

Abstract

Silver Nanoparticles Formation and PVP Crosslinking Using UV-Radiation [†]

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Silver nanoparticles (AgNPs) can be easily incorporated into wound dressings, exhibiting a broad range of antimicrobial activity, without possessing the acute cytotoxicity of ionic silver, and improving wound healing [1].

AgNPs can be produced using physical or chemical methods. In physical methods, energy is supplied to the Ag precursor in the form of heat or radiation, without using toxic reducing agents. Usually, a polymer is used to host the silver nanoparticles and act as stabilizers or surface capping agents. UV irradiation have been used to obtain cellulose acetate nanofibers [2] and polyvinylpyrrolidone (PVP) nanofibers [3] containing AgNPs on their surfaces.

In this work, we optimized the production of AgNPs using PVP as a stabilizing agent, AgNO₃ as the Ag source, and UV radiation as source of energy to produce the AgNPs and simultaneously to crosslink the PVP polymer.

Different parameters were evaluated: PVP concentration (16, 18, 20, 22 and 24 wt.%); water to ethanol solvent ratio (1:3, 1:1 and 3:1); PVP to AgNO₃ ratio (400:1, 200:1, 100:1 and 50:1); and irradiation time (15, 30, 60, 90, 120 and 150 min). The UV-Vis spectrum of each solution was acquired to evaluate the intensity of the absorption peak at 420 nm regarding the surface plasmon resonance of the AgNPs.

The results reveal that different conditions were able to induce the production of stable AgNPs but the highest production rate was obtained from the solution with 20 wt.% PVP dissolved in 1:1 water to ethanol ratio, with PVP to AgNO₃ ratio of 200:1 irradiated with UV. In addition, the obtained irradiated solution has the characteristics (viscosity and conductivity) adequate to be used in the electrospinning process to produce an antibacterial fibrous mat that can be used as wound dressing.

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