

Supplementary Materials

Biocompatible MgFeCO_3 Layered Double Hydroxide (LDH) for Bone Regeneration—Low-Temperature Processing through Cold Sintering and Freeze-Casting

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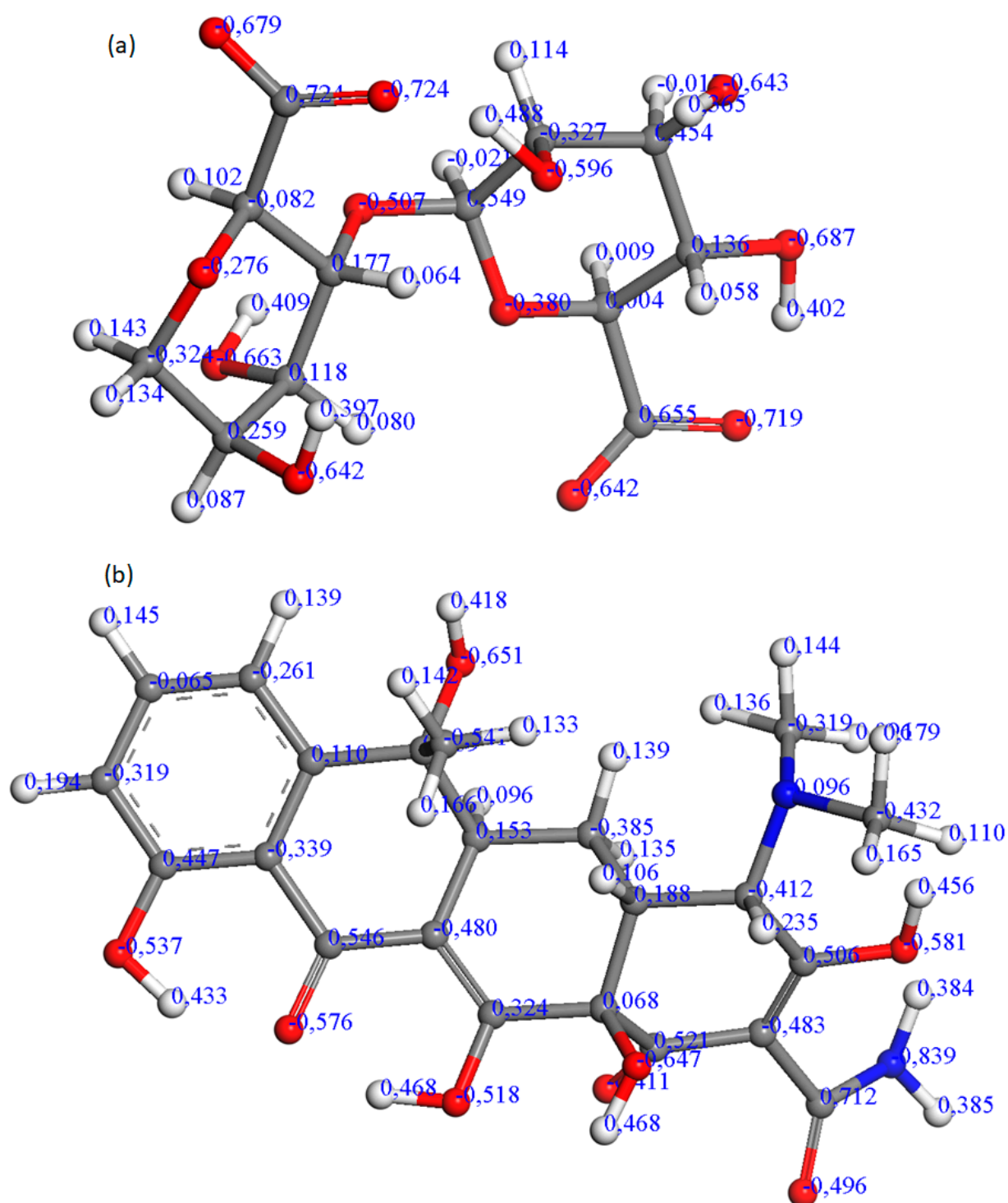


Figure S1. Partial charges obtained for tetracycline and alginate (charged 2-) from DFT calculations.

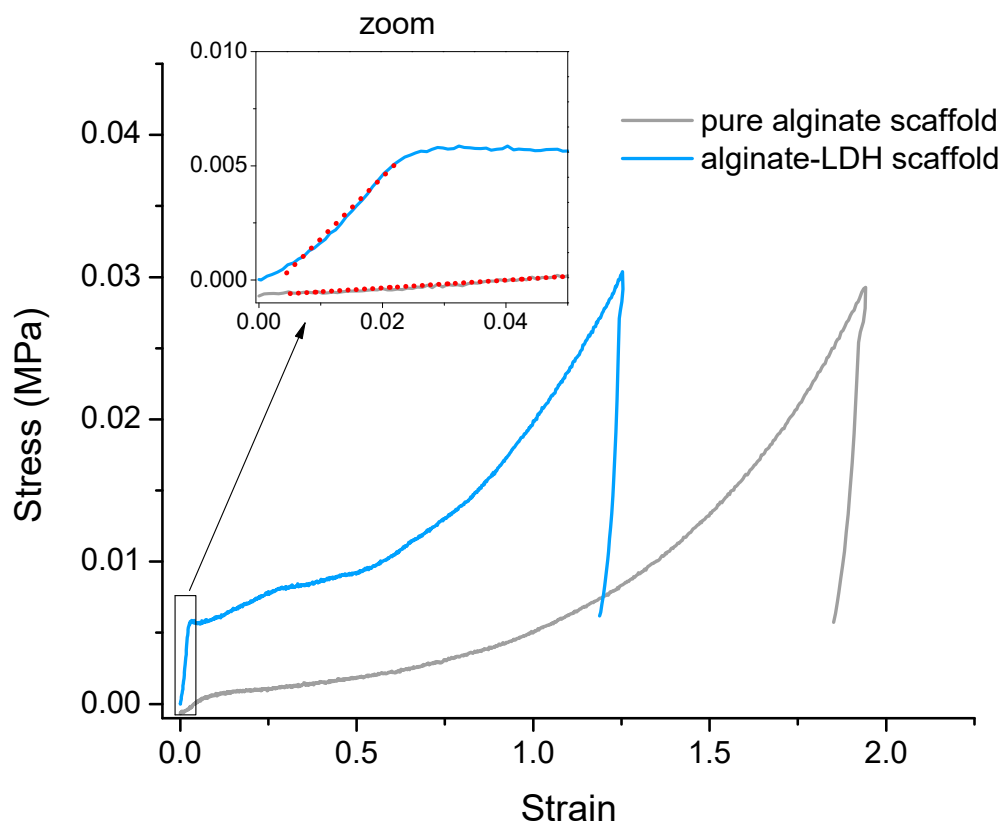


Figure S2. Evaluation of mechanical properties (stress-strain curves) through uniaxial compression for typical freeze-cast alginate/MgFeCO₃ LDH and pure alginate scaffolds. The zoomed inlet view allows better visualizing the elastic part of the curves to decipher the variation in Young modulus (slope of the red dotted lines).

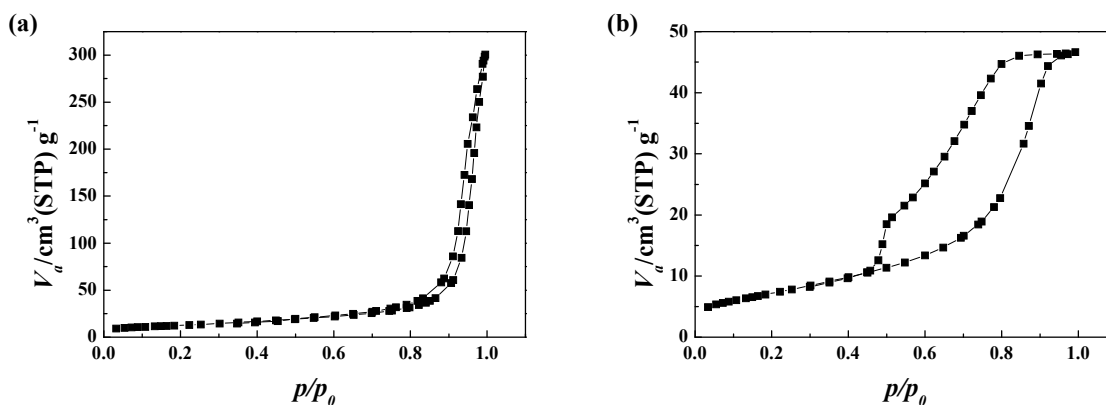
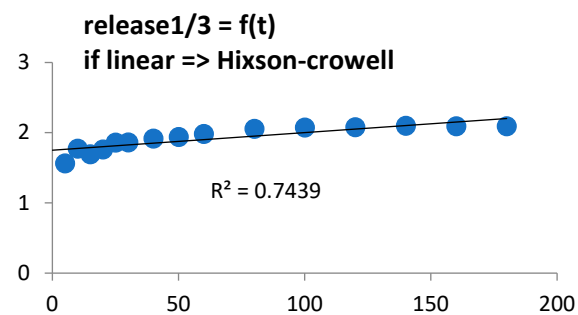
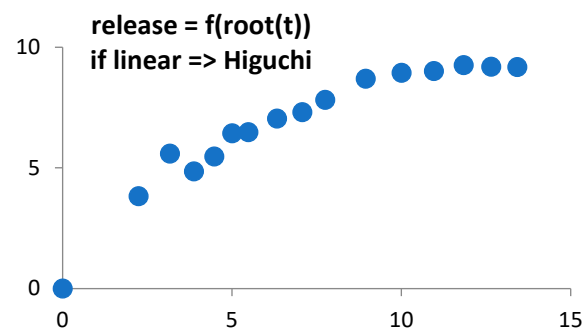
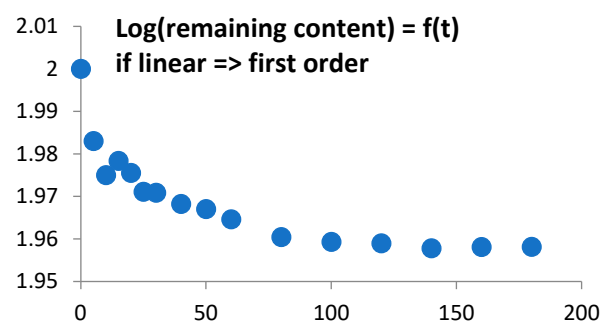
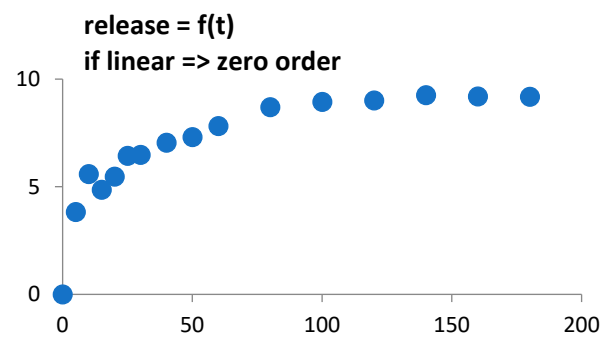


Figure S3. Evaluation of the porous volume by the BJH method, from processing nitrogen (N₂) adsorption/desorption data: (a) MgFeCO₃ LDH powder, (b) SPS cold-sintered monolith.



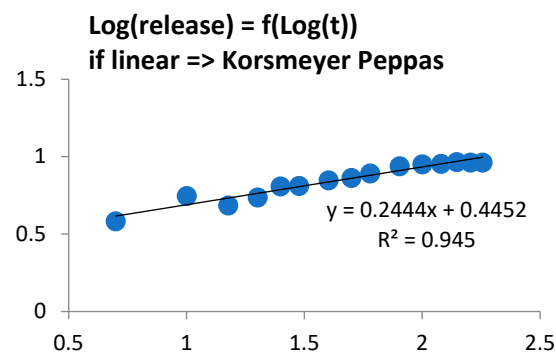


Figure S4. Mathematical fitting of MO release from MgFeCO₃ LDH monolith (from cold sintering) relatively to several models (t is given in minutes and the release rate in %).

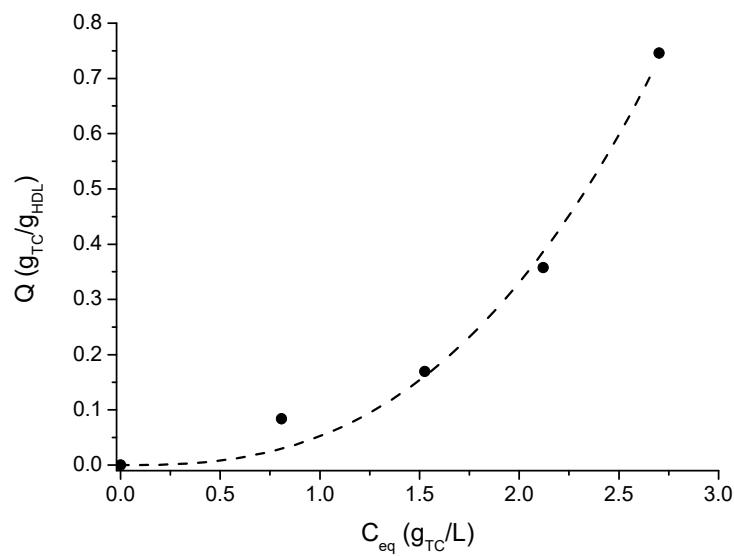


Figure S5. Isotherm of interaction between TC molecules and MgFeCO₃ LDH powder. The dotted line depicts the fit with Sips isotherm. Error bars are included but not visible (standard deviations correspond to 0.6% of the mean).

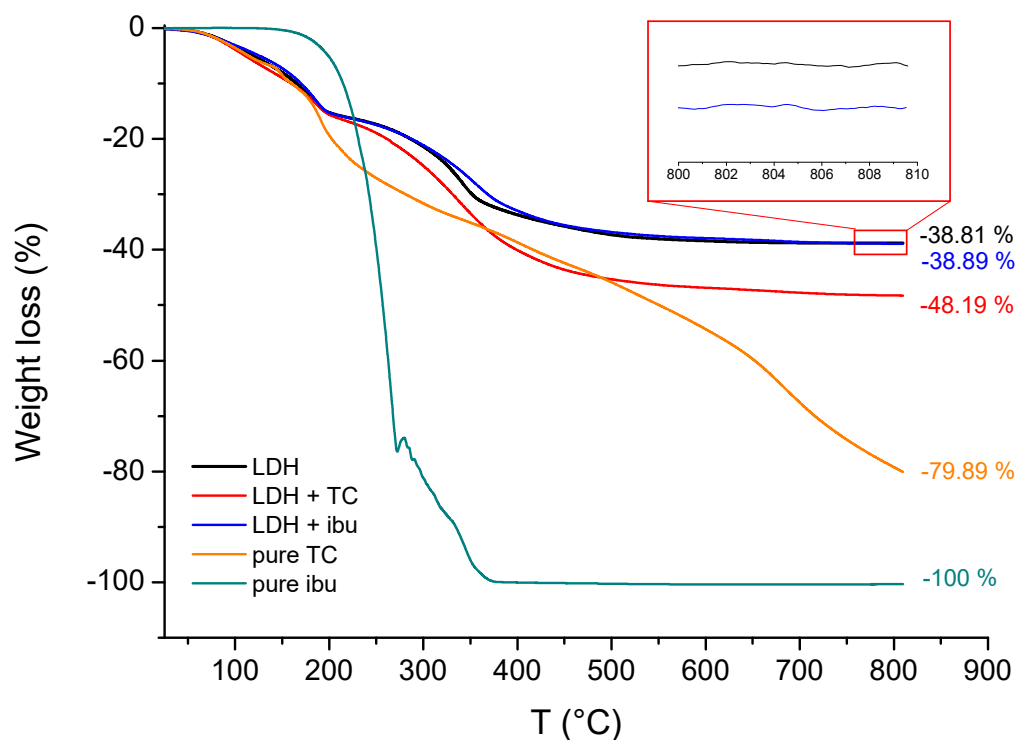


Figure S6. Thermogravimetry data on pure MgFeCO_3 LDH powder, tetracycline (TC), ibuprofene (ibu), and combined LDH-TC and LDH-ibu compounds.

Table S1. Correlation coefficient for the mathematical fit of the kinetics of adsorption of MO on MgFeCO_3 in powder or monoliths forms, using the pseudo-first order, pseudo-second order and Elovich kinetic models.

Sample	Kinetic model	R ²
MgFe-CO ₃ -LDH powder	Pseudo first order	0.9153
	Pseudo second order	0.9505
	Elovich	0.9689
MgFe-CO ₃ -LDH monolith	Pseudo first order	0.9742
	Pseudo second order	0.9895
	Elovich	0.9992