.5141204 285,0.23735 04368 H,0,- 0.580785200 7,0.5888125 413,0.27181 50823 H,0,- 1.803676031	Charge = 0 Multiplicity = 2 C,0,0.86839 96768,0.916 0541372,- 0.297405945 7 H,0,0.82705 86272,- 0.117582258 6,- 1.122886798 7 H,0,1.84964 26871,1.375 0771343,- 0.383394712 3 H,0,- 0.036975971 6,1.4852749 183,- 0.491194025 2 O,0,0.75360 02182,- 0.007413880 8,0.7243472 696 O,0,0.69072 88555,- 1.064120371 7,- 0.277075983 5 O,0,- 2.019627810 9,0.0784147 994,- 0.050269244 3	Charge = 0 Multiplicity = 2 C,0,- 1.204644425 2,1.3464885 854,0.42781 12487 O,0,- 1.383458305 2,0.4214636 796,- 0.536826062 2 O,0,- 1.370078089 3,- 0.870322759 2,0.0330891 711 H,0,- 0.408156635 ,- 1.039714900 8,0.0737981 063 H,0,- 1.295884361 8,2.3506376 926,0.04003 68336 H,0,- 1.517547498 3,1.0784058 796,1.42763 64445 O,0,1.34318 02731,- 0.415703404 5,0.0995154 707 H,0,1.62045 37091,- 0.291673324 7,-	Charge = 0 Multiplicity = 2 C,0,1.82315 43739,- 0.965038976 1,0.1093036 112 O,0,0.98866 89498,- 0.267409878 3,- 0.588526944 9 O,0,1.29414 23622,1.247 1427378,- 0.378473175 5 H,0,0.39201 3057,1.5225 281222,- 0.138081711 4 H,0,2.45226 74208,- 1.676558097 2,- 0.413478656 7 H,0,1.97055 69916,- 0.720362967 6,1.1543839 736 O,0,- 1.477056109 5,1.0348928 162,0.02463 92083 H,0,- 1.179911592	Charge = 0 Multiplicity = 2 C,0,-1.6122629331,-1.3436095392,- 0.084820079 O,0,-0.5210833955,-1.7446048934,- 0.3934505332 O,0,- 1.2972870327,1.981907949,0.4491861397 H,0,- 0.4053931507,1.5682205643,0.3794271931 H,0,-2.4877381609,-2.0091141187,- 0.1377457547 H,0,-1.7694562693,- 0.3081356201,0.2539383693 O,0,1.0008741071,0.5056466964,0.182075 1427 H,0,1.7096625242,0.7367278558,- 0.4195391864 H,0,0.6725663109,-0.3650698941,- 0.0952382916

4042496,0.1 03558306	07,- 2.16040097 53	9,1.5133581 406,0.36477 67418	H,0,- 1.459292747 9,- 0.641686663 9,- 0.355607889 2 H,0,- 1.720179534 4,0.2119361 858,0.85227 53293	0.811321376 8 H,0,1.05973 63328,0.458 7305521,0.3 947391642	6,0.1697894 282,- 0.276300924 4 H,0,- 2.046191453 1,1.3680348 147,- 0.671712380 2	
H,0,2.09732 71408,- 1.000170673 ,- 0.144596192 5	H,0,- 2.94279206 92,- 0.67084493 02,- 1.86144981 44 H,0,- 4.06674756 57,- 0.74149612 54,- 2.89163417 57					

**Table S2.** Rotational -vibrational parameters obtained from vibrational analysis.

	Int-1	Int-2	TS2	TS3	PC	RC	TS1
	146.53	213.15	-1852.31	-540.95	159.5	68.69	-335.06
	523.15	323.73	273.88	253.54	216.51	131.16	78.2
	978.2	519.98	777.59	432.04	221.23	141.98	225.5
	1151.17	706.47	913.57	481.86	443.92	227	486.76
	1204.56	962.11	1046.05	638.91	595.72	327.61	506.26
	1311.29	1190.62	1114.31	1172.8	1230.77	517.35	797.08
	1462.75	1233.29	1154.6	1196.02	1289.01	1417.06	1421.23
	1474.73	1423.45	1180.84	1296.6	1539.28	1421.94	1422.54
	1485.2	1456.33	1467.94	1490.13	1851.53	1765.77	1586.67
	3050.05	3151.17	2017.66	3116.27	2995.26	3120.34	3122.15
	3160.59	3294.63	3078.21	3253.22	3087.37	3302.3	3297.67
	3161.14	3818.26	3203.36	3852.9	3671.35	3309.78	3301.61
D1	53.65327	54.40369	36.00364	52.97019	40.59308	38.69067	48.3499
2D1	11.60987	11.54995	14.84479	10.85359	5.47897	5.39689	7.11037
2D2	10.16483	9.91932	11.349	9.47676	4.8274	4.89403	6.47351
2D	10.86335	10.70363	12.97974	10.14184	5.142877	5.139313	6.784471
with CO2							
RC-c	Int-1- c	Int-2- c	TS1-c	TS2-c	TS3-c	PC-c	
27.95	12.79	75.03	- 291.63	- 1844.6 2	- 627.36	25.51	
52.82	83.03	110.7	15.11	72.36	42.88	45.08	
71.61	107.98	146.48	66.4	117.1	64.36	94.2	
99.95	125.31	173.98	78.29	142.98	97.94	108.99	
111.57	134.58	199.5	89.1	168.93	133.03	121.98	
127.25	151.76	315.53	111.47	176	155.77	128.46	
150.06	526.43	399.25	119.84	316.72	365.34	172.15	
179.66	687.58	525.46	232.53	696.08	442.71	175.47	
202.4	702.13	691.09	427.18	698.43	478.82	364.7	
250.59	975.5	696.02	475.84	777.55	687.11	411.41	

531.08	1154.2 3	701.67	694.95	910.21	694.92	698.69	
	686.37	1200.5 5	970.51	702.63	1042.8 5	700.58	703.03
	697.88	1322.5 1	1196.7	794.26	1113.6 7	1186.3 1	1229.3 4
	1414.5 5	1417.0 3	1236.3 2	1416.8 2	1147.1 1	1207.2 7	1289.9 2
	1415.7 7	1465.2	1413.9 9	1417.8 4	1187.7 7	1314.9 5	1418.5 1
	1426.2 5	1472.4 2	1455.7 8	1424.4 3	1416.4 8	1414.7 6	1546.0 3
	1767.5 6	1475.7 7	1462.8	1589.5 4	1470.1 3	1497.8 8	1861.8 8
	2458.2 1	2465.0 6	2461.4 6	2461.4 9	2023.0 3	2459.9	2467.1 6
	3114.8 5	3055.3 5	3156.5 2	3126.3	2464.5 9	3134.3 6	2989.4 2
	3292.3 7	3160.2 7	3301.5 1	3297.3 5	3086.3 7	3272.1 5	3079.7 6
	3312.7 9	3176.6 7	3788.2 2	3318.4 5	3216.3 1	3814.9 1	3752.0 9
D1	3.7352 7	6.6719 5	5.7989 8	5.1627 8	6.3406 8	5.7516 5	4.1924 5
2D1	2.1894 1	1.9743	2.4165 4	1.9017 9	2.4227 3	2.0911 9	1.9853 6
2D2	1.8719	1.5383 5	1.8835 1	1.4514	2.0182 9	1.5780 1	1.3660 3
2D	2.0244 4	1.7427 46	2.1334 43	1.6614 02	2.2112 83	1.8165 68	1.6468 34
with H2O							
RC-h	Int-1-h	Int-2-h	TS1-h	TS2-h	TS3-h	PC-h	TS1-ah
33.92	114.43	124.97	-227.12	-1839.1	-691.5	56.71	-278.56
59.76	132.05	162.9	48.36	156.55	31.87	75.33	63.95
73.21	166.51	205.83	66.06	198.82	138.93	170.76	97.58
109.86	178.3	249.08	88.52	216.08	191.97	224.59	120.29
134.85	203.37	303.05	146.96	220.5	220.82	248.06	140.16
164.89	382.37	346.5	178.02	342.15	247.57	249.89	182.43
175.12	489.12	507.03	199.25	359.32	386.1	262.18	218.12
184.83	536.05	535.51	240.38	501.38	459.18	412.52	234.3
226.4	965.85	737.68	270.79	775.2	504.84	508.03	283.32
258.87	1139.77	762.49	412.13	909.39	601.01	638.97	481.41
268.31	1206.89	969.48	480.23	1051.29	722.12	807.99	499.02
504.29	1337.45	1201.04	768.86	1120.01	1183.16	1245.2	807.96
1405.35	1462.73	1256.09	1416.21	1143.6	1233.84	1292.85	1417.94
1416.58	1474.34	1457.36	1421.91	1192.91	1361.56	1552.15	1421.39
1634.52	1480.23	1532.87	1598.78	1460.92	1499.37	1641.2	1583.59
1748.11	1636.03	1631.89	1644.11	1655.85	1623.61	1844.98	1626.49
3113.78	3052.37	3141.51	3119.56	2017.39	3105.54	2993.4	3130.46
3293.15	3163.39	3283.58	3300.78	3094.13	3241.9	3099.79	3306.55
3306.85	3172.02	3579.23	3312.63	3228.06	3690.38	3481.74	3316.6
3899.57	3787.86	3817.19	3896.65	3869.68	3859.92	3682.82	3885.76
3997.12	3961.99	3948.9	3988.122	3955.5	3971.45	3966.72	3985.35
18.10487	11.34254	10.20971	17.97533	11.00459	13.22988	6.98875	7.81707
1.82558	4.08591	4.60744	2.1712	4.45225	3.54628	3.32137	3.27206

1.68733	3.06194	3.57528	1.96377	3.79697	3.00099	2.26552	2.43621
1.755094	3.537063	4.058681	2.064882	4.111576	3.262262	2.743106	2.823371

**Table S3.** Calculated equilibrium constants ( $K_{eq}$  in  $\text{cm}^3 \text{ molecule}^{-1}$ ) for the formation of two-body and three body complexes.

Temp	RC	Ke-1-c	Ke-2-c	Ke-1-h	Ke-2h
500	8.48E-16	9.11E-25	2.85E-27	2.07E-26	7.11E-23
600	5.09E-18	9.88E-25	3.02E-27	2.33E-26	5.29E-23
700	1.34E-19	1.11E-24	3.37E-27	2.68E-26	4.59E-23
800	8.95E-21	1.27E-24	3.86E-27	3.10E-26	4.35E-23
900	1.11E-21	1.46E-24	4.47E-27	3.60E-26	4.35E-23
1000	2.13E-22	1.68E-24	5.19E-27	4.16E-26	4.50E-23
1100	5.60E-23	1.92E-24	6.03E-27	4.79E-26	4.75E-23
1200	1.87E-23	2.19E-24	6.98E-27	5.49E-26	5.09E-23
1300	7.48E-24	2.49E-24	8.06E-27	6.25E-26	5.49E-23
1400	3.45E-24	2.82E-24	9.26E-27	7.09E-26	5.97E-23
1500	1.78E-24	3.18E-24	1.06E-26	8.00E-26	6.50E-23

### Details of chemical kinetics analysis

For pressure-dependent reactions, Rice–Ramsperger–Kassel–Marcus (RRKM)/master equation (ME) was used and the energy-dependent specific unimolecular rate constants  $k_i(E, J)$  is given by [39]

$$k(E) = L^\ddagger \frac{1}{h} \frac{G^\ddagger(E - E_0)}{\rho(E)} \quad (\text{S1})$$

where  $L^\ddagger = \frac{m^\ddagger}{m} \times \frac{\sigma_{ext}}{\sigma_{ext}^\ddagger}$  is the reaction path degeneracy;  $\sigma_{ext}^\ddagger$  and  $\sigma_{ext}$  are the ex-

ternal rotation symmetry numbers of the transition state and reactant;  $m^\ddagger$  and  $m$  are the numbers of optical isomers (mirror image) of the transition state and reactant, respectively,  $h$  is Planck's constant;  $G^\ddagger(E - E_0)$  is the sum-of-states of the transition state;  $E_0$  is the reaction threshold energy, and  $\rho(E)$  is the density of states of the reactant molecule. The sums and densities of states were computed using DenSum as implemented in MultiWell Program. The internal energy  $E$  is measured relative to the zero point energy of the reactant molecule and the reaction threshold energy (critical energy) is the difference between the zero point energies of reactant and transition state. The pressure-dependent rate constants were simulated using estimated energy transfer parameters for the intermediates. The tunnelling corrections were implemented based on asymmetric Eckart potential. The approximated temperature-dependent exponential down  $\langle \Delta E \rangle_{\text{down}} = 200 \cdot (T/300)^{0.85} \text{ cm}^{-1}$  [28,32] was used to calculate the rate constants. The Lennard-Jones parameters for collider gases ( $\text{N}_2$ )  $\epsilon/\text{kJ}$ ,  $\sigma(\text{N}_2) = 3.74 \text{ \AA}$ ,  $\epsilon/\text{kJ}(\text{N}_2) = 82 \text{ K}$  were used in our calculation based on previous studies.[28,32] The bimolecular rate constant describing collisions between  $\text{N}_2$  and the complex, intermediate ("Well") was based on Lennard-Jones collisions with net parameters obtained using the usual combining rules from parameters for the two collision partners:  $\text{N}_2$  gas ( $\sigma = 3.74 \text{ \AA}$  and  $\epsilon/\text{kJ} = 82 \text{ K}$ ) [52] and the Well ( $\sigma = 5.20 \text{ \AA}$  and  $\epsilon/\text{kJ} = 212 \text{ K}$ , used for all Wells). The double arrays used in MultiWell for the master equation simulations consisted of 5000 array elements with  $10 \text{ cm}^{-1}$  energy grains, with the quasi-continuum regime evaluated up to  $50000 \text{ cm}^{-1}$ . At each temperature and pressure, master equation simulations were initiated using the chemical activation energy distribution, which is appropriate for recombination reactions.

---

The calculated rate constants in the high-pressure limit ( $k_{\infty}$ ) were fit to the modified Arrhenius expression  $k_{\infty}^{bimol}(T) = A \times T^n \left( \exp - \left( \frac{E_a}{RT} \right) \right)$  in the temperature range of 500 – 1500 K.