





## Article

# Scalable Screen-Printed TiO<sub>2</sub> Compact Layers for Fully Printable Carbon-Based Perovskite Solar Cells

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**Abstract:** Fully printable carbon-based perovskite solar cells (C-PSCs) represent some of the most promising perovskite solar cell (PSC) architectures. Highly scalable, stable, and low in cost—these devices consist of a TiO<sub>2</sub> compact layer (C-TiO<sub>2</sub>) and three sequentially screen-printed mesoporous layers of TiO<sub>2</sub>, ZrO<sub>2</sub>, and carbon, through which perovskite is infiltrated. While there has been remarkable progress in optimizing and scaling up deposition of mesoporous layers and perovskite, few publications have focused on optimizing C-TiO<sub>2</sub>. In this work, we investigate the potential for substituting commonly used spray pyrolysis with more easily scaled screen-printing. It was found that when comparing layers of similar thickness, 1 cm<sup>2</sup> devices fabricated with printed C-TiO<sub>2</sub> exhibited similar power conversion efficiency (PCE) to those fabricated with spray pyrolysis. In contrast, thicker-printed C-TiO<sub>2</sub> led to lower efficiency. The influence of TiCl<sub>4</sub> treatment on the quality of produced compact layers was also examined. This proved beneficial, mostly in the printed films, where a champion PCE of 13.11% was attained using screen-printed, TiCl<sub>4</sub> treated C-TiO<sub>2</sub>. This work proves that screen-printing is a viable replacement for spray pyrolysis in C-PSCs fabrication.

**Keywords:** compact TiO<sub>2</sub> layer; screen printed C-TiO<sub>2</sub>; TiCl<sub>4</sub> treatment; carbon-based perovskite solar cells; enhanced efficiency



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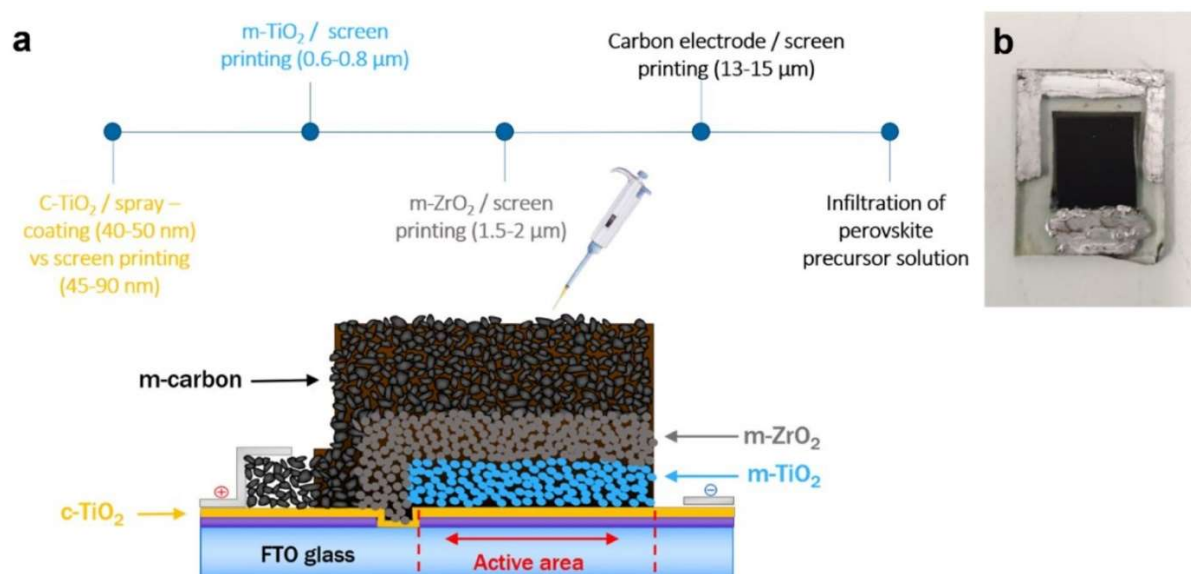


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## 1. Introduction

Organic–inorganic lead halide perovskite materials have been successfully used as light absorbers in efficient photovoltaic devices due to their exceptional optoelectronic properties. The power conversion efficiency (PCE) of perovskite solar cells (PSCs) has risen very rapidly since 2009, with the latest record standing at 25.7%, competing with the performance of commercially available photovoltaic devices [1–7]. However, problems such as the high cost of hole transporting materials (HTMs) and metal electrodes, as well as issues related to stability and upscaling devices, still represent significant barriers to commercialization [8,9].

All printable, hole transporter-free carbon-based perovskite solar cells (C-PSCs), first presented by Ku et al. in 2013 [10], currently show the most potential to overcome the above issues. This device (Figure 1) consists of a conductive fluorine-doped tin dioxide (FTO) glass substrate, a layer of compact titania (C-TiO<sub>2</sub>), an n-type mesoporous layer (usually TiO<sub>2</sub>), a mesoporous insulating layer (usually ZrO<sub>2</sub> or Al<sub>2</sub>O<sub>3</sub>), and a porous conductive carbon top contact. The perovskite precursor solution is deposited by liquid infiltration through the stack [11,12]. The absence of expensive noble metals and HTMs significantly decreases the device cost. Additionally, replacing noble metals with a hydrophobic carbon electrode prevents moisture accessing the perovskite and improves device stability [13,14]. In fact, in 2020 a printable (5-AVA)<sub>x</sub>MA<sub>1-x</sub>PbI<sub>3</sub> C-PSC passed >9000 h IEC61215:2016 standard maximum power point tests without significant performance loss [14].



**Figure 1.** (a) Schematic representation and fabrication method of standard C-PSC. (b) Image of a 1 cm<sup>2</sup> active area C-PSC.

C-PSCs are often cited as promising candidates for commercialization, because of their low cost, high stability, and the use of industrially applicable screen-printing processes [15,16]. Multiple publications exist on enhancing efficiency by optimizing the functional layers (electron transporting layer, insulator, and carbon) [17–21] and modifying the perovskite formulation, with the current highest reported PCEs standing at more than 18% [22–24]. The use of mixed cations/anions perovskites has typically delivered solar cells with high efficiency and stability; however, in the case of triple mesoporous C-PSCs, MAPbI<sub>3</sub> perovskite is the most well-established absorber due to its convenience and effectiveness in infiltration [15,16]. In addition, the use of (5-AVA)<sub>x</sub>MA<sub>1-x</sub>PbI<sub>3</sub> has led to the most stable carbon perovskite solar cell reported, achieved through the strengthening of the MAPbI<sub>3</sub> with the bifunctional organic molecule 5-AVA leading to a reduction in decomposition or reconstruction [14]. In terms of the industrially applicable screen-printed processes, large-area modules have been produced already using MAPbI<sub>3</sub>·AVA perovskites, with recorded PCEs of 10 and 11% at 10 × 10 cm<sup>2</sup> [25,26], ~6% at A4 size [27], and >9% at 220 cm<sup>2</sup> with the use of an alternative, nontoxic solvent system [28].

In the majority of these publications, the mesoporous layers are deposited using screen-printing, whereas the C-TiO<sub>2</sub> layer is usually deposited via spray pyrolysis. This fact represents one significant barrier to the fabrication of large area C-PSC devices, since spray pyrolysis does not allow high versatility in pattern control, and it is not as easily scalable as screen-printing.

The compact layer is essential to PSC performance as it prevents direct contact between holes formed in the perovskite and FTO electrode. Such contact results in significant charge recombination and voltage loss for the device [29–34]. Therefore, an efficient C-TiO<sub>2</sub> needs to be uniform and pinhole-free. Furthermore, it has to be transparent: if the transmittance of C-TiO<sub>2</sub> reduces, the PCE falls due to decreased light absorption. Moreover, the compact layer thickness must be minimised to reduce series resistance. It has also been proven that the thickness of this layer can strongly affect the hysteresis and the performance of C-PSCs [35] and of planar perovskite devices as well [36].

Due to the limited scalability of spray pyrolysis, alternative deposition methods for C-TiO<sub>2</sub> have been studied. In conventional architectures, Smith et al. used commercially available glass substrates that incorporate chemical vapor deposited fluorine doped tin oxide C-TiO<sub>2</sub> [37]. Perovskite devices produced with this substrate proved superior to those where C-TiO<sub>2</sub> was deposited via spray pyrolysis. In C-PSCs, the atomic layer deposition

(ALD) has been reported as a successful alternative method for depositing the compact layer [38]. Screen-printing has been trialled in other perovskite solar cell architectures but was previously found to be less effective than chemical bath deposition or spin coating due to low C-TiO<sub>2</sub> coverage on the FTO substrate [39]. In this work, we overcome these previous challenges and present pinhole-free screen-printed compact layers, with optimum thickness that leads to highly efficient C-PSCs, proving that this is a viable method for device fabrication.

It has also been proven that the high temperature annealing of C-TiO<sub>2</sub> forms oxygen vacancy defects, which act as charge recombination centres [40]. In the case of screen-printed compact layers, the TiCl<sub>4</sub> treatment is potentially more important, if we consider the less even screen-printed areas, which could be more prone to oxygen vacancies at high temperature as they are rougher and, therefore, have higher surface areas.

Post-annealing treatment with TiCl<sub>4</sub> has been widely used for reducing such surface traps, decreasing charge recombination at the TiO<sub>2</sub>/perovskite interface in perovskite solar cells and C-PSCs [41–45]. However, while some publications have previously reported highly efficient C-PSCs that incorporate TiCl<sub>4</sub> treated screen-printed C-TiO<sub>2</sub> [46,47], they did not focus on the compact layers and the effect of the TiCl<sub>4</sub> treatments was not discussed. Here, we investigate the impact of such treatments on different deposited compact layers and subsequent device performance, to further understand the influence of TiCl<sub>4</sub>, in MAPbI<sub>3</sub>.AVA triple mesoporous C-PSCs.

## 2. Materials and Methods

### 2.1. Materials

Conductive fluorine-doped tin oxide glass (FTO, TEC7, XOP) was used as the substrate. Samples and devices were prepared using the following: anhydrous 2-propanol (IPA, 99.5%), carbon paste (Gwent electronic materials (UK)), ZrO<sub>2</sub> paste (Solaronix (Switzerland)), TiO<sub>2</sub> paste (30NR-D, Dyesol (Australia)), terpineol (95%, Sigma-Aldrich (UK)), titanium diisopropoxide bis(acetylacetonate) (TAA, 75% in IPA, Sigma-Aldrich), Ti-Nanoxide BL/SP (Solaronix), and Titanium (IV) chloride tetrahydrofuran complex (Sigma Aldrich). PbI<sub>2</sub> (99%, Sigma-Aldrich), MAI (CH<sub>3</sub>NH<sub>3</sub>I, anhydrous, Dyesol), 5-ammonium valeric acid iodide (5-AVAI, Dyesol), and  $\gamma$ -Butyrolactone (Sigma Aldrich) were used as received for the preparation of perovskite precursors.

### 2.2. Fabrication of TiO<sub>2</sub> Compact Layers

FTO substrates were cleaned with a solution of ~2% Hellmanex in deionised water, rinsed with acetone and IPA, and a 5 min plasma clean in O<sub>2</sub>. Samples with sprayed compact TiO<sub>2</sub> layers were prepared by spray pyrolysis of 10% titanium diisopropoxide bis(acetylacetonate) in IPA at 300 °C. Two different screen-printed compact layers, in terms of thickness, were prepared as well. One printing of Ti-Nanoxide BL/SP commercial paste with a screen of 130-34 led to similar thickness to the sprayed one, while two printings with the same conditions led to higher thickness. After the printings, the samples were annealed at 550 °C.

For the TiCl<sub>4</sub> treatment, samples were dipped into a 0.05 M TiCl<sub>4</sub> solution in H<sub>2</sub>O for 30 min. Samples were then copiously rinsed and calcined at 500 °C.

### 2.3. Device Fabrication

For the device fabrication, FTO substrates were patterned with a Nb:YVO<sub>4</sub> laser (532 nm) before the cleaning procedure. After the deposition of C-TiO<sub>2</sub> (by spray pyrolysis or screen-printing together with the TiCl<sub>4</sub> treatment), the mesoporous TiO<sub>2</sub> paste was diluted 1:1 by weight in terpineol, screen-printed, and sintered at 550 °C. Next, the mesoporous ZrO<sub>2</sub> and carbon were sequentially screen-printed, and each annealed at 400 °C. A perovskite precursor solution (0.439 g PbI<sub>2</sub>, 0.1514 g MAI and 0.0067 g 5-AVAI in 1 mL  $\gamma$ -Butyrolactone) was then drop cast (15  $\mu$ L) onto the cooled stack. Devices were left in air for 10 min to allow the solution to percolate through the stack, before annealing in an oven

for 1 h at 50 °C (Figure 1). The finished cells were then exposed to a standard 70% relative humidity process at 40 °C for 24 h to induce a recrystallisation [48] and then dried under vacuum before measuring.

#### 2.4. Characterization Techniques

Optical transmission spectra of the tested TiO<sub>2</sub> compact layers was examined using a Perkin Elmer UV/vis/NIR spectrophotometer Lambda 750 with a 100 mm InGaAs integrating sphere. Scans from 800 to 300 nm were taken with a 10 nm data interval at 266.75 nm min<sup>−1</sup> with transmission mode.

The morphology of the tested films was examined using a HITACHI scanning electron microscope (SEM), while films thicknesses (Figure 1) were measured with a DEKTAK 150 profilometer system.

Cyclic voltammetry (CV) was conducted using a Zahner ZENNIUM X electrochemical workstation. Samples were mounted against an O-ring to define the area exposed to the electrolyte (~1 cm<sup>2</sup>). Measurements were performed at a scan rate of 50 mV s<sup>−1</sup> in an electrolyte containing 0.5 mM potassium ferricyanide/ferrocyanide in aqueous 0.5 M KCl against a Ag/AgCl (3M KCl) reference electrode

IV testing of the cells was performed under a class AAA solar simulator (Newport Oriel Sol3A) at AM1.5 and 100 mW cm<sup>−2</sup> illumination, calibrated using a KG5 filtered reference cell. The cell area was masked to 0.49 cm<sup>2</sup> and devices were light soaked for 3 min before current density (J)–voltage (V) sweeps were performed from open-circuit voltage (V<sub>OC</sub>) to short-circuit current density (J<sub>SC</sub>) and vice-versa, at a rate of 330 mV/s using a Keithley 2400 source meter. For stabilised power output measurements, device bias was set to the maximum power point voltage as determined by the J–V sweep and current monitored under 100 mW cm<sup>−2</sup> illumination for 200 s.

### 3. Results and Discussion

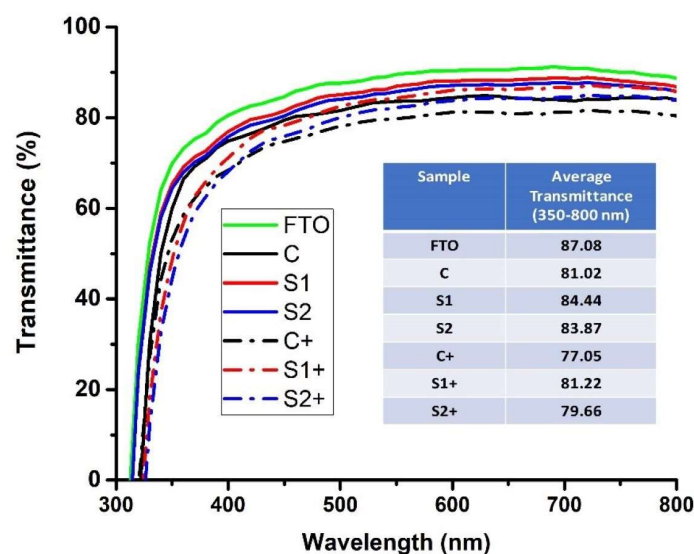
To investigate the viability of screen-printing for the C-TiO<sub>2</sub> deposition, two different thicknesses were screen-printed on FTO substrates and compared to those deposited with the standard spray pyrolysis method. The sprayed control samples (C) were found to be 40–50 nm. For the screen printed, a single print (S1) formed a 45–60 nm film, while two prints (S2) (without drying between printing) formed a 70–90 nm layer. Samples of each type were also exposed to a TiCl<sub>4</sub> treatment, to produce C+, S1+, and S2+ (Table 1).

**Table 1.** Nomenclature of the different C-TiO<sub>2</sub> samples.

Type of C-TiO <sub>2</sub> Samples	Nomenclature
Sprayed control samples	C
Single printing samples	S1
Two printing samples	S2
Control samples (sprayed) + TiCl <sub>4</sub> treatment	C+
Single printing samples + TiCl <sub>4</sub> treatment	S1+
Two printing samples + TiCl <sub>4</sub> treatment	S2

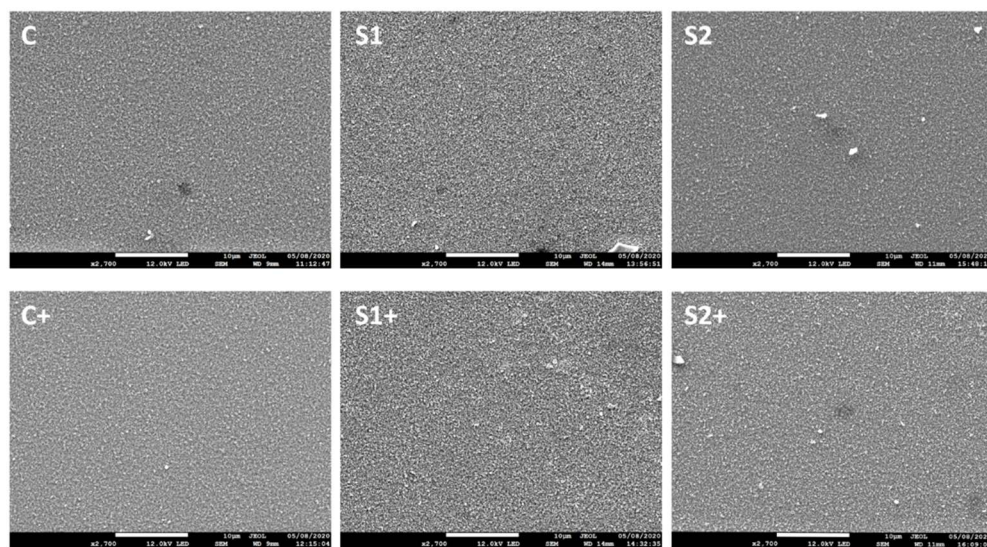
During PSC operation, light accesses the device via the glass/FTO substrate. Therefore, the most efficient compact layers will exhibit high transmission. Figure 2 shows optical transmission spectra for all tested compact layers, as well as plain FTO glass. Although coated samples exhibit lower transmission to plain FTO glass, they still exhibit high average optical transmissions of between 77% and 84% across the visible spectrum. Printed layers record higher transmission than sprayed, while the TiCl<sub>4</sub> treatment reduces the value by around 4% for each case.





**Figure 2.** Optical transmission spectra for the tested compact layers and plain FTO glass. The embedded table shows the average transmission (AT) of the samples across the visible spectrum.

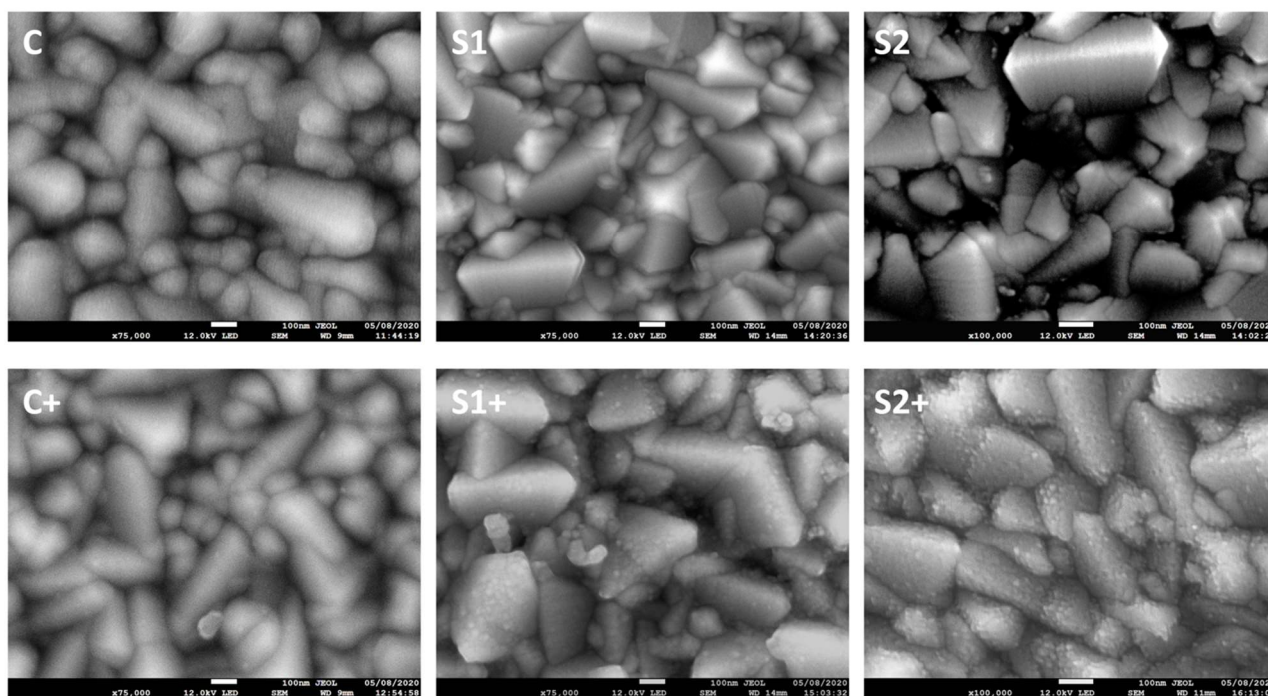
Surface SEM was performed to examine the C-TiO<sub>2</sub> surface coverage and check for pinholes. As shown in Figure 3, the sprayed sample and S1 printed sample look very similar, with both showing good FTO coverage. However, in the case of S2, there are some visible white spots, which likely correspond to pinholes. These pinholes can be attributed to the excessive thickness of the S2 sample: previous reports have proved that the cracking of TiO<sub>2</sub> layers increases with the thickness of the layer, especially at FTO crystal boundaries, due to higher tension forces exerted on films in these areas [36,49]. Such pinholes could act as charge recombination centres and compromise  $V_{oc}$  and the fill factor (FF) of the corresponding S2 C-PSCs.



**Figure 3.** Surface SEM images of the tested C-TiO<sub>2</sub> layers that showing comparative coverage of the FTO glass. Scale bar is 10 µm.

Surface SEM images with higher resolution are shown in Figure 4, where we can more clearly observe individual C-TiO<sub>2</sub> crystals and see the impact of TiCl<sub>4</sub> treatment. The printed samples are very conformal, with more angular crystals compared to the sprayed one. In addition, the impact of the TiCl<sub>4</sub> treatment is significantly evident on the printed samples (especially on the S2), whereas not so evident on the surface of the sprayed sample.

This fact suggests that the angular crystals of the printed layers are potentially a more suitable surface for the binding of  $\text{TiCl}_4$ .



**Figure 4.** Higher resolution surface SEM images that show the impact of  $\text{TiCl}_4$  treatment in the tested films. Scale bar is 100 nm.

Cyclic voltammetry is an electrochemical technique that can provide useful information for the quality of a compact layer on an FTO surface [50]. Figure 5 shows cyclic voltammograms of different C- $\text{TiO}_2$  films and plain FTO exposed to potassium ferricyanide/ferrocyanide electrolyte. In this experiment, the ferricyanide/ferrocyanide solution acts as a model redox system in a three-electrode cell, while the tested sample acts as a working electrode. The molecules of the potassium ferricyanide/ferrocyanide redox couple are small enough to penetrate small pinholes and cracks in the C- $\text{TiO}_2$  to reach the underlying FTO. The C- $\text{TiO}_2$  surface coverage can therefore be estimated using the magnitude of the anodic current density in the cyclic voltammogram, as the C- $\text{TiO}_2$  should form a dielectric contact with the redox couple and so repress the anodic current. Therefore, any anodic current will be a consequence of exposed FTO.

As there is minimal current, all the C- $\text{TiO}_2$  samples present blocking ability over the voltage range compared to bare FTO.  $\text{TiCl}_4$  treatment decreases the anodic current in all the samples but has a higher effect in the case of the S2 printed layer, which seems to be of worse quality compared to S1 and sprayed layers. This result confirms our observations from the surface SEM images where the presence of some pinholes on S2 surface were evident and the  $\text{TiCl}_4$  treatment had higher impact on it.

C-PSCs fabricated with the different C- $\text{TiO}_2$  layers and the statistical results of their performance are shown in Figure 6 and Table 2. S1 devices recorded average PCEs of  $11.61 \pm 0.37\%$ , which is comparable with the average PCEs of the sprayed C- $\text{TiO}_2$  devices ( $11.72 \pm 0.24\%$ ). Printed devices exhibited higher currents compared to the cells with sprayed C- $\text{TiO}_2$ , which could be due to the slightly higher transmittance. Higher transmittance results in higher light absorption into the perovskite layer and, therefore, higher current. In contrast, devices with thicker S2 layers recorded lower average efficiency ( $9.60 \pm 0.20\%$ ) due to lower values of  $V_{oc}$  and FF, likely a consequence of the pinholes observed in Figure 3.  $\text{TiCl}_4$  treatment was beneficial for all devices, particularly impacting those with printed compact layers. More specifically,  $\text{TiCl}_4$  treatment increased the average

PCE from  $11.61 \pm 0.24\%$  for S1 to  $12.87 \pm 0.18\%$  for S1+, while  $\text{TiCl}_4$  in thicker S2 layers increased the average efficiency from  $9.60 \pm 0.2\%$  to  $11.55 \pm 0.22\%$ . In the case of the sprayed compact layer, the improvement is less marked (from  $11.72 \pm 0.24\%$  to  $12.2 \pm 0.18\%$ ).

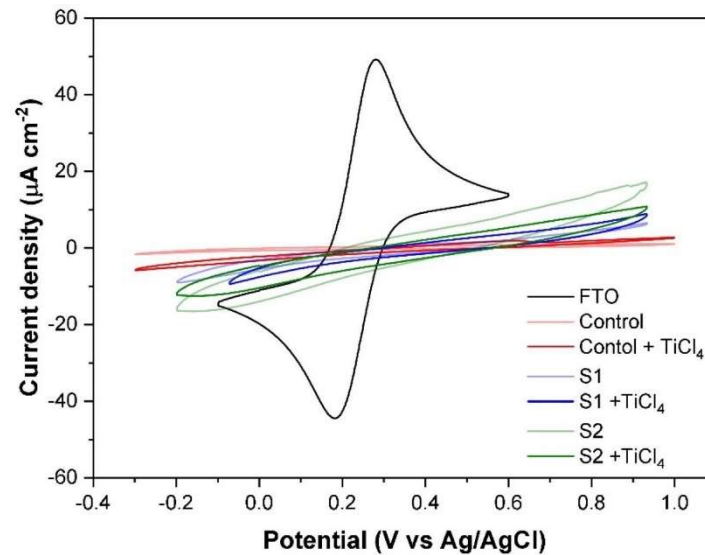


Figure 5. CV measurements of all the tested layers and FTO glass.

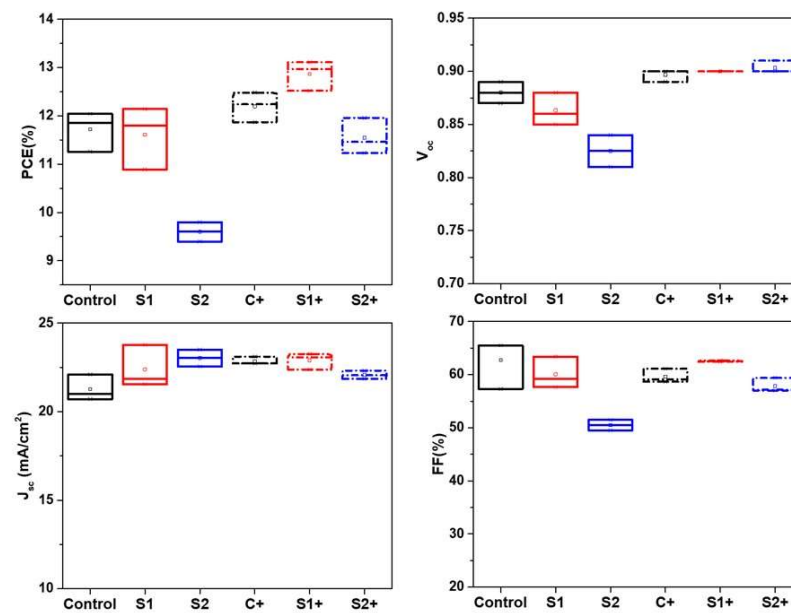


Figure 6. Statistical results of C-PSCs with different compact layers of  $\text{TiO}_2$ . Black represents the control, red represents those with a single printed layer and blue represents those with a double printed layer.

Table 2. Average values of photovoltaic parameters of C-PSCs with different  $\text{TiO}_2$  compact layers.

Device	PCE (%)	$J_{sc}$ (mA/cm <sup>2</sup> )	$V_{oc}$ (Volts)	FF (%)
C	$11.72 \pm 0.24$	$21.27 \pm 0.42$	$0.88 \pm 0.01$	$62.78 \pm 2.71$
S1	$11.61 \pm 0.37$	$22.39 \pm 0.70$	$0.86 \pm 0.01$	$60.08 \pm 1.71$
S2	$9.60 \pm 0.20$	$23.03 \pm 0.48$	$0.83 \pm 0.01$	$50.5 \pm 0.99$
C+	$12.20 \pm 0.18$	$22.85 \pm 0.13$	$0.9 \pm 0.00$	$59.63 \pm 0.76$
S1+	$12.87 \pm 0.18$	$22.9 \pm 0.33$	$0.9 \pm 0.00$	$62.52 \pm 0.06$
S2+	$11.55 \pm 0.22$	$22.08 \pm 0.13$	$0.9 \pm 0.00$	$57.85 \pm 0.77$

These results correlate well with the CV measurements, where  $\text{TiCl}_4$  had the greatest impact on the thick, lower quality S2 samples. The higher impact of  $\text{TiCl}_4$  can also be explained by the surface SEM images in Figure 4, where  $\text{TiCl}_4$  is more evident on the angular crystals.

The current density (J)–voltage (V) of the best performing C-PSCs, are shown in Figure 7, with detailed photovoltaic parameters presented in Table 3. A champion PCE of 13.11% was recorded for an S1+ device, while the untreated S1 champion exhibited a PCE of 12.14%. In the case of S2 samples the PCE of the champion device was lower (9.8%). However, the use of  $\text{TiCl}_4$  increased the efficiency to 11.96%. The device with the sprayed compact layer, recorded a champion PCE at 11.86%, which increased with the  $\text{TiCl}_4$  treatment (12.48%). The higher  $J_{sc}$  in the case of printed C- $\text{TiO}_2$  can be attributed to the higher transmittance of these layers, while the higher  $V_{oc}$  and FF of the sprayed sample can be attributed to the lower incidence of pinholes.

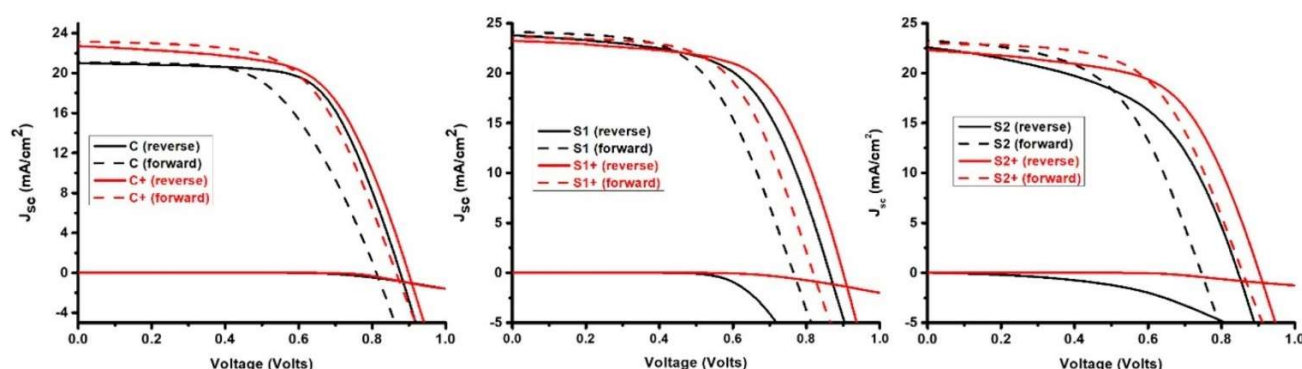


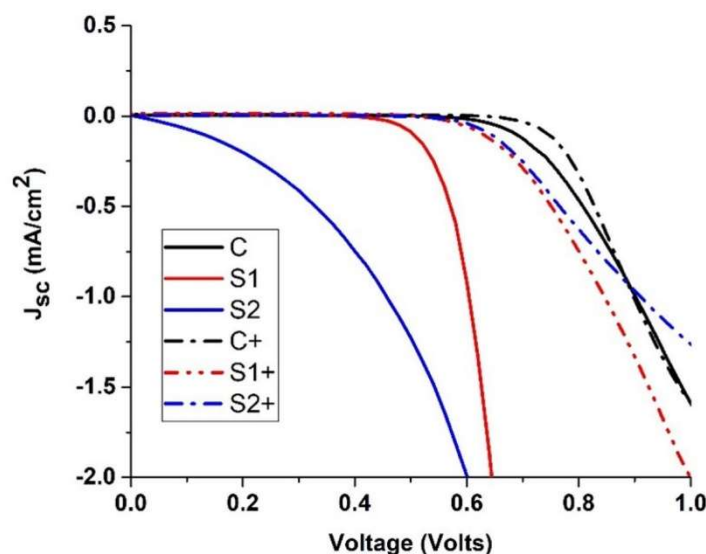
Figure 7. JV curves of the champion devices with the different C- $\text{TiO}_2$  blocking layers.

Table 3. Photovoltaic parameters of champion devices with the different C- $\text{TiO}_2$  blocking layers.

Samples	PCE (%)	$J_{sc}$ ( $\text{mA}/\text{cm}^2$ )	$V_{oc}$ (Volts)	FF (%)
C rev.	11.86	20.71	0.87	65.46
C for.	9.68	20.78	0.81	57.21
S1 rev.	12.14	23.77	0.86	59.19
S1 for.	10.38	24.1	0.76	56.29
S2 rev.	9.8	22.55	0.84	51.48
S2 for.	9.22	23.21	0.74	53.317
C+ rev.	12.48	22.72	0.9	61.12
C+ for.	11.9	23.14	0.87	59.46
S1+ rev.	13.11	23.26	0.9	62.64
S1+ for.	11.61	23.6	0.82	60.09
S2+ rev.	11.96	22.32	0.9	59.38
S2+ for.	11.59	22.95	0.86	58.88

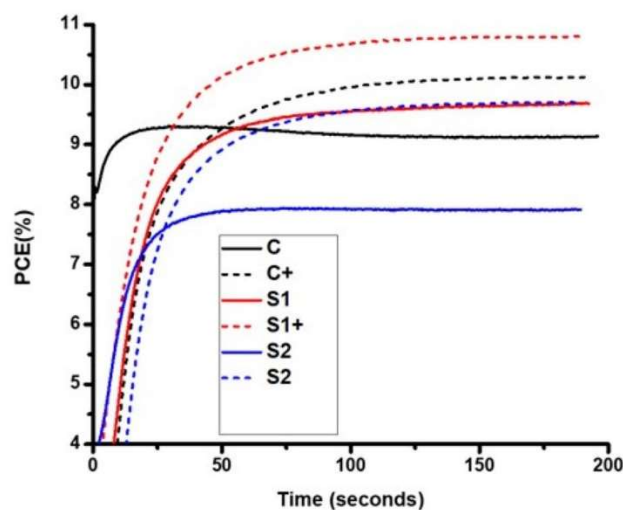
Dark current measurements of the champion devices are shown in Figure 8. In these measurements, increased onset voltage values indicate a reduction in the electron–hole recombination between perovskite and the  $\text{TiO}_2$  and FTO conduction bands [38]. It is obvious that the devices with printed compact layers show increased recombination phenomena since the onset voltage value is low, especially in the case of the S2 device. This is reasonable if we consider observed pinholes, which act as charge recombination centres (Figure 3). However, this behaviour is not apparent after  $\text{TiCl}_4$  treatment, where the onset voltage is increased in all the samples, particularly S2. This confirms our previous conclusions from SEM and CV characterisations regarding the contribution of  $\text{TiCl}_4$  in improving the printed C- $\text{TiO}_2$  quality compact layers.





**Figure 8.** J-V curves of the champion devices with the different blocking layers in the dark.

To provide additional confidence in the device performance, stabilised current measurements were also performed on the devices and the corresponding graphs of the champion cells are shown in Figure 9.



**Figure 9.** Stabilised PCEs at one sun of devices with different TiO<sub>2</sub> blocking layers.

These measurements confirm the observed trend from the J–V measurements, where devices with S1 and sprayed compact layers recorded similar PCEs, while thicker blocking layers (S2) led to lower efficiencies. As with other measurements, all devices with TiCl<sub>4</sub> treated layers showed improved PCEs, with the highest improvement obtained in the case of the S2 device. Once again, the use of S1+ layer led to the highest efficiency (10.81%). The conversion efficiencies become stable between 50 and 100 s after the start of the measurement, and in all the samples, the recorded J–V PCE is higher than the stabilised PCE. This behaviour is frequently observed in C-PSCs and is attributed to imbalanced charge extraction, as electrons are generated close to TiO<sub>2</sub> electron transporting layer while holes need to travel through a thick ZrO<sub>2</sub> layer to access the carbon electrode [51,52]. It is also noteworthy that devices were light soaked for 3 min before current density (J)–voltage (V) sweeps were performed from open-circuit voltage (V<sub>OC</sub>) to short-circuit current density (J<sub>SC</sub>) and vice-versa. This light soaking is important for C-PSCs to achieve their highest performances as the presence of the AVA molecule inhibits the movement of ions [53].

All the above results confirm the criticality of C-TiO<sub>2</sub> layer thickness, in accordance with previous publications [34,36,44,49]. Films with thickness from 40–60 nm (sprayed and S1) are pinhole-free (Figure 3), with high blocking ability (Figure 5). The comparable PCE results of the devices with the S1 and sprayed C-TiO<sub>2</sub> layers show that screen-printing is a viable substitution for spray pyrolysis in depositing C-TiO<sub>2</sub> films for highly efficient C-PSCs. The fact that S1 cells exhibit higher  $J_{sc}$  is also encouraging. Films with thicknesses of more than 70 nm present pinholes (Figure 3) due to higher tension forces exerted on the films, which result in C-TiO<sub>2</sub> cracking of layers at FTO crystal edges. These pinholes act as charge recombination centres and, thus, the corresponding cells record lower  $V_{oc}$  and FF, and as a result, lower PCE. It should also be noted that TiCl<sub>4</sub> treatments proved more beneficial in the case of printed compact layers. The more angular crystals in these printed layers (Figure 4) may provide cleaner, more easily accessible binding sites for the TiCl<sub>4</sub> reducing the impact of random printing flaws and improving the layer quality.

#### 4. Conclusions

In this work, we study the surface morphology and the optical and electrochemical properties of screen-printed TiO<sub>2</sub> compact layers, in order to investigate their viability for application in C-PSCs. A high quality, pinhole-free, screen-printed compact layer with over 84% transmittance and similar thickness to the standard sprayed blocking layer was obtained. C-PSCs fabricated with such printed layers exhibited 11.86% average PCE, comparable to that of devices produced via spray pyrolysis. Furthermore, TiCl<sub>4</sub> treatment was found to drastically improve the quality of the screen-printed films, with the corresponding devices producing a champion PCE of 13.11%.

In summary, the screen-printing was found to be a suitable method for depositing C-TiO<sub>2</sub> in C-PSCs. Additionally, layer quality and device performance may be significantly enhanced by employing a TiCl<sub>4</sub> treatment. As screen-printing is cheap and easy to scale up, these results could enhance the production of highly efficient solar cells and modules, hence facilitating the production of commercially competitive C-PSCs.

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