

# Biomimetic material for quantification of methotrexate using sensor based on molecularly imprinted polypyrrole film and MWCNT/GCE

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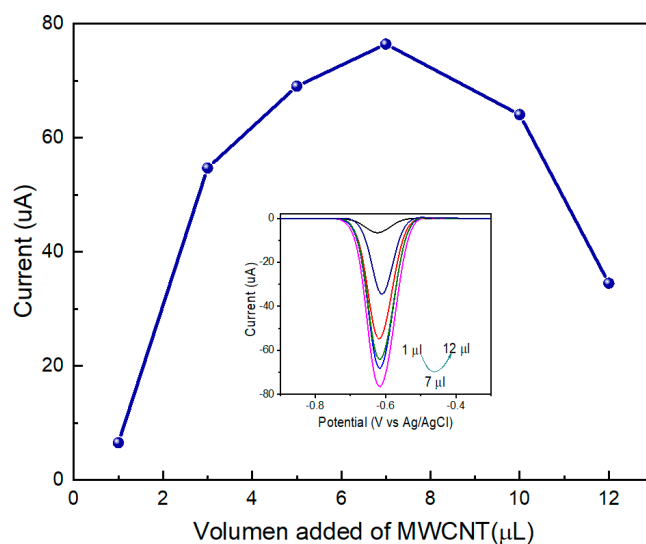


Figure S1. Deposited volume of 1mg/ml MWCNT.

Figure S2 shows the electrochemical profile of methotrexate in not modified electrodes. It can be observed that four peaks were found. The oxidation peak (1) could not be used in the detection, since it could oxidize the polypyrrole polymer that would affect the stability of the sensor. Peak (4) is another methotrexate oxidation peak; however, it only appears if methotrexate reduction has been performed. peak (3), therefore, remains as options to peaks (2) and (3), peak (2) is not stable due to the formation of intermediate species under this potential, for this reason it is not used for a detection method. Peak (3) is a reduction peak and is stable without affecting the sensor or the analyte, so this potential is used for the detection method.

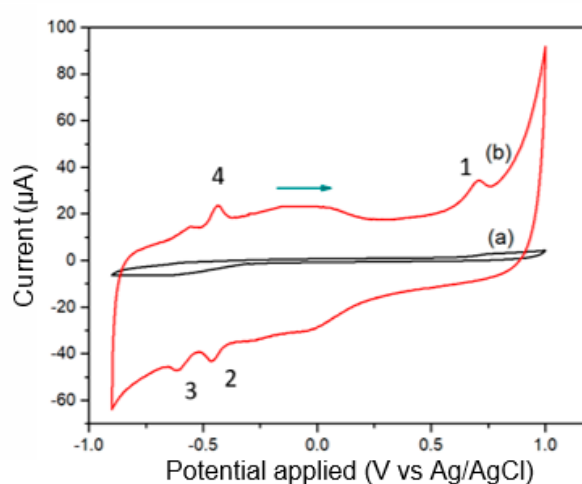
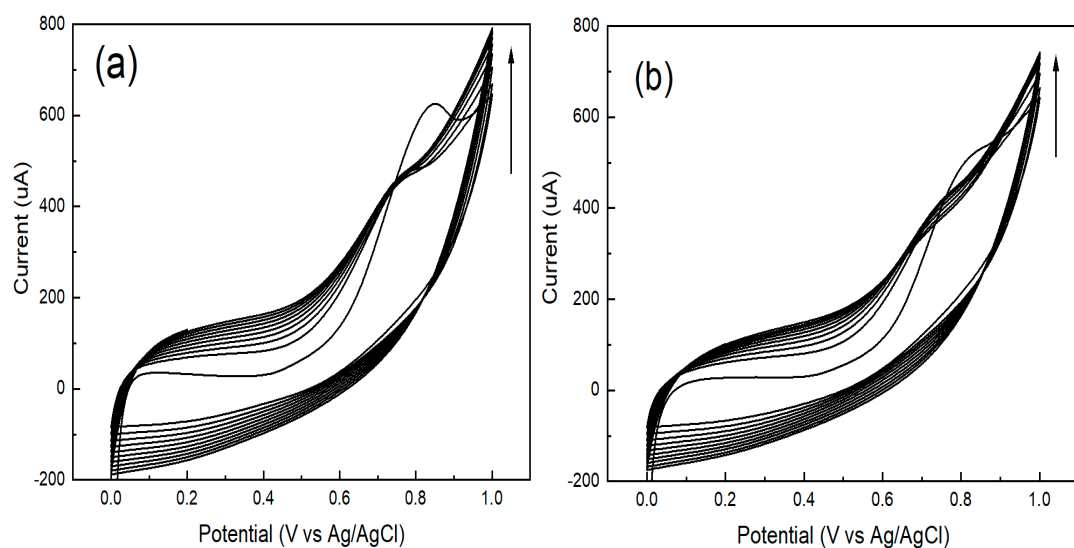


Figure S2. Cyclic voltammetry for (a) GCE and (b) MWCNT/GCE in the standard solution of methotrexate  $100 \mu\text{mol L}^{-1}$ ,  $v = 50 \text{ mVs}^{-1}$ .



**Figure S3.** Electropolymerization of 20 mmol L<sup>-1</sup> pyrrole solution, 0.1 mol L<sup>-1</sup> LiClO<sub>4</sub>, and 100 μmol L<sup>-1</sup> methotrexate on the surface of MWCNT/GCE: (a) MIP/MWCNT and (b) NIP/MWCNT.

In Figure S4, can be evidenced that the application of negative or positive potentials during the pre-treatment produces a better repeatability in the pyrrole oxidation, therefore this pre-treatment could favor the orderly growth of the polypyrrole. In other research, it is shown that polypyrrole has a characteristic morphology called "cauliflower", this morphology is due to polymeric growth in the  $\alpha$  and  $\beta$  position. Therefore, in the case of applying a potential as a pre-treatment, it can be expected that the morphology will change, since the nucleation is being oriented and a horizontal or vertical growth can occur depending on the applied potential. In this work, a potential of 0.4V was chosen for the pretreatment in order to obtain a vertical growth polymer, as shown in the SEM figure of this manuscript.

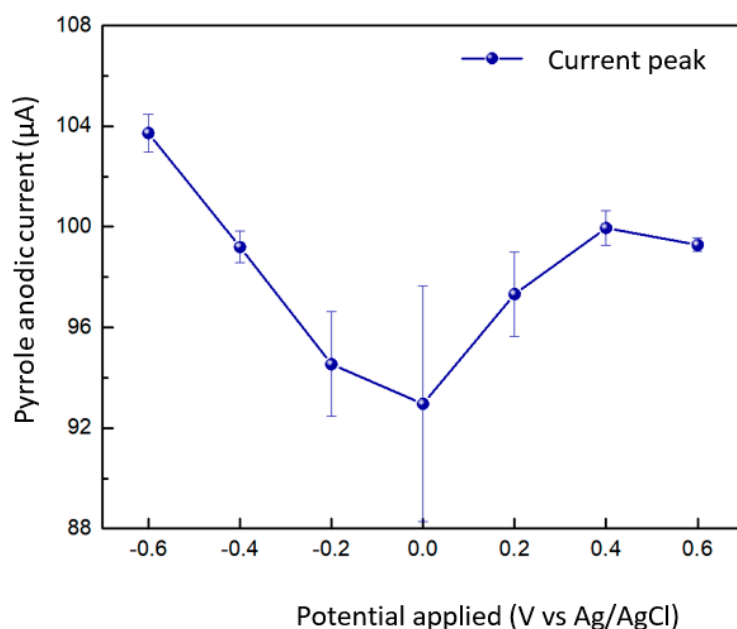
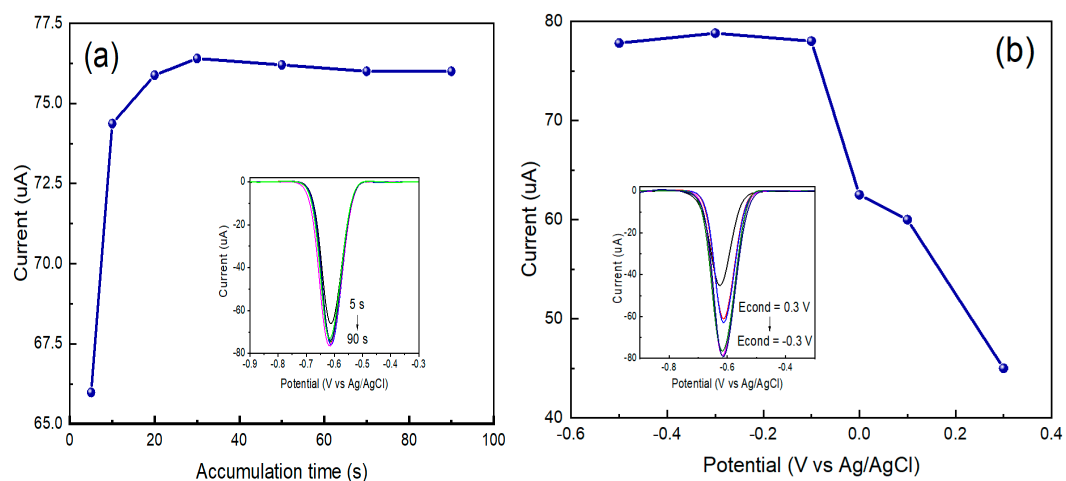
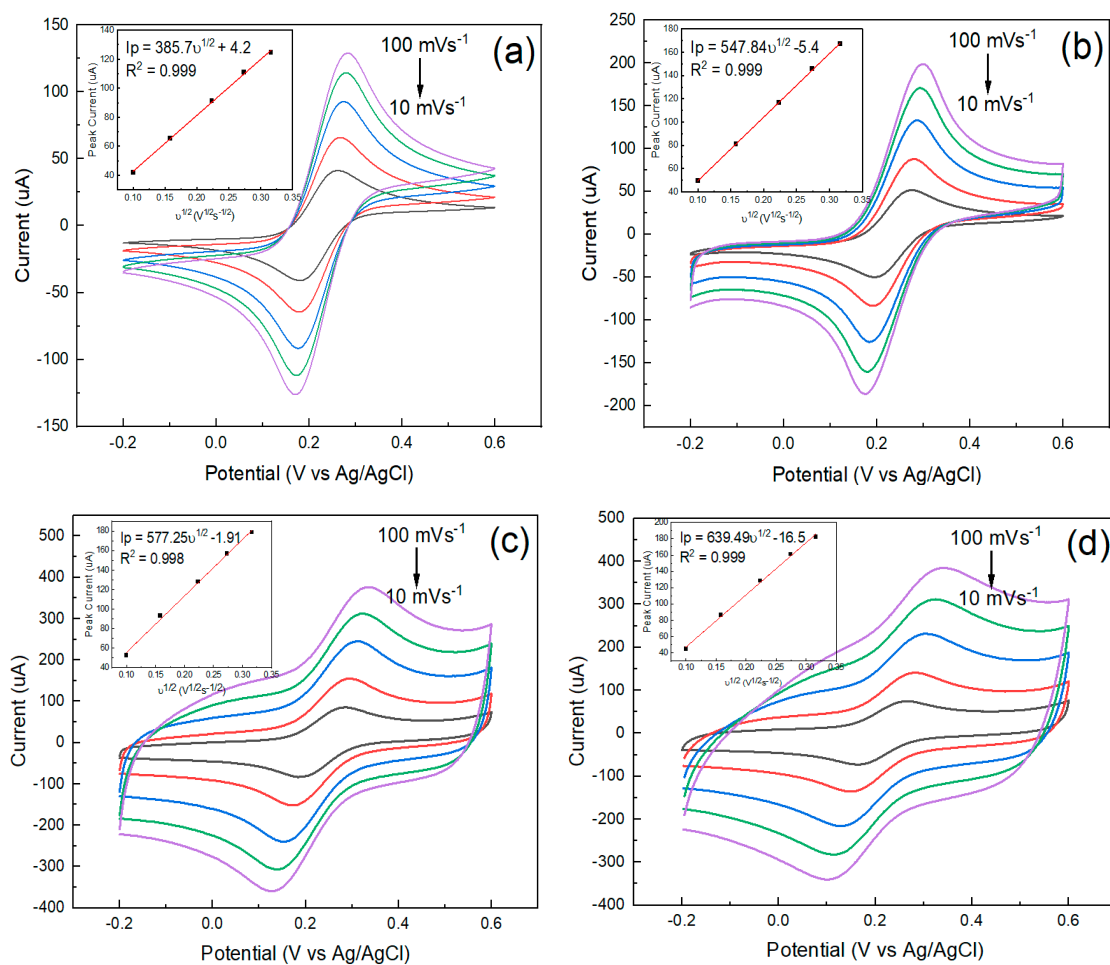


Figure S4. Influence of conditioning potential prior to the electropolymerization of pyrrole.



**Figure S5.** Conditioning parameters: (a) accumulation time and (b) applied accumulation potential.



**Figure S6.** Cyclic voltammograms obtained from the application of (a) GCE, (b) MWCNT/GCE, (c) NIP/MWCNT/GCE, and (d) MIP/MWCNT/GCE at the following scan rates: 10, 25, 50, 75, and 100 mVs<sup>-1</sup>.

Figure S7 shows like the pH affect the structure of the MTX. Through this mechanism it is possible to verify why the pH was adjusted up to pH 3 to carry out the detection of this drug and consequently to obtain the optimized conditions for the electrochemical detection.

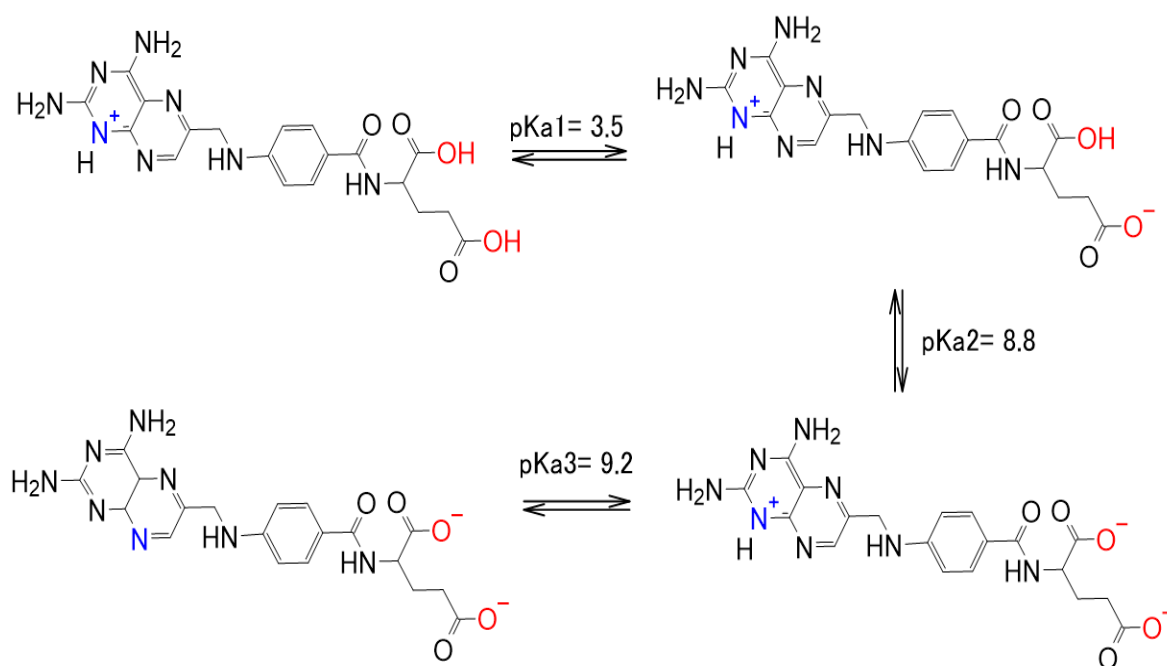


Figure S7. Mechanismo of the acid-base balance in an aqueous methotrexate solution.

Table S1. Summary of detection methods for methotrexate.

Method/sensor	Analyte*	Analytical measure	LOD (mol L <sup>-1</sup> )	Reference
<b>Graphene Oxide/Polypyrrole</b>	MTX	UV-visible	2.2 x 10 <sup>-11</sup>	[S1]
<b>QD CdTe – MIP</b>	MTX	Fluorescence	3.2 x 10 <sup>-8</sup>	[S2]
	FA		3.4 x 10 <sup>-8</sup>	
<b>HPLC- HRMS</b>	MTX	-----	4.5 x 10 <sup>-11</sup>	[S3]
<b>LC-MS/MS</b>	MTX	----	2.2 x 10 <sup>-11</sup>	[S4]
<b>Poly (l-Lysine)/ GCE</b>	MTX	SWV	1.7 x 10 <sup>-9</sup>	[S5]
<b>f-CNT / GCE</b>	MTX	DPV	5 x 10 <sup>-7</sup>	[S6]
<b>Boron-Doped Diamond Electrode</b>	MTX	DPV	1 x10 <sup>-8</sup>	[S7]
<b>MIP/MWCNT/GCE</b>	MTX	DPV	2.7 × 10 <sup>-9</sup>	This work

\*Analyte: MTX = methotrexate; QD CdTe – MIP: molecularly imprinted polymers on dual-color CdTe quantum dots; GCE: Glassy carbon electrode; FA: Folic acid; HPLC: high performance liquid chromatography; HRMS: high resolution mass spectrometry; SWV: Square wave voltammetry; f-CNT: functionalized carbon nanotube; DPV: differential pulse voltammetry.



## References:

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