

**Supporting Information**

**SiO<sub>2</sub>-Alginate-Based Gel Polymer Electrolytes for Zinc-Ions**

**Batteries**

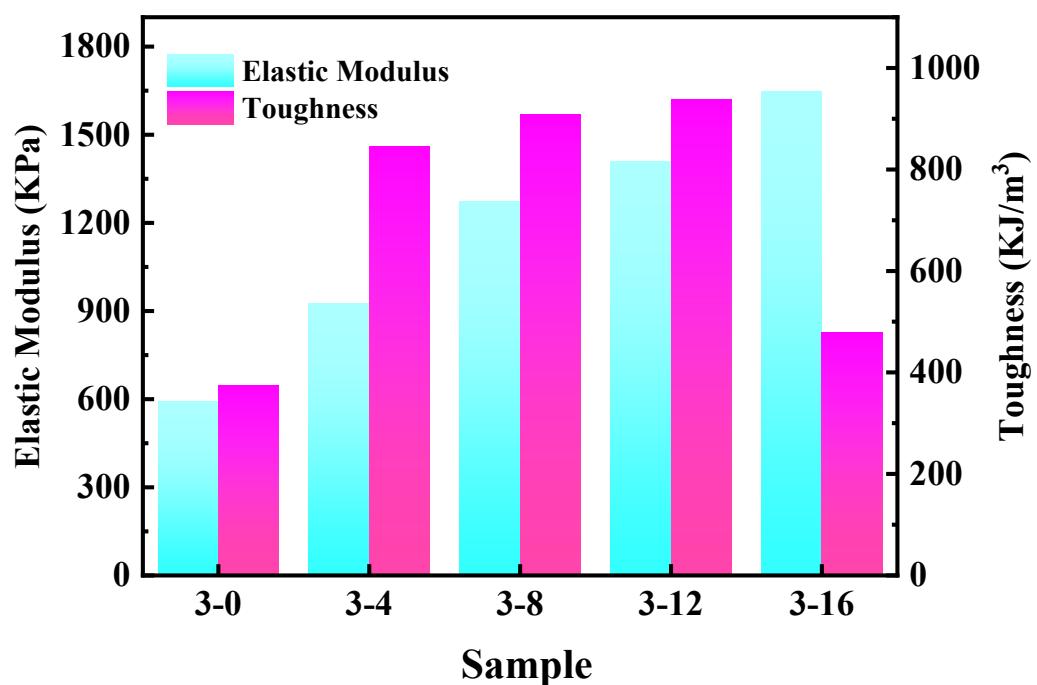
Peishu Tian<sup>1</sup>, Xin Zhong<sup>1</sup>, Caiting Gu<sup>1</sup>, Zhe Wang<sup>2</sup> and Fengwei Shi<sup>1, 2\*</sup>

1 School of Chemical Engineering, Changchun University of Technology, Changchun

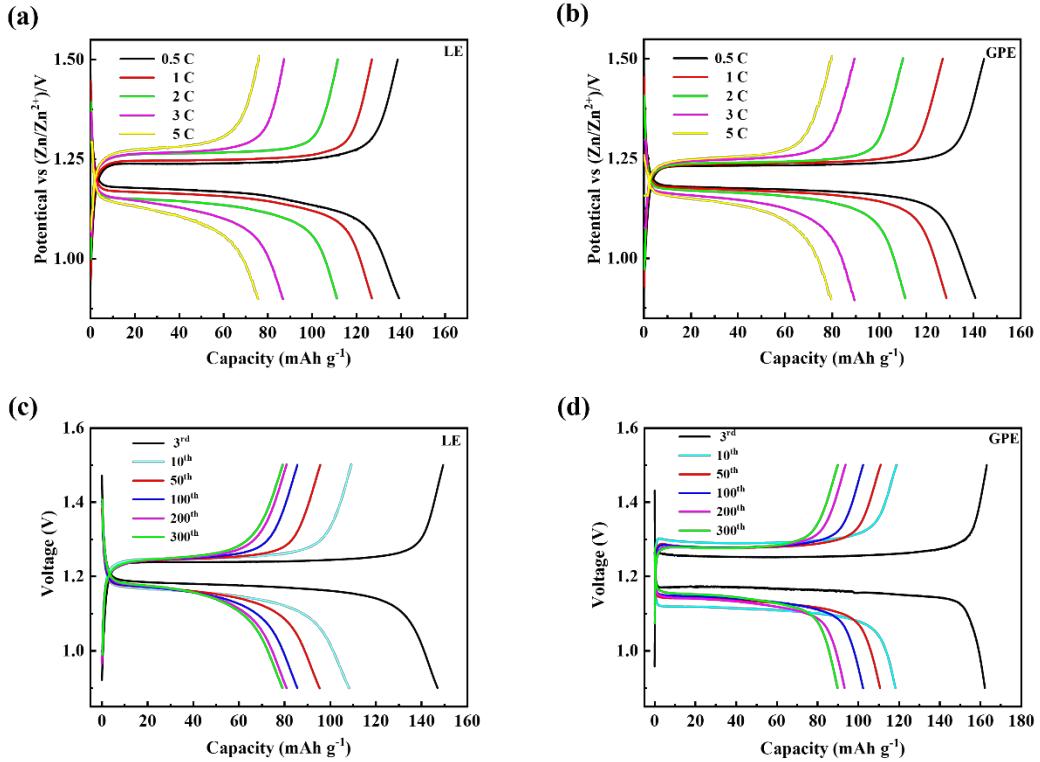
130012, PR China

2 Key Laboratory of Advanced Functional Polymer Membrane Materials of Jilin

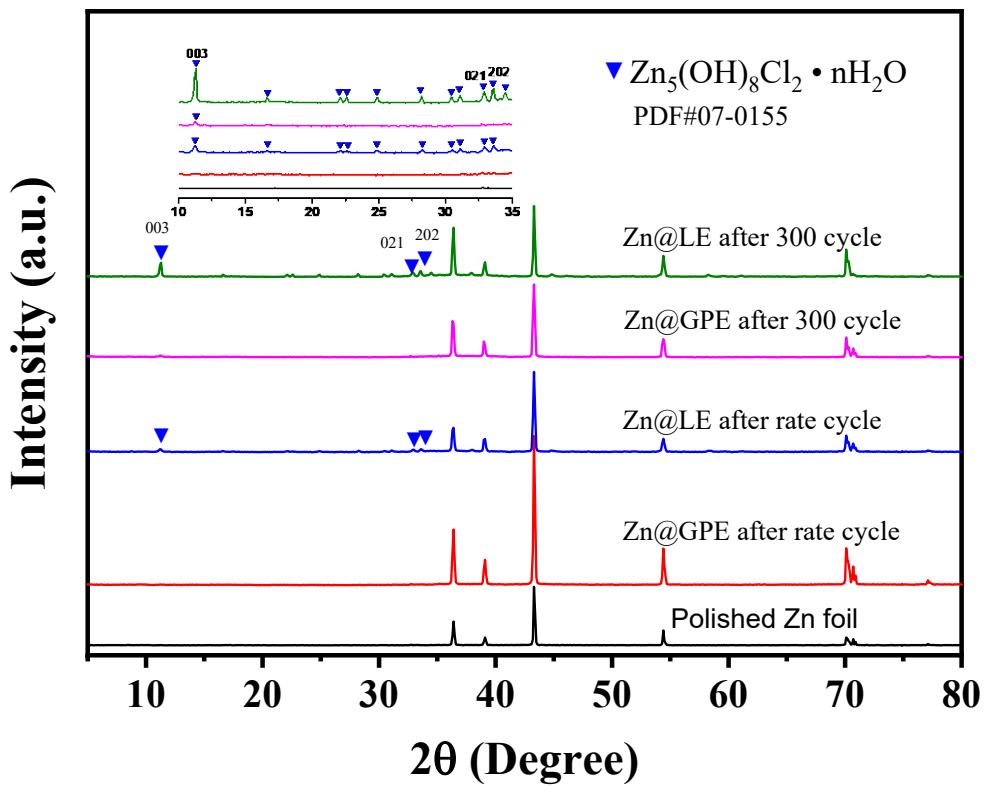
Province, Changchun 130012, PR China



**Figure S1.** Elastic modulus and toughness of GPE hydrogel matrix.



**Figure S2.** Charge/discharge profiles of (a) Zn/LE/LFP and (b) Zn/GPE/LFP, respectively. Charge/Discharge profiles of (c) Zn/LE/LFP and (d) Zn/GPE/LFP batteries at 3<sup>rd</sup>, 10<sup>th</sup>, 50<sup>th</sup>, 100<sup>th</sup>, 200<sup>th</sup> and 300<sup>th</sup> at the current density of 1 C; cut off voltage 0.9–1.5 V (vs.  $Zn/Zn^{2+}$ ).



**Figure S3.** XRD patterns of the anodes of the Zn//LFP batteries after 300cycles at 1C.

**Table S1.** Comparison of the performance of zinc-ion polymer electrolytes

gel polymer electrolyte	Salts	Ionic conductivities (S cm <sup>-1</sup> )	Ref.
SiO <sub>2</sub> -SA GPE	5M ZnCl <sub>2</sub> +4M LiCl	1.14×10 <sup>-2</sup>	This work
PVA	3M LiCl+2M ZnCl <sub>2</sub> +0.4M MnSO <sub>4</sub>	0.897×10 <sup>-2</sup>	[46]
Guar gum	2M ZnSO <sub>4</sub> +0.1M MnSO <sub>4</sub>	1.07×10 <sup>-2</sup>	[47]
PEO	ZnCl <sub>2</sub>	2~4×10 <sup>-3</sup>	[48]
Poly-ε-caprolactone	Zn(CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub>	8.8×10 <sup>-4</sup>	[49]