

## **Supporting Information**

### **Design of hierarchical NiCoP/NiO with tunable electronic structure and strong chemical interface for advanced supercapacitors**

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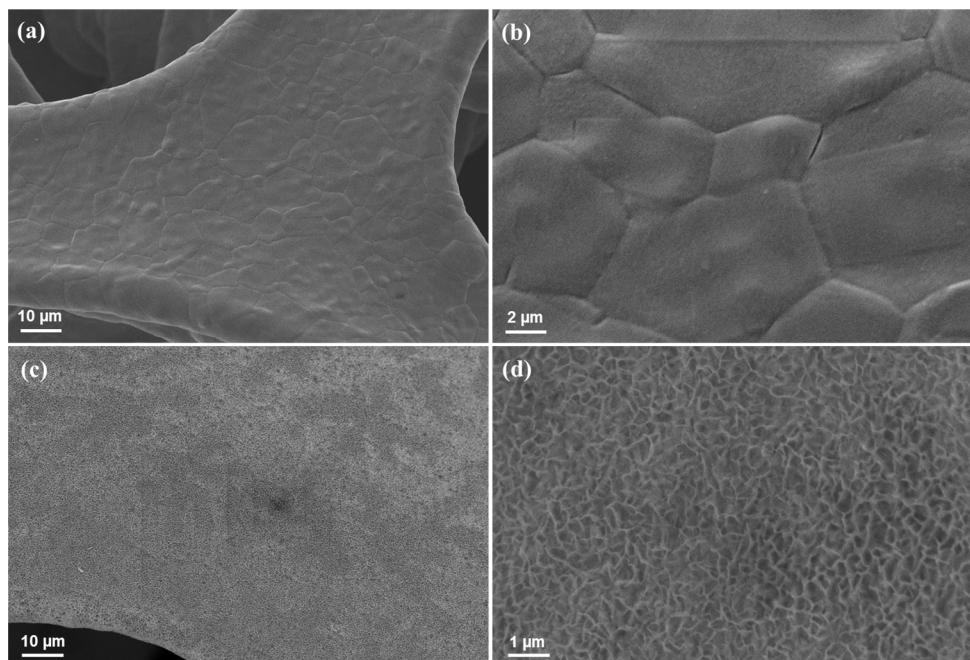
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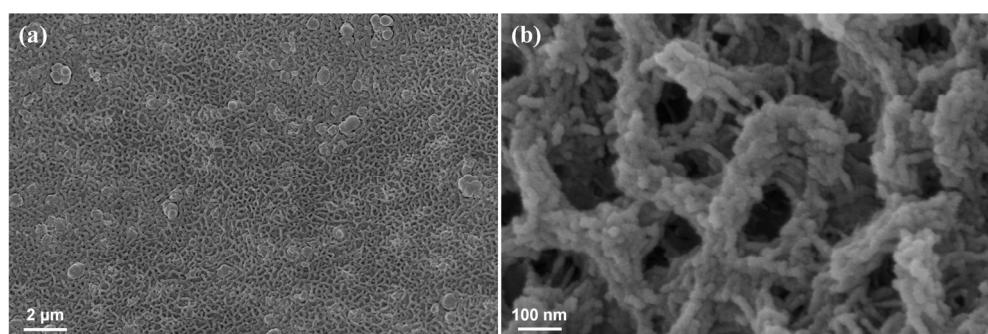
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## 1. Chemicals

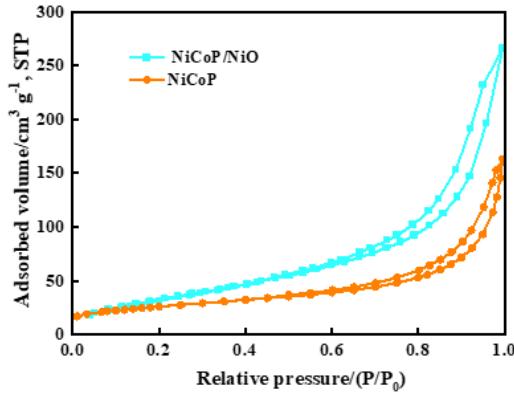
All the chemical reagents with analytical grade utilized directly without any treatment. Sodium hydroxide (NaOH), ammonium persulphate ( $(\text{NH}_4)_2\text{S}_2\text{O}_8$ ), cobaltous nitrate hexahydrate ( $\text{Co}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ ), nickel nitrate hexahydrate ( $\text{Ni}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ ) and sodium hypophosphite ( $\text{NaH}_2\text{PO}_2 \cdot \text{H}_2\text{O}$ ) were supplied by Sinopharm Chemical Reagents Co., Ltd. **Porous activated carbon (PAC, C196579)** purchased from Aladdin. Nickel foam (NF) were purchased from Suzhou Keshenghe Metal Materials Co., Ltd.



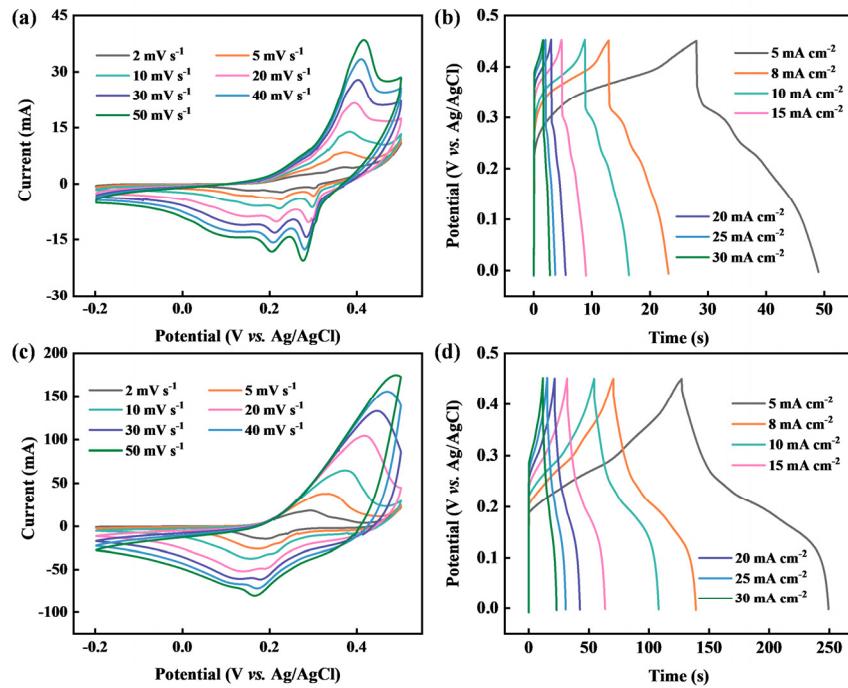
**Figure S1.** SEM images of (a, b) pure nickel foam and (c, d) NiO/NF at different magnifications.



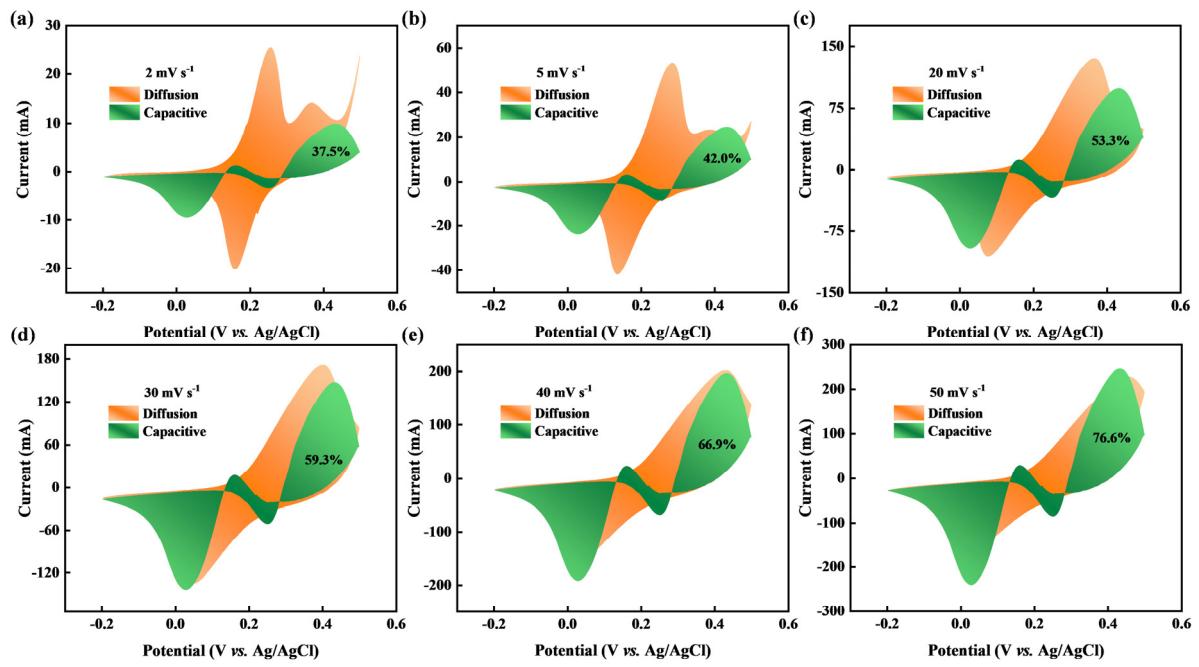
**Figure S2.** SEM images of NiCoP/NiO nanocomposites with **electrodeposition time of 15 min.**



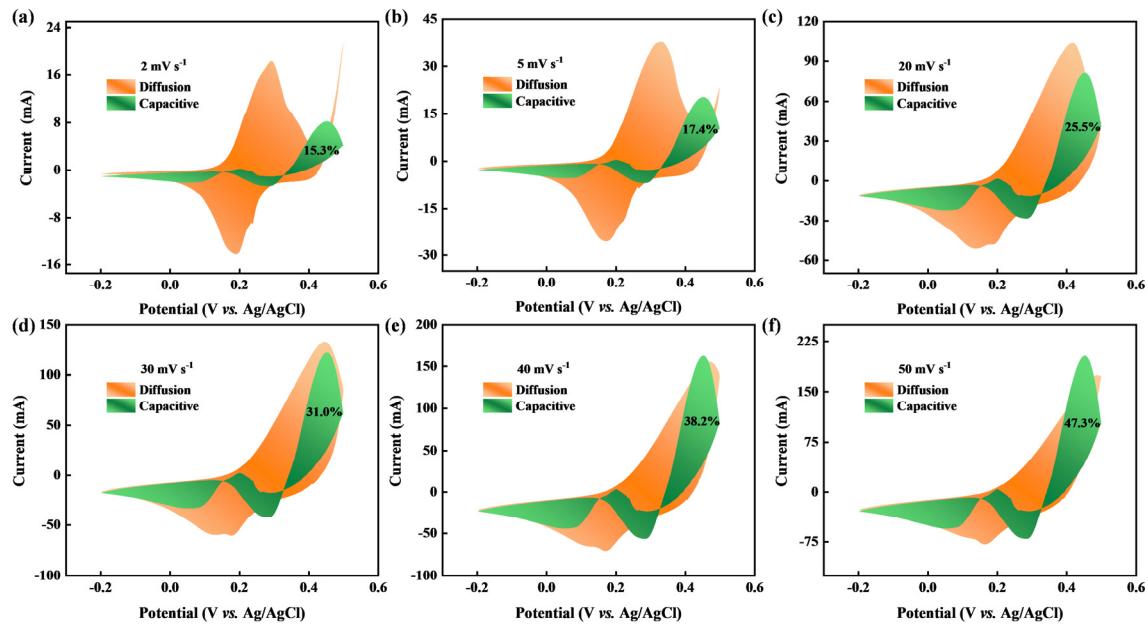
**Figure S3.** N<sub>2</sub> adsorption-desorption isotherms of the NiO/NiCoP and NiCoP.



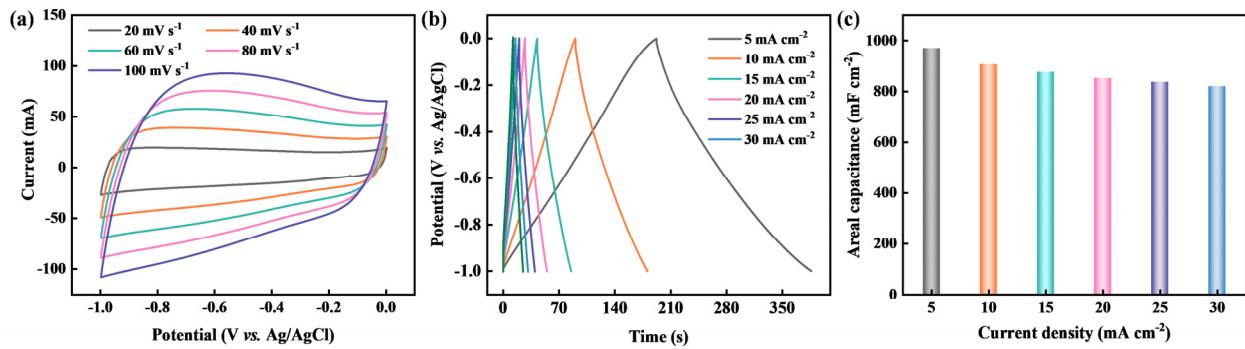
**Figure S4.** Electrochemical performance measured in a three-electrode system in 2.0 M KOH electrolyte: CV curves and GCD curves of the (a, b) NiO/NF and (c, d) NiCoP/NF electrodes, respectively.



**Figure S5.** Separation of the surface and diffusion-controlled currents at different scan rates for the NiCoP/NiO electrode.



**Figure S6.** Separation of the surface and diffusion-controlled currents at different scan rates for the NiCoP/NF electrode.



**Figure S7.** (a) CV curves, (b) GCD curves, and (c) areal specific capacitance of the PAC electrode.

**Table S1.** Comparative electrochemical performance of nickel-cobalt based nanocomposite materials reported by the literature.

Devices	Voltage (V)	Energy density ( $\mu\text{Wh}/\text{cm}^2$ )	Power density ( $\text{mW}/\text{cm}^2$ )	Cycling stability
NF/NiO/NiCo <sub>2</sub> S <sub>4</sub> //AC/NF <sup>[1]</sup>	1.2	25.4	0.782	90% (20 mA cm <sup>-2</sup> , 2000 C)
NiCo <sub>2</sub> S <sub>4</sub> @NiCoP/NF//AC <sup>[2]</sup>	1.5	135	0.755	/
NiCo <sub>2</sub> S <sub>4</sub> @CNT//Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> @CCT <sup>[3]</sup>	1.5	180	2	87% (/, 2000 C)
NiCo <sub>2</sub> O <sub>4</sub> NG@CF//PC@CF <sup>[4]</sup>	1.65	9.46	0.6084	92% (2 mA cm <sup>-2</sup> , 3000 C)
Co <sub>0.8</sub> Ni <sub>0.2</sub> Se <sub>2</sub> //AC <sup>[5]</sup>	1.8	71	0.75	48% (1 mA cm <sup>-2</sup> , 8000 C)
Ni <sub>a</sub> CobS@NF//AC <sup>[6]</sup>	1.2	1.25	6	88% (/, 6500 C)
NiCoP/NiO//PAC(This work)	1.6	173.7	1.6	74% (40 mA cm <sup>-2</sup> , 5000 C)

CNT: carbon nanotube film, CCT: carbon cloth threads, NG: nanograss, PC: porous carbon

## References

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