

Supporting Information

5-Chlorocoumaranone- Conjugates as Chemiluminescent Protecting Groups (CLPG) and Precursors to Fluorescent Protecting Groups (FPG)

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Funding Acquisition: Lead
T.L Conceptualization: Assist,
Investigation: Lead
R.H. Investigation: Assist
M.S.W. Conceptualization: Assist
L.S. Methodology: Lead
N.T.F. Methodology: Assist

Table of contents

1. NMR and IR spectra.....	1
1.1 NMR and IR spectra of compound 5A	1
1.2 NMR and IR spectra of compound 5E.....	5
1.3 NMR and IR spectra of compound 6	9
1.4 NMR and IR spectra of compound 8	13
2. Decomposition Experiments.....	17
2.1 NMR and ESI-spectra of 5E.....	20
2.2 NMR-spectra of 5A.....	23
2.3 NMR-spectra of 12.....	28

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1. NMR and IR spectra

1.1 NMR and IR spectra of compound 5A

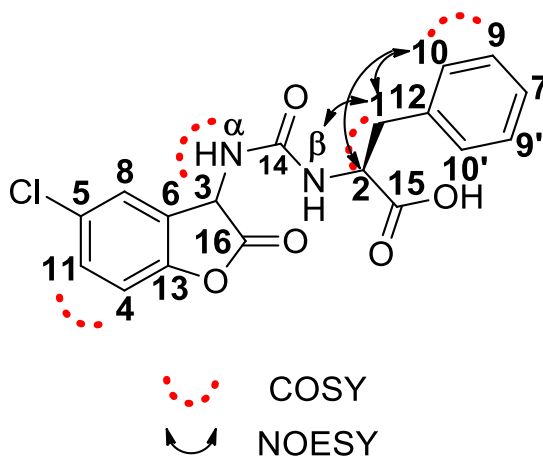


Table S1: 1D and 2D-NMR data of (2S)-2-(3-(5-chloro-2-oxo-2,3-dihydrobenzofuran-3-yl)ureido)-3-phenylpropanoic acid (5A) in DMSO- d_6 , at 298 K and 500 MHz for ^1H and 125 MHz for ^{13}C .

No.	δ_{H} [ppm], J in [Hz]	δ_{C} [ppm], mult.	HMBC ($^{\circ}\text{J}$)
1	3.26+3.36 (2H, m)	33.3+33.4, CH_2	$\text{C1} \rightarrow \text{H2}(^2\text{J}), \text{H10}(^3\text{J})$
2	4.83+4.84 (1H, ψt , $J = 10.9$ Hz)	52.92+52.98, CH	$\text{C2} \rightarrow \text{H1}(^2\text{J})$
3	5.21+5.30 (1H, s)	54.01+54.04, CH	$\text{C3} \rightarrow \text{H4}(^4\text{J}), \text{H8}(^3\text{J}), \text{NH}_\alpha(^2\text{J})$
4	6.84+6.87 (1H, d, $J = 8.7$ Hz)	117.0+117.2, CH_{arom}	$\text{C4} \rightarrow \text{NH}_\beta(^7\text{J})$
5	-	122.42+122.47, C_q	$\text{C5} \rightarrow \text{H8}(^2\text{J}), \text{H4}(^3\text{J}), \text{H11}(^2\text{J})$
6	-	123.59+123.60, C_q	$\text{C6} \rightarrow \text{H3}(^2\text{J}), \text{H4}(^3\text{J}), \text{NH}_\beta(^5\text{J})$
7	7.22 (1H, m)	126.6+126.7, CH_{arom}	$\text{C7} \rightarrow \text{H10}(^3\text{J})$
8	6.63+6.77 (1H, d, $J = 2.5$ Hz)	127.3+127.4, CH_{arom}	$\text{C8} \rightarrow \text{H3}(^3\text{J}), \text{H11}(^3\text{J})$
9	7.27 (2H, m)	128.28+128.31, $2 \times \text{CH}_{\text{arom}}$	$\text{C9} \rightarrow \text{H9}(^3\text{J})$
10	7.15 +7.21 (2H, m)	128.81+128.86, $2 \times \text{CH}_{\text{arom}}$	$\text{C10} \rightarrow \text{H1}(^3\text{J}), \text{H10}(^3\text{J})$
11	7.13+7.19 (1H, m)	129.30+129.33, CH_{arom}	$\text{C11} \rightarrow \text{H8}(^3\text{J})$
12		137.34+137.36, C_q	$\text{C12} \rightarrow \text{H1}(^2\text{J}), \text{H2}(^3\text{J}), \text{H9}(^3\text{J})$
13		154.6, C_q	$\text{C13} \rightarrow \text{H3}(^3\text{J}), \text{H4}(^2\text{J}), \text{H8}(^3\text{J}), \text{H11}(^3\text{J}), \text{NH}_\beta(^6\text{J})$
14	-	155.96+156.07, C_q	$\text{C14} \rightarrow \text{H2}(^3\text{J}), \text{H3}(^3\text{J}), \text{NH}_\alpha(^2\text{J})$
15	-	170.1, C_q	$\text{C15} \rightarrow \text{H1}(^3\text{J}), \text{H2}(^2\text{J})$
16	-	171.87+171.95, C_q	$\text{C16} \rightarrow \text{H2}(^6\text{J}), \text{H3}(^2\text{J}), \text{NH}_\alpha(^3\text{J})$
NH_α	8.48 (1H, s)	-	-
NH_β	10.19+10.21 (1H, s)	-	-
COOH	13.24 (1H, br)	-	-

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Note: The ^1H - and ^{13}C -NMR-spectra show a mixture of rotamers.

Chemical Formula: $\text{C}_{18}\text{H}_{15}\text{ClN}_2\text{O}_5$

Molecular Weight: 374.78 g/mol

IR: $\tilde{\nu}[\text{cm}^{-1}]$ = 3854 (w), 3650 (w), 3620 (w), 3600 (w), 3315 (br), 2363 (w), 1772 (w), 1737 (w), 1716 (m), 1694 (vs), 1599 (w), 1559 (w), 1497 (m), 1438 (m), 1426 (m), 1359 (w), 1281 (m), 1203 (m), 1143 (w), 1121 (w), 1017 (w), 959 (w), 932 (s), 906 (w), 821 (m).

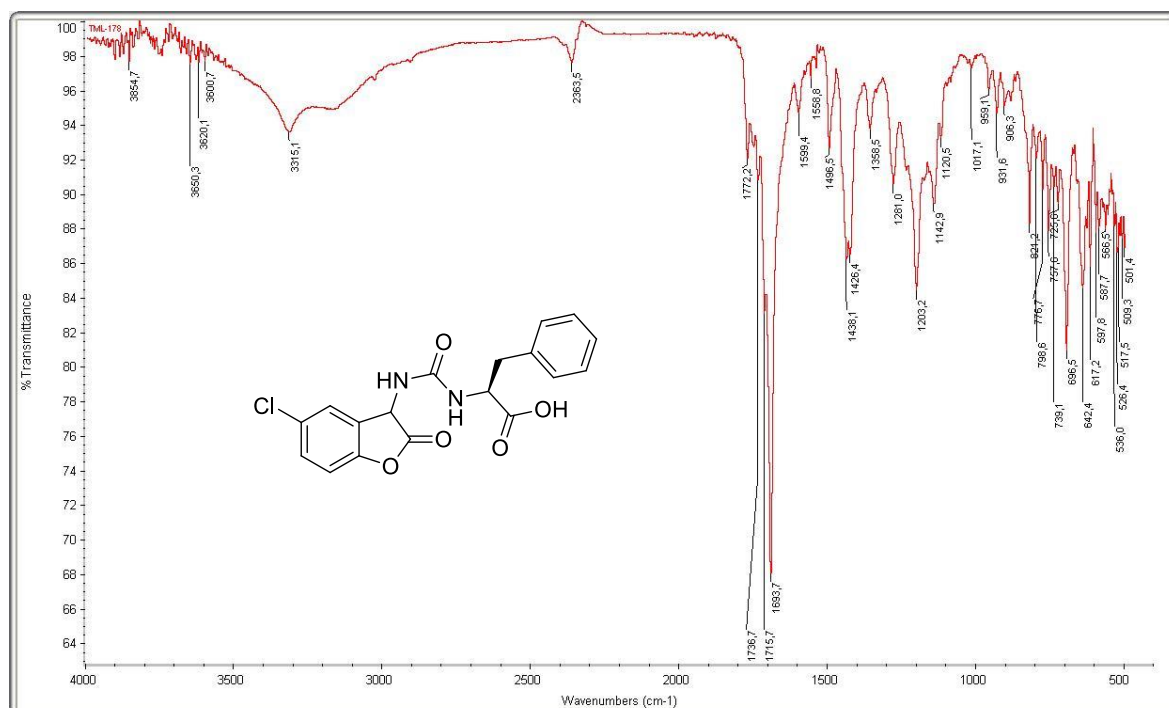
HR-MS: Theor. $[\text{M}+\text{H}]^+$: 375.07422, found: 375.07423.

Theor. $[\text{M}+\text{Na}]^+$: 397.05617, found: 397.05640.

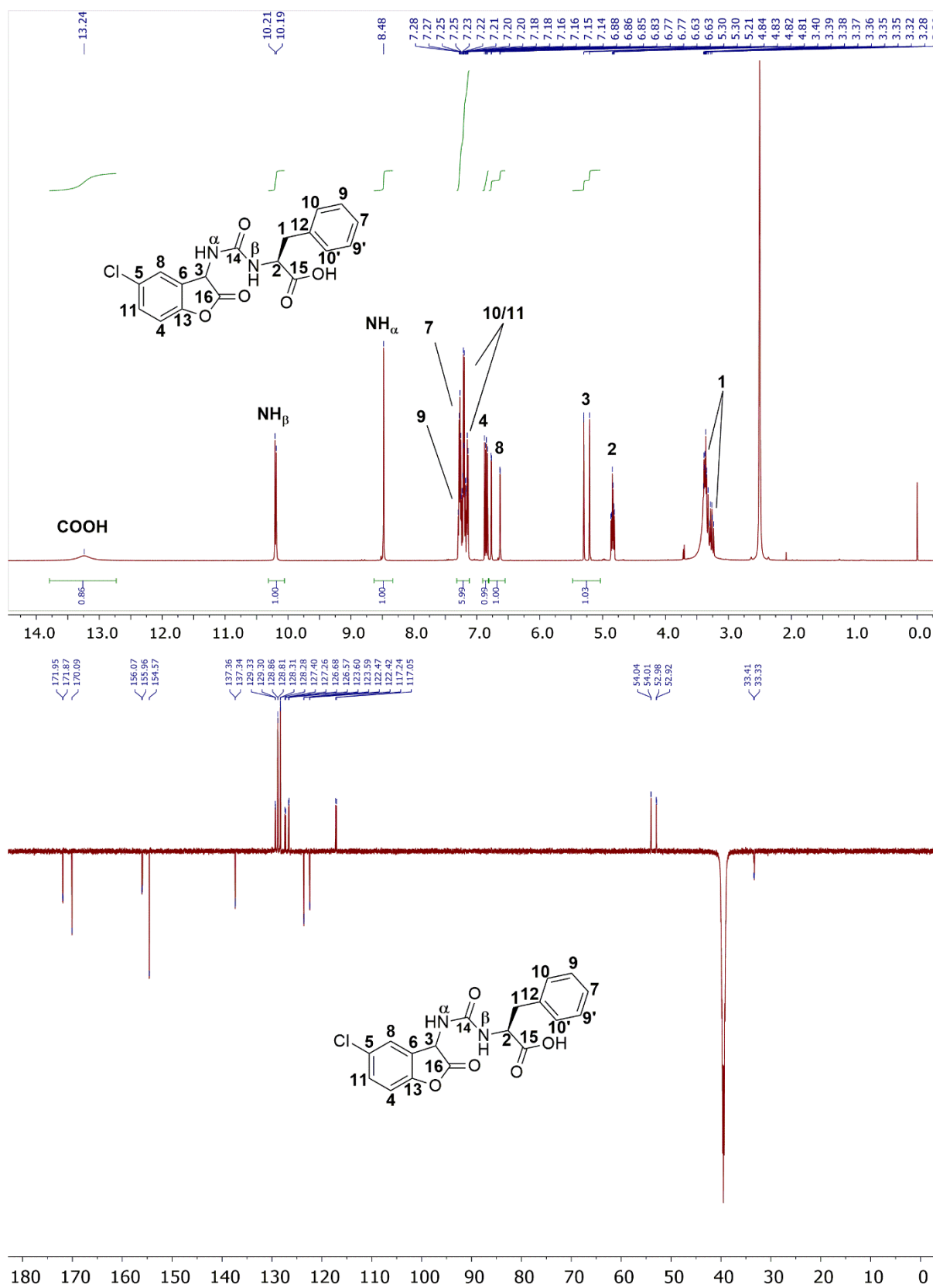
MP: Decomposition above 190 °C.

R_f: 0.25 (SiO_2 , 20:1 DCM:MeOH + 1% AcOH).

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Figure S1: NMR- and IR-spectra of 5A.

1.2 NMR and IR spectra of compound 5E

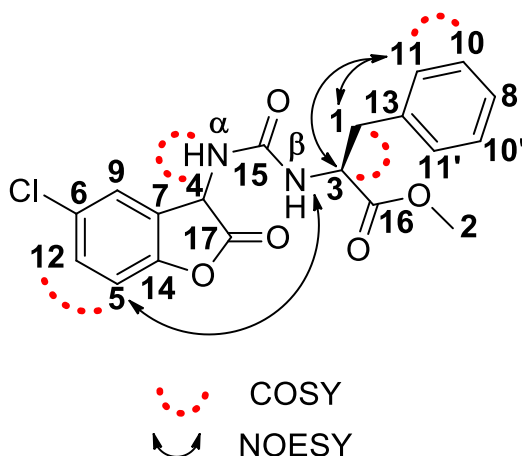


Table S2: 1D and 2D-NMR data of (2S)-methyl 2-(3-(5-chloro-2-oxo-2,3-dihydrobenzofuran-3-yl)ureido)-3-phenylpropanoate (5E) in DMSO- d_6 , at 298 K and 500 MHz for ^1H and 125 MHz for ^{13}C .

No.	δ_{H} [ppm], J in [Hz]	δ_{C} [ppm], mult.	HMBC ($^{\circ}\text{J}$)
1	3.21-3.32 (1H, m) 3.38+3.40 (1H, m)	33.45+33.41, CH_2	$\text{C1} \rightarrow \text{H3}^{(2)\text{J}}, \text{H11}^{(3)\text{J}}$
2	3.70+3.72 (3H, s)	52.62+52.56, CH_3	-
3	4.95-5.00 (1H, m)	52.66, CH	$\text{C3} \rightarrow \text{H1}^{(2)\text{J}},$
4	5.19+5.30 (1H, d, $J = 1.1$ Hz)	54.5+54.4, CH	$\text{C4} \rightarrow \text{H5}^{(4)\text{J}}, \text{H9}^{(3)\text{J}}, \text{NH}_{\alpha}^{(2)\text{J}}$
5	6.84+6.86 (1H, d, $J = 8.7$ Hz)	117.3+117.1, CH_{arom}	$\text{C5} \rightarrow \text{NH}_{\beta}^{(7)\text{J}}$
6		122.45+122.42, C_q	$\text{C6} \rightarrow \text{H5}^{(3)\text{J}}, \text{H9}^{(2)\text{J}}, \text{H12}^{(2)\text{J}}$
7		123.51+123.50, C_q	$\text{C7} \rightarrow \text{H4}^{(2)\text{J}}, \text{H5}^{(3)\text{J}}, \text{NH}_{\beta}^{(5)\text{J}}$
8	7.23 (1H, m)	126.8+126.7, CH_{arom}	$\text{C8} \rightarrow \text{H11}^{(3)\text{J}}$
9	6.67+6.82 (1H, d, $J = 2.7$ Hz)	127.8+127.6, CH_{arom}	$\text{C9} \rightarrow \text{H4}^{(3)\text{J}}, \text{H12}^{(3)\text{J}}$
10	7.27 (2H, m)	128.36+128.35, $2 \times \text{CH}_{\text{arom}}$	$\text{C10} \rightarrow \text{H10}^{(3)\text{J}}$
11	7.16+7.21 (2H, m)	128.97+128.93, $2 \times \text{CH}_{\text{arom}}$	$\text{C11} \rightarrow \text{H1}^{(3)\text{J}}, \text{H11}^{(3)\text{J}}$
12	7.21 (1H, m)	129.45+129.42, CH_{arom}	$\text{C12} \rightarrow \text{H9}^{(3)\text{J}}$
13	-	136.92+136.89, C_q	$\text{C13} \rightarrow \text{H1}^{(2)\text{J}}, \text{H10}^{(3)\text{J}}$
14	-	154.64+154.63, C_q	$\text{C14} \rightarrow \text{H9}^{(3)\text{J}}, \text{H12}^{(3)\text{J}}, \text{NH}_{\beta}^{(6)\text{J}}$
15	-	155.9+155.7, C_q	$\text{C15} \rightarrow \text{H3}^{(3)\text{J}}, \text{H4}^{(3)\text{J}}, \text{NH}_{\alpha}^{(2)\text{J}}$
16	-	169.14+169.07, C_q	$\text{C16} \rightarrow \text{H1}^{(3)\text{J}}, \text{H2}^{(3)\text{J}}, \text{H3}^{(2)\text{J}}$
17	-	171.9+171.8, C_q	$\text{C17} \rightarrow \text{H3}^{(6)\text{J}}, \text{H4}^{(2)\text{J}}, \text{NH}_{\alpha}^{(3)\text{J}}$
NH_{α}	8.52 (1H, s)		
NH_{β}	10.17+10.18 (1H, s)		

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Note: The ^1H - and ^{13}C -NMR-spectra show a mixture of rotamers.

Chemical Formula: $\text{C}_{19}\text{H}_{17}\text{ClN}_2\text{O}_5$

Molecular Weight: 388.80 g/mol

IR: $\tilde{\nu}$ [cm^{-1}] = 3325 (w), 2955 (w), 2330 (w), 1773 (w), 1705 (s), 1647 (w), 1635 (w), 1597 (w), 1556 (w), 1541 (w), 1497 (w), 1454 (w), 1430 (m), 1363 (w), 1326 (w), 1279 (m), 1249 (w), 1204 (w), 1175 (w), 1118 (w), 1032 (w), 996 (w), 967 (w), 960 (w), 950 (w), 907 (w), 879 (w), 816 (w).

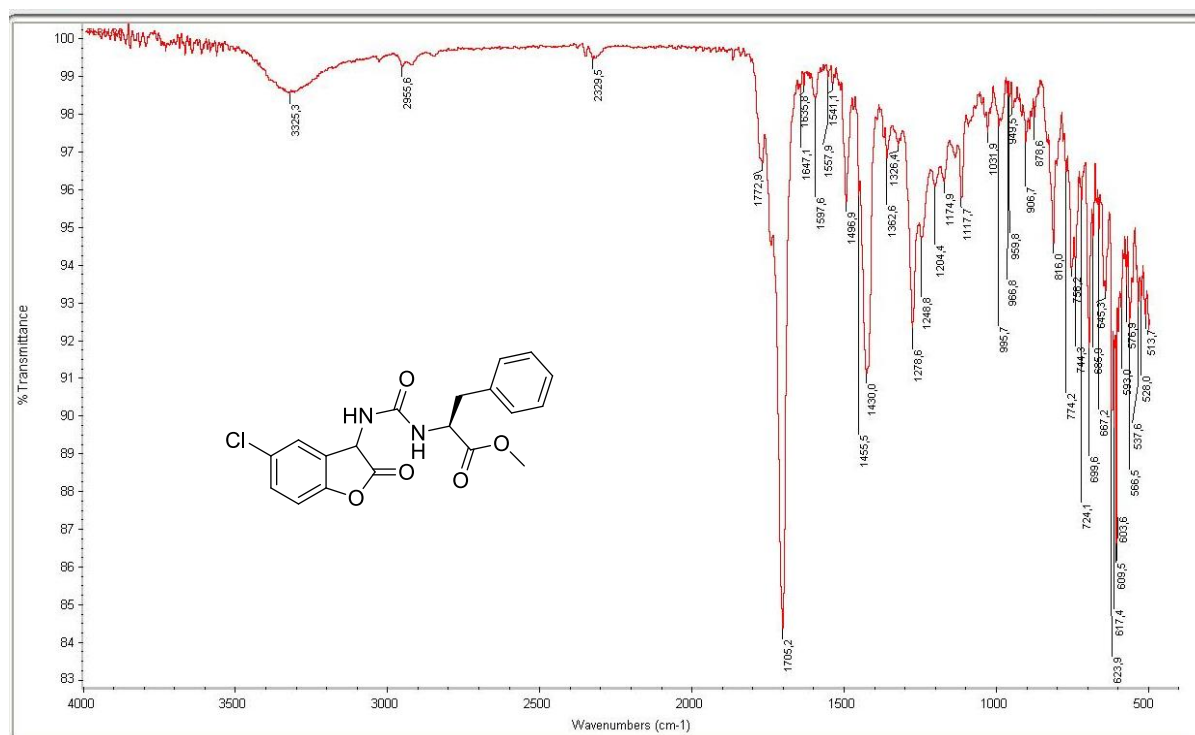
HR-MS: Theor. $[\text{M}+\text{H}]^+$: 398.08987, found: 397.08991.

Theor. $[\text{M}+\text{Na}]^+$: 411.07182, found: 411.07179.

MP: 206 °C.

R_f: 0.35 (SiO_2 , 20:1 DCM:MeOH).

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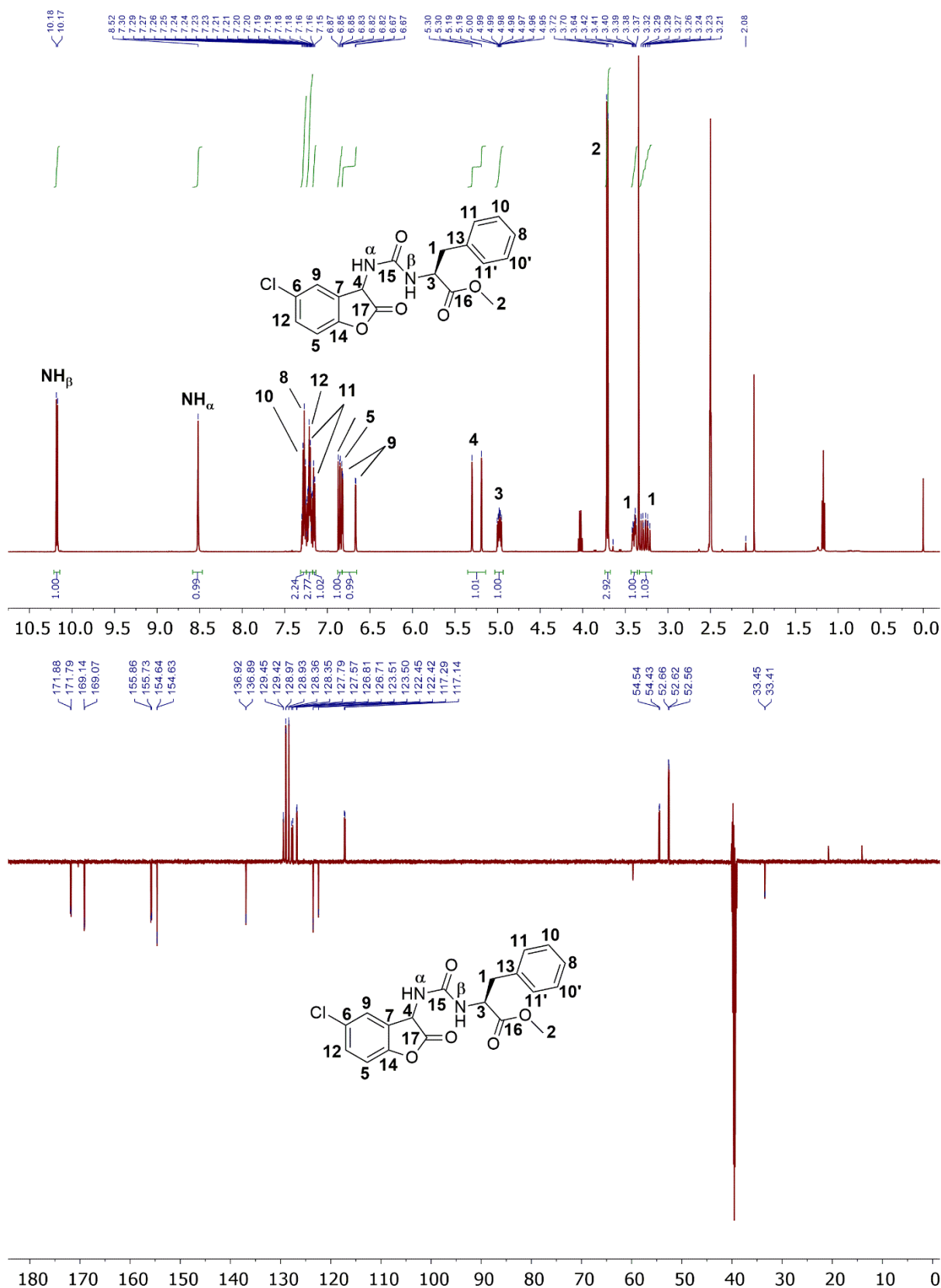


Figure S2: NMR- and IR-spectra of 5E.

1.3 NMR and IR spectra of compound 6

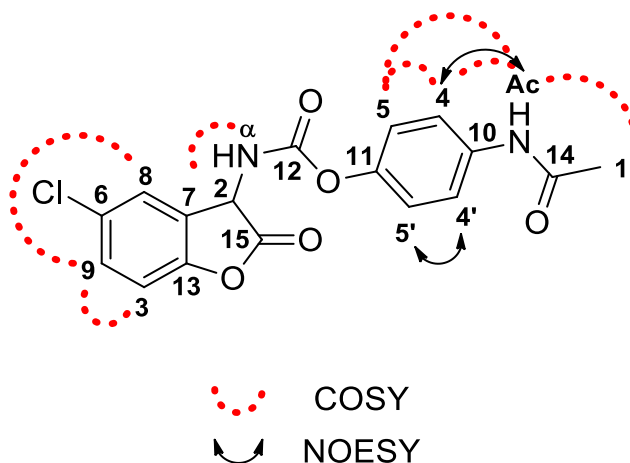


Table 3: 1D and 2D-NMR data of 4-acetamidophenyl (5-chloro-2-oxo-2,3-dihydrobenzofuran-3-yl)carbamate (6) in DMSO-d₆, at 298 K and 500 MHz for ¹H and 125 MHz for ¹³C.

No.	δ_H [ppm], J in [Hz]	δ_C [ppm], mult.	HMBC ($\times J$)
1	2.03 (3H, s)	23.9, CH ₃	-
2	5.37 (1H, d, J = 8.2 Hz)	53.0, CH	C2→H3(⁴ J), H8(³ J), NH _α (² J)
3	6.84 (1H, d, J = 8.7 Hz)	117.5, CH _{arom}	C3→H8(⁴ J)
4	7.56 (2H, d, J = 8.9 Hz)	119.8, 2xCH _{arom}	C4→H4(³ J), H5(² J), NH _{Ac} (³ J)
5	7.03 (2H, d, J = 9.0 Hz)	121.9, 2xCH _{arom}	C5→H4(² J), H5(³ J)
6	-	122.2, C _q	C6→H3(³ J), H8(² J), H9(² J)
7	-	126.7, C _q	C7→H2(² J), H3(³ J)
8	7.27 (1H, d, J = 2.7 Hz)	127.1, CH _{arom}	C8→H9(³ J)
9	7.16 (1H, dd, J = 8.7, 2.7 Hz)	128.3, CH _{arom}	C9→H8(³ J)
10	-	136.4, C _q	C10→H4(² J), H5(³ J), NH _{Ac} (² J)
11	-	146.2, C _q	C11→H4(³ J), H5(² J)
12	-	154.2, C _q	C12→H2(³ J), NH _α (² J)
13	-	154.4, C _q	C13→H3(² J), H8(³ J), H9(³ J)
14	-	168.2, C _q	C14→H1(² J), NH _{Ac} (² J)
15	-	171.71, C _q	C15→H2(² J)
NH _α	8.11 (1H, d, J = 8.2 Hz)	-	-
NH _{Ac}	9.97 (1H, s)	-	-

Chemical Formula: C₁₇H₁₃ClN₂O₅

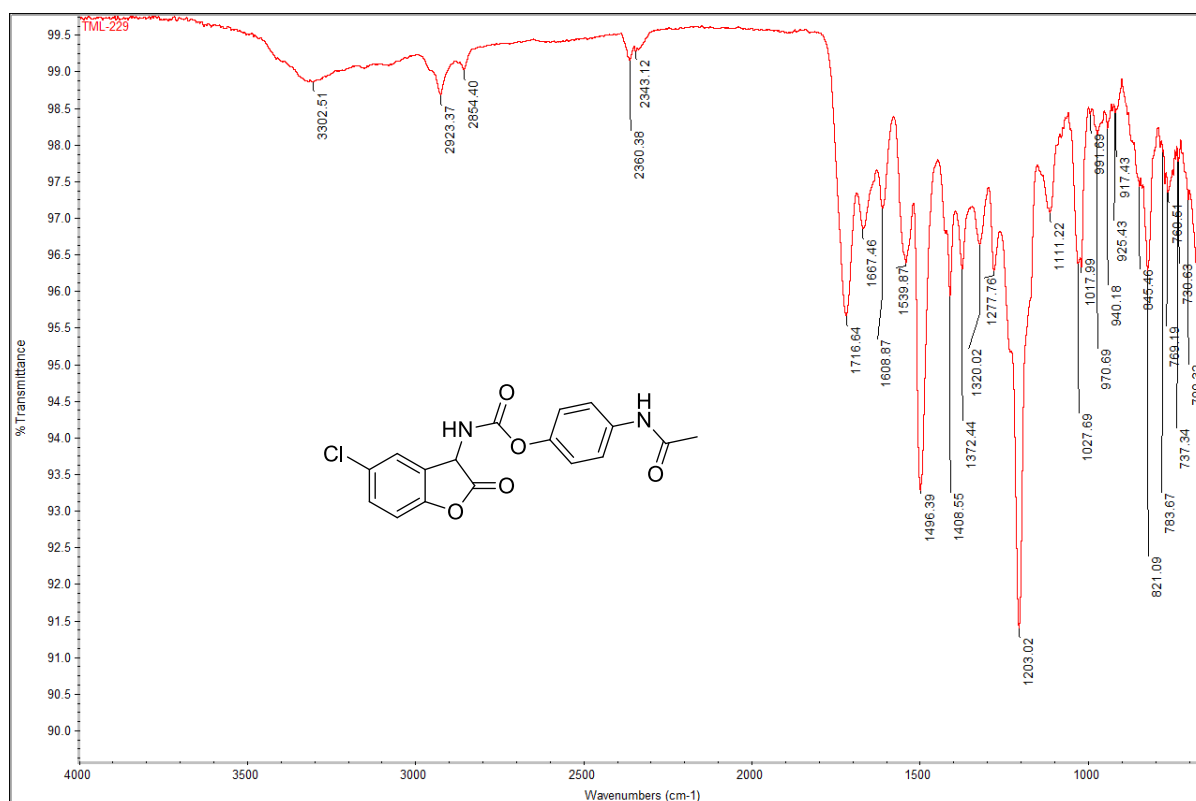
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Molecular Weight: 360.75 g/mol

IR: $\tilde{\nu}$ [cm⁻¹] = 3303 (br), 2923 (w), 2854 (w), 2360 (w), 2343 (w), 1717 (m), 1667 (w), 1609 (w), 1540 (w), 1496 (s), 1409 (w), 1372 (w), 1320 (w), 1278 (w), 1203 (vs), 1111 (w), 1028 (m), 1018 (m), 992 (w), 971 (w), 940 (w), 925 (w), 917 (w), 845 (w), 821 (m).

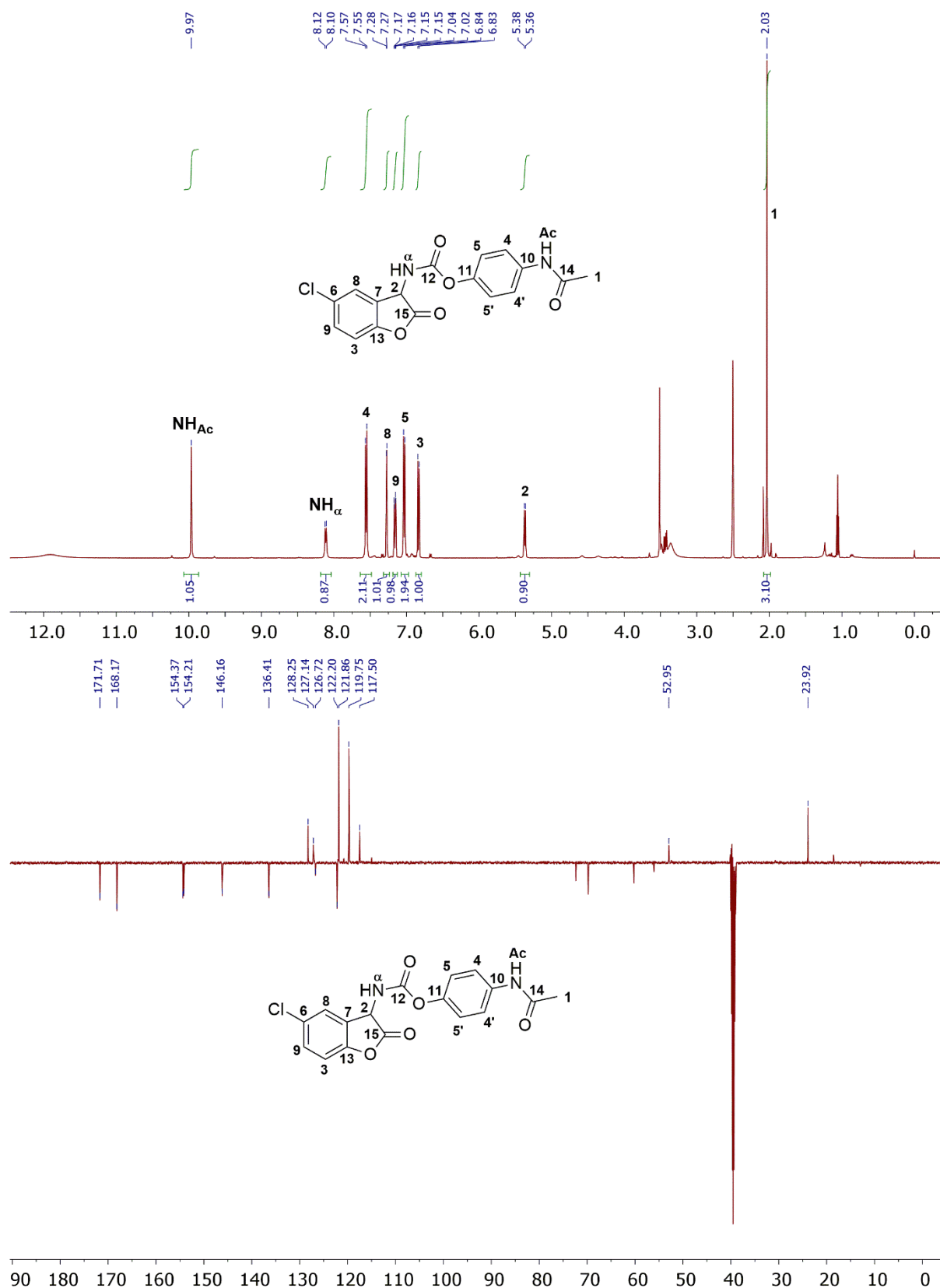
MP: Change of colour: 148 °C; Melting at 175 °C.

R_f: 0.10 (SiO₂, 8:1 DCM:MeOH).



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Figure S3: NMR- and IR-spectra of 6.

1.4 NMR and IR spectra of compound 8

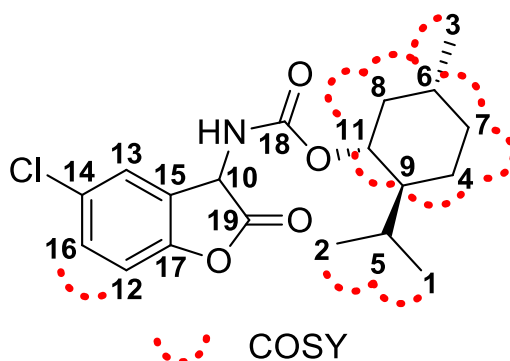


Table S4: 1D and 2D-NMR data of (1*R*,2*S*,5*R*)-2-isopropyl-5-methylcyclohexyl (5-chloro-2-oxo-2,3-dihydrobenzofuran-3-yl)carbamate (8) in CDCl₃, at 298 K and 500 MHz for ¹H and 125 MHz for ¹³C.

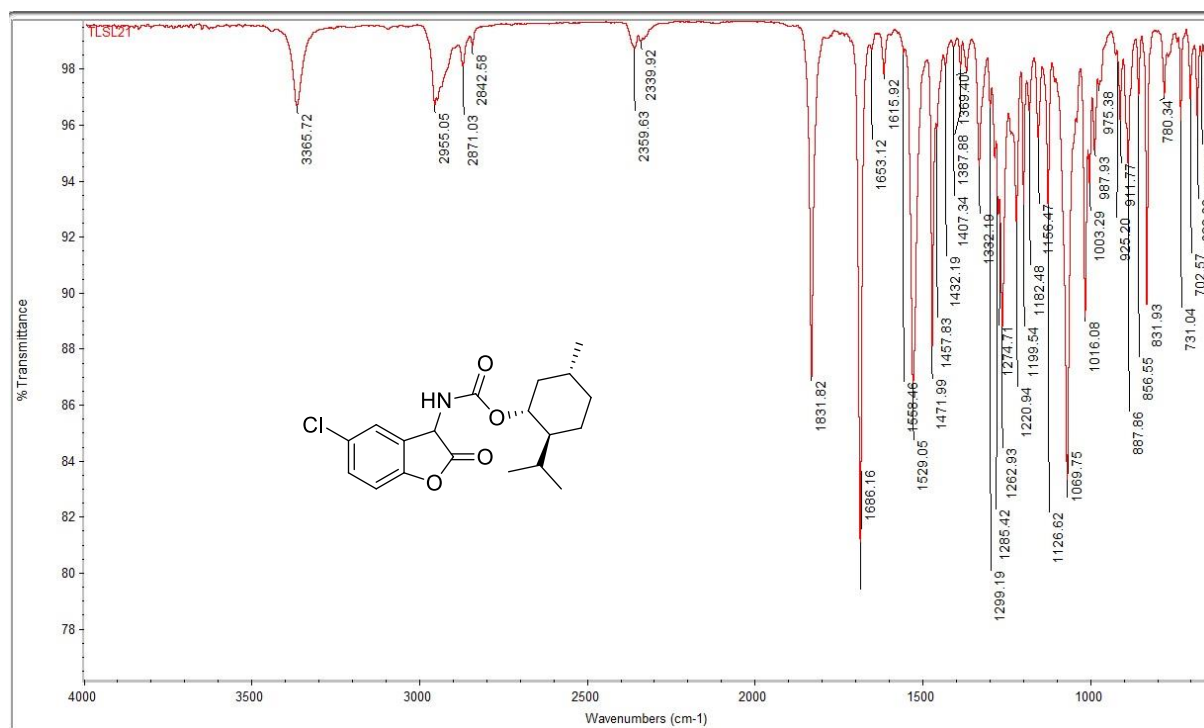
No.	δ_H [ppm], <i>J</i> in [Hz]	δ_C [ppm], mult.	HMBC ($\times J$)
1	0.74-0.80 (3H, s)	16.5, CH ₃	C1→H2(³ J)
2	0.90 (3H, m)	20.85+20.88, CH ₃	C2→H1(³ J)
3	0.89 (3H, m)	22.0, CH ₃	C3→H8(³ J)
4	1.04 (1H, m); 1.67 (1H, m)	23.7, CH ₂	-
5	1.89 (1H, m)	26.5, CH	C5→H1(² J), H2(² J)
6	1.43 (1H, m)	31.5, CH	C6→H3(² J), H7(² J), H8(² J)
7	0.84 (1H, m); 1.65 (1H, m)	34.3, CH ₂	C7→H3(³ J), H8(³ J)
8	0.98 (1H, m); 1.99 (1H, m)	41.2, CH ₂	C8→H3(³ J)
9	1.33 (1H, m)	47.4+47.5, CH	C9→H1(³ J), H2(³ J)
10	5.20-5.60 (1H, br)	52.4+52.8, CH	C10→H12(⁴ J), H13(³ J)
11	4.53 (1H, m)	76.9, CH	C11→H8(² J), H9(² J)
12	7.05 (1H, d, <i>J</i> = 8.3 Hz)	112.3+112.4, CH _{arom}	C12→H13(⁴ J)
13	7.34 (1H, m)	124.85+124.94, CH _{arom}	C13→H16(³ J)
14	-	126.6, C _q	C14→H12(³ J)
15	-	130.10+130.12, C _q	C15→H12(³ J), H16(⁴ J)
16	7.33 (1H, m)	130.4, CH _{arom}	C16→H13(³ J)
17	-	152.19+152.24, C _q	C17→H12(² J), H13(³ J), H16(³ J)
18	-	155.9, C _q	C18→H10(³ J)
19	-	172.9+173.0, C _q	C19→H10(² J)

Note: The ¹H- and ¹³C-NMR-spectra show a mixture of rotamers.

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<u>Chemical Formula:</u>	C ₁₉ H ₂₄ ClNO ₄
<u>Molecular Weight:</u>	365.85 g/mol
<u>IR:</u>	$\tilde{\nu}$ [cm ⁻¹] = 3366 (br), 2955 (w), 2871 (w), 2843 (w), 2360 (w), 2340 (w), 1832 (s), 1686 (vs), 1653 (w), 1616 (w), 1558 (w), 1529 (s), 1472 (s), 1458 (w), 1432 (w), 1388 (w), 1369 (w), 1332 (w), 1299 (w), 1285 (w), 1275 (w), 1263 (m), 1221 (w), 1200 (w), 1183 (w), 1157 (w), 1070 (s), 1016 (m), 1003 (w), 988 (w), 975 (w), 925 (w), 912 (w), 888 (w), 857 (w), 832 (m).
<u>HR-MS:</u>	Theor. [M+Na] ⁺ : 388.12860, found: 388.12883.
<u>Melting Point:</u>	172 °C.
<u>R_f:</u>	0.27 (SiO ₂ , 100:1 Toluene:MeOH + 1% AcOH).

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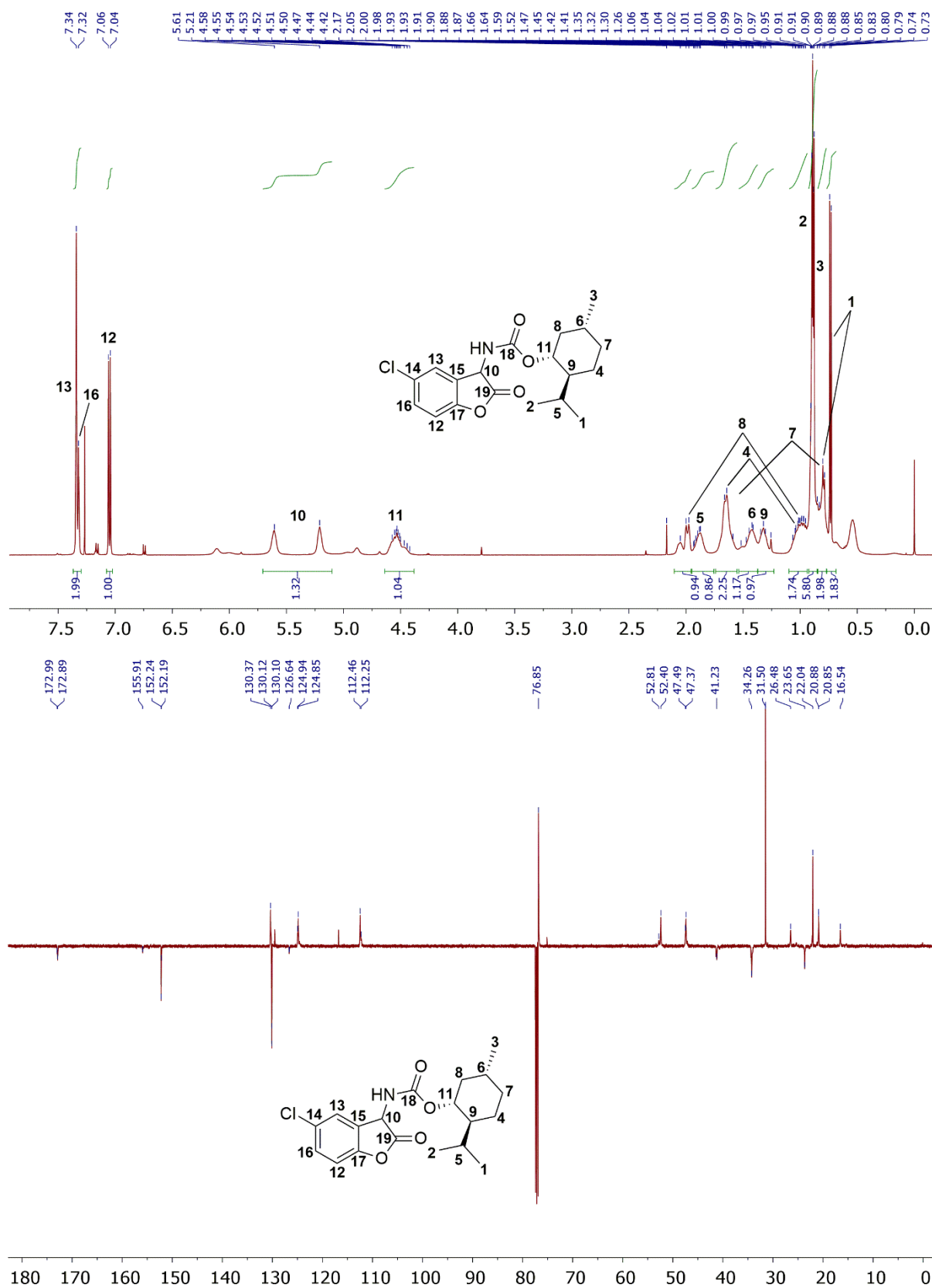
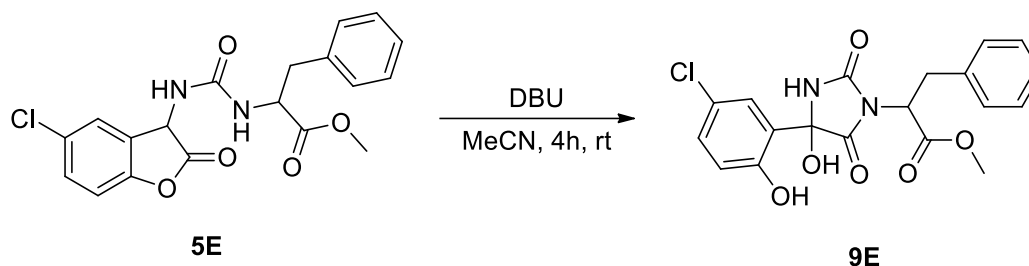
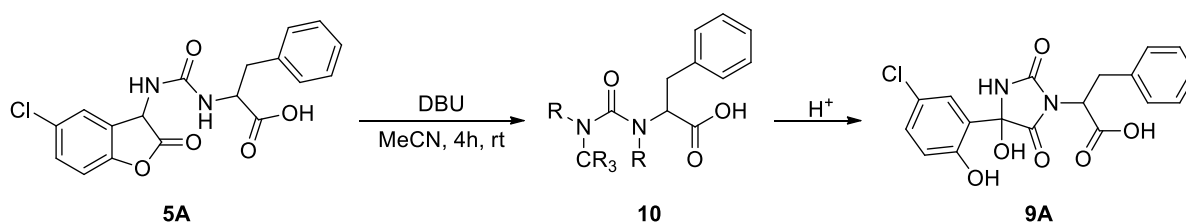


Figure S4: NMR- and IR-spectra of 8.

2. Decomposition Experiments

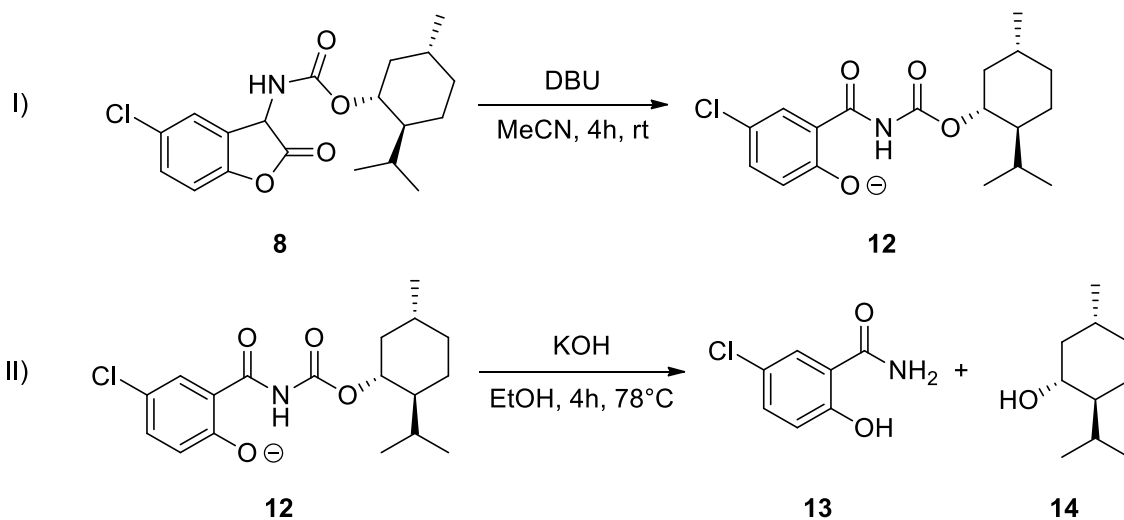


Decomposition experiment 5E-DE: 200 mg (0.51 mmol, 1.0 eq.) of **5E** were dissolved in 10.0 ml MeCN and 0.15 ml (1.03 mmol, 2.0 eq.) of DBU were added. At the begin of the addition a small flash of blue light could be seen and the clear mixture slowly turned orange and then dark brown. After 4 hours the solvent was removed under reduced pressure and 10 ml of water were added to the crude product. The aqueous phase was extracted three times with ethylacetate (20 ml per extraction) and the combined organic layers were washed four times with water and subsequently dried over sodium sulfate. Finally, the solvent was removed under reduced pressure and 50 mg of a brownish solid were obtained, which were analysed via NMR-spectroscopy and ESI-MS.



Decomposition experiment 5A-DE: 200 mg (0.53 mmol, 1.0 eq.) of **5A** were dissolved in 10.0 ml MeCN and 0.16 ml (1.07 mmol, 2.0 eq.) of DBU were added. At the begin of the addition a small flash of blue light could be seen and the clear

mixture slowly turned orange and then dark brown. After 4 hours the solvent was removed under reduced pressure. 10 ml of water and 1.0 ml of 2 M HCl solution were added to the crude product, which then changed its colour to orange. The aqueous phase was extracted three times with ethylacetate (20 ml per extraction) and the combined organic layers dried over sodium sulfate. Finally, the solvent was removed under reduced pressure. 22 mg of a brownish, orange solid was obtained, which were analysed via NMR.



Decomposition experiment 8-DE: 50 mg (0.14 mmol, 1.0 eq.) of **6** were dissolved in 5.0 ml MeCN and 42 μ l (0.28 mmol, 2.0 eq.) of DBU were added. Over a time period of 20 minutes a strong emission of blue light could be seen and the clear mixture slowly turned yellow, then orange and finally dark red. After 4 hours the solvent was removed under reduced pressure and 10 ml of water and ethylacetate were added to the crude product. The organic phase was washed four times with water and subsequently dried over sodium sulfate. Finally, the solvent was removed under reduced pressure and 45 mg of a colourless solid were obtained, which was analysed via NMR. After compound **12** was verified, it was dissolved in 5 ml EtOH and 24 mg

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KOH (0.42 mmol, 3.0 eq.) were added and the solution was refluxed for 4 hours. After that the solvent was removed under reduced pressure and 10 ml of water were added to the crude product. The aqueous phase was extracted three times with ethylacetate (15 ml per extraction) and the combined organic layers were washed with brine and subsequently dried over sodium sulfate. Finally, the solvent was removed under reduced pressure and 20 mg (0.11 mmol, 79% minus the contamination by 5-chlorosalicylamide **13**) of L-menthol **14** were obtained.

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2.1 NMR and ESI-spectra of 5E

Results from Experiment 5E-DE (Mainproduct)

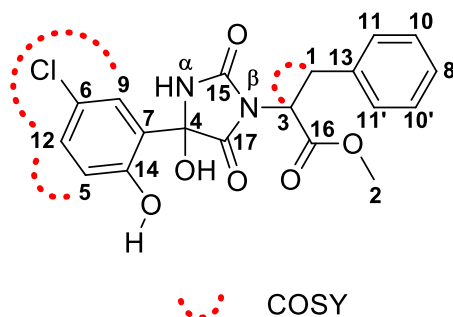


Table S4: 1D and 2D-NMR data of methyl 2-(4-(5-chloro-2-hydroxyphenyl)-4-hydroxy-2,5-dioxoimidazolidin-1-yl)-3-phenylpropanoate (9E) in DMSO- d_6 , at 298 K and 499 MHz for ^1H and 126 MHz for ^{13}C .

No.	δ_{H} [ppm], J in [Hz]	$\delta_{\text{C}}, \delta_{\text{N}}$ [ppm], mult.	HMBC ($\times J$)
1	3.04 (1H, dd, $J = 14.2, 1.6$ Hz) +3.45 (1H, dd, $J = 14.2, 7.2$ Hz)	33.38+33.61, CH_2	$\text{C1} \rightarrow \text{H3}^{(2J)}, \text{H11}^{(3J)}$
2	3.64 (3H, s)	52.30+52.32, CH_3	-
3	4.78 (1H, dd, $J = 7.8, 1.8$ Hz)	52.7, CH	$\text{C3} \rightarrow \text{H1}^{(2J)},$
4	-	81.76+81.92, C_q	$\text{C4} \rightarrow \text{H5}^{(4J)}, \text{H9}^{(3J)}, \text{NH}_\alpha^{(2J)}$
5	6.80 (1H, d, $J = 8.6$ Hz)	117.1, CH_{arom}	-
6	-	122.0, C_q	$\text{C6} \rightarrow \text{H5}^{(3J)}, \text{H9}^{(2J)}, \text{H12}^{(2J)}$
7	-	126.25+126.33, C_q	$\text{C7} \rightarrow \text{H5}^{(3J)}$
8	7.20 (1H, m)	126.5, CH_{arom}	$\text{C8} \rightarrow \text{H11}^{(3J)}$
9	7.49 (1H, d, $J = 2.6$ Hz)	127.37, CH_{arom}	$\text{C9} \rightarrow \text{H12}^{(3J)}$
10	7.27 (2H, m)	128.31+128.34, $2 \times \text{CH}_{\text{arom}}$	$\text{C10} \rightarrow \text{H10}^{(3J)}$
11	7.25 (2H, m)	129.02+129.07, $2 \times \text{CH}_{\text{arom}}$	$\text{C11} \rightarrow \text{H1}^{(3J)}, \text{H8}^{(3J)}, \text{H11}^{(3J)}$
12	7.24 (1H, m)	129.5, CH_{arom}	$\text{C12} \rightarrow \text{H9}^{(3J)}$
13	-	137.7, C_q	$\text{C13} \rightarrow \text{H1}^{(2J)}, \text{H10}^{(3J)}$
14	-	153.48+153.54, C_q	$\text{C14} \rightarrow \text{H5}^{(2J)}, \text{H9}^{(3J)}, \text{H12}^{(3J)}$
15	-	154.93+154.97, C_q	$\text{C15} \rightarrow \text{H3}^{(3J)}, \text{NH}_\alpha^{(2J)}$
16	-	168.92+169.04, C_q	$\text{C16} \rightarrow \text{H1}^{(3J)}, \text{H2}^{(3J)}, \text{H3}^{(2J)}$
17	-	172.16+172.29, C_q	$\text{C17} \rightarrow \text{H3}^{(3J)}, \text{NH}_\alpha^{(3J)}$
NH_α	8.86+8.83 (1H, s)	112.6, NH	-
N_β	-	147.8, NR_3	$\text{N}_\beta \rightarrow \text{H1}^{(3J)}, \text{H3}^{(2J)}, \text{NH}_\alpha^{(3J)}$
Ph-OH	10.13 (1H, br)	-	-

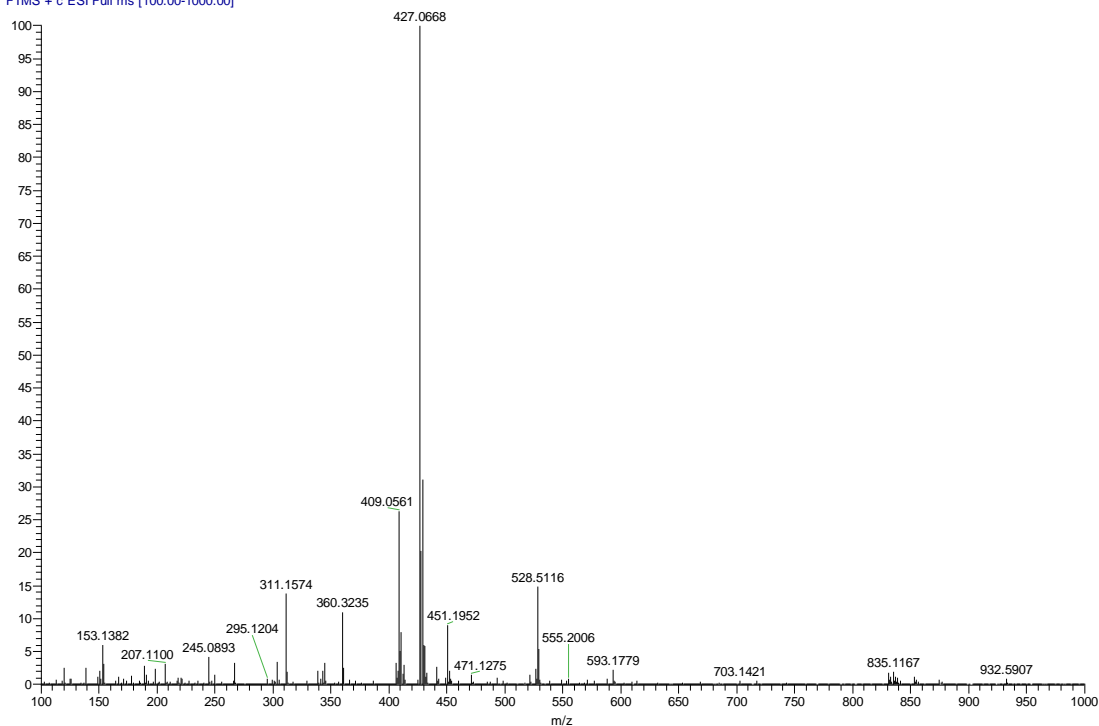
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Chemical Formula: C₁₉H₁₇ClN₂O₆

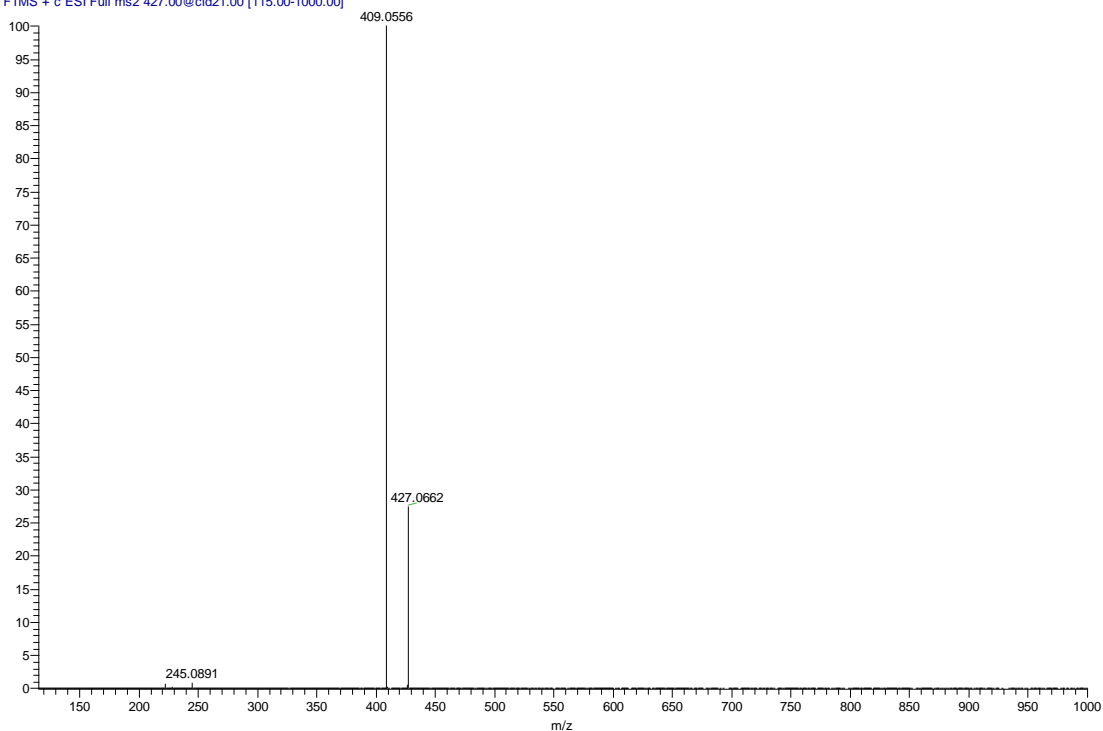
Molecular Weight: 404.80 g/mol

MS (ESI): m/z (%) 427.0668 [M+Na⁺] (27), 409.556 [M+Na⁺-H₂O] (100).

TML_284b #1-60 RT: 0.01-0.47 AV: 60 NL: 1.45E7
T: FTMS + c ESI Full ms [100.00-1000.00]



TML_284b@CID #1-58 RT: 0.01-0.64 AV: 58 NL: 4.07E6
T: FTMS + c ESI Full ms2 427.00@cid21.00 [115.00-1000.00]



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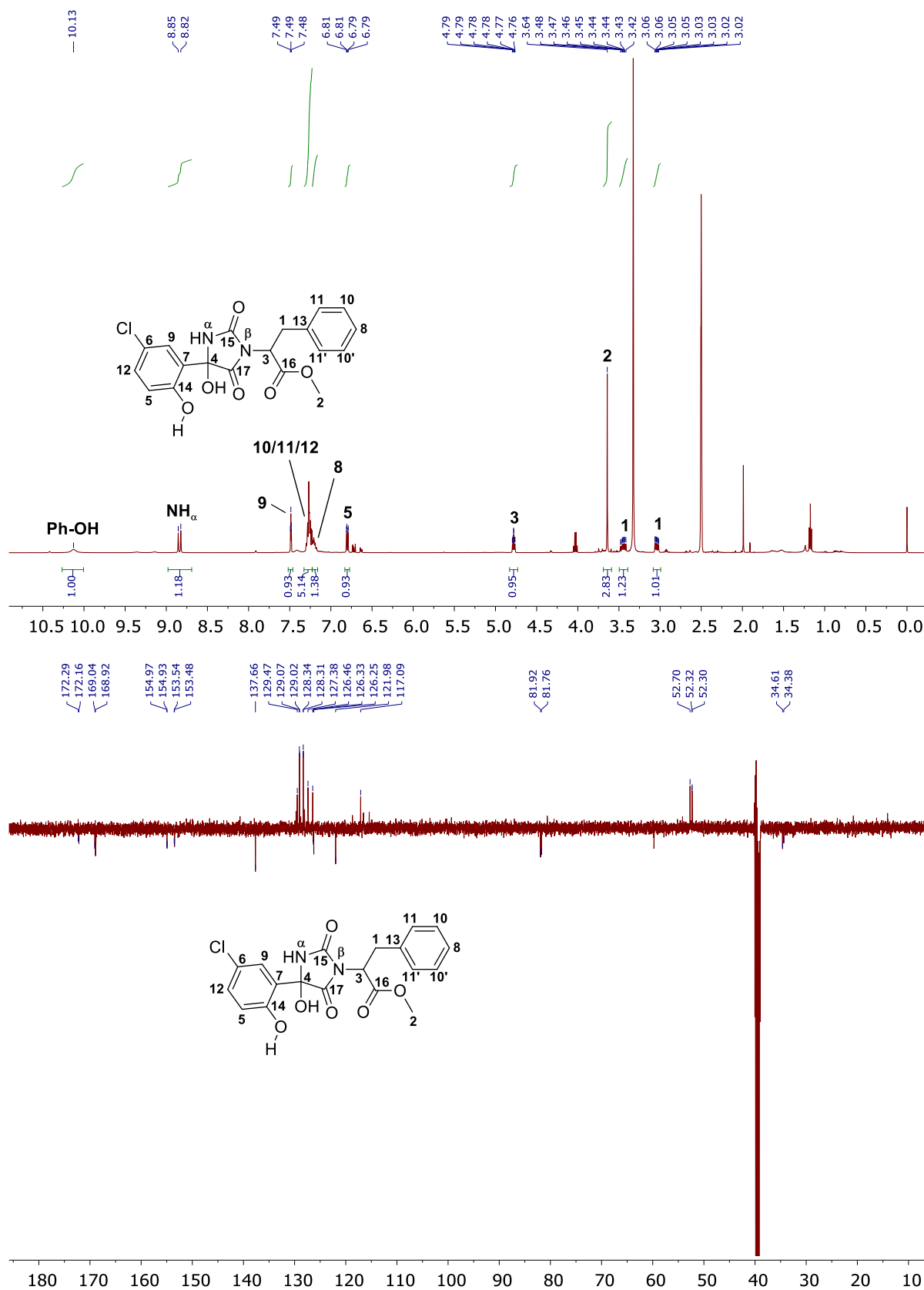


Figure S5: NMR- and ESI-spectra of **9E**.

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2.2 NMR-spectra of 5A

Results from Experiment 5A-DE (Product after protonation)

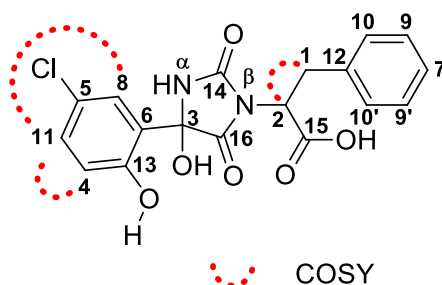


Table S6: 1D and 2D-NMR data of 2-(4-(5-chloro-2-hydroxyphenyl)-4-hydroxy-2,5-dioxoimidazolidin-1-yl)-3-phenylpropanoic acid (9A) in DMSO- d_6 , at 298 K and 499 MHz for ^1H and 126 MHz for ^{13}C .

No.	δ_{H} [ppm], J in [Hz]	δ_{C} [ppm], mult.	HMBC ($^{\circ}\text{J}$)
1	3.02 ($\frac{1}{2}$ H, dd, $J = 14.2, 6.6$ Hz)/3.18 ($\frac{1}{2}$ H dd, $J = 14.2, 10.8$ Hz) + 3.44 (1H, dd, $J = 14.2, 5.6$ Hz)	33.79+34.63, CH_2	$\text{C1} \rightarrow \text{H2}(^{\circ}\text{J}), \text{H10}(^{\circ}\text{J})$
2	4.67 ($\frac{1}{2}$ H, m) + 5.01 ($\frac{1}{2}$ H, dd, $J = 10.8, 5.2$ Hz)	52.92/53.00+53.76, CH	$\text{C2} \rightarrow \text{H1}(^{\circ}\text{J})$
3	-	81.72+82.17, C_q	$\text{C3} \rightarrow \text{H4}(^{\circ}\text{J}), \text{H8}(^{\circ}\text{J}), \text{NH}_{\alpha}(^{\circ}\text{J})$
4	6.80 (1H, d, $J = 8.6$ Hz)	117.2, CH_{arom}	-
5	-	122.0, C_q	$\text{C5} \rightarrow \text{H4}(^{\circ}\text{J}), \text{H8}(^{\circ}\text{J}), \text{H11}(^{\circ}\text{J})$
6	-	126.26+126.37, C_q	$\text{C6} \rightarrow \text{H4}(^{\circ}\text{J}), \text{Ph-OH}(^{\circ}\text{J})$
7	7.20 (1H, m)	126.32, CH_{arom}	$\text{C7} \rightarrow \text{H10}(^{\circ}\text{J})$
8	7.46 (1H, d, $J = 2.7$ Hz)	126.8+127.37/127.39, CH_{arom}	$\text{C8} \rightarrow \text{H11}(^{\circ}\text{J})$
9	7.19 (1H, m) + 7.26 (1H, m)	128.27/128.30+128.47, $2 \times \text{CH}_{\text{arom}}$	$\text{C9} \rightarrow \text{H9}(^{\circ}\text{J})$
10	7.19 (1H, m) + 7.25 (1H, m)	128.8+128.95/128.99, $2 \times \text{CH}_{\text{arom}}$	$\text{C10} \rightarrow \text{H1}(^{\circ}\text{J}), \text{H7}(^{\circ}\text{J}), \text{H10}(^{\circ}\text{J})$
11	7.23 (1H, m)	129.4, CH_{arom}	$\text{C11} \rightarrow \text{H8}(^{\circ}\text{J})$
12	-	137.0+138.30/138.31, C_q	$\text{C12} \rightarrow \text{H1}(^{\circ}\text{J}), \text{H2}(^{\circ}\text{J}), \text{H9}(^{\circ}\text{J})$
13	-	153.39+153.58/153.66, C_q	$\text{C13} \rightarrow \text{H4}(^{\circ}\text{J}), \text{H8}(^{\circ}\text{J}), \text{H11}(^{\circ}\text{J})$
14	-	155.11+155.15, C_q	$\text{C14} \rightarrow \text{H2}(^{\circ}\text{J}), \text{NH}_{\alpha}(^{\circ}\text{J})$
15	-	169.14+169.94/169.96, C_q	$\text{C15} \rightarrow \text{H1}(^{\circ}\text{J}), \text{H2}(^{\circ}\text{J})$
16	-	172.23+172.34, C_q	$\text{C16} \rightarrow \text{H2}(^{\circ}\text{J}), \text{NH}_{\alpha}(^{\circ}\text{J})$
NH_{α}	8.78+8.83 (1H, s)	-	-
N_{β}	-	-	-
Ph-OH	10.14 +10.16(1H, br)	-	-
COOH	12.95 (1H, br)	-	-

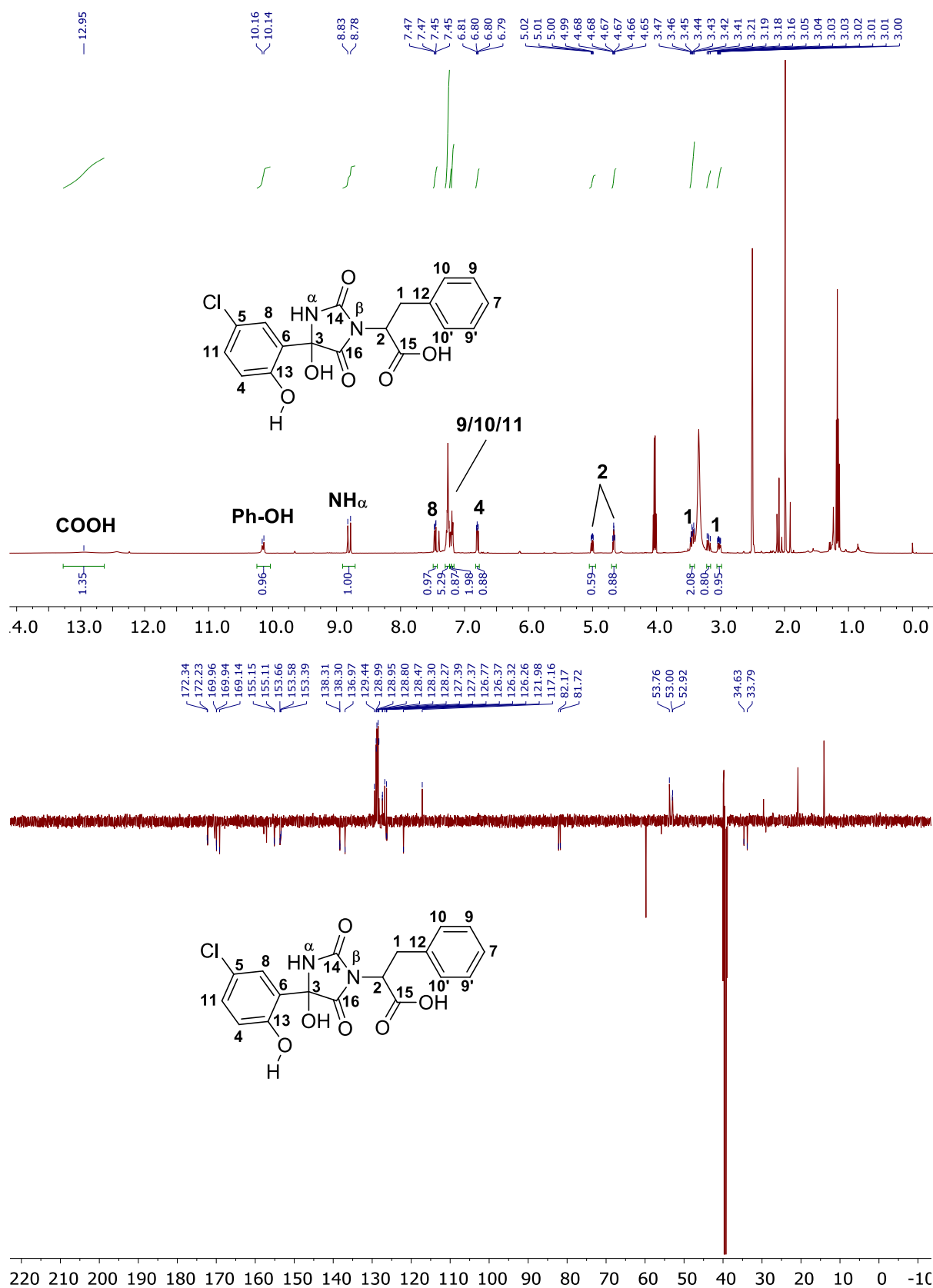
Chemical Formula:



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Molecular Weight: 390.06g/mol

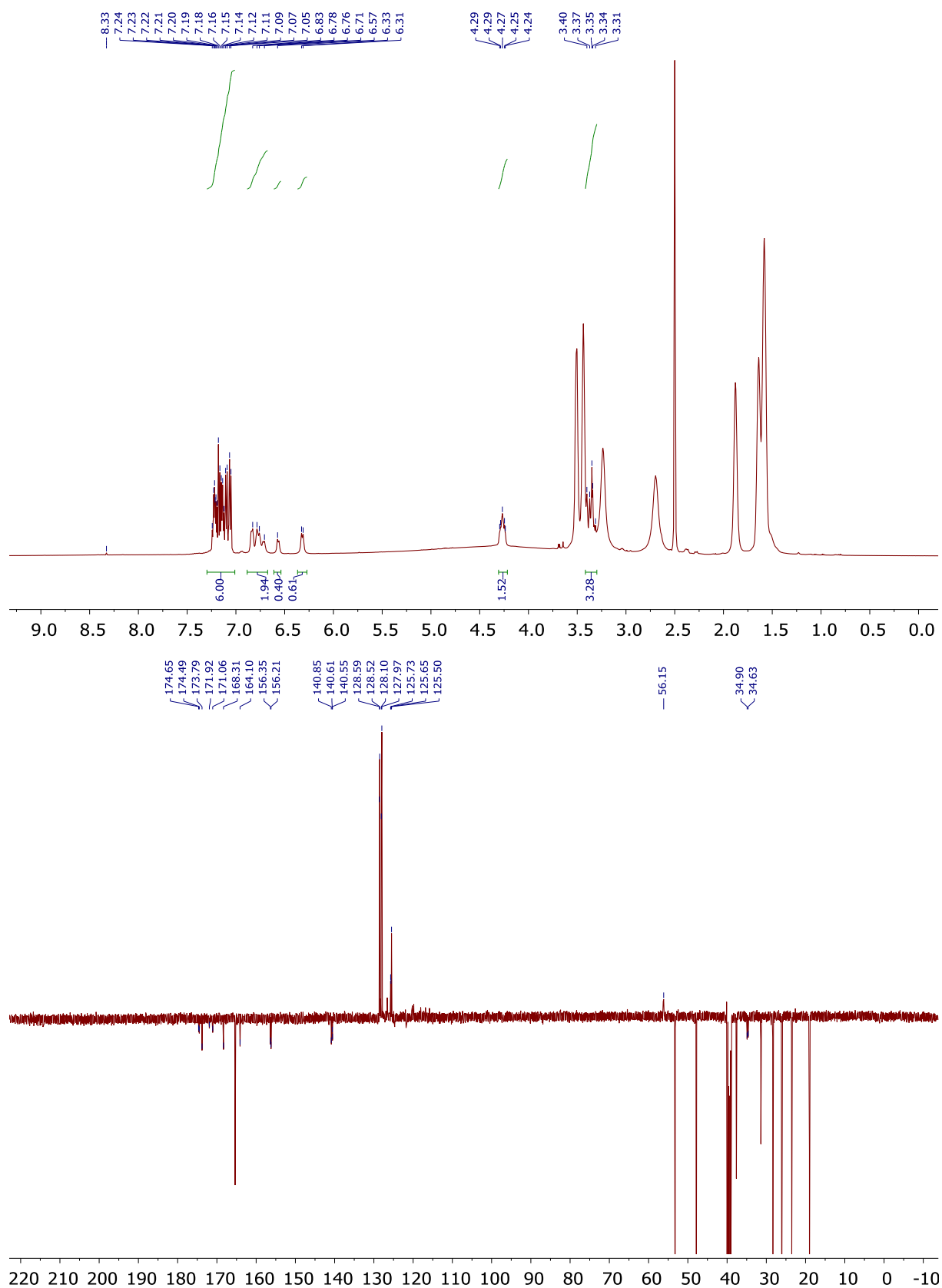
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Figure S6: NMR-spectra of **9A**.

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Figure S7: NMR-spectra of 5A-DE before protonation (Contaminated with DBU).

2.3 NMR-spectra of 12

Results from Experiment 8-DE (Product after 1 hour of oxidation)

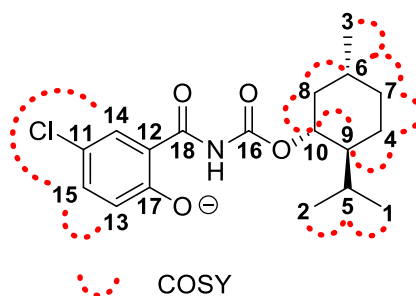


Table S5: 1D and 2D-NMR data of 4-chloro-2-((((1R,2S,5R)-2-isopropyl-5-methylcyclohexyl)oxy)carbonyl)carbamoylphenolate (12) in DMSO- d_6 , at 298 K and 500 MHz for ^1H and 125 MHz for ^{13}C .

No.	δ_{H} [ppm], J in [Hz]	$\delta_{\text{C}}, \delta_{\text{N}}$ [ppm], mult.	HMBC ($^{\circ}J$)
1	0.75 (3H, d, $J = 7.0$ Hz)	16.3, CH_3	$\text{C1} \rightarrow \text{H2}(^3J), \text{H5}(^2J)$
2	0.87 (3H, m)	20.5, CH_3	$\text{C2} \rightarrow \text{H1}(^3J), \text{H5}(^2J)$
3	0.89 (3H, m)	22.0, CH_3	$\text{C3} \rightarrow \text{H8}(^3J)$
4	1.04 (1H, m); 1.62 (1H, m)	23.1, CH_2	$\text{C4} \rightarrow \text{H9}(^2J)$
5	1.90 (1H, m)	25.8, CH	$\text{C5} \rightarrow \text{H1}(^2J), \text{H2}(^2J), \text{H10}(^3J)$
6	1.45 (1H, m)	30.9, CH	$\text{C6} \rightarrow \text{H3}(^2J), \text{H8}(^2J)$
7	0.84 (1H, m); 1.65 (1H, m)	33.8, CH_2	$\text{C7} \rightarrow \text{H3}(^3J), \text{H8}(^3J)$
8	0.98 (1H, m); 1.92 (1H, m)	40.9, CH_2	$\text{C8} \rightarrow \text{H3}(^3J), \text{H10}(^2J)$
9	1.34 (1H, m)	46.6, CH	$\text{C9} \rightarrow \text{H1}(^3J), \text{H2}(^3J), \text{H4}(^2J), \text{H5}(^2J), \text{H7}(^3J), \text{H8}(^3J), \text{H10}(^2J)$
10	4.50 (1H, td, $J = 10.9, 4.3$ Hz)	72.9, CH	$\text{C10} \rightarrow \text{H4}(^3J), \text{H8}(^2J), \text{H9}(^2J)$
11	-	112.5, C_q	$\text{C11} \rightarrow \text{H13}(^3J), \text{H14}(^2J), \text{H15}(^2J)$
12	-	117.6, C_q	$\text{C12} \rightarrow \text{H13}(^3J)$
13	6.34 (1H, d, $J = 8.9$ Hz)	124.0, CH_{arom}	-
14	7.50 (1H, d, $J = 3.1$ Hz)	128.2, CH_{arom}	$\text{C14} \rightarrow \text{H15}(^3J)$
15	6.92 (1H, dd, $J = 9.0, 3.1$ Hz)	132.5, CH_{arom}	$\text{C15} \rightarrow \text{H14}(^3J)$
16	-	151.8, C_q	$\text{C16} \rightarrow \text{H10}(^3J)$
17	-	165.9, C_q	$\text{C17} \rightarrow \text{H13}(^2J), \text{H14}(^3J)$
18	-	170.8, C_q	$\text{C18} \rightarrow \text{H13}(^4J), \text{H14}(^3J), \text{H15}(^3J)$
NH	15.91 (1H, s)	139.9, NH	-

Chemical Formula: $\text{C}_{18}\text{H}_{23}\text{ClNO}_4^-$

Molecular Weight: 352.83 g/mol

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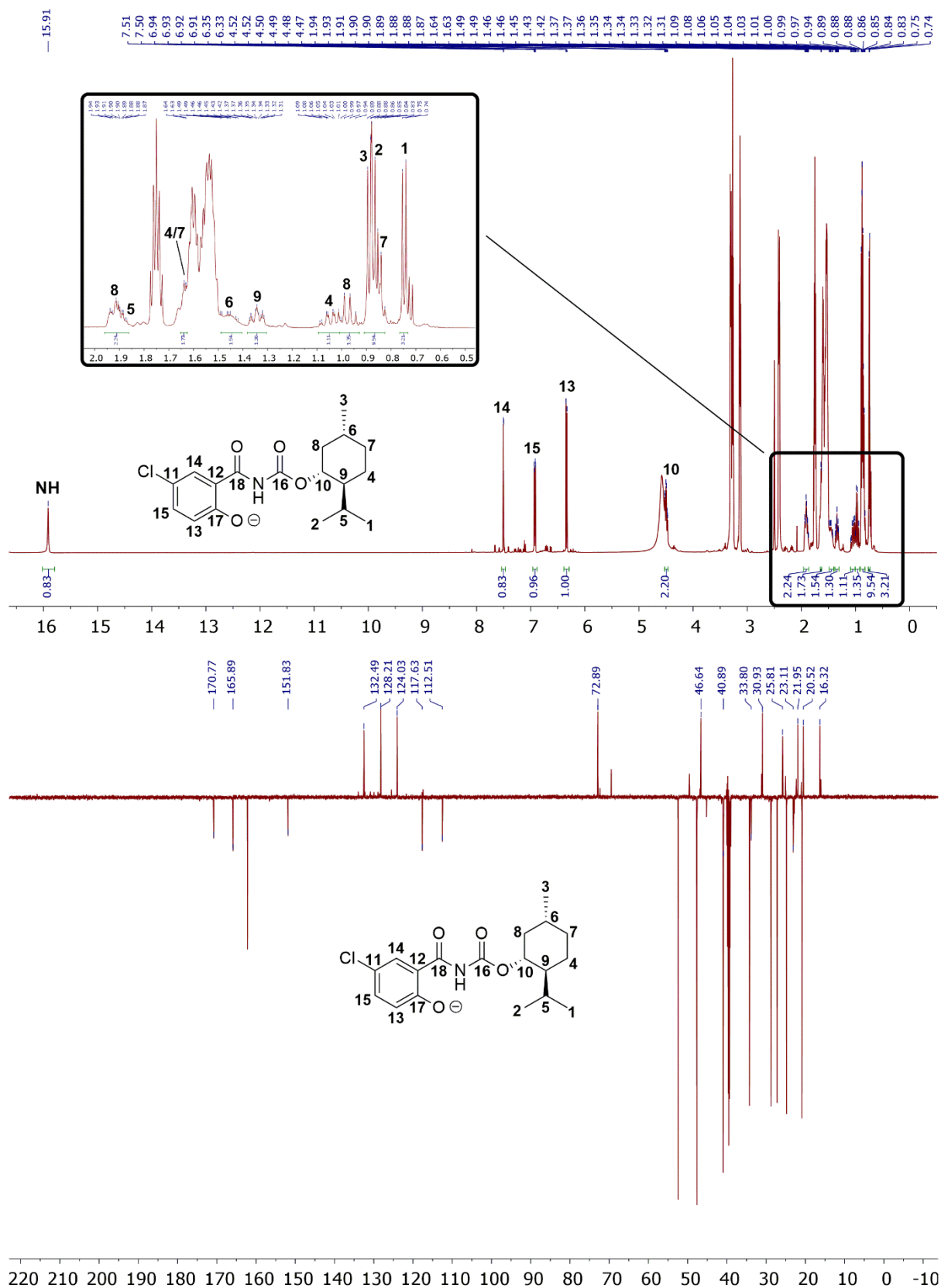


Figure S8: NMR-spectra of **12** (Contaminated with DBU).