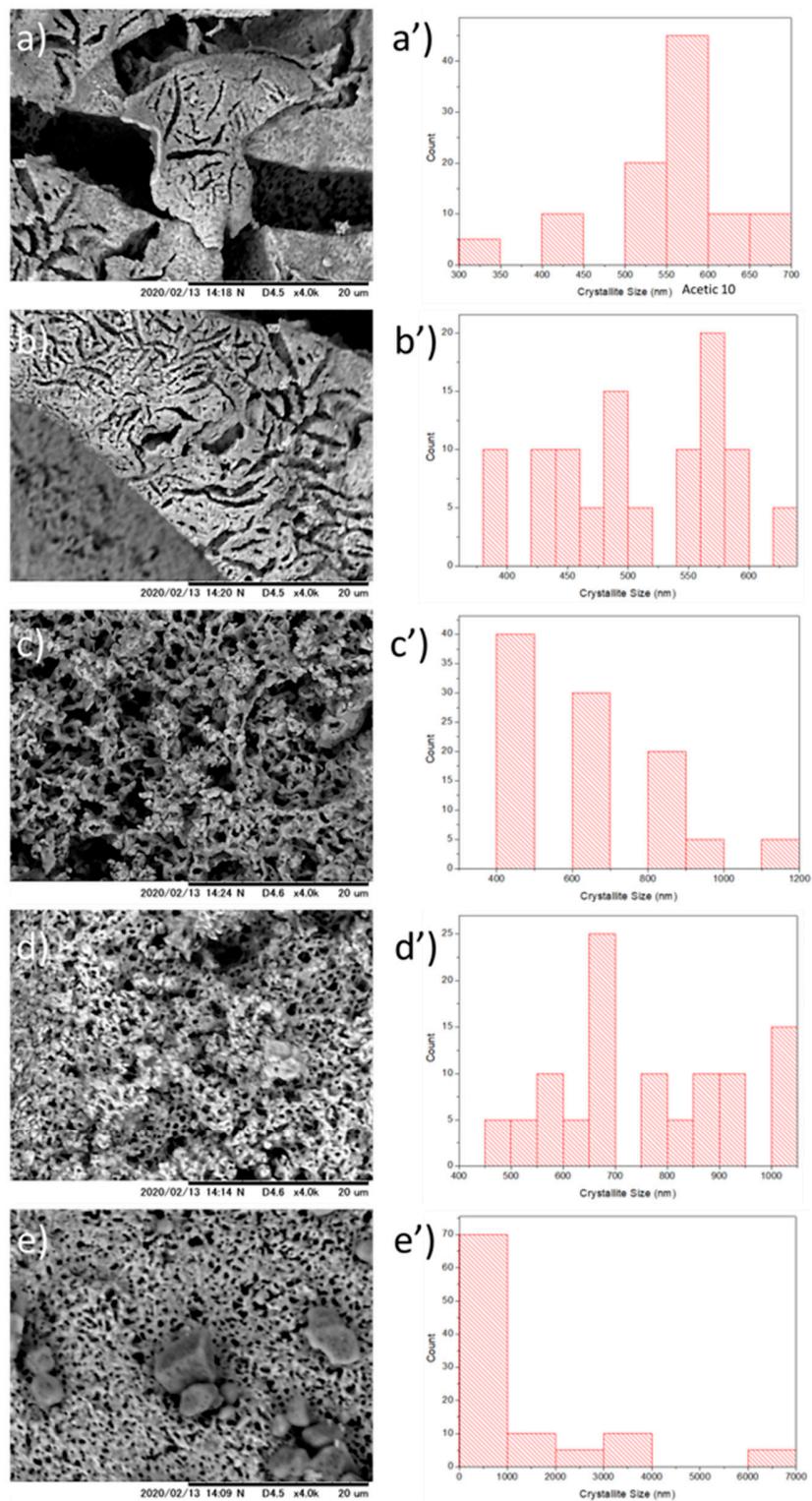
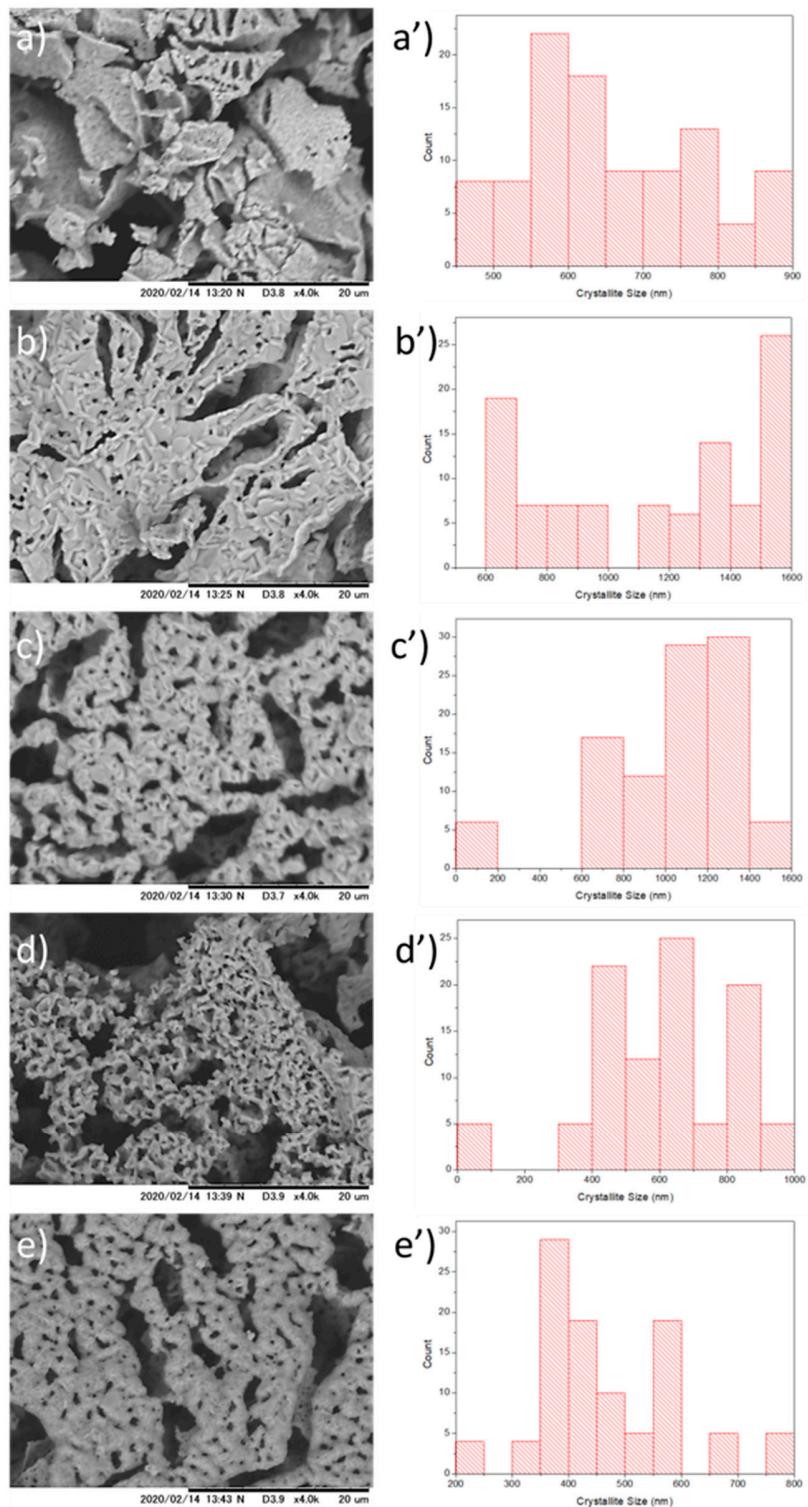


15301	BaO
26961	BaO
68217	Ba <sub>2</sub> CuO <sub>3</sub>
65801	BaCuO <sub>2</sub>
202998	Ba <sub>0.98</sub> CuO <sub>2.07</sub>
69757	CuO
67860	YBa <sub>2</sub> Cu <sub>2.78</sub> O <sub>7</sub>
32707	Y <sub>2</sub> BaCuO <sub>5</sub>
65550	Y <sub>0.5</sub> Ba <sub>3</sub> Cu <sub>1.5</sub> O <sub>5.5</sub>

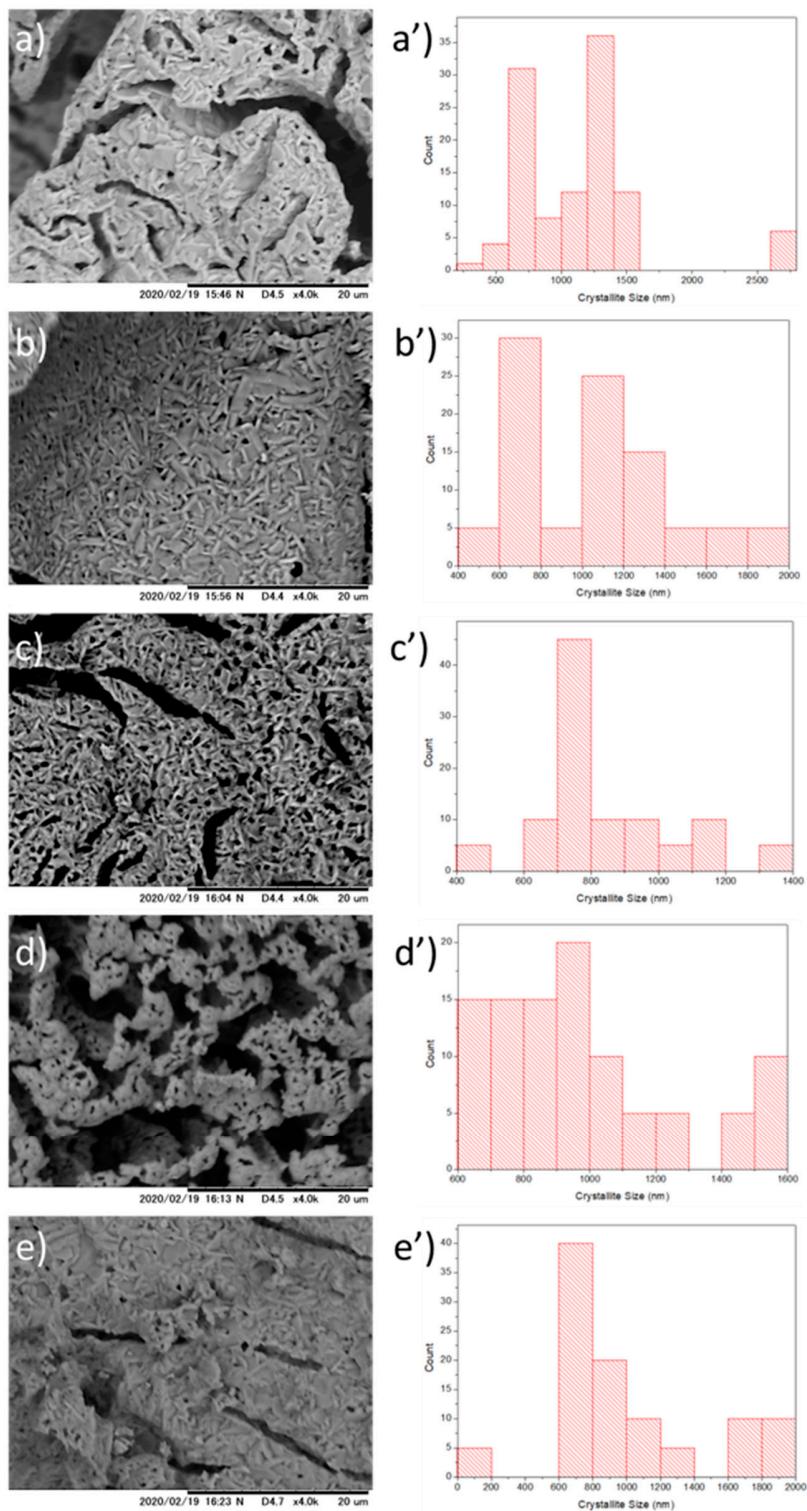
**Table S1.** ICSD numbers (left) with the corresponding crystal phases (right)



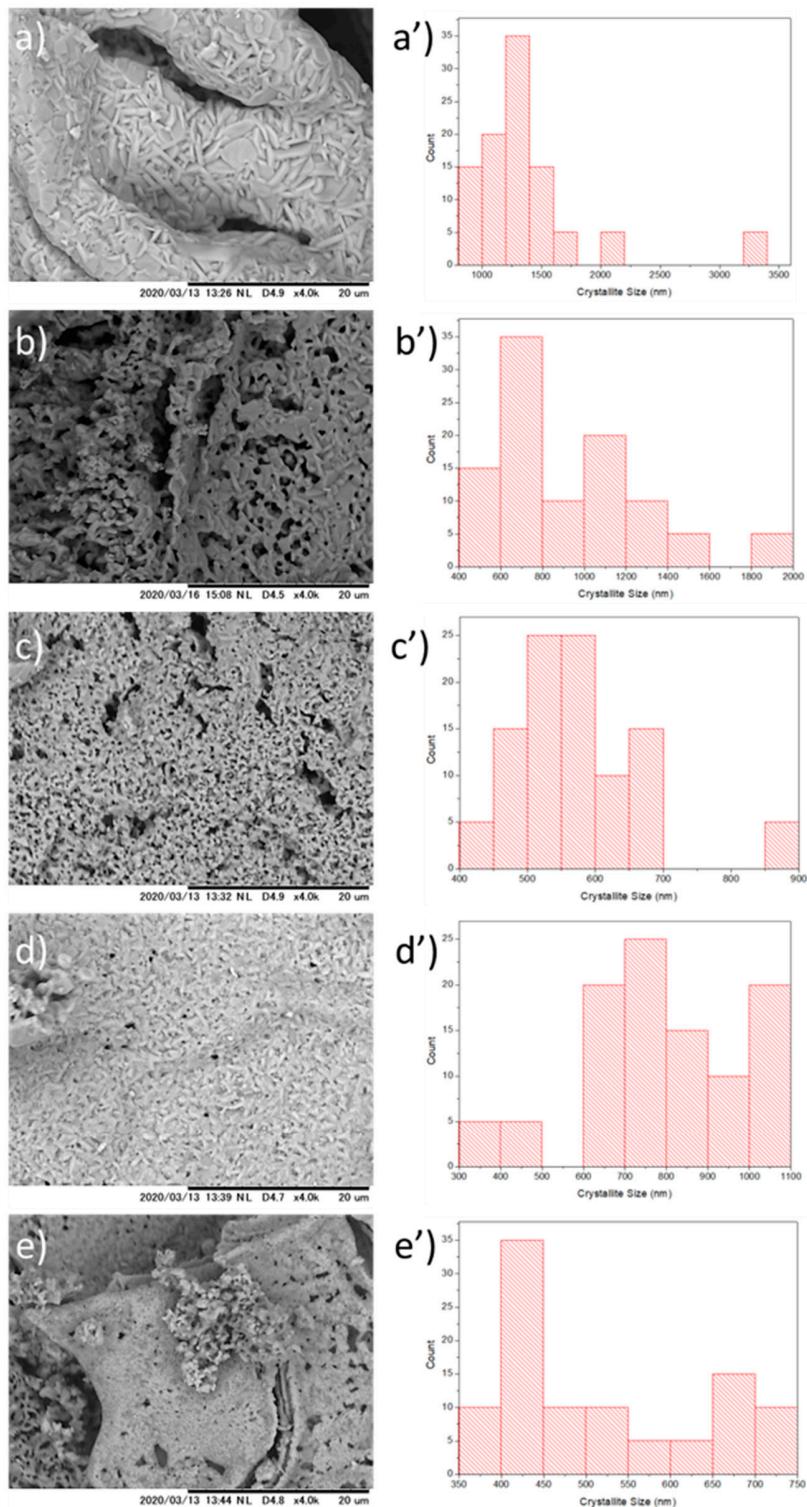
**Figure S1.** a-e) SEM micrographs showing the morphology exhibited by the crystallites, synthesized using choline hydroxide 1:1 acetic acid and molar ratios of DES (X):1 YBaCu metal nitrates molar ratios use in the reaction, with X values  $20 \leq X \leq 60$ , and X values increasing in tens. a'-e') Histograms showing crystallite size distributions of their respective SEM micrographs.



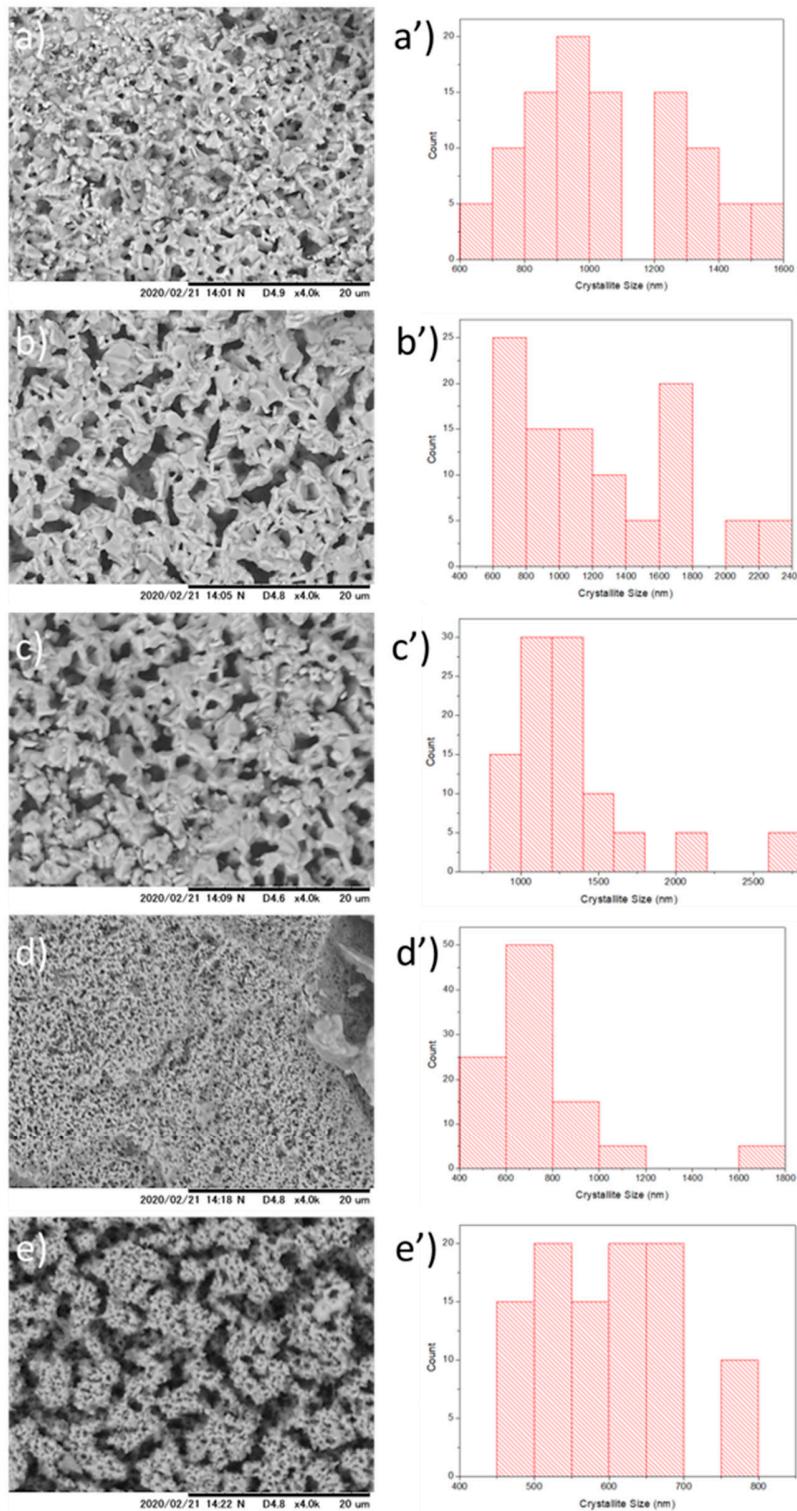
**Figure S2.** a-e) SEM micrographs showing the morphology exhibited by the crystallites, synthesized using choline hydroxide 1:1 propionic acid and molar ratios of DES (X):1 YBaCu metal nitrates molar ratios use in the reaction, with X values  $20 \leq X \leq 60$ , and X values increasing in tens. a'-e') Histograms showing crystallite size distributions of their respective SEM micrographs.



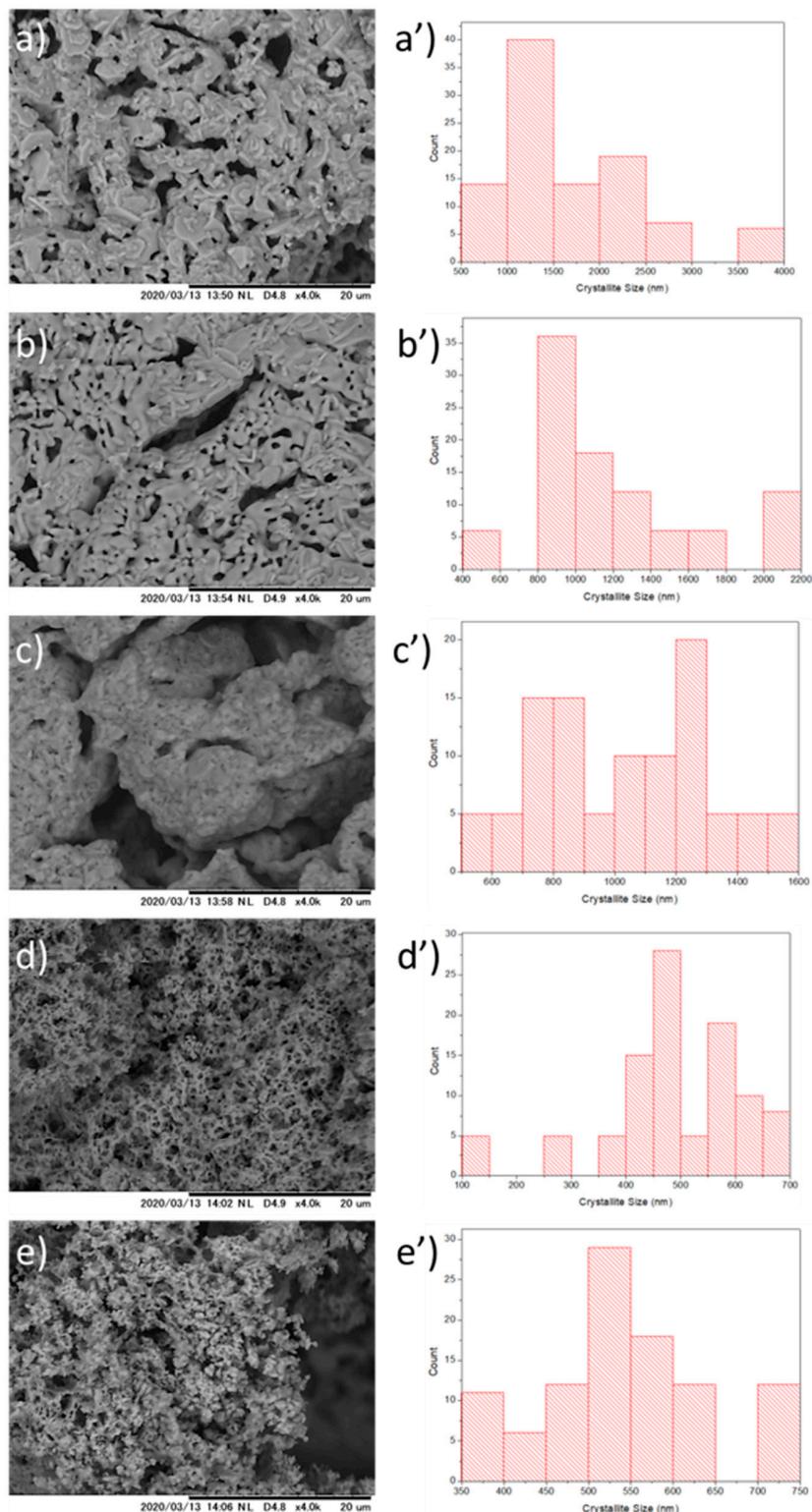
**Figure S3.** a-e) SEM micrographs showing the morphology exhibited by the crystallites, synthesized using choline hydroxide 1:1 butyric acid and molar ratios of DES (X):1 YBaCu metal nitrates molar ratios use in the reaction, with X values  $20 \leq X \leq 60$ , and X values increasing in tens. a'-e') Histograms showing crystallite size distributions of their respective SEM micrographs.



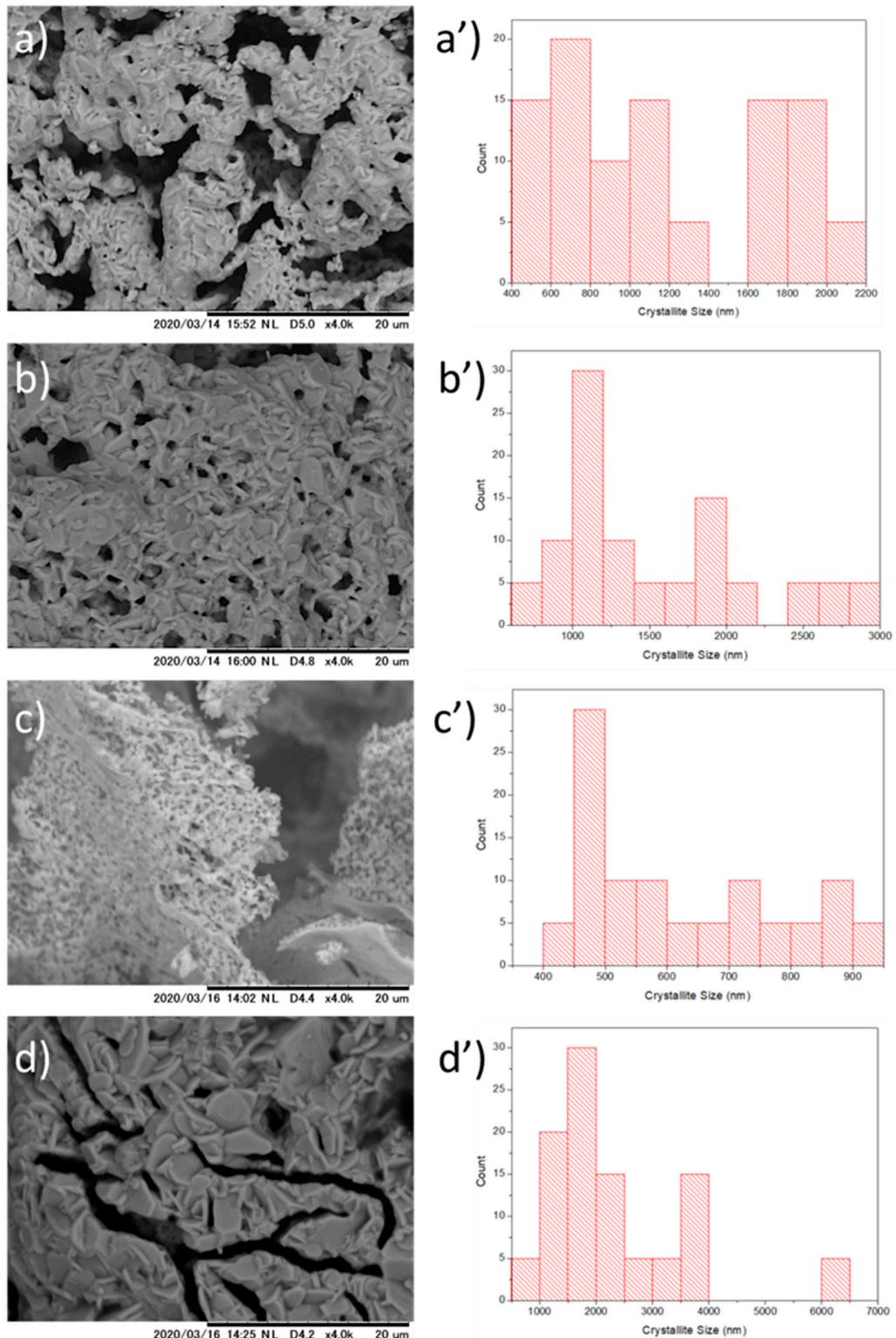
**Figure S4.** a-e) SEM micrographs showing the morphology exhibited by the crystallites, synthesized using choline hydroxide 1:1 valeric acid and molar ratios of DES (X):1 YBaCu metal nitrates molar ratios use in the reaction, with X values  $20 \leq X \leq 60$ , and X values increasing in tens. a'-e') Histograms showing crystallite size distributions of their respective SEM micrographs.



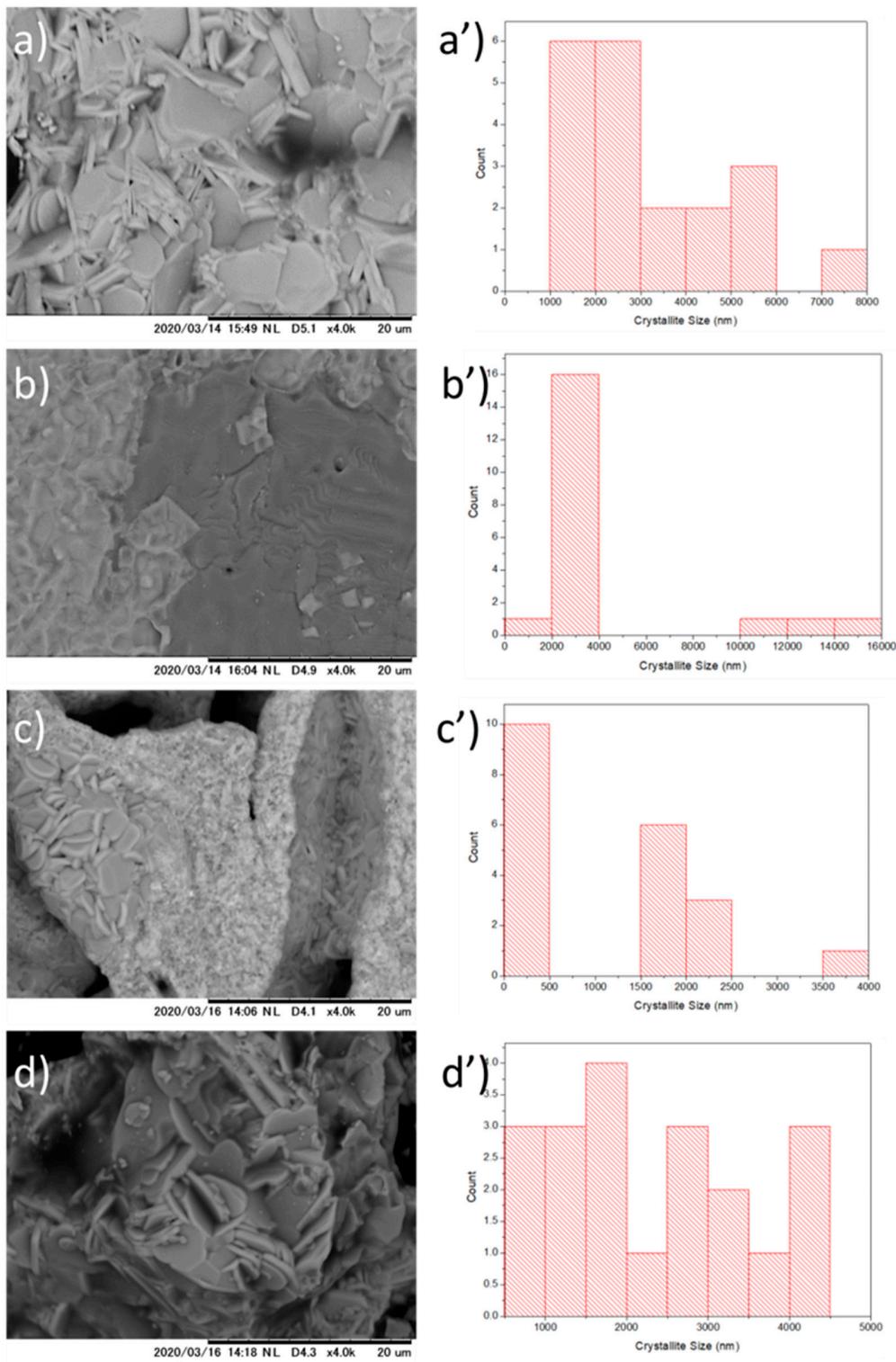
**Figure S5.** a-e) SEM micrographs showing the morphology exhibited by the crystallites, synthesized using choline hydroxide 1:1 nonanoic acid and molar ratios of DES ( $X$ ):1 YBaCu metal nitrates molar ratios use in the reaction, with  $X$  values  $20 \leq X \leq 60$ , and  $X$  values increasing in tens. a'-e') Histograms showing crystallite size distributions of their respective SEM micrographs.



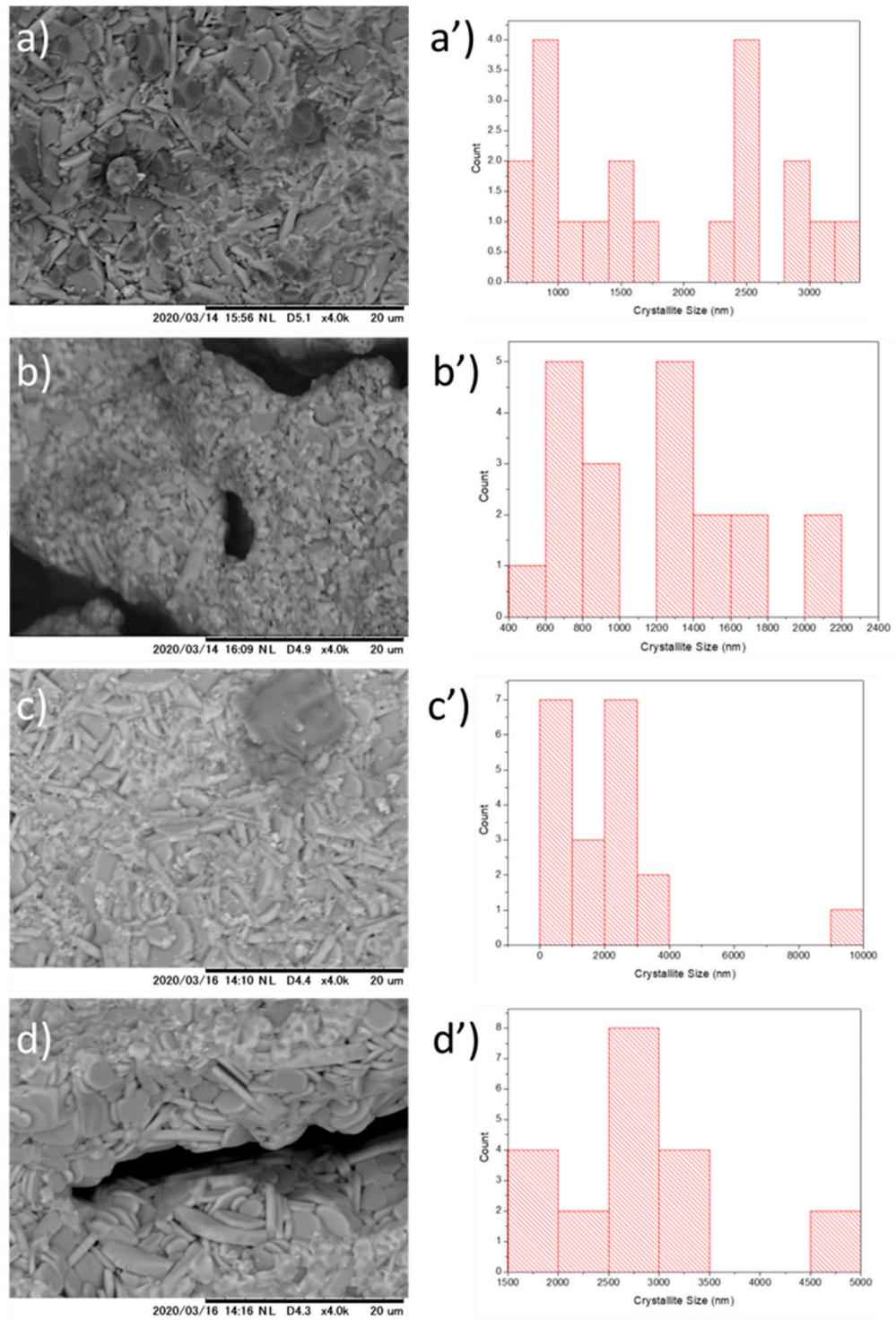
**Figure S6.** a-e) SEM micrographs showing the morphology exhibited by the crystallites, synthesized using choline hydroxide 1:1 decanoic acid and molar ratios of DES (X):1 YBaCu metal nitrates molar ratios use in the reaction, with X values  $20 \leq X \leq 60$ , and X values increasing in tens. a'-e') Histograms showing crystallite size distributions of their respective SEM micrographs.



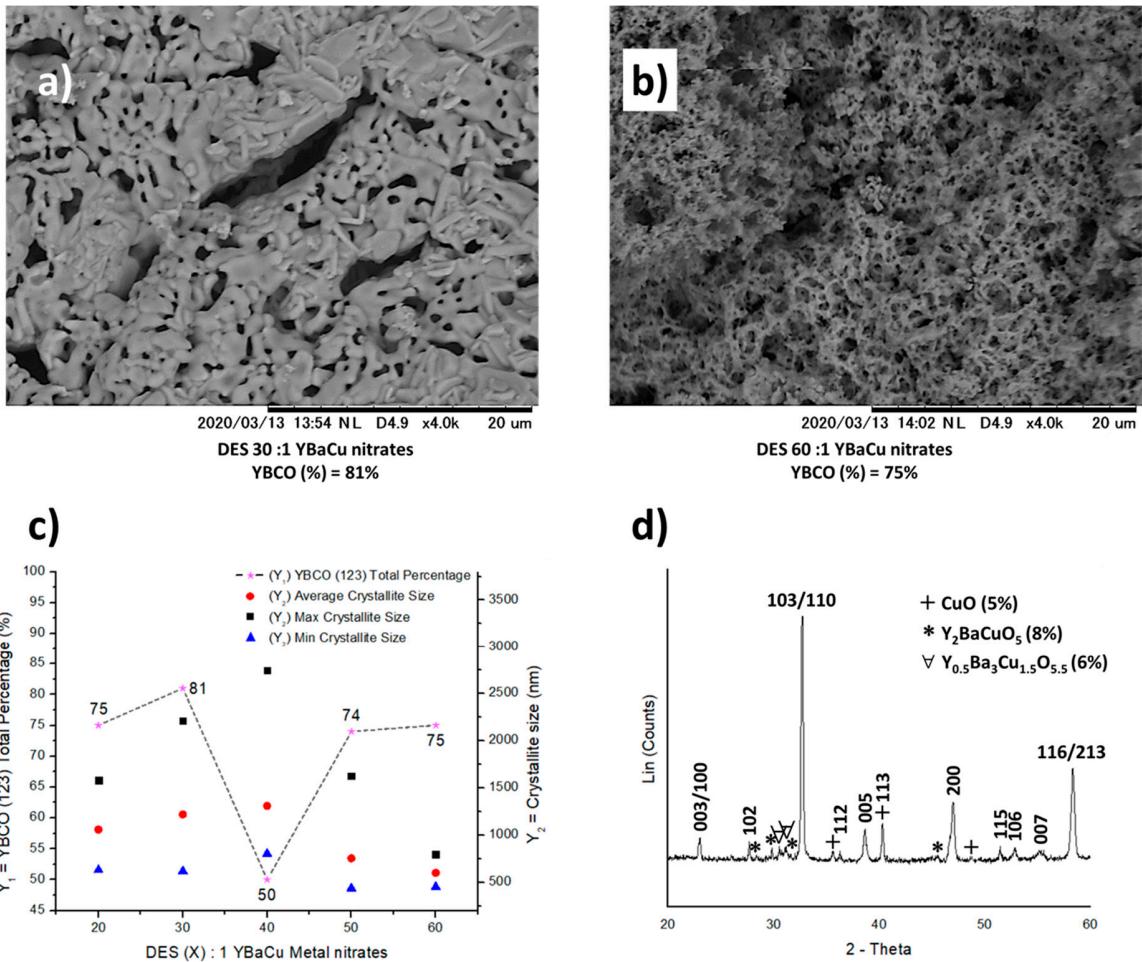
**Figure S7.** a-e) SEM micrographs showing the morphology exhibited by the crystallites, synthesized using choline hydroxide 1:1 acetic acid and molar ratio of DES 60:1 YBaCu metal nitrates, with dehydrating temperatures of a) 150 °C, b) 180 °C, c) 250 °C, and d) 300 °C. a'-d') Histograms showing crystallite size distributions of their respective SEM micrographs.



**Figure S8.** a-e) SEM micrographs showing the morphology exhibited by the crystallites, synthesized using choline hydroxide 1:1 valeric acid and molar ratio of DES 30:1 YBaCu metal nitrates, with dehydrating temperatures of a) 150 °C, b) 180 °C, c) 250 °C, and d) 300 °C. a'-d') Histograms showing crystallite size distributions of their respective SEM micrographs.



**Figure S9.** a-e) SEM micrographs showing the morphology exhibited by the crystallites, synthesized using choline hydroxide 1:1 nonanoic acid and molar ratio of DES 20:1 YBaCu metal nitrates, with dehydrating temperatures of a) 150 °C, b) 180 °C, c) 250 °C, and d) 300 °C. a'-d') Histograms showing crystallite size distributions of their respective SEM micrographs.



**Figure S10.** a & b) SEM micrographs showing the morphology exhibited by the crystallites, synthesized using choline hydroxide 1:1 decanoic acid and molar ratios of a) DES 30:1 YBaCu metal nitrates, and b) DES 60:1 YBaCu metal nitrates. c) graphical representation of the total phase percentage of YBCO (123), and crystallite sizes per DES (X):1 YBaCu metal nitrates molar ratios use in the reaction, with X values  $20 \leq X \leq 60$ . d) indexed power diffraction pattern of the synthesis of YBCO (123) via DES 30:1 YBaCu metal nitrates.

	Acetic acid - Total Phase Percentages (%)				
X =	20	30	40	50	60
<b>YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-d</sub></b>	82	86	86	88	92
<b>CuO</b>	6	3	3	5	5
<b>Y<sub>2</sub>BaCuO<sub>5</sub></b>	8	2	7	0	3
<b>Y<sub>0.5</sub>Ba<sub>3</sub>Cu<sub>1.5</sub>O<sub>5.5</sub></b>	4	9	4	7	0
<b>Total</b>	100	100	100	100	100

**Table S2** Summary of Total phase percentages obtained via Rietveld analysis of the powder diffraction patterns, collected from samples synthesized via choline hydroxide 1:1 acetic acid, DES (X):1 YBaCu metal nitrates molar ratios.

	Propionic acid - Total Phase Percentages (%)				
X =	20	30	40	50	60
<b>YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-d</sub></b>	91	90	87	88	89
<b>CuO</b>	4	5	7	4	5
<b>Y<sub>2</sub>BaCuO<sub>5</sub></b>	5	5	6	3	3
<b>Y<sub>0.5</sub>Ba<sub>3</sub>Cu<sub>1.5</sub>O<sub>5.5</sub></b>	0	0	0	5	3
<b>Total</b>	100	100	100	100	100

**Table S3** Summary of Total phase percentages obtained via Rietveld analysis of the powder diffraction patterns, collected from samples synthesized via choline hydroxide 1:1 propionic acid, DES (X):1 YBaCu metal nitrates molar ratios.

	Butyric acid - Total Phase Percentages (%)				
X =	20	30	40	50	60
<b>YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-d</sub></b>	91	92	91	93	92
<b>CuO</b>	3	4	4	4	5
<b>Y<sub>2</sub>BaCuO<sub>5</sub></b>	6	4	5	2	2
<b>Y<sub>0.5</sub>Ba<sub>3</sub>Cu<sub>1.5</sub>O<sub>5.5</sub></b>	0	0	0	1	1
<b>Total</b>	100	100	100	100	100

**Table S4** Summary of Total phase percentages obtained via Rietveld analysis of the powder diffraction patterns, collected from samples synthesized via choline hydroxide 1:1 butyric acid, DES (X):1 YBaCu metal nitrates molar ratios.

	Valeric acid - Total Phase Percentages (%)				
X =	20	30	40	50	60
<b>YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-d</sub></b>	89	88	88	88	82
<b>CuO</b>	3	4	6	8	7
<b>Y<sub>2</sub>BaCuO<sub>5</sub></b>	5	8	0	0	3
<b>Y<sub>0.5</sub>Ba<sub>3</sub>Cu<sub>1.5</sub>O<sub>5.5</sub></b>	3	0	6	4	8
<b>Total</b>	100	100	100	100	100

**Table S5** Summary of Total phase percentages obtained via Rietveld analysis of the powder diffraction patterns, collected from samples synthesized via choline hydroxide 1:1 valeric acid, DES (X):1 YBaCu metal nitrates molar ratios.

	Nonanoic acid - Total Phase Percentages (%)				
X =	20	30	40	50	60
<b>YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-d</sub></b>	87	89	89	86	81
<b>CuO</b>	5	3	5	6	8
<b>Y<sub>2</sub>BaCuO<sub>5</sub></b>	7	5	6	8	7
<b>Y<sub>0.5</sub>Ba<sub>3</sub>Cu<sub>1.5</sub>O<sub>5.5</sub></b>	1	3	0	0	4
<b>Total</b>	100	100	100	100	100

**Table S6** Summary of Total phase percentages obtained via Rietveld analysis of the powder diffraction patterns, collected from samples synthesized via choline hydroxide 1:1 nonanoic acid, DES (X):1 YBaCu metal nitrates molar ratios.

	Decanoic acid - Total Phase Percentages (%)				
X =	20	30	40	50	60
<b>YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-d</sub></b>	75	81	50	74	75
<b>CuO</b>	8	6	0	7	10
<b>Y<sub>2</sub>BaCuO<sub>5</sub></b>	9	5	14	9	9
<b>Y<sub>0.5</sub>Ba<sub>3</sub>Cu<sub>1.5</sub>O<sub>5.5</sub></b>	8	8	11	10	6
<b>Ba<sub>0.98</sub>CuO<sub>2.07</sub></b>	0	0	25	0	0
<b>Total</b>	100	100	100	100	100

**Table S7** Summary of Total phase percentages obtained via Rietveld analysis of the powder diffraction patterns, collected from samples synthesized via choline hydroxide 1:1 decanoic acid, DES (X):1 YBaCu metal nitrates molar ratios.