

# Platinum-Containing Nanoparticles on N-Doped Carbon Supports as an Advanced Electrocatalyst for the Oxygen Reduction Reaction

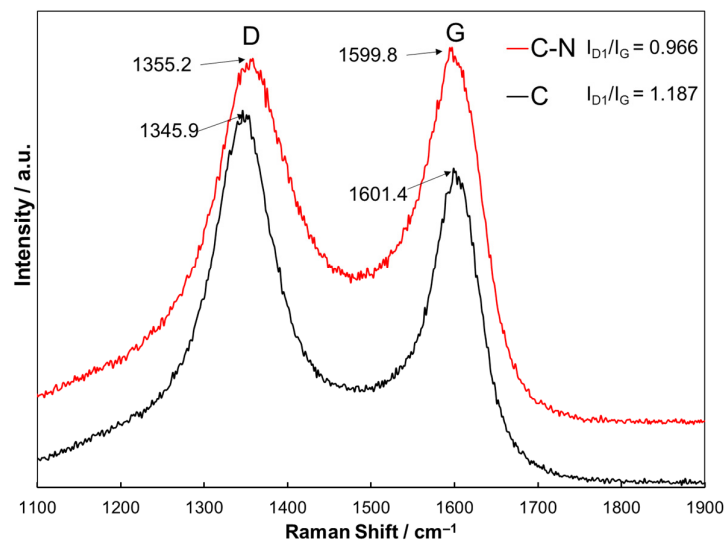
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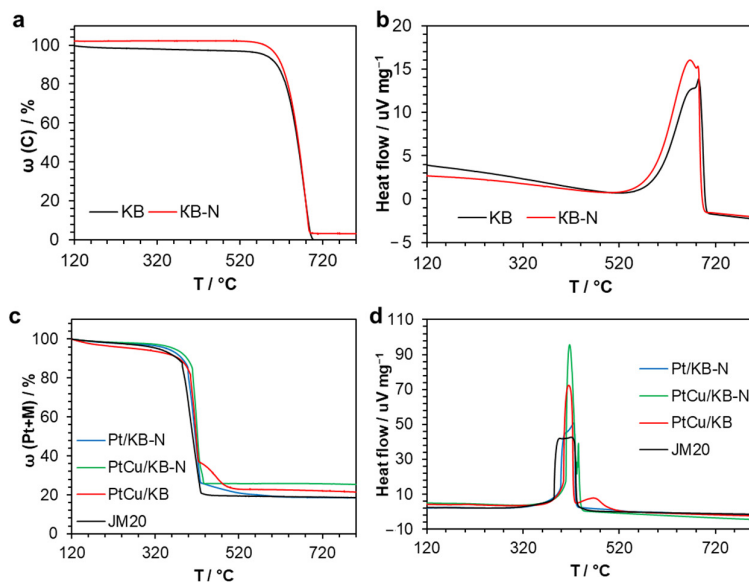
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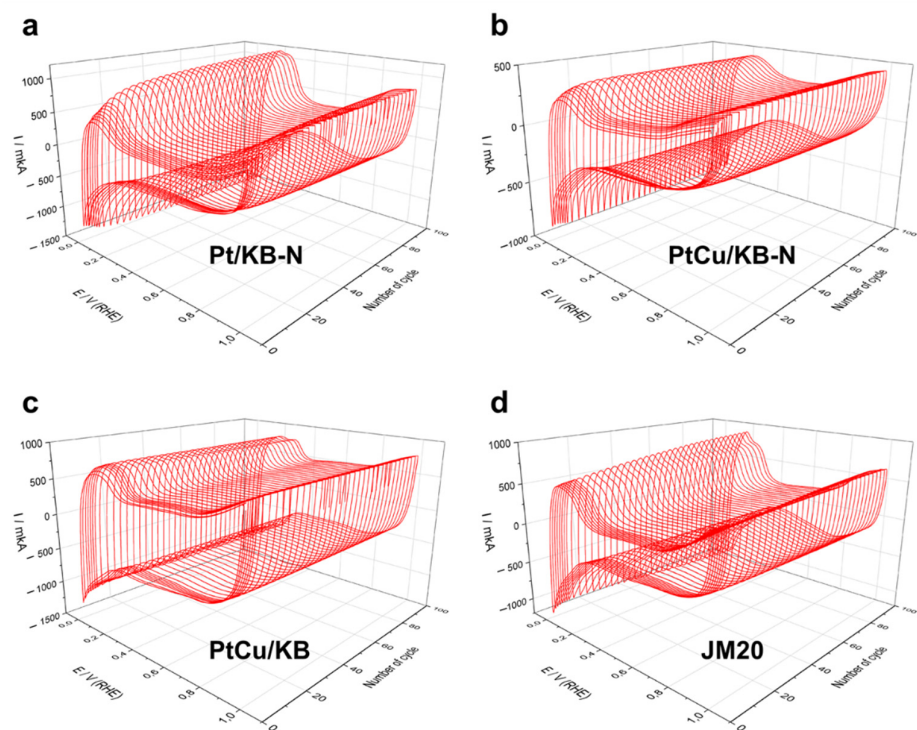
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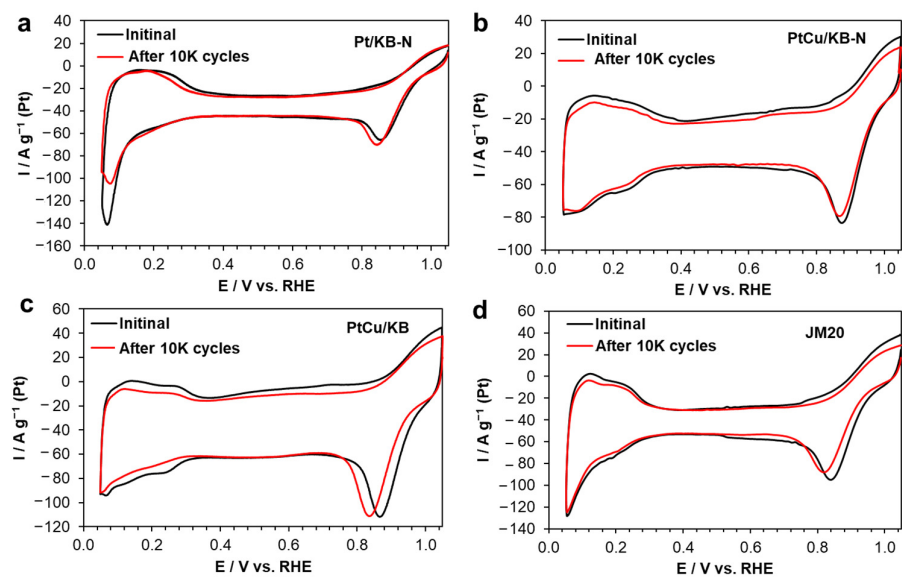
**Figure S1.** Raman spectra of undoped and N-doped carbons materials.



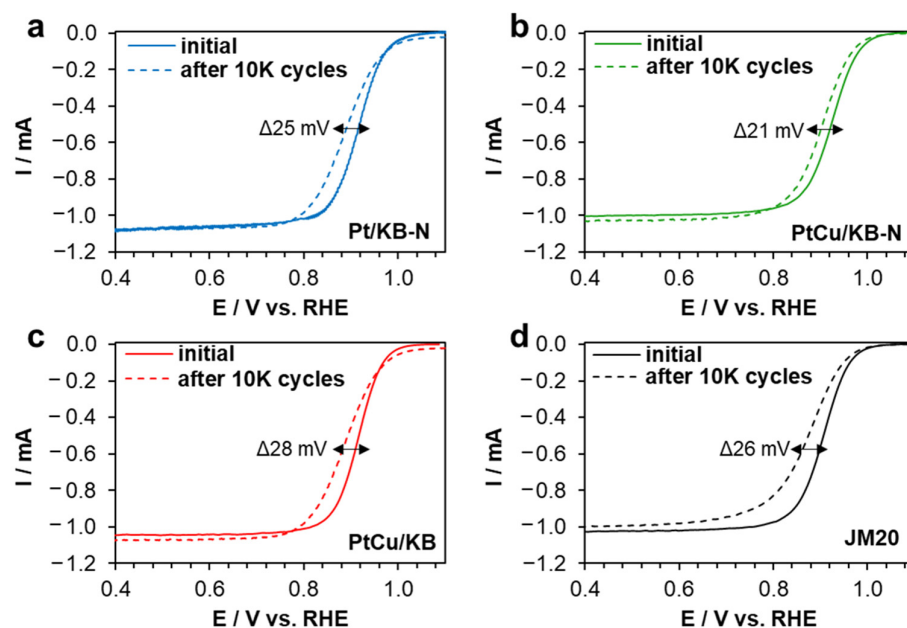
**Figure S2.** (a, c) TGA and (b, d) DSC curves of carbon support (a, b) and catalysts (c, d) oxidation.



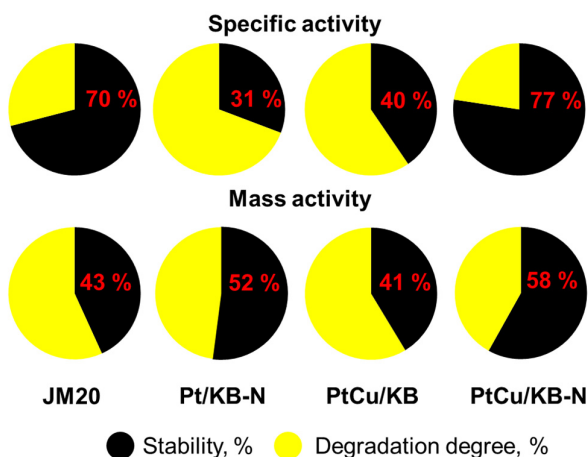
**Figure S3.** CVs at the stage of catalysts electrochemical activation: Pt/KB-N(a), PtCu/KB-N(b), PtCu/KB(c), JM20(d). The sweep rate of the potential is  $100 \text{ mV s}^{-1}$ .



**Figure S4.** CVs: initial and after the stress test in oxygen atmosphere: Pt/KB-N(a), PtCu/KB-N(b), PtCu/KB(c), JM20(d). Electrolyte is  $0.1 \text{ M HClO}_4$ . Black - initial curves; Red - end curves.



**Figure S5.** LSV ORR curves before and after the stress test: Pt/KB-N(a), PtCu/KB-N(b), PtCu/KB(c), JM20(d). Rotation speed of RDE is 1600 rpm. Electrolyte is 0.1 M  $\text{HClO}_4$ .



**Figure S6.** Relative stability and degradation degree of such parameters: mass and specific activity of obtained and commercial samples. The red color indicates the calculated value of corresponding stability for each sample.