

Supplementary Material

Chemical, antioxidant, and antimicrobial properties of the peel and male flowers by-products of four varieties of *Punica granatum* L. cultivated in the Marche region given their exploitation in cosmetic products

Summary of Tables

Table S1. Identification of polyphenols in pomegranate by-product extracts by UPLC-ESI-MS/MS analysis.

Table S2. Identification of anthocyanins in pomegranate by-products extracts by UPLC-ESI-MS/MS analysis.

Polyphenols identification

The first large group of hydrosysable tannins identified in our work contains three main subclasses of compound: gallotannins and ellagitannins belonging to the phenol groups that are esterified to the hydroxyl groups of glucose: gallic acid in gallotannins and hexahydroxydiphenic acid (HHDP) in ellagitannins [1] and gallagyl esters.

Compound **C1** and **C2** belonging to the classes of gallotannins showing a [M-H]⁻ ions at m/z 331 and 483, respectively. **C1** reports a fragment ions in MS² at m/z 169 related to the loss of hexose fraction (162Da) and a fragment at m/z 125 which is typical for the fragmentation of gallic acid [M-H-162 Da-44 Da]. According to the literature **C1** was assigned to galloyl-hexoside, while **C2** was identified as digalloyl-hexoside according to its precursor ion at m/z 483 [2, 3].

Among hydrolysable tannins, ellagitannins compounds from **C3** to **C18** were identified. For this class of compound, characteristic fragments ions were identified as m/z 301 and attributed to the aglycone of ellagic acid that was obtained after a spontaneous-lactonization of the HHDP residue (**C16**) [3]. So, **C16** was identified as ellagic acid showing a typical signal at m/z 301 and characteristic fragment ions at m/z 229 (corresponding to the loss of four H₂O molecules) and m/z 185 (corresponding to a loss of CO₂).

C3 at 2.3 min. produced a molecular ion at m/z 481 [M-H]⁻ and was identified as HHDP-hex [3-5]. The ellagic acid hexoside moiety was also detected in the compounds **C4**, **C11**, **C12**, **C13**. **C4** showed an [M-H]⁻ ion at m/z 633 and fragments at m/z 301 and 169 which are typical for the loss of ellagic acid and the galloyl moiety. **C11** showed an [M-H]⁻ ion at m/z 935 and a typical fragments at m/z 633 and 301 and was identified as galloyl-bis-HHDP-hexoside (casuaricitin) [3]. **C12** showed an [M-H]⁻ ion at m/z 463 and typical fragments at m/z 301 indicated the loss of hexoside and was identified as ellagic acid hex[3, 4]. **C13** exhibited an [M-H]⁻ ion at m/z 951 producing fragments at m/z 933 and at m/z 301 (ellagic acid) in the MS² experiment. This fragment (m/z 933), generating fragments at m/z 915 from the loss of water, is typical for castalagin/vescalagin or galloyl-gallagyl-hexoside (galloylpunicalin, pedunculagin III) as already reported in literature [3, 5]. **C9** produced an [M-H]⁻ ion at m/z 799 and fragments at m/z 479 (loss of ellagic acid), fragments at m/z 301 (ellagic acid). Furthermore, the established fragments of ellagic acid were confirmed. This compound may be attributed to granatin A (HHDP-DHHDP-hexoside). **C6** (m/z 1415) and **C15** (m/z 433) produced fragments at m/z 933 and at m/z 301 (ellagic acid) in the MS² experiment, respectively. **C6** was identified as di(HHDP-galloyl)glucose-pentose [6]. **C15** was assigned to an ellagic acid-pentoside that

in the MS2 experiment, the ion at m/z 301 was generated by the loss of 132 Da, reasonably assigned as the elimination of pentose. In addition, ellagitannins with a gluconic acid core were also found in pomegranates. Among these, **C5** was identified as galloyl-HHDP-gluconic acid. Its $[M-H]^-$ ion at m/z 649 showed fragments at m/z 497 and 301, resulting from the loss of gallic acid (releasing HHDP-gluconic acid) and ellagic acid, respectively. Galloyl-HHDP-gluconic acid (m/z 649) also formed part of **C10** exhibiting an $[M-H]^-$ ion at m/z 801. Further fragments at m/z 348 and 497 resulted from the loss of ellagic acid and gallic acid. This compound was identified as digalloyl-HHDP-gluconic acid (punigluconin). Compound **C7** exhibited an $[M-H]^-$ ion at m/z 783. The loss of ellagic acid in the MS2 experiment produced fragments at m/z 481 and the fragment at m/z 301 of ellagic acid. Based on this fragmentation pathway compound **C7** was identified as bis-HHDP-hexoside (pedunculagin I). **C8** was characterised as digalloyl-HHDP-hexoside (pedunculagin II). This assignment is based on its $[M-H]^-$ ion at m/z 785 and the release of typical ellagitannin and gallotannin fragments at m/z 301 (ellagic acid) and 169 (galloyl group). Each of the two different retention times corresponded to an isomeric structure, also differing in their fragmentation patterns. The $[M-H]^-$ ion of **C15** was obtained at m/z 433. In the MS2 experiment, the ion at m/z 301 was generated by the loss of 132 Da, reasonably assigned as the elimination of pentose. The occurrence of the ion at m/z 300 was attributed to a homolytic rupture of the glycosidic bond [5]. In addition to ellagic acid pentoside, further monoglycosylated ellagic acid derivatives were observed as an ellagic acid deoxyhexoside (m/z 447; **C14**) and an ellagic acid derivate (der) (m/z 441; **C17**) all of them showing the typical fragments of ellagic acid (m/z 301). An $[M-H]^-$ ion at m/z 469 was observed for the isomeric compound **C18**, producing fragment ions at m/z 425 indicating the loss of a carboxyl group and typical fragments of ellagic acid. Compound **C18** was therefore assigned to the valoneic acid bilactone [3]. **C19** eluted at 4.0 min with a precursor ion at m/z 781 and a fragment at m/z 601 (gallagyl residue) were identified as gallagyl-hexoside (punicalin). Identification of punicalagin was based on its MS spectrum which showed an exact $[M-H]^-$ ion at m/z 1083. Punicalagin α and β eluted in peaks **C20** (9.2 min) and **21** (11.7 min). MS spectra of both isomers showed a $[M-H]^-$ ion at m/z 1083, which fragmented at m/z 781 and m/z 601 (corresponding to the loss of ellagic acid and gallagic acid moiety, respectively). The 781 m/z peak **C19** corresponded to the punicalin $[M-H]^-$ ion and a 302 m/z peak to the ellagic acid $[M-H]^-$ ion [7]. Gallic acid (**C22**) is a signal at m/z 169 and showed a typical fragment at m/z 125 corresponding to the loss of CO₂. **C23** showed a signal at m/z 355 and fragment ions at m/z 193,175,217 and 236. This compound was identified as ferulic acid hex due to its fragmentation revealed the formation of an aglycone fragment at m/z 193 (ferulic acid) and fragments at m/z 175 from the loss of water. **C24** was identified as dihydrokaempferol-hex and produced an ion at m/z 449 and a fragments at m/z 287 (loss of hexose moiety), 259 and 269 (loss of water). Among the class of gallotechnin compounds **25** and **26** were identified as gallocatechin and catechin-galocatechin, respectively. **C25** showed a ion at m/z 306 producing a fragments at m/z 125,137,165 and 219, while **C26** (m/z 593) showed fragments at m/z 441,305,423 and 137 suggesting that the component of aglycone in these compounds was either catechin (290 Da) or epicatechin [2]. Compounds **C30** and **C31** showed an ion at m/z 447 reporting the same fragment ions at m/z 285 in the MS2 experiment. In case of **C30** were also detect the fragments ions at m/z 255 and 227 already detected for the astralagin[2]. The **C31** were identified as luteolin-glucoside reporting a in its MS2 experiment, the fragment ions at m/z 285 which led to the identification of the aglycone as lutelolin [2]. **C29** (rt = 24.1 min) exhibited a $[M-H]^-$ ion at m/z 593. This compound was identified as kaempferol-7-orhahmano-glucoside due to in the MS2 experiment showed the fragment ion at m/z 285 and characteristic fragments at m/z 255, 227 belonging to kaempferol [2]. **C28** exhibited $[M-H]^-$ ions at m/z 609 with fragments ions at m/z 301 and 300 matched with the loss of rhamno-glucose (146 + 162 Da) and was identified as quercetin-3-O-rhahmano-glucoside (rutin) [2]. We also revealed the presence of **C27** (m/z 343, RT=1.54 min.) never found in pomegranate peel extracts, according to the literature. The identification of this compound will be further confirmed by HRMS analysis.

Table S1. Identification of polyphenols in pomegranate by-product extracts by UPLC-ESI-MS/MS analysis.

| Peak number | R.T. (min.) | Precursor ion m/z [M-H] ⁻ | MS ² ion m/z | Identification |
|------------------------------|-------------|----------------------------------------|---------------------------|-------------------------------------|
| Hydrolysable tannins | | | | |
| Gallotannins | | | | |
| C1 | 2.6 | 331 | 125,169 | Galloyl-hex |
| C2 | 11.2 | 483 | 169,125,439 | Digalloyl-hex |
| Ellagitannins | | | | |
| C3 | 2.3 | 481 | 301,275,257 | HHDP ² -hex |
| C4 | 3.6 | 633 | 301, 275, 249, 169 | Galloyl-HHDP-hex |
| C5 | 3.6 | 649 | 301, 497 | Galloyl-HHDP-glucoside |
| C6 | 6.3 | 1415 | 633, 613, 783 | Di(HHDP-galloylglucose)-pent |
| C7 | 6.8 | 783 | 301, 481, 275 | Pedunculagina I |
| C8 | 11.7 | 785 | 301, 275, 249, 169 | PedunculaginaII |
| C9 | 12.9 | 799 | 479, 301, 247 | HHDP-DHHDP-hexoside Granatin A |
| C10 | 13.7 | 801 | 649, 348 | Digalloyl-HHDP-gluc (Punigluconin) |
| C11 | 13.8 | 935 | 633, 275, 708, 301 | Galloyl-bis-HHDP-hex (Casuarinin) |
| C12 | 16.2 | 463 | 301, 300, 271, 255 | Ellagic acid-hex |
| C13 | 19.3 | 951 | 301, 933, 273, 463 | Galloyl-HHDP-DHHDP-hex (Granatin B) |
| C14 | 20.5 | 447 | 300, 301 | Ellagic acid-deoxyhex |
| C15 | 20.7 | 433 | 300, 301 | Ellagic acid-pent |
| C16 | 21.8 | 301 | 145,185,229 | Ellagic acid |
| C17 | 22.4 | 441 | 397, 398, 301 | Ellagic acid der |
| C18 | 24.2 | 469 | 425, 301, 426 | Valoneic acid bilactone |
| Gallagyl esters | | | | |
| C19 | 4.0 | 781 | 601, 721 | Punicalin |
| C20 | 9.2 | 1083 | 601, 302, 781, 575 | Punicalagin |
| C21 | 11.7 | 1083 | 601, 302, 781, 575 | Punicalagin |
| Hydroxybenzoic acids | | | | |
| C22 | 3.8 | 169 | 125 | Gallic acid |
| Hydroxycinnamic acids | | | | |
| C23 | 2.7 | 355 | 193, 175, 217, 236 | Ferulic acid-hex |
| Dihydroflavonol | | | | |
| C24 | 16.4 | 449 | 287, 259, 269 | Dihydrokaempferol-hex |
| Gallotechnin | | | | |
| C25 | 6.8 | 306 | 125, 137, 165, 219 | Gallocatechin |
| C26 | 7.0 | 593 | 441, 305, 423, 137 | Catechin-gallocatechin |
| C27 | 1.54 | 343 | 181, 149, 113, 119, 89,59 | Unknown |
| C28 | 3.2 | 609 | 301, 300, 271 | Quercetin rutinoside |
| C29 | 24.1 | 593 | 285,255,227 | Kaempferol rutinoside |
| C30 | 25.4 | 447 | 255,227,285 | Astralagine |
| C31 | 26.3 | 447 | 285 | Luteolin-glucoside |

hex=hexoside; HHDP=hexahydroxydiphenoyl; pent=pentoside; der=derivate

The ESI source in positive ion mode was selected to detect anthocyanins extracted from pomegranate peel and male flowers extracts. MS and MS2 spectra of A1-A4 peaks showed

molecular ions characteristic of cyanidin (m/z 287), pelargonidin (m/z 271) confirmed by external standards. The anthocyanins revealed the typical mass spectrometric behavior in ESI(+)-experiments, i.e. they showed M⁺ experiments and the sequential loss of their saccharide moieties, releasing the aglycones in the MS⁺ ions.

Table S2. Identification of anthocyanins in pomegranate by-products extracts by UPLC-ESI-MS/MS analysis.

| Peak number | R.T. (min.) | Precursor ion m/z [M] ⁺ | MS ² ion m/z | Identification |
|-------------|-------------|------------------------------------|-------------------------|------------------------------|
| A1 | 6.2 | 611 | 287, 449 | Cyanidin 3,5-diglucoside |
| A3 | 8.4 | 595 | 433 | Pelargonidin 3,5-diglucoside |
| A2 | 10.3 | 449 | 487 | Cyanidin 3-glucoside |
| A4 | 12 | 433 | 271 | Pelargonidin 3-glucoside |

References

1. Bar-Ya'akov Irit; Tian Li; Amir Rachel; Holland Doron, Primary Metabolites, Anthocyanins, and Hydrolyzable Tannins in the Pomegranate Fruit. *2019*, 10, (620).
2. Abdulla Rahima; Mansur Sanawar; Lai Haizhong; Ubul Ablikim; Sun Guangying; Huang Guozheng; Aisa Haji Akber, Qualitative Analysis of Polyphenols in Macroporous Resin Pretreated Pomegranate Husk Extract by HPLC-QTOF-MS. *Phytochemical Analysis* **2017**, 28, (5), 465-473.
3. Fischer Ulrike A.; Carle Reinhold; Kammerer Dietmar R., Identification and quantification of phenolic compounds from pomegranate (*Punica granatum* L.) peel, mesocarp, aril and differently produced juices by HPLC-DAD-ESI/MSn. *Food Chemistry* **2011**, 127, (2), 807-821.
4. Hernández-Corroto Ester; Marina M^a Luisa; García M^a Concepción, Extraction and identification by high resolution mass spectrometry of bioactive substances in different extracts obtained from pomegranate peel. *Journal of Chromatography A* **2019**, 1594, 82-92.
5. Abid Mouna; Yaich HÉla; Cheikhrouhou Salma; Khemakhem Ibtihel; Bouaziz Mohamed; Attia Hamadi; Ayadi M. A., Antioxidant properties and phenolic profile characterization by LC-MS/MS of selected Tunisian pomegranate peels. *J Food Sci Technol* **2017**, 54, (9), 2890-2901.
6. Mena Pedro; Calani Luca; Dall; Asta Chiara; Galaverna Gianni; García-Viguera Cristina; Bruni Renato; Crozier Alan; Del Rio Daniele, Rapid and Comprehensive Evaluation of (Poly)phenolic Compounds in Pomegranate (*Punica granatum* L.) Juice by UHPLC-MSn. *Molecules* **2012**, 17, (12).
7. Gosset-Erard Clarisse; Zhao Minjie; Lordel-Madeleine Sonia; Ennahar Saïd, Identification of punicalagin as the bioactive compound behind the antimicrobial activity of pomegranate (*Punica granatum* L.) peels. *Food Chemistry* **2021**, 352, 129396.